

Supporting Information

Fe-MIL Tuned and Binded by Bi₄O₅Br₂ for Boosting Photocatalytic Reduction of CO₂ to CH₄ under Simulated Sunlight

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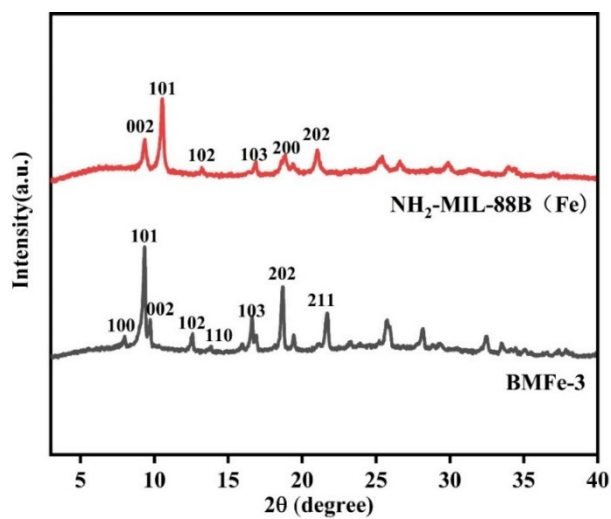


Fig. S1. XRD patterns of $\text{NH}_2\text{-MIL-88B(Fe)}$ and BMFe-3 .

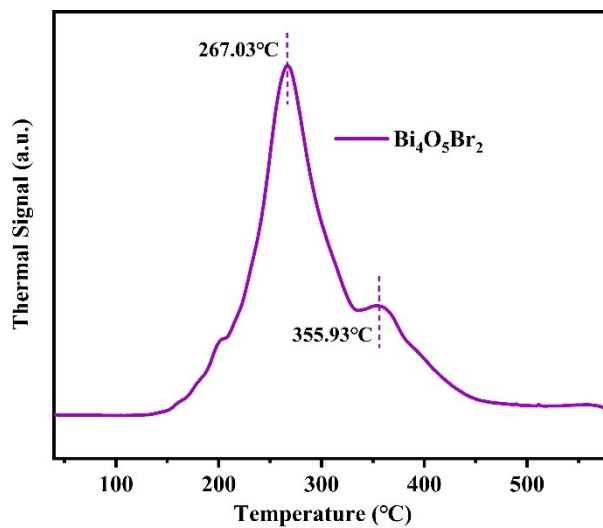


Fig. S2. NH_3 -TPD analysis of $\text{Bi}_4\text{O}_5\text{Br}_2$

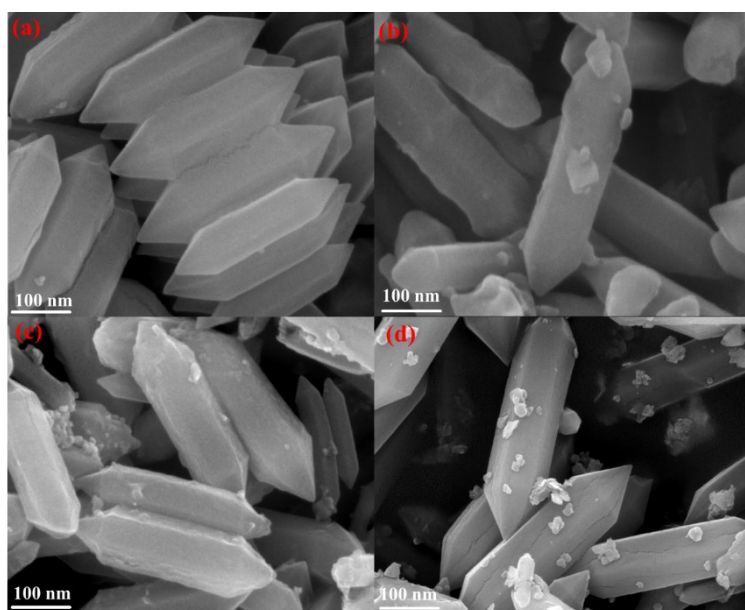


Fig. S3. SEM images of (a) BMFe-3 (b) BMFe-4(c) BMFe-5 (d) BMFe-6

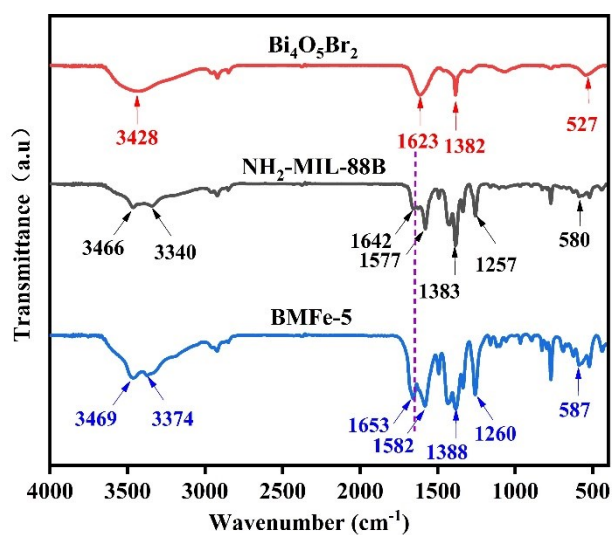


Fig. S4. FT-IR spectra of $\text{Bi}_4\text{O}_5\text{Br}_2$, $\text{NH}_2\text{-MIL-88B(Fe)}$ and BMFe-5.

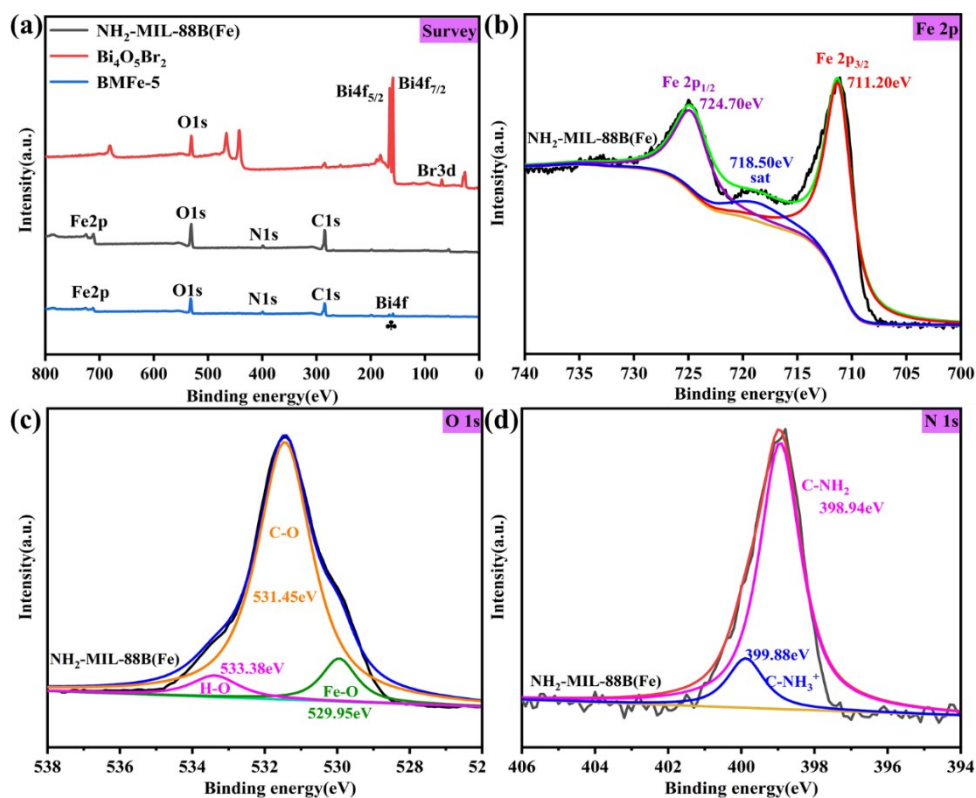


Fig. S5. XPS spectra: (a) Survey scan of $\text{NH}_2\text{-MIL-88B(Fe)}$, $\text{Bi}_4\text{O}_5\text{Br}_2$ and BMFe-5 , (b) Fe 2p, (c) O 1s and (d) N 1s of $\text{NH}_2\text{-MIL-88B(Fe)}$.

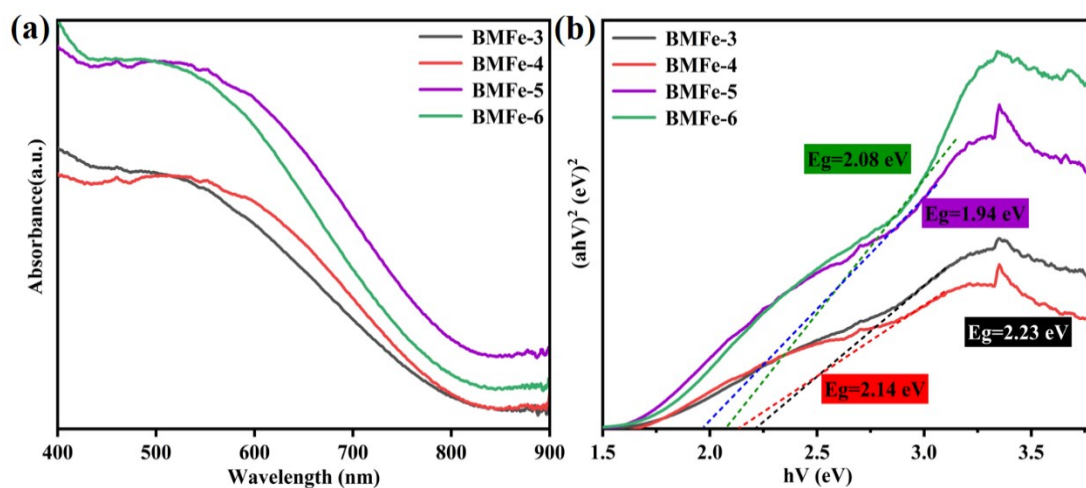


Fig. S6. (a) The UV-vis DRS and (b) the plot of $(ah\nu)^2$ vs photo energy ($h\nu$) of BMFe-X ($X=3, 4, 5, 6$).

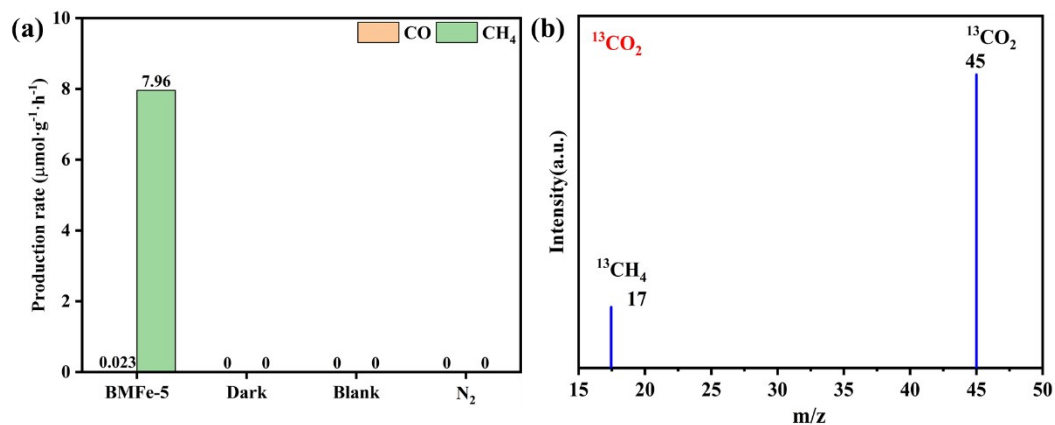


Fig. S7. (a) The control experiment of BMFe-5 under different reaction condition, (b) GC-MS analysis for the control experiment of $^{13}\text{CO}_2$

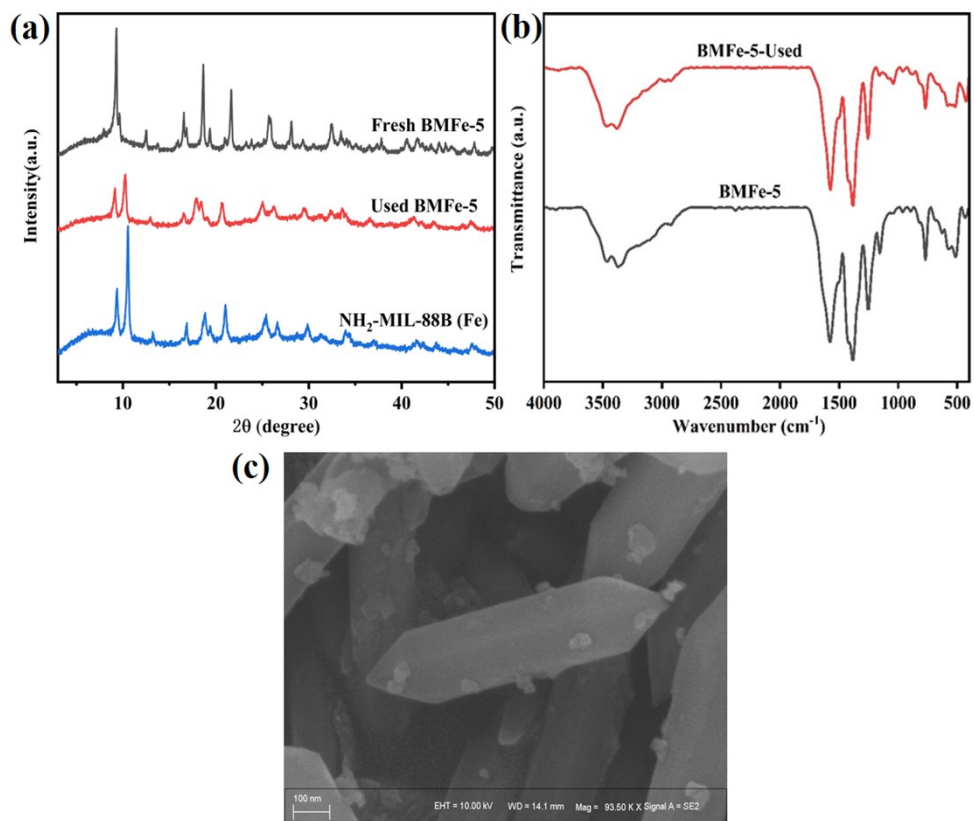


Fig. S8. (a) XRD, (b) FTIR patterns and (c) SEM image of BMFe-5 after four recycling tests

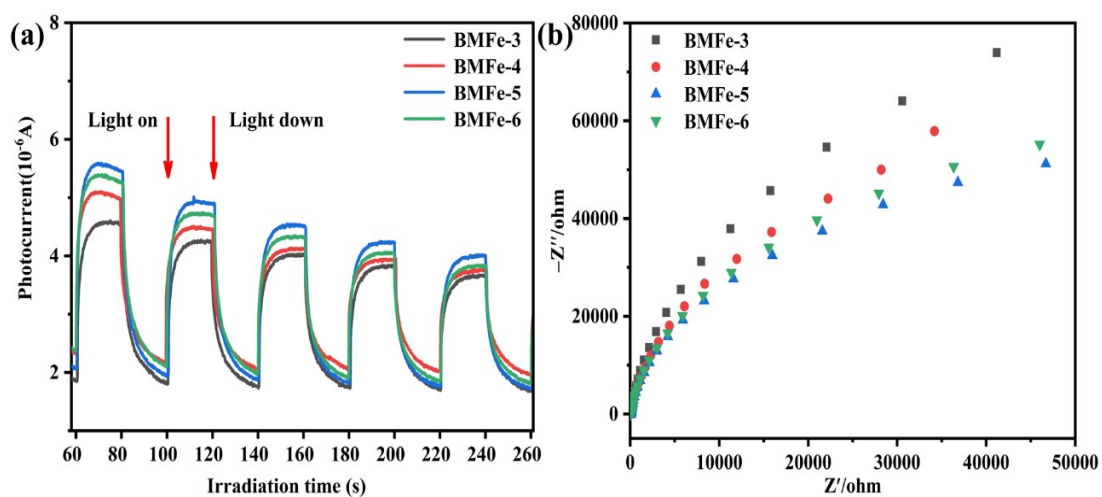


Fig. S9. (a) transient photocurrent responses and (b) EIS Nyquist plots of BMFe-X (X=3, 4, 5, 6).

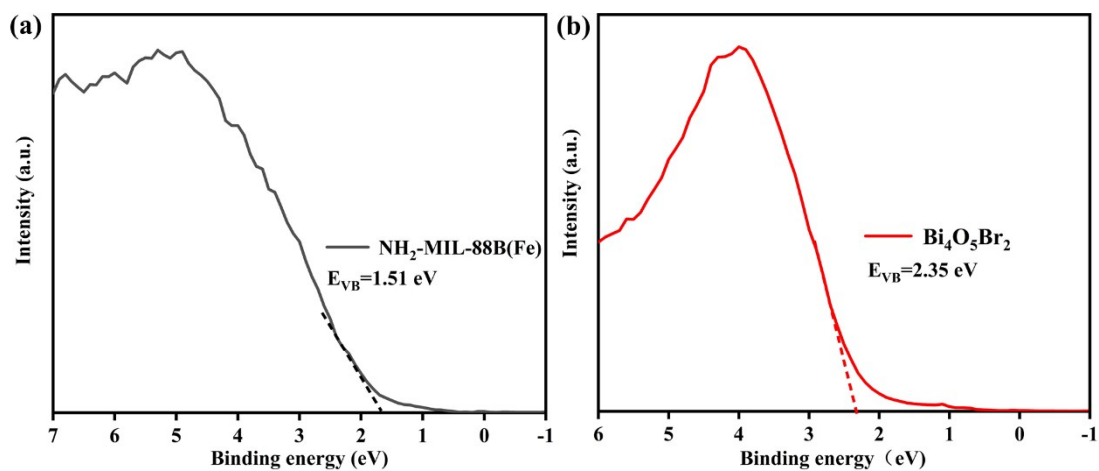


Fig. S10. The XPS valence band spectrum of (a) NH₂-MIL-88B(Fe) and (b) Bi₄O₅Br₂.

Table S1. The content of Bi₄O₅Br₂ in BMFe-X from ICP-MS test

Samples	The content of Bi ₄ O ₅ Br ₂ (wt%)
BMFe-1	0.04
BMFe-2	0.06
BMFe-3	0.07
BMFe-4	0.40
BMFe-5	3.28
BMFe-6	8.93

Table S2. The textural properties of the samples

Samples	S _{BET} (m ² /g)	Total Pore Volume (cm ³ /g)	Pore Size (nm)
NH ₂ -MIL-88(Fe)	72.39	0.127	7.01
Bi ₄ O ₅ Br ₂	148.72	0.358	9.63
BMFe-5	21.15	0.052	9.82

Table S3 Comparisons of the amount of CO and CH₄ produced over different photocatalyst

Photocatalyst	The amount of CO ($\mu\text{mol}\cdot\text{g}^{-1}\cdot\text{h}^{-1}$)	The amount of CH ₄ ($\mu\text{mol}\cdot\text{g}^{-1}\cdot\text{h}^{-1}$)	References
This work	0.023	7.96	--
CdS/ZIF-8	32.13	--	S1
UiO-66/CNNS	9.90	--	S2
Ni ₃ HITP ₂ /rGO	12.82	--	S3
Co-MOF/g-C ₃ N ₄	6.82	4.22	S4
Fe ³⁺ -doped TiO ₂	--	0.23	S5
Hollow Bi ₄ O ₅ Br ₂	3.16	0.5	S6
Ultrathin Bi ₄ O ₅ Br ₂	31.56	--	S7
Bi ₂ WO ₆ -OV/BiOI	--	2.29	S8
TiO ₂ /NH ₂ -MIL-125(Ti)	--	1.18	S9
CsPbBr ₃ QDs/UiO-66(NH ₂)	8.21	0.26	S10

Table S4. The formation rates of CH₄ and CO, and selectivity of CH₄

Photocatalyst	Formation rate of CH ₄ ($\mu\text{mol}\cdot\text{g}^{-1}\cdot\text{h}^{-1}$)	Formation rate of CO ($\mu\text{mol}\cdot\text{g}^{-1}\cdot\text{h}^{-1}$)	Selectivity of CH ₄ ^a
NH ₂ -MIL-88(Fe)	0.05	0.02	71.43%
Bi ₄ O ₅ Br ₂	0.42	0.471	47.19%
BMFe-3	4.21	0.013	99.62%
BMFe-4	6.38	0.021	99.79%
BMFe-5	7.96	0.023	99.74%
BMFe-6	7.51	0.018	99.76%

^a Selectivity of CH₄ = (formation rate of CH₄) / (formation rate of CH₄ + formation rate of CO).

Table S5 The Band gap (E_g), conduction band (E_{CB}), valence band (E_{VB}) and XPS valence band of NH₂-MIL-88B(Fe) and Bi₄O₅Br₂.

Samples	E _g (eV)	E _{CB} (V)	E _{VB} (V)	E _{VB-XPS} (eV)
Bi ₄ O ₅ Br ₂	2.68	-0.28	2.40	2.35
NH ₂ -MIL-88B(Fe)	1.86	-0.34	1.52	1.51

References

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