

Homo- and Heteroleptic trimethoxy terpyridine-Cu(II) complexes: Synthesis, Characterization, DNA/BSA Binding, DNA Cleavage and Cytotoxicity Studies

Surbhi Jain^a, Kishalay Bhar^a, Sandeep Kumar^b, Shreetama Bandyopadhyaya^c, Suman Tapryal^b, Chandi C. Mandal^c, Anuj K. Sharma^{a*}

^aDepartment of Chemistry, School of Chemical Sciences and Pharmacy, Central University of Rajasthan, Bandarsindri, District Ajmer, Rajasthan, 305817, India.

^bDepartment of Biotechnology, School of Chemical Sciences and Pharmacy, Central University of Rajasthan, Bandarsindri, District Ajmer, Rajasthan, 305817, India.

^cDepartment of Biochemistry, School of Chemical Sciences and Pharmacy, Central University of Rajasthan, Bandarsindri, District Ajmer, Rajasthan, 305817, India.

Supporting Information

Contents		
1.	Figure S1.	Data and graph for elemental analysis of C-I .
2.	Figure S2.	Data and graph for elemental analysis of C-II .
3.	Figure S3.	Data and graph for elemental analysis of C-III .
4.	Figure S4.	Data and graph for elemental analysis of C-IV .
5.	Figure S5.	FTIR spectrum of C-I .
6.	Figure S6.	FTIR spectrum of C-II .
7.	Figure S7.	FTIR spectrum of C-III .
8.	Figure S8.	FTIR spectrum of C-IV .
9.	Figure S9.	¹ H NMR spectrum of L in CDCl ₃ .
10.	Figure S10.	¹³ C NMR spectrum of ligand L in CDCl ₃ .
11.	Figure S11.	ESI-Mass spectrum of C-I recorded in MeCN.
12.	Figure S12.	ESI-Mass spectrum of C-II recorded in MeCN.
13.	Figure S13.	ESI-Mass spectrum of C-III recorded in MeCN.
14.	Figure S14.	UV-vis absorption spectra of L in MeCN.
15.	Figure S15.	UV- Visible absorption spectra of C-I recorded in MeCN and MeCN:PBS buffer (1:1) upto 48 h at 298 K.
16.	Figure S16.	UV- Visible absorption spectra of C-II recorded in MeCN and MeCN:PBS buffer (1:1) upto 48 h at 298 K.
17.	Figure S17.	UV- Visible absorption spectra of C-III recorded in MeCN and MeCN:PBS buffer (1:1) upto 48 h at 298 K.
18.	Figure S18.	UV- Visible absorption spectra of C-IV recorded in MeCN and MeCN:PBS buffer (1:1) upto 48 h at 298 K.
19.	Figure S19.	Crystal packing of C-I showing π - π interaction and H-bonding.
20.	Figure S20.	Crystal packing of C-II showing H-bonding.
21.	Figure S21.	Crystal packing of C-IV showing π - π interaction and H-bonding.
22.	Figure S22.	X-band EPR spectrum of complex C-II in MeCN glass at 77 K (frequency 9.1 GHz and 100 kHz field).
23.	Figure S23.	X-band EPR spectrum of complex C-III in MeCN glass at 77 K (frequency 9.1 GHz and 100 kHz field).
24.	Figure S24.	X-band EPR spectrum of complex C-IV in MeCN glass at 77 K (frequency 9.1 GHz and 100 kHz field).
22.	Figure S25.	Cyclic voltammogram of C-I (1 mM) solution in dry CH ₃ CN and 0.1 M TBAP as supporting electrolyte vs. Ag wire at varying scan rates.

23.	Figure S26.	Cyclic voltammogram of C-II (1 mM) solution in dry CH ₃ CN and 0.1 M TBAP as supporting electrolyte vs. Ag wire at varying scan rates.
24.	Figure S27.	Cyclic voltammogram of C-III (1 mM) solution in dry CH ₃ CN and 0.1 M TBAP as supporting electrolyte vs. Ag wire at varying scan rates.
25.	Figure S28.	Cyclic voltammogram of C-IV (1 mM) solution in dry CH ₃ CN and 0.1 M TBAP as supporting electrolyte vs. Ag wire at varying scan rates.
26.	Figure S29.	Plot of fluorescence emission intensity <i>I</i> versus wavelength λ for DNA-EtBr at different concentrations of C-II . The arrow shows the change in intensity of emission on increasing amount of the complex from 0 to 50 μ M.
27.	Figure S30.	Plot of fluorescence emission intensity <i>I</i> versus wavelength λ for DNA-EtBr at different concentrations of C-III . The arrow shows the change in intensity of emission on increasing amount of the complex from 0 to 50 μ M.
28.	Figure S31.	Plot of fluorescence emission intensity <i>I</i> versus wavelength λ for DNA-EtBr at different concentrations of C-IV . The arrow shows the change in intensity of emission on increasing amount of the complex from 0 to 50 μ M.
29.	Figure S32.	Plot of fluorescence emission intensity <i>I</i> versus wavelength λ for BSA at different concentrations of C-II . The arrow shows the change in intensity of emission on increasing amount of the complex from 0 to around 50 μ M.
30.	Figure S33.	Plot of fluorescence emission intensity <i>I</i> versus wavelength λ for BSA at different concentrations of C-III . The arrow shows the change in intensity of emission on increasing amount of the complex from 0 to around 50 μ M.
31.	Figure S34.	Plot of fluorescence emission intensity <i>I</i> versus wavelength λ for BSA at different concentrations of C-IV . The arrow shows the change in intensity of emission on increasing amount of the complex from 0 to around 50 μ M.
32.	Figure S35.	Agarose gel electrophoresis for complexes C-I to C-IV showing DNA aggregation in presence of H ₂ O ₂ .
33.	Figure S36.	Agarose gel electrophoresis for terpyridine ligand and co-ligands in presence of 3-MPA.
34.	Figure S37.	
35.	Table S1.	Consolidated characterization data for copper(II) complexes (C-I - C-IV)
36.	Table S2.	Summary of crystal data for complexes C-I , C-II and C-IV
37.	Table S3.	Selected bond lengths (\AA) and bond angles ($^\circ$) for complexes C-I , C-II and C-IV
38.	Table S4.	Summary of IC ₅₀ values for similar Cu(II) complexes in cancer cell line.

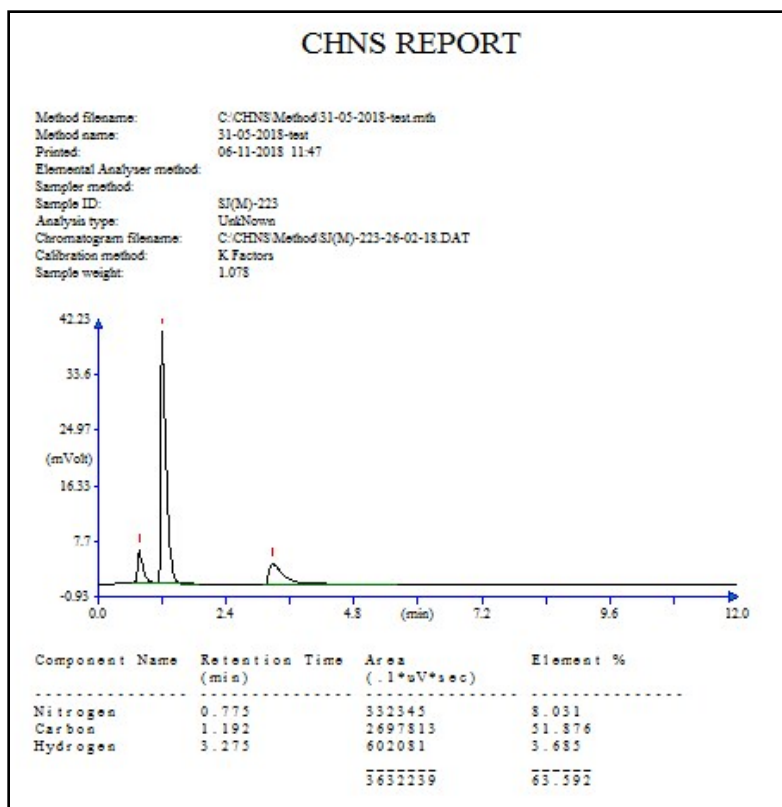


Figure S1. Data and graph for elemental analysis of C-I.

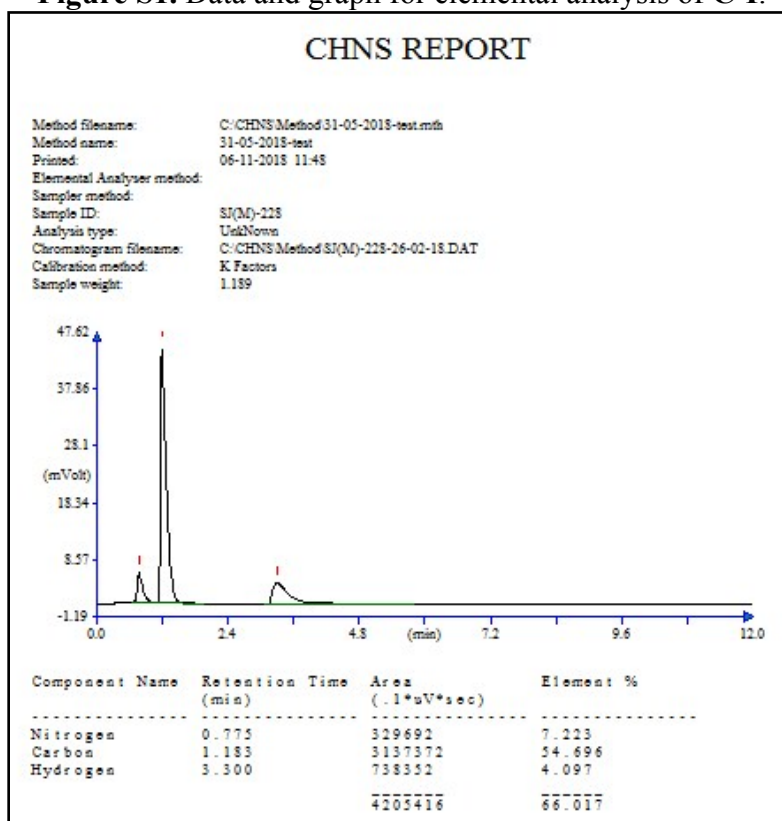


Figure S2. Data and graph for elemental analysis of C-II.

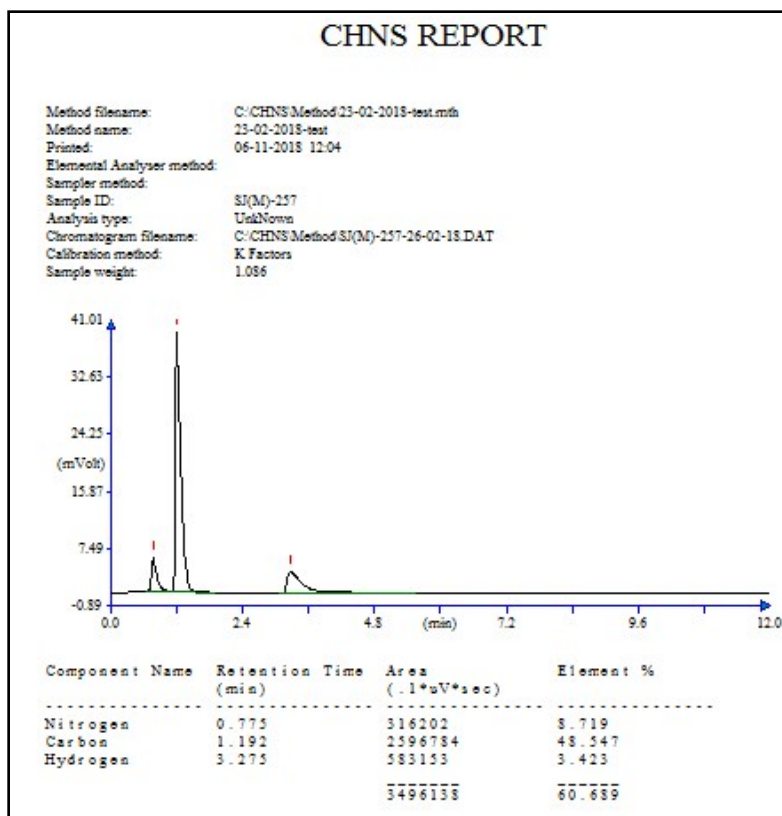


Figure S3. Data and graph for elemental analysis of C-III.

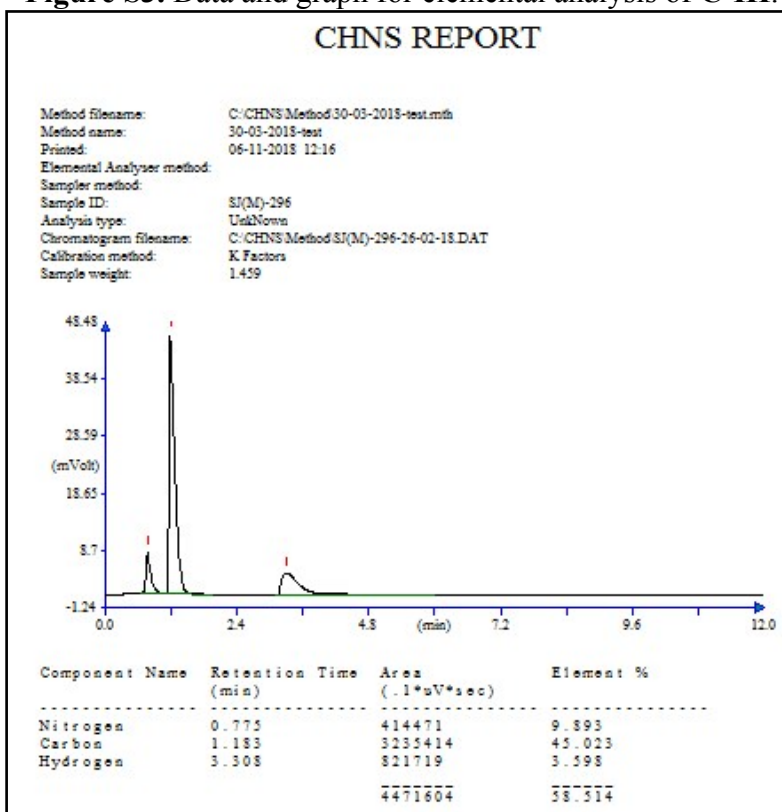


Figure S4. Data and graph for elemental analysis of C-IV

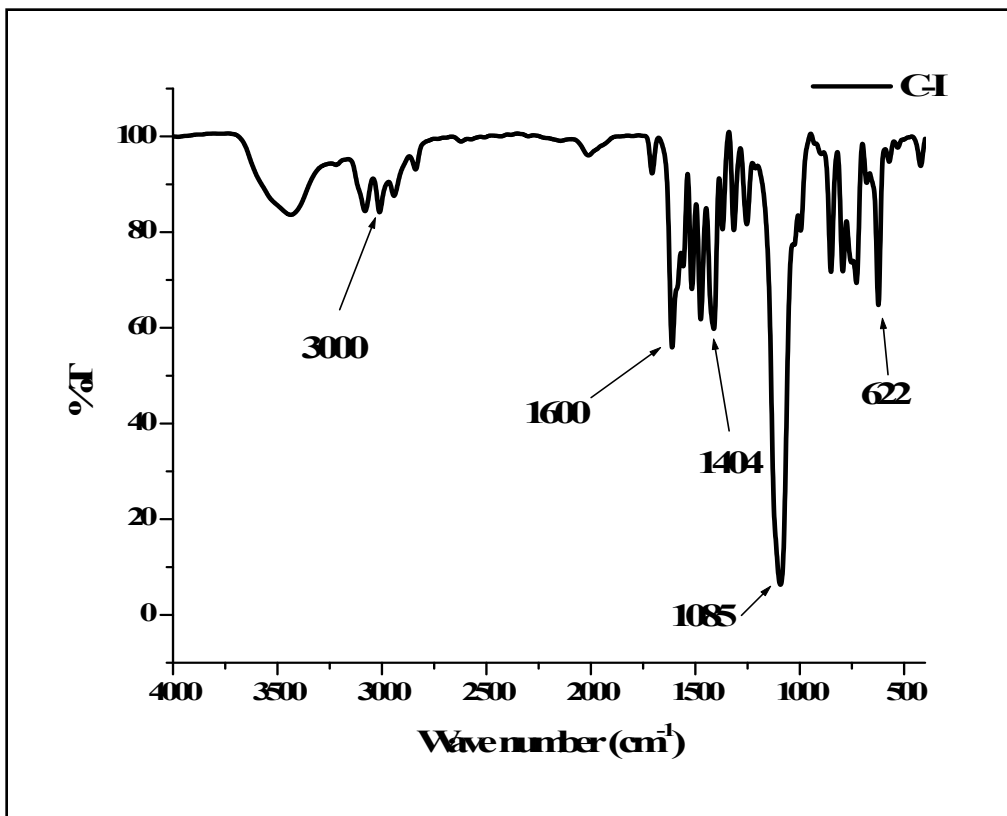


Figure S5. IR spectrum of C-I.

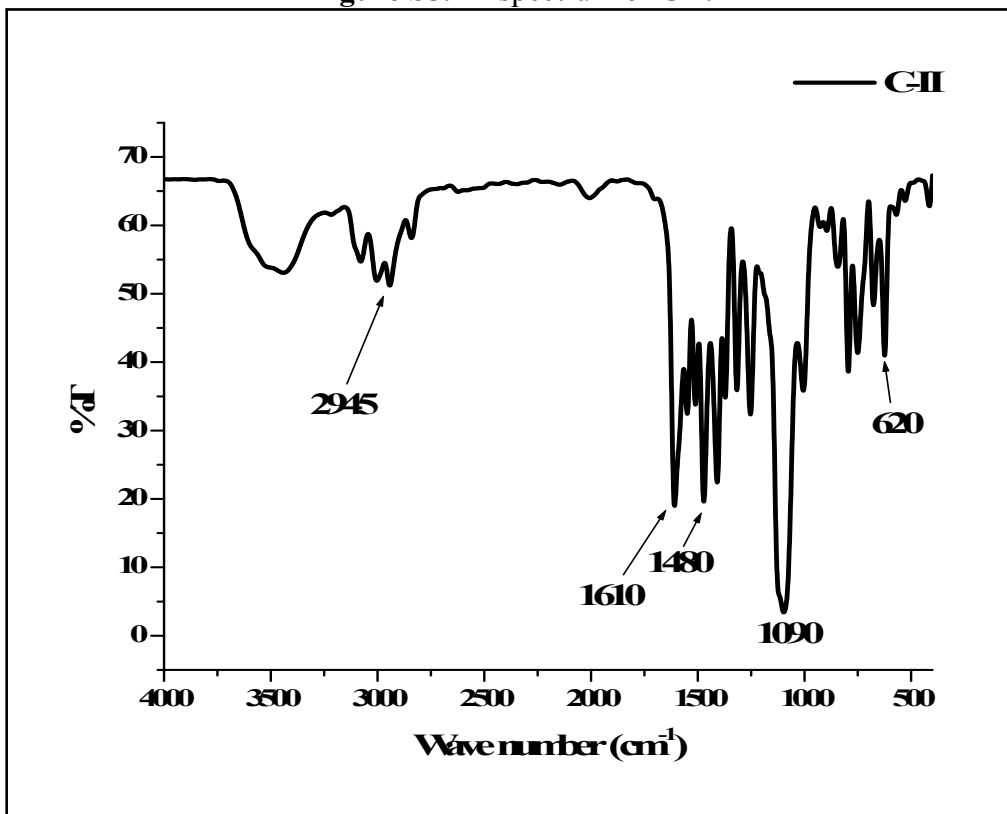


Figure S6. IR spectrum of C-II.

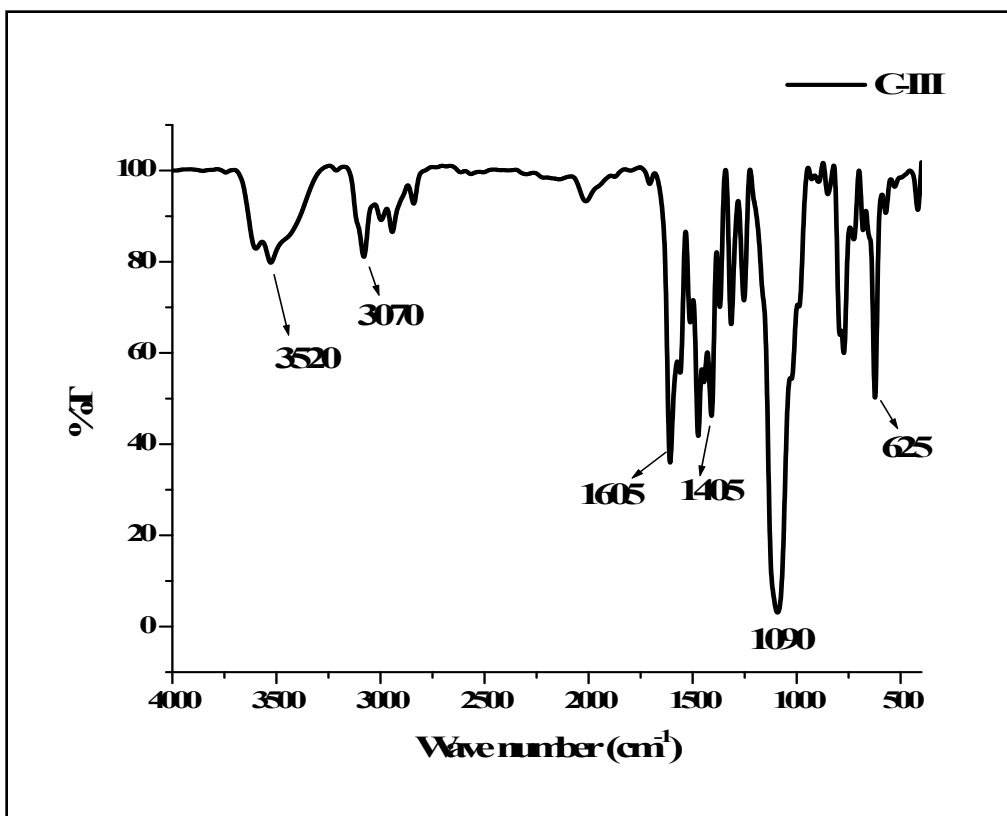


Figure S7. IR spectrum of C-III.

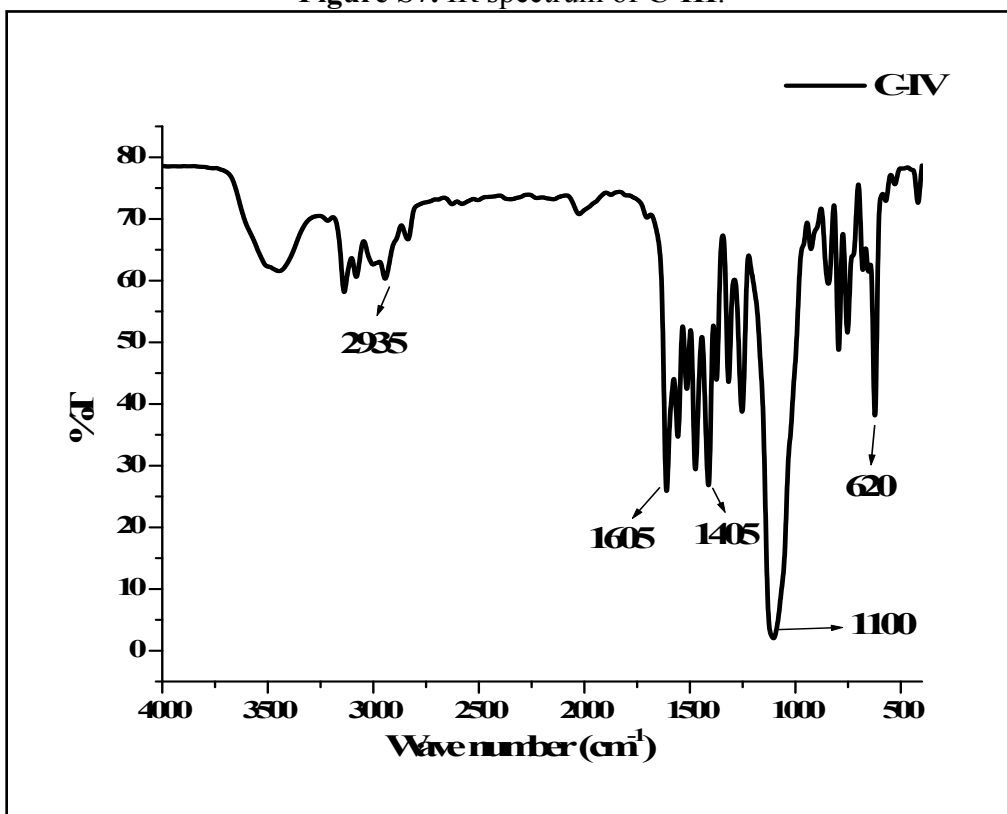


Figure S8. IR spectrum of C-IV.

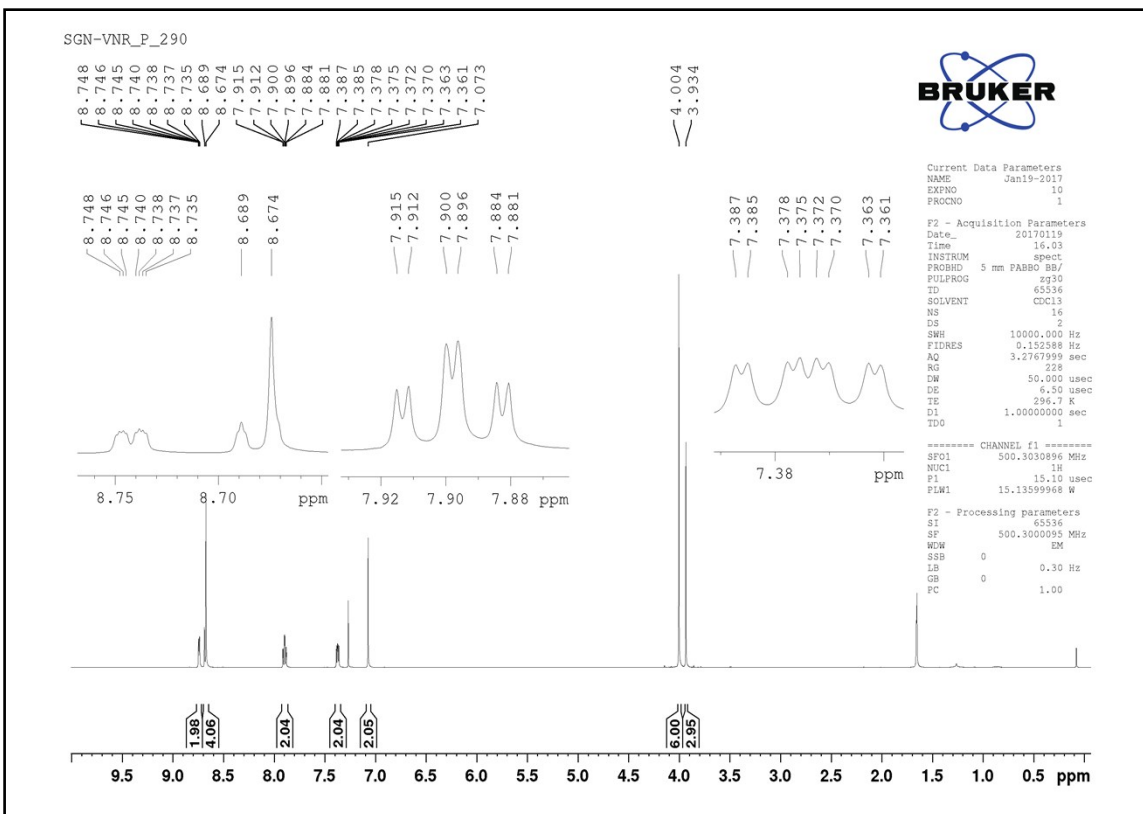


Figure S9. ^1H NMR spectrum of ligand L in CDCl_3 .

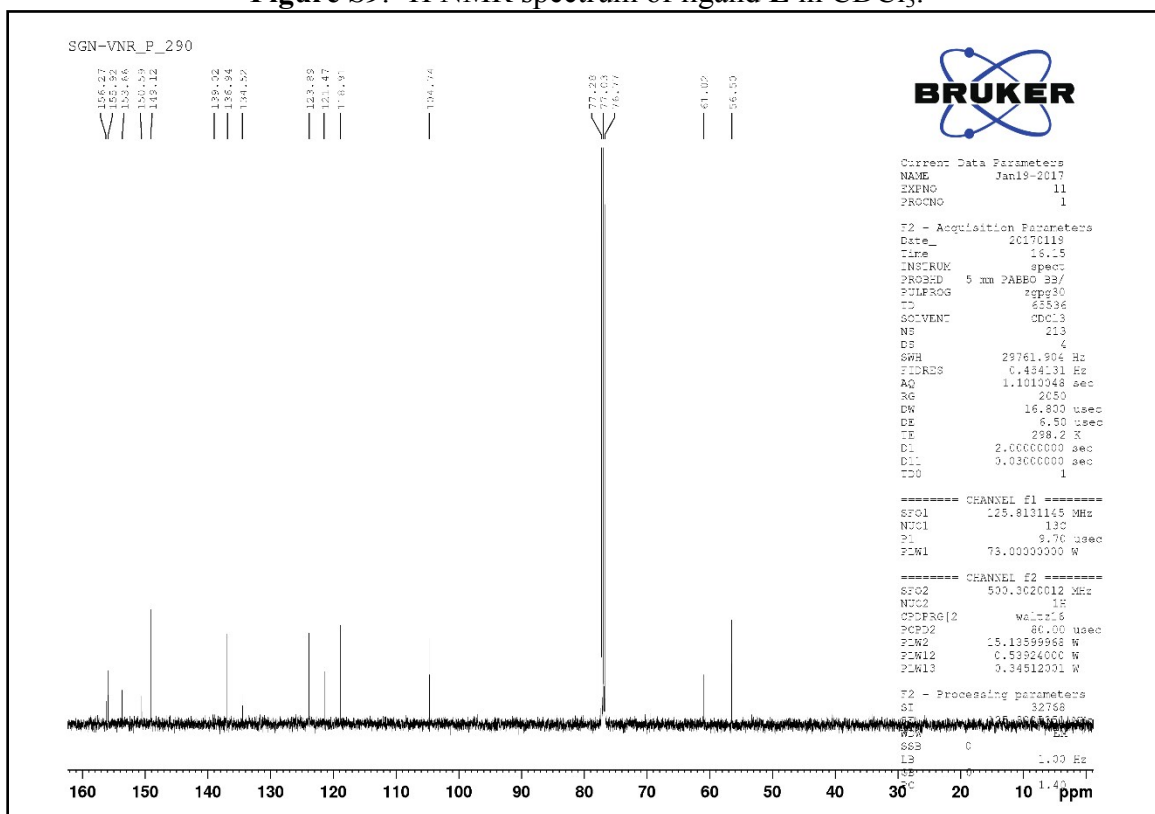


Figure S10. ^{13}C NMR spectrum of ligand L in CDCl_3 .

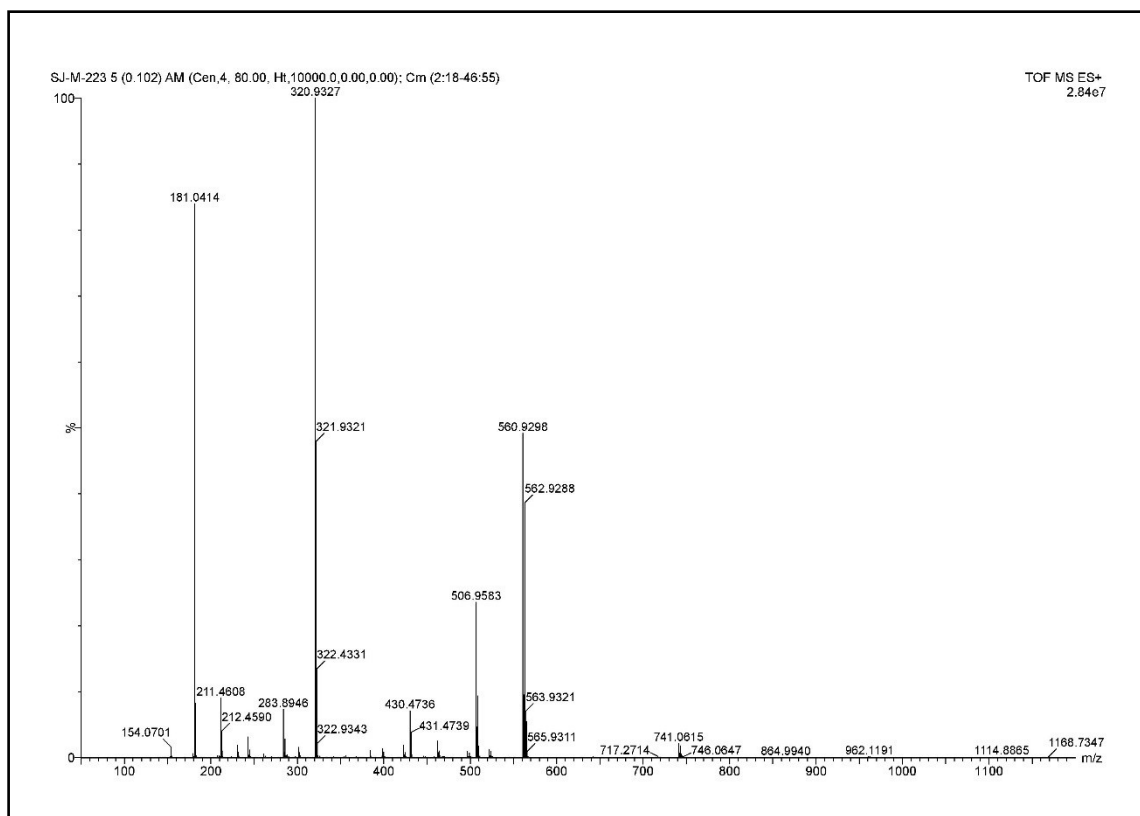


Figure S11. ESI-Mass spectrum of **C-I** recorded in MeCN.

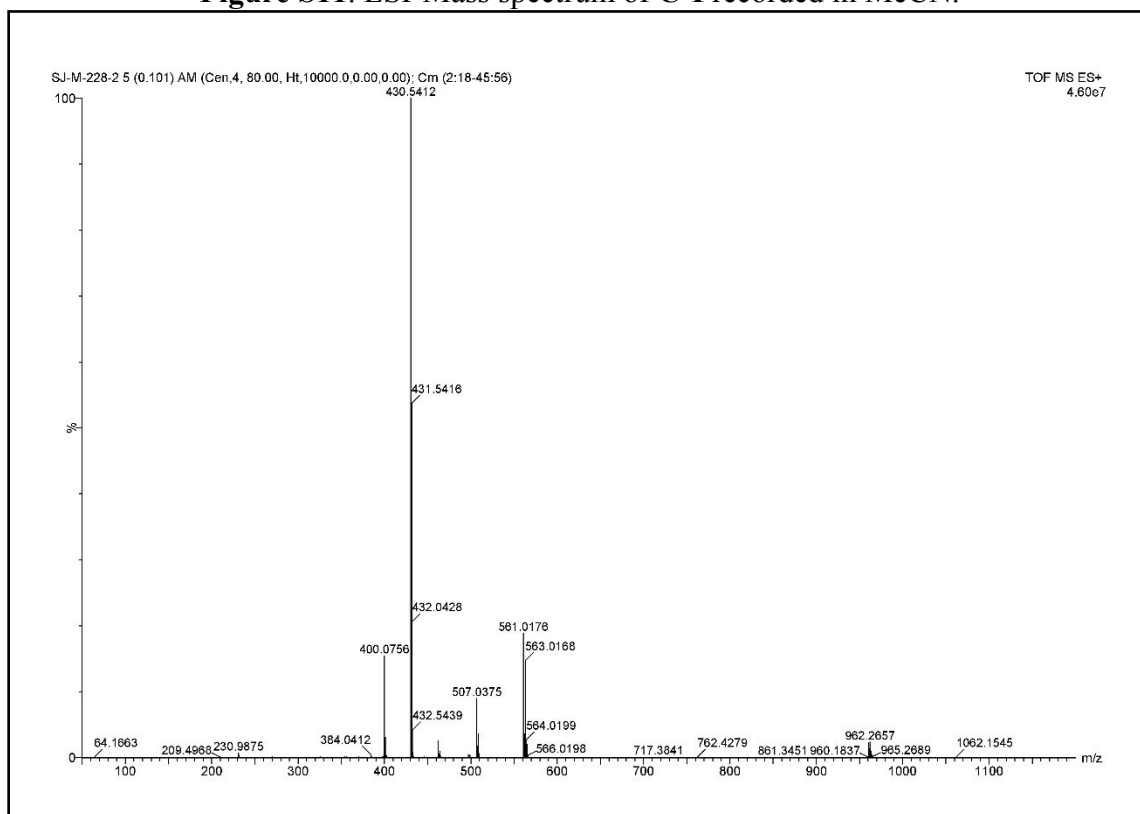


Figure S12. ESI-Mass spectrum of **C-II** recorded in MeCN.

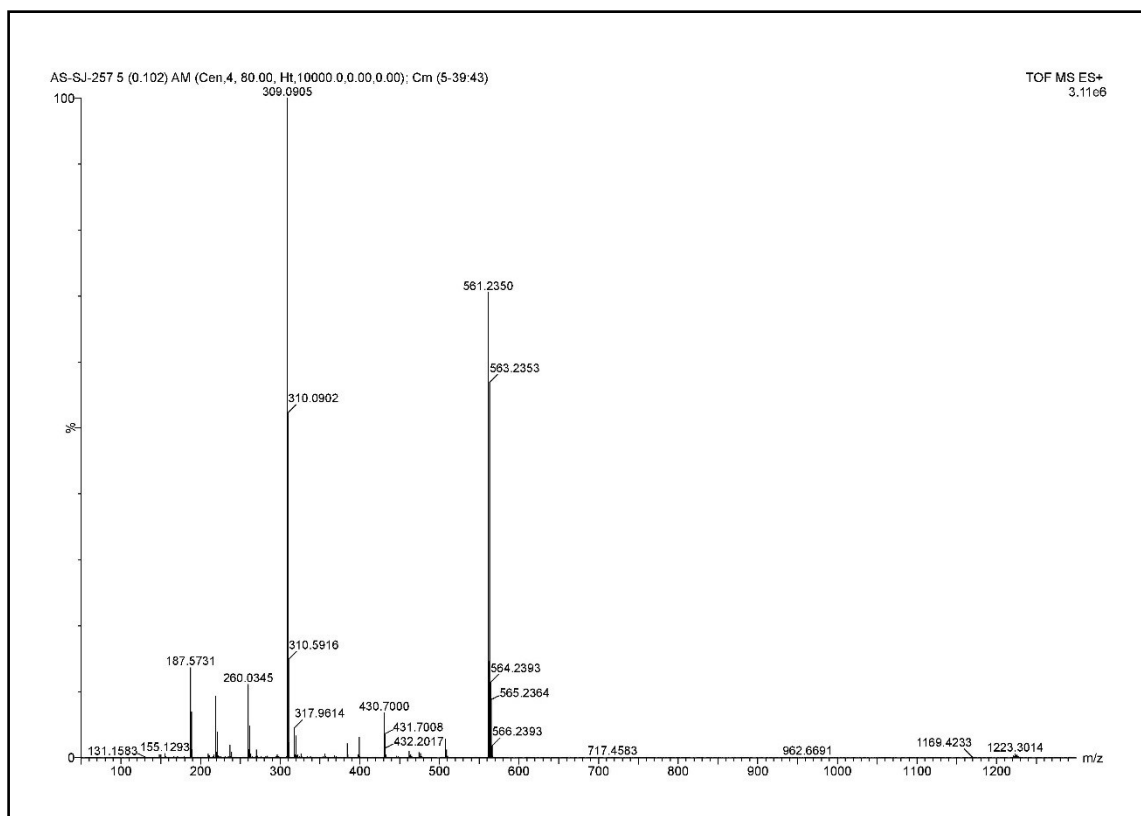


Figure S13. ESI-Mass spectrum of C-III recorded in MeCN.

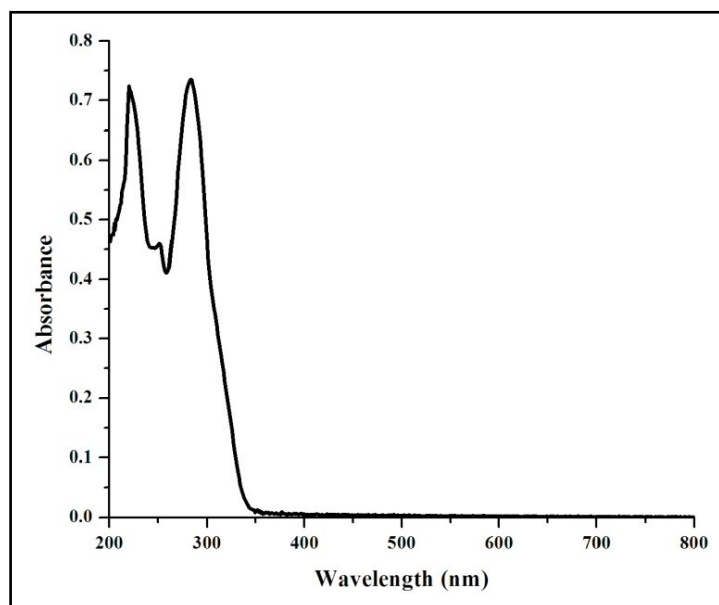


Figure S14. UV-Vis spectrum of L (20 μ M) in MeCN

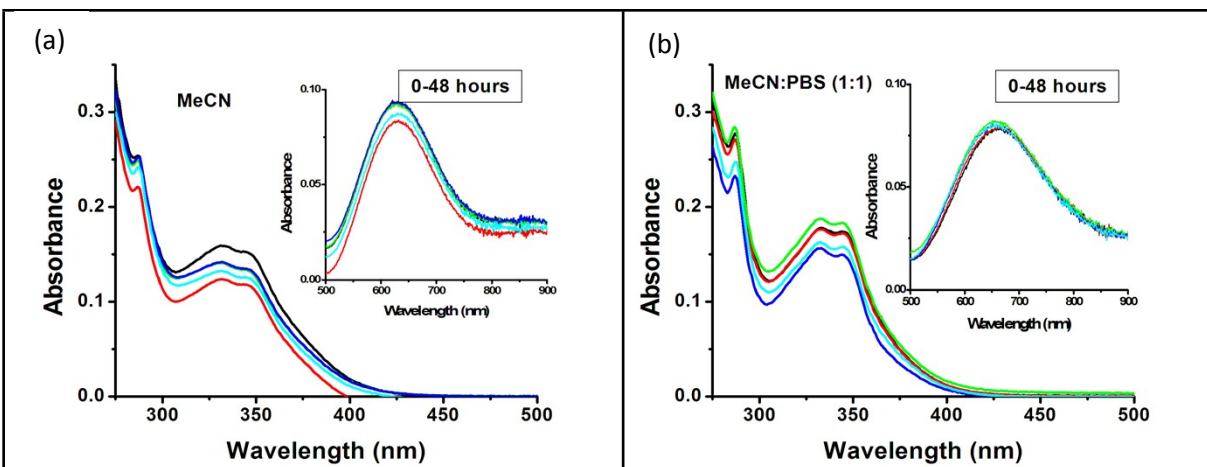


Figure S15. UV- Visible absorption spectra of C-I recorded in (a) MeCN and; (b) MeCN:PBS buffer (1:1) upto 48 h at 298 K.

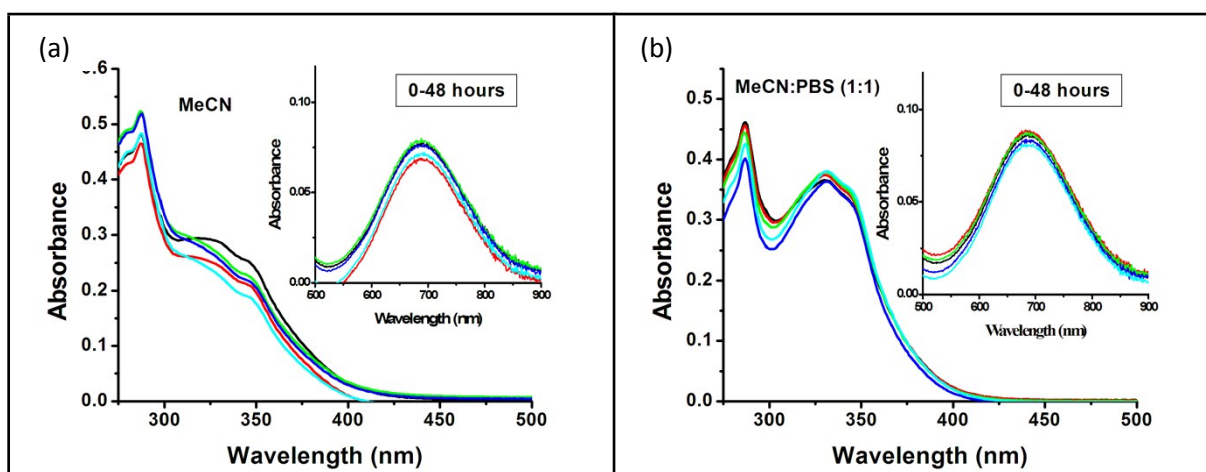


Figure S16. UV- Visible absorption spectra of C-II recorded in (a) MeCN and; (b) MeCN:PBS buffer (1:1) upto 48 h at 298 K.

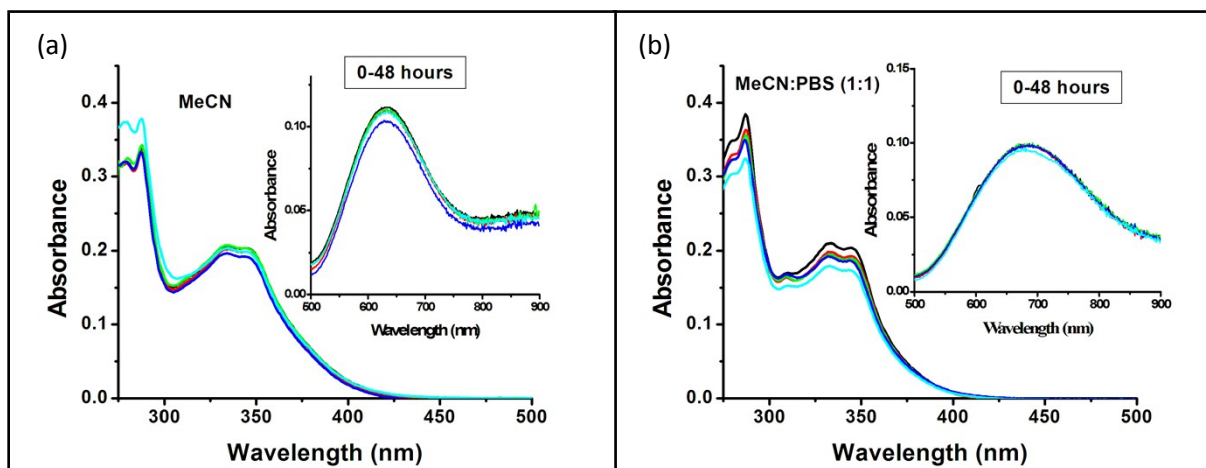


Figure S17. UV- Visible absorption spectra of C-III recorded in (a) MeCN and; (b) MeCN:PBS buffer (1:1) upto 48 h at 298 K.

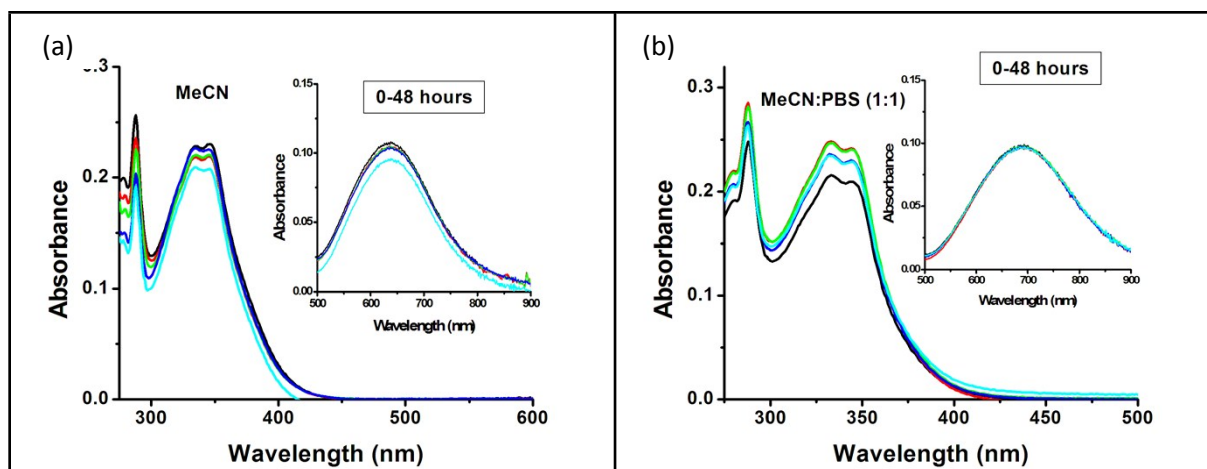


Figure S18. UV- Visible absorption spectra of C-IV recorded in (a) MeCN and; (b) MeCN:PBS buffer (1:1) upto 48 h at 298 K

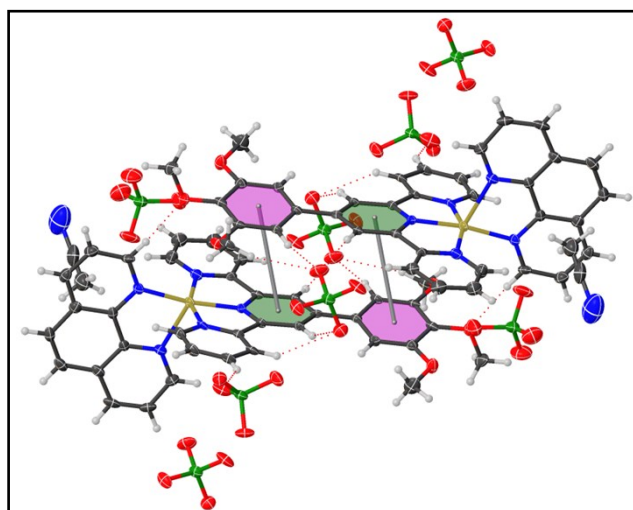


Figure S19. Crystal packing of C-I showing π - π interaction and H-bonding.

π - π interactions and hydrogen bond parameters (\AA , $^\circ$) for C-I

Cg(i)...Cg(j)	Distance	Dihedral angle	Slippage	Symmetry	
Cg(8)...Cg(9)	3.974(3)	14.00	1.857	1-x, 1-y, -z	
D-H...A	D-H	A...H	D...A	<D-A...H	Symmetry
C1-H1A...O4	0.9600	2.6000	3.531(10)	164.00	1-x,1-y,1-z
C4-H4...O4	0.9300	2.5200	3.422(8)	163.00	1-x,1-y,1-z
C11-H11...O4	0.9300	2.4800	3.408(7)	173.00	1-x,1-y,1-z
C19-H19...O7	0.9300	2.5800	3.384(8)	145.00	x,y,-1+z
C24-H24...O7	0.9300	2.4700	3.314(7)	150.00	x,y,-1+z
C26-H26...O6	0.9300	2.3400	3.149(8)	146.00	1-x,-1/2+y,1/2-z
C34-H34...O2	0.9300	2.5100	3.217(8)	133.0	1-x,1-y,-z
C36-H36...O9	0.9300	2.6000	3.393(9)	144.00	-x,1-y,-z

Cg(8): N5-C12-C24-C10-C11-C13; Cg(9): C2-C3-C4-C9-C8-C5

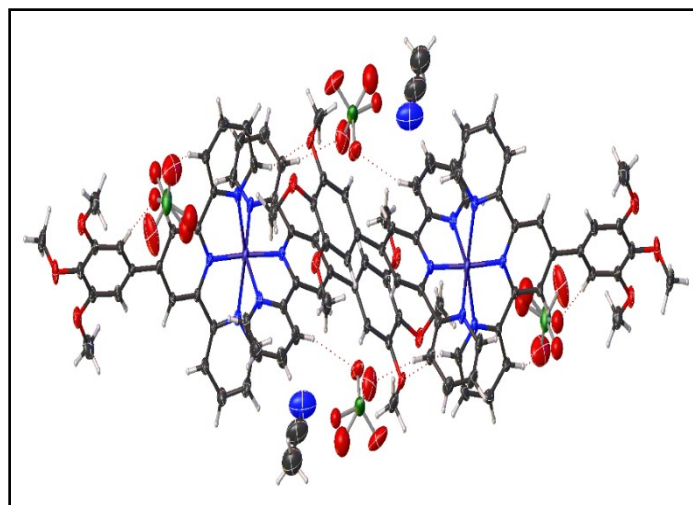


Figure S20. Crystal packing of **C-II** showing H-bonding.

Hydrogen bond parameters (\AA , $^\circ$) for **C-II**

D-H...A	D-H	A...H	D...A	<D-A...H	Symmetry
C2-H2A...O10	0.9600	2.3100	2.788(7)	110.00	
C3-H3A...O3	0.9600	2.5800	3.227(10)	125.00	x,y,1+z
C3-H3B...O11	0.9600	2.5500	3.492(7)	166.00	x,y,1+z
C016-H016...O5	0.9300	2.5500	3.366(10)	147.00	
C017-H017...O1	0.9300	2.5500	3.468(11)	171.00	1-x,1-y,1-z
C01B-H01B...O3	0.9300	2.5800	3.486(9)	166.00	x,y,1+z
C01E-H01E...O8	0.9300	2.4100	3.181(7)	140.0	1-x,1-y,2-z
C01N-H01N...O3	0.9300	2.4400	3.219(12)	141.00	2-x,2-y,1-z

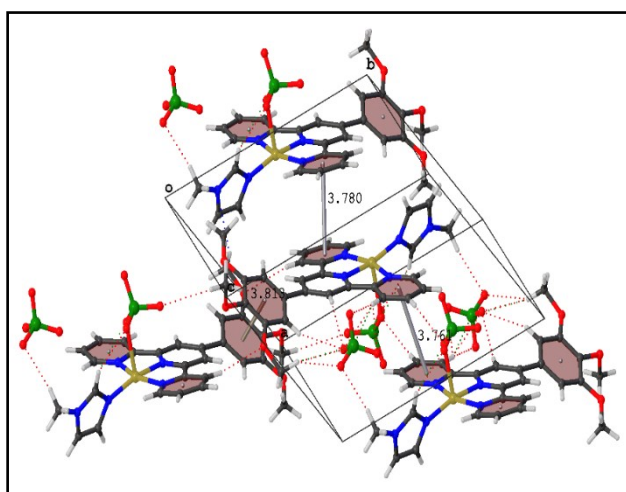


Figure S21. Crystal packing of **C-IV** showing π - π interaction and H-bonding.

π - π interactions and hydrogen bond parameters (\AA , $^\circ$) for C-IV

Cg(i)...Cg(j)	Distance	Dihedral angle	Slippage	Symmetry	
Cg(4)...Cg(4)	3.761(2)	0.00	1.844	2-x,1-y,1-z	
Cg(6)...Cg(6)	3.7803(18)	0.00	1.839	1-x,1-y,-z	
Cg(7)...Cg(7)	3.8130(19)	0.00	1.457	1-x,-y,1-z	
D-H...A	D-H	A...H	D...A	<D-A...H	Symmetry
C7-H7...O11	0.9300	2.5900	3.427(6)	150.00	2-x,1-y,1-z
C14-H14...O3	0.9300	2.5900	3.517(4)	172.00	-1+x,y,z
C21-H21...O4	0.9300	2.3800	3.093(4)	133.00	2-x,1-y,-z
C24-H24A...O7	0.9600	2.4400	3.223(6)	139.00	-x,-y,1-z
C25-H25...O4	0.9300	2.4500	3.285(5)	149.00	
C28-H28B...O9	0.9600	2.5700	3.482(6)	160.00	
C28-H28C...O2	0.9600	2.4700	3.425(5)	173.0	2-x,1-y,-z

Cg(4): N1-C1-C2-C3-C4-C5; Cg(6): N3-C17-C18-C19-C20-C21; Cg(7): C9-C10-C11-C12-C13-C14

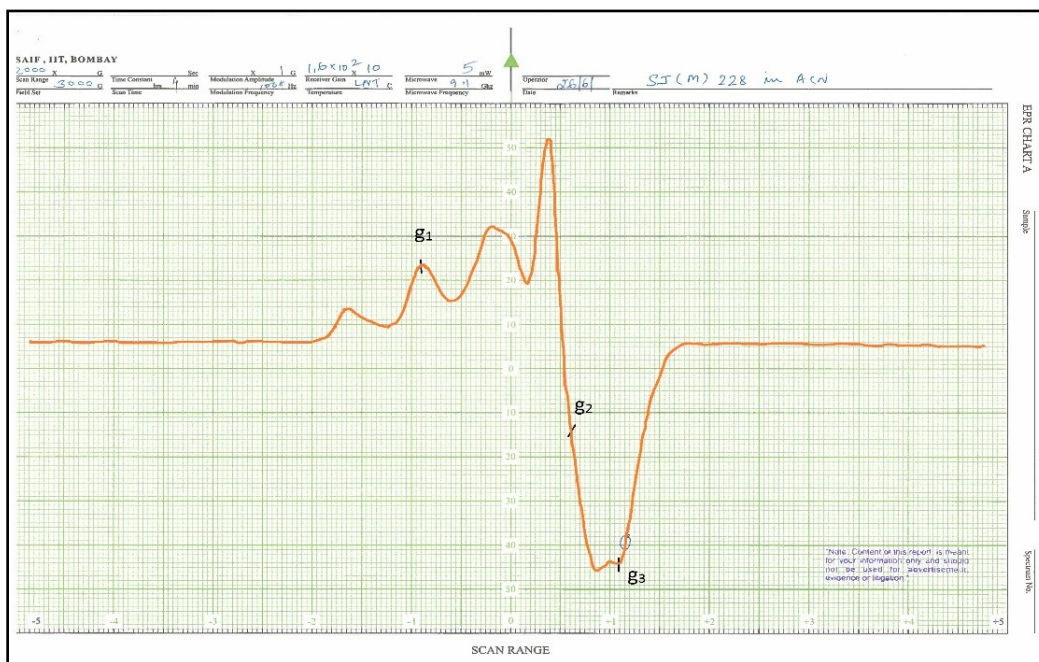


Figure S22. X-band EPR spectrum of complex C-II in MeCN glass at 77 K (frequency 9.1 GHz and 100 kHz field).

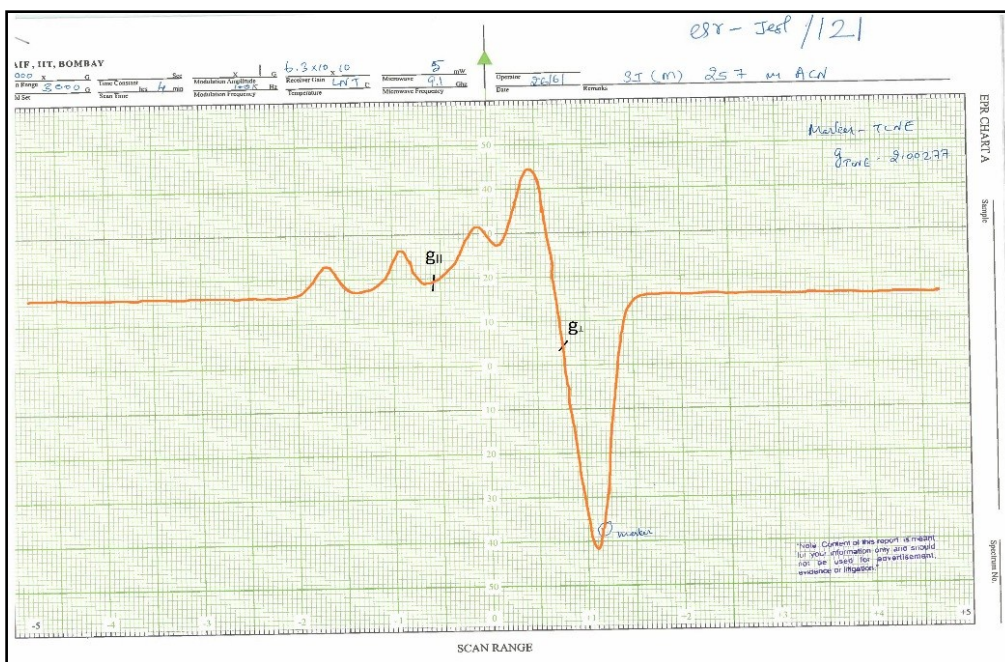


Figure S23. X-band EPR spectrum of complex **C-III** in MeCN glass at 77 K (frequency 9.1 GHz and 100 kHz field).

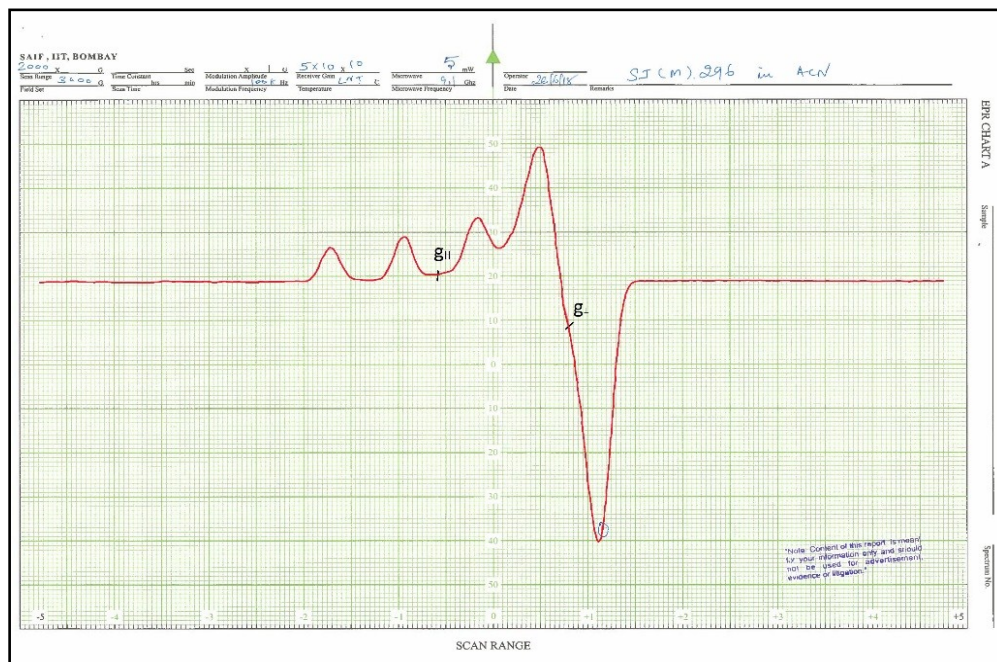


Figure S24. X-band EPR spectrum of complex **C-IV** in MeCN glass at 77 K (frequency 9.1 GHz and 100 kHz field).

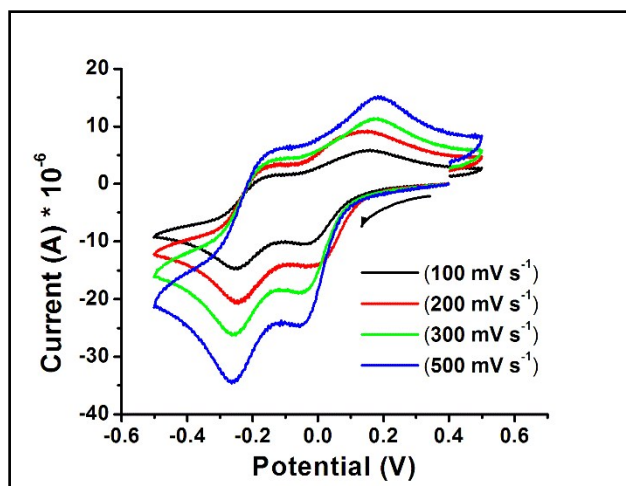


Figure S25. Cyclic voltammogram of C-I (1 mM) solution in dry CH_3CN and 0.1 M TBAP as supporting electrolyte vs. Ag wire at varying scan rates.

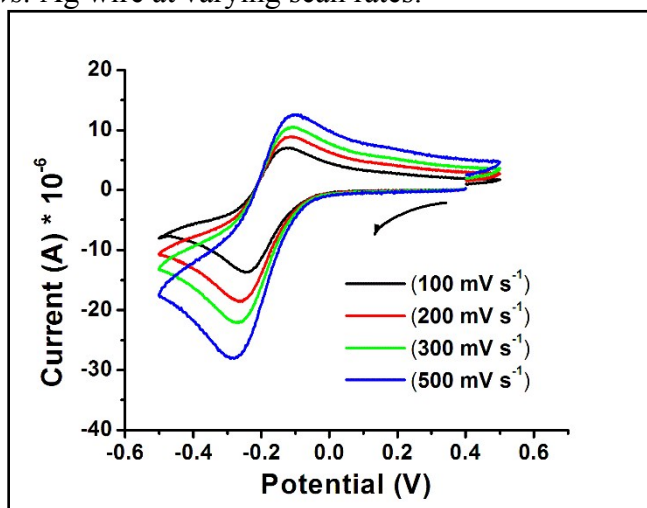


Figure S26. Cyclic voltammogram of C-II (1 mM) solution in dry CH_3CN and 0.1 M TBAP as supporting electrolyte vs. Ag wire at varying scan rates.

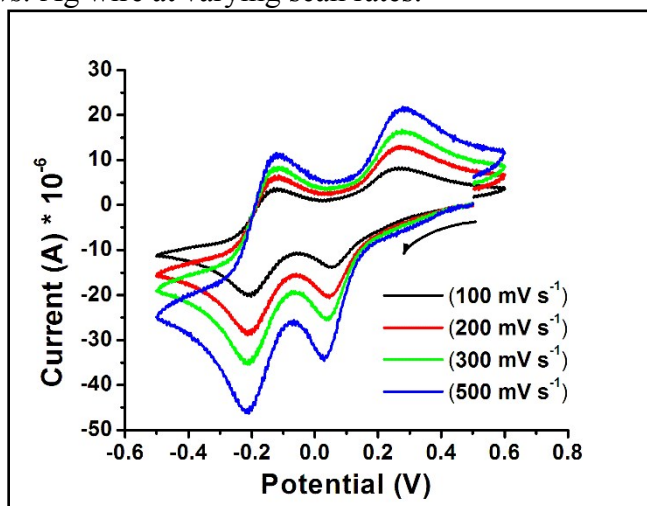


Figure S27. Cyclic voltammogram of C-III (1 mM) solution in dry CH_3CN and 0.1 M TBAP as supporting electrolyte vs. Ag wire at varying scan rates.

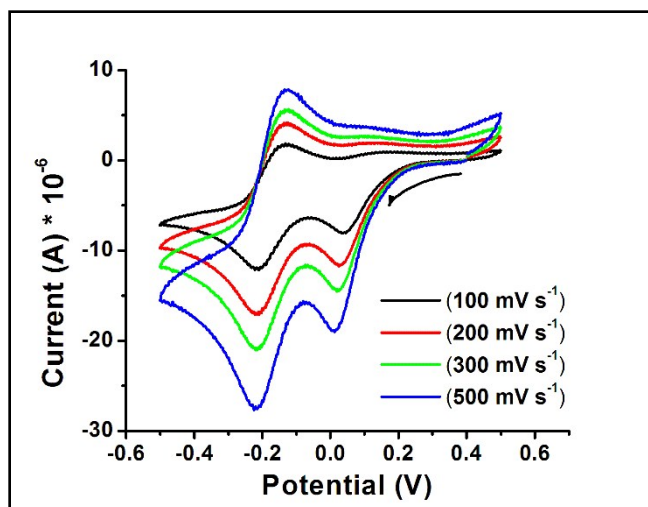


Figure S28. Cyclic voltammogram of **C-IV** (1 mM) solution in dry CH_3CN and 0.1 M TBAP as supporting electrolyte vs. Ag wire at varying scan rates.

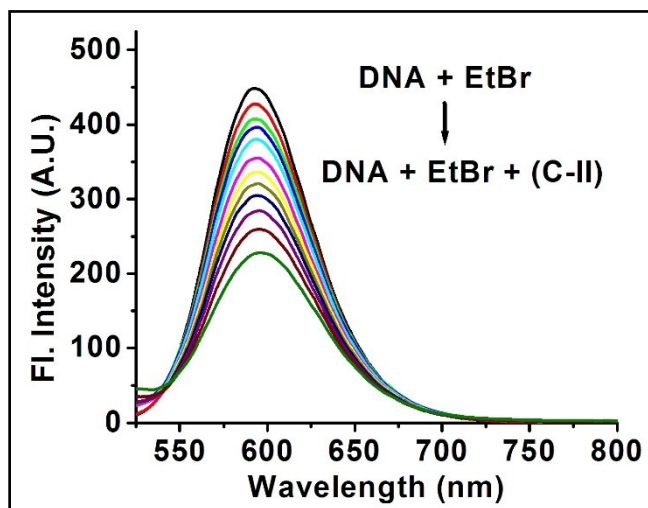


Figure S29. Plot of fluorescence emission intensity I versus wavelength λ for DNA-EtBr at different concentrations of **C-II**. The arrow shows the change in intensity of emission on increasing amount of the complex from 0 to 50 μM .

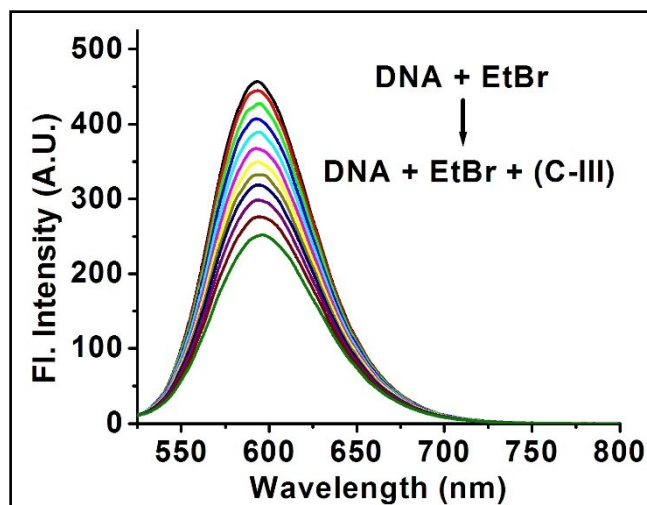


Figure S30. Plot of fluorescence emission intensity I versus wavelength λ for DNA-EtBr at different concentrations of C-III. The arrow shows the change in intensity of emission on increasing amount of the complex from 0 to 50 μM .

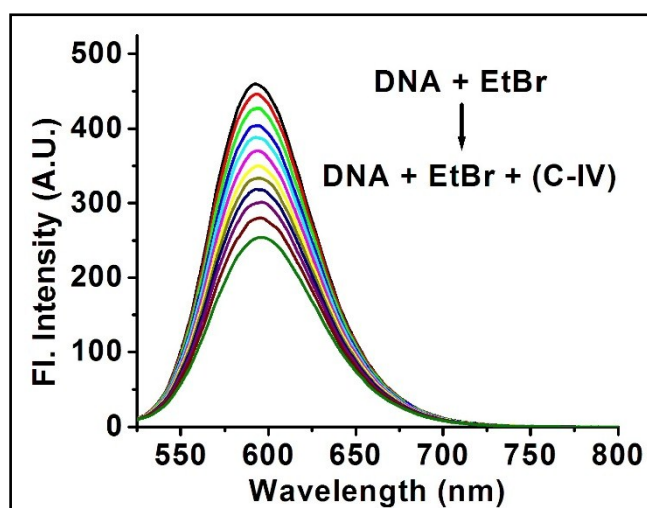


Figure S31. Plot of fluorescence emission intensity I versus wavelength λ for DNA-EtBr at different concentrations of C-IV. The arrow shows the change in intensity of emission on increasing amount of the complex from 0 to 50 μM .

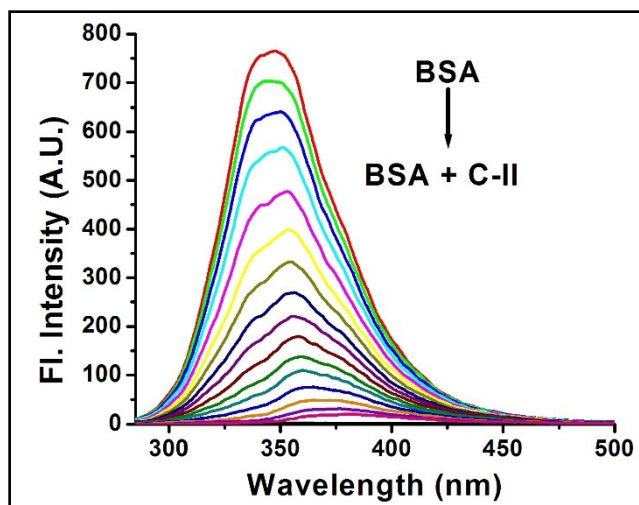


Figure S32. Plot of fluorescence emission intensity I versus wavelength λ for BSA at different concentrations of C-II. The arrow shows the change in intensity of emission on increasing amount of the complex from 0 to around 50 μM .

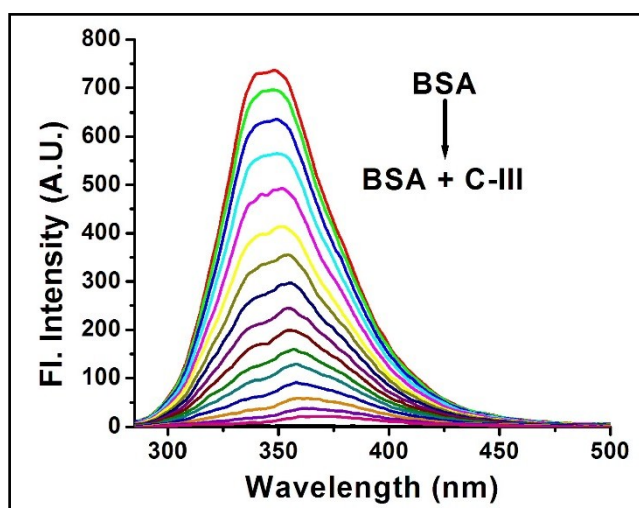


Figure S33. Plot of fluorescence emission intensity I versus wavelength λ for BSA at different concentrations of C-III. The arrow shows the change in intensity of emission on increasing amount of the complex from 0 to around 50 μM .

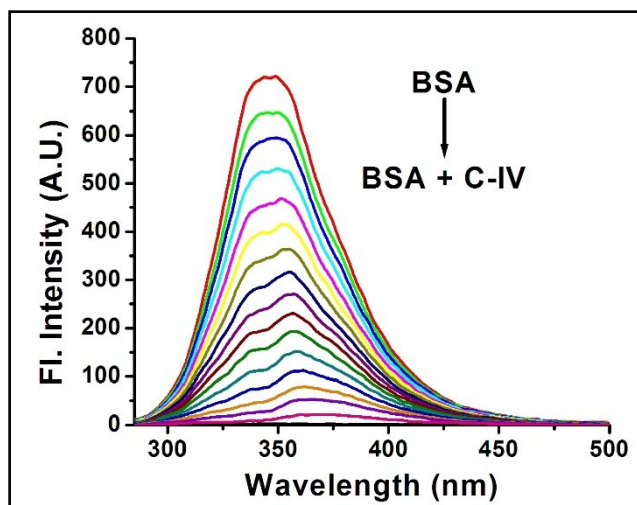


Figure S34. Plot of fluorescence emission intensity I versus wavelength λ for BSA at different concentrations of C-IV. The arrow shows the change in intensity of emission on increasing amount of the complex from 0 to around 50 μM .

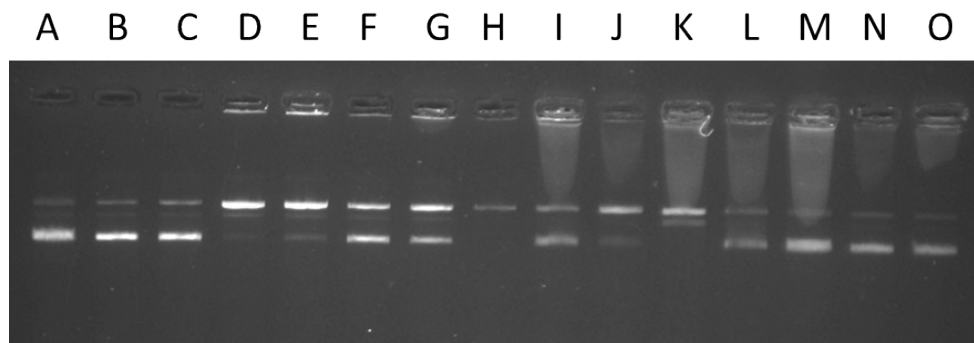


Figure S35. Agarose gel electrophoresis for complexes C-I to C-IV showing DNA aggregation in presence of H_2O_2 . Lane A: DNA only; Lane B: Buffer control; Lane C: H_2O_2 (500 μM) control; Lane D: C-I (500 μM) control; Lane E: C-II (500 μM) control; Lane F: C-III (500 μM) control; Lane G: C-IV (500 μM) control; Lane H: C-I (100 μM) + H_2O_2 (500 μM); Lane I: C-II (100 μM) + H_2O_2 (500 μM); Lane J: C-III (100 μM) + H_2O_2 (500 μM); Lane K: C-IV (100 μM) + H_2O_2 (500 μM); Lane L: C-I (500 μM) + H_2O_2 (500 μM); Lane M: C-II (500 μM) + H_2O_2 (500 μM); Lane N: C-III (500 μM) + H_2O_2 (500 μM); Lane O: C-IV (500 μM) + H_2O_2 (500 μM)

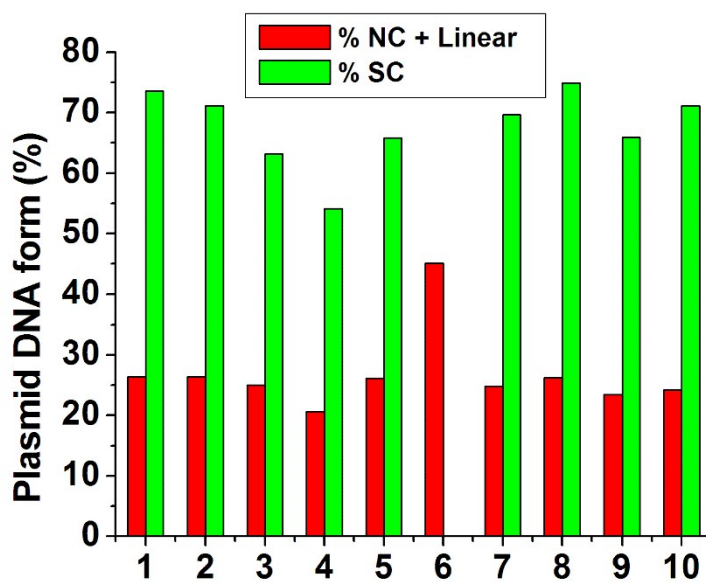
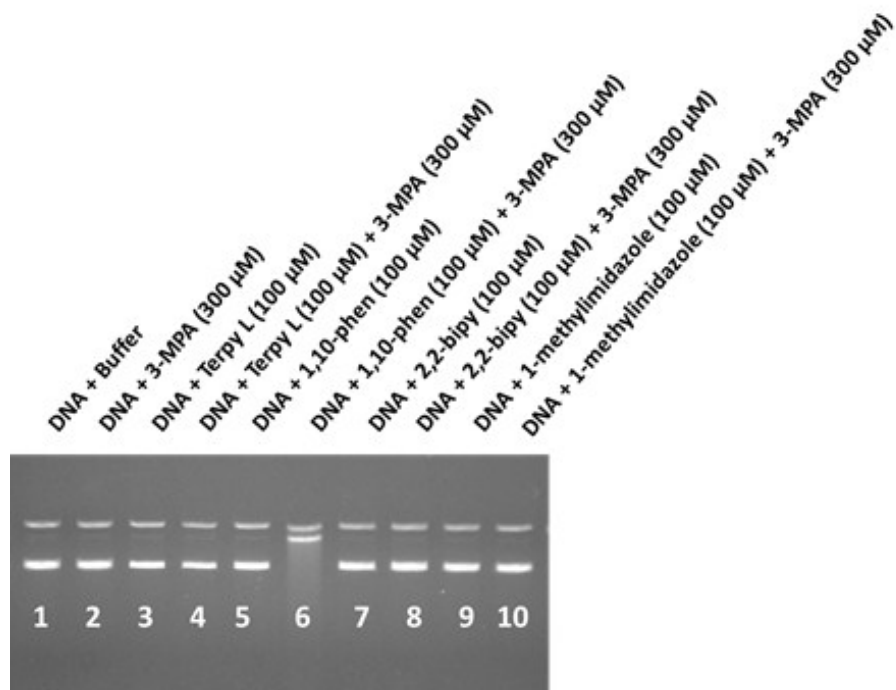


Figure S36. (Top): Agarose gel electrophoresis for terpyridine ligand (L) and co-ligands (1,10-phenanthroline, 2,2'-bipyridine, 3-methylimidazole) in absence and presence of 3-MPA; (Bottom): % Plasmid DNA form present in each lane.

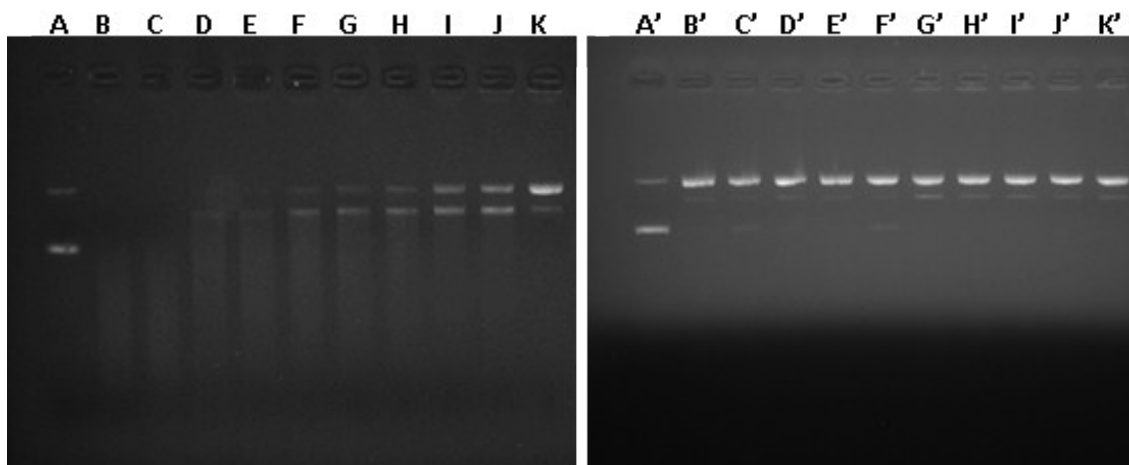


Figure S37 (Left): Gel electrophoresis diagram showing chemical nuclease activity of **C-I** to **C-IV** in the presence of various controls. Lane A: Buffer control; Lane B: DNA + **C-I** (100 μM) + MPA (500 μM); Lane C: DNA + **C-I** (100 μM) + MPA (500 μM) + D_2O (2.5 mM); Lane D: DNA + **C-I** (100 μM) + MPA (500 μM) + DMSO (2.5 mM); Lane E: DNA + **C-I** (100 μM) + MPA (500 μM) + NaN_3 (2.5 mM); Lane F: DNA + **C-I** (100 μM) + MPA (500 μM) + KI (2.5 mM); Lane G: DNA + **C-II** (100 μM) + MPA (500 μM); Lane H: DNA + **C-II** (100 μM) + MPA (500 μM) + D_2O (2.5 mM); Lane I: DNA + **C-II** (100 μM) + MPA (500 μM) + DMSO (2.5 mM); Lane J: DNA + **C-II** (100 μM) + MPA (500 μM) + NaN_3 (2.5 mM); Lane K: DNA + **C-II** (100 μM) + MPA (500 μM) + KI (2.5 mM).

(Right): Lane A': Buffer control; Lane B': DNA + **C-III** (100 μM) + MPA (500 μM); Lane C': DNA + **C-III** (100 μM) + MPA (500 μM) + D_2O (2.5 mM); Lane D': DNA + **C-III** (100 μM) + MPA (500 μM) + DMSO (2.5 mM); Lane E': DNA + **C-III** (100 μM) + MPA (500 μM) + NaN_3 (2.5 mM); Lane F': DNA + **C-III** (100 μM) + MPA (500 μM) + KI (2.5 mM); Lane G': DNA + **C-IV** (100 μM) + MPA (500 μM); Lane H': DNA + **C-IV** (100 μM) + MPA (500 μM) + D_2O (2.5 mM); Lane I': DNA + **C-IV** (100 μM) + MPA (500 μM) + DMSO (2.5 mM); Lane J': DNA + **C-IV** (100 μM) + MPA (500 μM) + NaN_3 (2.5 mM); Lane K': DNA + **C-IV** (100 μM) + MPA (500 μM) + KI (2.5 mM).

Table S1. Consolidated characterization data for copper(II) complexes (**C-I** - **C-IV**).

Complex	FTIR, cm^{-1} (KBr disc)	λ_{max} , nm (ϵ , $\text{M}^{-1} \text{cm}^{-1}$) (in MeCN)	ESI-MS m/z found (calculated): [(M^{2+} - 2ClO_4)/2] in MeCN	CHN found (calculated)	Conductance ($\text{S}\cdot\text{cm}^{-1}\text{M}^{-1}$)
C-I	3000 (w), 1600 (m), 1404 (m), 1085 (vs), 622 (m)	223 (7.5×10^4); 270 (5.1×10^4); 330 (1.9×10^4); 690 (1.2×10^2)	320.933 (321.080)	C=51.87 (51.35); H=3.68 (3.47); N=8.03 (8.32)	310
C-II	2945 (w),	222 ($8.4 \times$	430.541	C=54.69 (54.32);	280

	1610 (m), 1480 (m), 1090 (vs), 620 (m)	10 ⁴); 265 (4.0 x 10 ⁴); 287 (3.8 x 10 ⁴); 332 (3.9 x 10 ⁴); 689 (8.7 x 10 ¹)	(430.625)	H=4.09 (3.99); N=7.22 (7.72)	
C-III.H₂O	3520 (w), 3070 (w), 1605 (m), 1405 (m), 1090 (vs), 625 (m)	222 (6.3 x 10 ⁴); 288 (3.3 x 10 ⁴); 340 (2.0 x 10 ⁴); 637 (1.1 x 10 ²)	309.0905 (309.080)	C=48.54 (48.84); H=3.42 (3.74); N=8.72 (8.38)	350
C-IV	2935 (w), 1605 (m), 1405 (m), 1100 (vs), 620 (m)	222 (5.7 x 10 ⁴); 288 (2.0 x 10 ⁴); 340 (1.9 x 10 ⁴); 640 (1.1 x 10 ²)	----	C=45.02 (45.20); H=3.59 (3.66); N=9.89 (9.41)	320

Table S2: Summary of crystal data for complexes **C-I**, **C-II** and **C-IV**

Compound	C-I	C-II	C-IV
Formula	C ₇₆ H ₆₄ Cl ₄ Cu ₂ N ₁₂ O ₂₂	C ₅₀ H ₄₅ Cl ₂ CuN ₇ O ₁₄	C ₂₈ H ₂₇ Cl ₂ CuN ₅ O ₁₁
Formula weight/ g.mol ⁻¹	1766.27	1102.36	743.98
T (K)	296(2)	296(2)	293(2)
Crystal system	Monoclinic	Triclinic	Triclinic
Space group	P2 ₁ /c	p $\bar{1}$	p $\bar{1}$
<i>a</i> /Å	19.078(5)	9.6483(4)	9.1079(4)
<i>b</i> /Å	12.031(3)	12.9618(6)	12.5746(4)
<i>c</i> /Å	17.935(4)	20.9170(9)	14.8504(7)
α /°	90	91.460(3)	70.650(4)
β /°	113.406(12)	100.086(3)	88.016(4)
γ /°	90	107.065(3)	73.255(4)
V/Å ³	3778.0(16)	2453.82(19)	1533.18(12)
<i>Z</i>	2	2	2
ρ_{calcd} (mg/m ³)	1.553	1.436	1.612
Absorption coefficient (μ) (mm ⁻¹)	0.791	0.627	0.956
F (000)	1812	1094	762
R (int)	0.2052	0.0223	0.0198
Data/restraints/parameters	6633 / 0 / 527	8613 / 0 / 674	5385 / 0 / 424
Goodness-of-fit on <i>F</i> ²	1.117	1.169	1.057
Final R indices [<i>I</i> >2 σ (<i>I</i>)] (R ₁ , wR ₂)	0.0825, 0.2189	0.0768, 0.1962	0.0427, 0.1152
R indices (all data) (R ₁ , wR ₂)	0.0989, 0.2326	0.0860, 0.2029	0.0486, 0.1208
Largest diff. peak and hole	0.917 and -1.087	1.386 and -0.578	0.911 and -0.518

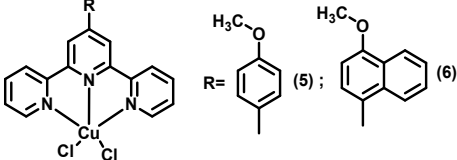
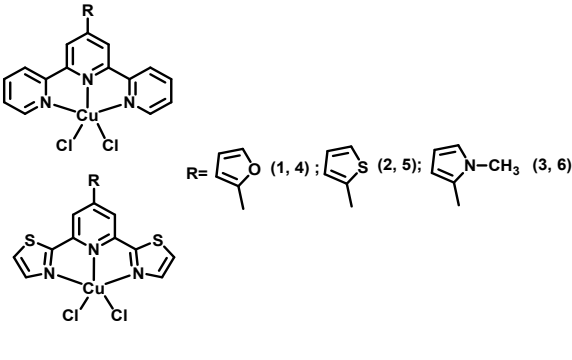
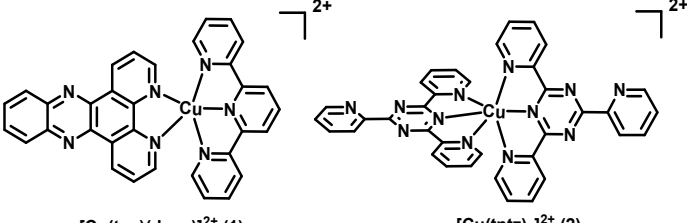
(e.Å ⁻³)			
----------------------	--	--	--

Table S3. Selected bond lengths (Å) and bond angles (°) for complexes **C-I**, **C-II** and **C-IV**

C-I					
Selected bonds		Selected angles			
Cu(1)-N(5)	1.944(4)	N(5)-Cu(1)-N(1)	170.58(19)	N(4)-Cu(1)-N(3)	158.5(2)
Cu(1)-N(1)	2.000(5)	N(5)-Cu(1)-N(4)	79.48(19)	N(5)-Cu(1)-N(2)	109.36(18)
Cu(1)-N(4)	2.044(5)	N(1)-Cu(1)-N(4)	100.5(2)	N(1)-Cu(1)-N(2)	79.98(19)
Cu(1)-N(3)	2.045(5)	N(5)-Cu(1)-N(3)	80.61(19)	N(4)-Cu(1)-N(2)	99.88(19)
Cu(1)-N(2)	2.215(5)	N(1)-Cu(1)-N(3)	97.77(19)	N(3)-Cu(1)-N(2)	94.38(18)
C-II					
Selected bonds		Selected angles			
Cu(01)-N(2)	1.947(4)	N(2)-Cu(01)-N(5)	176.74(16)	N(2)-Cu(01)-N(6)	103.24(16)
Cu(01)-N(5)	1.998(4)	N(2)-Cu(01)-N(3)	78.62(16)	N(5)-Cu(01)-N(6)	76.49(15)
Cu(01)-N(3)	2.109(5)	N(5)-Cu(01)-N(3)	104.63(16)	N(3)-Cu(01)-N(6)	94.14(17)
Cu(01)-N(1)	2.112(4)	N(2)-Cu(01)-N(1)	78.64(16)	N(1)-Cu(01)-N(6)	89.75(17)
Cu(01)-N(6)	2.223(4)	N(5)-Cu(01)-N(1)	98.11(16)	N(2)-Cu(01)-N(4)	103.42(16)
Cu(01)-N(4)	2.240(4)	N(3)-Cu(01)-N(1)	157.22(16)	N(5)-Cu(01)-N(4)	76.97(15)
		N(3)-Cu(01)-N(4)	89.75(17)	N(1)-Cu(01)-N(4)	96.80(17)
		N(6)-Cu(01)-N(4)	153.30(15)		
C-IV					
Selected bonds		Selected angles			
Cu(01)-N(2)	1.923(2)	N(2)-Cu(01)-N(4)	167.10(11)	N(3)-Cu(01)-N(1)	159.88(10)
Cu(01)-N(4)	1.950(2)	N(2)-Cu(01)-N(3)	80.20(9)	N(2)-Cu(01)-O(1)	95.26(10)
Cu(01)-N(3)	2.025(2)	N(4)-Cu(01)-N(3)	101.98(10)	N(4)-Cu(01)-O(1)	97.25(10)
Cu(01)-N(1)	2.035(3)	N(2)-Cu(01)-N(1)	79.75(10)	N(3)-Cu(01)-O(1)	94.06(9)
Cu(01)-O(1)	2.389(2)	N(4)-Cu(01)-N(1)	97.94(11)	N(1)-Cu(01)-O(1)	86.17(9)

Table S4. Summary of IC₅₀ values for similar Cu(II) complexes in cancer cell line.

Complex	IC ₅₀ (μM)	Ref.
 <chem>[Cu(terpy)(ClO4)2(H2O)]</chem> (1)	(1) 1.9 ± 0.1 ^a (2) 1.2 ± 0.1 ^a	1
 <chem>[Cu(terpy)2](CF3SO3)2·2H2O</chem> (2)		

	<p>(5) 1.02 ± 0.04^b 2.54 ± 0.05^c (6) 0.40 ± 0.06^b 0.95 ± 0.05^c</p>	2
	<p>(1) 17.3 ± 0.3^b 20.5 ± 0.2^a 43 ± 0.3^d 2.9 ± 0.1^c (2) 2.1 ± 0.1^b 4.2 ± 0.2^a $>100^d$ $>100^c$ (3) $>100^b$ $>100^a$ $>100^d$ $>100^c$ (4) 81.7 ± 0.3^b 64.3 ± 0.3^a $>100^d$ 5.9 ± 0.1^c (5) 20.5 ± 0.4^b 22.6 ± 0.3^a 62.3 ± 0.5^d 4.6 ± 0.1^c (6) $>100^b$ 70.1 ± 0.4^a 89 ± 0.5^d 2.6 ± 0.1^c</p>	3
 <p>[Cu(tpy)(dppz)]²⁺ (1) [Cu(tptz)₂]²⁺ (2)</p>	<p>(1) $4.57 \mu\text{M}^d$ (2) $1.98 \mu\text{M}^d$</p>	4

^a A549 cell; ^b HCT116 cell; ^c A2780 cell; ^d MCF-7 cell

1. B. Đ. Glišić, J. Nikodinovic-Runic, T. Ilic-Tomic, H. Wadepohl, A. Veselinović, I. M. Opsenica and M. I. Djuran, *Polyhedron*, 2018, **139**, 313-322.
2. A. Maroń, K. Czerwińska, B. Machura, L. Raposo, C. Roma-Rodrigues, A. R. Fernandes, J. G. Małecki, A. Szlapa-Kula, S. Kula and S. Krompiec, *Dalton Trans.*, 2018, **47**, 6444-6463.
3. K. Czerwińska, B. Machura, S. Kula, S. Krompiec, K. Erfurt, C. Roma-Rodrigues, A. R. Fernandes, L. S. Shul'pina, N. S. Ikonnikov and G. B. Shul'pin, *Dalton Trans.*, 2017, **46**, 9591-9604.
4. M. Salimi, K. Abdi, H. M. Kandelous, H. Hadadzadeh, K. Azadmanesh, A. Amanzadeh and H. Sanati, *BioMetals*, 2015, **28**, 267-278.