

Inference of Principal Species in Caustic Aluminate Solutions Through Solid-State Spectroscopic Characterization

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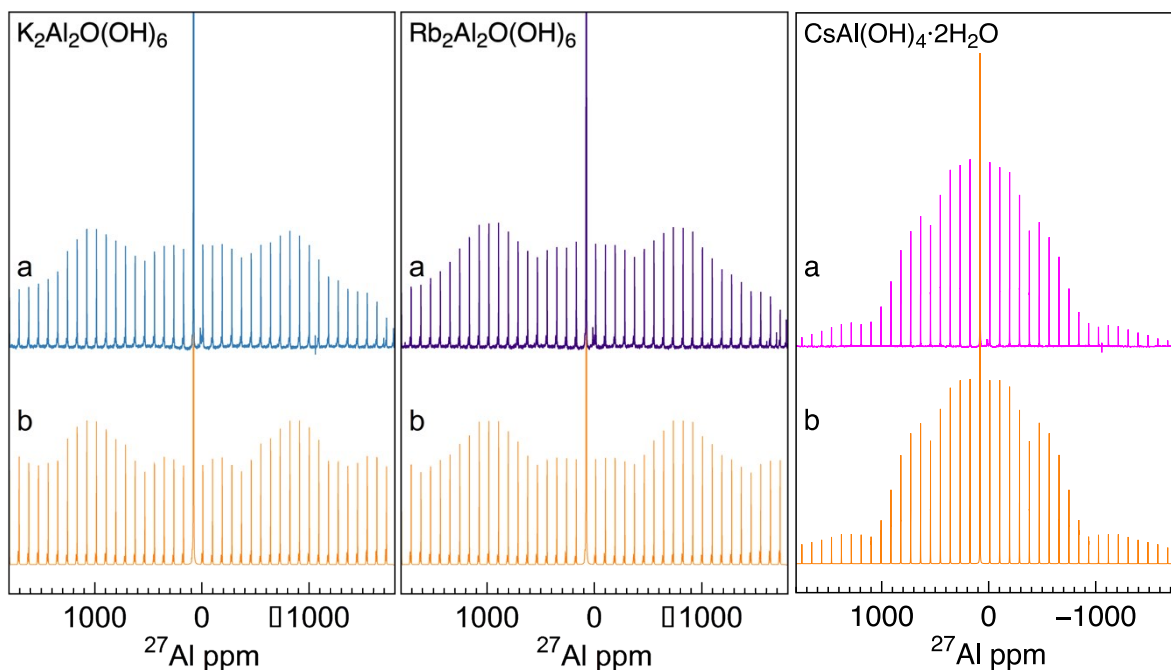


Figure S1. Experimental (top) and model fits (bottom) ^{27}Al MAS-NMR of $\text{K}_2\text{Al}_2\text{O}(\text{OH})_6$ (left), $\text{Rb}_2\text{Al}_2\text{O}(\text{OH})_6$ (center), and $\text{CsAl}(\text{OH})_4 \cdot 2\text{H}_2\text{O}$ collected at 19.975 T and 20 kHz spin rate.

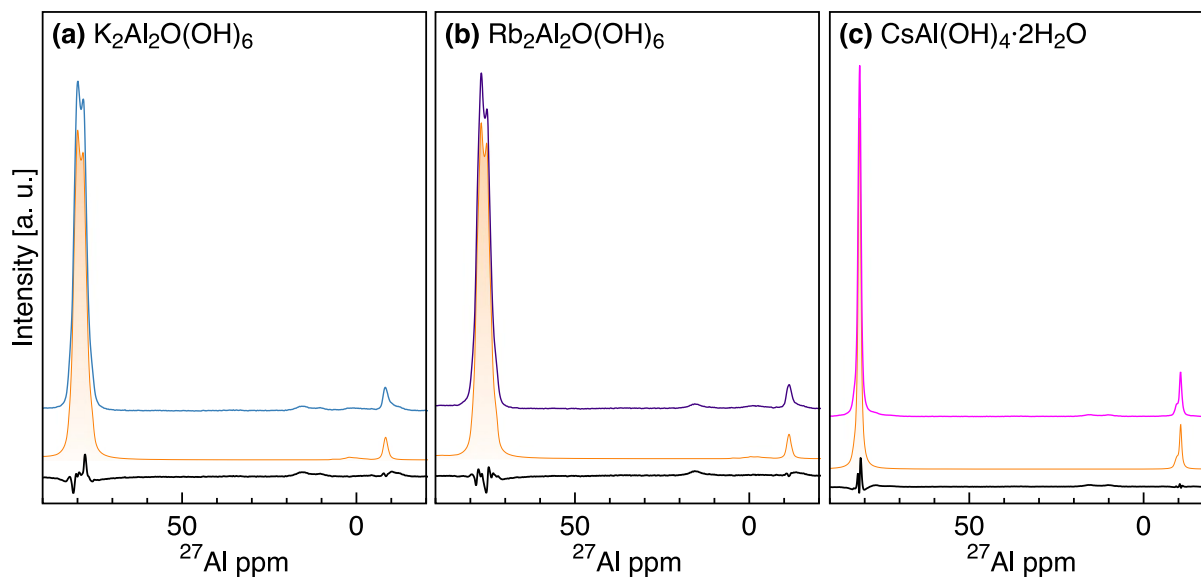


Figure S2. Difference (black), experimental (blue, purple, magenta) and model fits (orange) ^{27}Al MAS-NMR of $\text{K}_2\text{Al}_2\text{O}(\text{OH})_6$ (left), $\text{Rb}_2\text{Al}_2\text{O}(\text{OH})_6$ (center), and $\text{CsAl}(\text{OH})_4 \cdot 2\text{H}_2\text{O}$ collected at 19.975 T and 20 kHz spin rate. Signals at ca. -10 ppm are the spinning side bands. Those at ca. 10-15 ppm correspond to trace octahedral impurities.

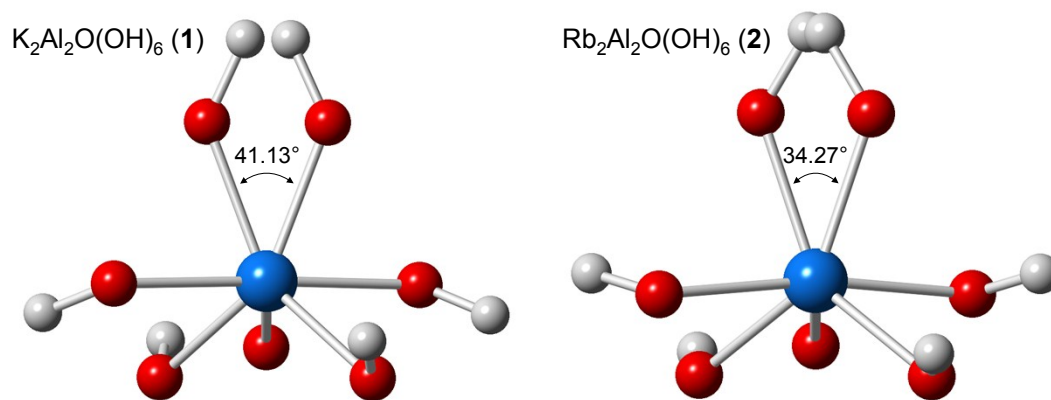


Figure S3. Comparison of conformations of the K (left) and Rb (right) salt of the $\text{Al}_2\text{O}(\text{OH})_6^{2-}$. Blue, red, and white spheres represent aluminum, oxygen, and hydrogen atoms, respectively.

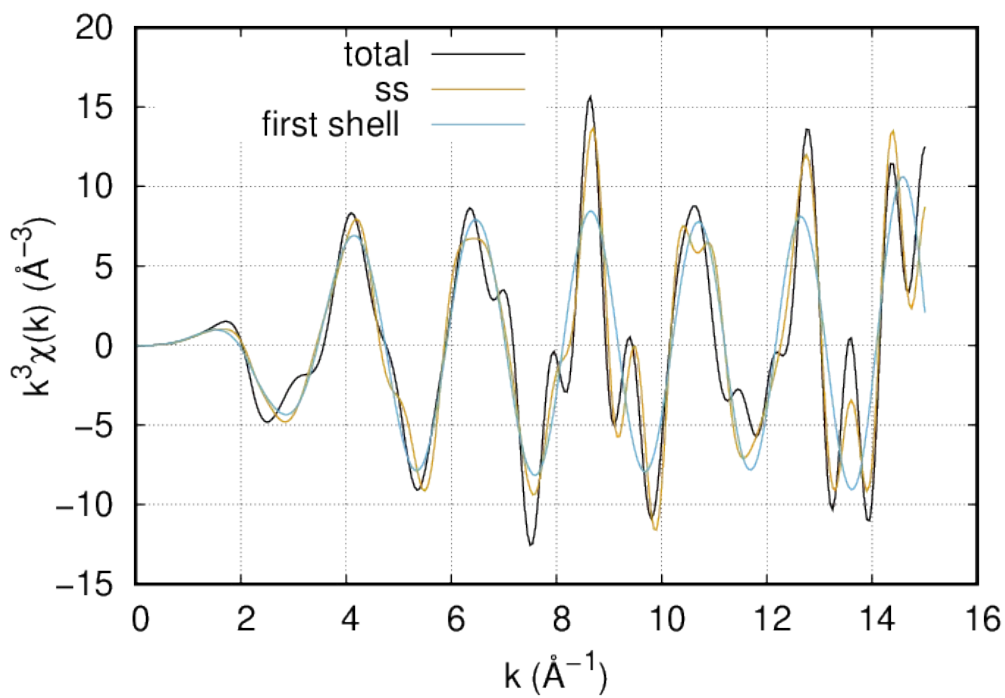


Figure S4. AIMD predicted EXAFS in k-space for the full set of paths (black), single-scattering paths (orange), and first-shell single-scattering paths (blue) for a single structure.

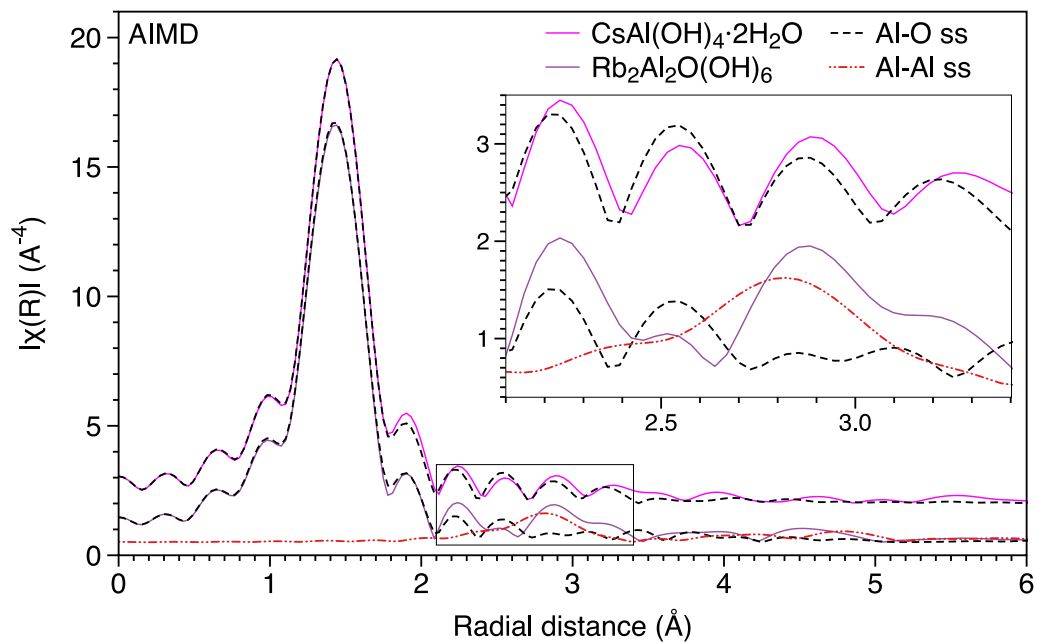


Figure S5. R-space EXAFS showing magnitude for **2** and **3**. Forward Fourier transform k-range between 3 and 13 \AA^{-1} used.

Crystallographic Tables

Crystal data for $K_2Al_2O(OH)_6$ (1)

$K_2Al_2O_7H_6$	$D_x = 2.169 \text{ Mg m}^{-3}$
$M_r = 250.21$	Ag $K\alpha$ radiation, $\lambda = 0.56086 \text{ \AA}$
Orthorhombic, $Aba2$	Cell parameters from 2593 reflections
$a = 10.1605 (12) \text{ \AA}$	$q = 3.1\text{--}26.8^\circ$
$b = 7.5416 (8) \text{ \AA}$	$m = 0.75 \text{ mm}^{-1}$
$c = 9.9990 (12) \text{ \AA}$	$T = 110 \text{ K}$
$V = 766.19 (15) \text{ \AA}^3$	Blocky, colorless
$Z = 4$	$0.04 \times 0.03 \times 0.02 \text{ mm}$
$F(000) = 504$	

Data collection

Bruker APEX-II CCD diffractometer	1478 reflections with $I > 2s(I)$
f and ω scans	$R_{\text{int}} = 0.085$
Absorption correction: multi-scan SADABS (Krause 2015)	$q_{\text{max}} = 26.9^\circ$, $q_{\text{min}} = 3.1^\circ$
$T_{\text{min}} = 0.663$, $T_{\text{max}} = 0.746$	$h = -16^{\circ}16$
9194 measured reflections	$k = -12^{\circ}12$
1687 independent reflections	$l = -16^{\circ}16$

Refinement

Refinement on F^2	Hydrogen site location: difference Fourier map
Least-squares matrix: full	All H-atom parameters refined
$R[F^2 > 2s(F^2)] = 0.037$	$w = 1/[s^2(F_o^2) + (0.0241P)^2]$ where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.068$	$(D/s)_{\text{max}} < 0.001$
$S = 1.12$	$D\rho_{\text{max}} = 0.46 \text{ e \AA}^{-3}$
1687 reflections	$D\rho_{\text{min}} = -0.53 \text{ e \AA}^{-3}$

63 parameters	Absolute structure: Flack x determined using 623 quotients [(+)-(I-)]/[(+)+(I-)] (Parsons, Flack and Wagner, Acta Cryst. B69 (2013) 249-259).
4 restraints	Absolute structure parameter: 0.00 (6)

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2) for (K2Al2O8H6)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
K1	0.30153 (6)	0.02766 (8)	0.59783 (6)	0.00989 (12)
Al1	0.49154 (8)	0.20877 (10)	0.86079 (9)	0.00461 (16)
O1	0.6432 (2)	0.3220 (2)	0.8575 (3)	0.0091 (4)
H1	0.718 (3)	0.259 (6)	0.827 (6)	0.058 (19)*
O2	0.3787 (2)	0.3316 (3)	0.7611 (2)	0.0078 (4)
H2	0.380 (6)	0.446 (3)	0.799 (5)	0.054 (18)*
O3	0.4299 (2)	0.2183 (3)	1.0253 (2)	0.0103 (4)
H3	0.469 (7)	0.202 (8)	1.111 (4)	0.10 (3)*
O4	0.500000	0.000000	0.7927 (3)	0.0072 (5)

Atomic displacement parameters (\AA^2) for (K2Al2O8H6)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
K1	0.0100 (2)	0.0108 (2)	0.0089 (2)	0.0012 (2)	0.0005 (3)	0.0011 (3)
Al1	0.0050 (4)	0.0042 (3)	0.0047 (4)	0.0001 (3)	-0.0006 (4)	-0.0005 (4)
O1	0.0046 (9)	0.0080 (8)	0.0149 (10)	-0.0007 (6)	-0.0013 (10)	-0.0023 (10)

O2	0.0069 (10)	0.0071 (9)	0.0094 (10)	0.0008 (7)	-0.0027 (8)	0.0005 (7)
O3	0.0115 (11)	0.0132 (10)	0.0062 (9)	0.0018 (8)	0.0008 (8)	-0.0011 (8)
O4	0.0109 (14)	0.0028 (12)	0.0078 (12)	0.0003 (10)	0.000	0.000

Geometric parameters (Å, °) for (K2Al2O8H6)

K1—O1 ⁱ	2.716 (3)	K1—Al1	3.5365 (11)
K1—O3 ⁱⁱ	2.770 (2)	K1—Al1 ⁱ	3.7402 (11)
K1—O4	2.812 (2)	K1—Al1 ^{vi}	3.8097 (11)
K1—O3 ⁱⁱⁱ	2.850 (2)	K1—Al1 ⁱⁱ	3.8895 (11)
K1—O2 ^{iv}	2.865 (2)	Al1—O4	1.7177 (15)
K1—O2	2.921 (2)	Al1—O1	1.761 (2)
K1—O1 ^v	3.259 (3)	Al1—O3	1.762 (3)
K1—O3 ⁱ	3.412 (2)	Al1—O2	1.779 (2)
O1 ⁱ —K1—O3 ⁱⁱ	91.30 (7)	O3 ⁱⁱ —K1—Al1 ⁱⁱ	23.91 (5)
O1 ⁱ —K1—O4	119.71 (7)	O4—K1—Al1 ⁱⁱ	91.16 (4)
O3 ⁱⁱ —K1—O4	77.35 (6)	O3 ⁱⁱⁱ —K1—Al1 ⁱⁱ	124.50 (5)
O1 ⁱ —K1—O3 ⁱⁱⁱ	74.64 (7)	O2 ^{iv} —K1—Al1 ⁱⁱ	110.21 (5)
O3 ⁱⁱ —K1—O3 ⁱⁱⁱ	138.32 (8)	O2—K1—Al1 ⁱⁱ	133.83 (5)
O4—K1—O3 ⁱⁱⁱ	143.66 (6)	O1 ^v —K1—Al1 ⁱⁱ	161.07 (4)
O1 ⁱ —K1—O2 ^{iv}	148.45 (7)	O3 ⁱ —K1—Al1 ⁱⁱ	79.68 (4)
O3 ⁱⁱ —K1—O2 ^{iv}	90.88 (7)	Al1—K1—Al1 ⁱⁱ	114.89 (3)
O4—K1—O2 ^{iv}	91.44 (6)	Al1 ⁱ —K1—Al1 ⁱⁱ	70.33 (3)
O3 ⁱⁱⁱ —K1—O2 ^{iv}	82.99 (7)	Al1 ^{vi} —K1—Al1 ⁱⁱ	81.793 (14)
O1 ⁱ —K1—O2	96.40 (7)	O4—Al1—O1	113.15 (9)
O3 ⁱⁱ —K1—O2	132.91 (7)	O4—Al1—O3	115.17 (12)
O4—K1—O2	58.56 (5)	O1—Al1—O3	107.97 (13)
O3 ⁱⁱⁱ —K1—O2	88.16 (6)	O4—Al1—O2	106.69 (11)

O2 ^{iv} —K1—O2	104.94 (8)	O1—Al1—O2	107.52 (11)
O1 ⁱ —K1—O1 ^v	131.44 (7)	O3—Al1—O2	105.83 (11)
O3 ⁱⁱ —K1—O1 ^v	137.25 (6)	O4—Al1—K1	51.57 (7)
O4—K1—O1 ^v	80.09 (6)	O1—Al1—K1	130.60 (10)
O3 ⁱⁱⁱ —K1—O1 ^v	67.65 (6)	O3—Al1—K1	121.06 (8)
O2 ^{iv} —K1—O1 ^v	53.83 (5)	O2—Al1—K1	55.44 (7)
O2—K1—O1 ^v	54.13 (6)	O4—Al1—K1 ^{vii}	135.26 (6)
O1 ⁱ —K1—O3 ⁱ	53.98 (6)	O1—Al1—K1 ^{vii}	42.49 (9)
O3 ⁱⁱ —K1—O3 ⁱ	92.34 (7)	O3—Al1—K1 ^{vii}	65.57 (8)
O4—K1—O3 ⁱ	67.38 (5)	O2—Al1—K1 ^{vii}	116.14 (8)
O3 ⁱⁱⁱ —K1—O3 ⁱ	108.82 (4)	K1—Al1—K1 ^{vii}	169.51 (3)
O2 ^{iv} —K1—O3 ⁱ	157.24 (7)	O4—Al1—K1 ^{vi}	43.08 (6)
O2—K1—O3 ⁱ	57.56 (6)	O1—Al1—K1 ^{vi}	74.41 (8)
O1 ^v —K1—O3 ⁱ	111.64 (6)	O3—Al1—K1 ^{vi}	149.28 (8)
O1 ⁱ —K1—Al1	112.55 (5)	O2—Al1—K1 ^{vi}	102.29 (8)
O3 ⁱⁱ —K1—Al1	105.24 (6)	K1—Al1—K1 ^{vi}	66.88 (3)
O4—K1—Al1	28.59 (2)	K1 ^{vii} —Al1—K1 ^{vi}	112.08 (3)
O3 ⁱⁱⁱ —K1—Al1	116.41 (5)	O4—Al1—K1 ^{viii}	146.44 (6)
O2 ^{iv} —K1—Al1	97.15 (5)	O1—Al1—K1 ^{viii}	98.38 (8)
O2—K1—Al1	30.10 (5)	O3—Al1—K1 ^{viii}	39.60 (7)
O1 ^v —K1—Al1	62.79 (4)	O2—Al1—K1 ^{viii}	72.51 (8)
O3 ⁱ —K1—Al1	60.30 (4)	K1—Al1—K1 ^{viii}	114.89 (3)
O1 ⁱ —K1—Al1 ⁱ	25.98 (5)	K1 ^{vii} —Al1—K1 ^{viii}	64.17 (2)
O3 ⁱⁱ —K1—Al1 ⁱ	90.98 (5)	K1 ^{vi} —Al1—K1 ^{viii}	169.71 (3)
O4—K1—Al1 ⁱ	94.32 (4)	Al1—O1—K1 ^{vii}	111.53 (12)
O3 ⁱⁱⁱ —K1—Al1 ⁱ	91.98 (5)	Al1—O1—K1 ^{ix}	127.98 (12)
O2 ^{iv} —K1—Al1 ⁱ	174.21 (5)	K1 ^{vii} —O1—K1 ^{ix}	117.22 (7)
O2—K1—Al1 ⁱ	77.67 (5)	Al1—O2—K1 ^x	179.30 (13)
O1 ^v —K1—Al1 ⁱ	126.70 (4)	Al1—O2—K1	94.46 (9)

O3 ⁱ —K1—Al1 ⁱ	28.04 (4)	K1 ^x —O2—K1	85.12 (6)
Al1—K1—Al1 ⁱ	87.650 (16)	Al1—O3—K1 ^{viii}	116.49 (11)
O1 ⁱ —K1—Al1 ^{vi}	133.78 (5)	Al1—O3—K1 ^{xi}	120.69 (11)
O3 ⁱⁱ —K1—Al1 ^{vi}	61.74 (5)	K1 ^{viii} —O3—K1 ^{xi}	88.26 (6)
O4—K1—Al1 ^{vi}	24.664 (17)	Al1—O3—K1 ^{vii}	86.39 (9)
O3 ⁱⁱⁱ —K1—Al1 ^{vi}	150.12 (5)	K1 ^{viii} —O3—K1 ^{vii}	81.25 (6)
O2 ^{iv} —K1—Al1 ^{vi}	73.60 (5)	K1 ^{xi} —O3—K1 ^{vii}	152.75 (9)
O2—K1—Al1 ^{vi}	80.40 (5)	Al1—O4—Al1 ^{vi}	133.26 (19)
O1 ^v —K1—Al1 ^{vi}	83.42 (4)	Al1—O4—K1 ^{vi}	112.25 (6)
O3 ⁱ —K1—Al1 ^{vi}	88.17 (4)	Al1 ^{vi} —O4—K1 ^{vi}	99.84 (6)
Al1—K1—Al1 ^{vi}	50.68 (3)	Al1—O4—K1	99.84 (6)
Al1 ⁱ —K1—Al1 ^{vi}	112.08 (3)	Al1 ^{vi} —O4—K1	112.25 (6)
O1 ⁱ —K1—Al1 ⁱⁱ	67.44 (5)	K1 ^{vi} —O4—K1	92.28 (10)

Symmetry codes: (i) $-x+1, -y+1/2, z-1/2$; (ii) $x, y-1/2, z-1/2$; (iii) $-x+1/2, y, z-1/2$; (iv) $-x+1/2, y-1/2, z$; (v) $x-1/2, -y+1/2, z$; (vi) $-x+1, -y, z$; (vii) $-x+1, -y+1/2, z+1/2$; (viii) $x, y+1/2, z+1/2$; (ix) $x+1/2, -y+1/2, z$; (x) $-x+1/2, y+1/2, z$; (xi) $-x+1/2, y, z+1/2$.

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Crystal Data for Rb₂Al₂O(OH)₆ (2)

Rb ₂ Al ₂ O ₇ H ₆	$D_x = 2.795 \text{ Mg m}^{-3}$
$M_r = 342.95$	Ag Ka radiation, $\lambda = 0.56086 \text{ \AA}$
Orthorhombic, <i>Aba</i> 2	Cell parameters from 5220 reflections
$a = 10.3511 (8) \text{ \AA}$	$q = 3.0\text{--}28.6^\circ$
$b = 7.6823 (7) \text{ \AA}$	$m = 6.58 \text{ mm}^{-1}$
$c = 10.2481 (8) \text{ \AA}$	$T = 110 \text{ K}$
$V = 814.93 (12) \text{ \AA}^3$	Blocky, colorless
$Z = 4$	$0.04 \times 0.03 \times 0.02 \text{ mm}$
$F(000) = 648$	
<i>Data collection</i>	
Bruker APEX-II CCD diffractometer	2057 reflections with $I > 2s(I)$
f and w scans	$R_{\text{int}} = 0.088$
Absorption correction: multi-scan SADABS (Krause 2015)	$q_{\text{max}} = 31.8^\circ$, $q_{\text{min}} = 3.0^\circ$
$T_{\text{min}} = 0.639$, $T_{\text{max}} = 0.746$	$h = -19^{*}18$
17980 measured reflections	$k = -14^{*}14$
2754 independent reflections	$l = -19^{*}17$
<i>Refinement</i>	
Refinement on F^2	Hydrogen site location: difference Fourier map
Least-squares matrix: full	All H-atom parameters refined
$R[F^2 > 2s(F^2)] = 0.043$	$w = 1/[s^2(F_o^2) + (0.0182P)^2]$ where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.062$	$(D/s)_{\text{max}} < 0.001$
$S = 1.07$	$D\rho_{\text{max}} = 1.14 \text{ e \AA}^{-3}$
2754 reflections	$D\rho_{\text{min}} = -1.49 \text{ e \AA}^{-3}$
63 parameters	Absolute structure: Flack x determined using 773

quotients $[(I+)-(I-)]/[(I+)+(I-)]$ (Parsons, Flack and Wagner, Acta Cryst. B69 (2013) 249-259).

4 restraints

Absolute structure parameter: 0.004 (10)

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2) for (Rb2Al2O8H6)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
Rb1	0.20111 (3)	0.97386 (4)	0.35079 (4)	0.00902 (6)
Al1	0.00211 (11)	0.79223 (14)	0.08628 (11)	0.00433 (19)
O1	0.1157 (3)	0.6689 (3)	0.1785 (3)	0.0078 (5)
O2	0.0558 (3)	0.7775 (4)	-0.0766 (3)	0.0108 (5)
O3	-0.1470 (3)	0.6832 (4)	0.0991 (3)	0.0094 (5)
O4	0.000000	1.000000	0.1489 (4)	0.0076 (7)
H2	0.010 (5)	0.808 (7)	-0.155 (3)	0.033 (16)*
H1	0.119 (7)	0.548 (3)	0.156 (6)	0.027 (17)*
H3	-0.213 (7)	0.765 (10)	0.079 (13)	0.11 (4)*

Atomic displacement parameters (\AA^2) for (Rb2Al2O8H6)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Rb1	0.00935 (11)	0.00892 (11)	0.00880 (11)	0.00096 (11)	0.00087 (19)	0.0012 (2)
Al1	0.0053 (4)	0.0021 (4)	0.0056 (5)	0.0003 (4)	0.0002 (4)	-0.0003 (4)

O1	0.0069 (11)	0.0050 (11)	0.0114 (12)	0.0007 (9)	-0.0030 (10)	0.0004 (9)
O2	0.0120 (14)	0.0138 (14)	0.0065 (12)	0.0024 (11)	0.0008 (10)	-0.0006 (9)
O3	0.0068 (11)	0.0062 (12)	0.0151 (13)	-0.0005 (8)	-0.0001 (11)	-0.0019 (10)
O4	0.0125 (17)	0.0026 (17)	0.0077 (15)	0.0010 (12)	0.000	0.000

Geometric parameters (Å, °) for (Rb₂Al₂O₈H₆)

Rb1—O3 ⁱ	2.871 (3)	Rb1—Al1	3.6794 (12)
Rb1—O2 ⁱⁱ	2.874 (3)	Rb1—Al1 ⁱ	3.7985 (13)
Rb1—O4	2.942 (3)	Rb1—Al1 ^{vi}	3.8733 (12)
Rb1—O1 ⁱⁱⁱ	2.993 (3)	Rb1—Rb1 ^{vii}	3.9722 (4)
Rb1—O2 ^{iv}	3.027 (3)	Al1—O4	1.7203 (18)
Rb1—O1	3.064 (3)	Al1—O3	1.761 (3)
Rb1—O3 ^v	3.253 (3)	Al1—O2	1.763 (3)
Rb1—O2 ⁱ	3.369 (3)	Al1—O1	1.781 (3)
O3 ⁱ —Rb1—O2 ⁱⁱ	90.56 (8)	O1—Rb1—Rb1 ^{vii}	48.25 (5)
O3 ⁱ —Rb1—O4	120.94 (7)	O3 ^v —Rb1—Rb1 ^{vii}	61.19 (5)
O2 ⁱⁱ —Rb1—O4	75.89 (7)	O2 ⁱ —Rb1—Rb1 ^{vii}	69.32 (5)
O3 ⁱ —Rb1—O1 ⁱⁱⁱ	148.95 (8)	Al1—Rb1—Rb1 ^{vii}	77.051 (18)
O2 ⁱⁱ —Rb1—O1 ⁱⁱⁱ	94.48 (8)	Al1 ⁱ —Rb1—Rb1 ^{vii}	67.710 (17)
O4—Rb1—O1 ⁱⁱⁱ	89.95 (7)	Al1 ^{vi} —Rb1—Rb1 ^{vii}	125.946 (17)
O3 ⁱ —Rb1—O2 ^{iv}	74.62 (8)	O4—Al1—O3	113.71 (11)
O2 ⁱⁱ —Rb1—O2 ^{iv}	140.91 (10)	O4—Al1—O2	114.61 (16)
O4—Rb1—O2 ^{iv}	142.67 (7)	O3—Al1—O2	108.43 (16)
O1 ⁱⁱⁱ —Rb1—O2 ^{iv}	82.40 (8)	O4—Al1—O1	107.72 (14)
O3 ⁱ —Rb1—O1	97.65 (8)	O3—Al1—O1	106.63 (14)
O2 ⁱⁱ —Rb1—O1	128.23 (8)	O2—Al1—O1	105.08 (15)
O4—Rb1—O1	56.13 (5)	O4—Al1—Rb1	51.73 (9)

O1 ⁱⁱⁱ —Rb1—O1	103.03 (9)	O3—Al1—Rb1	128.04 (11)
O2 ^{iv} —Rb1—O1	90.03 (8)	O2—Al1—Rb1	123.07 (11)
O3 ⁱ —Rb1—O3 ^v	129.87 (9)	O1—Al1—Rb1	56.05 (9)
O2 ⁱⁱ —Rb1—O3 ^v	139.41 (8)	O4—Al1—Rb1 ^{viii}	136.81 (8)
O4—Rb1—O3 ^v	79.02 (7)	O3—Al1—Rb1 ^{viii}	46.05 (10)
O1 ⁱⁱⁱ —Rb1—O3 ^v	53.96 (7)	O2—Al1—Rb1 ^{viii}	62.50 (11)
O2 ^{iv} —Rb1—O3 ^v	66.94 (7)	O1—Al1—Rb1 ^{viii}	114.59 (10)
O1—Rb1—O3 ^v	53.04 (7)	Rb1—Al1—Rb1 ^{viii}	169.18 (3)
O3 ⁱ —Rb1—O2 ⁱ	53.82 (7)	O4—Al1—Rb1 ^{vi}	45.71 (8)
O2 ⁱⁱ —Rb1—O2 ⁱ	89.71 (8)	O3—Al1—Rb1 ^{vi}	72.08 (10)
O4—Rb1—O2 ⁱ	68.66 (6)	O2—Al1—Rb1 ^{vi}	149.71 (11)
O1 ⁱⁱⁱ —Rb1—O2 ⁱ	156.50 (7)	O1—Al1—Rb1 ^{vi}	103.54 (10)
O2 ^{iv} —Rb1—O2 ⁱ	108.44 (3)	Rb1—Al1—Rb1 ^{vi}	67.20 (2)
O1—Rb1—O2 ⁱ	57.41 (7)	Rb1 ^{viii} —Al1—Rb1 ^{vi}	113.18 (3)
O3 ^v —Rb1—O2 ⁱ	110.11 (7)	O4—Al1—Rb1 ^{ix}	142.90 (8)
O3 ⁱ —Rb1—Al1	112.61 (6)	O3—Al1—Rb1 ^{ix}	101.85 (10)
O2 ⁱⁱ —Rb1—Al1	101.86 (6)	O2—Al1—Rb1 ^{ix}	39.51 (10)
O4—Rb1—Al1	27.33 (2)	O1—Al1—Rb1 ^{ix}	69.83 (10)
O1 ⁱⁱⁱ —Rb1—Al1	96.30 (6)	Rb1—Al1—Rb1 ^{ix}	112.80 (3)
O2 ^{iv} —Rb1—Al1	117.23 (6)	Rb1 ^{viii} —Al1—Rb1 ^{ix}	64.75 (2)
O1—Rb1—Al1	28.83 (5)	Rb1 ^{vi} —Al1—Rb1 ^{ix}	169.77 (3)
O3 ^v —Rb1—Al1	62.97 (5)	O4—Al1—Rb1 ^x	85.08 (9)
O2 ⁱ —Rb1—Al1	60.23 (5)	O3—Al1—Rb1 ^x	148.48 (11)
O3 ⁱ —Rb1—Al1 ⁱ	26.21 (6)	O2—Al1—Rb1 ^x	40.31 (10)
O2 ⁱⁱ —Rb1—Al1 ⁱ	88.99 (6)	O1—Al1—Rb1 ^x	89.93 (10)
O4—Rb1—Al1 ⁱ	95.26 (5)	Rb1—Al1—Rb1 ^x	83.48 (2)
O1 ⁱⁱⁱ —Rb1—Al1 ⁱ	174.32 (6)	Rb1 ^{viii} —Al1—Rb1 ^x	102.80 (3)
O2 ^{iv} —Rb1—Al1 ⁱ	92.07 (6)	Rb1 ^{vi} —Al1—Rb1 ^x	130.79 (3)
O1—Rb1—Al1 ⁱ	78.16 (6)	Rb1 ^{ix} —Al1—Rb1 ^x	58.273 (17)

O3 ^v —Rb1—Al1 ⁱ	124.88 (5)	Al1—O1—Rb1 ^{vii}	175.84 (15)
O2 ⁱ —Rb1—Al1 ⁱ	27.66 (5)	Al1—O1—Rb1	95.12 (11)
Al1—Rb1—Al1 ⁱ	87.361 (9)	Rb1 ^{vii} —O1—Rb1	81.95 (7)
O3 ⁱ —Rb1—Al1 ^{vi}	135.18 (6)	Al1—O2—Rb1 ^{ix}	117.52 (14)
O2 ⁱⁱ —Rb1—Al1 ^{vi}	61.33 (6)	Al1—O2—Rb1 ^x	117.56 (14)
O4—Rb1—Al1 ^{vi}	24.74 (2)	Rb1 ^{ix} —O2—Rb1 ^x	84.58 (8)
O1 ⁱⁱⁱ —Rb1—Al1 ^{vi}	72.48 (6)	Al1—O2—Rb1 ^{viii}	89.84 (12)
O2 ^{iv} —Rb1—Al1 ^{vi}	148.73 (6)	Rb1 ^{ix} —O2—Rb1 ^{viii}	83.73 (7)
O1—Rb1—Al1 ^{vi}	78.16 (5)	Rb1 ^x —O2—Rb1 ^{viii}	152.56 (10)
O3 ^v —Rb1—Al1 ^{vi}	83.08 (5)	Al1—O3—Rb1 ^{viii}	107.74 (13)
O2 ⁱ —Rb1—Al1 ^{vi}	89.53 (5)	Al1—O3—Rb1 ^{xi}	131.23 (14)
Al1—Rb1—Al1 ^{vi}	49.93 (3)	Rb1 ^{viii} —O3—Rb1 ^{xi}	116.91 (9)
Al1 ⁱ —Rb1—Al1 ^{vi}	113.17 (3)	Al1—O4—Al1 ^{vi}	136.2 (2)
O3 ⁱ —Rb1—Rb1 ^{vii}	69.10 (6)	Al1—O4—Rb1	100.95 (7)
O2 ⁱⁱ —Rb1—Rb1 ^{vii}	156.68 (6)	Al1 ^{vi} —O4—Rb1	109.55 (8)
O4—Rb1—Rb1 ^{vii}	104.258 (15)	Al1—O4—Rb1 ^{vi}	109.54 (8)
O1 ⁱⁱⁱ —Rb1—Rb1 ^{vii}	108.82 (5)	Al1 ^{vi} —O4—Rb1 ^{vi}	100.95 (7)
O2 ^{iv} —Rb1—Rb1 ^{vii}	46.07 (6)	Rb1—O4—Rb1 ^{vi}	90.61 (10)

Symmetry codes: (i) $-x, -y+3/2, z+1/2$; (ii) $x, y+1/2, z+1/2$; (iii) $-x+1/2, y+1/2, z$; (iv) $-x+1/2, y, z+1/2$; (v) $x+1/2, -y+3/2, z$; (vi) $-x, -y+2, z$; (vii) $-x+1/2, y-1/2, z$; (viii) $-x, -y+3/2, z-1/2$; (ix) $x, y-1/2, z-1/2$; (x) $-x+1/2, y, z-1/2$; (xi) $x-1/2, -y+3/2, z$.

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Crystal Data for CsAl(OH)₄·2H₂O (3)

CsAlO₆H₈

$F(000) = 228$

$M_r = 245.94$

$D_x = 2.333 \text{ Mg m}^{-3}$

Monoclinic, <i>C</i> 2	Ag <i>K</i> α radiation, $\lambda = 0.56086 \text{ \AA}$
$a = 10.4124 (15) \text{ \AA}$	Cell parameters from 3749 reflections
$b = 6.6986 (10) \text{ \AA}$	$q = 3.1\text{--}21.7^\circ$
$c = 6.2156 (10) \text{ \AA}$	$m = 2.83 \text{ mm}^{-1}$
$\beta = 126.129 (4)^\circ$	$T = 110 \text{ K}$
$V = 350.16 (9) \text{ \AA}^3$	Acicular, colorless
$Z = 2$	$0.04 \times 0.01 \times 0.01 \text{ mm}$

Data collection

Bruker APEX-II CCD diffractometer	527 reflections with $I > 2s(I)$
ϕ and ω scans	$q_{\max} = 21.7^\circ$, $q_{\min} = 3.1^\circ$
Absorption correction: multi-scan <i>SADABS</i> (Krause 2015)	$h = -13^{\circ}11$
$T_{\min} = 0.625$, $T_{\max} = 0.745$	$k = 0^{\circ}8$
527 measured reflections	$l = 0^{\circ}8$
527 independent reflections	

Refinement

Refinement on F^2	H-atom parameters not defined
Least-squares matrix: full	$w = 1/[s^2(F_o^2) + (0.0539P)^2]$ where $P = (F_o^2 + 2F_c^2)/3$
$R[F^2 > 2s(F^2)] = 0.027$	$(D/s)_{\max} < 0.001$
$wR(F^2) = 0.079$	$D\rho_{\max} = 0.94 \text{ e \AA}^{-3}$
$S = 1.21$	$D\rho_{\min} = -1.02 \text{ e \AA}^{-3}$
527 reflections	Absolute structure: No quotients, so Flack parameter determined by classical intensity fit
39 parameters	Absolute structure parameter: $-0.03 (12)$
1 restraint	

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Refinement. Refined as a 2-component twin.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2) for (CsAlOH₄)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
Cs1	0.500000	0.17830 (9)	0.500000	0.0100 (2)
Al1	0.500000	0.662 (3)	1.000000	0.007 (2)
O1	0.3636 (10)	0.5095 (12)	0.7282 (14)	0.0138 (14)
O2	0.5884 (10)	0.8230 (12)	0.9015 (16)	0.0133 (15)
O1W	0.2690 (9)	0.2201 (11)	0.7680 (13)	0.017 (2)

Atomic displacement parameters (\AA^2) for (CsAlOH₄)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Cs1	0.0101 (4)	0.0092 (4)	0.0118 (3)	0.000	0.0070 (3)	0.000
Al1	0.0081 (14)	0.008 (6)	0.0050 (13)	0.000	0.0036 (11)	0.000
O1	0.022 (4)	0.009 (3)	0.007 (3)	-0.006 (3)	0.007 (3)	-0.001 (3)
O2	0.020 (4)	0.006 (3)	0.018 (3)	-0.005 (3)	0.014 (3)	-0.003 (3)
O1W	0.021 (4)	0.017 (8)	0.014 (3)	0.008 (3)	0.011 (3)	0.003 (2)

Geometric parameters (\AA , $^\circ$) for (CsAlOH₄)

Cs1—O2 ⁱ	3.167 (8)	Cs1—O1W ^{vi}	3.696 (6)
Cs1—O2 ⁱⁱ	3.167 (8)	Cs1—O1W ^{vii}	3.696 (6)
Cs1—O1 ⁱⁱⁱ	3.360 (9)	Cs1—O2 ^{viii}	3.788 (8)
Cs1—O1 ^{iv}	3.360 (9)	Cs1—O2 ^{ix}	3.788 (8)
Cs1—O1	3.366 (9)	Al1—O2	1.747 (15)

Cs1—O1 ^v	3.366 (9)	Al1—O2 ^{vi}	1.747 (15)
Cs1—O1W ^v	3.656 (7)	Al1—O1	1.756 (14)
Cs1—O1W	3.656 (7)	Al1—O1 ^{vi}	1.756 (14)
O2 ⁱ —Cs1—O2 ⁱⁱ	82.6 (3)	O1W ^{vi} —Cs1—O1W ^{vii}	171.3 (2)
O2 ⁱ —Cs1—O1 ⁱⁱⁱ	71.52 (19)	O2 ⁱ —Cs1—O2 ^{viii}	126.9 (2)
O2 ⁱⁱ —Cs1—O1 ⁱⁱⁱ	79.13 (19)	O2 ⁱⁱ —Cs1—O2 ^{viii}	77.50 (14)
O2 ⁱ —Cs1—O1 ^{iv}	79.13 (19)	O1 ⁱⁱⁱ —Cs1—O2 ^{viii}	147.30 (16)
O2 ⁱⁱ —Cs1—O1 ^{iv}	71.52 (19)	O1 ^{iv} —Cs1—O2 ^{viii}	47.98 (17)
O1 ⁱⁱⁱ —Cs1—O1 ^{iv}	140.7 (3)	O1—Cs1—O2 ^{viii}	86.08 (18)
O2 ⁱ —Cs1—O1	146.0 (2)	O1 ^v —Cs1—O2 ^{viii}	74.39 (18)
O2 ⁱⁱ —Cs1—O1	99.33 (18)	O1W ^v —Cs1—O2 ^{viii}	76.14 (16)
O1 ⁱⁱⁱ —Cs1—O1	75.48 (12)	O1W—Cs1—O2 ^{viii}	101.56 (16)
O1 ^{iv} —Cs1—O1	134.0 (2)	O1W ^{vi} —Cs1—O2 ^{viii}	39.94 (16)
O2 ⁱ —Cs1—O1 ^v	99.33 (18)	O1W ^{vii} —Cs1—O2 ^{viii}	136.71 (17)
O2 ⁱⁱ —Cs1—O1 ^v	146.0 (2)	O2 ⁱ —Cs1—O2 ^{ix}	77.50 (14)
O1 ⁱⁱⁱ —Cs1—O1 ^v	134.0 (2)	O2 ⁱⁱ —Cs1—O2 ^{ix}	126.9 (2)
O1 ^{iv} —Cs1—O1 ^v	75.48 (12)	O1 ⁱⁱⁱ —Cs1—O2 ^{ix}	47.98 (17)
O1—Cs1—O1 ^v	97.5 (2)	O1 ^{iv} —Cs1—O2 ^{ix}	147.30 (16)
O2 ⁱ —Cs1—O1W ^v	70.25 (18)	O1—Cs1—O2 ^{ix}	74.39 (18)
O2 ⁱⁱ —Cs1—O1W ^v	116.94 (18)	O1 ^v —Cs1—O2 ^{ix}	86.09 (18)
O1 ⁱⁱⁱ —Cs1—O1W ^v	135.53 (16)	O1W ^v —Cs1—O2 ^{ix}	101.56 (16)
O1 ^{iv} —Cs1—O1W ^v	48.54 (17)	O1W—Cs1—O2 ^{ix}	76.14 (16)
O1—Cs1—O1W ^v	134.11 (17)	O1W ^{vi} —Cs1—O2 ^{ix}	136.71 (17)
O1 ^v —Cs1—O1W ^v	37.14 (18)	O1W ^{vii} —Cs1—O2 ^{ix}	39.94 (16)
O2 ⁱ —Cs1—O1W	116.94 (18)	O2 ^{viii} —Cs1—O2 ^{ix}	150.3 (2)
O2 ⁱⁱ —Cs1—O1W	70.25 (18)	O2—Al1—O2 ^{vi}	103.6 (12)
O1 ⁱⁱⁱ —Cs1—O1W	48.54 (17)	O2—Al1—O1	108.4 (3)
O1 ^{iv} —Cs1—O1W	135.54 (16)	O2 ^{vi} —Al1—O1	113.7 (4)

O1—Cs1—O1W	37.14 (18)	O2—Al1—O1 ^{vi}	113.7 (4)
O1 ^v —Cs1—O1W	134.11 (17)	O2 ^{vi} —Al1—O1 ^{vi}	108.4 (3)
O1W ^v —Cs1—O1W	171.2 (2)	O1—Al1—O1 ^{vi}	109.0 (12)
O2 ⁱ —Cs1—O1W ^{vi}	134.65 (18)	O2—Al1—Cs1 ^x	64.2 (3)
O2 ⁱⁱ —Cs1—O1W ^{vi}	53.93 (19)	O2 ^{vi} —Al1—Cs1 ^x	113.8 (5)
O1 ⁱⁱⁱ —Cs1—O1W ^{vi}	107.38 (17)	O1—Al1—Cs1 ^x	132.3 (6)
O1 ^{iv} —Cs1—O1W ^{vi}	75.66 (17)	O1 ^{vi} —Al1—Cs1 ^x	50.0 (3)
O1—Cs1—O1W ^{vi}	63.97 (17)	O2—Al1—Cs1 ^{xi}	113.8 (5)
O1 ^v —Cs1—O1W ^{vi}	109.80 (17)	O2 ^{vi} —Al1—Cs1 ^{xi}	64.2 (3)
O1W ^v —Cs1—O1W ^{vi}	115.4 (2)	O1—Al1—Cs1 ^{xi}	50.0 (3)
O1W—Cs1—O1W ^{vi}	63.8 (2)	O1 ^{vi} —Al1—Cs1 ^{xi}	132.3 (6)
O2 ⁱ —Cs1—O1W ^{vii}	53.93 (19)	Cs1 ^x —Al1—Cs1 ^{xi}	177.0 (5)
O2 ⁱⁱ —Cs1—O1W ^{vii}	134.65 (18)	Al1—O1—Cs1 ^{xi}	106.4 (5)
O1 ⁱⁱⁱ —Cs1—O1W ^{vii}	75.66 (17)	Al1—O1—Cs1	119.0 (5)
O1 ^{iv} —Cs1—O1W ^{vii}	107.38 (16)	Cs1 ^{xi} —O1—Cs1	134.0 (2)
O1—Cs1—O1W ^{vii}	109.80 (17)	Al1—O2—Cs1 ^{xii}	140.6 (5)
O1 ^v —Cs1—O1W ^{vii}	63.97 (17)	Al1—O2—Cs1 ^x	91.3 (4)
O1W ^v —Cs1—O1W ^{vii}	63.8 (2)	Cs1 ^{xii} —O2—Cs1 ^x	126.9 (2)
O1W—Cs1—O1W ^{vii}	115.4 (2)	Cs1—O1W—Cs1 ^{xiii}	115.4 (2)

Symmetry codes: (i) $-x+1, y-1, -z+1$; (ii) $x, y-1, z$; (iii) $-x+1/2, y-1/2, -z+1$; (iv) $x+1/2, y-1/2, z$; (v) $-x+1, y, -z+1$; (vi) $-x+1, y, -z+2$; (vii) $x, y, z-1$; (viii) $-x+3/2, y-1/2, -z+2$; (ix) $x-1/2, y-1/2, z-1$; (x) $x+1/2, y+1/2, z+1$; (xi) $x-1/2, y+1/2, z$; (xii) $x, y+1, z$; (xiii) $x, y, z+1$.

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