

## Supporting Information

### Facile Synthesis of Porous $\text{Co}_3\text{O}_4$ Nanoflakes as an Interlayer for High Performance Lithium-Sulfur Batteries

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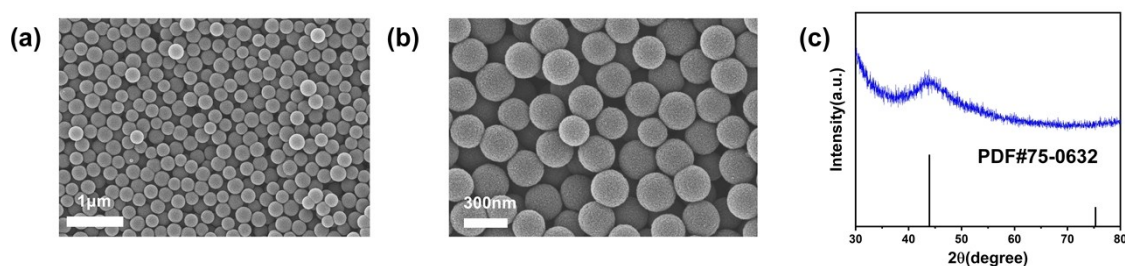


Fig. S1 (a-b) SEM images and (c) XRD of carbon spheres.

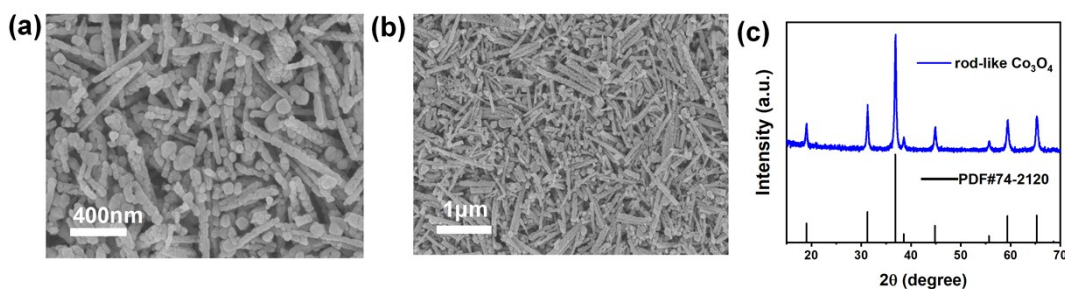


Fig. S2 (a-b) SEM images and (c) XRD of rod-like  $\text{Co}_3\text{O}_4$ .

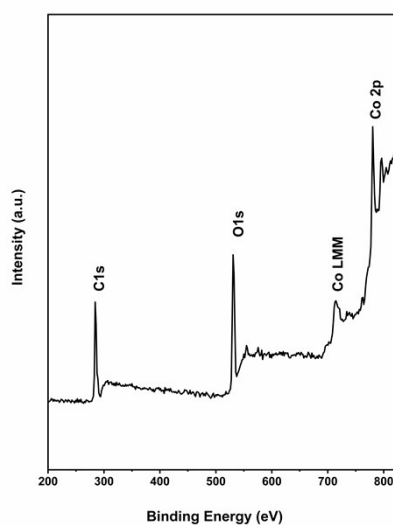
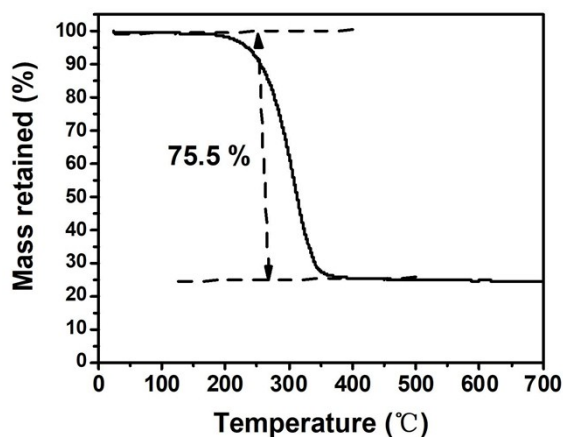
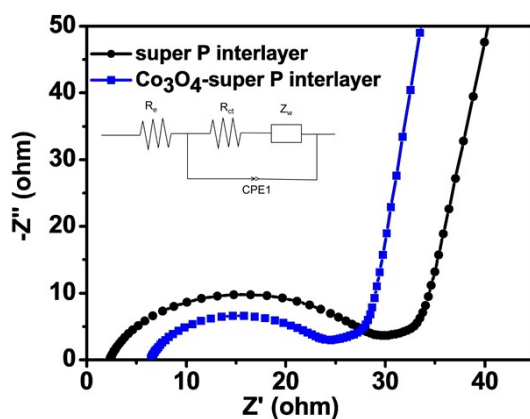


Fig. S3 XPS spectrum of  $\text{Co}_3\text{O}_4$

**Table S1** the relative content of  $\text{Co}^{2+}$  and  $\text{Co}^{3+}$  before and after adsorption

Adsorption state	$\text{Co}^{2+}$ (%)	$\text{Co}^{3+}$ (%)
Before adsorption	40.4	59.6
After adsorption	48.6	51.4

**Fig. S4** Thermogravimetric analysis (TGA) curve of carbon black-sulfur cathode with a heating rate of  $5\text{ }^\circ\text{C min}^{-1}$ .**Fig. S5** Nyquist plots of  $\text{Co}_3\text{O}_4$ -super P interlayer and super P interlayer and equivalent circuit (inset).**Table S2** EIS fitting results of cells with  $\text{Co}_3\text{O}_4$ -super P interlayer and super P interlayer

Interlayer	$R_{ct}$ ( $\Omega$ )	$R_s$ ( $\Omega$ )
$\text{Co}_3\text{O}_4$ -super P	19.4	6.4
Super P	26.9	2.3

