

C-H...Au interaction and optical properties in [(P,P)₄Au₆]²⁺ molecular gold nanoclusters.

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Supplementary Material

Table S1. Main geometrical features (in Å and °) of the [core+exo] Au₆ units together with PP distance and PAuP bond angle of the main systems studied here.

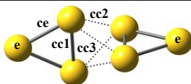
	<i>cc1</i>	<i>cc2</i>	<i>cc3</i>	<i>ce</i>	<i>ee</i>	d(PP)	PAu_eP
1	2.689	2.964	2.989	2.885	7.394	5.527	137.5
1 (gas phase)	2.687	2.957	3.010	2.880	7.396	5.535	137.4
1_{Me-op}	2.665	2.964	3.029	2.882	7.441	5.511	142.7
1_{H-op}	2.647	2.994	3.063	2.872	7.479	5.490	144.2
2	2.669	2.987	3.043	2.901	7.502	5.550	139.7
2 (gas phase)	2.670	2.948	3.064	2.889	7.461	5.555	138.0
2_{Me-op}	2.655	2.981	3.011	2.893	7.475	5.519	143.0
2_{H-op}	2.643	3.001	3.002	2.873	7.450	5.456	143.8
[Au₆]²⁺	2.722	2.876	2.876	2.680	6.754	---	---
[Au₆]²⁺ (gas phase)	2.690	2.942	2.944	2.703	6.933	---	---

Table S2. Main geometry features of the C-H \cdots Au interaction in optimized geometries of **1** and **2** in CH₂Cl₂ and in the gas phase.

Cluster	Distances (Å)				Angle (°)
	C-H	H \cdots Au	BPL	$\Delta_{\text{BPL-d(H}\cdots\text{Au)}}$	
1 (gas phase)	1.088	2.636	2.675	0.039	174.8
1	1.088	2.630	2.669	0.038	175.2
2 (gas phase)	1.098	2.829	2.948	0.119	173.0
2	1.098	2.777	2.828	0.051	175.3

BPL: Bond path length.

Table S3. ¹H NMR chemical shifts (reference to TMS).

	1	mPhDP	1_H	1_{H-op}
H2	12.07	8.37	11.04	10.73
H4, H6	7.52, 7.27	7.36, 7.19	8.22, 7.98	8.18, 7.92
H5	7.70	7.51	7.75	7.70
	2	TMDP	2_H	2_{H-op}
H2	4.47	1.55	3.87	3.82
H1, H3	2.53	2.49	1.84	1.89
H2'	1.51	1.50	1.36	1.41

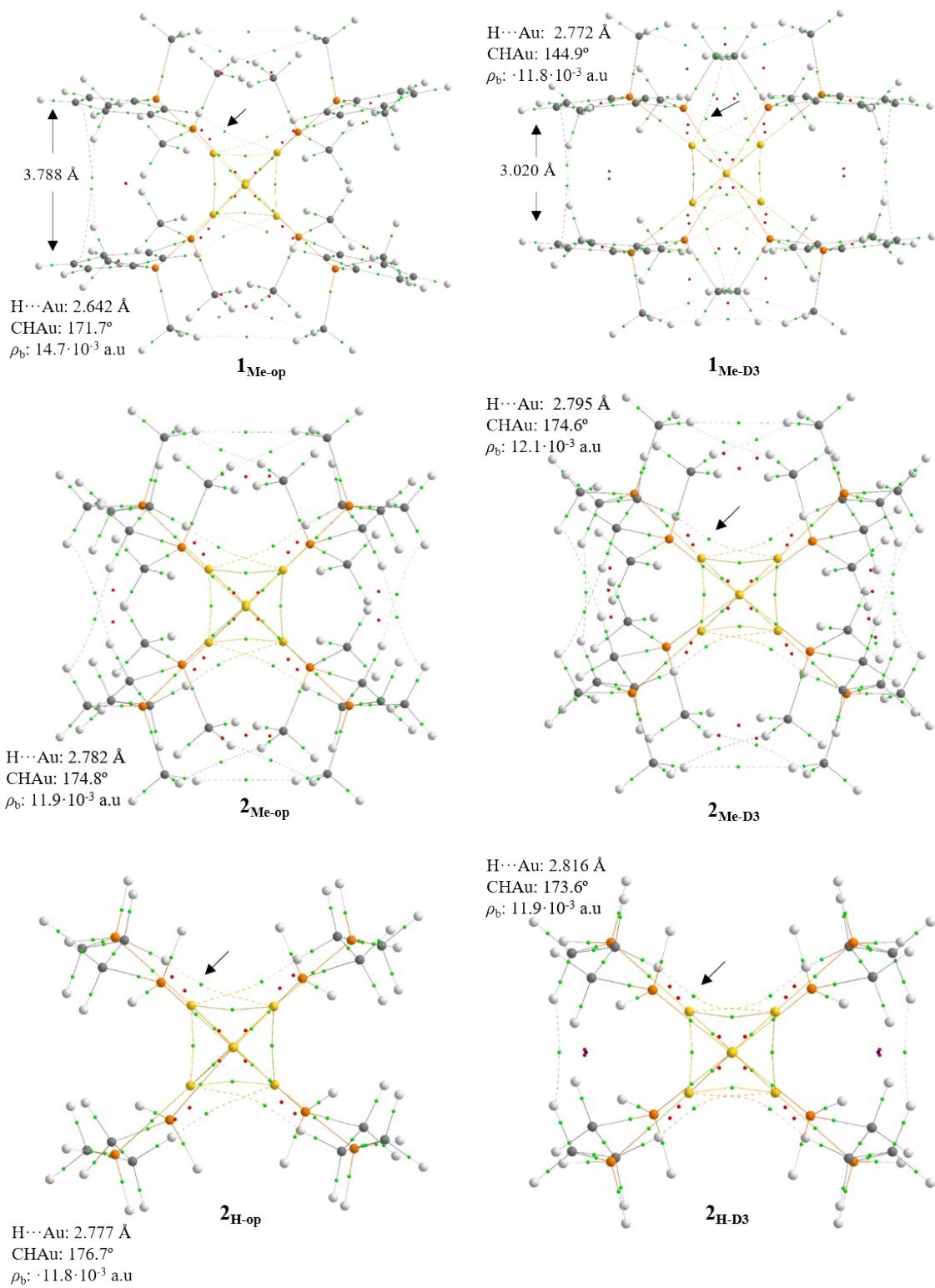


Figure S1. QTAIM plots of **1_{Me-op}**, **1_{Me-D3}**, **2_{Me-op}**, **2_{Me-D3}**, **2_{H-op}** and **2_{H-D3}**. An arrow is shown to localize one of the four equivalent CH...Au BCP's.

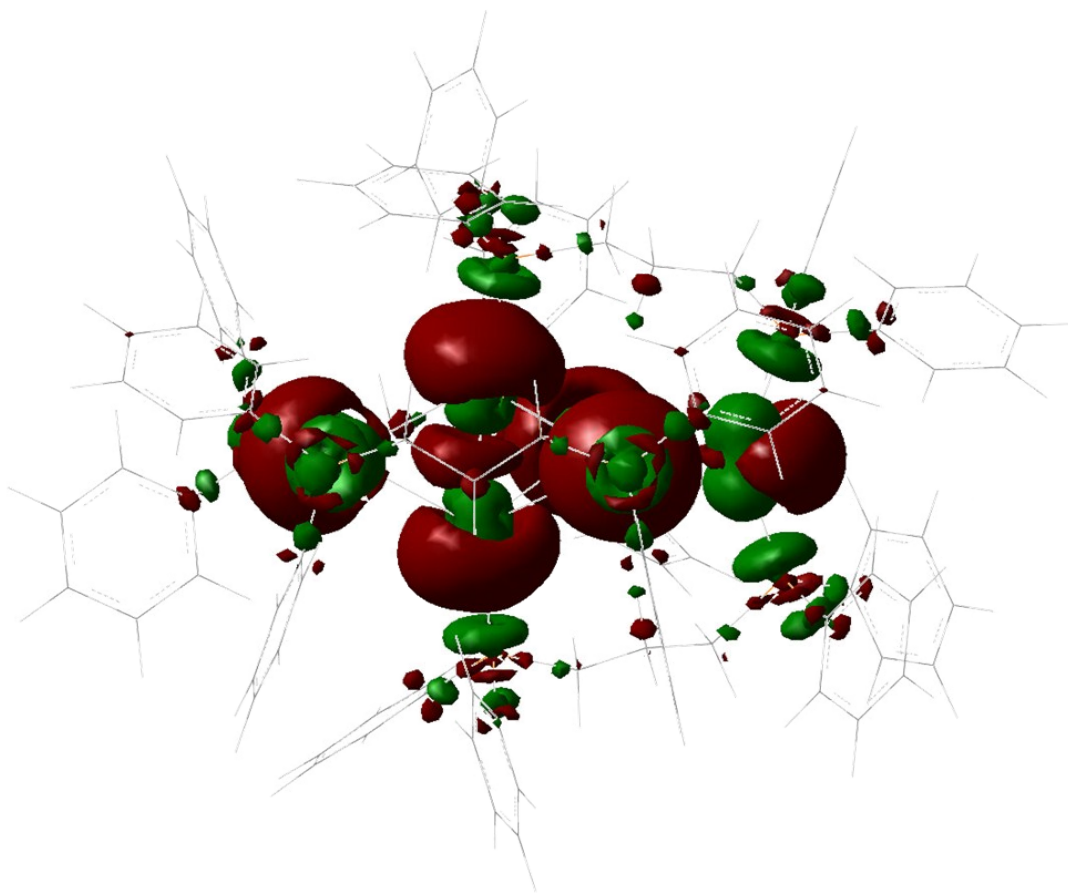


Figure S2. $\Delta\rho^{\text{def}}(\mathbf{r})$ of cluster 2 where red and green denote, respectively, enhancement and depletion of the electron density upon L-Au₆ binding (cutoff 0.0003 a.u.).