Supplementary Information

## Plasma-liquid synthesis of MoOx and WO<sub>3</sub> as potential photocatalysts

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I. Analysis of dyes solution after photocatalysis

The products of the destruction process were analyzed by gas chromatography-mass spectrometry (GC/MS) (Shimadzu GCMS QP2010 Ultra) in positive electrospray mode. Samples were extracted by organic solvents before analysis.



FIG. S1 Chromato-mass-spectrum of dyes solution after photocatalysis (destruction degree <100%)

Two kinetic models are analyzed and proved for all dyes adsorption. Figures S1-S2 (a-d) show the  $ln(q_e-q_t)=f(t)$  and  $t/q_t$  versus t for MoO<sub>x</sub> and WO<sub>3</sub>, respectively.



FIG. S2 The pseudo-first (1) and pseudo-second (2) order kinetic models plots for the adsorption of MB (a), RhB (b), and RR6C (c) dyes on the  $MoO_x$ 



FIG. S3 The pseudo-first (1) and pseudo-second (2) order kinetic models plots for the adsorption of MB (a), RhB (b), and RR6C (c) dyes on the WO<sub>3</sub>

Figure S4 (a-b) presents the  $q_t$  versus  $t^{1/2}$  for intraparticle diffusion model.



FIG. S4 Intraparticle diffusion model for adsorption of four dyes on MoO<sub>x</sub> (a) and WO<sub>3</sub> (b)