

Supporting Information

Accelerative charge transfer of Cd_{0.5}Zn_{0.5}S@ZnS core-shell nano-spheres via decoration of Ni₂P and g-C₃N₄ toward efficient visible-light-driven H₂ production

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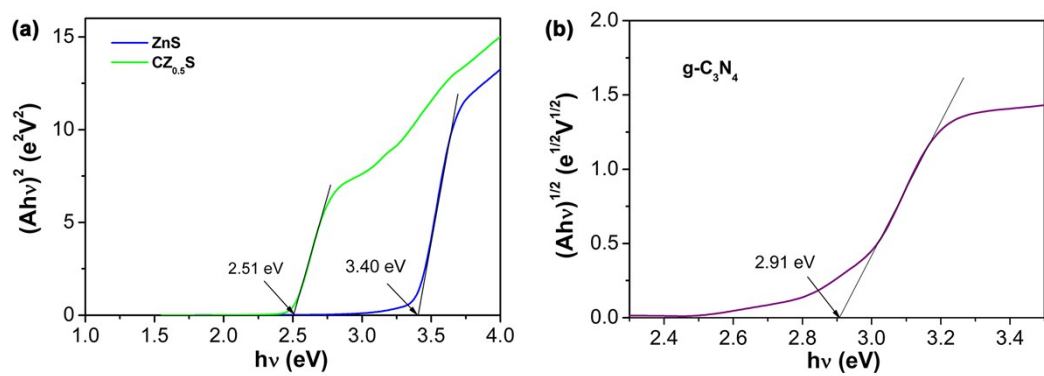


Fig. S1 Calculated bandgaps of (a) CZ_{0.5}S and ZnS, (b) g-C₃N₄.

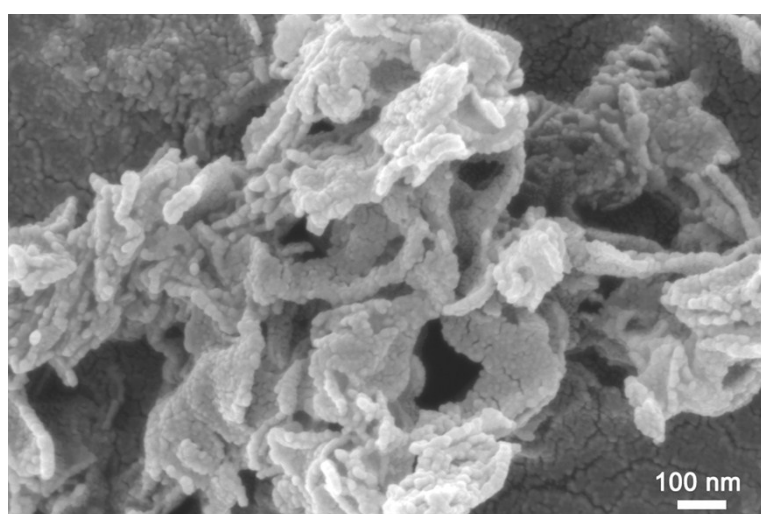


Fig. S2 SEM image of g-C₃N₄ nanosheets.

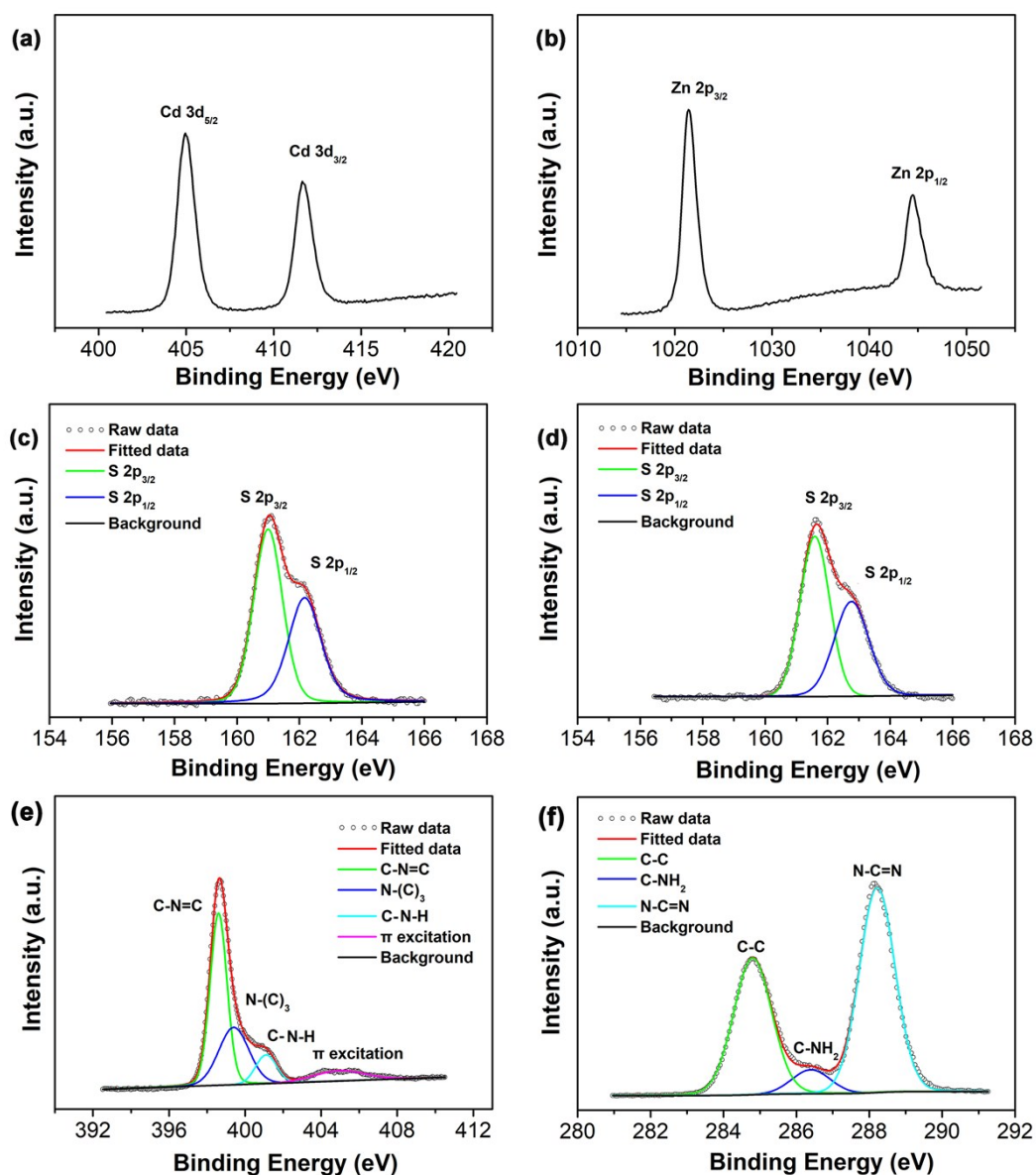


Fig. S3 (a) Cd 3d, (b) Zn 2p, and (c) S 2p XPS spectra of $\text{CZ}_{0.5}\text{S}@50\text{ZS}$. (d) S 2p XPS spectrum of $\text{CZ}_{0.5}\text{S}@50\text{ZS}-6\text{N}/8\text{CN}$. (e) N 1s and (f) C 1s XPS spectra of $\text{g-C}_3\text{N}_4$.

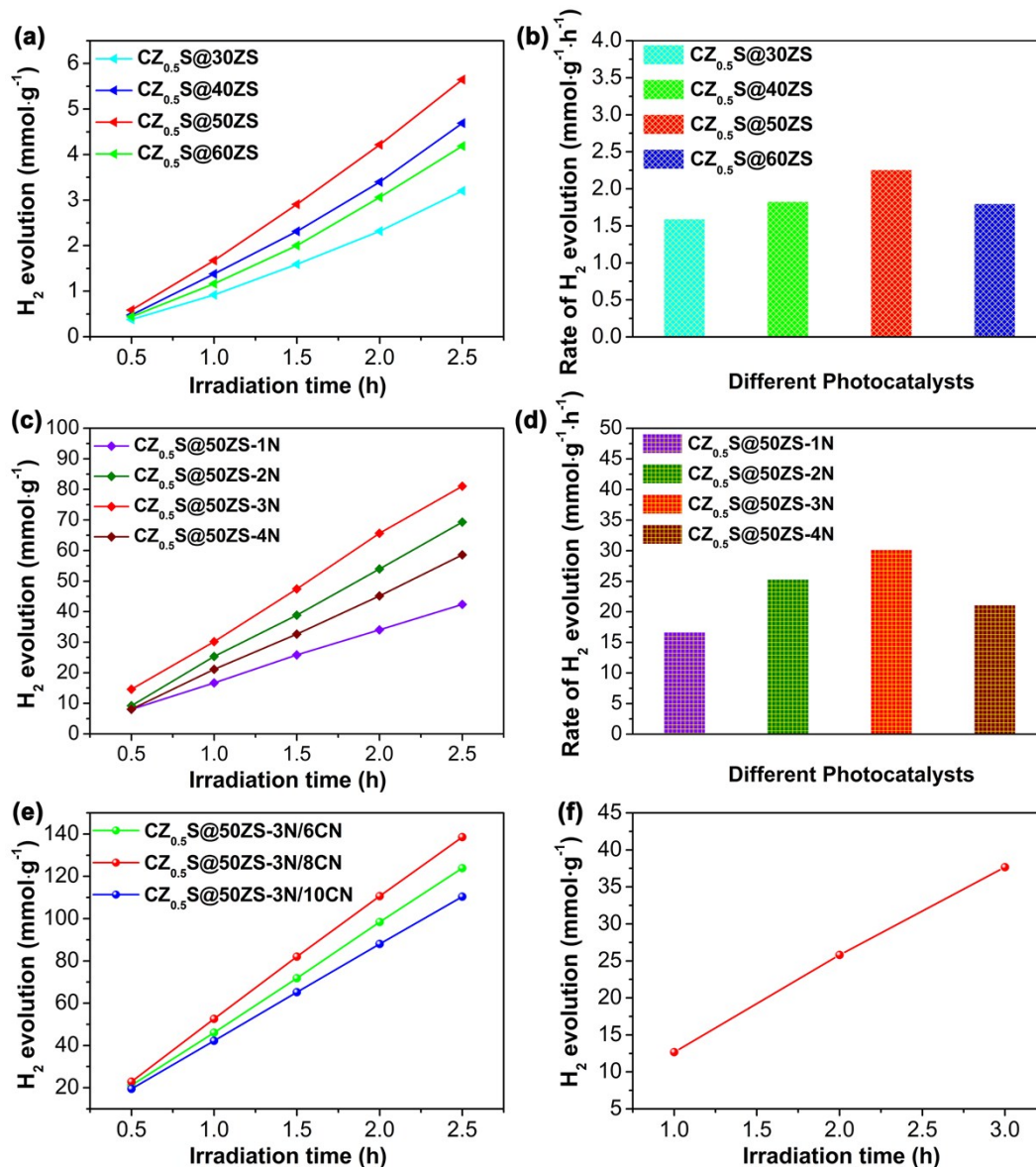


Fig. S4 (a) Visible-light-induced HER activities and (b) corresponding average rates of the $\text{CZ}_{0.5}\text{S}@Z\text{nS}$ composites containing different ZnS contents. (c) H_2 generation activities and (d) corresponding average rates of the $\text{CZ}_{0.5}\text{S}@50\text{ZS}-\text{Ni}_2\text{P}$ photocatalysts with varying Ni_2P concentrations. (e) The influence of g-C₃N₄ loading amount on the HER activity of $\text{CZ}_{0.5}\text{S}@50\text{ZS}-3\text{N}/\text{g-C}_3\text{N}_4$ hybrids. (f) Apparent quantum yield calculating curve of $\text{CZ}_{0.5}\text{S}@50\text{ZS}-3\text{N}/8\text{CN}$.

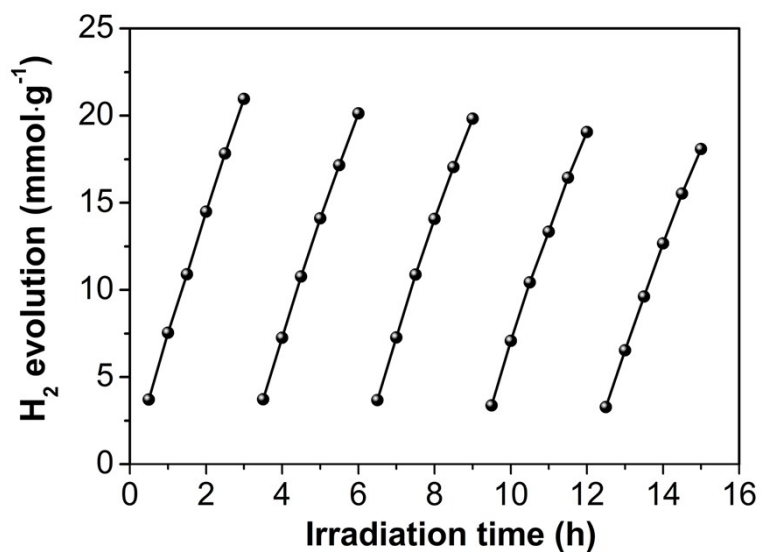


Fig. S5 Cycling H₂ evolution of the CZ_{0.5}S-3N/8CN composite.

Table S1 Comparison on the photocatalytic HER activities of CdS-based photocatalysts.

Photocatalyst	Hole scavenger (aqueous solution)	Light source (Xe lamp)	Maximum rate (mmol·h ⁻¹ ·g ⁻¹)	AQY (420 nm)	Referenc e
CZ _{0.5} S@ZS-Ni ₂ P/CN	Na ₂ S/Na ₂ SO ₃	λ > 420 nm	55.43	21%	This work
MnOx@CdS@GR	Na ₂ S/Na ₂ SO ₃	λ > 420 nm	5.45	12.4%	1
Ni ₂ P/MCdS-DETA	Na ₂ S/Na ₂ SO ₃	λ > 420 nm	6.84	-	2
ZnO/CdS-T120	Na ₂ S/Na ₂ SO ₃	λ > 420 nm	2.07	-	3
CdS/ZnS	Na ₂ S/Na ₂ SO ₃	λ > 420 nm	0.79	-	4
MoS ₂ -NiS/CdS	Lactic acid	λ > 420 nm	25.25	-	5
CdS-Ag ₂ S	TEOA	λ > 420 nm	7.5	-	6
SiCN@2CdS	Triethanolamine	λ > 420 nm	2.73	-	7
CdS/VN	Lactic acid	λ > 420 nm	6.24	5.3%	8
CdS-Co ₃ O ₄	Lactic acid	λ > 420 nm	10.14	9.7%	9
CoPe@CdS/rGO	Na ₂ S/Na ₂ SO ₃	λ > 420 nm	29.4	-	10
CdS/Mo ₂ C@C	Lactic acid	λ > 420 nm	5.54	4.9%	11
Cd _{0.5} Zn _{0.5} S/Ni ₂ P	Na ₂ S/Na ₂ SO ₃	λ > 420 nm	41.26	-	12
CDs/CdS	Lactic acid	λ > 420 nm	6.7	19.3%	13
CuS/CdS(H)/CdS(C)	Lactic acid	λ > 420 nm	2.03	-	14
CdS/CuS	Triethanolamine	λ > 420 nm	0.22	-	15
2D/2D Ni ₂ P/CdS	Lactic acid	λ ≥ 420 nm	17.95	4.2%	16
NiCd/CdS	Na ₂ S/Na ₂ SO ₃	λ > 410 nm	11.57	-	17
NiCo-LDH/P-CdS	Lactic acid	λ > 420 nm	8.66	14%	18
CdS/WS ₂	Lactic acid	λ > 420 nm	14.1	-	19
CdS/VC	Lactic acid	λ > 420 nm	14.2	8.7%	20

Ni ₂ P/Zn _{0.5} Cd _{0.5} S	Na ₂ S/Na ₂ SO ₃	λ > 420 nm	5.33	18.1%	21
MoS ₂ /CdS	Na ₂ S/Na ₂ SO ₃	λ > 420 nm	4.65	7.31%	22

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