

Electronic Supporting Information

**Impacts of Hydrogen Bonding Interactions with Np(V/VI)O<sub>2</sub>Cl<sub>4</sub> Complexes: Vibrational Spectroscopy, Redox Behavior, and Computational Analysis.**

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1. Crystallographic tables

**Table S1.** Select bond lengths (Å) and angles (°) for Compound **Np(V)Morph**.

Np(1)-O(1)	1.8304(13)	O(1)-Np(1)-O(2)	178.66(5)
Np(1)-O(2)	1.8356(13)	O(1)-Np(1)-Cl(4)	92.99(4)
Np(1)-Cl(4)	2.7358(7)	O(2)-Np(1)-Cl(4)	88.27(4)
Np(1)-Cl(2)	2.7466(7)	O(1)-Np(1)-Cl(2)	86.29(4)
Np(1)-Cl(3)	2.7497(8)	O(2)-Np(1)-Cl(2)	92.44(4)
Np(1)-Cl(1)	2.7622(8)	Cl(4)-Np(1)-Cl(2)	177.376(12)
O(3)-C(2)	1.424(2)	O(1)-Np(1)-Cl(3)	92.87(4)
O(3)-C(3)	1.425(2)	O(2)-Np(1)-Cl(3)	87.58(4)
O(4)-C(6)	1.422(2)	Cl(4)-Np(1)-Cl(3)	91.396(12)
O(4)-C(7)	1.425(2)	Cl(2)-Np(1)-Cl(3)	91.158(12)
O(5)-C(10)	1.422(2)	O(1)-Np(1)-Cl(1)	88.72(4)
O(5)-C(11)	1.422(2)	O(2)-Np(1)-Cl(1)	90.84(4)
N(1)-C(4)	1.487(2)	Cl(4)-Np(1)-Cl(1)	88.076(12)
N(1)-C(1)	1.492(2)	Cl(2)-Np(1)-Cl(1)	89.388(12)
N(2)-C(8)	1.492(2)	Cl(3)-Np(1)-Cl(1)	178.357(13)
N(2)-C(5)	1.494(2)	C(2)-O(3)-C(3)	110.12(13)
		C(6)-O(4)-C(7)	109.81(14)
		C(10)-O(5)-C(11)	110.05(14)
		C(4)-N(1)-C(1)	110.60(14)
		C(10)-O(5)-C(11)	110.05(14)
		C(8)-N(2)-C(5)	112.13(13)
		C(12)-N(3)-C(9)	110.75(14)

**Table S2.** Select bond lengths (Å) and angles (°) for Compound **Np(VI)Morph**

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Np(1)-O(1) <sup>a</sup>	1.7526(18)	O(1) <sup>a</sup> -Np(1)-O(1)	180.0
Np(1)-O(1)	1.7526(18)	O(1) <sup>a</sup> -Np(1)-Cl(2) <sup>a</sup>	90.65(7)
Np(1)-Cl(2) <sup>a</sup>	2.6558(8)	O(1)-Np(1)-Cl(2) <sup>a</sup>	89.35(7)
Np(1)-Cl(2)	2.6558(8)	O(1) <sup>a</sup> -Np(1)-Cl(2)	89.35(7)
Np(1)-Cl(1) <sup>a</sup>	2.6884(8)	O(1)-Np(1)-Cl(2)	90.65(7)
Np(1)-Cl(1)	2.6884(8)	Cl(2) <sup>a</sup> -Np(1)-Cl(2)	180.00(3)
O(2)-C(3)	1.417(4)	O(1) <sup>a</sup> -Np(1)-Cl(1) <sup>a</sup>	91.05(7)
O(2)-C(2)	1.423(4)	O(1)-Np(1)-Cl(1) <sup>a</sup>	88.95(7)
O(3)-C(8)	1.398(4)	Cl(2) <sup>a</sup> -Np(1)-Cl(1) <sup>a</sup>	91.14(3)
O(3)-C(5)	1.412(4)	Cl(2)-Np(1)-Cl(1) <sup>a</sup>	88.86(3)
N(1)-C(1)	1.474(4)	O(1) <sup>a</sup> -Np(1)-Cl(1)	88.95(7)
N(1)-C(4)	1.489(4)	O(1)-Np(1)-Cl(1)	91.05(7)
N(2)-C(6)	1.469(4)	Cl(2) <sup>a</sup> -Np(1)-Cl(1)	88.86(3)
N(2)-C(7)	1.485(4)	Cl(2)-Np(1)-Cl(1)	91.14(3)
C(1)-C(2)	1.498(5)	Cl(1) <sup>a</sup> -Np(1)-Cl(1)	180.00(3)
C(3)-C(4)	1.508(5)	C(3)-O(2)-C(2)	110.7(2)
C(5)-C(6)	1.475(5)	C(8)-O(3)-C(5)	110.3(3)
C(7)-C(8)	1.475(5)	C(1)-N(1)-C(4)	110.5(2)
		C(6)-N(2)-C(7)	110.9(2)

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Symmetry transformations used to generate equivalent atoms: a: -x,-y+1,-z+1

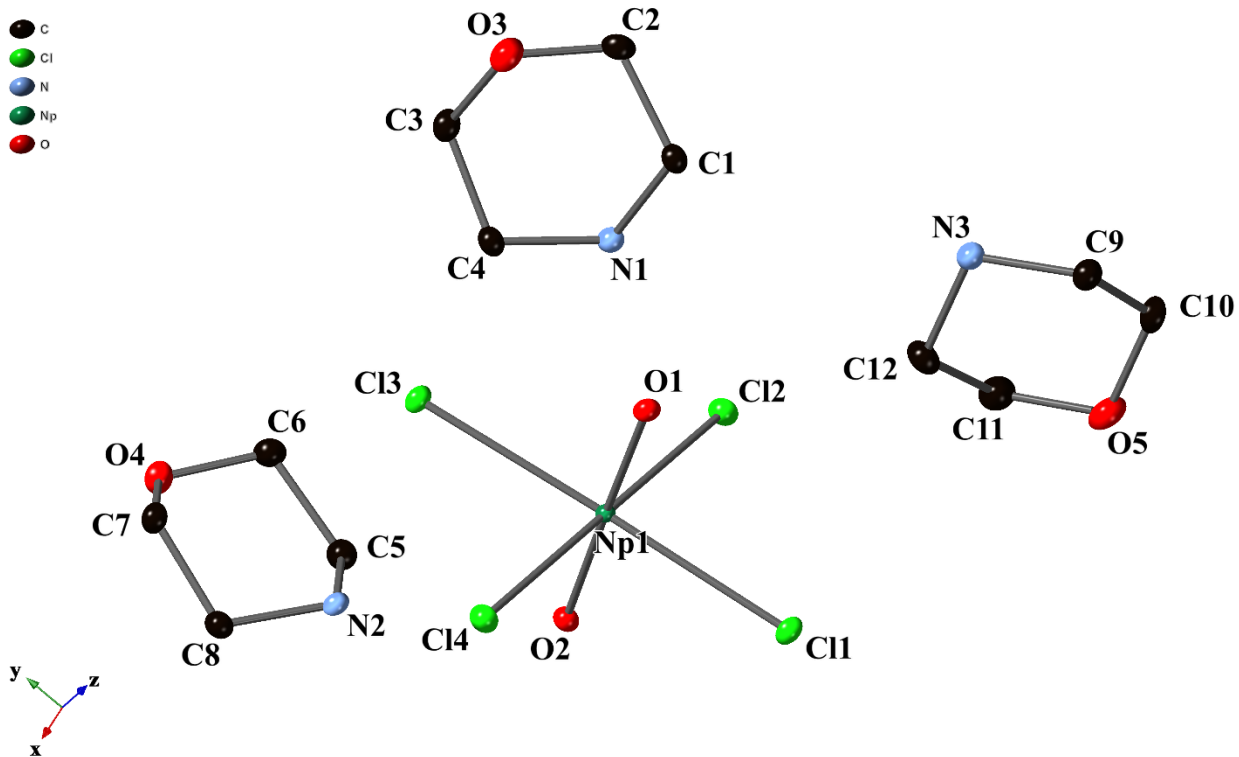


Figure S1. Thermal ellipsoid plots of Np(V)morph drawn at 50% probability.

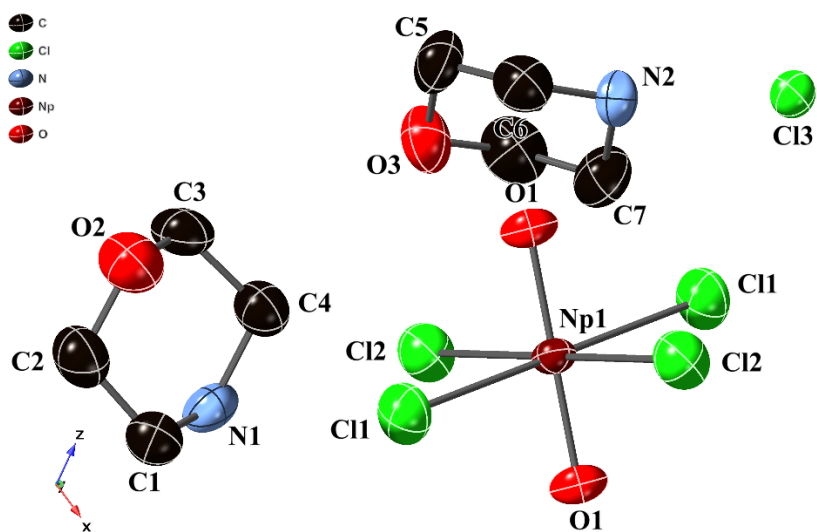


Figure S2. Thermal ellipsoid plots of Np(VI)morph drawn at 50% probability.

## I. Vibrational Spectra of Stocks

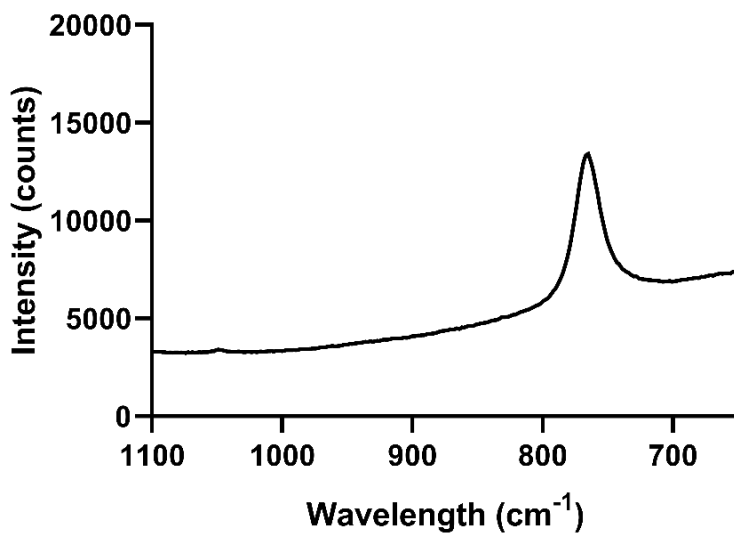


Figure S3. Solution state Raman spectra for Np(V) in 1M HCl stock.

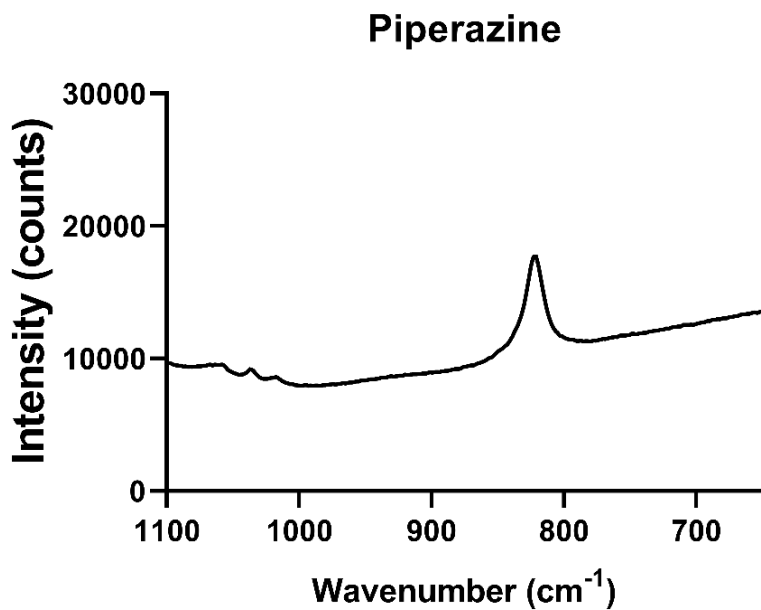
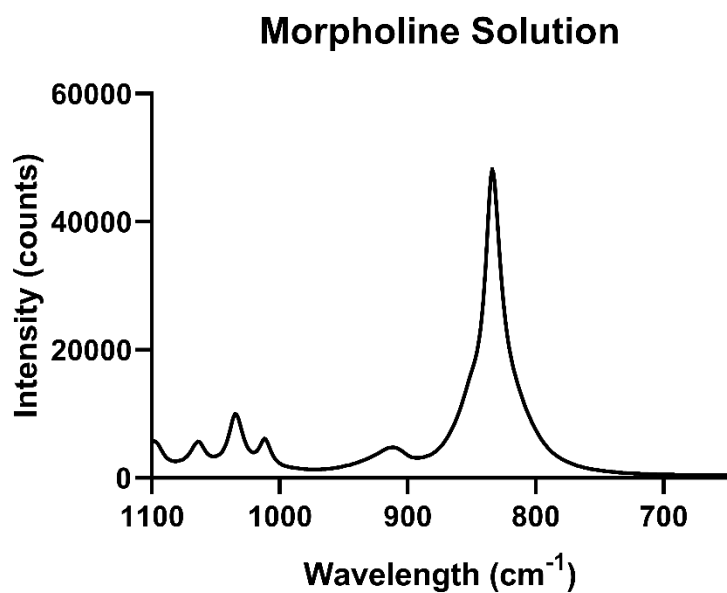
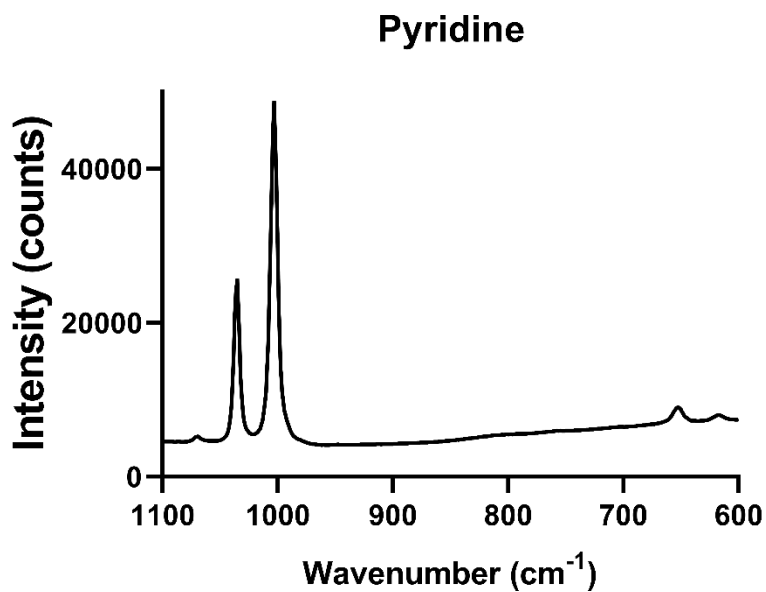


Figure S4. Solution state Raman spectra for piperazine.



**Figure S5.** Solution state Raman spectra for morpholine.



**Figure S6.** Solution state Raman spectra for pyridine.

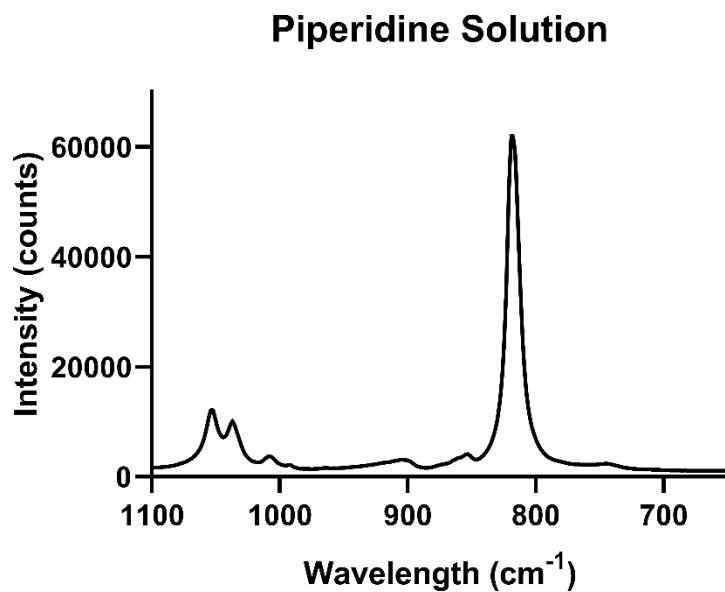


Figure S7. Solution state Raman spectra for piperidine.

## II. Solid-state Raman Spectroscopy of Np Compounds

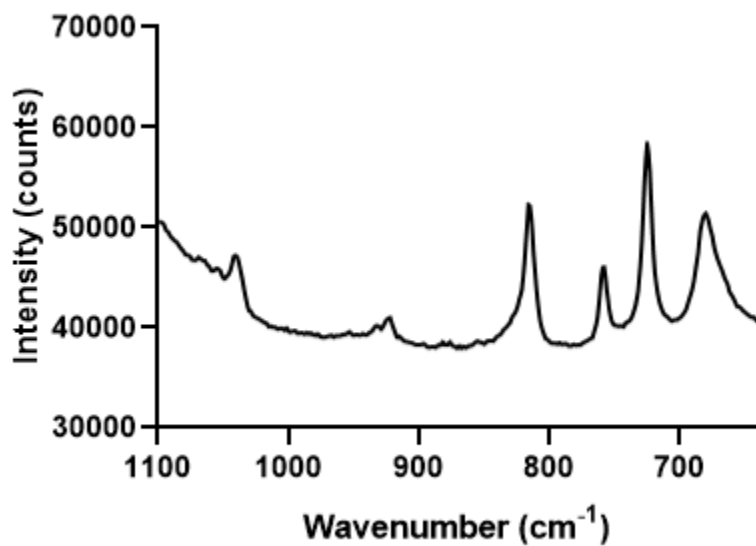
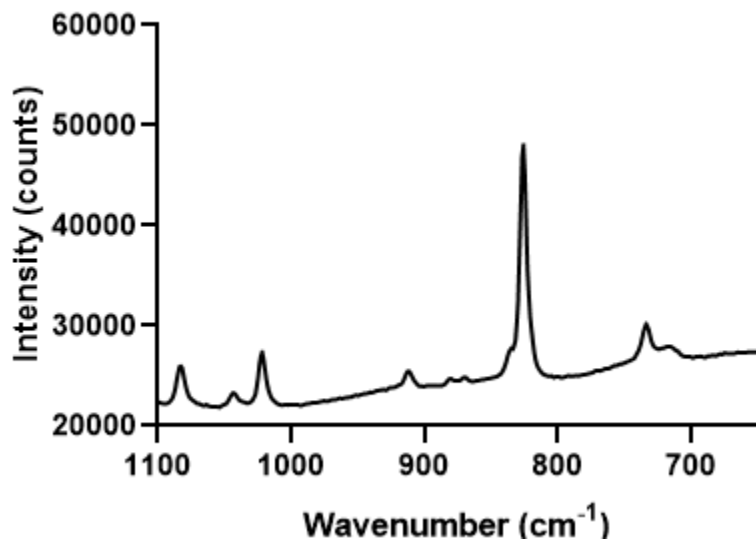
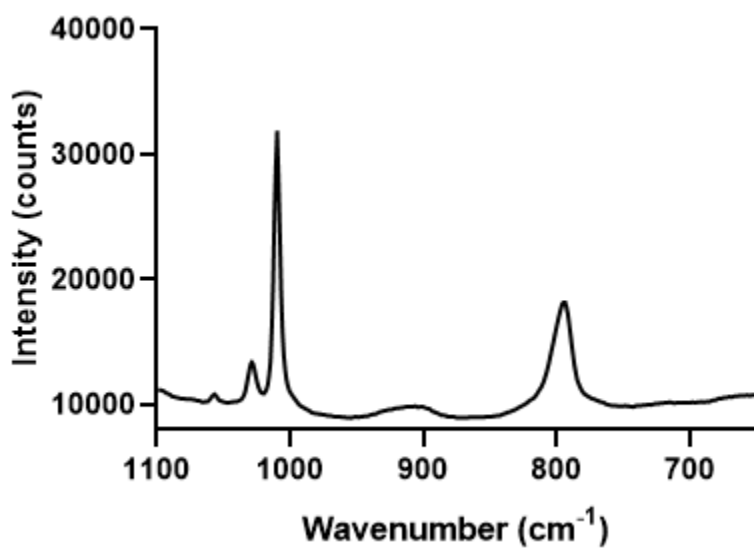


Figure S8. Solid state Raman spectra for compound **Np(V)pipz** containing Piperazine.

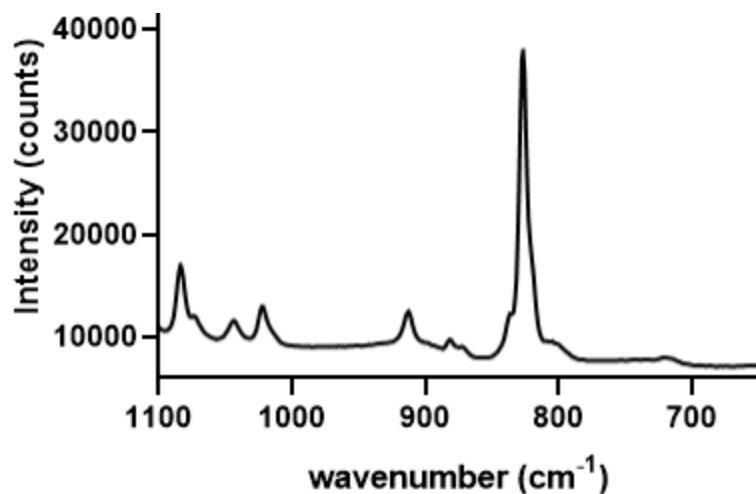


**Figure S9.** Solid state Raman spectra for **Np(V)morph** containing Morpholine.



**Figure S10.** Solid state Raman spectra for compound **Np(VI)pyr** containing Pyridine.



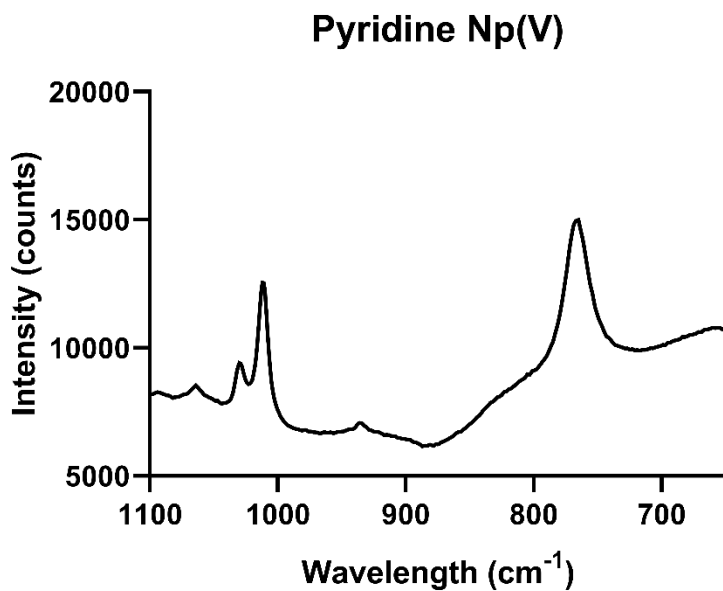


**Figure S11.** Solid state Raman spectra for **Np(VI)morph** containing morpholine.

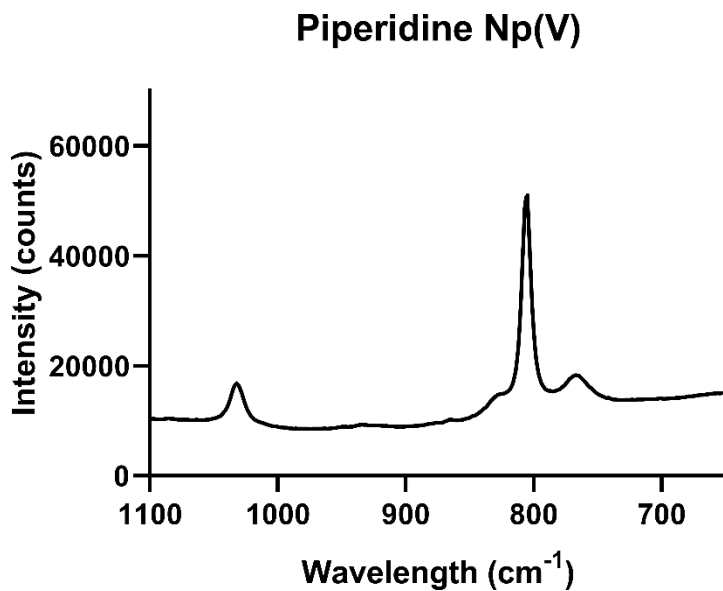
**Table S3:** Spectral results from figure S8-S10, including the peaks and their associated peak centers, full width half maxes.

		Peak 1	Peak 2	Peak 3	Peak 4	Peak 5	Peak 6
Np 1	Peak Centroid	$670 \pm 2$	$680 \pm 2$	$724 \pm 2$	$758 \pm 2$	$815 \pm 2$	$823 \pm 2$
	FWHM	$20 \pm 0.6$	$12 \pm 0.6$	$8 \pm 0.6$	$6 \pm 0.6$	$7 \pm 0.6$	$18 \pm 0.6$
Np2	Peak Centroid	$717 \pm 2$	$734 \pm 2$	$826 \pm 2$	$836 \pm 2$	-	-
	FWHM	$12 \pm 0.6$	$9 \pm 0.6$	$6 \pm 0.6$	$4 \pm 0.6$	-	-
Np3	Peak Centroid	$609 \pm 2$	$636 \pm 2$	$795 \pm 2$	-	-	-
	FWHM	$6 \pm 0.6$	$7 \pm 0.6$	$17 \pm 0.6$	-	-	-
Np4	Peak Centroid	$760 \pm 2$	$770 \pm 2$	$837 \pm 2$	-	-	-
	FWHM	$5 \pm 0.6$	$6 \pm 0.6$	$5 \pm 0.6$	-	-	-

### III. Solution Raman Spectroscopy of Np Heterocycles



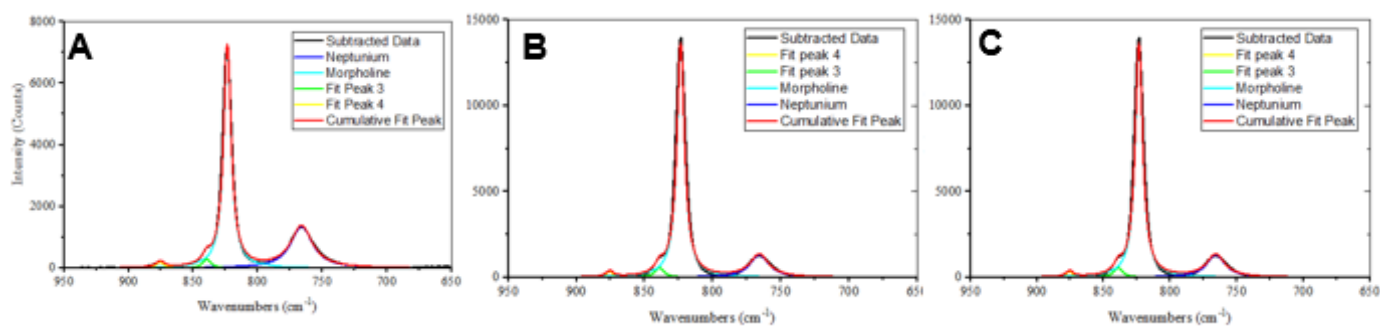
**Figure S12.** Solution state Raman spectra for Np(V) stock containing Pyridine.



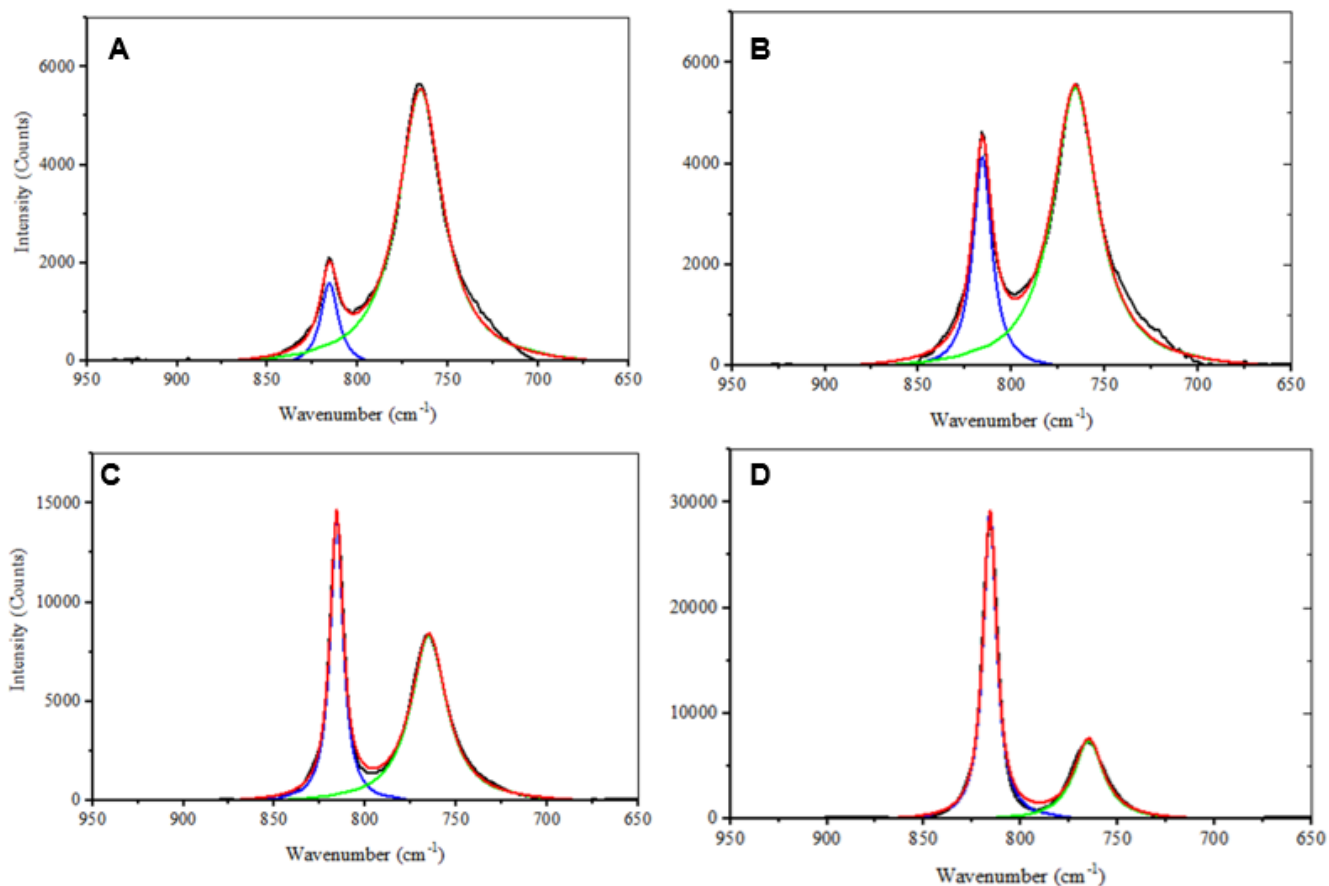
**Figure S13.** Solution state Raman spectra for Np(V) stock containing Piperidine.

**Table S4.** Raman band positions associated with the fitting of the Raman spectra from the Np-heterocycle solutions.

Raman Mode (cm <sup>-1</sup> )	Np(V)Piperazine	Np(V) Morpholine	Np(V) Pyridine	Np(V) Piperidine
	765	765	767	766
	815	823	805	805
	-	839	-	827
	-	875	-	-
	1044	1022	1011	1032
	1064	1046	1031	
		1077		



**Figure S14.** Peak fitting for morpholine titration experiments A (Np to morph 1:2), B (Np to morph 1:6) and C (Np to morph 1:8)



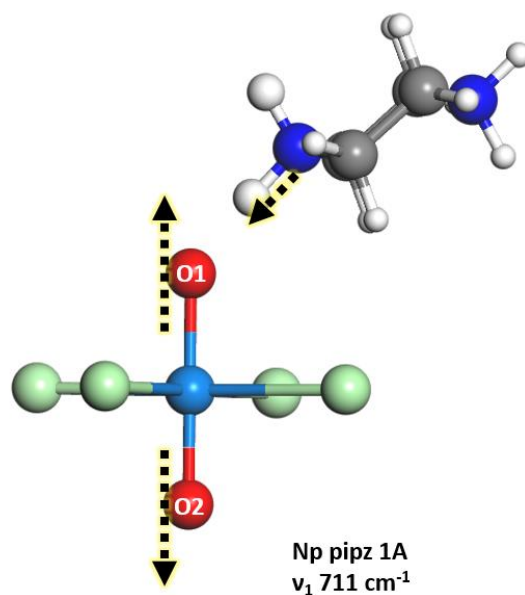
**Figure S15.** Peak fitting for piperazine titration experiments A (Np to Pipz 1:2), B (Np to Pipz 1:4), C (Np to Pipz 1:6), and D (Np to Pipz 1:9)

**Table S5:** The peak center, full width half max, and total area underneath the titration experiments for Np(V) piperazine, and Np(V) Morpholine.

<b>Piperazine</b>	<b>Center</b>	<b>FWHM</b>	<b>Area</b>
Np:Pipz 1:2	815	12.	31787
	765	30.	249642
Np:Pipz 1:4	815	12.5	80775.
	765	29.3	241147.
Np:Pipz 1:6	815.	8.	193276.
	765	23	300480.
Np:Pipz 1:9	815	8.	389697.
	764	18.	215567
<b>Morpholine</b>	<b>Peak center</b>	<b>FWHM</b>	<b>Area</b>
Np:Morph 1:2	875	7.5	2299
	839	7.5	3440
	823	7.3	81682

	766	20.4	41518
Np:Morph 1:6	875	6.5	3435
	839	8	7068
	823	7	147716
	765	17	33669
Np:Morph 1:8	875	7.5	7829
	839	7.5	10320
	823	7.3	280538
	765	14	24219

#### IV. Details on Computational Analysis



**Figure S16.** Example of the **NP pipz 1A** computational model which shows the concerted motions of Np=O stretch and NH<sub>2</sub> motions.

##### *a. XYZ Coordinates of DFT-Optimized Structures*

[NpO<sub>2</sub>Cl<sub>4</sub>]<sup>3-</sup>

Np	-0.0000318	-0.0002877	-0.0000670	Cl	-0.0296995	-2.7957120	-0.4139549
Cl	-0.5015225	0.4209561	-2.7483018	O	-1.7865627	-0.0265642	0.3141238
O	1.7865493	0.0259561	-0.3136038	Cl	0.0299151	2.7964836	0.4133047
Cl	0.5013522	-0.4208319	2.7484991				

*[NpO<sub>2</sub>Cl<sub>4</sub>]-1APipz (Long Pip)*

N	2.2217106	-1.1223717	-1.4598314	H	1.8121168	0.8502382	1.3026734
H	2.5114927	-0.4215675	-2.1462156	H	1.8800283	1.4152620	-0.3716958
H	2.7868979	-1.9494427	-1.6595999	C	2.5232980	-0.6330288	-0.0700152
N	0.2119316	0.2277783	0.0791736	H	3.5750074	-0.3623616	-0.0297132
H	-0.4189265	1.0436615	0.3198584	H	2.3440520	-1.4679958	0.6046710
H	-0.1070492	-0.4792286	0.7730667	Np	-3.1473369	1.1702687	1.7834446
C	0.7655941	-1.4510606	-1.6474980	Cl	-4.6153819	3.4780259	1.2781240
H	0.5482350	-2.3016071	-1.0035146	Cl	-3.9530373	0.0755799	-0.6640060
H	0.6174595	-1.7428241	-2.6837406	Cl	-2.2297610	2.1598566	4.2150128
C	-0.0883461	-0.2479621	-1.3042673	Cl	-1.5928250	-1.2035397	2.2104922
H	0.0965527	0.5815464	-1.9858805	O	-4.5627479	0.4167332	2.6128425
H	-1.1447918	-0.5082903	-1.3516709	O	-1.6867773	1.9105770	0.9323349
C	1.6526046	0.5617530	0.2659548				

*[NpO<sub>2</sub>Cl<sub>4</sub>]-1BPipz (Short Pip)*

N	-1.8014701	1.2144320	1.6768841	Cl	4.2096425	-3.0502267	-0.9830966
H	-1.0584588	1.9079669	1.7931177	O	1.4462333	-1.2913953	-1.6156703
H	-2.5746804	1.5444381	2.2571584	O	4.8217008	-0.0717247	-2.3315593
C	-0.2164056	-0.6499254	1.2944602	N	-0.6342150	-0.6925759	-0.1408709
H	0.6801618	-0.0302951	1.3467452	H	-1.3607735	-1.3974122	-0.2753638
H	0.0468243	-1.6585478	1.6037786	H	0.2089336	-0.9896786	-0.7434595
C	-1.3348310	-0.1264023	2.1729185	C	-2.2362912	1.1653657	0.2383765
H	-2.2025286	-0.7837923	2.1643120	H	-3.1166284	0.5265794	0.1965769
H	-0.9894636	-0.0050471	3.1962064	H	-2.5109593	2.1719069	-0.0656561
Np	3.1531803	-0.6705959	-1.9756154	C	-1.1121514	0.6365993	-0.6292587
Cl	2.0694381	1.7167543	-2.9384930	H	-0.2510458	1.3051519	-0.6363183
Cl	2.9728776	-1.7859615	-4.5318953	H	-1.4632777	0.5205196	-1.6515086
Cl	3.2541881	0.4938666	0.5782312				

*NpO<sub>2</sub>Cl<sub>4</sub>-2Pipz*

N	5.5107435	-2.1692555	-2.7883066	H	3.8721919	-3.3760152	-2.2820564
H	5.8197189	-1.5621412	-3.5514304	H	4.0518712	-3.0712957	-4.0182567
H	6.1351225	-2.9773987	-2.8134436	C	3.1503034	-1.4379638	-2.9477458
N	3.3014130	-0.7304490	-1.6381633	H	3.3476235	-0.7117476	-3.7344641
H	2.6309090	0.0899438	-1.6025554	H	2.1186850	-1.7722768	-3.0363721
H	2.9516283	-1.3327929	-0.8780407	C	4.7054637	-0.2863996	-1.3929705
C	4.0961426	-2.6185715	-3.0312644	H	4.7591806	0.1631236	-0.4041250

H	4.9402059	0.4741412	-2.1360138	Cl	-1.3130395	3.0122252	-1.9151109
C	5.6638154	-1.4580213	-1.4721828	Cl	-0.6167000	-0.5676757	-3.4057382
H	6.6913443	-1.1129716	-1.3935884	Cl	-0.0223081	1.7986703	1.5639473
H	5.4725969	-2.1922850	-0.6920252	Cl	0.7608065	-1.7674779	0.0297866
N	-4.7320813	2.0347765	3.6650496	O	-2.0333666	0.1133244	-0.4690300
H	-4.0258324	2.7738046	3.7044132	O	1.3829305	1.1322018	-1.3878471
H	-5.3955951	2.2523864	4.4105381	N	-3.8245887	0.3830064	1.4824928
C	-3.1380259	0.3406313	2.8130148	H	-4.5148024	-0.3673485	1.4185993
H	-2.2905284	1.0248699	2.7554880	H	-3.1102249	0.2120909	0.7105455
H	-2.7581442	-0.6662496	2.9680800	C	-5.4371498	2.0681433	2.3381736
C	-4.0946369	0.6992194	3.9320876	H	-6.2664325	1.3659379	2.4028501
H	-4.9036853	-0.0228405	4.0271596	H	-5.8296288	3.0704473	2.1885883
H	-3.5618533	0.7637753	4.8772042	C	-4.4752950	1.7060378	1.2250195
Np	-0.3291664	0.6353648	-0.9300323	H	-3.6765638	2.4408821	1.1255224
				H	-5.0130471	1.6461734	0.2822032

*NpO<sub>2</sub>Cl<sub>2</sub>-3Pipz*

N	2.5285188	-5.4664634	-2.0627774	O	-0.4683776	-0.9868652	-0.5341283
H	3.1979888	-4.9826577	-2.6664435	N	-0.1591873	3.4903995	-3.7016106
H	2.7584922	-6.4587367	-2.1437275	H	0.1128401	4.1601341	-4.4237123
N	0.9192306	-3.3547691	-0.9699857	H	-0.7924876	4.0016234	-3.0819316
H	0.6200092	-2.3595495	-0.8573161	N	-0.4313189	3.0596143	5.8041652
H	0.2220423	-3.8464433	-0.3706857	H	-0.6186005	4.0645142	5.8451744
C	1.1285507	-5.2369980	-2.5594407	H	-0.0543556	2.8162104	6.7220972
H	0.4762408	-5.8953185	-1.9882917	C	-0.8818005	2.3343879	-4.3364549
H	1.0907117	-5.5218563	-3.6073455	H	-0.2202469	1.9270741	-5.0990005
C	0.7449109	-3.7802971	-2.3936423	H	-1.7796050	2.7191746	-4.8126573
H	1.3632798	-3.1320164	-3.0138776	C	0.0409337	3.0608956	3.3711777
H	-0.2983855	-3.6408286	-2.6688361	H	-0.1941138	4.1164566	3.2432148
C	2.3138175	-3.5719688	-0.4792295	H	0.7650417	2.7689318	2.6141279
H	2.3590722	-3.2918687	0.5707689	C	1.0607988	3.0495168	-2.9423401
H	2.9707093	-2.9124659	-1.0450887	H	1.4924983	3.9222707	-2.4596980
C	2.7101772	-5.0263736	-0.6368373	H	1.7667099	2.6610500	-3.6741221
H	3.7551995	-5.1636669	-0.3727435	C	0.6003208	2.7704633	4.7496916
H	2.0984769	-5.6831442	-0.0212063	H	0.8839455	1.7251918	4.8611521
Np	-2.0909990	-1.1483472	0.3884726	H	1.4672635	3.3957661	4.9453176
Cl	-2.3673012	1.6586190	0.2373180	N	-0.0272552	0.8538091	-2.5397780
Cl	-3.3080869	-1.4830641	-2.0387854	H	-0.2857656	0.1559308	-1.7909655
Cl	-0.7844392	-0.8010736	2.8720488	H	0.6044181	0.3688409	-3.1795688
Cl	-1.6694348	-3.9077255	0.6597406	N	-1.2184356	2.2837137	3.1544220
O	-3.6513954	-1.2940923	1.2598257	H	-1.0121438	1.2567049	3.1557926

H	-1.6009583	2.4496844	2.2073227	C	-1.2488769	1.3028015	-3.2897835
C	0.6896708	2.0096890	-1.9042717	H	-1.6986588	0.4340066	-3.7626918
H	0.0204214	2.4101110	-1.1435590	H	-1.9455236	1.6957721	-2.5508999
H	1.5907143	1.6340499	-1.4257038	C	-2.2537027	2.5945008	4.1897539
C	-1.7128144	2.3065352	5.5759092	H	-2.5326909	3.6406516	4.0736815
H	-1.4908387	1.2498970	5.7161594	H	-3.1278932	1.9757179	4.0003449
H	-2.4273120	2.6218796	6.3314586				

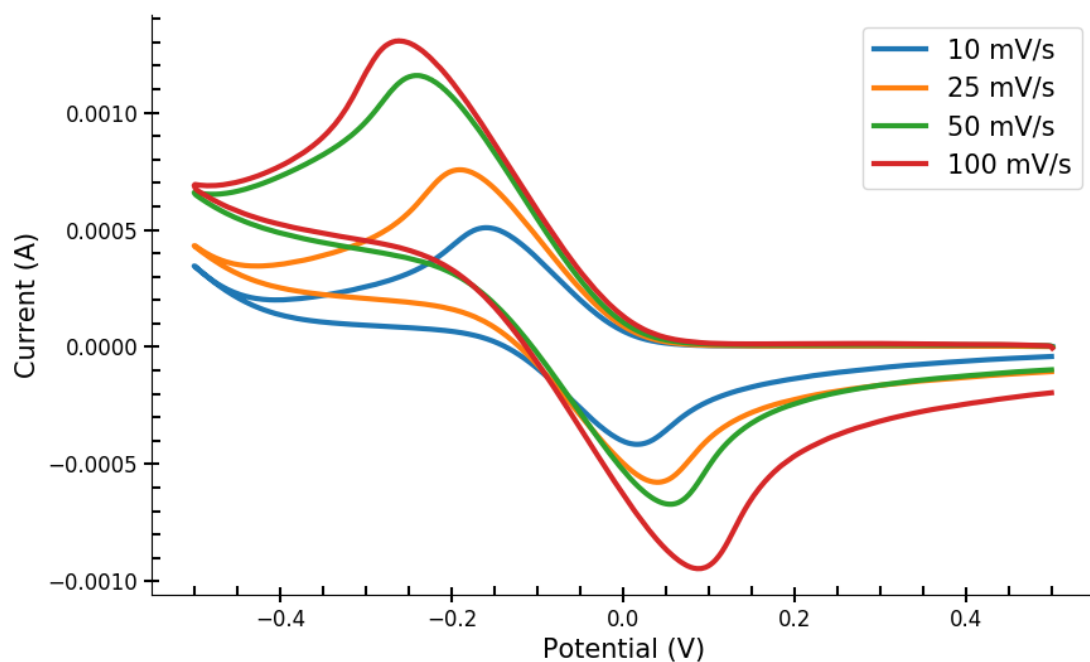
*NpO<sub>2</sub>Cl<sub>4</sub>-4Pipz*

N	4.1928635	-4.4701144	-1.6539485	O	0.6085248	-0.3479499	-0.3414514
H	4.7989783	-3.8746972	-2.2239706	N	-5.1693185	-1.3506574	1.3740671
H	4.6085736	-5.4025744	-1.6996665	H	-5.8851895	-1.8920363	0.8846573
N	2.1539048	-2.6687419	-0.7162650	H	-4.3636604	-1.2492986	0.7045405
H	1.7010802	-1.7273265	-0.6572038	C	-6.8420228	-0.0981205	2.7043322
H	1.5046627	-3.2544031	-0.1610252	H	-7.6926476	-0.6313068	2.2841860
C	2.8096316	-4.5077592	-2.2405880	H	-7.1621006	0.8916795	3.0185591
H	2.2522007	-5.2658392	-1.6936028	C	-5.6985957	0.0078979	1.7161412
H	2.8912637	-4.8086199	-3.2814839	H	-4.8663364	0.5840418	2.1205246
C	2.1601023	-3.1434207	-2.1388240	H	-6.0442217	0.4847820	0.8025880
H	2.6958449	-2.4012941	-2.7294726	N	1.4585411	3.8032710	-4.7513562
H	1.1306482	-3.1903728	-2.4864612	H	1.9798704	4.2990813	-5.4771835
C	3.5276979	-2.6356609	-0.1233821	H	0.8782517	4.5154518	-4.3008219
H	3.4458286	-2.3607756	0.9258245	N	2.6022715	3.2623062	3.5816285
H	4.0935911	-1.8678919	-0.6490710	H	3.4570739	3.5096823	3.0763488
C	4.1933364	-3.9934820	-0.2293776	H	2.6445773	3.7855137	4.4583522
H	5.2245995	-3.9324969	0.1072050	C	0.5899152	2.7553026	-5.3882577
H	3.6704877	-4.7450135	0.3583085	H	1.2411077	2.1235202	-5.9891574
N	-6.4077632	-0.8580231	3.9269899	H	-0.1205100	3.2558555	-6.0406350
H	-5.7170691	-0.3143658	4.4503058	C	2.4602644	0.9809885	2.6044109
H	-7.2028980	-0.9647572	4.5599096	H	3.3532815	1.0628900	1.9867144
C	-4.7170587	-2.0898237	2.5962794	H	2.3020288	-0.0634921	2.8652729
H	-3.8689221	-1.5416576	3.0068867	C	2.4216455	3.2173409	-3.7599136
H	-4.3815906	-3.0793497	2.2966062	H	2.9681674	4.0328359	-3.2941746
C	-5.8517861	-2.2146435	3.5925222	H	3.1182767	2.5959731	-4.3192614
H	-6.6755454	-2.8061501	3.1970359	C	2.5844318	1.7908708	3.8808763
H	-5.4968406	-2.6695646	4.5133548	H	1.7491617	1.6132152	4.5554478
Np	-1.2379535	-0.6827780	-0.2953427	H	3.5106727	1.5388082	4.3903397
Cl	-1.5318252	1.9822083	-0.9169943	N	0.8228144	1.3696342	-3.3476905
Cl	-1.0837800	-1.2403944	-2.9760376	H	0.2980651	0.8456676	-2.6299551
Cl	-1.3501964	-0.0707560	2.4268353	H	1.4079767	0.6796851	-3.8244674
Cl	-0.7191085	-3.3289603	0.3071579	N	1.2915317	1.4367622	1.7839337
O	-3.0323969	-0.9948862	-0.2707791	H	0.3979506	1.1535852	2.2420448

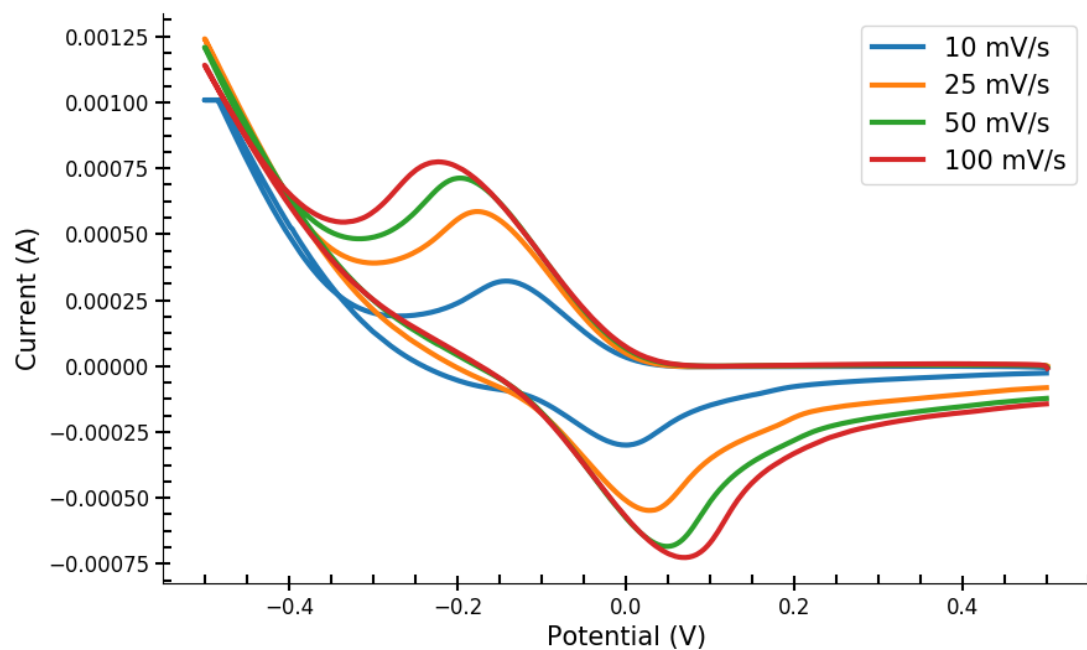


H	1.2496816	0.8968990	0.8952747	C	-0.1490826	1.9653058	-4.3294899
C	1.6765369	2.4228665	-2.7062893	H	-0.7034756	1.1514004	-4.7889040
H	1.0087399	3.0482817	-2.1160022	H	-0.8367139	2.5829422	-3.7540546
H	2.3903750	1.9318860	-2.0491664	C	1.3103017	2.9088816	1.5048603
C	1.3965503	3.6931774	2.7969806	H	2.1652458	3.1116834	0.8607607
H	0.5279179	3.5275734	3.4315081	H	0.3951108	3.1641690	0.9743305
H	1.4864514	4.7555389	2.5878259				

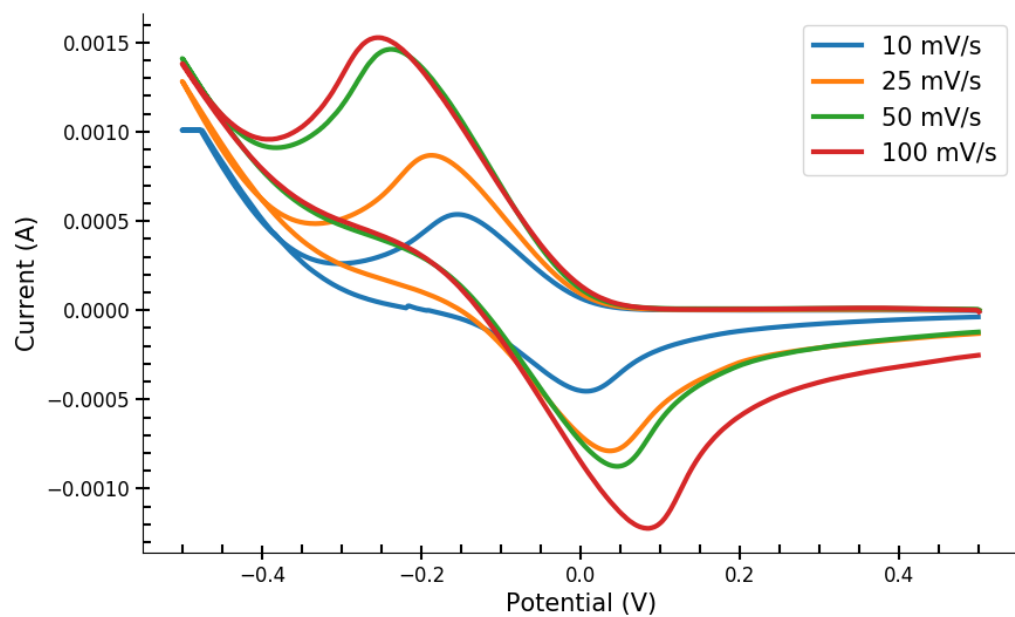
## V. Electrochemical analysis



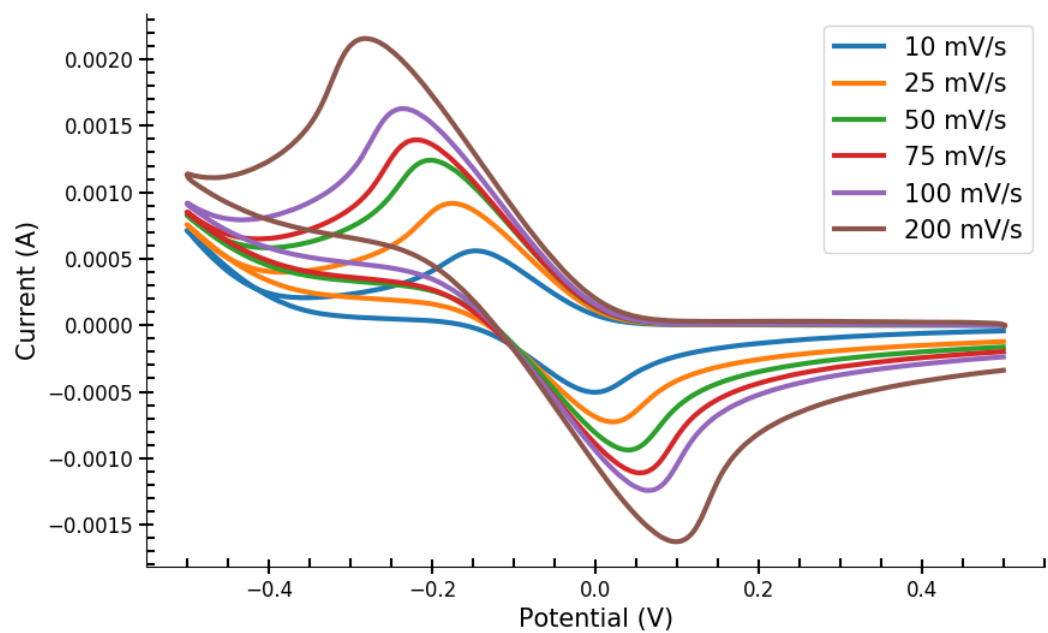
**Figure S17.** Overlay displaying varying scan rates of a 33.3mM  $\text{Np(V)O}_2$  solution with added morpholine. The potential was measured against Ag|AgCl.



**Figure S18.** Overlay displaying varying scan rates of a 33.3mM Np(V)O<sub>2</sub> solution with added piperidine. The potential was measured against Ag|AgCl.



**Figure S19.** Overlay displaying varying scan rates of a 33.3mM Np(V)O<sub>2</sub> solution with added pyridine. The potential was measured against Ag|AgCl.



**Figure S20.** Overlay displaying varying scan rates of the 33.3mM  $\text{Np(V)O}_2$  stock solution . The potential was measured against Ag|AgCl.