

Supporting Information

for

Temperature-dependent hysteretic two-step spin crossover in two-dimensional Hofmann-type compounds

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Table S1. Crystal structure and refinement details for compound **1** at 298, 195, 85 K and **2** at 298, 190, 89 K.

	Compound 1			Compound 2		
Temp / K	298	195	85	298	190	89
formula	FePdC ₂₈ N ₈ O ₃	Fe ₂ Pd ₂ C ₅₆ N ₁₆	Fe ₂ Pd ₂ C ₅₆ N ₁	Fe ₂ Pt ₂ C ₅₆ N ₁₆	FePtC ₂₈ N ₈ O ₃	FePtC ₂₈ N ₈ O ₃
	H ₂₆	O ₆ H ₅₆	₆ O ₆ H ₅₆	O ₆ H ₅₆	H ₂₆	H ₂₆
<i>M_r</i>	684.82	1369.64	1369.64	1546.98	773.49	773.49
cryst syst	orthorhombic			orthorhombic		
space group	<i>Cmmm</i>	<i>Imma</i>	<i>Imma</i>	<i>Imma</i>	<i>Cmmm</i>	<i>Cmmm</i>
<i>a</i> , Å	7.4125(6)	7.2795(5)	14.6772(11)	7.4043(4)	7.2305(3)	7.1424(4)
<i>b</i> , Å	28.588(3)	14.9441(13)	7.1611(4)	15.1012(6)	28.3930(14)	28.216(3)
<i>c</i> , Å	7.5142(6)	28.394(2)	28.221(2)	28.6158(14)	7.3865(5)	7.3155(7)
α , deg	90	90	90	90	90	90
β , deg	90	90	90	90	90	90
γ , deg	90	90	90	90	90	90
<i>V</i> , Å ³	1592.3(2)	3088.8(4)	2966.2(3)	3199.6(3)	1516.42	1474.3(2)
<i>Z</i>	2	2	2	2	2	2
<i>D_c</i> , g cm ⁻³	1.316	1.356	1.412	1.494	1.576	1.621
μ , mm ⁻¹	1.048	1.080	1.125	4.848	5.115	5.261
<i>F</i> (000)	632	1264	1264	1392	696	696
goodness-of-fit on <i>F</i> ²	1.014	1.177	1.169	1.060	1.067	1.132
<i>R</i> 1 (<i>I</i> ≥ 2σ(<i>I</i>)) ^a	0.0345	0.0741	0.0572	0.0348	0.0451	0.0295
w <i>R</i> 2 (all data) ^a	0.0892	0.1827	0.1481	0.0874	0.1176	0.0752

a: $I \geq 2\sigma(I)$: $R1 = \sum ||F_o| - |F_c|| / \sum |F_o|$, $wR2 = \{\sum [w(F_o^2 - F_c^2)^2] / \sum [w(F_o^2)^2]\}^{1/2}$.

Table S2. The Fe^{II}–N bond lengths in compounds **1** and **2** at different temperatures.

<i>T</i> / K	1			2		
	298	195	85	298	190	89
Fe–N1 (Å)	2.195(3)	2.120(4)	2.033(5)	2.206(3)	2.084(7)	2.020(4)
Fe–N3 (Å)	2.146(3)	2.068(5)	1.983(4)	2.160(4)	2.027(7)	1.966(4)
Fe–N _{ave} (Å)	2.170(8)	2.094(45)	2.008(45)	2.183(35)	2.056(2)	1.993(4)

Table S3. Fe···Fe distances in compounds **1** and **2** at different temperatures.

<i>T</i> / K	1			2		
	298	195	85	298	190	89
<i>d</i> ₁ (Å)	7.51(4)	7.47(2)	7.33(9)	7.55(1)	7.38(7)	7.31(6)
<i>d</i> ₂ (Å)	7.41(2)	7.27(9)	7.16(1)	7.40(4)	7.23(1)	7.14(2)
<i>d</i> _{ave} (Å)	7.46(3)	7.37(55)	7.25	7.47(75)	7.30(9)	7.22(9)

Table S4. The $\pi\cdots\pi$ interactions and the distortion parameter Σ_{Fe} of {FeN₆} moiety at different temperatures for compounds **1** and **2**.

<i>T</i> / K	1			2		
	298	195	85	298	190	89
$\pi\cdots\pi$ interactions (Å)	3.771(9)	3.692(11)	3.653(8)	3.732(12)	3.673(17)	3.653(9)
Σ_{Fe} of {FeN ₆ } (°)						
$\sum_{i=1}^{12} \varphi_i - 90^\circ $ (°)	30.7(26)	44.2(7)	37.0	28.4	37.6	32.8

Table S5. The pore sizes in compound **1** and **2** at different temperature.

		1			2		
<i>T</i> / K		298	195	85	298	190	89
pore size	A	14.294 ×	14.144 ×	14.110 ×	14.308 ×	14.196 ×	14.108 ×
		3.572	3.084	2.837	3.786	3.420	3.505
	B	14.294 ×	14.144 ×	14.110 ×	14.308 ×	14.196 ×	14.108 ×
		3.572	3.898	3.931	4.295	3.420	3.505

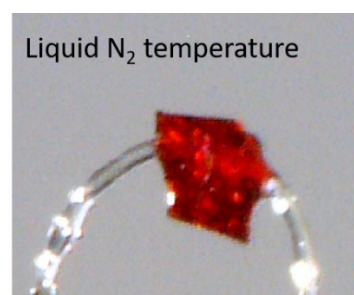
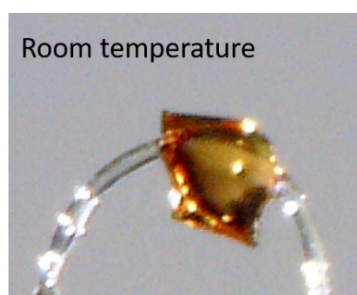


Fig. S1. Thermochromism of compound 1.

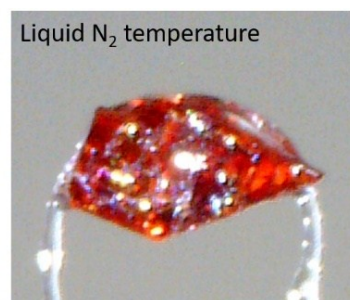


Fig. S2. Thermochromism of compound 2.

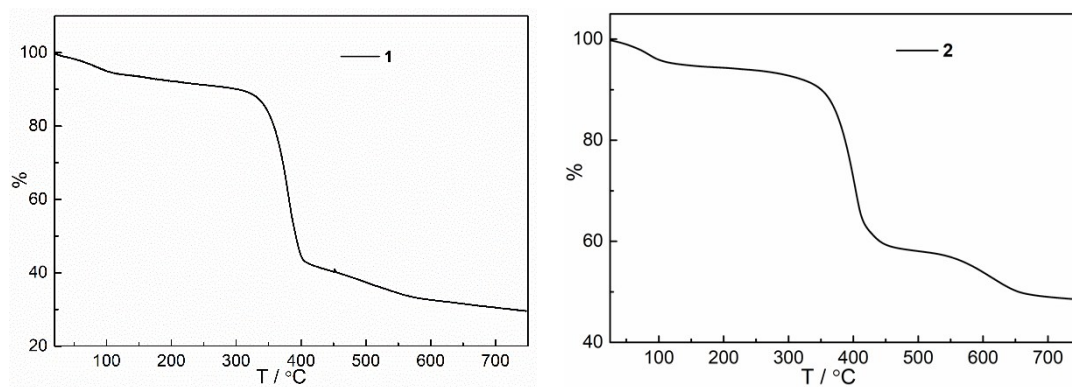


Fig. S3. TGA of compounds **1** and **2**.

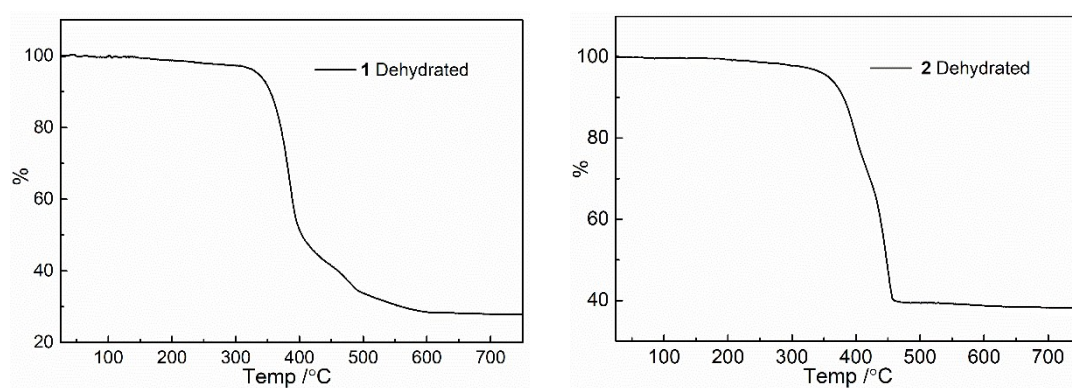


Fig. S4. TGA of dehydrated **1** and **2**.

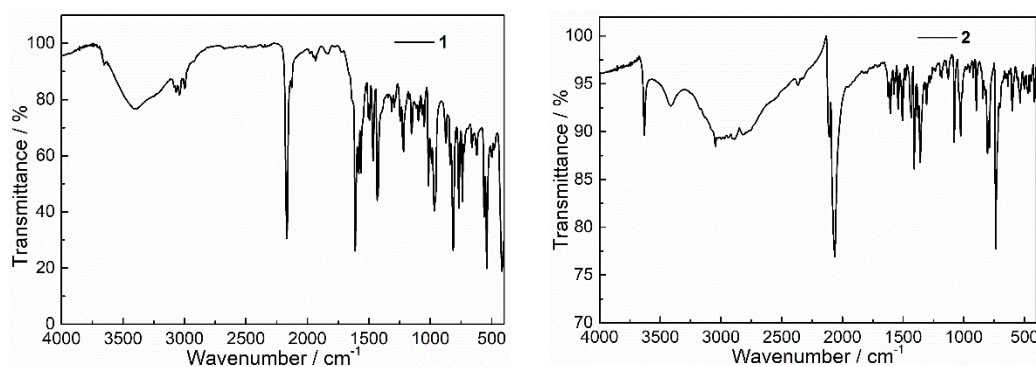


Fig. S5. The FT-IR spectra of **1** and **2**.

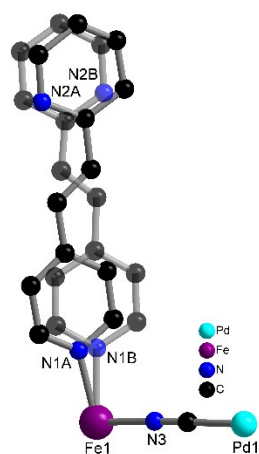


Fig. S6. The asymmetric unit of compound **1** at 195 K.

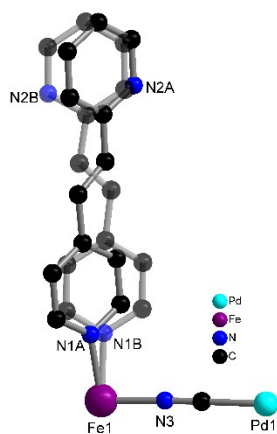


Fig. S7. The asymmetric unit of compound **1** at 85 K.

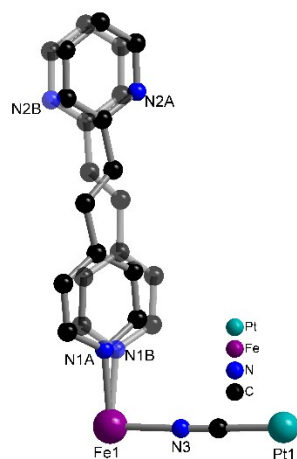


Fig. S8. The asymmetric unit of compound **2** at 298 K.

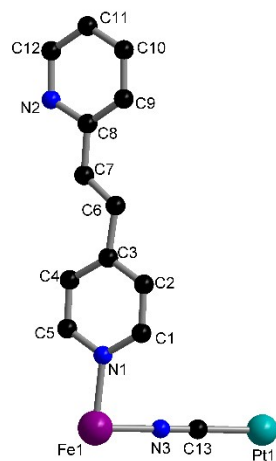


Fig. S9. The asymmetric unit of compound **2** at 190 K.

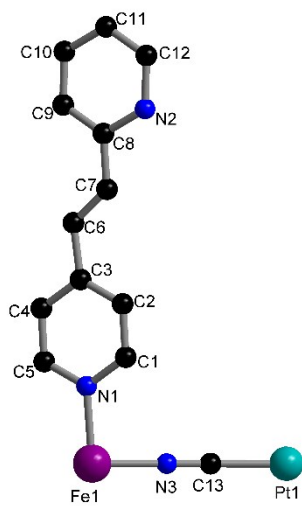


Fig. S10. The asymmetric unit of compound **2** at 89 K.