

Supporting Information for

Solid-State 1D → 3D Transformation of Polynitrile- Based Coordination Polymers by Dehydration Reaction

*Artem O. Dmitrienko,^{a,b} Mikhail I. Buzin,^b Zouaoui Setifi,^{c,d} Eugeny V. Alexandrov,^{e,f}
Eugenia D. Voronova,^a Fatima Setifi,^d and Anna V. Vologzhanina^{a,*}*

a A. N. Nesmeyanov Institute of Organoelement Compounds, RAS. 28 Vavilova str. 119991
Moscow, Russia.

b Lomonosov Moscow State University. GSP1 Leninskie Gory 119991 Moscow, Russia.

c Département de Technologie, Faculté de Technologie, Université 20 Août 1955-Skikda, Skikda
21000, Algeria.

d Laboratoire de Chimie, Ingénierie Moléculaire et Nanostructures (LCIMN), Université Ferhat
Abbas Sétif 1, Sétif 19000, Algeria.

e Samara Polytech, Molodogvardeyskaya str. 244, 443100 Samara, Russia.

f Samara Branch of P. N. Lebedev Physical Institute of the Russian Academy of Sciences, Novo-
Sadovaya str. 221, 443011 Samara, Russia

Powder XRD data of **1** and **2**.

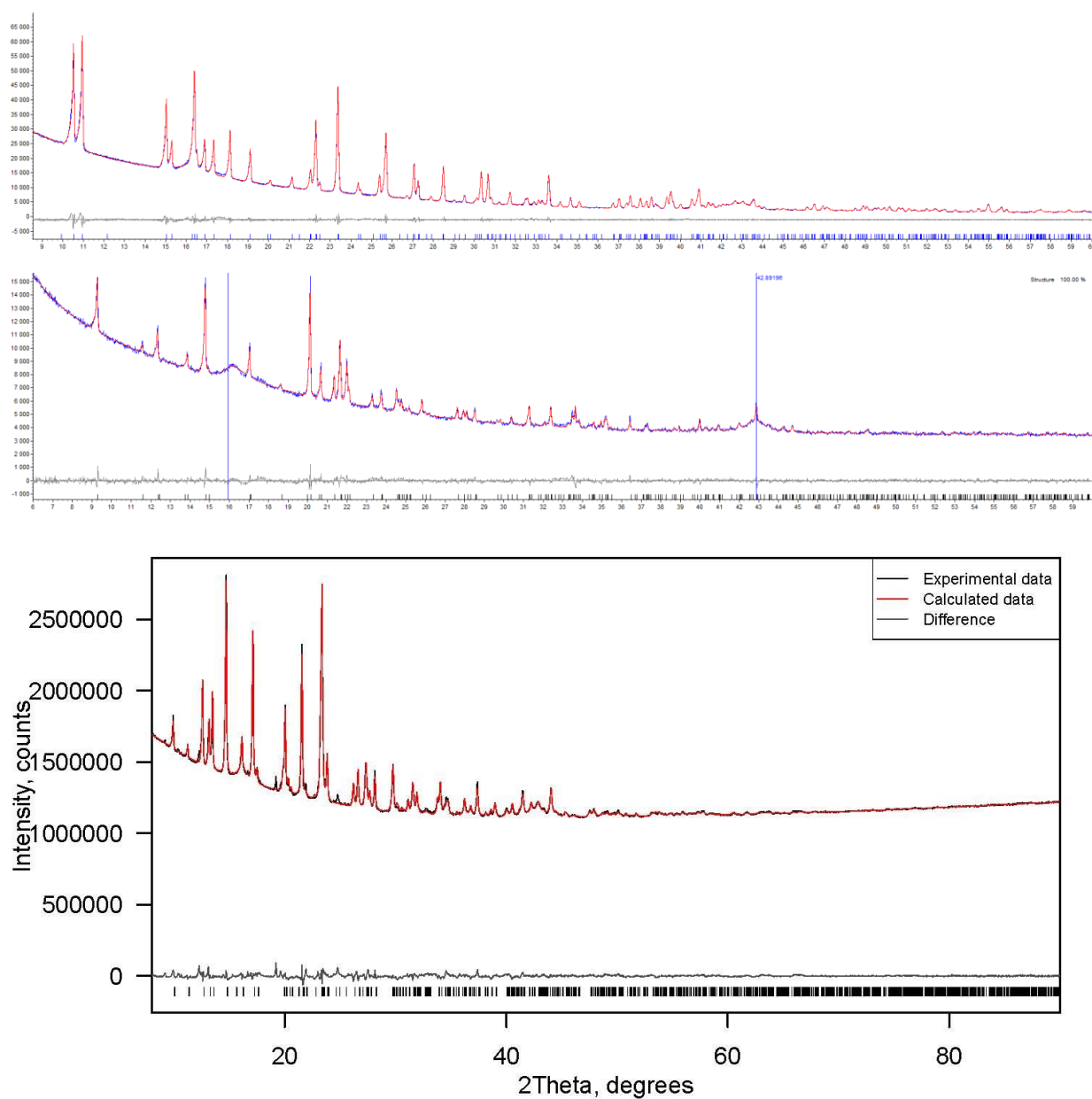


Figure S1. Observed (black), calculated (red) and difference profiles (gray) for the Rietveld refinement of (top) **1**, (middle) **2** and (bottom) **4**.

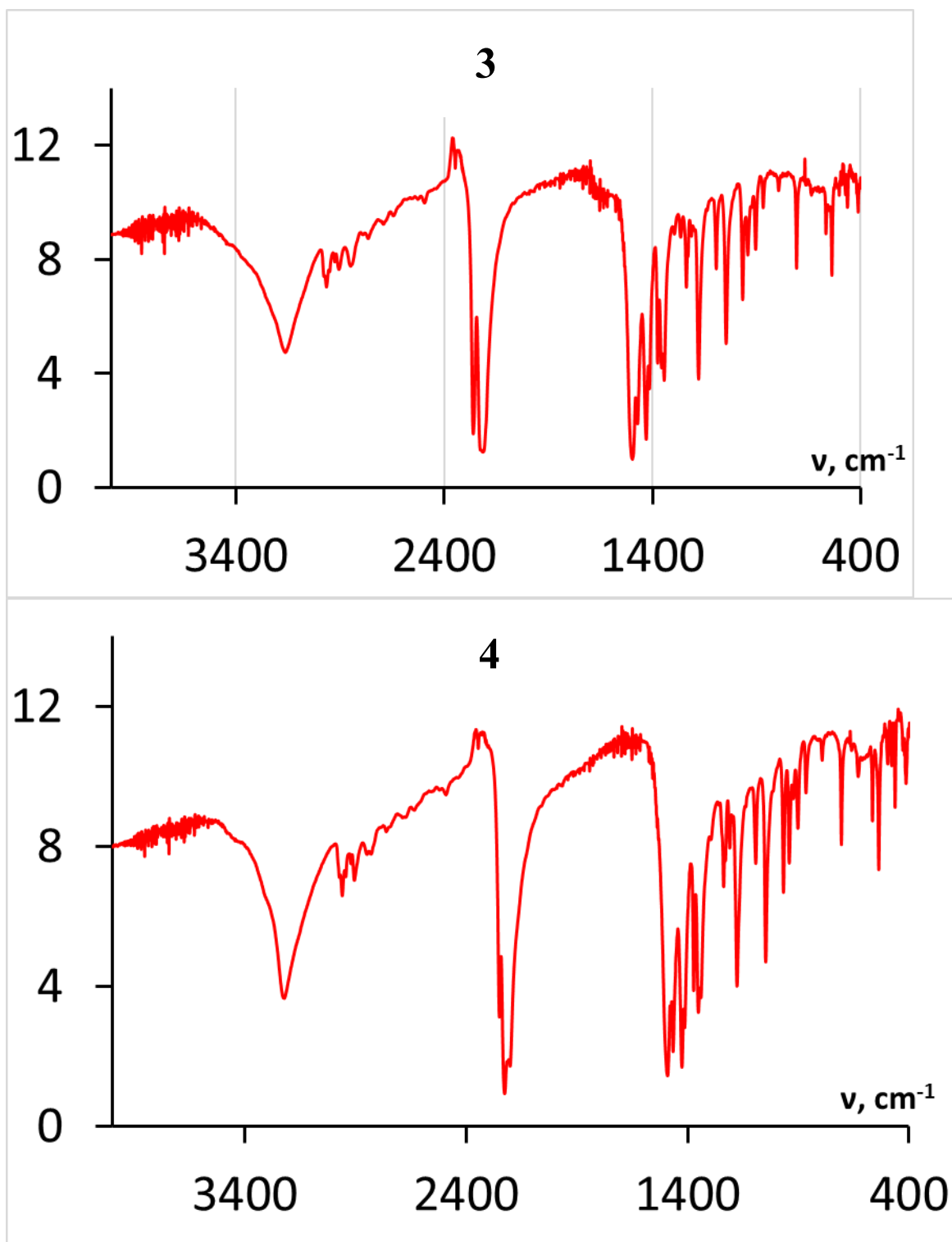


Figure S2. FT-IR spectra of studied compounds (top) **3** and (bottom) **4**.

TGA data

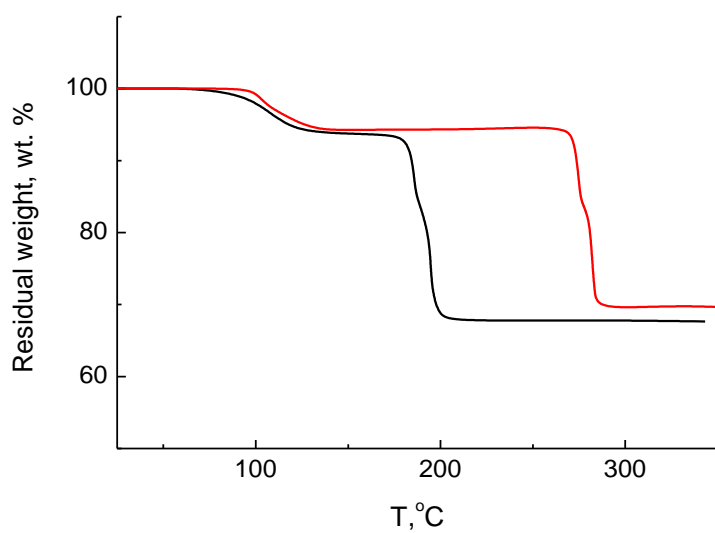


Figure S3. The thermal curves for (red) 1 and (black) 2.

Analysis of subnet/supernet relations.

Notation of coordination modes and networks.

There is strong dependence between coordination modes of ligands at one hand, and theoretically possible nets and their occurrences in crystalline materials at the other.^{S1-S4} Thus, there is need in notation of coordination modes that can be automatically processed to analyze and predict topology of underlying nets. We use notation described by some of us in Ref. ^{S5}. Within this notation the total denticity of a ligand is denoted as M for mono-, B for bi-, T for tri-, K for tetra-, P for penta-, G for hexa-, H for hepta-, O for octa-, N for nona- and D for decadenticity. The way in which A atoms surround the ligand is denoted by numerical superscripts ($mbtkpghond$). The superscripts define the 'partial' denticity of the ligand with respect to any A atom (m – mono-, b – bi-, t – tri-, ..., d – decadenticity). The number of A atoms with respect to the ligand that exhibits the corresponding partial denticity is denoted by the numerical value of the corresponding superscript. Then the coordination type of an i th ligand is given as $D_i^{mbtkpghond}$. This notation allows to calculate the total number of complexing atoms (Z) which surround the ligand, and the total number of chemical bonds that the ligand makes with the central atom (N_B). In the general case for a ligand with a $D_i^{mbtkpghond}$ coordination type

$$Z = m + b + t + k + p + g + h + o + n + d,$$

$$N_B = 1m + 2b + 3t + 4k + 5p + 6g + 7h + 8o + 9n + 10d.$$

Table S1 Potential contacts in the environment of Co1 in structure of compound 1

Atom	Symmetry code of edges	Distance, Å
N1	(-x-y-z)(-10-1); within unit cell	1.9902
N1	(-x-y-z)(00-1); (-100)	6.7929
N2	(-x-y-z)(-10-1); (00-1)	4.1737
N2	(-x-y-z)(-10-1); within unit cell	6.1161
N2	(-x-y-z)(-1-10); (0-1-1)	6.9115
N3	(-x-y-z); (-100)	4.6211
N3	(-x-y-z)(-100); (00-1)	4.4666
N4	(-x-y-z); (-10-1)	4.1849
N4	(-x-y-z)(0-10); (-11-1)	6.0434
O1	(-x-y-z)(00-1); (-100)	2.2903
O1	(-x-y-z)(-10-1); within unit cell	5.1537

Table S2 Potential contacts in the environment of Ni1 in structure of compound 2

Atom	Symmetry code of edges	Distance, Å
N1	(-x-y-z)(-100); (0-10)	2.1023
N1	(-x-y-z); (-1-10)	6.6732
N2	(-x-y-z)(0-10); (-100)	2.0907
N2	(-x-y-z)(-100); (0-10)	6.9348
N2	(-x-y-z); (-1-10)	6.8607
N3	(-x-y-z)(-100); (0-10)	3.9255
N3	(-x-y-z); (-1-10)	5.326
N3	(-x-y-z)(0-10); (-100)	6.1484
N4	(-x-y-z)(00-1); (-1-11)	5.1559
N4	(-x-y-z); (-1-10)	5.6816
N4	(-x-y-z)(-100); (0-10)	6.4896
O3	(-x-y-z)(-100); (0-10)	3.9473
O3	(-x-y-z)(-10-1); (0-11)	6.5739

Table S3. Supercells used for screening of the network transitions in structure of compound **1**, number of transitions (*N*), and their distribution over 3,6-c underlying nets with tridentate ligands.

Supercell	<i>N</i>	2D						3D					
		3,6C1	3,6L60	3,6L64	3,6L66	3,6L83	kgd	alb-3,6-C2/c	alb-3,6-P4 ₂ /mnm	ant	flu-3,6-C2/c	rtl	sit
(2a,b,c)	27225			208			311	156			698	342	
(2a,b,c; ½,0,0)	74613				8		1709			2268			
(-a,2b+2c,-c)	74613				184	360	1599			1150			
(-a,2b+2c,-c; 0,½,0)	27225			28			863	116	40		568	192	
(a,b,2b+2c)	27225			204			331	84	72		954	396	
(a,b,2b+2c; 0,0,½)	74613				4	8	1599			2374			
(a+c,-a-b+c,a-c)	27225			4			519	72			1000	324	
(a+c,-a-b+c,a-c; ½,0,0)	74613	216	448				663			1530			13
(-a-b,-b-c,2a)	74613	196	1272				637			1408			32
(-a-b,-b-c,2a; ½,0,0)	27225			24			499	44			928	214	
(2b,-a-b,c)	74613				8		1709			2268			
(2b,-a-b,c; ½,0,0)	27225			180			339	40			802	192	
(-a,-b-c,-2b)	74613				24	48	1233			2044			
(-a,-b-c,-2b; 0,½,0)	74613			184			391	112	44	894		250	

Unit cells that contain two independent metal atoms and two anions are marked with gray.

Table S4. Supercells used for screening of the network transitions in structure of compound **2**, number of transitions (*N*), and their distribution over 3,6-c underlying nets with tridentate ligands.

Supercell	<i>N</i>	kgd	ant	flu-3,6-C2/c	rtl
(-b,2a-2c,-a+2c)	230230	1009	1164		
(-b,2a-2c,-a+2c; 0,0,½)	81796	1009		1164	
(2a-b,2a-b-c,b)	81796	1009		1536	
(2a-b,2a-b-c,b; ½,0,0)	230230	1009	1536		
(-a,a-2b,-a-b+c)	230230	2005	1716		
(-a,a-2b,-a-b+c; 0,½,0)	81796	1009			996
(a+c,b-c,2c)	230230	1009	1692		
(a+c,b-c,2c; ½,0,0)	81796	1009		1692	
(a-2b,2b,c)	81796	1009		1032	
(a-2b,2b,c; 0,½,0)	230230	1009	1032		
(a-b,a-b-c,a+b)	81796	1009			864
(a-b,a-b-c,a+b; ½,0,0)	230230	1873	1848		
(-a-b+c,-b+2c,-a+b-c)	230230	1861	1860		
(-a-b+c,-b+2c,-a+b-c; ½,0,0)	81796	1009			852

Unit cells that contain two independent metal atoms and two anions are marked with gray.

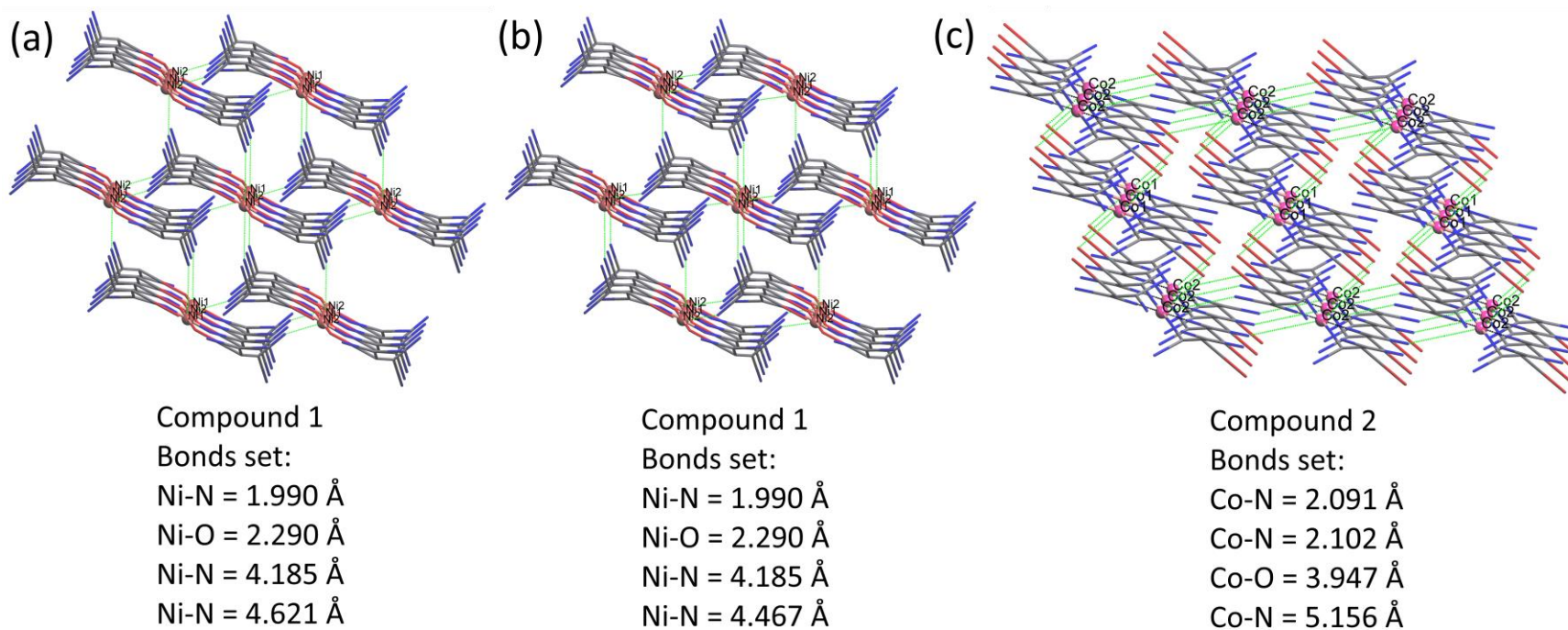


Figure S4. Closest neighborhood of metal atoms forming novel coordination bonds.

References:

- S1. T. G. Mitina and V. A. Blatov *Cryst Growth & Des.*, 2013, **13**, 1655.
- S2. E. V. Alexandrov, A. V. Virovets, V. A. Blatov and E. V. Peresypkina *Chem. Rev.* 2015, **115**, 12286.
- S3. E. V. Alexandrov, A. P. Shevchenko, A. A. Asiri and V. A. Blatov *CrystEngComm*, 2015, **17**, 2913.
- S4. E. V. Alexandrov, A. P. Shevchenko and V. A. Blatov *Cryst Growth & Des.*, 2019, **19**, 2604.
- S5. V. N. Serezhkin, A. V. Vologzhanina, L. B. Serezhkina, E. S. Smirnova, E. V. Grachova, P. V. Ostrova and M. Yu. Antipin, *Acta Crystallogr., Sect. B*, 2009, **65**, 45.

CIF files of **1** and **2** in doubled unit cells with removed water molecules and a corrected connectivity of atoms identical with that of, respectively, **3** and **4**.

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C8 C9 1.5076 1 0 0 0 1 0 0 1 V 2 17.83
C9 C10 1.5104 1 0 0 0 1 0 0 0 V 2 17.84
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 [T3] H2O[0D] H2O[0D]
 ATOMLIG: H1#1 H2#1 O5#1 H10#2 H11#2 O2#2 H3#3 H
 4#3 H5#3 H6#3 H7#3 H8#4 H9#4 H12#4 H13#4 H14#4 H15#4 H16#4 H17#3 H18#
 3 C1#4 C2#4 C3#4 C4#4 C5#4 C6#3 C7#4 C8#4 C9#3 C10#3 C11#3 C12#3 C13#
 3 C14#3 C15#3 C16#4 C17#3 C18#3 C19#4 C20#4 N1#4 N2#4 N3#4 N4#4 N5#3
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 SIMPPAR: METH=Stand; INTRA=V
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H4	H 2 0.85315 0.25975 0.87355 1.0000
H5	H 2 0.92330 0.34030 0.77220 1.0000
H6	H 2 0.04240 0.31480 0.92040 1.0000
H7	H 2 0.06790 0.21750 0.80280 1.0000
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H9	H 2 0.64765 0.05475 0.35965 1.0000
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H17	H 2 0.24030 0.42700 0.81100 1.0000
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C2	C 2 0.34757 0.24314 0.41129 1.0000
C3	C 2 0.12103 0.04606 0.34076 1.0000
C4	C 2 0.25488 0.34131 0.43560 1.0000

C5 C 2 0.31267 0.07491 0.40075 1.0000
 C6 C 2 0.92132 0.34415 0.84509 1.0000
 C7 C 2 0.13944 0.83877 0.41909 1.0000
 C8 C 2 0.47725 0.31763 0.39965 1.0000
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H14 N6 2.9377 1 0 0 0 2 1 1 1 W 2 7.10
H14 C13 3.3030 1 0 0 0 2 1 1 1 W 2 3.56
H15 C19 0.9700 1 0 0 0 1 0 0 0 V 2 27.92
H15 O1 2.6120 1 0 0 0 2 1 1 1 W 2 10.95
H15 C8 2.8672 1 0 0 0 2 1 1 1 W 2 7.66
H15 C2 3.1151 1 0 0 0 2 1 1 1 W 2 3.86
H16 C19 0.9699 1 0 0 0 1 0 0 0 V 2 27.95
H16 H17 2.5841 1 0 0 0 2 1 1 1 W 2 14.85
H16 N7 2.9558 1 0 0 0 2 1 1 1 W 2 11.19

H16 N4 3.4936 1 0 0 0 1 0 0 0 W 2 4.17
H17 C10 0.9703 1 0 0 0 1 0 0 0 V 2 27.86
H17 N7 3.4872 1 0 0 0 1 0 0 0 W 2 5.63
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H17 C17 4.1738 1 0 0 0 2 1 1 2 W 2 1.95
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H18 O4 2.5848 1 0 0 0 1 -1 0 0 W 2 9.15
H18 N8 2.9620 1 0 0 0 1 0 0 0 W 2 9.91
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C14 N5 1.1517 1 0 0 0 1 0 0 0 V 2 34.74
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C14 N8 3.5495 1 0 0 0 2 1 2 2 W 2 3.25
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C16 C19 1.5112 1 0 0 0 1 0 0 0 V 2 17.86
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C18 N8 1.1545 1 0 0 0 1 1 0 0 V 2 33.63
C18 O4 2.6895 1 0 0 0 1 0 0 0 W 2 5.21
C18 N3 3.2144 1 0 0 0 2 1 1 1 W 2 4.71
C18 N8 3.3664 1 0 0 0 2 1 2 2 W 2 5.84
C19 C20 1.5143 1 0 0 0 1 0 0 0 V 2 17.84
C20 O6 1.4279 1 0 0 0 1 0 0 0 V 2 20.05
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N1 N2 2.8943 1 0 0 0 2 0 1 1 W 2 7.04
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N1 O3 3.2496 1 0 0 0 1 0 0 0 W 2 1.97
N1 N3 3.3295 1 0 0 0 1 0 0 0 W 2 2.21
N1 Co2 6.6732 1 0 0 0 1 0 0 0 N 2 0.00
N2 Co1 2.0907 1 0 0 0 1 0 0 0 V 2 16.26
N2 O5 2.8956 1 0 0 0 1 -1 0 0 W 2 4.91
N2 N3 3.4271 1 0 0 0 2 0 1 1 W 2 1.92
N2 Co1 6.9348 1 0 0 0 1 0 1 0 N 2 0.00
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N4 Co1 6.4896 1 0 0 0 1 0 0 0 N 2 0.00
N5 N6 2.8943 1 0 0 0 2 1 1 2 W 2 7.04
N5 O2 2.9180 1 0 0 0 1 0 0 0 W 2 4.55
N5 O6 3.2496 1 0 0 0 1 0 0 1 W 2 1.97
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N5 Co2 2.1023 1 0 0 0 1 0 1 1 H 2 15.98
N5 Co1 6.6732 1 0 0 0 1 1 0 0 N 2 0.00
N6 Co2 2.0907 1 0 0 0 1 0 0 1 V 2 16.26
N6 O2 2.8956 1 0 0 0 1 0 -1 0 W 2 4.91
N6 N7 3.4271 1 0 0 0 2 1 1 2 W 2 1.92
N6 Co2 6.9348 1 0 0 0 1 0 1 1 N 2 0.00
N6 Co1 6.8607 1 0 0 0 1 1 0 0 N 2 0.00
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