Electronic Supplementary Material (ESI) for Dalton Transactions. This journal is © The Royal Society of Chemistry 2020

## **Supporting Information**

## Ultrabroadband Red Luminescence of Mn<sup>4+</sup> in MgAl<sub>2</sub>O<sub>4</sub> Peaking at 651 nm

Haipeng Ji,<sup>1</sup> Xinghui Hou,<sup>1</sup> Maxim S. Molokeev,<sup>2,3</sup> Jumpei Ueda,<sup>4</sup> Setsuhisa Tanabe,<sup>4</sup> Mikhail G. Brik,<sup>5,6</sup> Zongtao Zhang,<sup>1</sup> Yu Wang,<sup>1,\*</sup> Deliang Chen<sup>1,7,\*</sup>

- 1 School of Materials Science and Engineering, Zhengzhou University, Zhengzhou 450001, China
- 2 Laboratory of Crystal Physics, Kirensky Institute of Physics, Federal Research Center KSC SB RAS, Krasnoyarsk 660036, Russia
- 3 Siberian Federal University, Krasnoyarsk 660041, Russia
- 4 Graduate School of Human and Environmental Studies, Kyoto University, Kyoto 606-8501, Japan
- 5 CQUPT-BUL Innovation Institute & College of Sciences, Chongqing University of Posts and Telecommunications, Chongqing 400065, China
- 6 Institute of Physics, University of Tartu, Tartu 50411, Estonia
- 7 School of Materials Science and Engineering, Dongguan University of Technology, Dongguan 523808, China

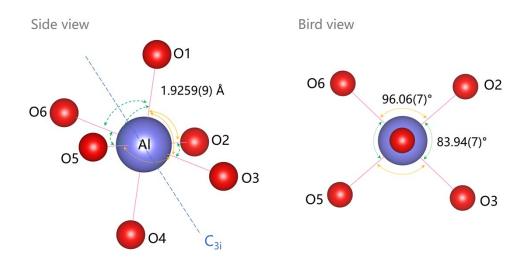


Figure S1 Al<sup>3+</sup> octahedral coordination in MgAl<sub>2</sub>O<sub>4</sub> (ICSD 31373) showing the interatomic distances and angles between the chemical bonds.  $d_{Al-O1} = d_{Al-O2} = d_{Al-O3} = d_{Al-O4} = d_{Al-O5} = d_{Al-O6} = 1.9259(9)$  Å;  $\angle O1$ -Al-O2 =  $\angle O1$ -Al-O3 =  $\angle O2$ -Al-O6 =  $\angle O3$ -Al-O5 =  $\angle O2$ -Al-O4 =  $\angle O3$ -Al-O4 =  $\angle O3$ -Al-O4 =  $\angle O3$ -Al-O5 =  $\angle O4$ -Al-O5 =  $\angle O4$ -Al-O6 = 96.06(7)°,  $\angle O1$ -Al-O5 =  $\angle O1$ -Al-O6 =  $\angle O2$ -Al-O3 =  $\angle O5$ -Al-O6 = 83.94(7)°.

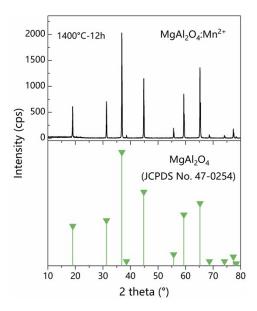
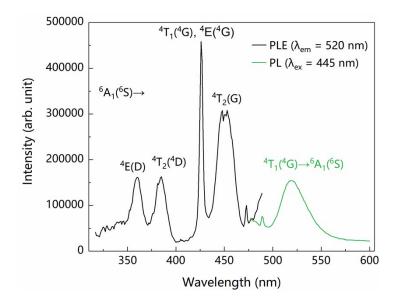


Figure S2 XRD pattern of the spinel phosphor calcined at 1400 °C for 12 h. The standard pattern for MgAl<sub>2</sub>O<sub>4</sub> (JCPDS card no. 47-0254) was provided as reference.



**Figure S3** Photoluminescence spectra of the MgAl<sub>2</sub>O<sub>4</sub>:Mn<sup>2+</sup> phosphor calcined at 1400 °C for 12 h. The corresponding energy transitions were assigned based on Ref. [1].

E. H. Song, Y. Y. Zhou, Y. Wei, X. X. Han, Z. R. Tao, R. L. Qiu, Z. G. Xia and Q. Y. Zhang. A thermally stable narrow-band green-emitting phosphor MgAl<sub>2</sub>O<sub>4</sub>:Mn<sup>2+</sup> for wide color gamut backlight display application. *J. Mater. Chem. C*, 2019, 7, 8192-8198.

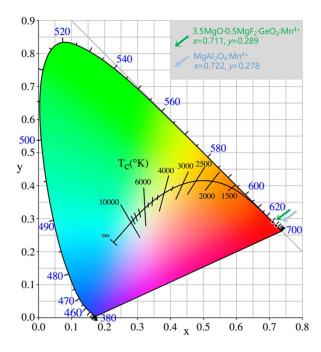


Figure S4 Chromaticity coordinate ( $\lambda_{ex}$  = 440 nm) of MgAl<sub>2</sub>O<sub>4</sub>:Mn<sup>4+</sup> in CIE 1931 diagram. The chromaticity of the commercial 3.5MgO·0.5MgF<sub>2</sub>·GeO<sub>2</sub>:Mn<sup>4+</sup> was also given.

## Preparation of MgAl<sub>2</sub>O<sub>4</sub>:Mn<sup>4+</sup> by the molten salt method:

MgO (A.R., Sinopharm, China) and Al<sub>2</sub>O<sub>3</sub> (99.99%,  $\gamma$ -phase,  $\leq$ 20 nm, Aladdin, China) were used as the raw chemicals. MnCO<sub>3</sub> (A.R., Sinopharm, China) and LiCl·H<sub>2</sub>O (A.R., Sinopharm, China) were used as the dopant source and the salt. MgO, Al<sub>2</sub>O<sub>3</sub>, and MnCO<sub>3</sub> were weighed according to the formula of MgAl<sub>1.998</sub>Mn<sub>0.002</sub>O<sub>4</sub> and then mixed with LiCl·H<sub>2</sub>O in a weight ratio of 1:5. The mixture was then transferred to corundum crucibles covered with a lid and heated at 950 °C for 3 h. After cooling to room temperature, the product was washed with hot deionized water for 5 times to remove the salt, concentrated by centrifugation, and then dried in an oven at 100 °C for 12 h.