

## Electronic Supplementary Information

### Microwave-Assisted Solid-State Synthesis of NaRE(MO<sub>4</sub>)<sub>2</sub> Phosphors (RE = La, Pr, Eu, Dy; M = Mo, W)

S. Sameera Perera,<sup>†</sup> Hashini N. Munasinghe,<sup>†</sup> Emily N. Yatooma, and Federico A. Rabuffetti\*

Department of Chemistry, Wayne State University, Detroit, MI 48202, USA

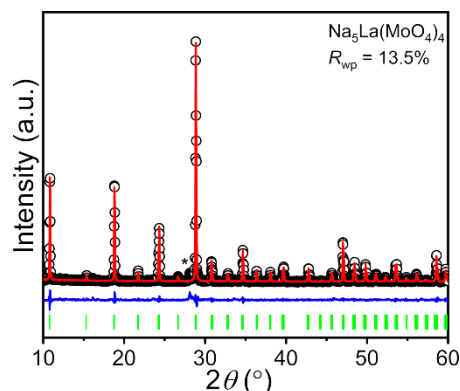
<sup>†</sup>These authors contributed equally

\*Corresponding Author. Email: [far@chem.wayne.edu](mailto:far@chem.wayne.edu)

**Table S1.** Structural Parameters of Na<sub>2</sub>MO<sub>4</sub> and NaRE(MO<sub>4</sub>)<sub>2</sub>

M	Na <sub>2</sub> MO <sub>4</sub>		NaEu(MO <sub>4</sub> ) <sub>2</sub>		NaLa <sub>0.95</sub> Eu <sub>0.05</sub> (MO <sub>4</sub> ) <sub>2</sub>		NaLa <sub>0.95</sub> Pr <sub>0.025</sub> Dy <sub>0.025</sub> (MO <sub>4</sub> ) <sub>2</sub>	
	Mo	W	Mo	W	Mo	W	Mo	W
<i>a</i> (Å)	9.10985(10)	9.12938(12)	5.24577(8)	5.25722(8)	5.33755(6)	5.35498(5)	5.33988(3)	5.34993(5)
<i>c</i> (Å)	–	–	11.4682(2)	11.4079(2)	11.71844(19)	11.65403(13)	11.72515(8)	11.64530(17)
<i>V</i> (Å <sup>3</sup> )	756.02(3)	760.90(3)	315.582(12)	315.296(12)	333.852(10)	334.189(8)	334.334(5)	333.309(9)
<i>x</i> O	0.26172(14)	0.2582(4)	0.1426(13)	0.1470(15)	0.1389(11)	0.1441(13)	0.1334(6)	0.1432(18)
<i>y</i> O	0.26172(14)	0.2582(4)	0.4663(12)	0.4721(12)	0.4744(9)	0.4790(11)	0.4847(6)	0.4763(14)
<i>z</i> O	0.26172(14)	0.2582(4)	0.2065(5)	0.2087(6)	0.2073(4)	0.2069(5)	0.2051(3)	0.2029(7)
<i>U</i> <sub>iso</sub> <sup>Na/RE</sup>	0.46(4)	0.10	1.34(6)	2.45(9)	1.84(12)	0.73(8)	0.30	1.07(5)
<i>U</i> <sub>iso</sub> <sup>M</sup>	0.45(4)	0.10	1.34(6)	1.20(5)	0.48(8)	1.61(5)	0.30	1.07(5)
<i>U</i> <sub>iso</sub> <sup>O</sup>	1.50(10)	0.15	0.6(3)	1.0(4)	3.0(3)	1.8(3)	0.45	2.9(4)
<i>R</i> <sub>wp</sub> (%)	5.5	10.0	5.8	6.4	8.0	8.4	7.0	9.6

*U*<sub>iso</sub> values given as 100 × *U*.



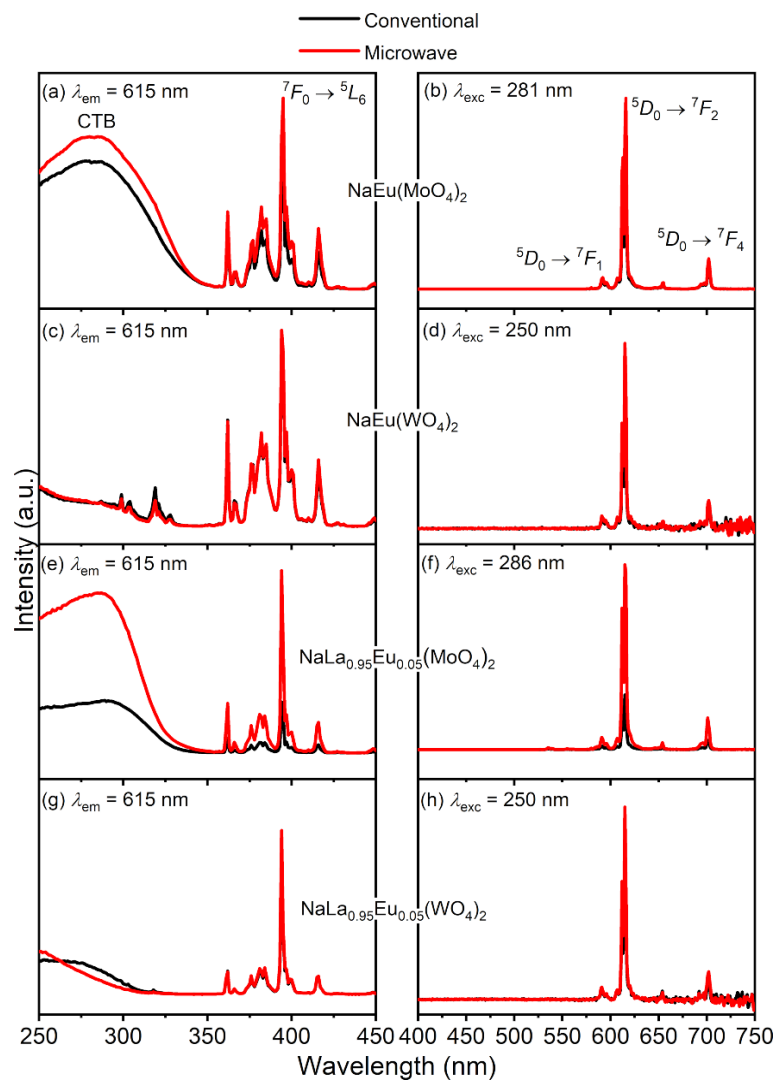
**Figure S1.** Rietveld analysis of the PXRD pattern of the quadruple molybdate  $\text{Na}_5\text{La}(\text{MoO}_4)_4$  (PDF No. 01–072–2158) obtained via microwave-assisted solid-state reaction (sample mass: 0.20 g; power: 600 W; heating profile: 9 cycles of 3 min each; total time: 27 min). Experimental data ( $\circ$ ), calculated pattern ( $—$ ), and difference curve ( $—$ , offset for clarity) are shown. Tick marks ( $|$ ) corresponding to the calculated position of the diffraction maxima are included. The peak depicted with a \* symbol corresponds to the double molybdate  $\text{NaLa}(\text{MoO}_4)_2$ , which appears as a secondary phase.

**Table S2.** Structural Parameters of  $\text{Na}_5\text{La}(\text{MoO}_4)_4$

	$x$	$y$	$z$	$U_{\text{iso}}$
Na1	0.0	0.25	0.625	0.7(3)
La1	0.0	0.25	0.125	0.7(3)
Na2	0.1270(18)	0.042(2)	0.3392(14)	0.7(3)
Mo1	0.1784(5)	0.3474(9)	0.3923(4)	1.8(3)
O1	0.168(3)	0.459(3)	0.484(3)	1.1(5)
O2	0.149(3)	0.221(3)	0.475(3)	1.1(5)
O3	0.058(3)	0.367(3)	0.295(3)	1.1(5)
O4	0.317(3)	0.332(3)	0.318(2)	1.1(5)

$\text{Na}_5\text{La}(\text{MoO}_4)_4$ : space group  $I4_1/a$ ,  $a = 11.5750(2)$  Å,  $c = 11.6210(3)$  Å,  $V = 1556.99(8)$  Å<sup>3</sup>.

$U_{\text{iso}}$  values given as  $100 \times U$ .



**Figure S2.** Excitation and emission spectra of  $\text{Eu}^{3+}$ -containing metalate phosphors synthesized using conventional and microwave heating. Metal-to-oxygen charge-transfer band (CTB) and selected f–f transitions of  $\text{Eu}^{3+}$  are indicated in (a) and (b).