Supporting information

Photochemical deposition of amorphous MoS_x on one-dimensional $NaNbO_3$-CdS heterojunction photocatalyst for highly efficient visible-light-driven hydrogen evolution

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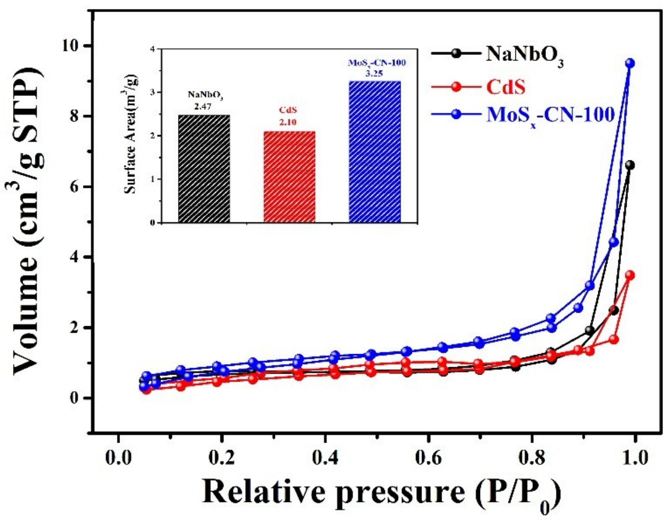

Fig. S1 Nitrogen adsorption–desorption isotherms of $NaNbO_3$, CdS and MoS_x-CN-100. Inset: The BET surface area of $NaNbO_3$, CdS and MoS_x-CN-100.

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Table S1. A comparison of the hydrogen evolution performance between our MoS_x-CN and CdS, $NaNbO_3$ based photocatalysts

No.	photocatalysts	Light source (Wavelength)	Sacrificial solution	H_2 mmol/h/g	Ref.
1	MoS_x-$NaNbO_3$-CdS	Xe lamp 300W(>400 nm)	Lactic acid	2.386	This work
2	CdS/Ni-MOF	Xe lamp 300W (>420nm)	Lactic acid	2.508	1
3	CdS/Zn_2GeO_4	Xe lamp 300W (>420nm)	Na_2SO_3&Na_2S	1.72	2
4	CdS/MoS_2	Xe lamp 300W (>420nm)	Lactic acid	1.4	3
5	g-C_3N_4/Au/CdS	Xe lamp 300W (>420nm)	Lactic acid	1.1	4
6	Pt-rGO-$NaNbO_3$	Xe lamp 300W (>420nm)	Methanol	2.342	5
7	Pt-$NaNbO_3$	Xe lamp 300W	Methanol	0.266	6
8	Ca-$NaNbO_3$-C_3N_4	Solar irradiation (~900W/m^2)	Methanol	2.95	7
9	$NaNbO_3$-Pt	Xe lamp 400W (>300nm)	Lactic acid	1.12	8

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