

Electronic supplementary information for

**Addition of CF<sub>2</sub> group to endohedral fullerene Sc<sub>3</sub>N@I<sub>h</sub>-C<sub>80</sub>**

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**Table of contents**

**Figure S1.** Negative ion and positive ion MALDI mass spectra of isolated Sc<sub>3</sub>N@C<sub>80</sub>(CF<sub>2</sub>)...S2

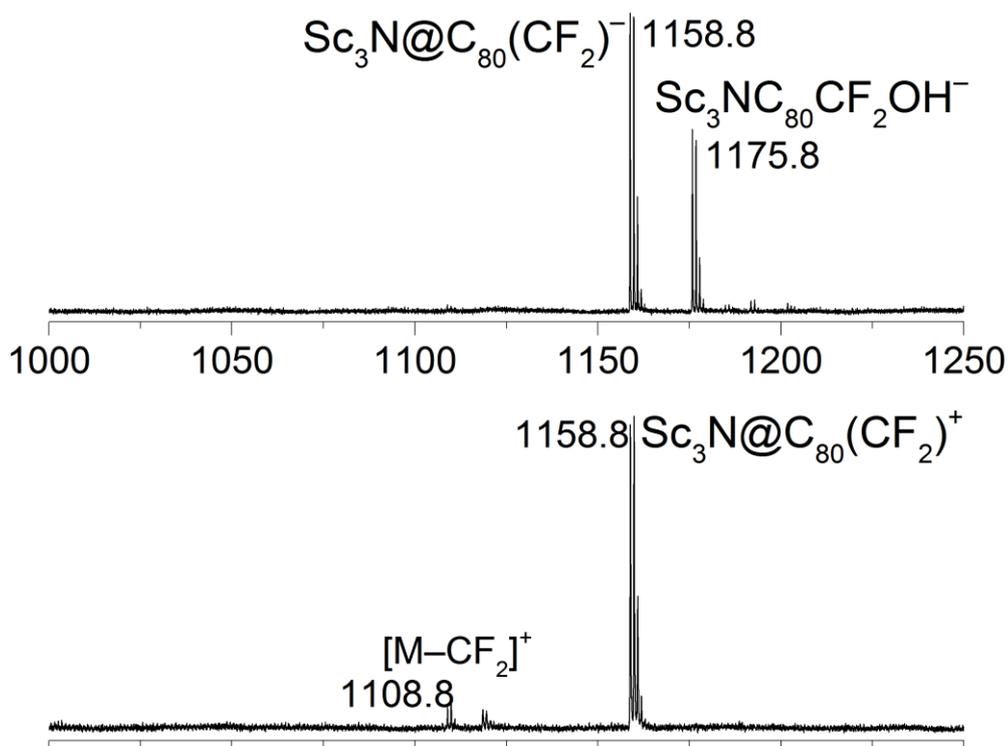
**Figure S2.** The temperature-dependent full width at half-maximum (FWHM) of a peak at ca. 190 ppm of <sup>45</sup>Sc NMR spectra of Sc<sub>3</sub>N@C<sub>80</sub>(CF<sub>2</sub>).....S2

**Figure S3.** CV and deconvoluted CV curves of Sc<sub>3</sub>N@C<sub>80</sub>(CF<sub>2</sub>) and Sc<sub>3</sub>N@I<sub>h</sub>-C<sub>80</sub>. .....S3

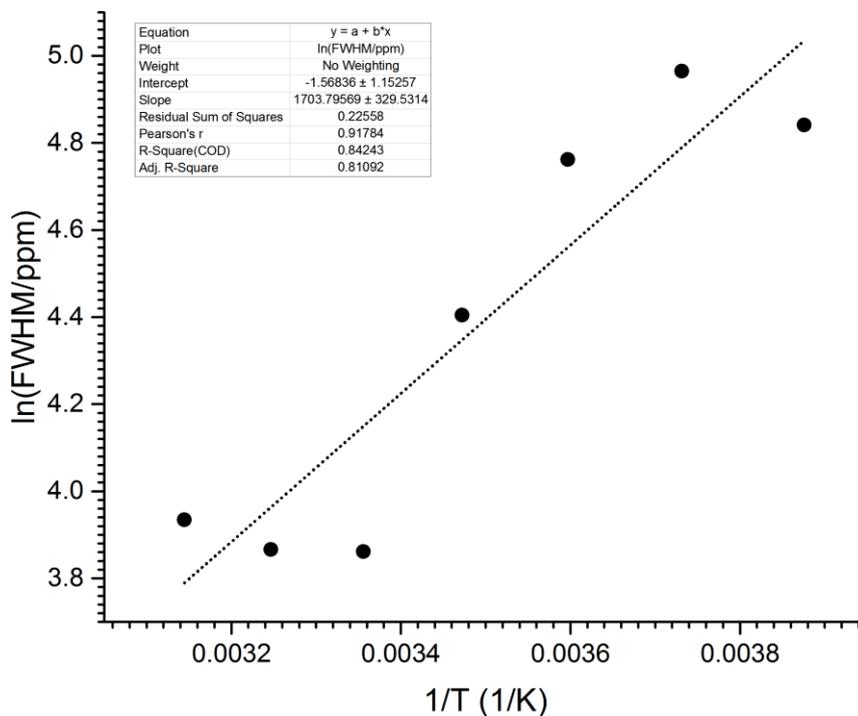
**Figure S4.** Experimental and calculated <sup>19</sup>F NMR chemical shifts correlation for CF<sub>2</sub>-derivitized fullerenes .....S4

**Figure S5.** Experimental and calculated <sup>13</sup>C NMR chemical shifts of bridgehead and CF<sub>2</sub> carbon atoms for CF<sub>2</sub>-derivitized fullerenes.....S4

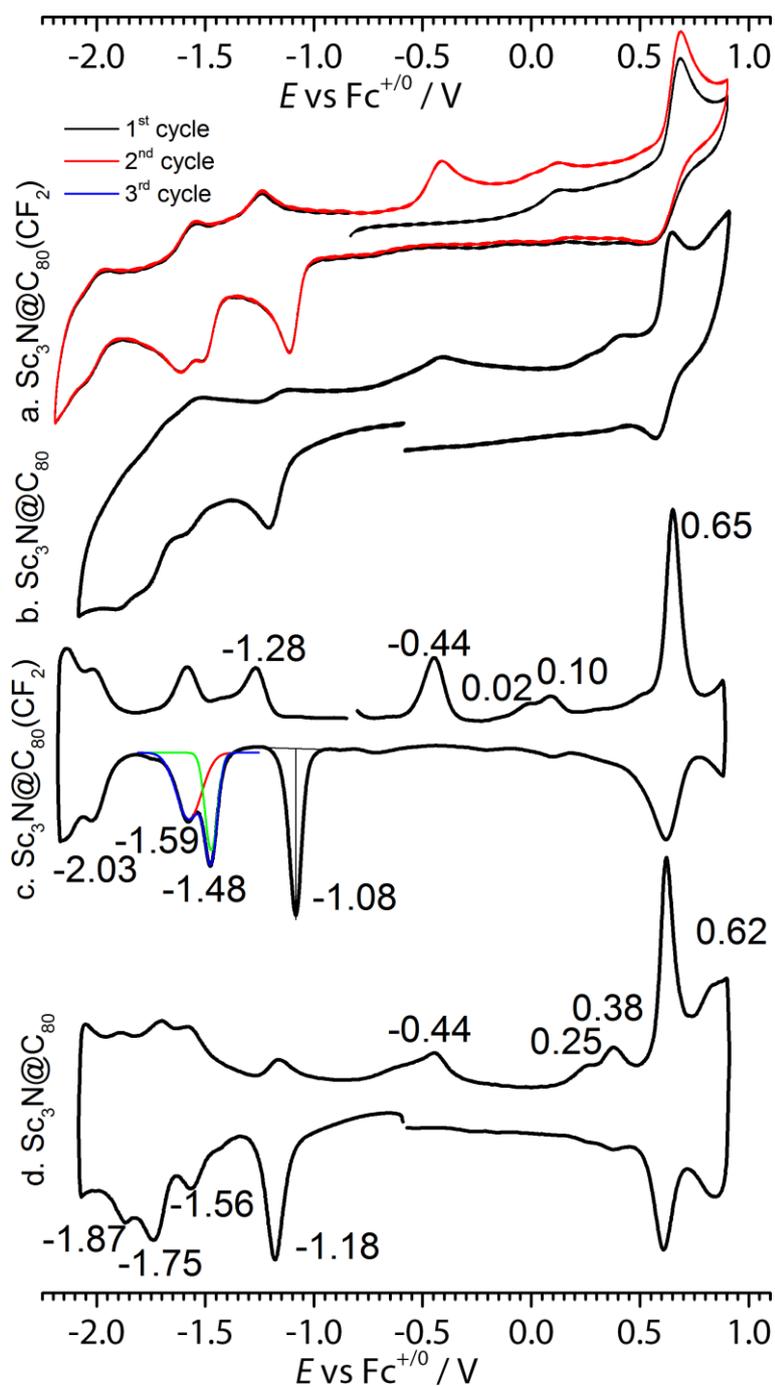
**Figure S6.** The conformers of Sc<sub>3</sub>N@C<sub>80</sub>(CF<sub>2</sub>), THJ and PHHJ Sc<sub>3</sub>N@C<sub>80</sub>(CF<sub>2</sub>Cl)<sup>-</sup> intermediates .....S5



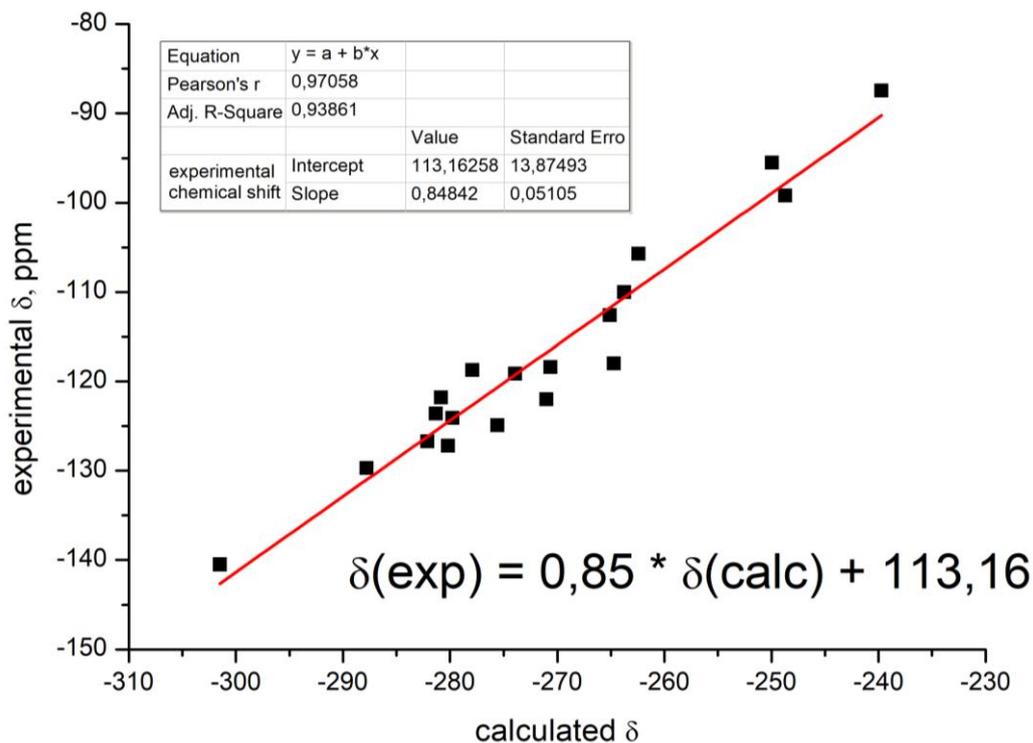
**Figure S1.** Negative ion (top) and positive ion (bottom) MALDI mass spectra of isolated Sc<sub>3</sub>N@C<sub>80</sub>(CF<sub>2</sub>).



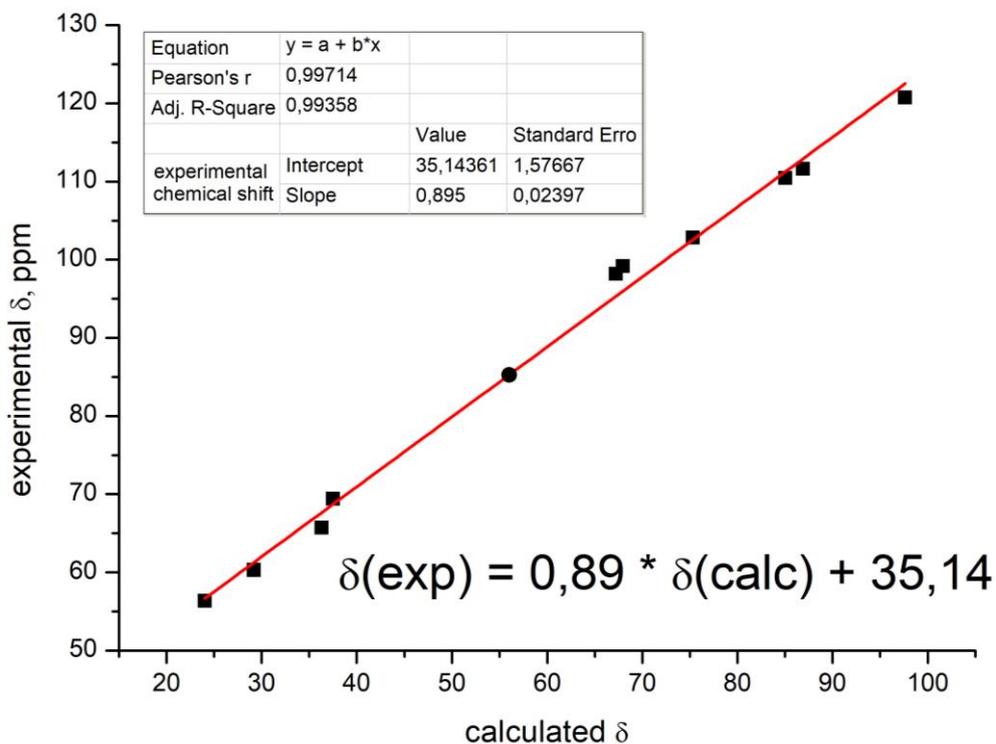
**Figure S2.** The temperature-dependent full width at half-maximum (FWHM) of a peak at ca. 190 ppm of <sup>45</sup>Sc NMR spectra of Sc<sub>3</sub>N@C<sub>80</sub>(CF<sub>2</sub>)



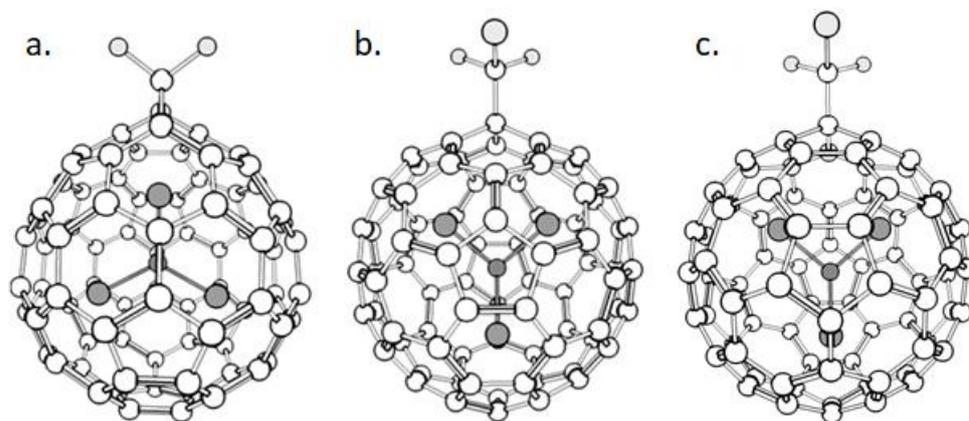
**Figure S3.** CV curves of (a)  $\text{Sc}_3\text{N}@C_{80}(\text{CF}_2)$  and (b)  $\text{Sc}_3\text{N}@I_h\text{-C}_{80}$  (Pt, oDCB, 0.15 M  $\text{Bu}_4\text{NBF}_4$ , vs  $\text{Fc}^{+/0}$ ,  $100 \text{ mV s}^{-1}$ ,  $30 \text{ }^\circ\text{C}$ ). The deconvoluted CV curves are shown for (c, 2<sup>nd</sup> scan)  $\text{Sc}_3\text{N}@C_{80}(\text{CF}_2)$  and (d)  $\text{Sc}_3\text{N}@I_h\text{-C}_{80}$  (the redox potentials are shown for deconvoluted curves).



**Figure S4.** Experimental and calculated  $^{19}\text{F}$  NMR chemical shifts correlation for  $\text{CF}_2$ -derivitized fullerenes:  $\text{C}_{60}(\text{CF}_2)$ , *cis*-2- $\text{C}_{60}(\text{CF}_2)_2$ ,  $\text{C}_{60}(\text{CF}_2)\text{H}_2$ ,  $\text{C}_{70}(\text{CF}_2)$ ,  $\text{C}_s\text{-C}_{70}(\text{CF}_3)_8(\text{CF}_2)$ ,  $\text{C}_s\text{-C}_{70}(\text{CF}_3)_8(\text{CF}_2)(\text{OH})_2$ ,  $\text{C}_{70}(\text{CF}_3)_8(\text{CF}_2)\text{H}(\text{OH})$ .



**Figure S5.** Experimental and calculated  $^{13}\text{C}$  NMR chemical shifts of bridgehead and  $\text{CF}_2$  carbon atoms for  $\text{CF}_2$ -derivitized fullerenes:  $\text{C}_{60}(\text{CF}_2)$ ,  $\text{C}_{60}(\text{CF}_2)\text{H}_2$ ,  $\text{C}_{70}(\text{CF}_2)$ ,  $\text{C}_s\text{-C}_{70}(\text{CF}_3)_8(\text{CF}_2)$ .



**Figure S6.** The conformers of (a) Sc<sub>3</sub>N@C<sub>80</sub>(CF<sub>2</sub>) as well as (b) THJ and (c) PHHJ Sc<sub>3</sub>N@C<sub>80</sub>(CF<sub>2</sub>Cl)<sup>-</sup> intermediates, respectively.