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Supplementary material of :

Structure, Polymorphism and Luminescence of Cyanate Iodides MI(OCN) (M = Ba, Eu, Sr)

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Figure S1. Differential thermal analysis (DTA) of the reaction Srl₂ and KOCN (1:1) in silica ampoule.



Figure S2. Differential thermal analysis (DTA) of the reaction Bal_2 and KOCN (1:1) in silica ampoule. Heating and cooling cycles are represented in red and blue respectively.



Figure S4. Crystal structure refinement plot of EuI(OCN) based on X-ray powder pattern data. Green ticks mark Bragg reflections relative to EuI(OCN) and KI phase. The difference curve (blue) highlights the difference between observed (red) and calculated pattern (black).

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Figure S5. Crystal structure refinement plot of Bal(OCN) based on X-ray powder pattern data. Green ticks mark Bragg reflections relative to Bal(OCN) and KI phase. The difference curve (blue) highlights the difference between observed (red) and calculated pattern (black).



Figure S6. Powder X-ray patterns of the products obtained at different temperatures of solid-state reaction of Srl_2 with KOCN (1:1).



Figure S7. Powder X-ray patterns of the products obtained at different temperatures of solid-state reaction of Eul_2 with KOCN (1:1).



Figure S8. Powder X-ray patterns of the products obtained at different temperatures of solid-state reaction of Bal₂ with KOCN (1:1).



Figure S9. FTIR spectra of α -Bal(OCN) (SG: *Pnma*) and β -Bal(OCN) (SG: *Cmcm*) compounds.



Figure S10. Fourier transform infrared spectrum of MI(OCN) compounds.



Figure S11. Excitation (red) and emission spectra (black) of Eu^{2+} (2%)-doped β -Bal(OCN) compound.



Figure S12. Decay curve of β -Bal(OCN): Eu²⁺recorded for the europium blue transition at 431 nm.