

Supporting Information

Bistable spin-crossover in a new series of $[\text{Fe}(\text{BPP-R})_2]^{2+}$ (BPP = 2,6-bis(pyrazol-1-yl)pyridine; R = CN) complexes

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S1. Representation of angular components of **1c**

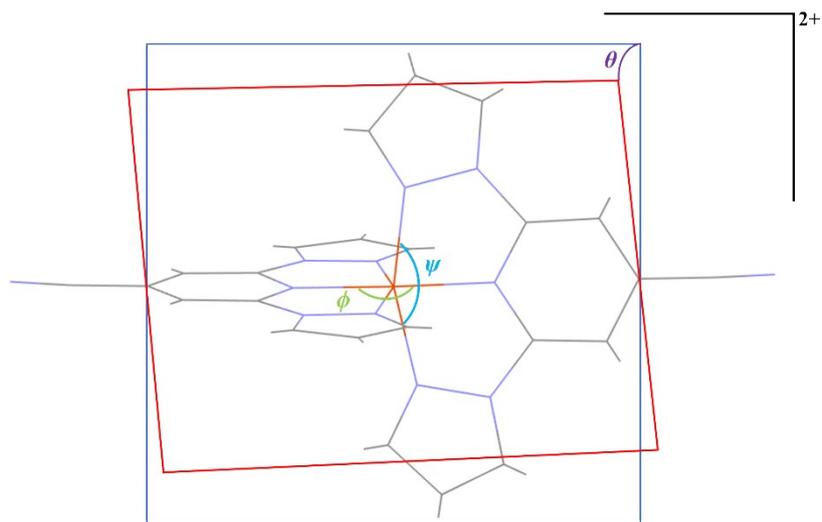


Chart S1. Structural model depicting angular components in $[\text{Fe}(\text{1-BPP-CN})_2](\text{BF}_4)_2 \cdot \text{CH}_3\text{CN}$ (**1c**). The angles ϕ and θ represent the degree of distortion; complexes with ideal octahedral geometry have $\phi = 180^\circ$ and $\theta = 90^\circ$. The N{pyrazole}-Fe-N{pyrazole} clamp angle is represented as ψ .

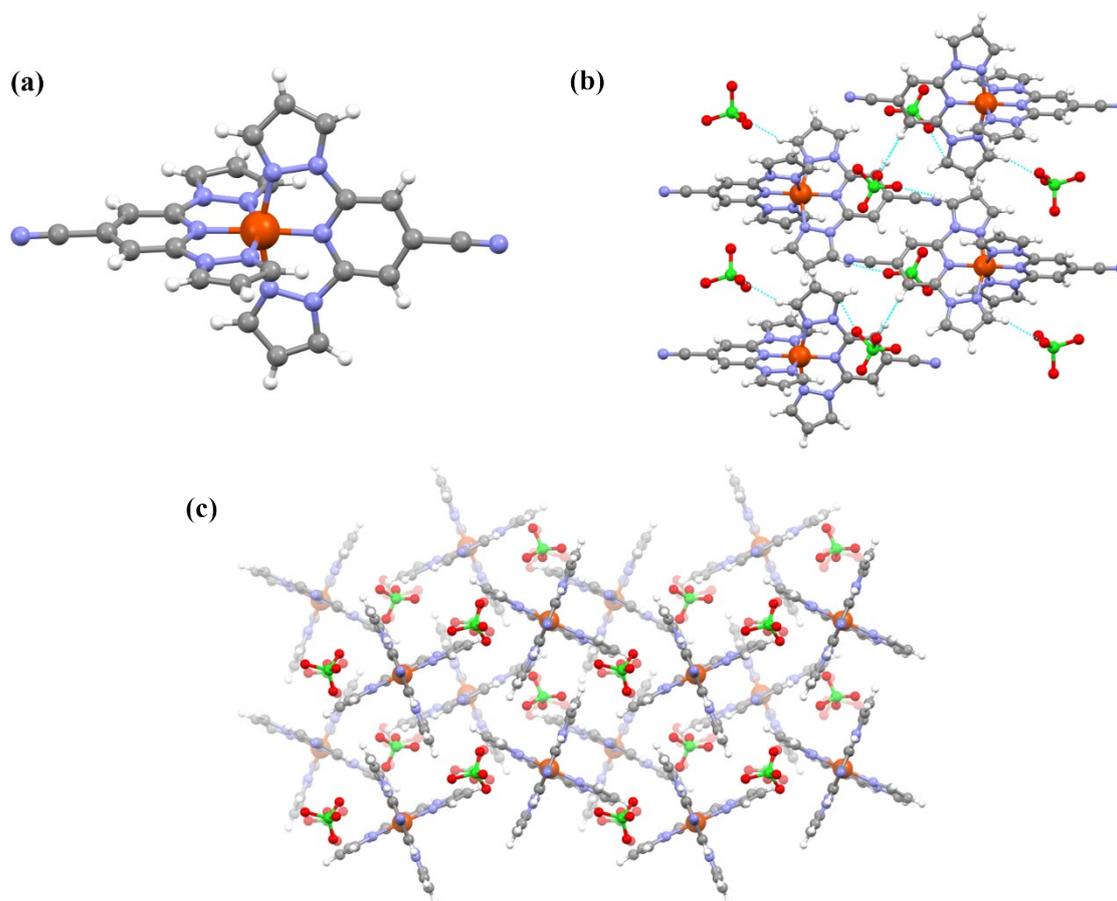


Figure S1. (a) X-ray structure of complex **2-cryst** (CCDC 2010806). Counter anions are omitted for clarity, (b) Intermolecular short contacts observed in the crystal lattice of **2-cryst**, and (c) Non-terpyridine embrace lattice packing pattern observed for complex **2-cryst** in the crystal lattice.

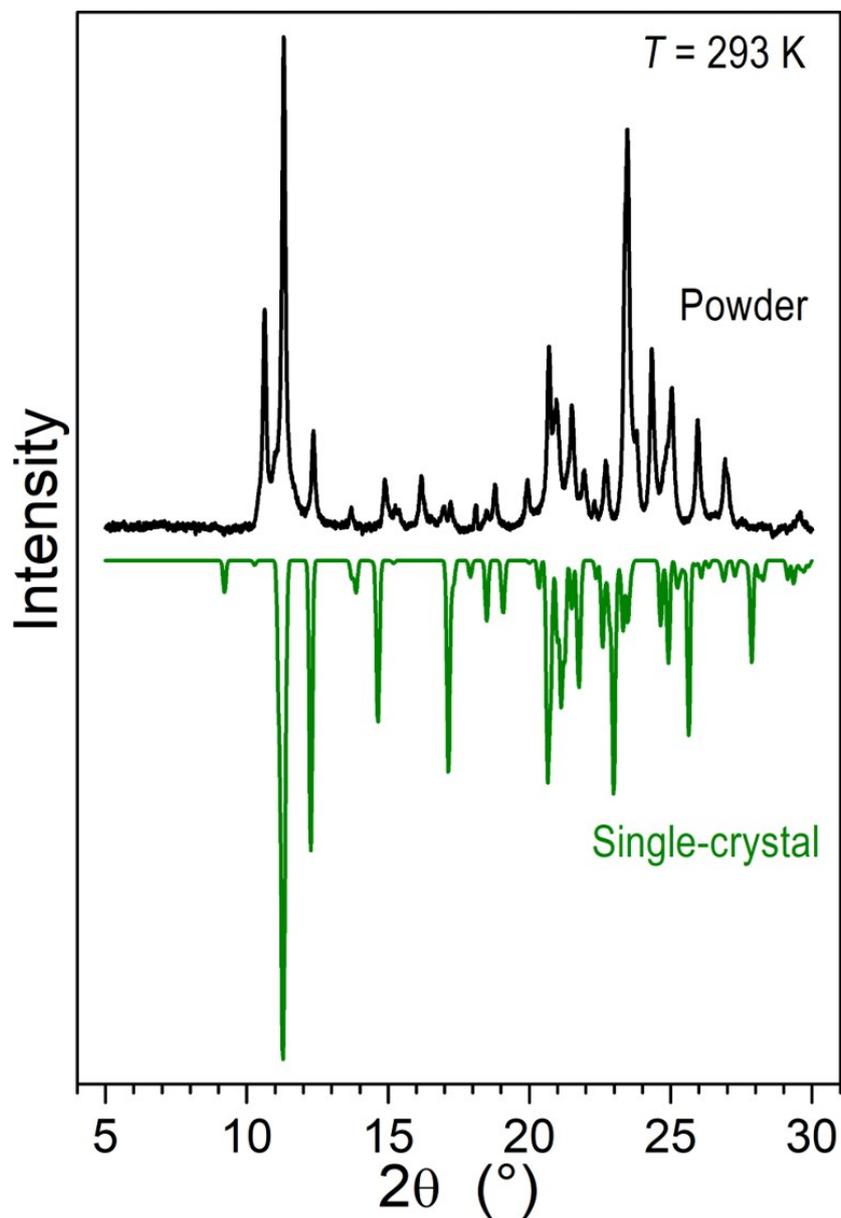


Figure S2. Comparison between the powder XRD pattern of **2** and SCXRD pattern of **2-cryst**. The remarkable difference between the patterns indicate different organizations of the molecules in the crystal lattice of **2** and **2-cryst**. Such different molecular organizations caused the HS and LS states of **2** and **2-cryst**, respectively, at 293 K.

S2. UV-vis spectra of the ligand and complexes **1a** and **2**

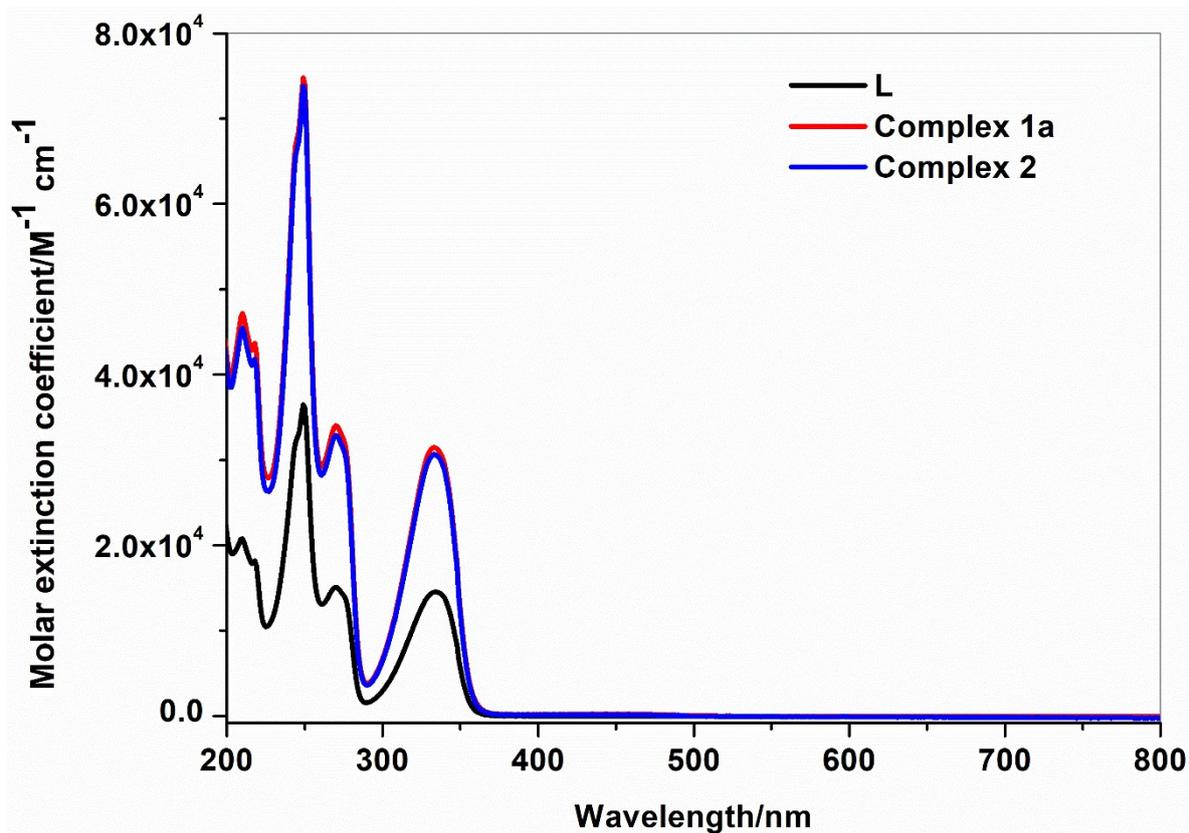


Figure S3. UV-vis absorption spectra of the ligand and complexes **1a** and **2**. Spectrophotometric grade acetonitrile was used as a solvent.

Table S1. Absorption maxima (λ_{\max}) and molar extinction coefficients (ϵ) of the ligands and complexes

| Entry | $\lambda_{\max, \text{abs}} (\epsilon)$ [nm, ($10^4 \text{ M}^{-1} \text{ cm}^{-1}$)] |
|-------------------|--|
| L | 210 (2.06), 218 (1.82), 245 (3.21), 249 (3.66), 270 (1.48), 276 (1.36), 335 (1.47) |
| Complex 1a | 210 (4.7), 218 (4.34), 244 (6.72), 249 (7.49), 270 (3.41), 276 (3.2), 334 (3.18) |
| Complex 2 | 210 (4.54), 218 (4.17), 245 (6.58), 249 (7.36), 270 (3.28), 276 (3.07), 333 (3.06) |

S3. Spin-crossover characteristics of complexes **1b-d**

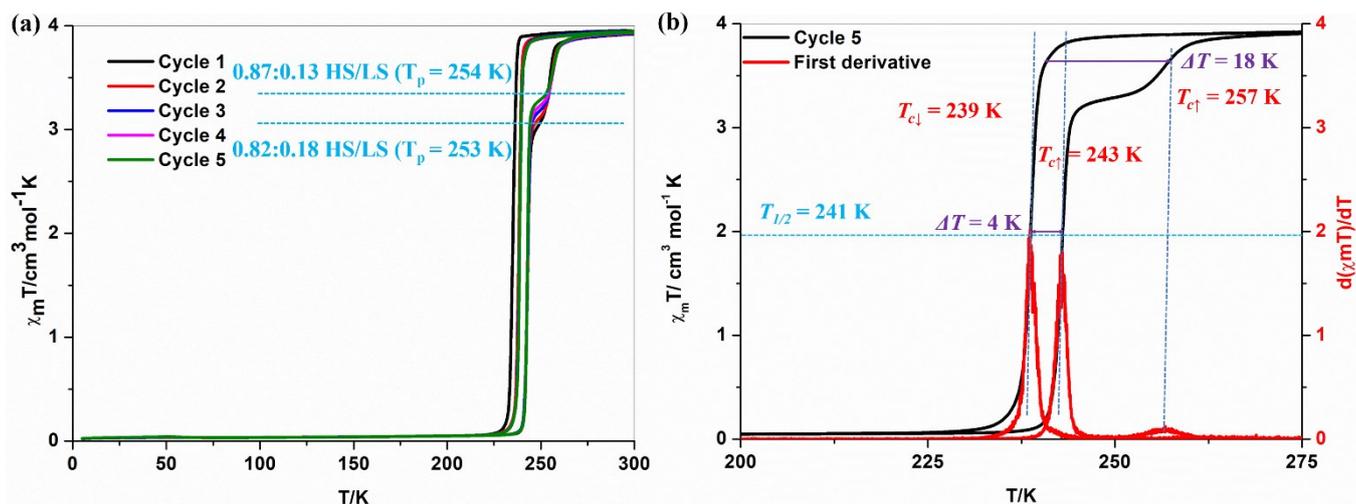


Figure S4. (a) $\chi_M T$ versus T plots of complex **1b**, and (b) $\chi_M T$ versus T and $d(\chi_M T)/dT$ versus T plots of complex **1b**, corresponding to cycle 5. T_p is the temperature, where the centre of the shoulder-like region situated. The critical temperatures (T_c) associated with the cooling and heating branches are designated as $T_{c\downarrow}$ and $T_{c\uparrow}$, respectively.

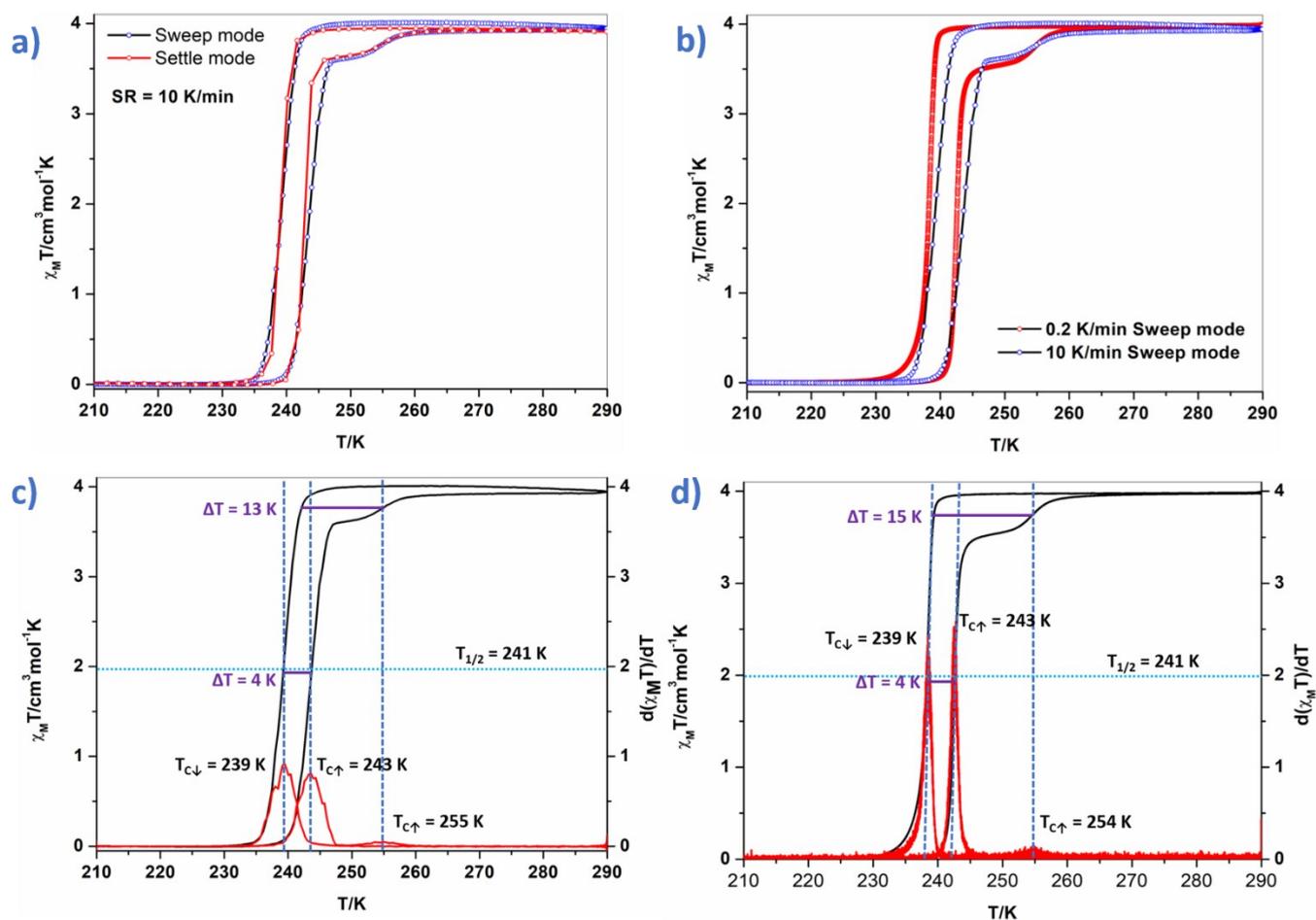


Figure S5. (a-b) $\chi_M T$ versus T plots of complex **1b** measured at scan rates of 0.2 K/min and 10 K/min. $\chi_M T$ versus T and $d(\chi_M T)/dT$ versus T plots of complex **1b** measured at (c) 10 K/min and (d) 0.2 K/min in sweep mode. The critical temperatures (T_c) associated with the cooling and heating branches are designated as $T_{c\downarrow}$ and $T_{c\uparrow}$, respectively. Note, the magnetic measurements with 0.2 K/min scan rate were performed in sweep mode. The comparable nature of $\chi_M T$ versus T profiles and $\Delta T = 4$ K observed for both the scan rates in the sweep mode indicates scan rate independent nature of ΔT in **1b**.

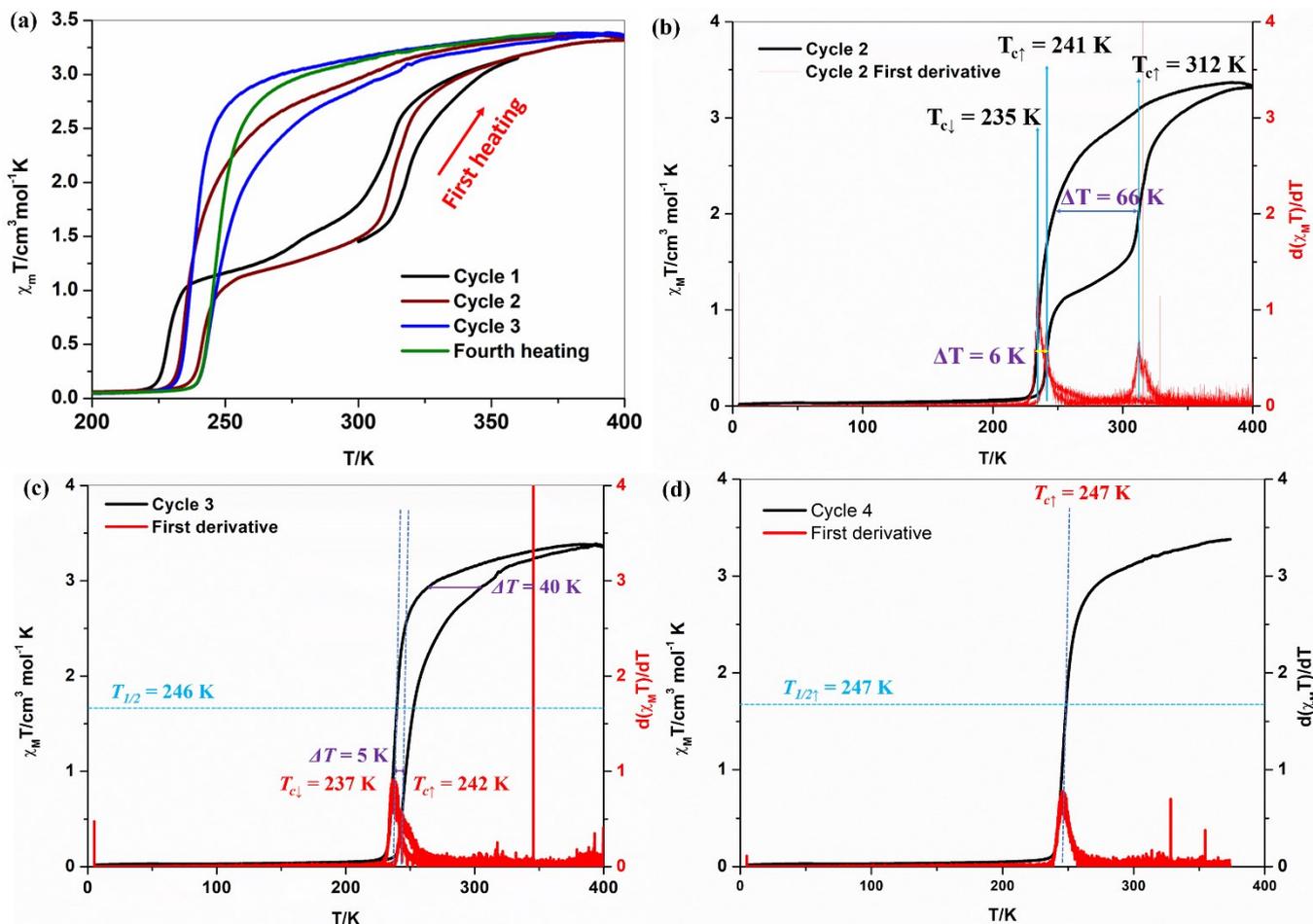


Figure S6. (a) $\chi_M T$ versus T plots and (b-c) $d(\chi_M T)/dT$ versus T and $\chi_M T$ versus T plots of complex **1c**. The critical temperatures (T_c) associated with the cooling and heating branches are designated as $T_{c\downarrow}$ and $T_{c\uparrow}$, respectively.

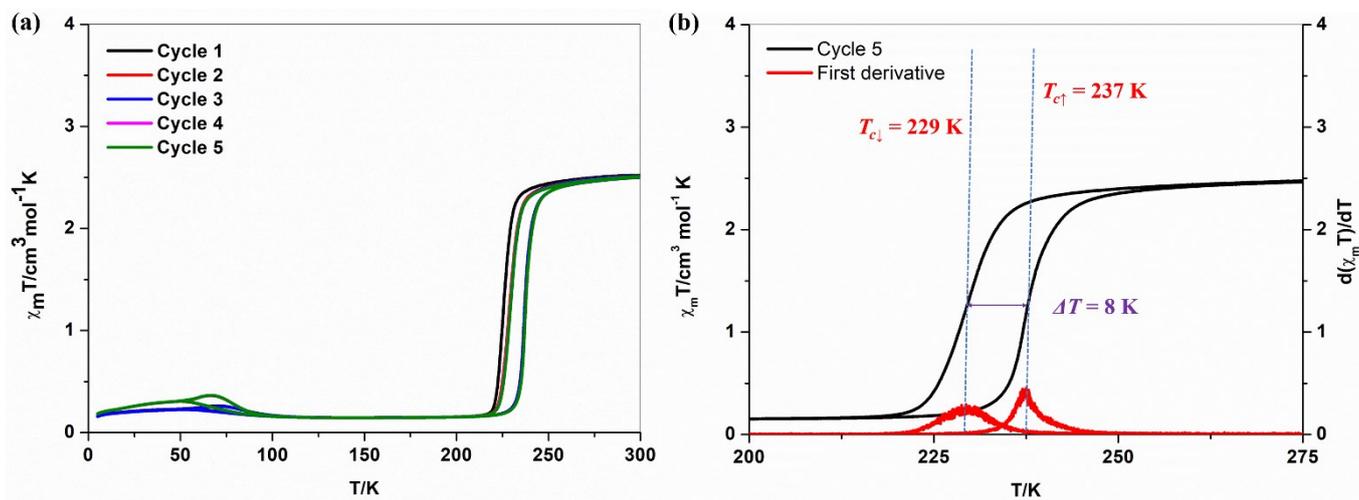


Figure S7. (a) $\chi_M T$ versus T plots, and (b) $d(\chi_M T)/dT$ versus T and $\chi_M T$ versus T plots of complex **1d** corresponding to cycle 5. The critical temperatures (T_c) associated with the cooling and heating branches are designated as $T_{c\downarrow}$ and $T_{c\uparrow}$, respectively.

S4. Differential scanning calorimetry studies of complexes **1c** and **1d**

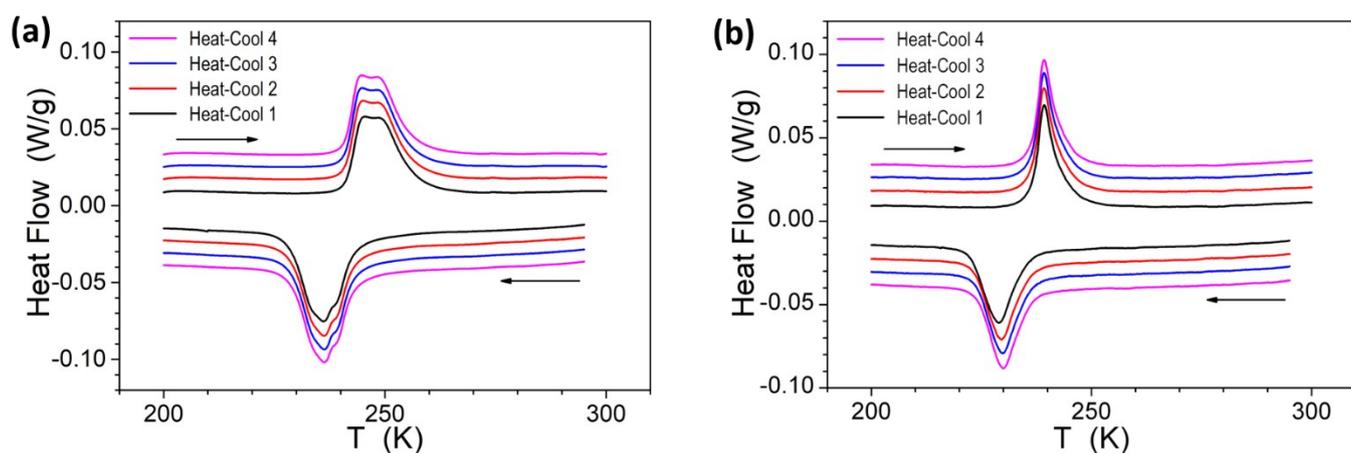


Figure S8. Differential scanning calorimetry (DSC) curves of complexes (a) solvent-free **1c** obtained after SQUID measurements and (b) **1d**. A scanning rate of 2 K/min, was employed; endotherm (up).