

Electronic Supplementary Information

Azo- and methylene-bridged mixed azoles for stable and insensitive energetic applications

Ajay Kumar Chinnam,^a Qiong Yu,^a Gregory H. Imler,^b Damon A. Parrish^b and Jean'ne M. Shreeve*^a

^a Department of Chemistry, University of Idaho, Moscow, Idaho, 83844-2343 USA.

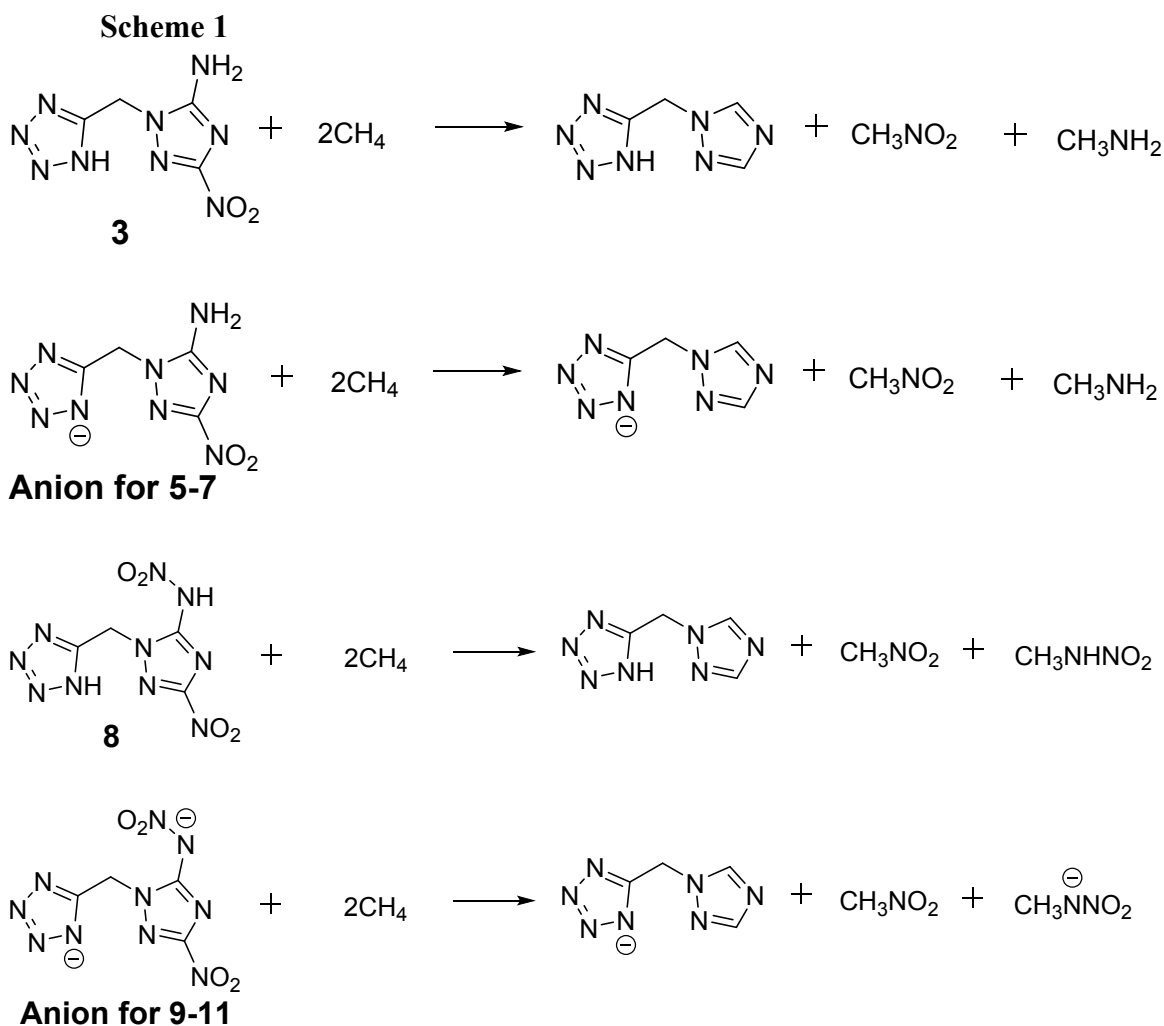
^b Naval Research Laboratory, 4555 Overlook Avenue, Washington, D.C. 20375 USA

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1. Theoretical calculations

Heats of formation for compounds **3**, **4**, **8** and anions of **5-9**, **9-11** were calculated based on isodesmic reactions (Scheme S1). The calculations were carried out using Gaussian 03 (Revision D.01) suite of programs.¹ The geometric optimization and frequency analyses of the structures were calculated using B3LYP/6-31+G** level. The gas phase enthalpy of formation was calculated, and the enthalpy of reaction was obtained by combining the MP2/6-311++G** energy difference for the reactions, the scaled zero-point energies (ZPE), values of thermal correction (HT), and other thermal factors.



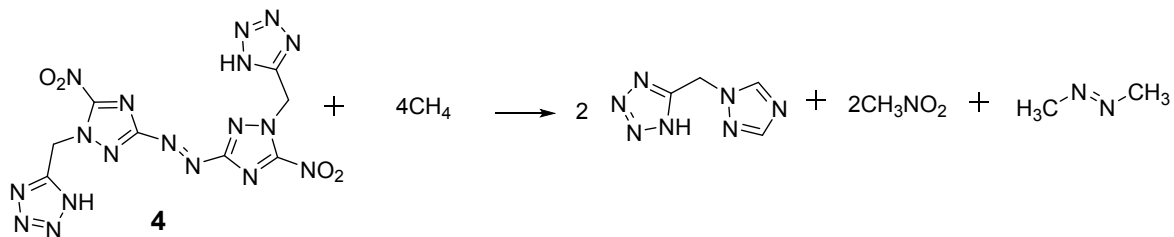


Table S1 Calculated zero point energy (ZPE), values of the correction (Hr), total energy (E0) and gas-state heats of formation (HOF).^a

Compound	ZPE [Hartree /Particle]	HT [Hartree /Particle]	E0 [kJ mol ⁻¹]	HOF (gas) [kJ mol ⁻¹]
2	0.134645	0.147861	-796.6711541	651.83
Anion of 5-7	0.121962	0.134489	-796.1815479	396.36
8	0.137093	0.152560	-1000.7422053	706.71
Anion of 9-11	0.110324	0.125400	-999.6214854	540.84
4	0.225204	0.249977	-1590.9076155	1292.81

^a The enthalpy of sublimation was calculated by using Trouton's rule. Solid-state heats of formation of the resulting compounds were calculated with Equation (1) in which T_m is the melting temperature.

$$\Delta H_f = \Delta H_f(g) - \Delta H_{sub} = \Delta H_f(g) - 188[\text{J mol}^{-1} \text{K}^{-1}] \times T_m \quad (1)$$

3. Crystal Structure Data

Table S2. Crystal data and structure refinement for 7.²⁻⁶

CCDC #	1985501	
Identification code	shrv813	
Empirical formula	C ₄ H ₈ N ₁₀ O ₃	
Formula weight	244.20	
Temperature	100(2) K	
Wavelength	1.54178 Å	
Crystal system	Triclinic	
Space group	P-1	
Unit cell dimensions	a = 4.2376(3) Å	α = 96.429(3)°.
	b = 7.3733(5) Å	β = 94.148(3)°.
	c = 15.1316(10) Å	γ = 96.090(3)°.
Volume	465.46(6) Å ³	
Z	2	
Density (-173°C)	1.742 Mg/m ³	
Density (23°C)	1.706 Mg/m ³	
Absorption coefficient	1.292 mm ⁻¹	
F(000)	252	
Crystal size	0.227 x 0.103 x 0.030 mm ³	
Theta range for data collection	2.949 to 74.798°.	
Index ranges	-5 ≤ h ≤ 4, -8 ≤ k ≤ 8, -15 ≤ l ≤ 18	
Reflections collected	3092	
Independent reflections	1531 [R _{int} = 0.0442]	
Completeness to theta = 67.679°	85.4 %	
Absorption correction	Semi-empirical from equivalents	
Max. and min. transmission	0.7538 and 0.5674	
Refinement method	Full-matrix least-squares on F ²	
Data / restraints / parameters	1531 / 0 / 158	
Goodness-of-fit on F ²	1.070	
Final R indices [I > 2σ(I)]	R ₁ = 0.0402, wR ₂ = 0.1059	
R indices (all data)	R ₁ = 0.0420, wR ₂ = 0.1090	
Largest diff. peak and hole	0.226 and -0.293 e.Å ⁻³	

Single-crystal X-ray structures of 7

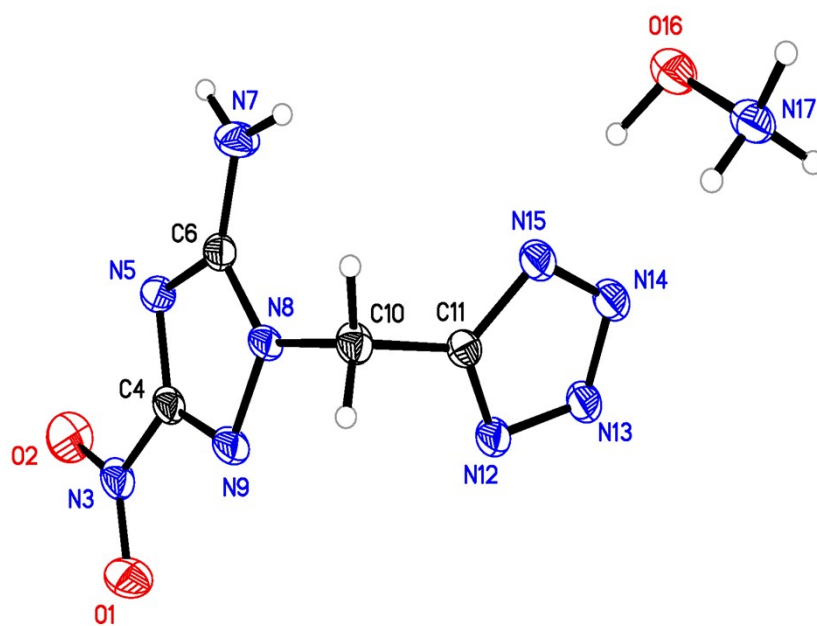


Figure S1. Single-crystal X-ray structure of 7 with numbering

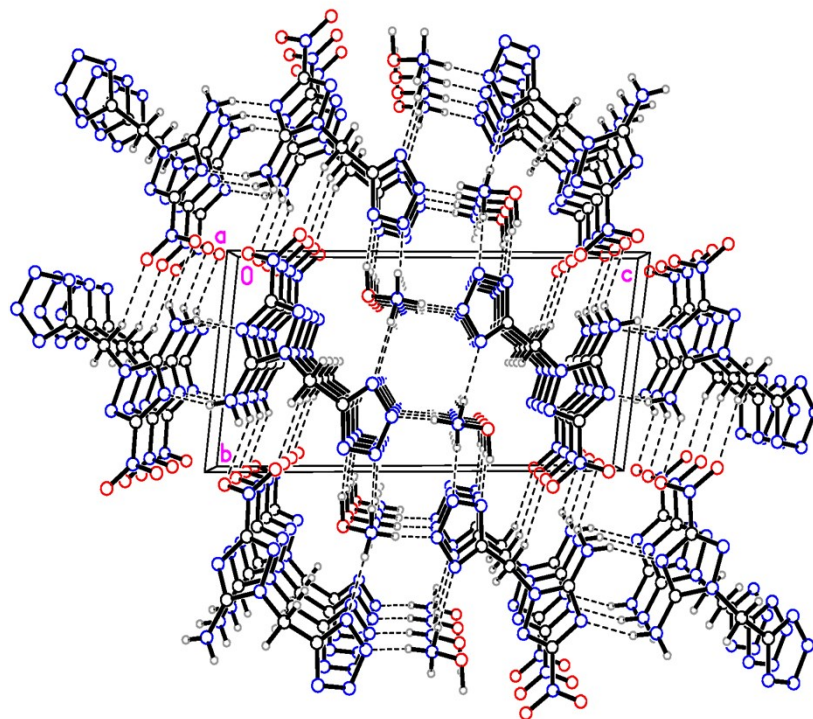


Figure S2. Unit cell view of **7** along a axis; hydrogen bonds are marked as dotted lines.

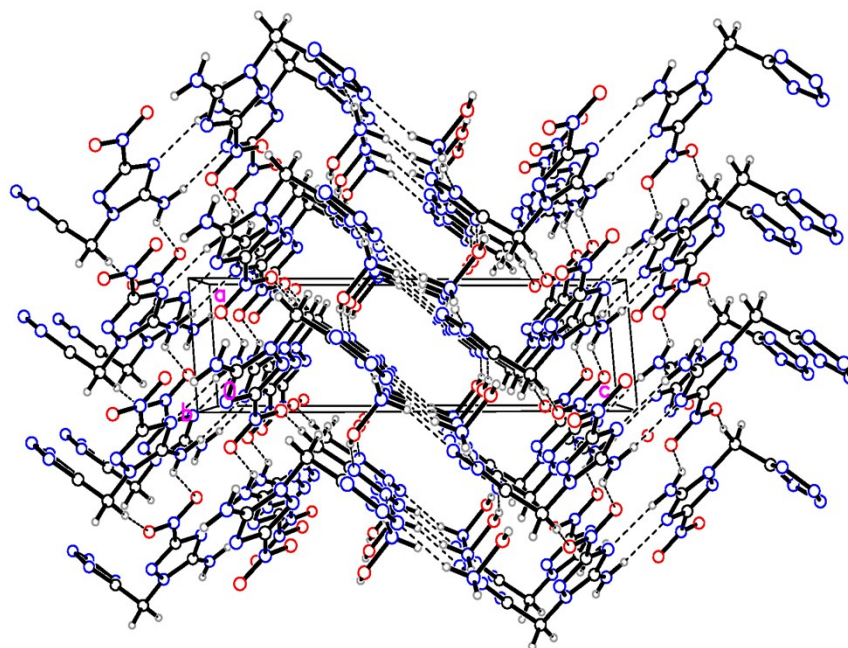


Figure S3. Unit cell view of **7** along b axis; hydrogen bonds are marked as dotted lines.

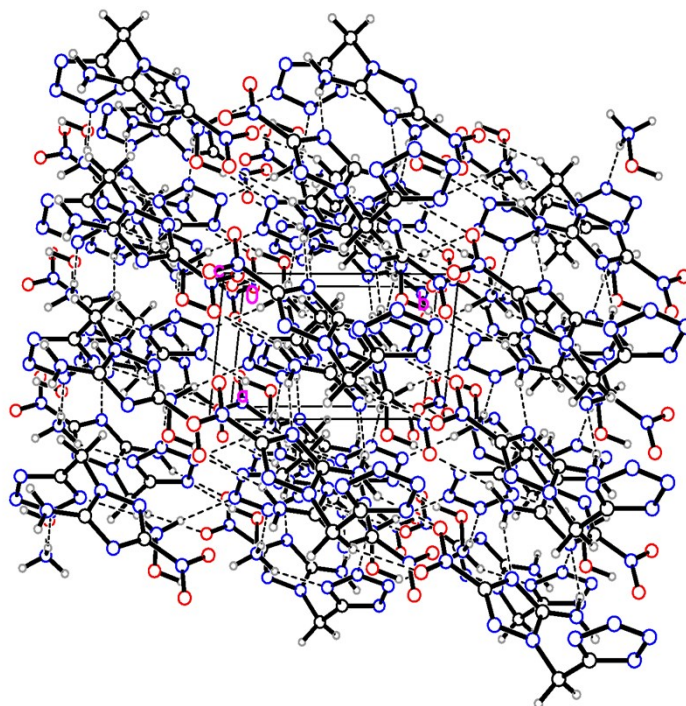


Figure S4. Unit cell view of **7** along *c* axis; hydrogen bonds are marked as dotted lines.

Table S3. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **7**. $U(\text{eq})$ is defined as one third of the trace of the orthogonalized U^{ij} tensor.

	x	y	z	U(eq)
O(1)	9918(3)	10398(2)	8150(1)	39(1)
O(2)	12547(3)	9670(2)	9309(1)	39(1)
N(3)	10469(3)	9342(2)	8694(1)	26(1)
C(4)	8563(4)	7569(2)	8599(1)	21(1)
N(5)	9039(3)	6366(2)	9183(1)	23(1)
C(6)	6852(4)	4952(2)	8857(1)	20(1)
N(7)	6361(4)	3379(2)	9200(1)	30(1)
N(8)	5240(3)	5390(2)	8112(1)	19(1)
N(9)	6375(3)	7111(2)	7936(1)	23(1)
C(10)	2866(3)	4251(2)	7479(1)	21(1)
C(11)	4440(3)	3251(2)	6742(1)	19(1)
N(12)	6134(3)	4086(2)	6161(1)	22(1)
N(13)	7278(3)	2711(2)	5664(1)	24(1)
N(14)	6275(3)	1137(2)	5935(1)	25(1)
N(15)	4455(3)	1430(2)	6618(1)	23(1)
O(16)	1537(3)	-2140(2)	6508(1)	31(1)
N(17)	-1033(3)	-2318(2)	5840(1)	25(1)

Table S4. Bond lengths [\AA] and angles [$^\circ$] for **7**.

O(1)-N(3)	1.221(2)	O(2)-N(3)	1.2191(19)
N(3)-C(4)	1.4491(19)	C(4)-N(9)	1.305(2)
C(4)-N(5)	1.340(2)	N(5)-C(6)	1.3430(19)
C(6)-N(7)	1.326(2)	C(6)-N(8)	1.362(2)
N(7)-H(7A)	0.8800	N(7)-H(7B)	0.8800
N(8)-N(9)	1.3708(18)	N(8)-C(10)	1.4618(18)
C(10)-C(11)	1.503(2)	C(10)-H(10A)	0.9900
C(10)-H(10B)	0.9900	C(11)-N(12)	1.336(2)
C(11)-N(15)	1.336(2)	N(12)-N(13)	1.3471(18)
N(13)-N(14)	1.312(2)	N(14)-N(15)	1.3458(19)
O(16)-N(17)	1.4163(18)	O(16)-H(16)	0.95(3)
N(17)-H(17A)	0.9100	N(17)-H(17B)	0.9100
N(17)-H(17C)	0.9100		
O(2)-N(3)-O(1)	124.42(14)	O(2)-N(3)-C(4)	117.44(14)
O(1)-N(3)-C(4)	118.13(14)	N(9)-C(4)-N(5)	118.97(14)
N(9)-C(4)-N(3)	120.38(15)	N(5)-C(4)-N(3)	120.65(14)
C(4)-N(5)-C(6)	101.41(13)	N(7)-C(6)-N(5)	125.42(15)
N(7)-C(6)-N(8)	125.56(14)	N(5)-C(6)-N(8)	109.02(14)
C(6)-N(7)-H(7A)	120.0	C(6)-N(7)-H(7B)	120.0
H(7A)-N(7)-H(7B)	120.0	C(6)-N(8)-N(9)	110.46(12)
C(6)-N(8)-C(10)	129.59(13)	N(9)-N(8)-C(10)	119.62(13)
C(4)-N(9)-N(8)	100.14(13)	N(8)-C(10)-C(11)	110.76(12)
N(8)-C(10)-H(10A)	109.5	C(11)-C(10)-H(10A)	109.5
N(8)-C(10)-H(10B)	109.5	C(11)-C(10)-H(10B)	109.5
H(10A)-C(10)-H(10B)	108.1	N(12)-C(11)-N(15)	111.86(14)
N(12)-C(11)-C(10)	123.97(13)	N(15)-C(11)-C(10)	124.07(15)
C(11)-N(12)-N(13)	104.54(13)	N(14)-N(13)-N(12)	109.45(13)
N(13)-N(14)-N(15)	109.65(12)	C(11)-N(15)-N(14)	104.49(13)
N(17)-O(16)-H(16)	107.1(14)	O(16)-N(17)-H(17A)	109.5
O(16)-N(17)-H(17B)	109.5	H(17A)-N(17)-H(17B)	109.5
O(16)-N(17)-H(17C)	109.5	H(17A)-N(17)-H(17C)	109.5
H(17B)-N(17)-H(17C)	109.5		

Table S5. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **7**. The anisotropic displacement factor exponent takes the form: $-2h^2a^*U^{11} + \dots + 2hk a^* b^* U^{12}$

	U^{11}	U^{22}	U^{33}	U^{23}	U^{13}	U^{12}
O(1)	56(1)	20(1)	40(1)	10(1)	3(1)	-8(1)
O(2)	40(1)	31(1)	41(1)	0(1)	-5(1)	-15(1)
N(3)	32(1)	17(1)	28(1)	-1(1)	8(1)	-4(1)
C(4)	25(1)	15(1)	23(1)	0(1)	6(1)	-2(1)
N(5)	28(1)	17(1)	22(1)	2(1)	4(1)	-4(1)
C(6)	24(1)	17(1)	20(1)	1(1)	6(1)	-1(1)
N(7)	41(1)	19(1)	27(1)	7(1)	-4(1)	-9(1)
N(8)	24(1)	12(1)	21(1)	2(1)	4(1)	-3(1)
N(9)	28(1)	15(1)	25(1)	3(1)	5(1)	-2(1)
C(10)	20(1)	18(1)	22(1)	1(1)	3(1)	-2(1)
C(11)	21(1)	16(1)	20(1)	2(1)	0(1)	-1(1)
N(12)	27(1)	18(1)	21(1)	2(1)	5(1)	-1(1)

N(13)	29(1)	19(1)	25(1)	0(1)	5(1)	0(1)
N(14)	31(1)	19(1)	26(1)	1(1)	7(1)	0(1)
N(15)	28(1)	18(1)	24(1)	1(1)	5(1)	-1(1)
O(16)	36(1)	21(1)	34(1)	4(1)	-1(1)	-3(1)
N(17)	28(1)	18(1)	29(1)	4(1)	7(1)	1(1)

Table S6. Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **7**.

	x	y	z	U(eq)
H(7A)	7521	3198	9682	36
H(7B)	4876	2515	8946	36
H(10A)	1373	5036	7221	25
H(10B)	1626	3349	7795	25
H(16)	2310(60)	-870(30)	6622(16)	46
H(17A)	-288	-2513	5292	37
H(17B)	-1998	-1270	5880	37

Table S7. Torsion angles [$^\circ$] for **7**.

O(2)-N(3)-C(4)-N(9)	-177.63(14)
O(1)-N(3)-C(4)-N(9)	1.5(2)
O(2)-N(3)-C(4)-N(5)	1.7(2)
O(1)-N(3)-C(4)-N(5)	-179.19(14)
N(9)-C(4)-N(5)-C(6)	-0.44(18)
N(3)-C(4)-N(5)-C(6)	-179.79(13)
C(4)-N(5)-C(6)-N(7)	179.51(15)
C(4)-N(5)-C(6)-N(8)	0.02(15)
N(7)-C(6)-N(8)-N(9)	-179.13(14)
N(5)-C(6)-N(8)-N(9)	0.36(16)
N(7)-C(6)-N(8)-C(10)	-5.9(2)
N(5)-C(6)-N(8)-C(10)	173.61(13)
N(5)-C(4)-N(9)-N(8)	0.64(17)
N(3)-C(4)-N(9)-N(8)	179.99(13)
C(6)-N(8)-N(9)-C(4)	-0.57(15)
C(10)-N(8)-N(9)-C(4)	-174.59(12)
C(6)-N(8)-C(10)-C(11)	-88.86(18)
N(9)-N(8)-C(10)-C(11)	83.86(16)
N(8)-C(10)-C(11)-N(12)	-63.1(2)
N(8)-C(10)-C(11)-N(15)	113.17(16)
N(15)-C(11)-N(12)-N(13)	-0.81(17)

C(10)-C(11)-N(12)-N(13)	175.88(13)
C(11)-N(12)-N(13)-N(14)	0.61(16)
N(12)-N(13)-N(14)-N(15)	-0.22(17)
N(12)-C(11)-N(15)-N(14)	0.69(17)
C(10)-C(11)-N(15)-N(14)	-176.01(13)
N(13)-N(14)-N(15)-C(11)	-0.28(17)

Table S8. Hydrogen bonds for **7** [\AA and $^\circ$].

D-H...A	d(D-H)	d(H...A)	d(D...A)	\angle (DHA)
N(7)-H(7A)...N(5)#1	0.88	2.14	2.9916(19)	161.7
N(7)-H(7B)...O(2)#2	0.88	2.36	3.0531(18)	135.7
O(16)-H(16)...N(15)	0.95(3)	1.84(3)	2.7688(18)	165(2)
N(17)-H(17A)...N(13)#4	0.91	2.00	2.876(2)	160.9
N(17)-H(17B)...N(14)#5	0.91	1.99	2.8927(19)	174.6
N(17)-H(17C)...N(12)#2	0.91	2.05	2.899(2)	155.3

Symmetry transformations used to generate equivalent atoms:

#1 $-x+2, -y+1, -z+2$ #2 $x-1, y-1, z$ #3 $x, y+1, z$

#4 $-x+1, -y, -z+1$ #5 $x-1, y, z$

Table S9. Crystal data and structure refinement for **10**.⁷⁻¹⁰

Formula	C ₄ H ₁₂ N ₁₄ O ₄
CCDC	1975376
$D_{calc.}/\text{g cm}^{-3}$	1.709
μ/mm^{-1}	1.287
Formula Weight	320.28
Colour	yellow
Shape	needle
Size/mm ³	0.37×0.14×0.05
T/K	173(1)
Crystal System	monoclinic
Space Group	$P2_1/c$
$a/\text{Å}$	8.63810(10)
$b/\text{Å}$	13.5120(2)
$c/\text{Å}$	10.66810(10)
$\alpha/^\circ$	90
$\beta/^\circ$	91.2230(10)
$\gamma/^\circ$	90
$V/\text{Å}^3$	1244.88(3)
Z	4
Z'	1
Wavelength/Å	1.541838
Radiation type	CuK $_{\alpha}$
$\theta_{min}/^\circ$	5.121
$\theta_{max}/^\circ$	68.400
Measured Refl.	10664
Independent Refl.	2288
Reflections with $I > 2(I)$	1956
R_{int}	0.0367
Parameters	247
Restraints	0
Largest Peak	0.229
Deepest Hole	-0.253
GooF	1.041
wR_2 (all data)	0.0885
wR_2	0.0837
R_1 (all data)	0.0401
R_1	0.0331

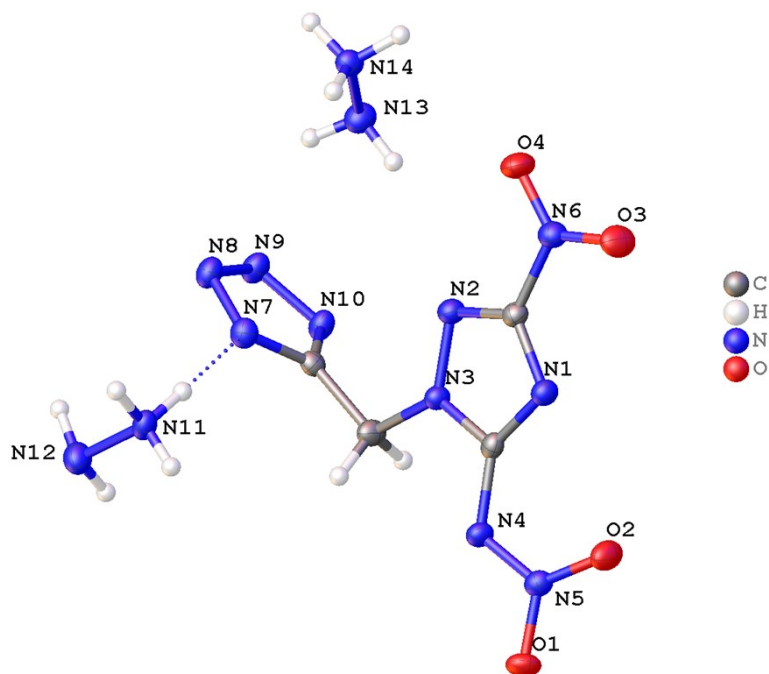


Figure S5. Single-crystal X-ray structure of **10** with numbering.

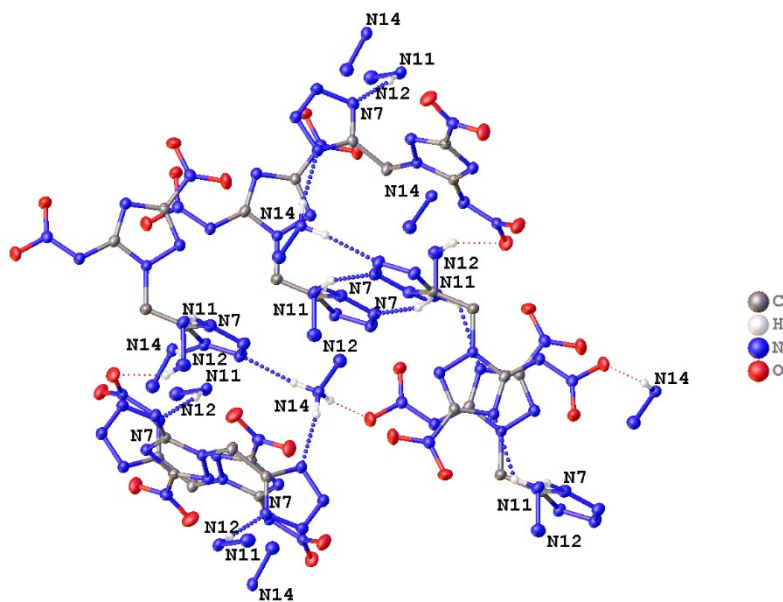


Figure S6. The following hydrogen bonding interactions with a maximum D-D distance of 3.1 Å and a minimum angle of 110° are present in **10**: N14–O1_1: 2.848 Å, N14–O3_2: 2.888 Å, N14–N10_3: 2.919 Å, N14–N9_4: 2.902 Å, N11–N1_1: 2.976 Å, N11–N7: 2.873 Å, N11–N8_5: 2.879 Å, N12–O1_6: 3.075 Å.

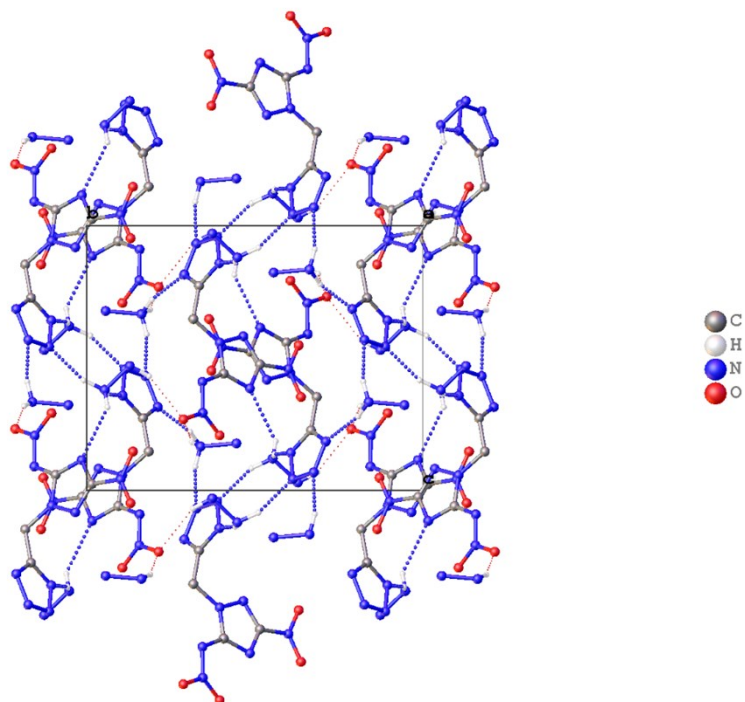


Figure S7. Packing diagram of **10**.

Table S10. Fractional Atomic Coordinates ($\times 10^4$) and Equivalent Isotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for **10**. U_{eq} is defined as 1/3 of the trace of the orthogonalized U_{ij} .

Atom	x	y	z	U_{eq}
O1	6791.1(12)	7132.0(8)	7316.2(11)	24.5(3)
O2	5087.4(14)	6054.4(9)	7918.0(10)	28.3(3)
O3	1331.9(14)	3633.5(9)	6468.7(11)	29.1(3)
O4	567.2(14)	3702.1(9)	4521.2(11)	30.5(3)
N3	3338.0(14)	6051.7(9)	4422.5(11)	16.1(3)
N1	3315.3(15)	5154.0(9)	6147.7(12)	18.2(3)
N14	245.2(16)	3380.7(10)	1659.6(13)	18.7(3)
N2	2259.8(14)	5348.2(9)	4178.5(12)	18.2(3)
N5	5663.2(14)	6555.0(9)	7070.8(12)	17.9(3)
N10	1724.1(14)	7104.2(10)	1847.4(12)	19.6(3)
N7	3611.9(15)	6063.4(10)	1420.1(12)	20.3(3)
N6	1336.5(15)	4003.4(10)	5417.0(12)	20.1(3)
N11	6813.7(16)	5527.7(10)	1203.9(12)	20.0(3)
N8	2630.7(15)	6124.6(10)	430.1(12)	20.9(3)
N9	1507.7(15)	6745.8(10)	686.0(12)	21.4(3)
N4	5158.8(14)	6566.5(10)	5902.1(12)	18.6(3)
N12	7227.7(17)	6272.9(11)	298.0(14)	25.7(3)
N13	118.4(17)	4445.7(10)	1775.8(15)	25.1(3)
C4	3020.7(16)	6665.4(11)	2269.9(14)	16.3(3)
C2	2321.9(17)	4856.8(11)	5241.2(13)	16.7(3)
C1	3970.3(16)	5932.0(11)	5590.7(13)	16.3(3)
C3	3737.4(19)	6836.8(12)	3531.1(14)	20.9(3)

Table S11. Anisotropic Displacement Parameters ($\times 10^4$) **10**. The anisotropic displacement factor exponent takes the form: $-2\sigma^2[h^2a^{*2} \times U_{11} + \dots + 2hka^* \times b^* \times U_{12}]$

Atom	U_{11}	U_{22}	U_{33}	U_{23}	U_{13}	U_{12}
O1	19.6(5)	25.9(6)	27.6(6)	-0.3(5)	-7.1(4)	-5.8(5)
O2	38.4(7)	27.9(6)	18.5(6)	4.1(5)	-3.5(5)	-9.8(5)
O3	37.0(7)	25.0(6)	25.1(6)	9.3(5)	-3.9(5)	-7.4(5)
O4	32.8(6)	33.3(7)	25.2(6)	-4.6(5)	-3.6(5)	-12.7(5)
N3	16.8(6)	16.4(6)	15.1(6)	0.8(5)	-1.6(5)	-0.6(5)
N1	19.6(6)	18.5(6)	16.5(6)	0.3(5)	-1.5(5)	0.0(5)
N14	18.0(7)	19.7(7)	18.3(7)	2.6(5)	-1.9(5)	-0.6(5)
N2	18.7(6)	17.8(6)	18.1(6)	-0.9(5)	-0.9(5)	-1.0(5)
N5	17.6(6)	16.7(6)	19.2(6)	-0.6(5)	-2.1(5)	2.1(5)
N10	21.4(6)	21.8(7)	15.6(6)	-0.8(5)	-0.4(5)	4.2(5)
N7	21.0(6)	19.4(7)	20.4(7)	1.0(5)	1.1(5)	3.6(5)
N6	20.3(6)	18.2(7)	21.8(7)	-1.4(5)	-0.3(5)	-0.6(5)
N11	20.4(7)	24.9(7)	14.7(6)	0.0(5)	0.4(5)	2.4(5)
N8	25.0(7)	20.8(7)	16.8(6)	-1.1(5)	0.6(5)	3.8(5)
N9	23.3(7)	23.5(7)	17.2(6)	-1.1(5)	-2.0(5)	4.3(5)
N4	19.2(6)	20.2(7)	16.4(6)	0.3(5)	-2.8(5)	-2.1(5)
N12	30.4(8)	24.5(8)	22.5(7)	0.7(6)	4.5(6)	-2.7(6)
N13	29.8(7)	21.5(7)	24.0(7)	-0.4(6)	-1.0(6)	-0.6(6)
C4	16.5(7)	15.1(7)	17.5(7)	2.9(6)	1.1(6)	-0.8(5)
C2	17.4(7)	15.9(7)	16.9(7)	-1.4(6)	0.1(6)	0.2(6)
C1	16.2(7)	18.8(7)	13.7(7)	-1.2(6)	0.6(5)	2.9(6)
C3	24.4(8)	20.3(8)	18.0(8)	3.8(6)	-3.1(6)	-4.3(6)

Table S12. Bond Lengths in Å for **10**.

Atom	Atom	Length/Å
O1	N5	1.2707(17)
O2	N5	1.2414(17)
O3	N6	1.2283(17)
O4	N6	1.2220(17)
N3	N2	1.3523(18)
N3	C1	1.3596(18)
N3	C3	1.4706(19)
N1	C2	1.3406(19)
N1	C1	1.339(2)
N14	N13	1.4487(19)
N2	C2	1.314(2)
N5	N4	1.3121(18)
N10	N9	1.3396(18)
N10	C4	1.3370(19)
N7	N8	1.3426(18)
N7	C4	1.328(2)
N6	C2	1.448(2)
N11	N12	1.4458(19)
N8	N9	1.3158(19)
N4	C1	1.3729(19)
C4	C3	1.487(2)

Table S13. Bond Angles in ° for **10**.

Atom	Atom	Atom	Angle/°
N2	N3	C1	110.75(12)
N2	N3	C3	123.56(12)
C1	N3	C3	125.69(12)
C1	N1	C2	100.66(12)
C2	N2	N3	100.01(12)
O1	N5	N4	115.22(12)
O2	N5	O1	120.11(12)
O2	N5	N4	124.62(13)
C4	N10	N9	104.53(12)
C4	N7	N8	104.70(12)
O3	N6	C2	117.11(13)
O4	N6	O3	124.50(14)
O4	N6	C2	118.39(13)
N9	N8	N7	109.38(12)
N8	N9	N10	109.49(12)
N5	N4	C1	117.00(12)
N10	C4	C3	124.44(14)
N7	C4	N10	111.90(13)
N7	C4	C3	123.66(14)
N1	C2	N6	120.94(13)
N2	C2	N1	119.00(14)
N2	C2	N6	120.06(13)
N3	C1	N4	115.54(13)
N1	C1	N3	109.57(13)
N1	C1	N4	134.74(14)
N3	C3	C4	111.95(12)

Table S14. Torsion Angles in ° for **10**.

Atom	Atom	Atom	Atom	Angle/°
O1	N5	N4	C1	178.16(12)
O2	N5	N4	C1	-4.1(2)
O3	N6	C2	N1	7.6(2)
O3	N6	C2	N2	-72.96(14)
O4	N6	C2	N1	-72.67(14)
O4	N6	C2	N2	6.8(2)
N3	N2	C2	N1	-0.21(17)
N3	N2	C2	N6	-79.70(12)
N2	N3	C1	N1	-0.57(16)
N2	N3	C1	N4	175.59(12)
N2	N3	C3	C4	-9.7(2)
N5	N4	C1	N3	171.68(12)
N5	N4	C1	N1	-13.4(2)
N10	C4	C3	N3	99.03(17)
N7	N8	N9	N10	-0.11(17)
N7	C4	C3	N3	-81.91(18)
N8	N7	C4	N10	-0.56(16)
N8	N7	C4	C3	-79.72(13)
N9	N10	C4	N7	0.50(17)
N9	N10	C4	C3	179.65(14)

Atom	Atom	Atom	Atom	Angle/°
C4	N10	N9	N8	-0.23(16)
C4	N7	N8	N9	0.40(16)
C2	N1	C1	N3	0.39(15)
C2	N1	C1	N4	-74.73(16)
C1	N3	N2	C2	0.45(15)
C1	N3	C3	C4	170.98(13)
C1	N1	C2	N2	-0.12(18)
C1	N1	C2	N6	179.37(13)
C3	N3	N2	C2	-79.00(13)
C3	N3	C1	N1	178.87(13)
C3	N3	C1	N4	-5.0(2)

Table S15. Hydrogen Fractional Atomic Coordinates ($\times 10^4$) and Equivalent Isotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for **10**. U_{eq} is defined as 1/3 of the trace of the orthogonalised U_{ij} .

Atom	x	y	z	U_{eq}
H11A	7030(20)	4898(16)	840(20)	37(6)
H13A	650(20)	4698(16)	1210(20)	32(5)
H11B	5790(20)	5580(14)	1365(17)	22(5)
H14A	-80(20)	3221(13)	894(18)	19(4)
H14B	-390(20)	3067(15)	2206(19)	30(5)
H14C	1210(20)	3124(15)	1820(18)	28(5)
H12A	7060(30)	6849(19)	680(20)	48(7)
H13B	560(20)	4593(15)	2510(20)	31(5)
H12B	6480(20)	6205(15)	-320(20)	31(5)
H3A	4840(20)	6856(15)	3486(18)	30(5)
H3B	3390(20)	7439(15)	3888(17)	24(5)
H11C	7330(20)	5614(16)	1860(20)	35(6)

Table S16. Hydrogen Bond information for **10**.

D	H	A	d(D-H)/\AA	d(H-A)/\AA	d(D-A)/\AA	D-H-A/deg
N11	H11A	N8 ¹	0.95(2)	1.96(2)	2.8790(19)	160.3(18)
N11	H11B	N7	0.91(2)	1.99(2)	2.8729(19)	162.8(17)
N14	H14A	N9 ²	0.885(19)	2.067(19)	2.9020(18)	156.9(16)
N14	H14B	N10 ³	0.91(2)	2.02(2)	2.9191(18)	166.9(18)
N14	H14C	O1 ⁴	0.91(2)	1.97(2)	2.8475(17)	159.9(18)
N14	H14C	O3 ⁵	0.91(2)	2.41(2)	2.8875(18)	112.9(15)
N12	H12A	O1 ⁶	0.89(3)	2.24(3)	3.0753(19)	156(2)
N11	H11C	N1 ⁴	0.83(2)	2.44(2)	2.9760(18)	122.8(18)

¹1-x,1-y,-z; ²-x,1-y,-z; ³-x,-1/2+y,1/2-z; ⁴1-x,1-y,1-z; ⁵+x,1/2-y,-1/2+z; ⁶+x,3/2-y,-1/2+z

3. DSC Curves

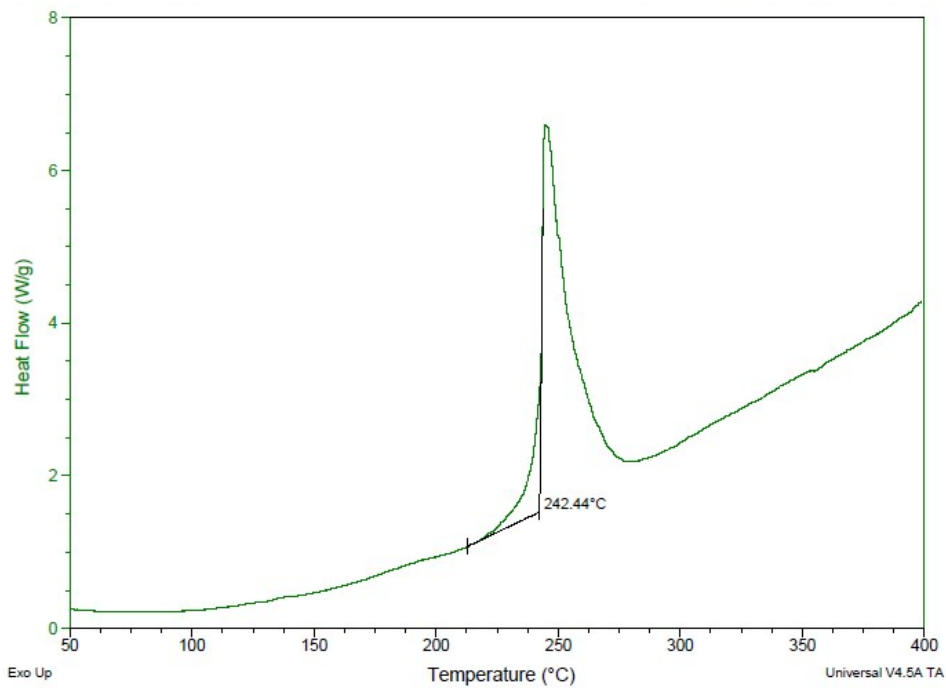


Figure S8. DSC Curve of 3

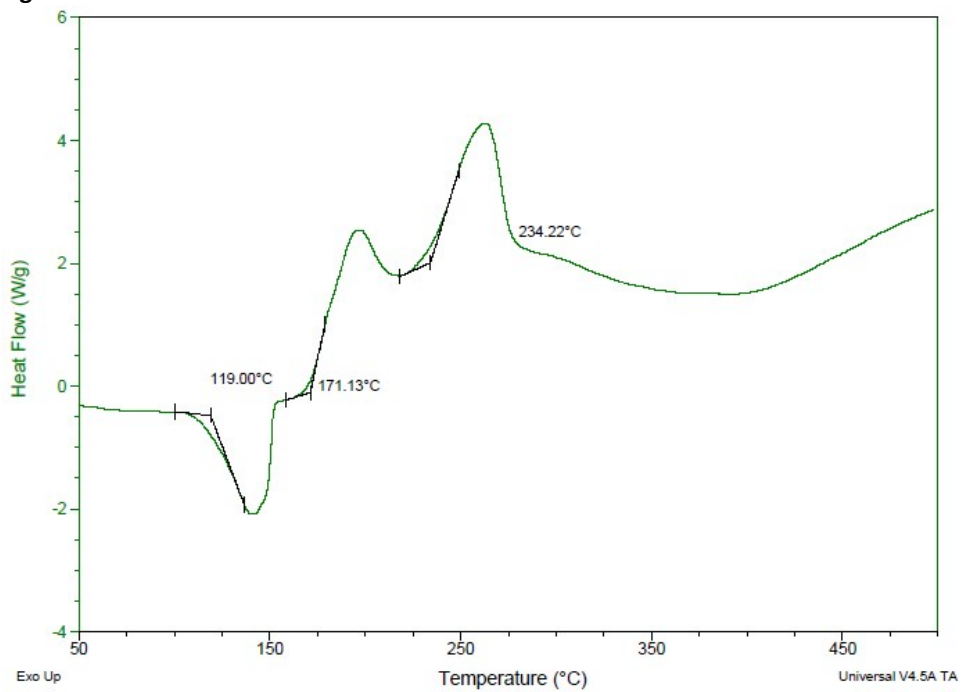


Figure S9. DSC Curve of 4

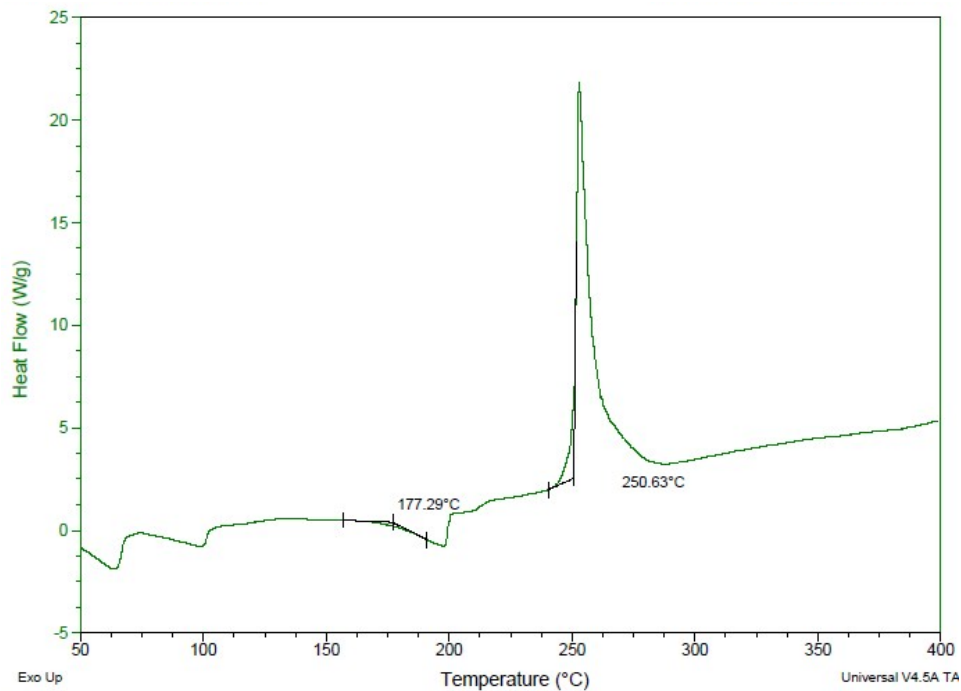


Figure S10. DSC Curve of 5

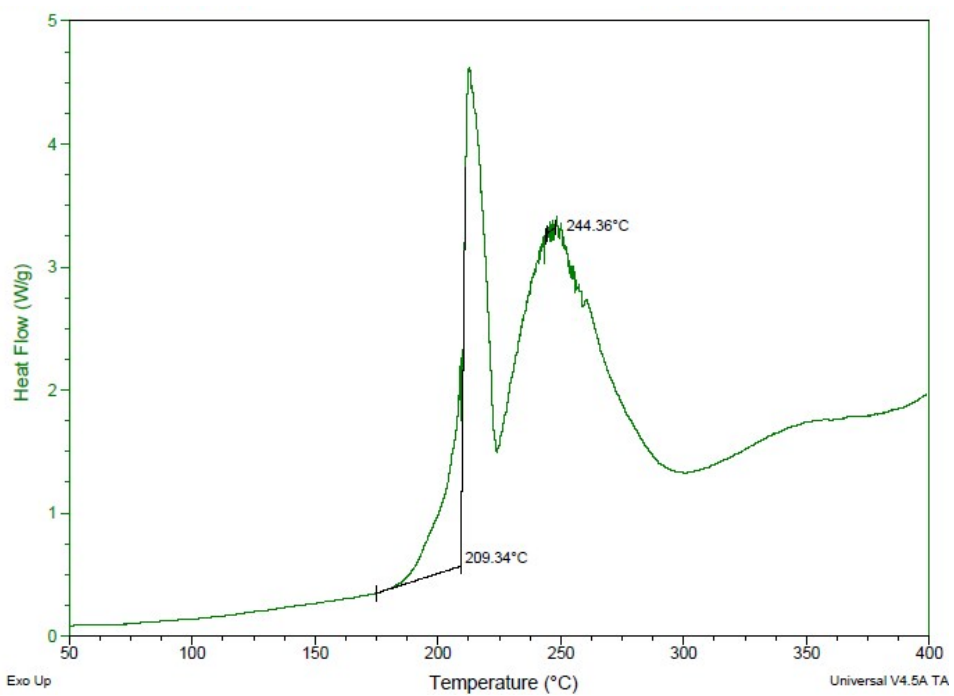


Figure S11. DSC Curve of 6

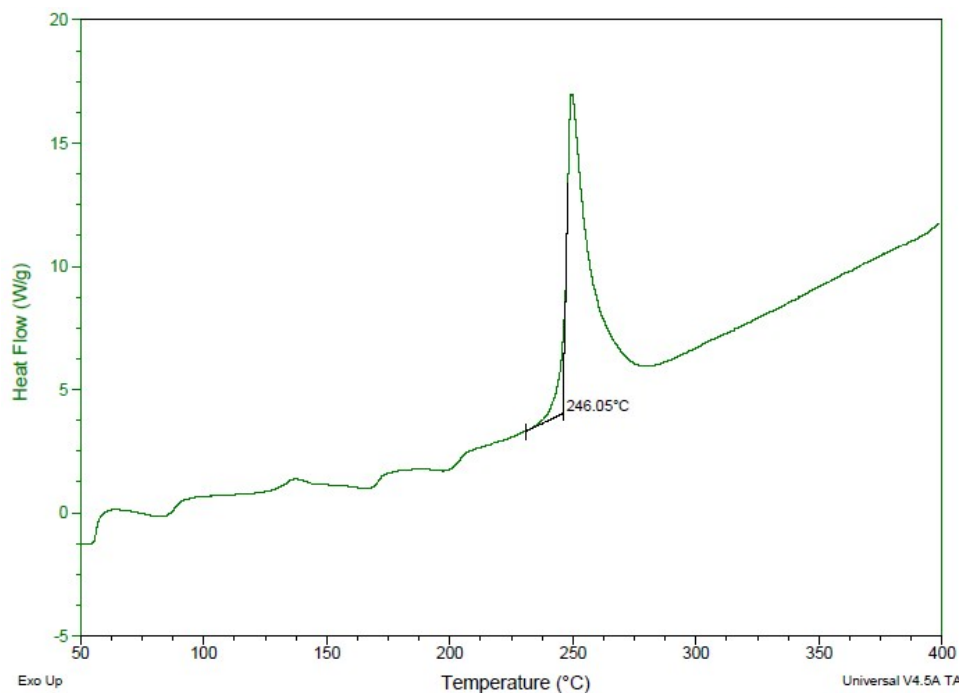


Figure S12. DSC Curve of 7

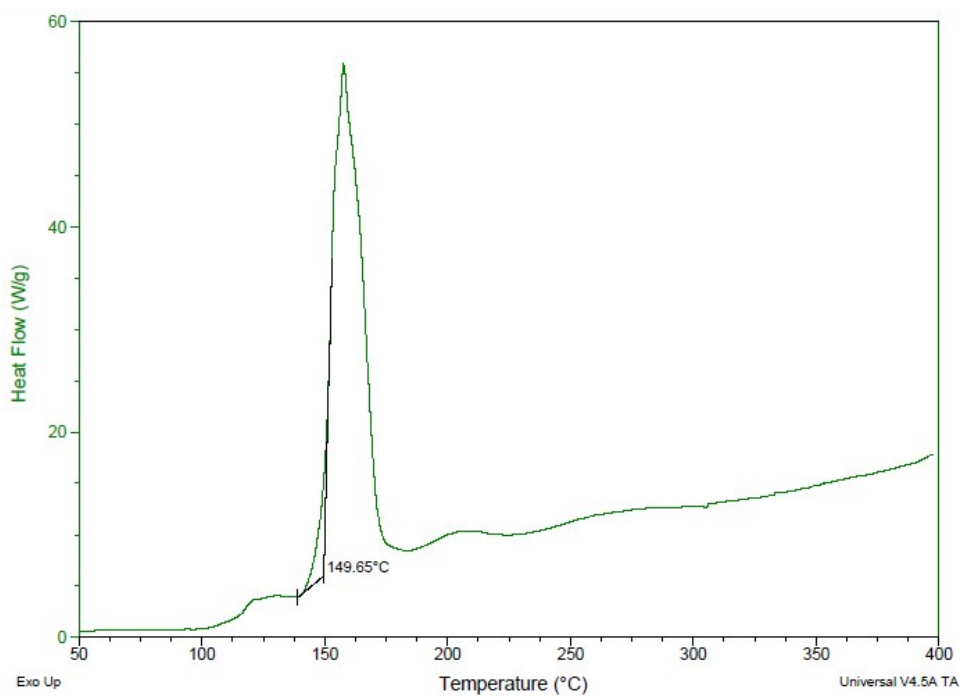


Figure S13. DSC Curve of 8

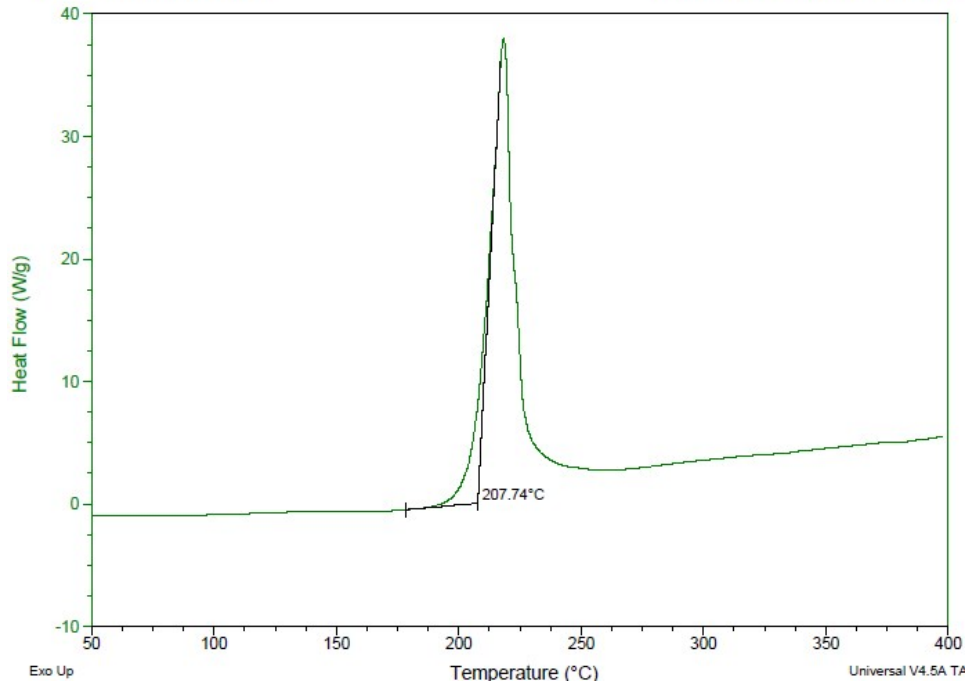


Figure S14. DSC Curve of 9

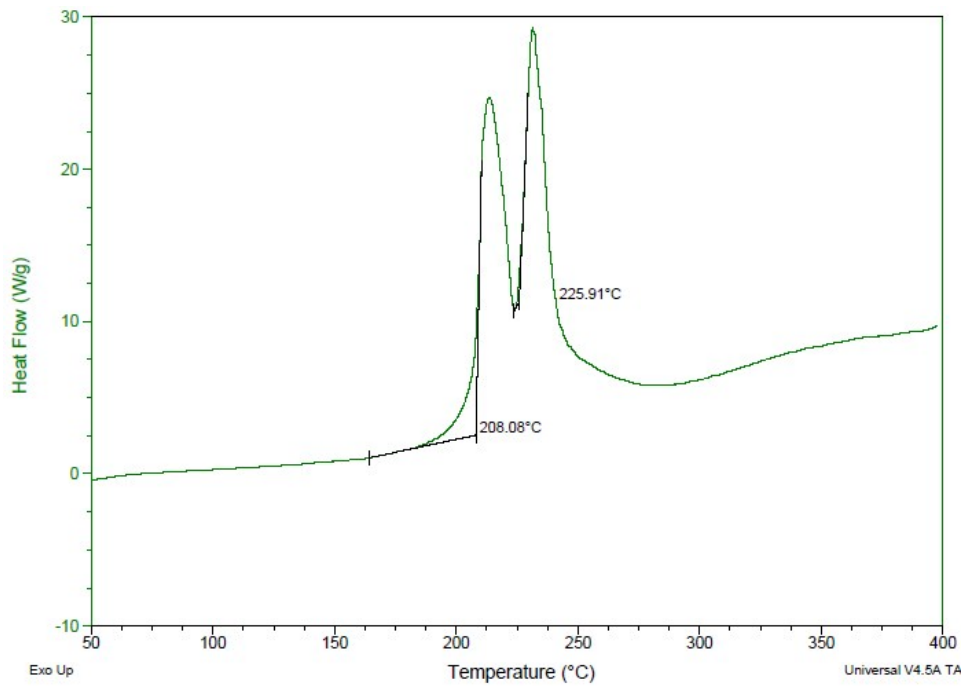


Figure S15. DSC Curve of 10

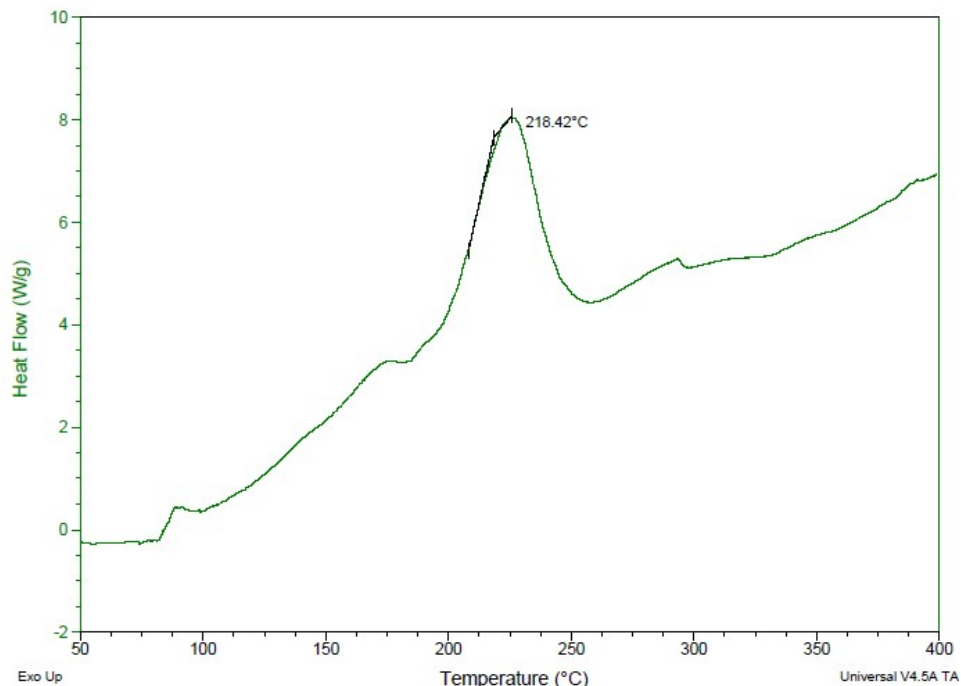


Figure S16. DSC Curve of 11

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