## **Supporting Information**

## A Free-Standing, Self-Healing Multi Stimuli Responsive Gel Showing Cryogenic Magnetic Cooling

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Fig. S1 Preparation of xerogel.



**Fig. S2** (a) Solution of the reaction of anhy.  $FeCl_3$  salt with  $H_2Pic$  in 1.5 : 1 molar ratio; (b) Solution of the reaction of anhy.  $FeCl_3$  salt with  $H_2Pic$  in 3 : 1 molar ratio (c) solution of only ligand.



**Fig. S3** (a) Solution of the reaction of  $Fe(NO_3)_3 \cdot 9H_2O$  and  $H_2L_3(b)$  Solution of the reaction of  $FeSO_4$  and  $H_2L_3(c)$  Solution of the reaction of  $Fe(CIO_4)_3 \cdot 6H_2O$  and  $H_2L_3$ .



**Fig. S4** (a) Molecular structure of  $[Dipic(H_2O)FeOH]_2$ ·H<sub>2</sub>O.



Fig. S5 PXRD pattern of xero gel from freshly prepared gel and the crystal of  $[Dipic(H_2O)FeOH]_2$ · $H_2O$ .



**Fig. S6** Packing diagram of [Dipic(H<sub>2</sub>O)FeOH]<sub>2</sub>·H<sub>2</sub>O along a-axis.



Fig. S7 TEM images showing (a) fibrous morphology, (b) HR image of freshly prepared gel.



Fig. S8 TG analysis of the freshly prepared xerogel.



Fig. S9 FT-IR plot of as synthesized gel.



**Fig. S10** Comparison of FT-IR plots of as synthesized gel and crystalline material of  $[Dipic(H_2O)FeOH]_2$ ·H<sub>2</sub>O.



Fig. S11 EPR spectrum of as synthesized gel.



Fig. S12 Probable self-assembly of the formation of synthesized gel.



Fig. S13 Stress sweep experiment of the gel.



Fig. S14 Frequency sweep experiment of the gel.



Fig. S15 Reduction of Fe(III) to Fe(II) of metallogel by ascorbic acid.



Fig. S16 Comparison of EPR spectra of as synthesized gel and the gel after addition of NH<sub>3</sub>.



Fig. S17 Probable change in geometry around Fe(III) during  $NH_3$  sensing.



Fig. S18 Comparison of strength between ammoniacal gel and gel as-synthesized.



Fig. S19 Isothermal normalized magnetizations vs. field/temperature plot of the xerogel.



**Fig. S20** Magnetic entropy changes of the  $[Dipic(H_2O)FeOH]_2$ · $H_2O$ .