

Supporting Information

Selenite Bromides Nonlinear Optical Materials **Pb₂GaF₂(SeO₃)₂Br and Pb₂NbO₂(SeO₃)₂Br: Synthesis and Characterization**

Huanke Zhao ^a, Pifu Gong ^b, Xinyuan Zhang ^{a*}, Zheshuai Lin ^b, Zhanggui Hu ^a and Yicheng Wu ^a

^a. *Tianjin Key Laboratory of Functional Crystal Materials, Institute of Functional Crystals, Tianjin University of Technology, Tianjin 300384, China.*

^b. *Key Laboratory of Functional Crystals and Laser Technology, Beijing Center for Crystal Research and Development, Technical Institute of Physics and Chemistry, Chinese Academy of Sciences, Beijing 100190, China.*

1. Tables

Table S1. Bond Lengths and Bond Angles for Pb₂GaF₂(SeO₃)₂Br.

Pb ₂ GaF ₂ (SeO ₃) ₂ Br			
Pb1-Br1 ¹	3.1407(18)	Se1-O5	1.717(13)
Pb1-Br1	3.0065(18)	Br1-Pb1 ⁴	3.1407(18)
Pb1-O1	2.567(15)	Ga1-F2	1.837(10)
Pb1-O1 ²	2.565(13)	Ga1-O6	1.974(12)
Pb1-O2 ³	2.669(13)	Ga1-F1 ⁵	1.995(9)
Pb1-O4 ¹	2.596(12)	Ga1-F1	1.962(9)
Pb2-Br1	3.095(2)	Ga1-O5 ⁴	1.985(12)
Pb2-O1 ⁴	2.650(14)	Ga1-O3 ⁶	1.911(13)
Pb2-O2	2.686(13)	Se2-O2	1.696(13)
Pb2-O4	2.573(11)	Se2-O3	1.721(12)
Pb2-O3 ⁴	2.649(12)	Se1-O6	1.730(12)
Se2-O1	1.711(14)	Se1-O4	1.672(10)
Br1-Pb1-Br1 ¹	123.27(6)	F1-Ga1-O6	171.2(5)
F2-Ga1-O6	96.7(5)	F1-Ga1-F1 ⁵	88.68(12)
F2-Ga1-F1 ⁵	88.3(5)	F1-Ga1-O5 ⁴	91.0(5)
F2-Ga1-F1	91.7(5)	O5 ⁴ -Ga1-F1 ⁵	177.7(5)
F2-Ga1-O5 ⁴	94.0(5)	O3 ⁶ -Ga1-O6	83.3(5)
F2-Ga1-O3 ⁶	176.7(5)	O3 ⁶ -Ga1-F1	88.2(5)
O6-Ga1-F1 ⁵	89.0(5)	O3 ⁶ -Ga1-F1 ⁵	88.4(5)
O6-Ga1-O5 ⁴	91.0(5)	O3 ⁶ -Ga1-O5 ⁴	89.3(6)
O1-Se2-O3	96.2(6)	Se2-O1-Pb1 ³	112.3(6)
O2-Se2-O1	99.0(6)	Se2-O1-Pb1	105.1(6)
O2-Se2-O3	103.1(6)	Se2-O1-Pb2 ¹	103.1(7)
O4-Se1-O6	100.1(6)	Pb1 ² -O2-Pb2	112.0(5)
O4-Se1-O5	98.6(6)	Se2-O2-Pb1 ²	108.6(6)
O5-Se1-O6	100.4(6)	Se2-O2-Pb2	122.3(7)
Pb1-Br1-Pb1 ⁴	123.27(6)	Se1-O6-Ga1	122.0(7)
Pb1-Br1-Pb2	92.27(5)	Pb2-O4-Pb1 ⁴	103.4(4)
Pb2-Br1-Pb1 ⁴	81.17(4)	Se1-O4-Pb1 ⁴	132.5(7)
Se2-O3-Pb2 ¹	102.9(5)	Se1-O4-Pb2	111.0(5)
Se2-O3-Ga1 ⁸	125.9(6)	Ga1-F1-Ga1 ⁷	159.2(5)
Ga1 ⁸ -O3-Pb2 ¹	122.0(6)	Se1-O5-Ga1 ¹	123.5(7)

¹+X,-1+Y,+Z; ²2-X,1/2+Y,2-Z; ³2-X,-1/2+Y,2-Z; ⁴+X,1+Y,+Z; ⁵-X,-1/2+Y,1-Z; ⁶1-X,3/2+Y,1-Z; ⁷-X,1/2+Y,1-Z; ⁸1-X,-3/2+Y,1-Z
¹+X,-1+Y,+Z; ²2-X,1/2+Y,2-Z; ³2-X,-1/2+Y,2-Z; ⁴+X,1+Y,+Z; ⁵-X,-1/2+Y,1-Z; ⁶1-X,3/2+Y,1-Z; ⁷-X,1/2+Y,1-Z; ⁸1-X,-3/2+Y,1-Z

Table S2. Bond Lengths and Bond Angles for $\text{Pb}_2\text{NbO}_2(\text{SeO}_3)_2\text{Br}$.

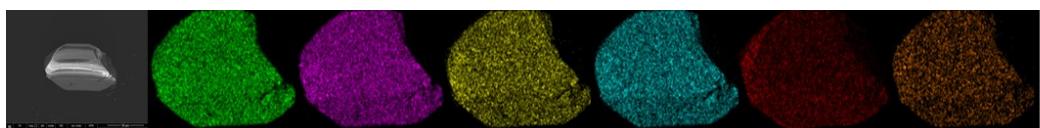
$\text{Pb}_2\text{NbO}_2(\text{SeO}_3)_2\text{Br}$			
Pb1-Br1 ¹	3.2051(14)	Nb1-O2 ⁶	2.220(8)
Pb1-Br1	3.0263(13)	Se1-O3	1.698(9)
Pb1-O3 ¹	2.664(9)	Se1-O1	1.731(9)
Pb1-O1	2.525(9)	Se1-O2	1.707(8)
Pb1-O1 ²	2.608(9)	Se2-O4	1.673(8)
Pb1-O4 ²	2.700(8)	Se2-O6	1.721(9)
Pb1-O8 ³	2.672(8)	Se2-O5	1.738(9)
Pb2-Br1 ²	3.0725(13)	Nb1-O6	2.144(9)
Pb2-O3 ¹	2.639(9)	Nb1-O7	1.938(8)
Pb2-O1	2.603(9)	Nb1-O7 ⁴	1.920(8)
Pb2-O4	2.524(9)	Nb1-O8	1.763(8)
Pb2-O2	2.594(9)	Nb1-O5 ⁵	2.110(9)
Br1-Pb1-Br1 ¹	122.17(4)	Pb2-O4-Pb1 ⁷	102.6(3)
O7-Nb1-O2 ⁴	86.4(3)	O2-Pb2-O3 ¹	108.9(3)
O75-Nb1-O2 ⁴	83.8(3)	O2-Pb2-O1	58.9(3)
O8-Nb1-O6	98.3(4)	O6-Nb1-O2 ⁴	75.1(3)
O8-Nb1-O7	98.6(4)	O7 ⁵ -Nb1-O6	158.9(4)
O8-Nb1-O7 ⁵	102.6(4)	O7-Nb1-O6	85.6(3)
O8-Nb1-O5 ⁶	92.6(4)	O7 ⁵ -Nb1-O7	94.32(12)
O8-Nb1-O2 ⁴	171.4(3)	O7 ⁵ -Nb1-O5 ⁶	91.5(4)
O5 ⁶ -Nb1-O6	84.4(3)	O7-Nb1-O5 ⁶	166.0(4)
O5 ⁶ -Nb1-O2 ⁴	81.5(3)	Se2-O4-Pb1 ⁷	127.8(5)
O3-Se1-O1	99.4(4)	Se2-O4-Pb2	114.2(4)
O3-Se1-O2	103.6(4)	Se2-O6-Nb1	119.2(5)
O2-Se1-O1	96.0(4)	Nb1 ⁸ -O7-Nb1	151.9(5)
O4-Se2-O6	100.5(4)	Nb1-O8-Pb1 ⁹	163.7(6)
O4-Se2-O5	99.2(4)	Se2-O5-Nb1 ¹	121.3(5)
O6-Se2-O5	101.0(4)	Nb1 ¹⁰ -O2-Pb2	126.2(3)
Pb1-Br1-Pb1 ⁶	122.17(4)	Se1-O2-Pb2	103.0(4)
Pb1-Br1-Pb2	793.61(4)	Se1-O2-Nb1 ¹⁰	120.0(4)
Pb2 ⁷ -Br1-Pb1	681.01(3)	Pb1-O1-Pb2	115.1(3)
Pb2 ⁶ -O3-Pb1 ⁶	109.4(3)	Pb2-O1-Pb1 ⁷	103.0(3)
Se1-O3-Pb1 ⁶	108.6(5)	Se1-O1-Pb1 ⁷	106.2(4)
Se1-O3-Pb2 ⁶	125.9(4)	Se1-O1-Pb1	111.3(4)
Pb1-O1-Pb1 ⁷	117.6(3)	Se1-O1-Pb2	102.0(4)

¹+X,-1+Y,+Z; ²-X,-1/2+Y,1-Z; ³-1+X,-1+Y,+Z; ⁴2-X,1/2+Y,2-Z; ⁵+X,1+Y,+Z; ⁶1-X,1/2+Y,2-Z; ⁷-X,1/2+Y,1-Z; ⁸2-X,-1/2+Y,2-Z; ⁹1+X,1+Y,+Z; ¹⁰1-X,-1/2+Y,2-Z
¹+X,-1+Y,+Z; ²-X,-1/2+Y,1-Z; ³-1+X,-1+Y,+Z; ⁴1-X,1/2+Y,2-Z; ⁵2-X,1/2+Y,2-Z;
⁶+X,1+Y,+Z; ⁷-X,1/2+Y,1-Z; ⁸2-X,-1/2+Y,2-Z; ⁹1+X,1+Y,+Z; ¹⁰1-X,-1/2+Y,2-Z

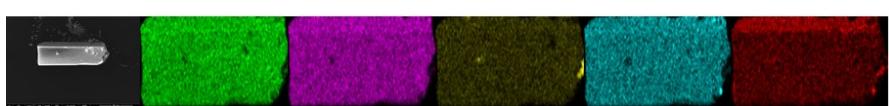
Table S3. The calculated SHG coefficients of $\text{Pb}_2\text{GaF}_2(\text{SeO}_3)_2\text{Br}$ and $\text{Pb}_2\text{NbO}_2(\text{SeO}_3)_2\text{Br}$.

	d_{16}	d_{14}	d_{22}	d_{33}	d_{powder}
$\text{Pb}_2\text{GaF}_2(\text{SeO}_3)_2\text{Br}$	-0.55	-1.17	5.37	-2.08	2.58
$\text{Pb}_2\text{NbO}_2(\text{SeO}_3)_2\text{Br}$	0.47	-0.08	1.95	-0.55	0.91

2. Figures



(a)



(b)

Figure S1. SEM images of $\text{Pb}_2\text{GaF}_2(\text{SeO}_3)_2\text{Br}$ (a) and $\text{Pb}_2\text{NbO}_2(\text{SeO}_3)_2\text{Br}$ (b).

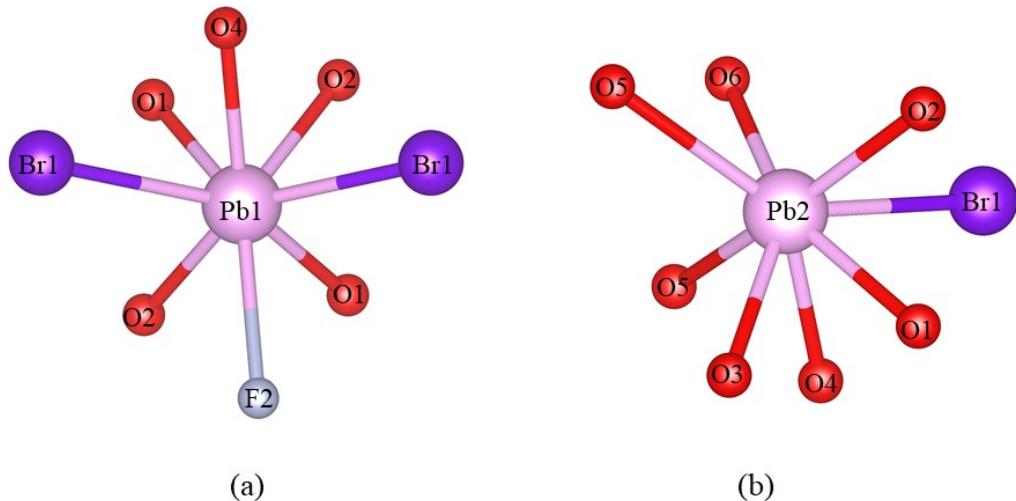


Figure S2. The coordination geometries around the Pb(1) (a) and Pb(2) (b) atoms in $\text{Pb}_2\text{GaF}_2(\text{SeO}_3)_2\text{Br}$.

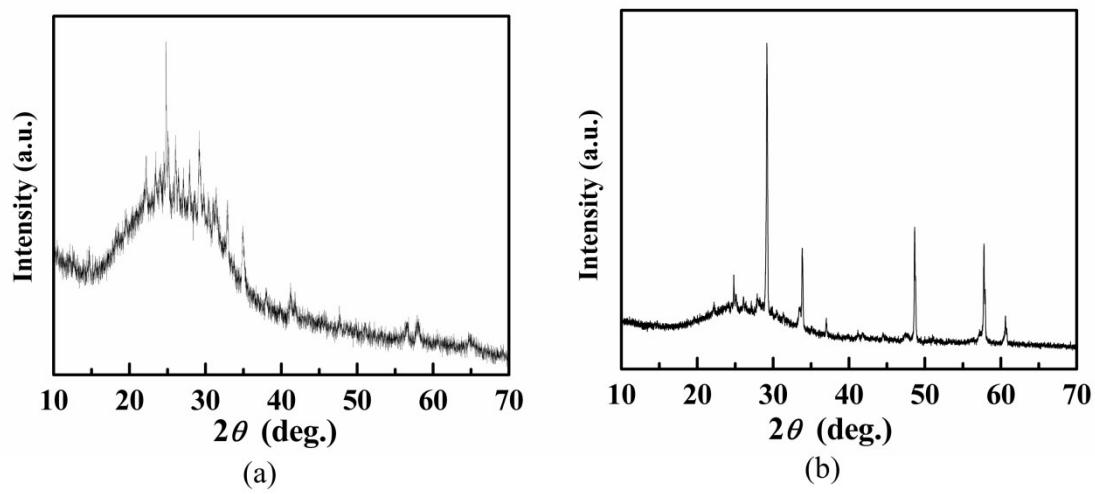


Figure S3. Powder XRD patterns for calcined products of $\text{Pb}_2\text{GaF}_2(\text{SeO}_3)_2\text{Br}$ (a), $\text{Pb}_2\text{NbO}_2(\text{SeO}_3)_2\text{Br}$ (b).

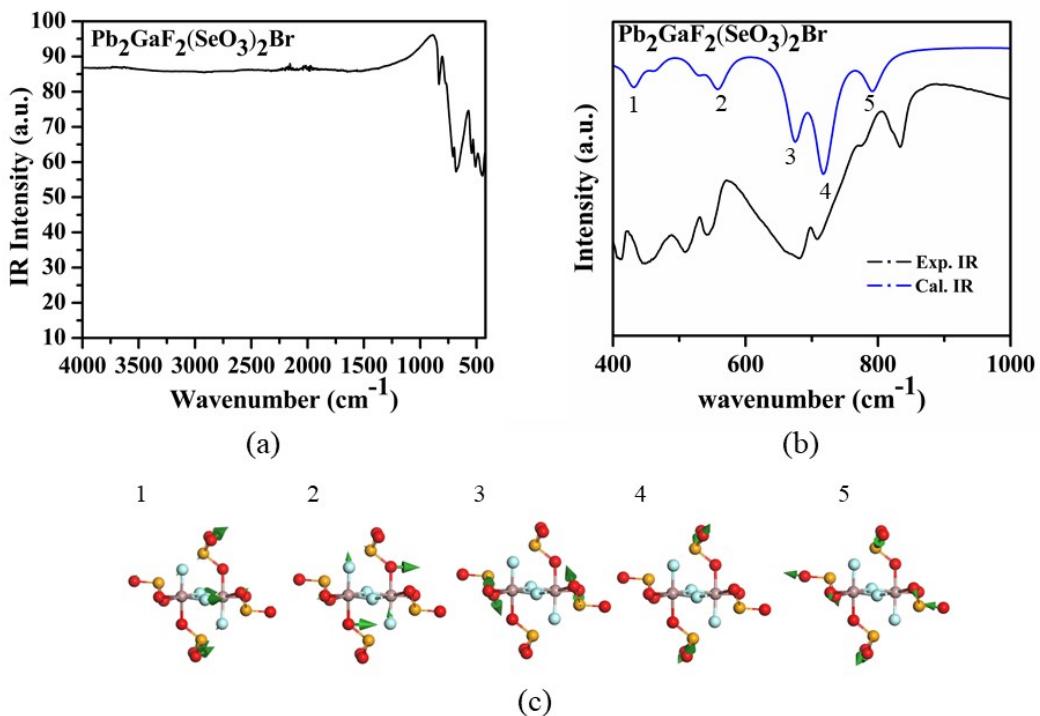


Figure S4. The experimental and simulated IR spectrum of $\text{Pb}_2\text{GaF}_2(\text{SeO}_3)_2\text{Br}$ (a, b); Assignments of peaks and the corresponding vibrational mode in the range of 400-1000 cm^{-1} in the IR spectrum for $\text{Pb}_2\text{GaF}_2(\text{SeO}_3)_2\text{Br}$ (c).

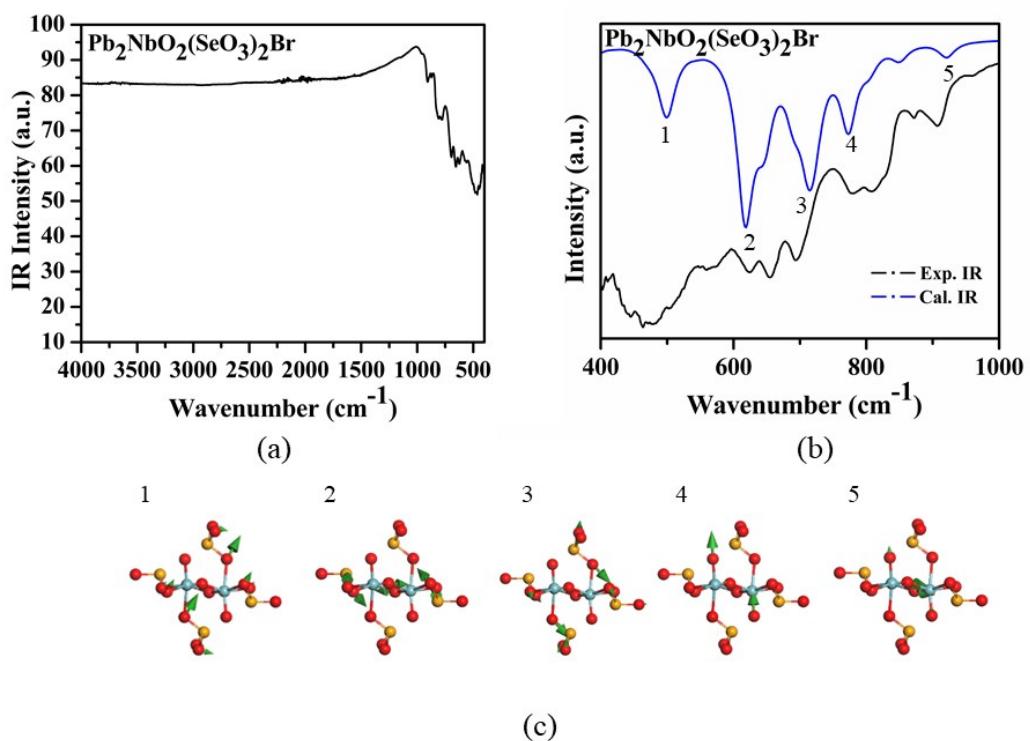


Figure S5. The experimental and simulated IR spectrum of $\text{Pb}_2\text{NbO}_2(\text{SeO}_3)_2\text{Br}$ (a, b); Assignments of peaks and the corresponding vibrational mode in the range of 400-1000 cm^{-1} in the IR spectrum for $\text{Pb}_2\text{NbO}_2(\text{SeO}_3)_2\text{Br}$ (c).

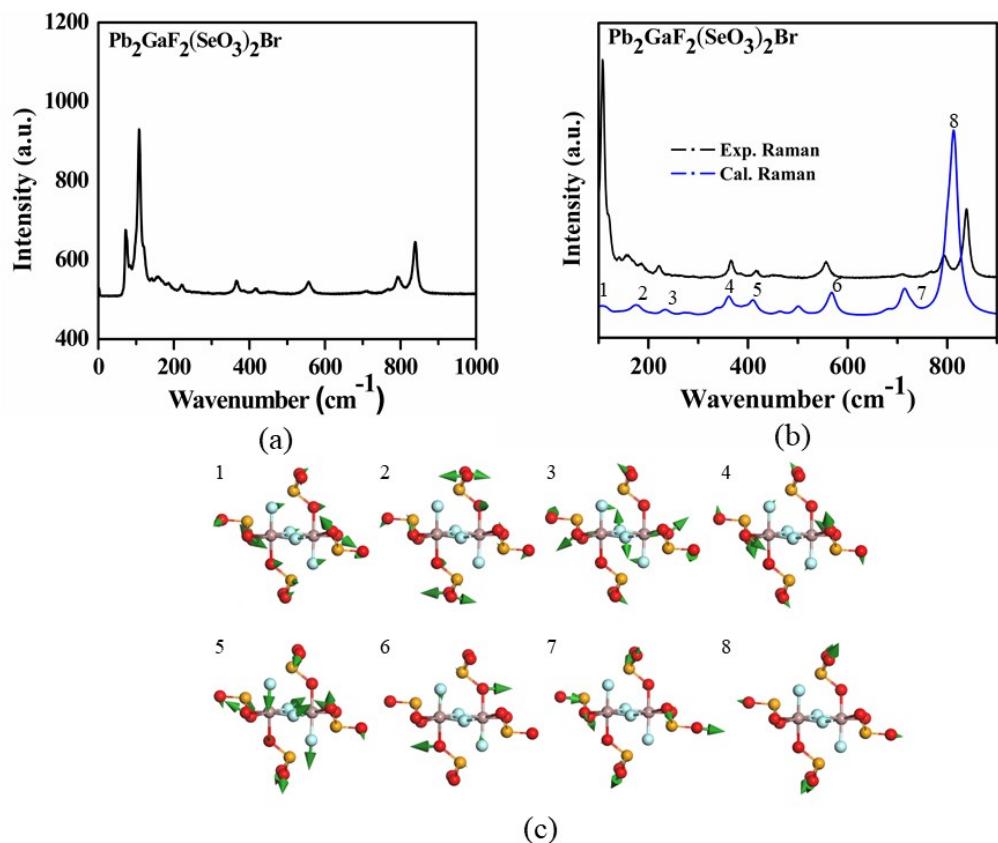


Figure S6. The experimental and simulated Raman spectrum of $\text{Pb}_2\text{GaF}_2(\text{SeO}_3)_2\text{Br}$ (a, b); Assignments of peaks and the corresponding vibrational mode in the range of 100-900 cm^{-1} in the Raman spectrum for $\text{Pb}_2\text{GaF}_2(\text{SeO}_3)_2\text{Br}$ (c).

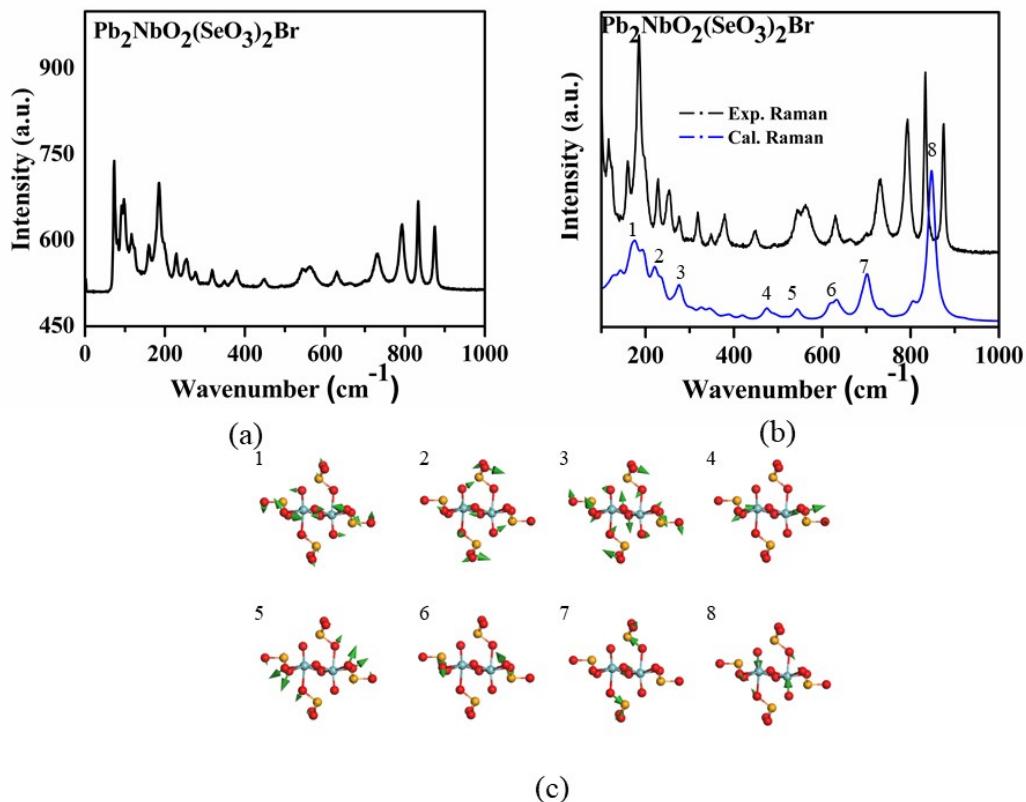


Figure S7. The experimental and simulated Raman spectrum of $\text{Pb}_2\text{NbO}_2(\text{SeO}_3)_2\text{Br}$ (a, b); Assignments of peaks and the corresponding vibrational mode in the range of 100-1000 cm^{-1} in the Raman spectrum for $\text{Pb}_2\text{NbO}_2(\text{SeO}_3)_2\text{Br}$ (c).

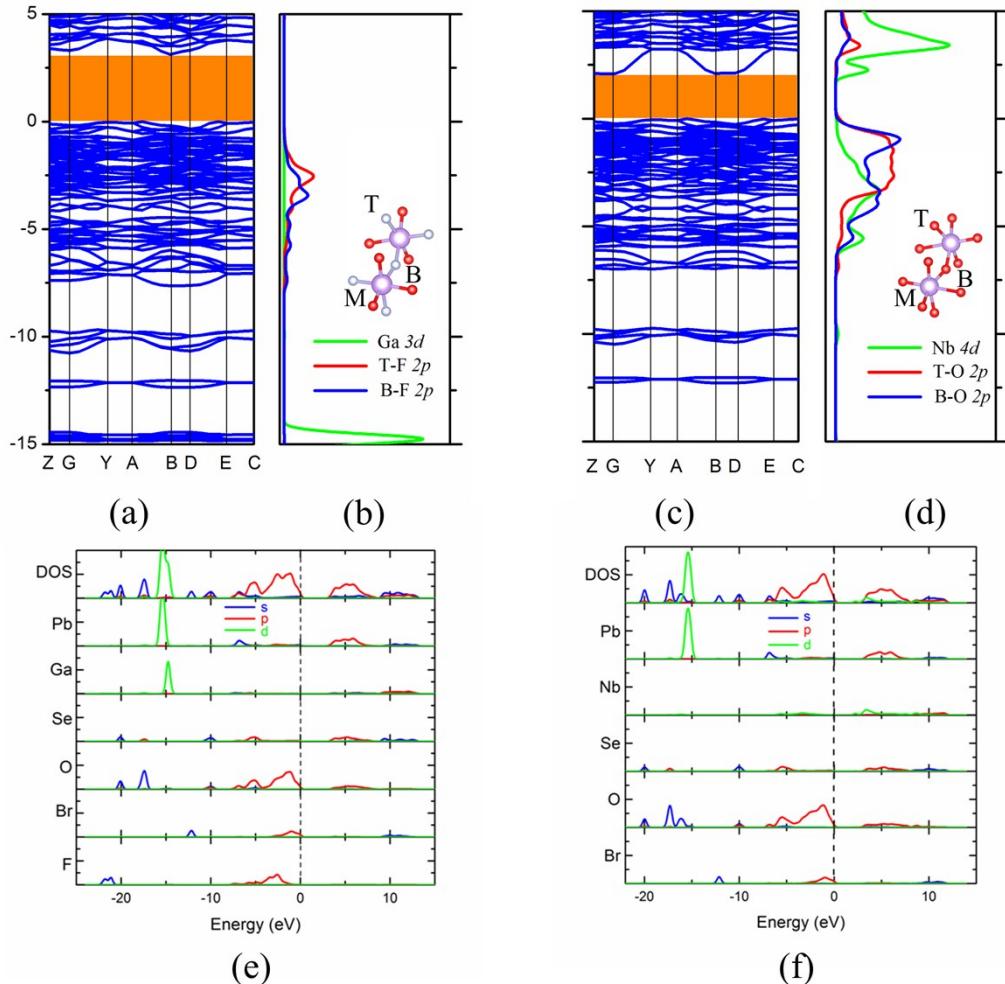


Figure S8. Band structures of (a) $\text{Pb}_2\text{GaF}_2(\text{SeO}_3)_2\text{Br}$ and (c) $\text{Pb}_2\text{NbO}_2(\text{SeO}_3)_2\text{Br}$; PDOS for (b) $(\text{GaO}_3\text{F}_3)^{6-}$ and (d) $(\text{NbO}_6)^{7-}$ groups. Three crucial atomic sites were highlighted, namely, metal cations (M site), terminal O/F (T site), and bridge O/F (B site). T-F and T-O indicate terminal F and O atoms, while B-F and B-O indicate bridged F and O atoms, respectively. The orange regions represent forbidden gap. The PDOS for all constituent ions of $\text{Pb}_2\text{GaF}_2(\text{SeO}_3)_2\text{Br}$ (e) and $\text{Pb}_2\text{NbO}_2(\text{SeO}_3)_2\text{Br}$ (f).

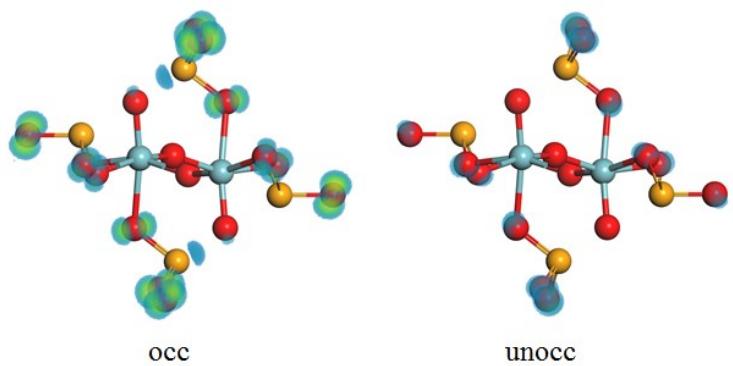


Figure S9. SHG-weighted electron density maps of the occupied and unoccupied states of $[Nb_2O_2(SeO_3)_4]^{6-}$ building blocks of $Pb_2NbO_2(SeO_3)_2Br$. The blue, red, and orange balls represent Nb, O, and Se, respectively.