Supplementary Information

Magnetic dynamics in an open-ring tridysprosium complex

employing mixed ligands

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Crystal Data and Structures

Bond length		Bond angle	
Dy1-010	2.299(4)	O10-Dy1-O11	69.60(15)
Dy1-011	2.342(4)	O10-Dy1-O14	137.77(17)
Dy1-013	2.265(4)	O10-Dy1-N5	78.24(17)
Dy1-014	2.299(5)	O10-Dy1-N6	83.25(16)
Dy1-N5	2.529(6)	O10-Dy1-N7	105.79(18)
Dy1-N6	2.521(6)	O10-Dy1-N8	70.22(18)
Dy1-N7	2.515(6)	O11-Dy1-N5	70.04(17)
Dy1-N8	2.532(6)	O11-Dy1-N6	131.21(18)
Dy2-04	2.476(4)	O11-Dy1-N7	86.48(18)
Dy2-05	2.314(5)	O11-Dy1-N8	120.49(17)
Dy2-06	2.576(5)	013-Dy1-010	149.60(17)
Dy2-07	2.247(5)	013-Dy1-011	81.22(16)
Dy2-08	2.234(5)	013-Dy1-014	72.60(18)
Dy2-09	2.563(5)	O13-Dy1-N5	84.07(19)
Dy2-010	2.311(4)	O13-Dy1-N6	111.39(18)
Dy2-011	2.436(4)	013-Dy1-N7	80.0(2)
Dy3-01	2.279(5)	O13-Dy1-N8	135.52(19)
Dy3-O2	2.293(5)	014-Dy1-011	150.40(17)
Dy3-O4	2.334(5)	O14-Dy1-N5	119.38(18)
Dy3-05	2.295(4)	O14-Dy1-N6	72.91(19)
Dy3-N1	2.504(6)	O14-Dy1-N7	75.74(19)
Dy3-N2	2.547(6)	O14-Dy1-N8	73.01(19)
Dy3-N3	2.546(7)	N5-Dy1-N8	138.09(19)
Dy3-N4	2.513(6)	N6-Dy1-N5	65.09(18)
		N6-Dy1-N8	84.05(19)
		N7-Dy1-N5	153.39(19)
		N7-Dy1-N6	141.0(2)
		N7-Dy1-N8	64.8(2)
		O4-Dy2-O6	114.33(15)
		O4-Dy2-O9	75.62(16)
		05-Dy2-O4	68.10(15)
		O5-Dy2-O6	62.47(16)
		O5-Dy2-O9	77.65(16)
		05-Dy2-011	130.57(15)
		07-Dy2-O4	78.03(16)
		07-Dy2-05	142.46(16)
		07-Dy2-O6	151.82(17)
		07-Dy2-O9	78.73(17)
		07-Dy2-O10	113.92(16)
		07-Dy2-011	86.03(16)
		08-Dy2-O4	83.61(16)
		08-Dy2-05	114.95(17)

Table S1. The selected bond length (Å) and bond angle (°) for 1.

08-Dy2-06	80.85(16)
08-Dy2-07	75.35(17)
O8-Dy2-O9	149.58(17)
08-Dy2-01	0 144.00(16)
08-Dy2-01	1 78.80(16)
O9-Dy2-O6	127.93(16)
O10-Dy2-O	4 131.72(15)
O10-Dy2-O	5 79.75(15)
O10-Dy2-O	6 77.78(15)
O10-Dy2-O	9 62.64(16)
O10-Dy2-O	11 67.79(15)
011-Dy2-0	4 158.77(15)
O11-Dy2-O	6 74.61(15)
011-Dy2-0	9 115.14(16)
O1-Dy3-O2	72.61(19)
O1-Dy3-O4	149.96(17)
O1-Dy3-O5	136.69(18)
O1-Dy3-N1	74.73(19)
O1-Dy3-N2	119.80(18)
O1-Dy3-N3	73.6(2)
O1-Dy3-N4	75.66(19)
O2-Dy3-O4	81.05(17)
O2-Dy3-O5	150.66(17)
O2-Dy3-N1	112.90(18)
O2-Dy3-N2	83.83(18)
O2-Dy3-N3	135.7(2)
O2-Dy3-N4	79.7(2)
O4-Dy3-N1	130.37(19)
O4-Dy3-N2	70.05(17)
O4-Dy3-N3	119.46(17)
O4-Dy3-N4	85.83(18)
O5-Dy3-O4	70.91(16)
O5-Dy3-N1	80.69(18)
O5-Dy3-N2	78.76(17)
O5-Dy3-N3	68.90(18)
O5-Dy3-N4	105.98(19)
N1-Dy3-N2	65.0(2)
N1-Dy3-N3	84.40(19)
N1-Dy3-N4	142.0(2)
N3-Dy3-N2	138.6(2)
N4-Dy3-N2	152.7(2)
N4-Dy3-N3	64.7(2)

Complex	OP-8	HPY-8	HBPY-8	CU-8	SAPR-8	TDD-8	JGBF-8
Dy1	30.10	21.76	15.48	10.15	1.25	2.65	14.57
Dy2	29.25	23.69	16.68	10.95	1.22	3.53	15.73
Dy3	29.84	21.74	15.44	9.84	1.19	2.85	14.58
	JETBPY-8	JBTPR-8	BTPR-8	JSD-8	TT-8	ETBPY-8	
Dy1	JETBPY-8 23.44	JBTPR-8 2.95	BTPR-8 2.40	JSD-8 4.80	TT-8 10.69	ETBPY-8 19.78	
Dy1 Dy2	JETBPY-8 23.44 23.10	JBTPR-8 2.95 3.16	BTPR-8 2.40 3.13	JSD-8 4.80 4.91	TT-8 10.69 11.76	ETBPY-8 19.78 19.66	

 Table S2. Continuous shape measures calculations (CShM)^{1,2} for rare-earth ions in 1 and 2.

OP-8 = Octagon (D_{8h}); HPY-8 = Heptagonal pyramid (C_{7v}); HBPY-8 = Hexagonal bipyramid (D_{6h}); CU-8 = Cube (O_h); SAPR-8 = Square antiprism (D_{4d}); TDD-8 = Triangular dodecahedron (D_{2d}); JGBF-8 = J Johnson gyrobifastigium J26 (D_{2d}). JETBPY-8 = Johnson elongated triangular bipyramid J14 (D_{3h}); JBTPR-8 = Biaugmented trigonal prism J50 (C_{2v}); BTPR-8 = Biaugmented trigonal prism (C_{2v}); JSD-8 = Snub diphenoid J84 (D_{2d}); TT-8 = Triakis tetrahedron (T_d) and ETBPY-8 = Elongated trigonal bipyramid (D_{3h}).

[1] Alvarez, S.; Alemany, P.; Casanova, D.; Cirera, J.; Llunell, M.; Avnir, D. Shape maps and polyhedral interconversion paths in transition metal chemistry. *Coord. Chem. Rev.* **2005**, 249 (17), 1693-1708.

[2] Casanova, D.; Llunell, M.; Alemany, P.; Alvarez, S. The Rich Stereochemistry of Eight-Vertex Polyhedra: A Continuous Shape Measures Study. *Chem. Eur. J.* 2005, 11 (5), 1479-1494.



Figure S1. The experimental and simulated powder X-ray diffraction (PXRD) patterns for 1.



Figure S2. Thermogravimetric analysis (TGA) curve at a heating rate of 10 K min⁻¹. Exptl. (200 °C): 5.7%, calc: 5.7%.

Magnetic Characterization



Figure S3. Frequency-dependence of the in-phase product (χ'_{M}) and out-of-phase (χ''_{M}) at 4 K under 0–2 kOe dc field for **1**.

Т/К	χ_0 / cm ³ mol ⁻¹	χ_{∞} / cm ³ mol ⁻¹	τ / s	α	R^2
2	23.29088	0.90336	1.29938	0.48437	0.99665
3	14.64284	0.75402	0.38529	0.43399	0.99755
4	10.50752	0.64213	0.13306	0.37506	0.9984
5	8.2069	0.55028	0.05086	0.3158	0.99897
6	6.79868	0.46771	0.02252	0.2763	0.99939
7	5.92623	0.37295	0.01175	0.26609	0.99967
8	5.15296	0.32737	0.00647	0.24298	0.99974
9	4.56045	0.29189	0.00386	0.22611	0.99976
10	4.09587	0.26518	0.00245	0.21378	0.99968
11	3.71955	0.25164	0.00163	0.20177	0.99962
12	3.37491	0.23164	0.00106	0.19501	0.99962
13	3.12871	0.2201	7.49304E-4	0.19136	0.9996
14	2.90814	0.21222	5.31377E-4	0.1857	0.99965
15	2.71485	0.23359	3.86419E-4	0.17616	0.99965
16	2.54256	0.24705	2.79308E-4	0.1669	0.99967
17	2.39545	0.25919	2.01906E-4	0.16437	0.99974
18	2.25538	0.26251	1.44327E-4	0.14923	0.99979
19	2.13429	0.28927	1.03925E-4	0.13967	0.99981
20	2.03082	0.23448	7.16994E-5	0.13189	0.99988

 Table S3. The best-fit parameters obtained from the generalized Debye model at zero dc field.

Т/К	χ_0 / cm ³ mol ⁻¹	χ_{∞} / cm ³ mol ⁻¹	τ/s	α	R^2
2	18.5201	0.28902	1.60361	0.39095	0.99884
3	12.03296	0.25634	0.47415	0.3576	0.99931
4	8.91401	0.22728	0.1688	0.31792	0.99948
5	7.04549	0.20654	0.06642	0.27843	0.99959
6	5.85907	0.18309	0.03015	0.25515	0.99952
7	5.12926	0.13518	0.01598	0.2605	0.99974
8	4.49526	0.12279	0.00893	0.24632	0.9994
9	3.97982	0.1047	0.00529	0.24211	0.99962
10	3.56376	0.10168	0.0033	0.23165	0.99955
11	3.23829	0.1069	0.00217	0.21949	0.99941
12	2.93218	0.10022	0.00139	0.2137	0.99921
13	2.70557	0.11666	9.58452E-4	0.19933	0.99928
14	2.52204	0.13252	6.77485E-4	0.18919	0.99926
15	2.35352	0.13169	4.71963E-4	0.1858	0.99933
16	2.20192	0.16378	3.37349E-4	0.16895	0.99921
17	2.06659	0.15552	2.33524E-4	0.1685	0.99942
18	1.95322	0.19121	1.69801E-4	0.15189	0.99941
19	1.84871	0.17935	1.16792E-4	0.14408	0.99972
20	1.74904	0.21125	8.40138E-5	0.1206	0.99957

 Table S4. The best-fit parameters obtained from the generalized Debye model at 500 Oe dc field.



Figure S4. The relaxation time versus *T* plots for **1**. The solid lines are the best fit with equation $\tau^{-1} = \tau_0^{-1} \exp(-U_{\text{eff}}/k_{\text{B}}T) + CT^n + \tau_{\text{QTM}}^{-1}$ and gives a share $U_{\text{eff}} = 114(7) \text{ cm}^{-1}$. The dashed line represents for the tendency of the Orbach-type process.

Ab initio calculation

Computational Details. Ab initio calculations were performed with OpenMOLCAS version 18.09^{1} and are of the CASSCF/RASSI type. The Cholesky decomposition threshold was set to 1×10^{-8} to save disk space. An entire molecule was included, and the coordinates of atoms were extracted from the experimentally determined crystal structure. The adjacent Dy(III) sites were computationally replaced by the diamagnetic Y(III). ANO-RCC basis set approximations have been employed (see Table S4).²⁻³ Active space of the CASSCF method included nine electrons in seven 4*f* orbitals of Dy(III). 21 sextets for Dy(III) were optimized in state-averaged calculations and then mixed by spin–orbit coupling using RASSI approach.⁴ The *g*-tensors, energy spectra, main magnetic axis as well as the magnetizations were obtained by SINGLE_ANISO routine⁵.

[1] Aquilante, F.; Autschbach, J.; Carlson, R. K.; Chibotaru, L. F.; Delcey, M. G.; De Vico, L.; Fdez. Galván, I.; Ferré, N.; Frutos, L. M.; Gagliardi, L.; Garavelli, M.; Giussani, A.; Hoyer, C. E.; Li Manni, G.; Lischka, H.; Ma, D.; Malmqvist, P. Å.; Müller, T.; Nenov, A.; Olivucci, M.; Pedersen, T. B.; Peng, D.; Plasser, F.; Pritchard, B.; Reiher, M.; Rivalta, I.; Schapiro, I.; Segarra-Martí, J.; Stenrup, M.; Truhlar, D. G.; Ungur, L.; Valentini, A.; Vancoillie, S.; Veryazov, V.; Vysotskiy, V. P.; Weingart, O.; Zapata, F.; Lindh, R. Molcas 8: New capabilities for multiconfigurational quantum chemical calculations across the periodic table. *J. Comput. Chem.* 2016, 37 (5), 506-541.

[2] Roos, B. O.; Lindh, R.; Malmqvist, P.-Å.; Veryazov, V.; Widmark, P.-O.; Borin, A. C. New Relativistic Atomic Natural Orbital Basis Sets for Lanthanide Atoms with Applications to the Ce Diatom and LuF3. *J. Phys. Chem. A* 2008, 112 (45), 11431-11435.
[3] Roos, B. O.; Lindh, R.; Malmqvist, P.-Å.; Veryazov, V.; Widmark, P.-O. Main group atoms and dimers studied with a new relativistic ANO basis set. *J. Phys. Chem. A* 2004, 108 (15), 2851-2858.

[4] Malmqvist, P. Å.; Roos, B. O.; Schimmelpfennig, B. The restricted active space (RAS) state interaction approach with spin– orbit coupling. *Chem. Phys. Lett.* 2002, 357 (3), 230-240.

[5] Chibotaru, L. F.; Ungur, L. Ab initio calculation of anisotropic magnetic properties of complexes. I. Unique definition of pseudospin Hamiltonians and their derivation. *J. Chem. Phys.* 2012, 137 (6), 064112.

	1
	Dy.ANO-RCC-VTZP
	Y.ANO-RCC-VDZP
ANO BCC basis sats	O.ANO-RCC-VDZ
ANO-RCC Dasis sets	N.ANO-RCC-VDZ
	C.ANO-RCC-VDZ
	H.ANO-RCC-MB

Table S5. The employed ANO-RCC basis sets.

	E/cm^{-1}	g _x	g_{y}	gz
	0	0.0083	0.0165	19.608
	133.4008	0.1560	0.2081	16.664
	220.7048	0.6559	0.8775	13.117
Dv1	325.8642	2.2202	3.7194	9.8892
Dyl	412.2692	2.6040	4.5305	11.615
	463.8032	0.0826	2.7287	12.830
	549.0074	0.6856	1.8792	16.256
	618.3796	0.3139	1.0819	18.449
	0	0.0099	0.0223	19.718
	122.1173	0.073	0.2369	16.874
	186.187	1.2241	2.9802	13.428
Dv2	221.292	1.8994	5.3407	10.249
Dyz	264.3722	1.1340	4.4627	10.897
	326.0537	1.9312	2.0886	14.726
	445.2158	0.1473	0.1628	17.081
	652.4321	0.0030	0.0040	19.260
	0	0.0079	0.0146	19.642
	125.0711	0.1299	0.1614	16.816
	214.6793	0.4435	0.6240	13.255
D. 2	315.2309	1.9627	3.1608	10.152
Dy3	389.1664	3.2829	4.1635	13.495
	451.2692	1.1047	2.3024	12.228
	526.8271	0.9981	2.6851	16.147
	617.5835	0.2344	0.7171	18.988

Table S6. The energy spectra and *g* tensors for the eight Kramers doublets of the ${}^{6}H_{15/2}$ multiplets for Dy(III) ions in **1**.