Supporting Information

Synthesis of Se-HPCF composite via a liquid-solution route and its stable cycling performance in Li-Se batteries

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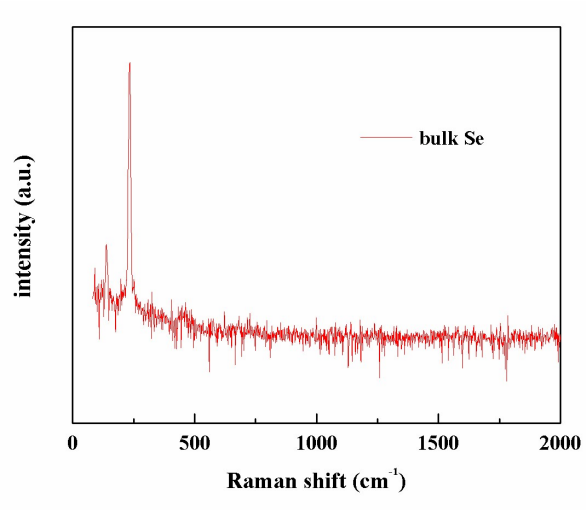

Fig. S1 Raman spectrum of Se crystals.

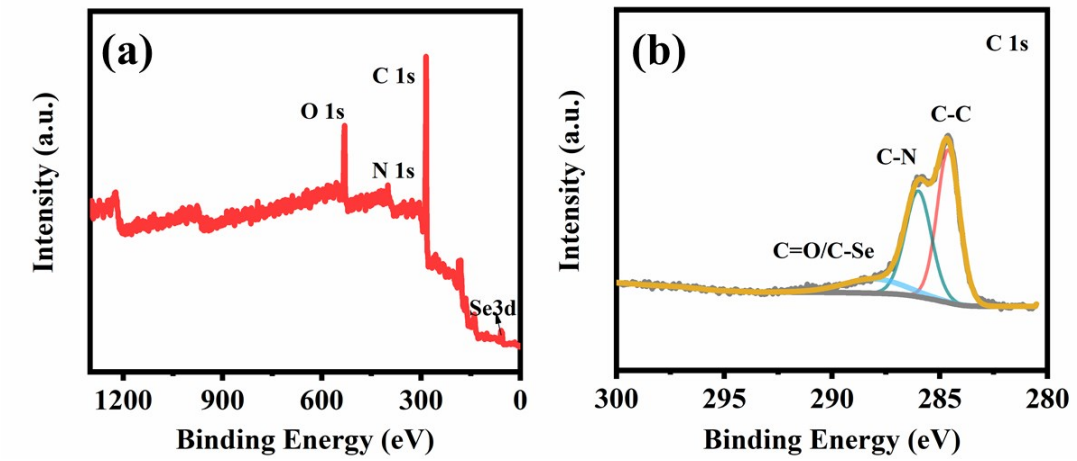

Fig. S2 High resolution XPS spectra of (a) Survey, (b) C 1s for Se-HPCF composite.

Table S1 Electrochemical performance comparison of Se-C composites reported in the literatures with present work.

Electrode Materials	Current density	Cycling capacity (mAh g^{-1})	Cycle numbers(n)	Rate capability (mAh g^{-1})	Ref.
Se/porous carbon nanofibers	0.74 C	516	900	637/0.147 C 306/5.9 C	[S1]
Se/CMCs	0.2 C 2C	425.2 166.3	100 460	218.1/5 C	[S2]
Se/CNSs	0.1C 0.5C	600 376	100 1000	700/0.1 C ~390/10 C	[S3]
Se/N-CSHPC-II	0.2 C 0.5 C	555 462	150 200	438/1 C	[S4]
Se/C/aniline	134 mA g^{-1}	462	200	600/67.2 mA g^{-1} 319/1.34 A g^{-1}	[S5]
Se/CNTs	5C	390	500	592/0.5C 390/5 C	[S6]
Se/PBC	0.2 C 0.5C	509 376	200 500	518/0.2C 323/5 C	[S7]
Se-HPCF	**0.2 C** **5C**	**533** **202**	**50** **2000**	**659/0.2C** **351/5 C**	**This work**

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