

Variable dimensionality in ‘hollow’ hybrid tin iodide perovskites

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Supplementary Information

Synthesis

Tin powder (~325 mesh, 99.8% metals basis), hydroiodic acid (HI, 57%, w/w aqueous solution stabilised with 1.5% HPA), hypophosphorous acid (H₃PO₂ (HPA), 50% w/w aqueous solution) and 1,2,4-triazole (C₂H₃N₃, 99%) were purchased from Alfa Aesar. All chemicals were directly used without further purification.

Both (TzH)₁₁(H₃PO₂)Sn₆I₂₃ and (TzH)_xSn₂I₇ were crystallised by a slow evaporation method.

For (TzH)₁₁(H₃PO₂)Sn₆I₂₃ [C₂N₃H₄]₁₁[H₃PO₂]Sn₆I₂₃: Tin powder (237 mg, 2 mmol) was dissolved in conc. HI (5 ml) and HPA (5 ml) with moderate heating. Once fully dissolved, 1,2,4-triazole (138 mg, 2 mmol) was added and the solution allowed to cool. After an hour orange/red crystals were obtained. These were filtered and washed with diethyl ether.

For (TzH)₃Sn₂I₇ [C₂N₃H₄]₃Sn₂I₇: The reaction mixture used in the synthesis of (TzH)₁₁(H₃PO₂)Sn₆I₂₃ was left for 2 months. Orange/red crystals were obtained. These were filtered and washed with diethyl ether. Elemental analysis: (Anal. Calc. (%)) for (TzH)₃Sn₂I₇: C 5.39; N 9.44; H 0.91. Found: C 5.48; N 9.51; H 0.86).

The synthesis of both (TzH)₁₁(H₃PO₂)Sn₆I₂₃ and (TzH)₃Sn₂I₇ was carried out in air. HPA was used in the reaction to prevent oxidation of Sn²⁺ to Sn⁴⁺ while the crystals remained in solution. After filtration the crystals were stored under argon. In air both (TzH)₁₁(H₃PO₂)Sn₆I₂₃ and (TzH)₃Sn₂I₇ oxidise rapidly with a clear colour change and evolution of I₂.

Characterisation

Single Crystal data were collected at 93 K and 173 K for (TzH)₁₁(H₃PO₂)Sn₆I₂₃ and (TzH)₃Sn₂I₇, respectively. Data were collected on a Rigaku Pilatus 200 K diffractometer using Mo-K α radiation. Data were collected using CrystalClear (Rigaku).¹ Structures were solved by direct methods and refined using SHELX-2014² incorporated in the WINGX program.³ Absorption corrections were performed semi-empirically from equivalent reflections on the basis of multi-scans.

In (TzH)₁₁(H₃PO₂)Sn₆I₂₃ non-H atoms were refined anisotropically and hydrogen atoms were treated as riding atoms. Due to the strong hydrogen bonding between H₃PO₂ and two TzH⁺ cations these are both known to exhibit disorder over two positions around the pseudo two-fold axis and C/N positions have been assigned accordingly. For all other TzH⁺ cations only the N with the lone pair could be assigned unambiguously and therefore all other C/N positions have been assigned as 50:50 occupancy to maintain the correct compositional formula. This model allows full anisotropic refinement of all TzH⁺ moieties.

In $(\text{TzH})_3\text{Sn}_2\text{I}_7$, only the tin iodide framework has been refined. Models with partial assignment of TzH^+ fragments were trialled, but no complete and satisfactory model could be obtained. The CHN analysis supports the formula proposed, with all TzH^+ moieties intact. Moreover, although some degradation of linear amines has been observed under similar reaction conditions, there is no evidence to suggest this would also take place with the cyclic, conjugated TzH^+ amine. Hence, we are confident of the formula proposed.

Preliminary UV-vis absorption was carried out on $(\text{TzH})_{11}(\text{H}_3\text{PO}_2)\text{Sn}_6\text{I}_{23}$ using a JASCO-V550 ultraviolet-visible spectrophotometer with the wavelength range at 200 nm to 900 nm (JASCO Corporation, Essex, UK).

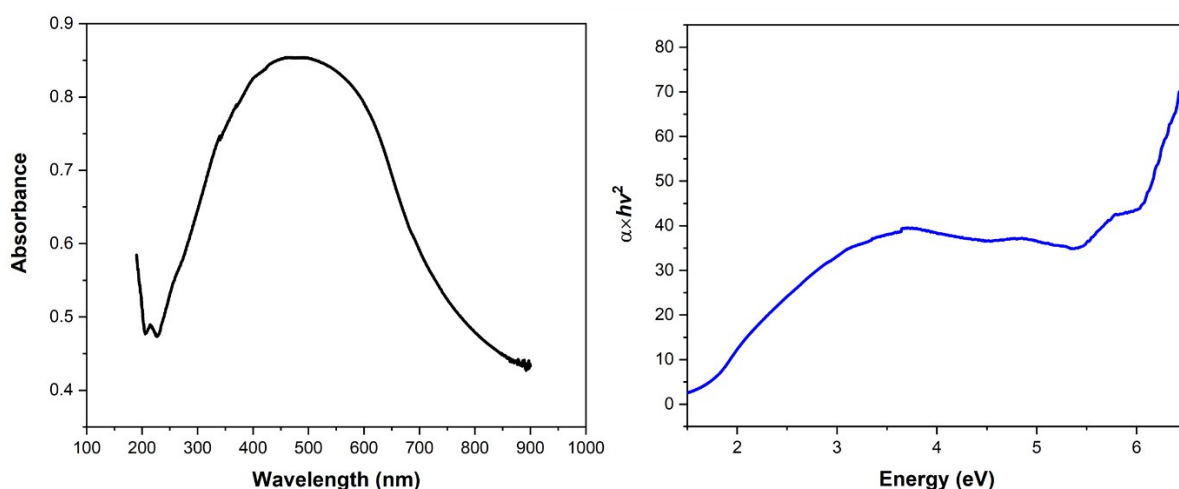


Figure S1. Preliminary UV-vis absorption spectra (left) and calculated Tauc plot (right) for $(\text{TzH})_{11}(\text{H}_3\text{PO}_2)\text{Sn}_6\text{I}_{23}$ collected at room temperature.

The bond length distortion of the SnI_6 octahedra of both compositions was calculated using eq. 1,⁴ where d is the average Sn-I bond distance and d_n are the six individual bond distances. The bond angle variance of each octahedron from the ideal 90° of an undistorted structure was calculated using eq. 2,⁵ where θ_i is the individual I-Sn-I angle.

$$\Delta d = \left(\frac{1}{6}\right) \sum \left[\frac{d_n - d}{d} \right]^2 \quad (1)$$

$$\sigma^2 = \sum_{i=1}^{12} \frac{(\theta_i - 90)^2}{11} \quad (2)$$

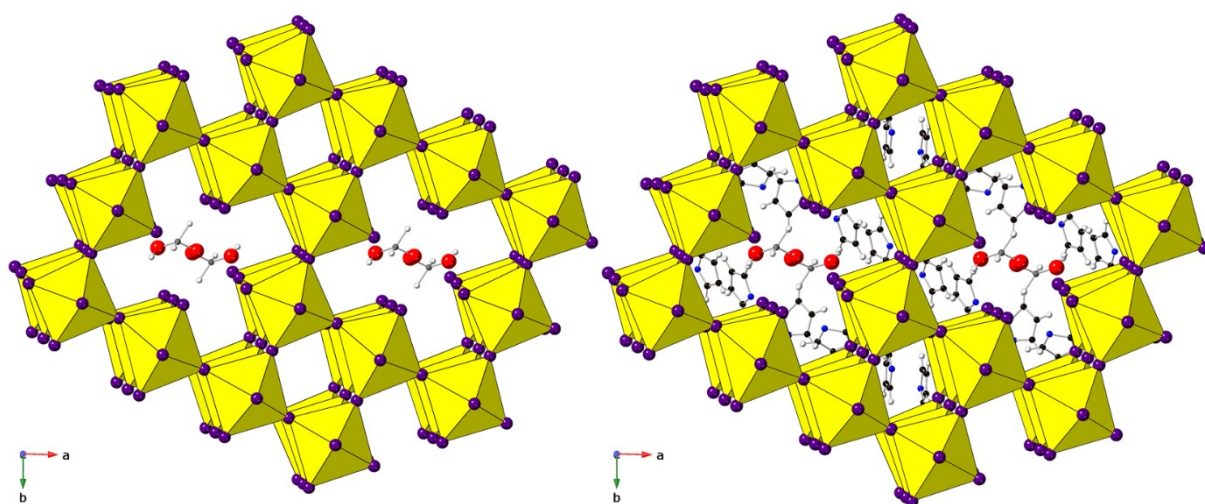


Figure S2. Crystal structure of $(\text{TzH})_{11}(\text{H}_3\text{PO}_2)\text{Sn}_6\text{I}_{23}$ viewed along the c -axis highlighting the position of H_3PO_2 (left) and the surrounding amines in the inter-layer site (right). H_3PO_2 is positioned between triple perovskite-like blocks in the inter-layer site and below the ‘hollow’ channel caused by the B-site deficiency.

Table S1 Selected bond lengths (Å) and angles (°) derived from single crystal data of (TzH)₁₁(H₃PO₂)Sn₆I₂₃ collected at 93 K. Bridging and terminal iodine atoms are denoted with *b* or *t*, respectively.

Sn-I						
	Sn1-I4 ^t	2.925(1)	Sn2-I7 ^t	2.981(1)	Sn3-I14 ^t	2.991(1)
	Sn1-I11 ^t	3.069(1)	Sn2-I11 ^b	3.026(1)	Sn3-I15 ^b	3.024(1)
	Sn1-I3 ^b	3.120(1)	Sn2-I10 ^t	3.115(1)	Sn3-I12 ^t	3.101(1)
	Sn1-I5 ^b	3.224(1)	Sn2-I5 ^b	3.207(1)	Sn3-I13 ^b	3.222(1)
	Sn1-I2 ^b	3.271(1)	Sn2-I9 ^b	3.312(1)	Sn3-I11 ^b	3.300(1)
	Sn1-I6 ^b	3.449(1)	Sn2-I8 ^b	3.365(1)	Sn3-I3 ^b	3.353(1)
Sn-I						
	Sn4-I18 ^t	2.887(1)	Sn5-I20 ^t	2.905(1)	Sn6-I24 ^t	2.952(1)
	Sn4-I16 ^t	3.097(1)	Sn5-I21 ^b	3.132(1)	Sn6-I22 ^b	3.047(1)
	Sn4-I6 ^b	3.170(1)	Sn5-I8 ^b	3.143(1)	Sn6-I2 ^b	3.081(1)
	Sn4-I9 ^b	3.223(1)	Sn5-I19 ^b	3.215(1)	Sn6-I13 ^b	3.191(1)
	Sn4-I17 ^b	3.267(1)	Sn5-I17 ^b	3.224(1)	Sn6-I23 ^b	3.264(1)
	Sn4-I15 ^b	3.619(1)	Sn5-I22 ^b	3.522(1)	Sn6-I21 ^b	3.398(1)
I-Sn-I						
I-Sn-I	Sn1		Sn2		Sn3	
	84.92(4)		83.49(4)		82.41(4)	
	85.10(4)		88.03(3)		86.10(4)	
	85.68(4)		88.04(4)		87.82(4)	
	87.11(4)		89.28(4)		88.84(4)	
	88.24(4)		90.07(4)		88.90(4)	
	89.54(4)		90.47(4)		89.25(4)	
	91.01(4)		90.58(4)		90.61(4)	
	91.54(4)		90.84(4)		92.05(4)	
	92.28(4)		90.88(4)		92.10(4)	
	92.43(4)		91.81(4)		92.64(4)	
	95.49(4)		92.73(4)		92.68(4)	
	95.77(4)		94.14(4)		96.19(4)	
I-Sn-I						
I-Sn-I	Sn4		Sn5		Sn6	
	81.85(4)		79.17(4)		85.56(4)	
	83.65(4)		84.16(4)		86.06(4)	
	86.27(4)		87.83(4)		87.78(4)	
	88.04(4)		87.99(4)		88.33(4)	
	88.13(4)		90.06(4)		89.21(4)	
	88.16(4)		90.48(4)		89.52(4)	
	90.45(4)		90.64(4)		89.63(4)	
	91.17(4)		91.21(4)		90.35(4)	
	92.92(4)		91.56(4)		91.75(4)	
	94.26(4)		91.81(4)		92.66(4)	
	97.37(4)		92.77(4)		93.97(4)	
	97.66(4)		102.66(4)		95.10(4)	
Sn-I-Sn						

Sn1-I5-Sn2	170.57(4)	Sn1-I3-Sn3	174.29(4)	Sn1-I6-Sn4	166.26(4)
Sn1-I2-Sn6	173.02(4)	Sn2-I11-Sn3	171.31(4)	Sn2-I9-Sn4	178.32(4)
Sn2-I8-Sn5	173.31(4)	Sn3-I15-Sn4	174.37(4)	Sn3-I13-Sn6	171.49(4)
Sn4-I17-Sn5	165.02(4)	Sn5-I19-Sn5	180	Sn5-I21-Sn6	177.50(4)
Sn5-I22-Sn6	175.30(4)	Sn6-I23-Sn6	180		

BVS					
Sn1	2.10	Sn2	2.09	Sn3	2.10
Sn4	2.04	Sn5	2.06	Sn6	2.17
I1 ^t	0.43	I2 ^b	0.66	I3 ^b	0.57
I4 ^t	0.63	I5 ^b	0.57	I6 ^b	0.47
I7 ^t	0.54	I8 ^b	0.54	I9 ^b	0.50
I10 ^t	0.38	I11 ^b	0.70	I12 ^t	0.39
I13 ^b	0.59	I14 ^t	0.52	I15 ^b	0.58
I16 ^t	0.39	I17 ^b	0.53	I18 ^t	0.69
I19 ^b	0.57	I20 ^t	0.66	I21 ^b	0.53
I22 ^b	0.58	I23 ^b	0.50	I24 ^t	0.58

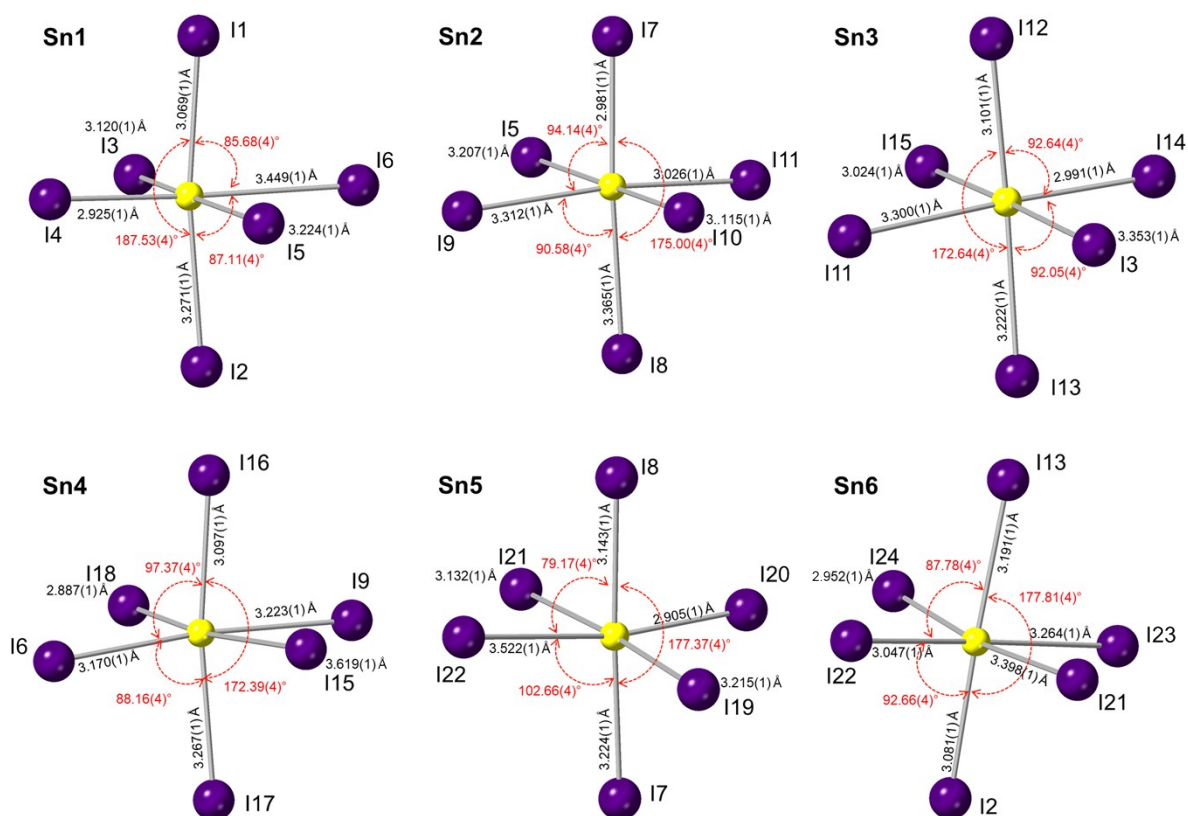


Figure S3. Ball and stick representations of the octahedra present in $(\text{TzH})_{11}(\text{H}_3\text{PO}_2)\text{Sn}_6\text{I}_{23}$ at 93K highlighting the distortion in both bond lengths and bond angles.

Table S2 Selected bond lengths (Å) and angles (°) derived from single crystal data of for (TzH)_xSn₂I₇ collected at 173 K.

Sn1		Sn2	
Sn-I			
Sn1-I1	2.928(4)	Sn2-I5	2.961(3)
Sn1-I3	3.096(4)	Sn2-I6	3.091(3)
Sn1-I2	3.150(4)	Sn2-I4	3.201(5)
Sn1-I2	3.255(4)	Sn2-I4	3.207(5)
Sn1-I8	3.302(4)	Sn2-I7	3.283(4)
Sn1-I6	3.499(4)	Sn2-I3	3.403(4)
I-Sn-I			
	84.43(10)		84.94(10)
	85.84(10)		86.42(10)
	87.58(10)		87.32(10)
	89.57(10)		87.44(10)
	90.16(10)		89.71(8)
	90.19(10)		90.99(10)
	90.86(10)		91.13(10)
	90.95(10)		91.39(10)
	91.92(8)		91.52(9)
	92.45(10)		92.09(9)
	92.98(10)		92.54(8)
	93.04(10)		94.17(10)
Sn-I-Sn			
Sn1-I2-Sn1	177.35(10)	Sn1-I8-Sn1	180
Sn2-I4-Sn2	175.63(10)	Sn2-I7-Sn2	180
Sn1-I3-Sn2	176.71(10)	Sn1-I6-Sn2	174.40(10)
BVS			
Sn1	1.97	Sn2	1.97
I1	0.62	I2	0.60
I3	0.57	I4	0.59
I5	0.57	I6	0.53
I7	0.48	I8	0.45

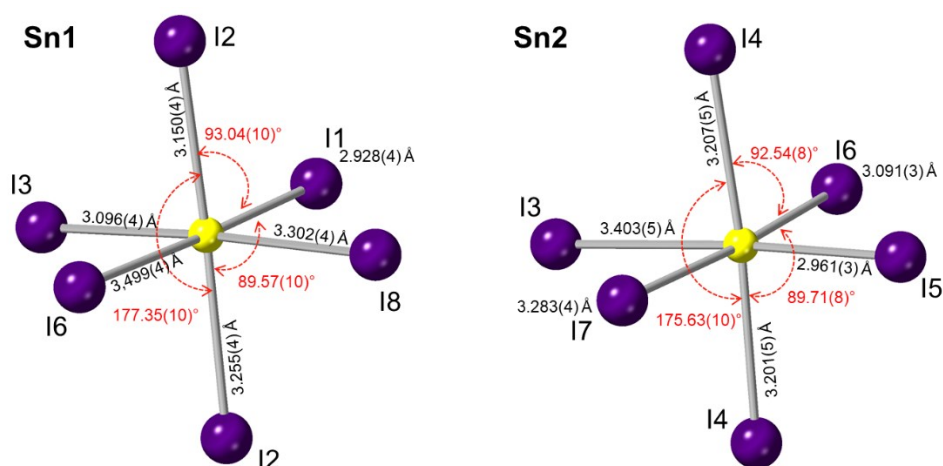


Figure S4. Ball and stick representations of the octahedra present in $(\text{TzH})_x\text{Sn}_2\text{I}_7$ at 173K highlighting the distortion in both bond lengths and bond angles.

References

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