

Electronic and steric effects of Platinum(II) di-yne and poly-yne substituents on the photo-switching behaviour of stilbene: Experimental and theoretical insights

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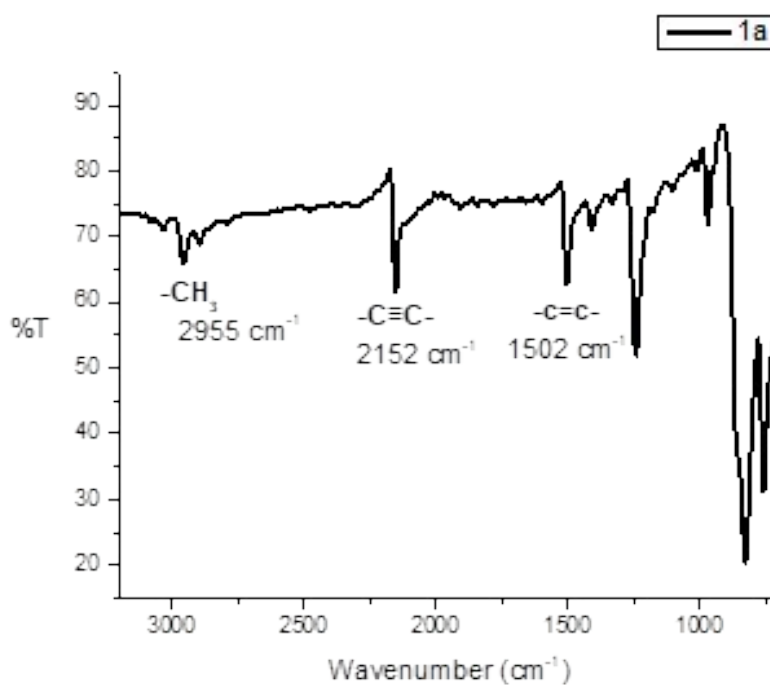


Figure S1. FTIR spectrum of **1a**.

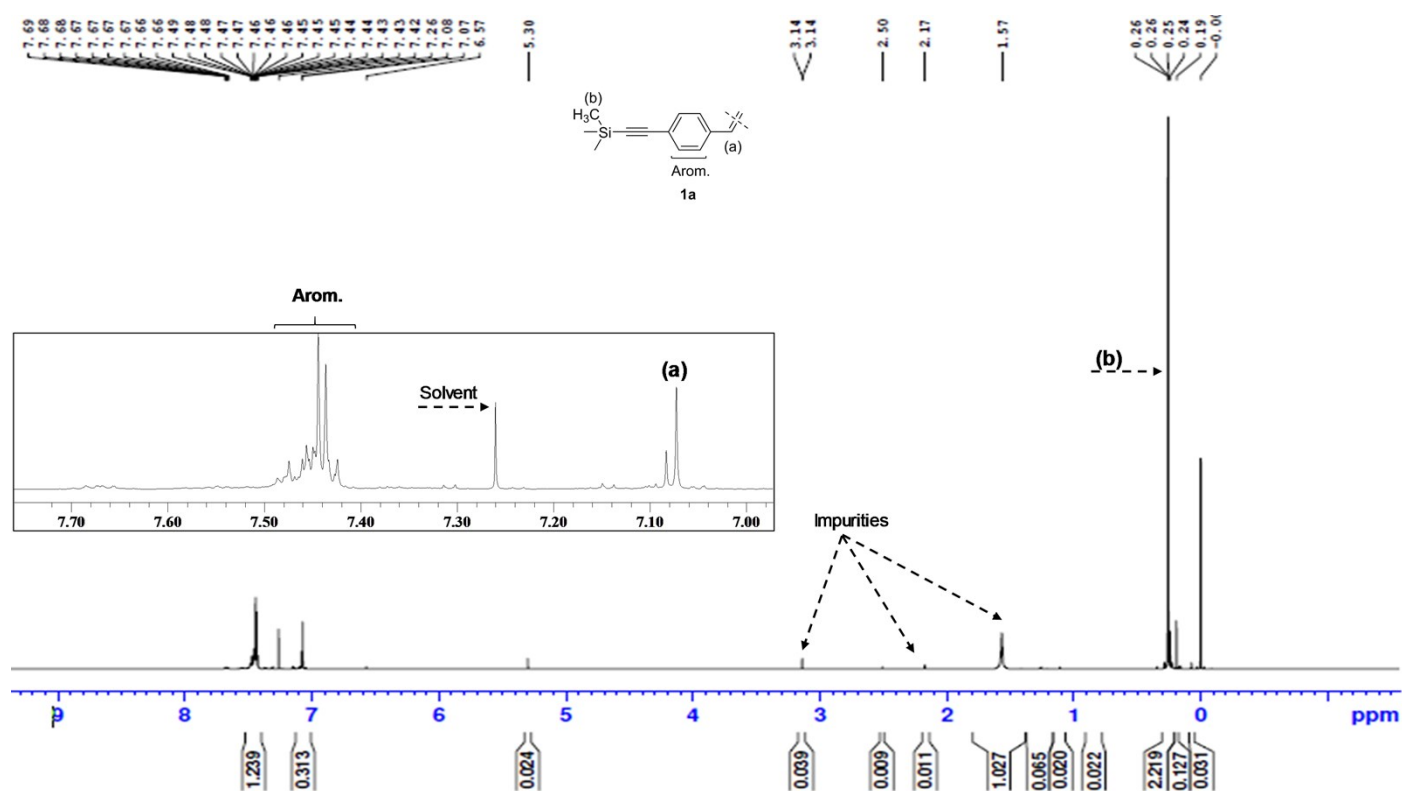


Figure S2. ^1H -NMR spectrum of **1a**.

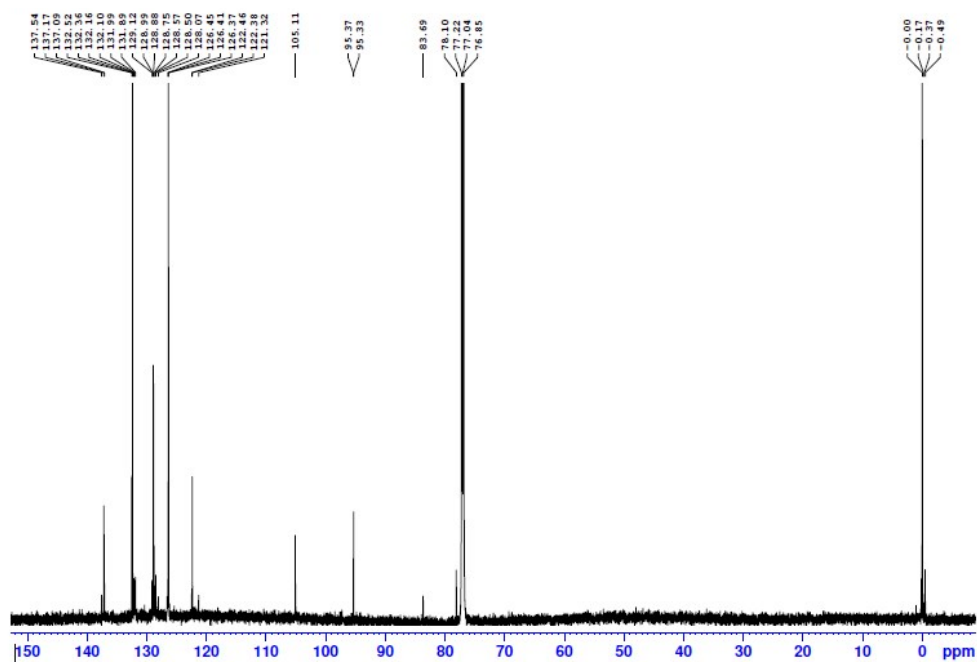


Figure S3. ^{13}C -NMR spectrum of **1a**.

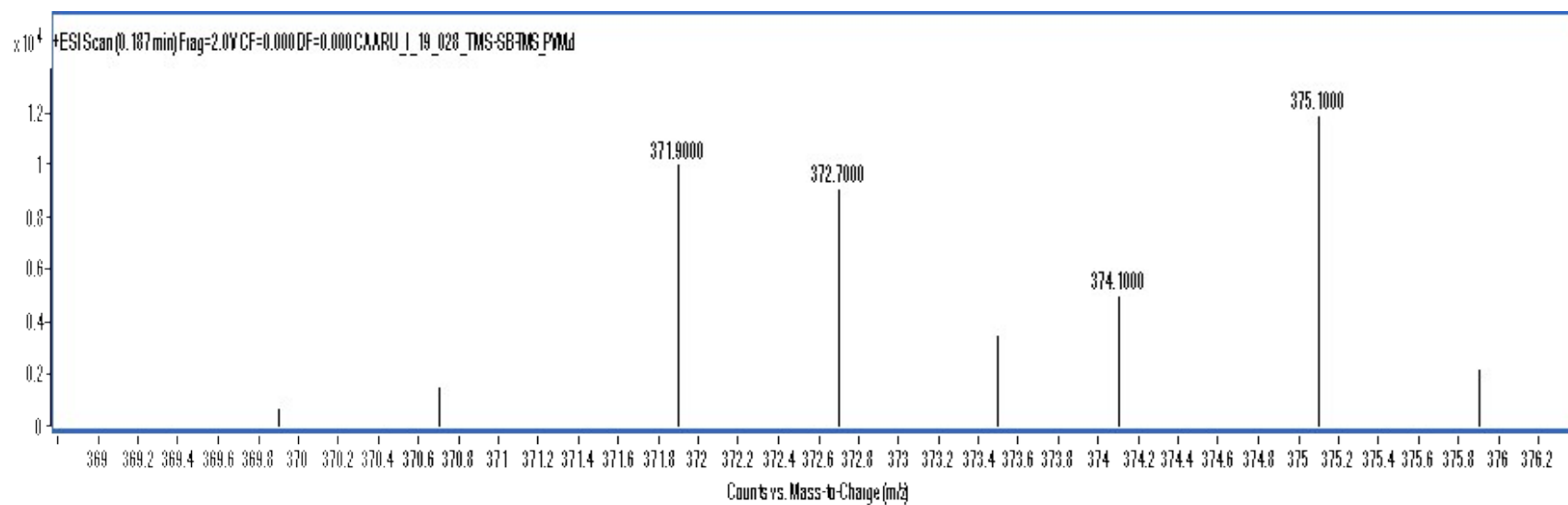


Figure S4. ESI-MS spectrum of **1a**.

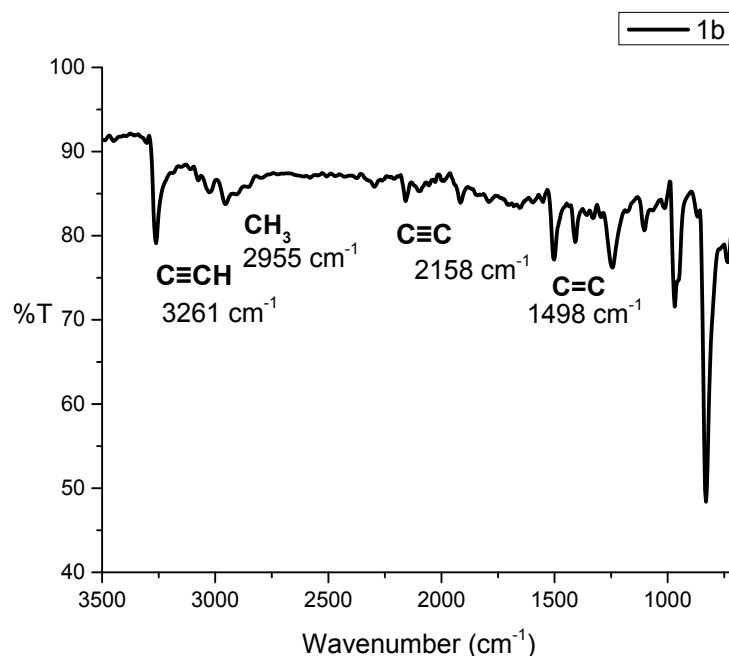


Figure S5. FTIR spectrum of **1b**.

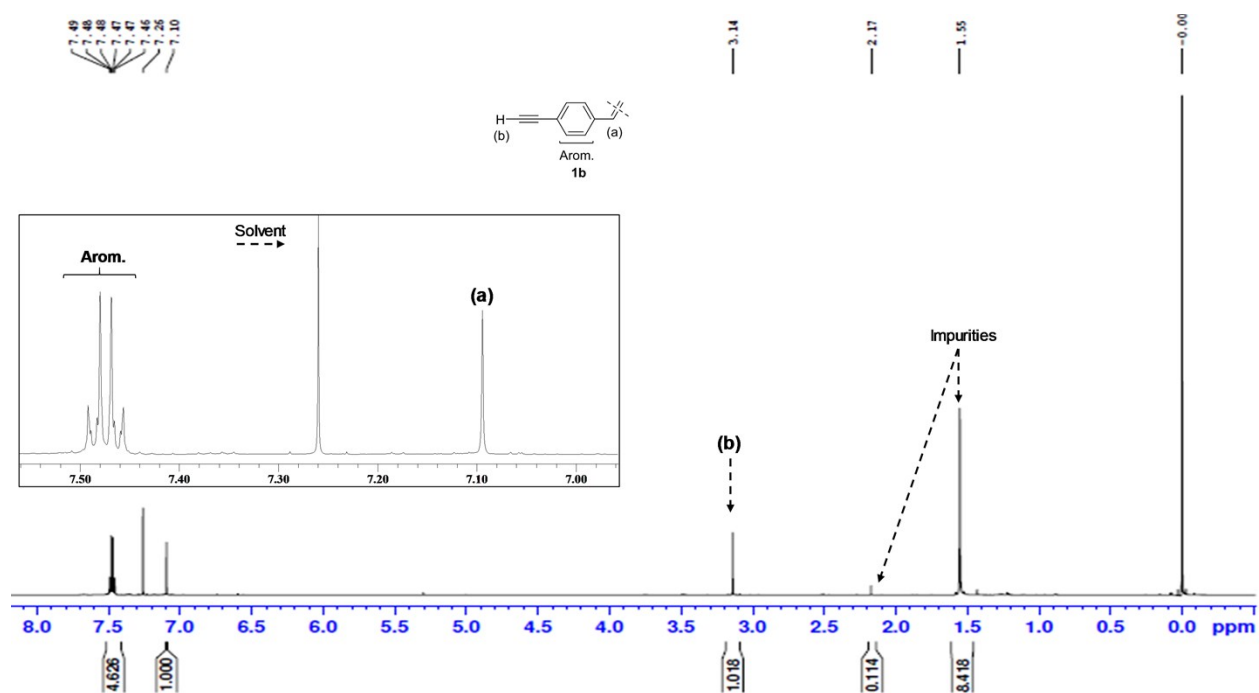


Figure S6. ^1H -NMR spectrum of **1b**.

H-Sb-H-C13CPD

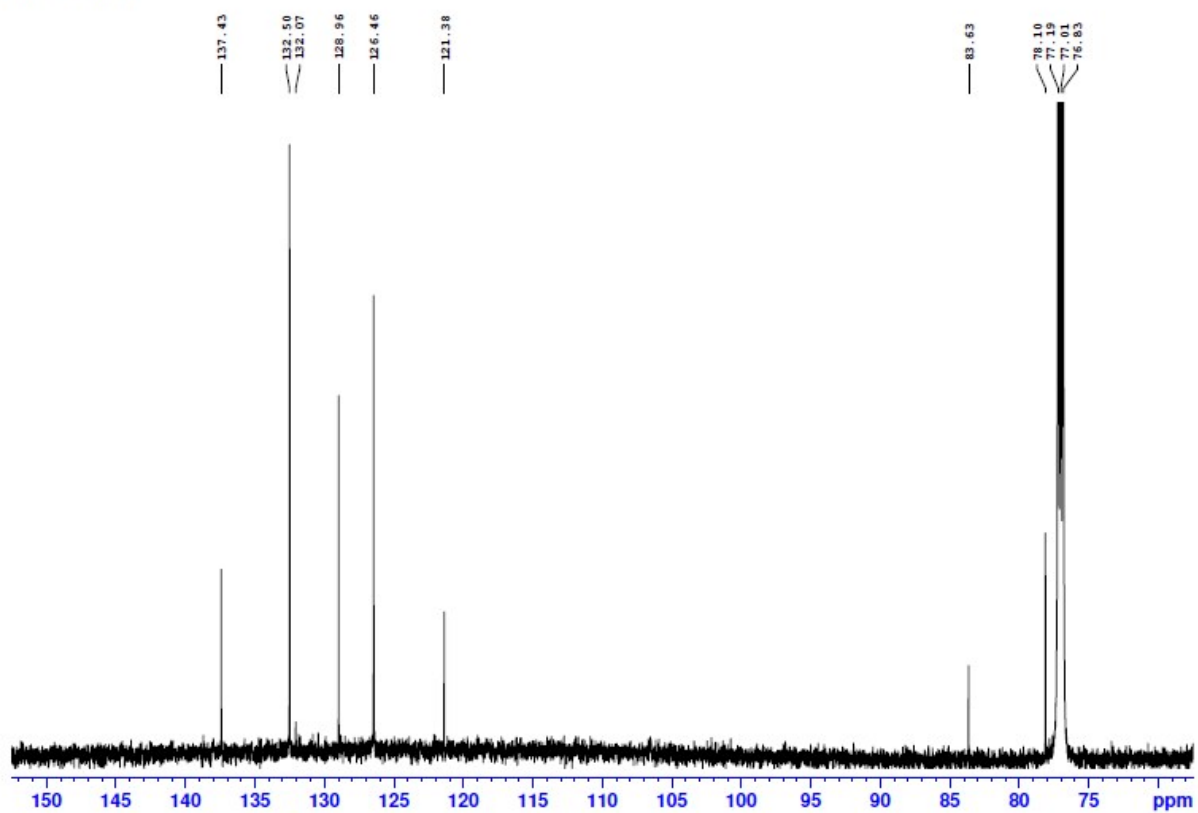


Figure S7. ¹³C-NMR spectrum of **1b**.

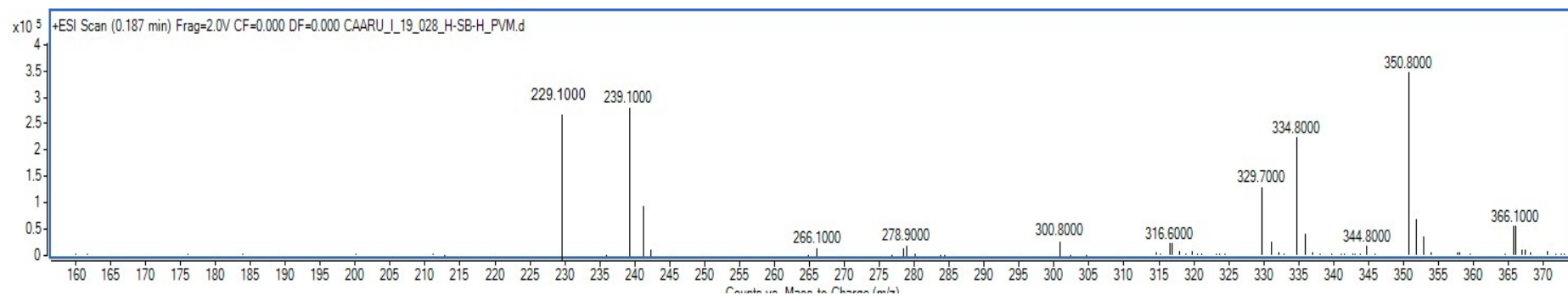


Figure S8. ESI-MS spectrum of **1b**.

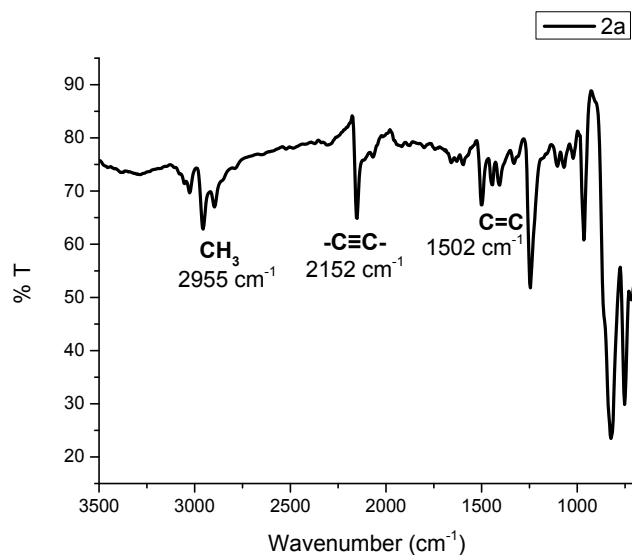


Figure S9. FTIR spectrum of **2a**.

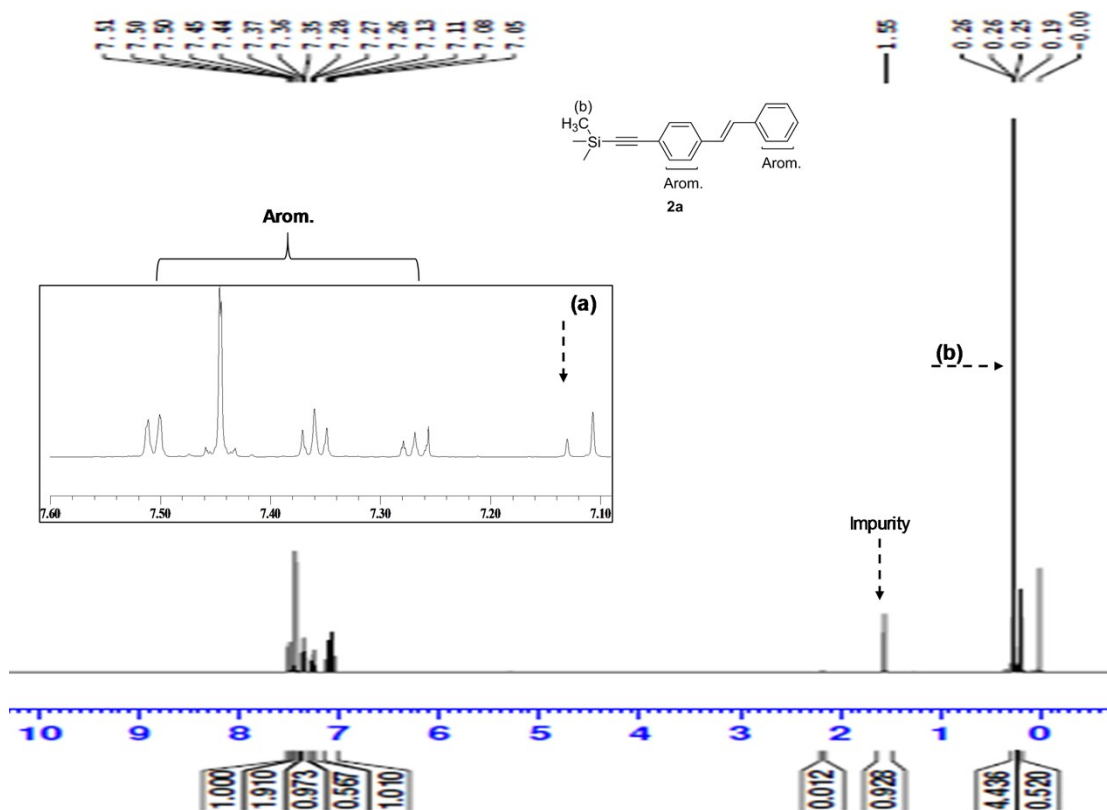


Figure S10. ^1H -NMR spectrum of **2a**.

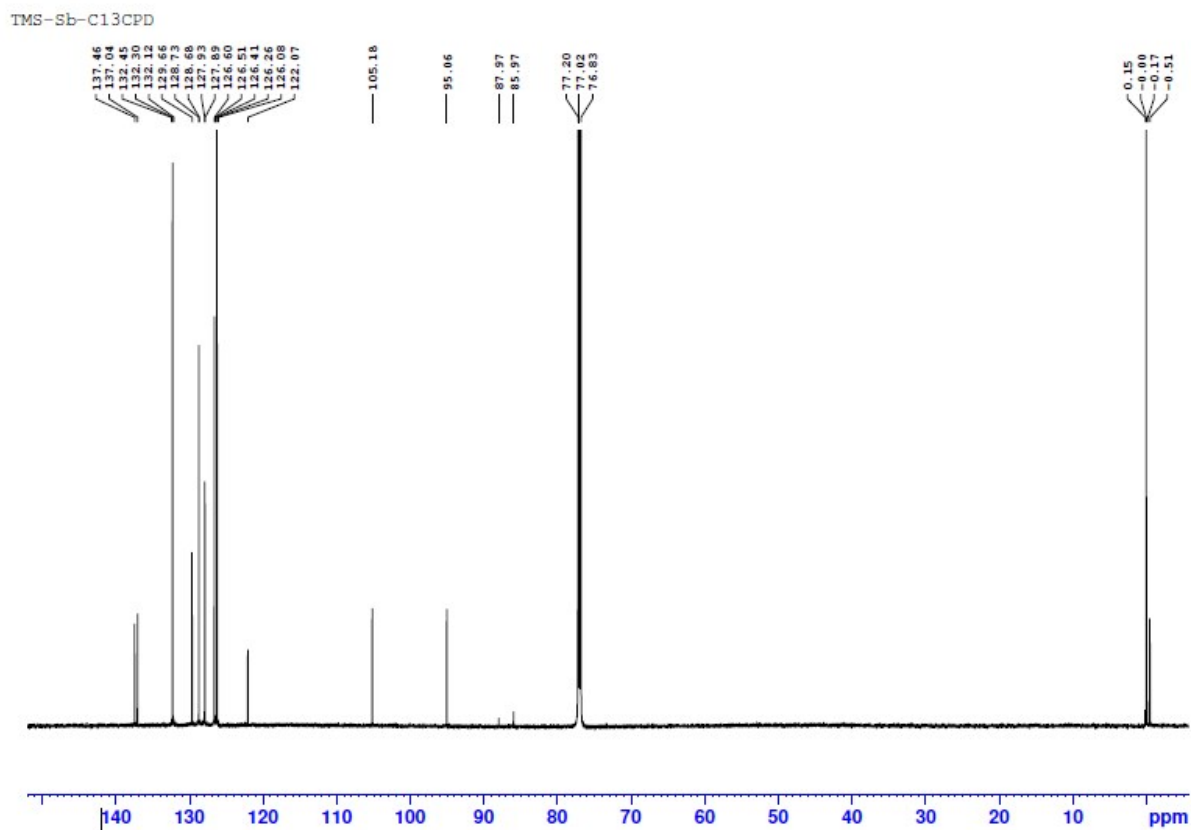


Figure S11. ^{13}C spectrum of 2a.

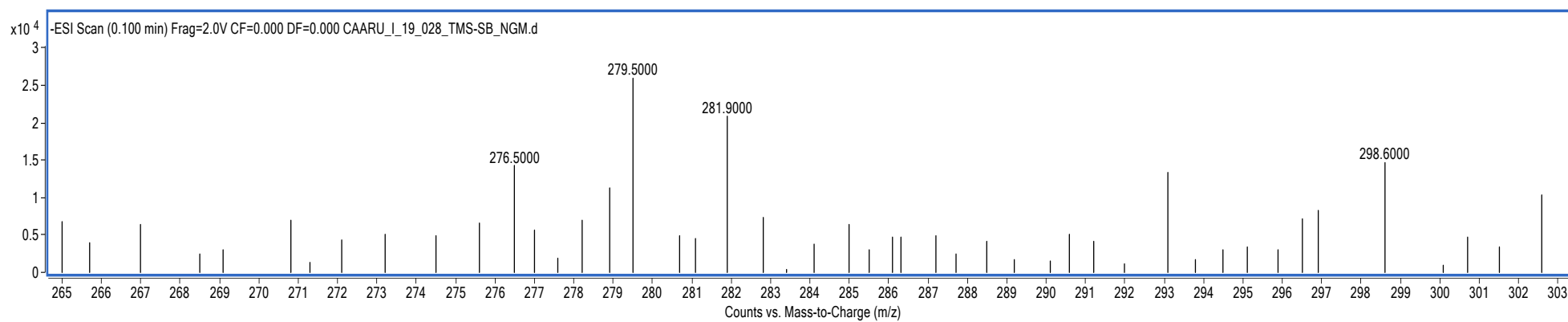


Figure S12. ESI-MS spectrum of **2a**.

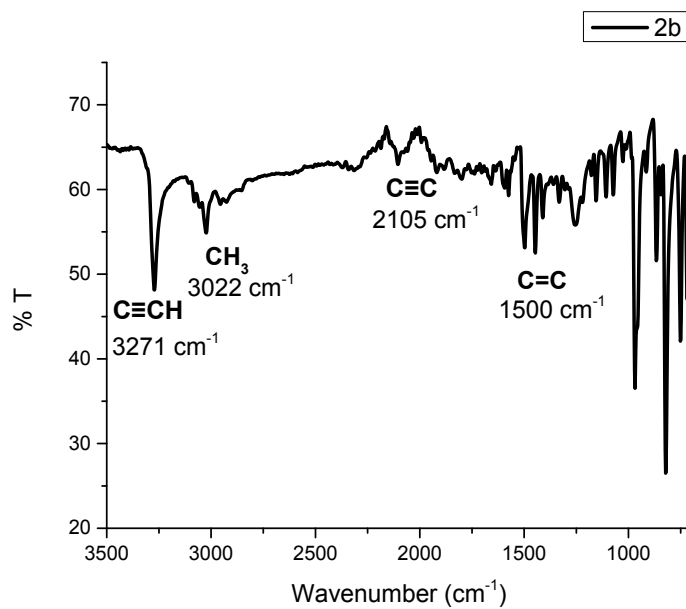


Figure S13. FTIR spectrum of **2b**.

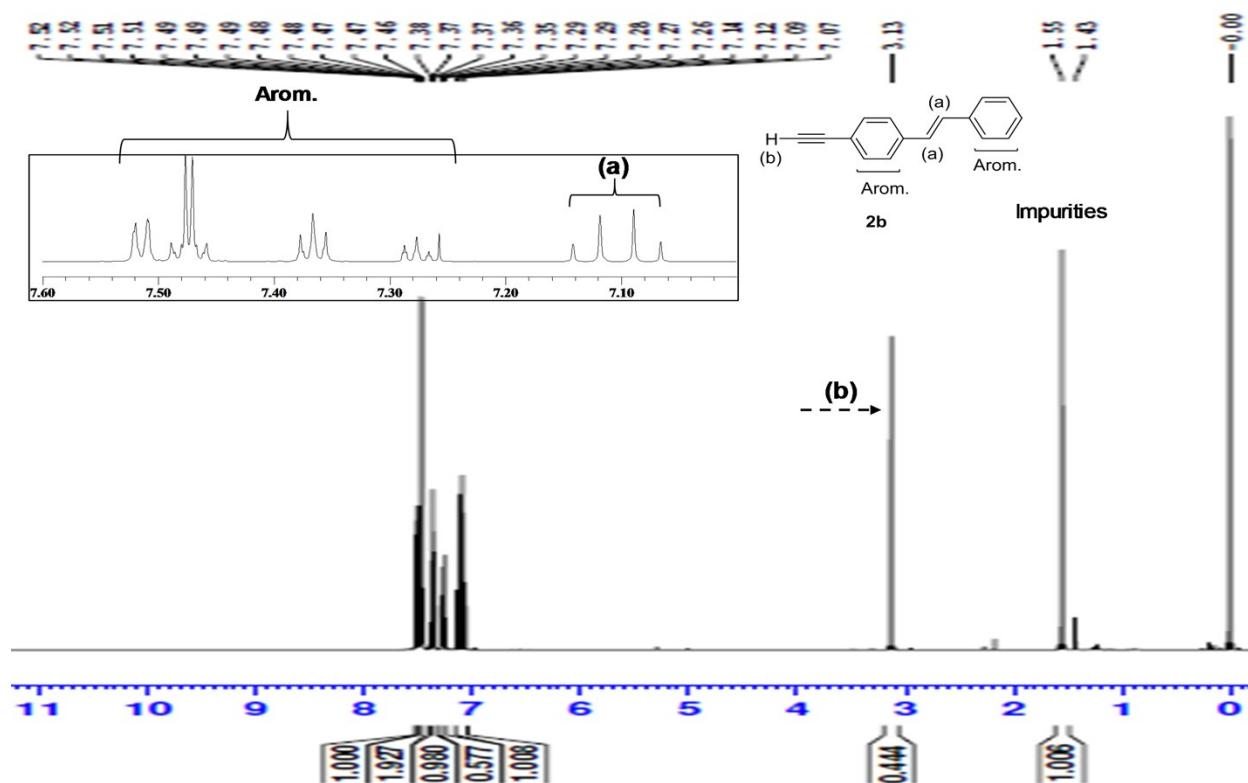


Figure S14. ^1H -NMR spectrum of **2b**.

H-Sb-C13CPD

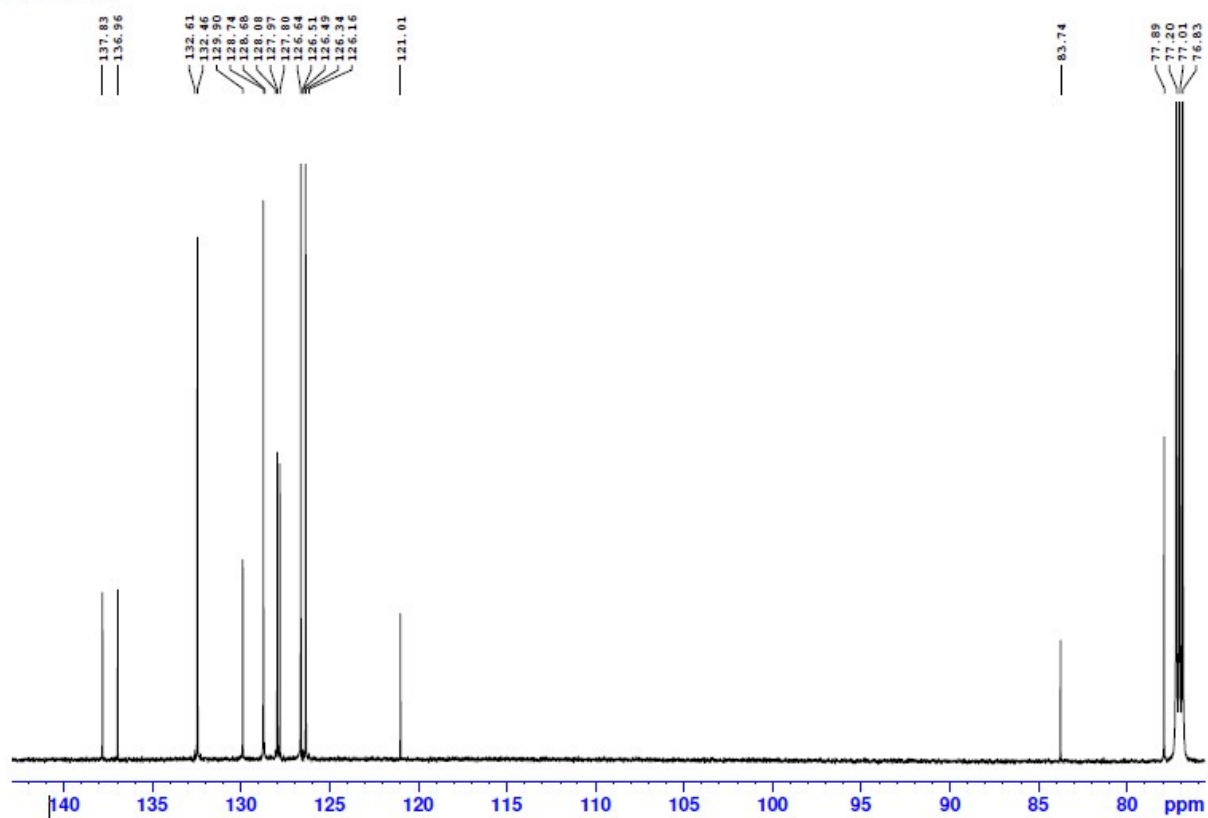


Figure S15. ^{13}C -NMR spectrum of **2b**.

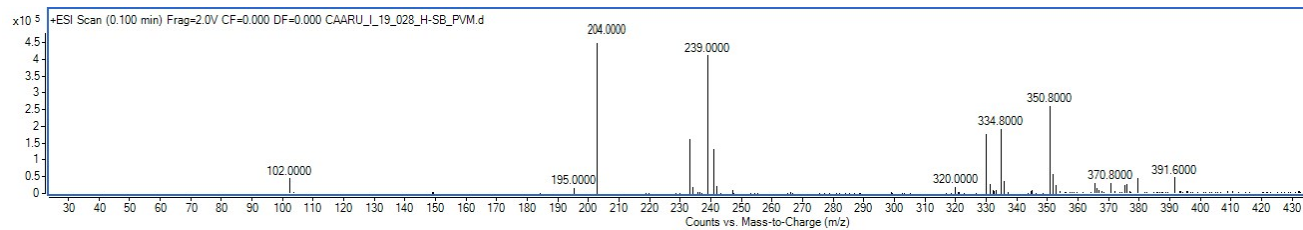


Figure S16. ESI-MS spectrum of **2b**.

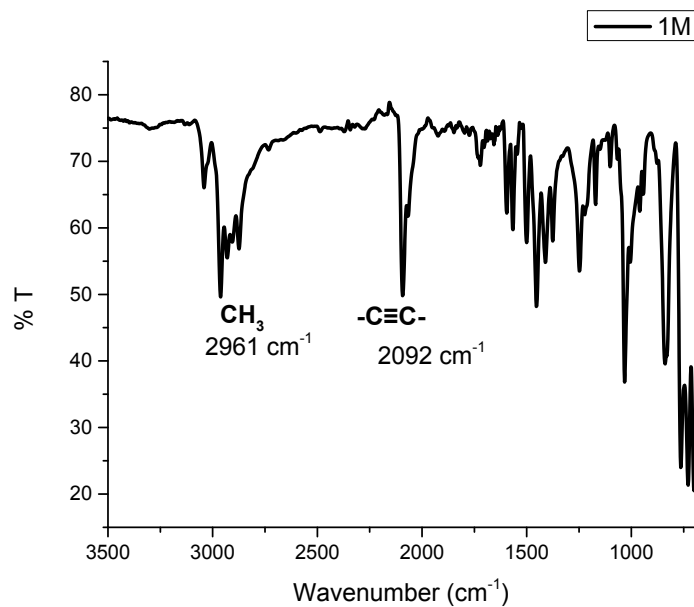


Figure S17. FTIR spectrum of **1M**.

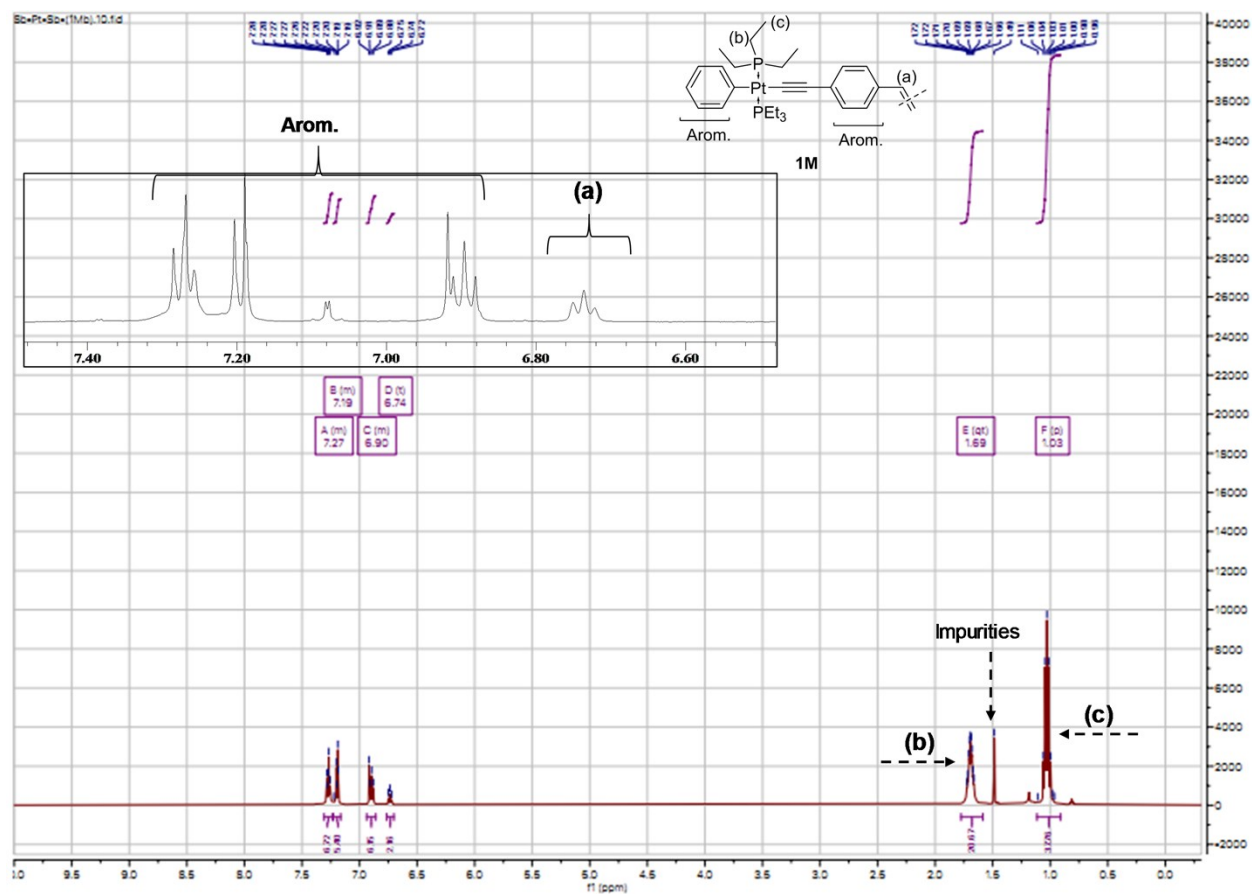


Figure S18. ^1H -NMR spectrum of **1M**.

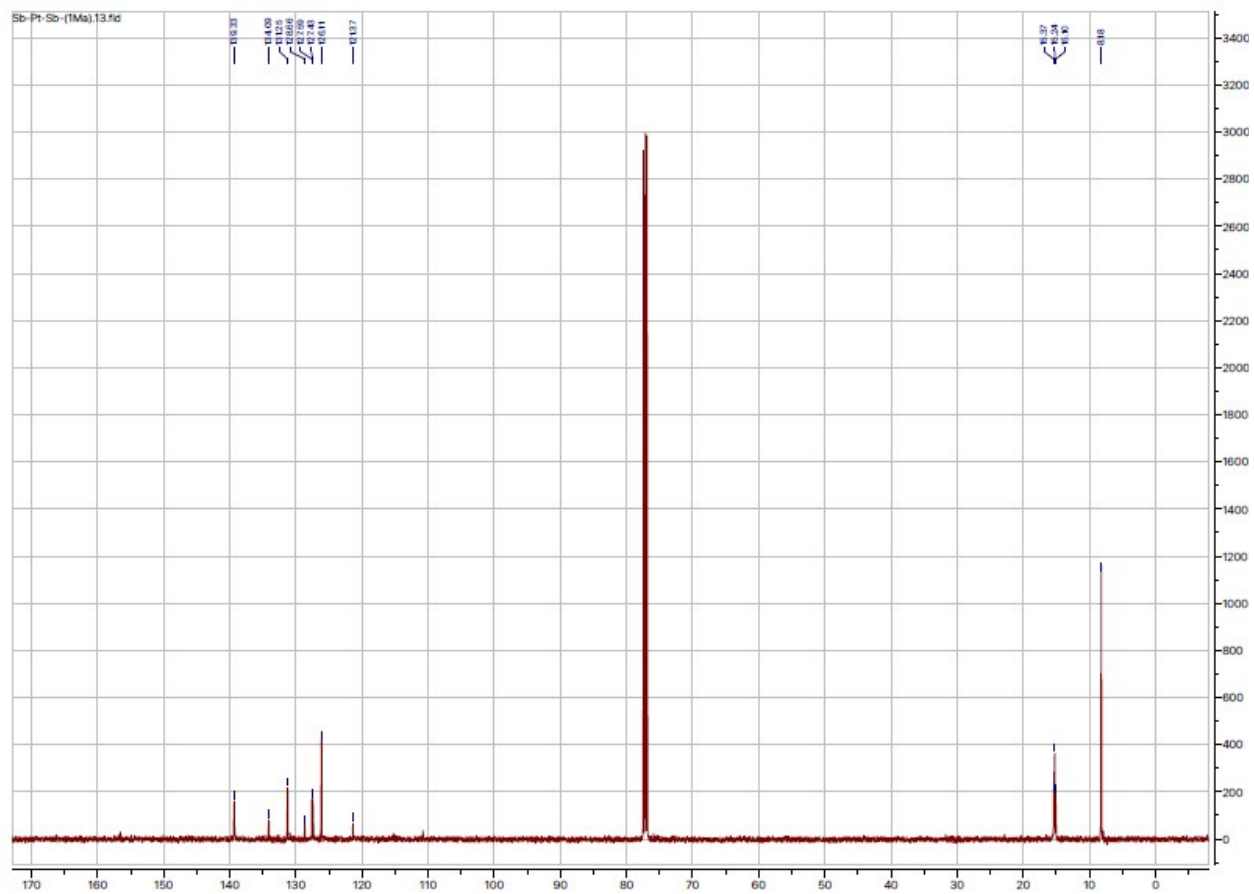


Figure S19. ¹³C-NMR spectrum of **1M**.

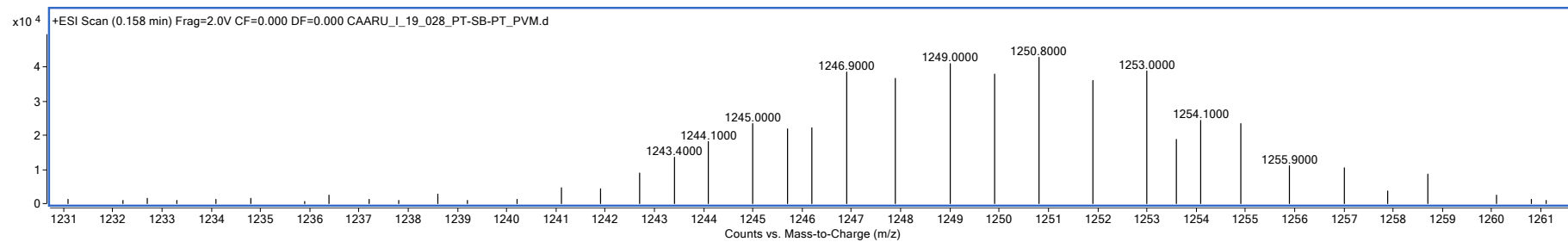


Figure S20. ESI-MS spectrum of **1M** .

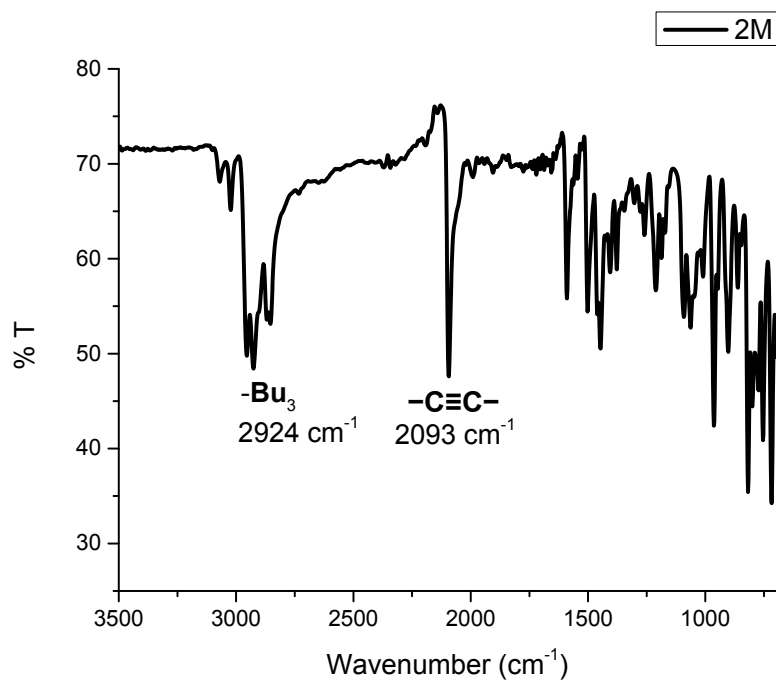


Figure S21. FTIR spectrum of **2M**.

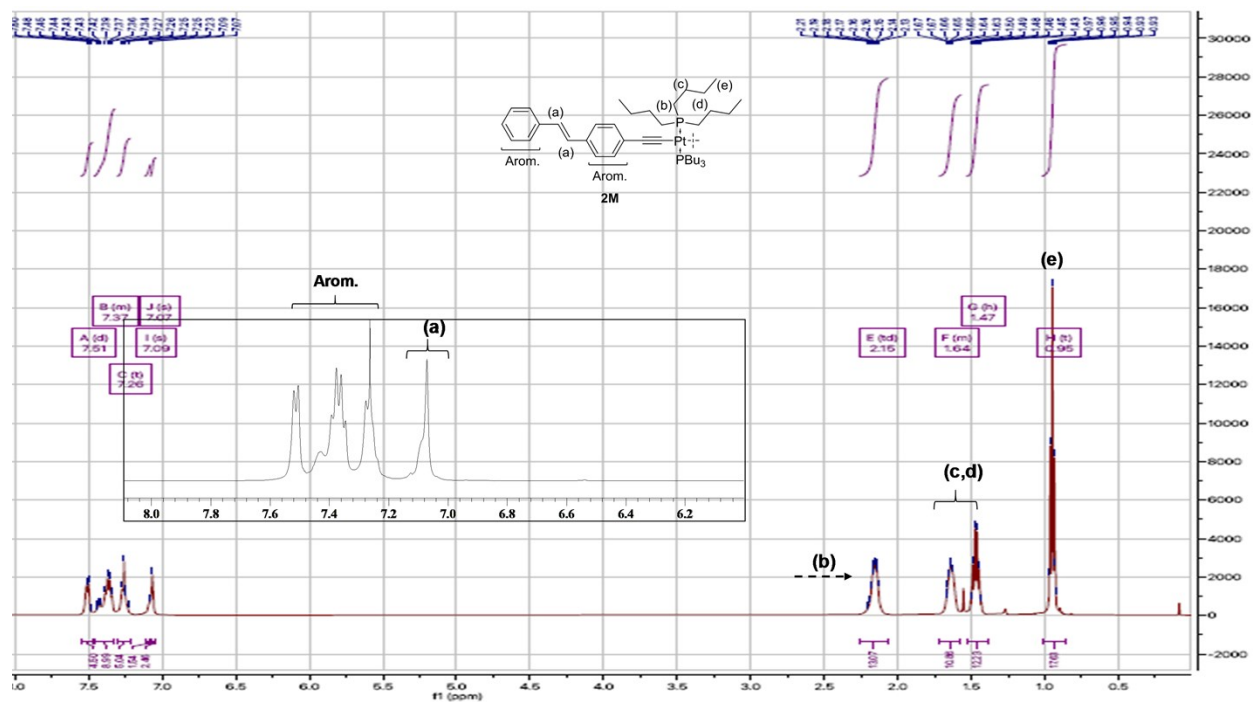


Figure S22. ^1H -NMR spectrum of **2M**.

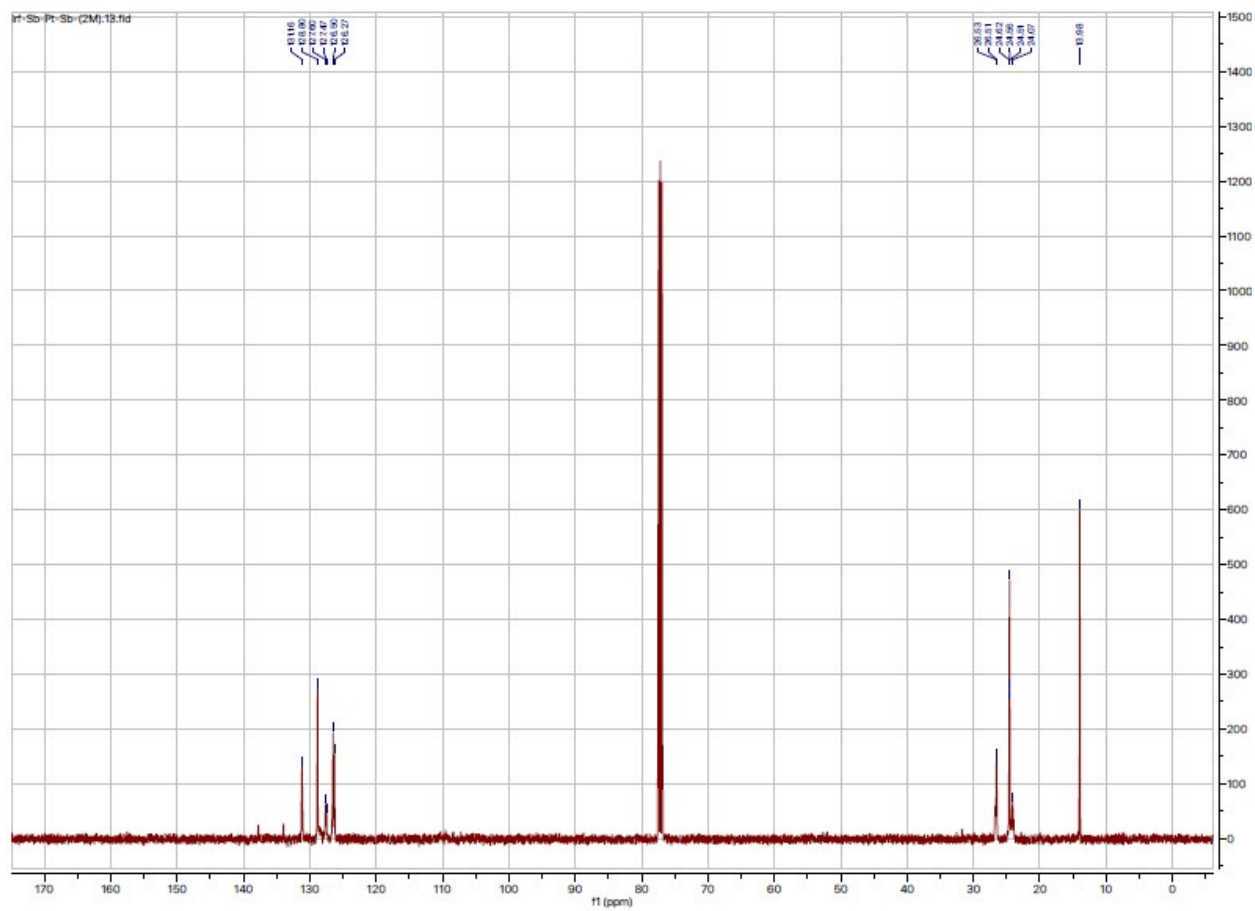


Figure S23. ¹³C-NMR spectrum of **2M**.

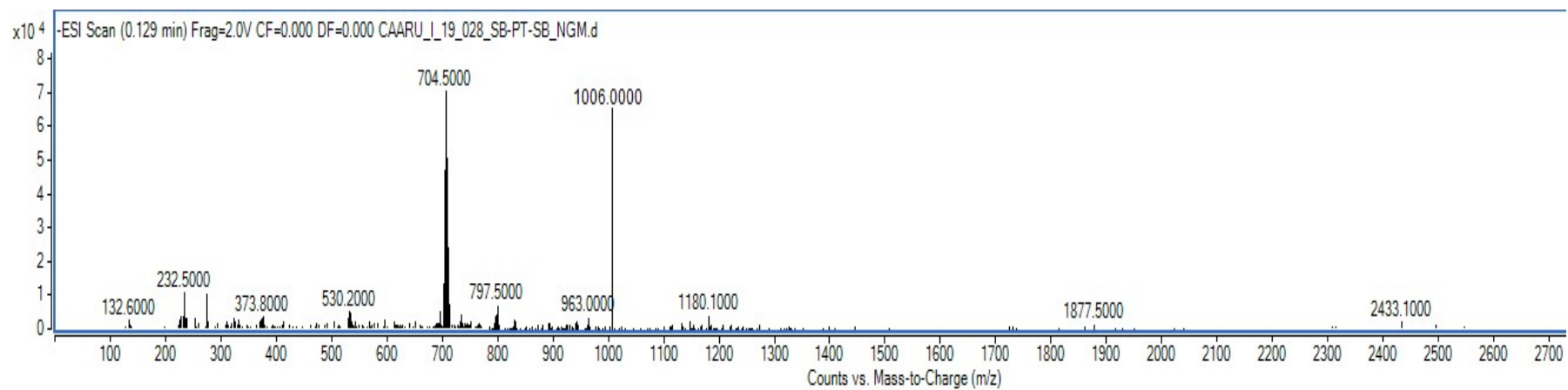


Figure S24. ESI-MS of **2M**.

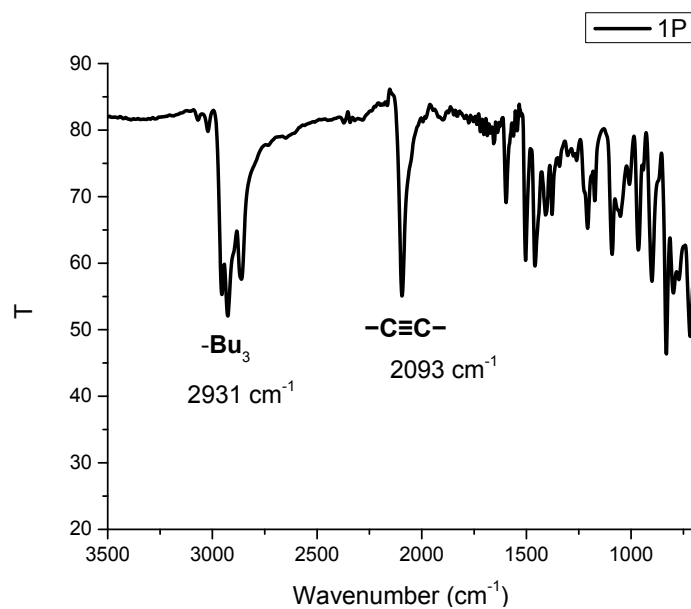


Figure S25. FTIR spectrum of **1P**.

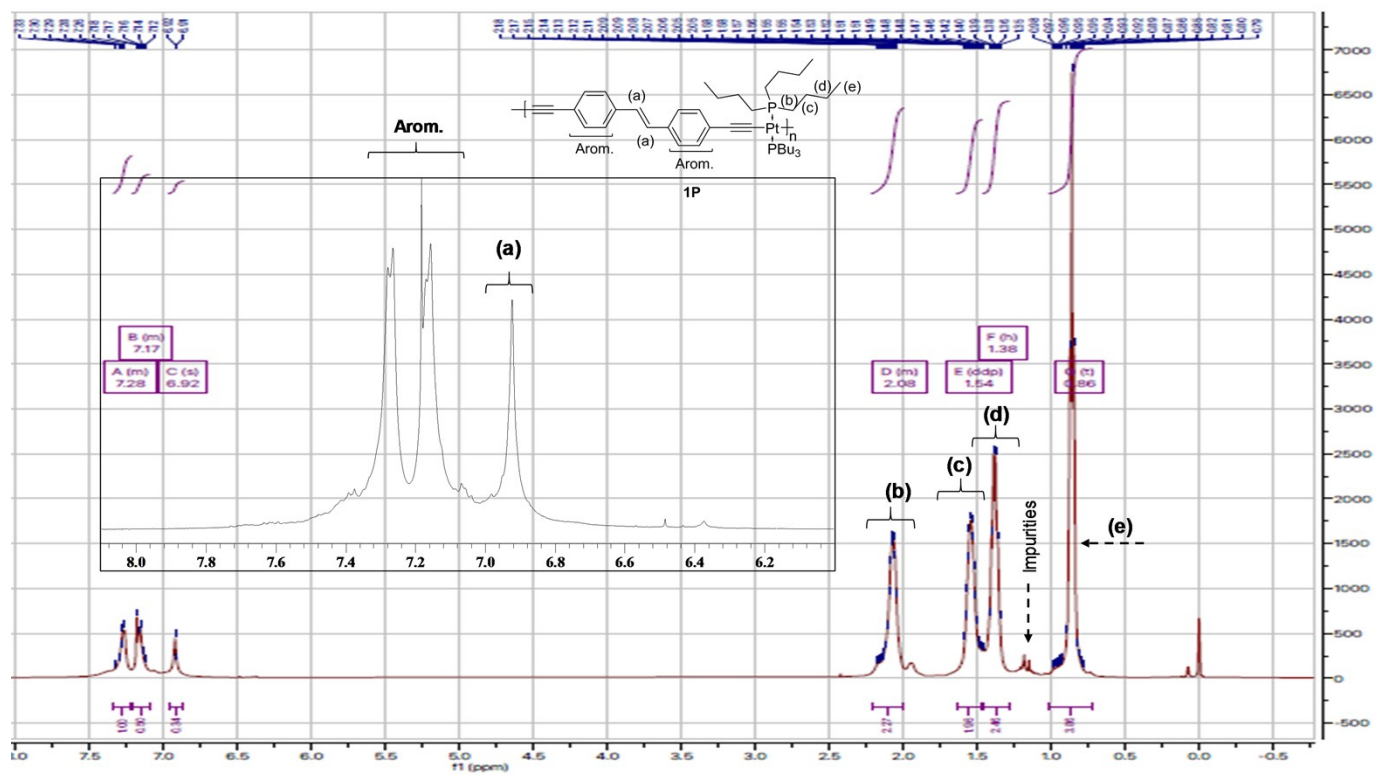


Figure S26. ^1H -NMR spectrum of **1P**.

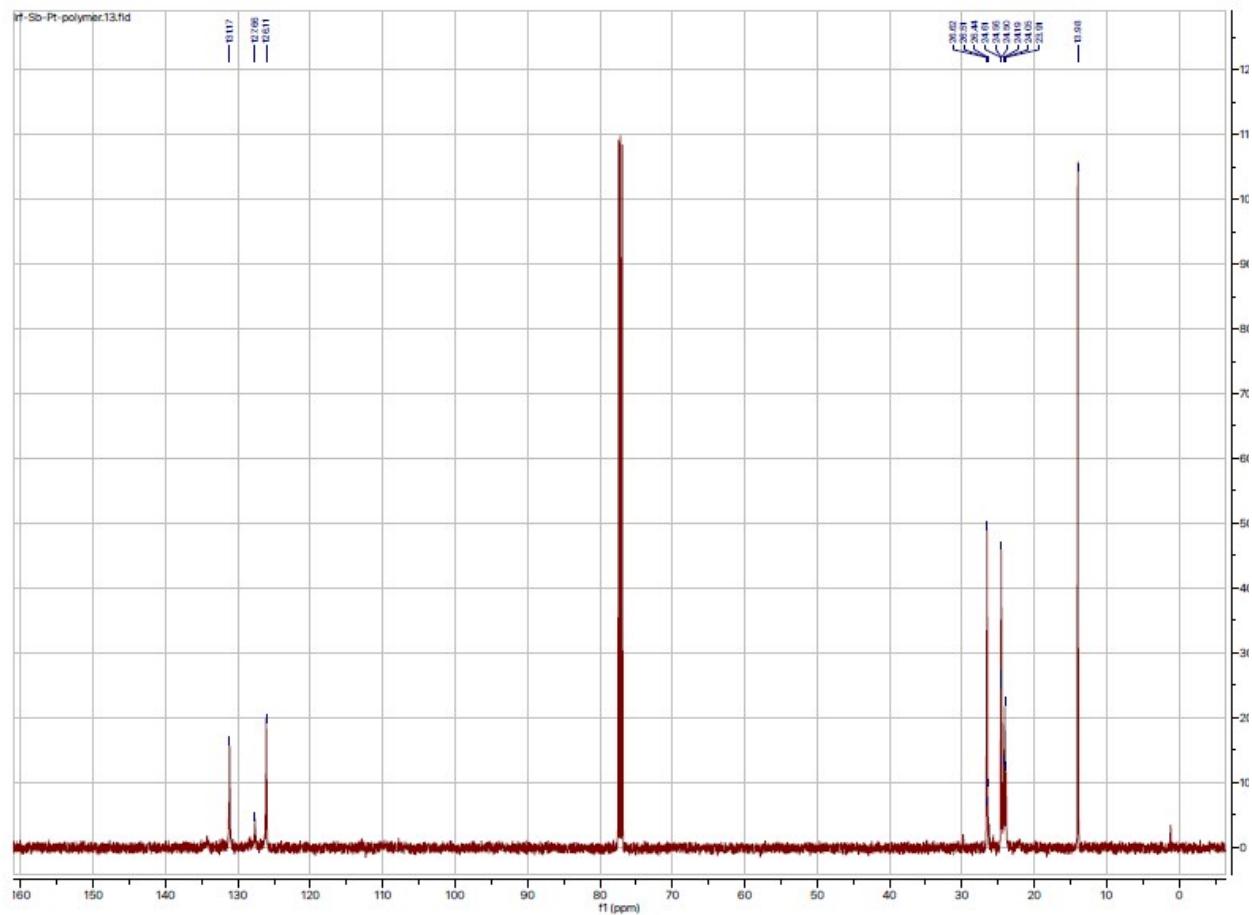


Figure S27. ^{13}C -NMR spectrum of 1P.



Figure S28. ¹H NMR of **1M** after it was irradiated at 365 nm for 25 min. (spectrum obtained with sample in CDCl₃ as a solvent).

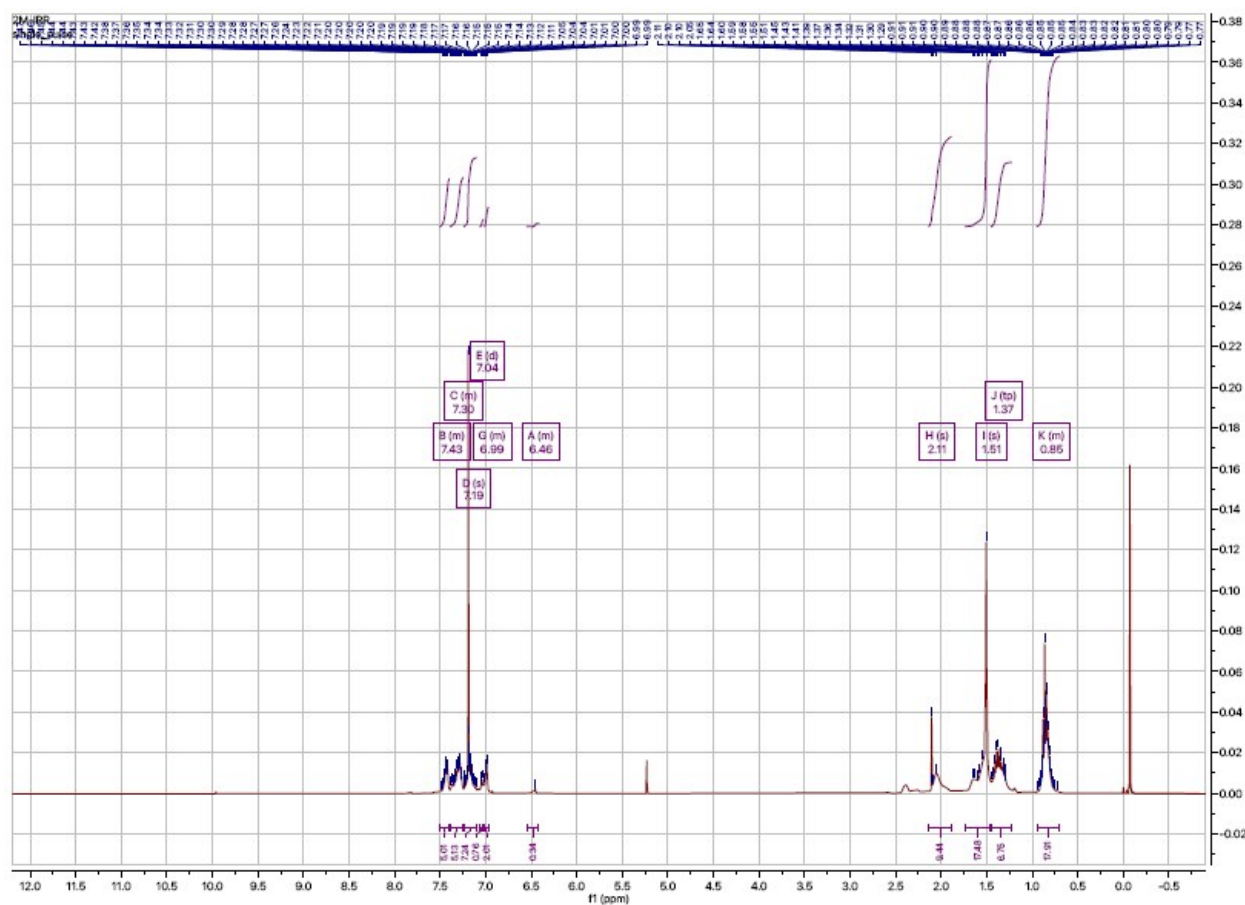


Figure S29. Aromatic region ^1H NMR of **2M** after it was irradiated at 365 nm for 25 min. Insets: ^1H NMR spectrum of olefinic hydrogen atoms (spectrum obtained with sample in CDCl_3 as a solvent).

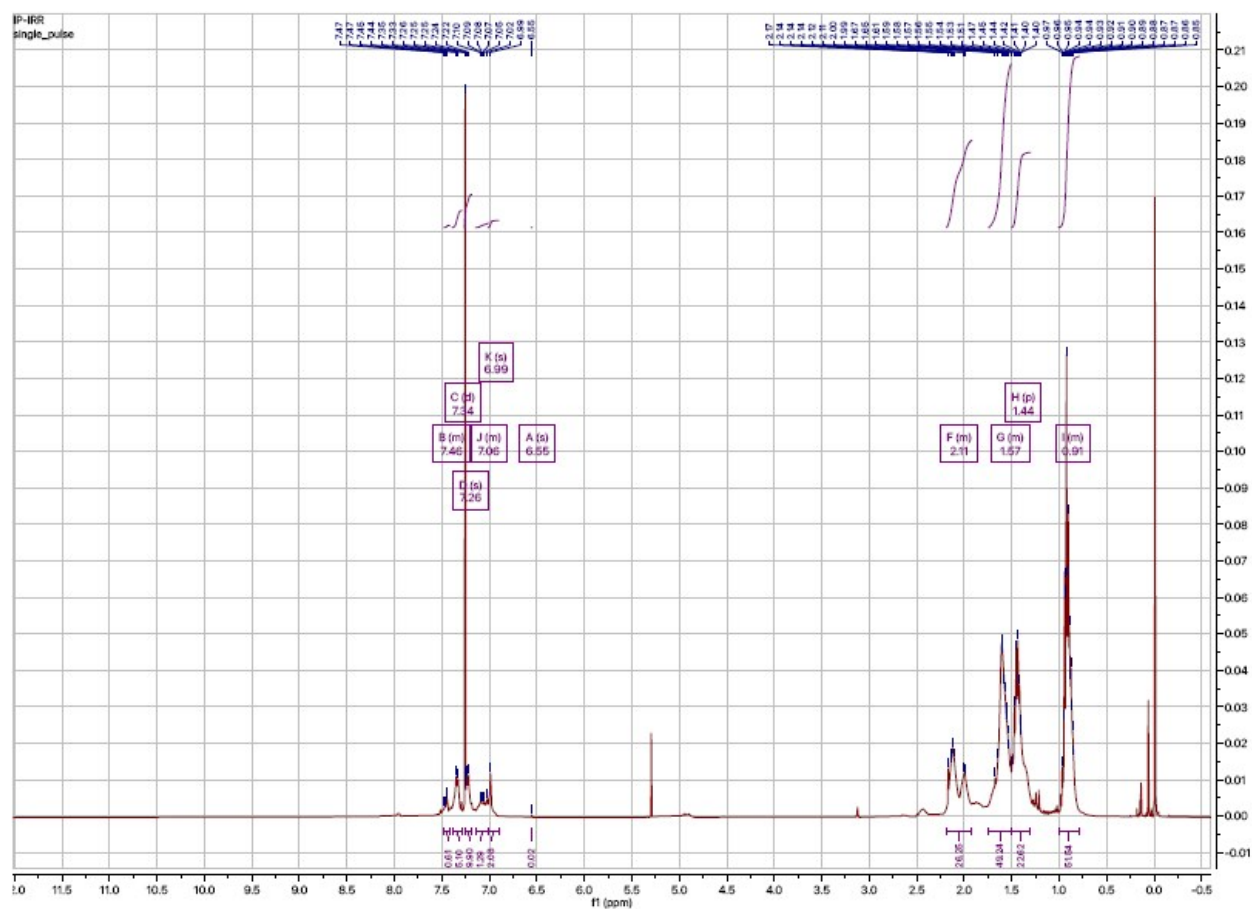


Figure S30. ¹H NMR of **1P** after it was irradiated at 365 nm for 25 min. (spectrum obtained with sample in CDCl₃ as a solvent)

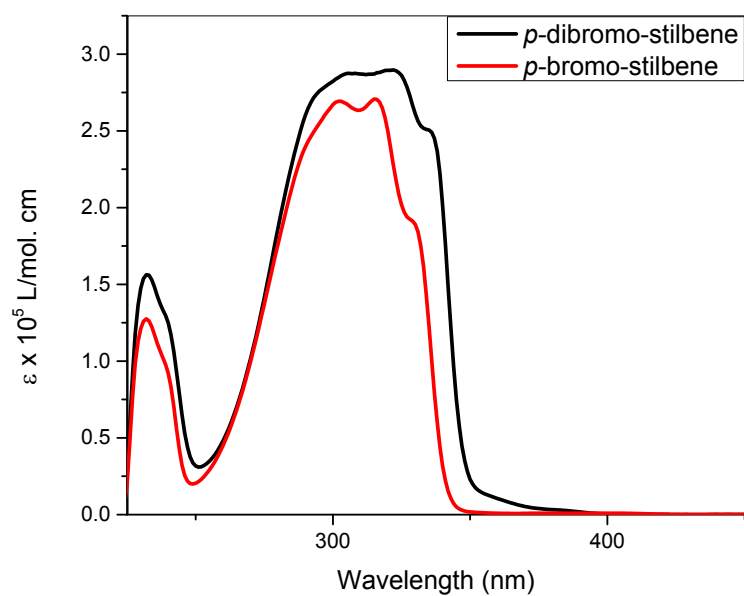


Figure S31. Electronic spectra of the stilbene ligands.

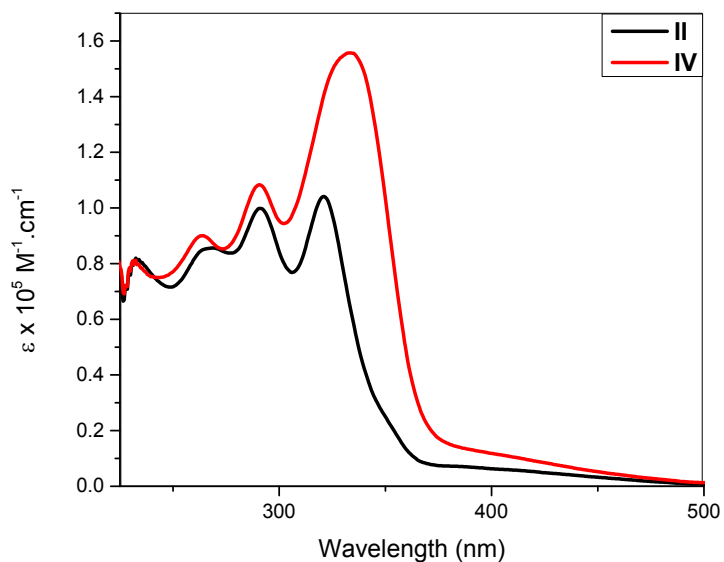


Figure S32. Electronic absorption spectra of meta-azobenzene diyne (**II**) and poly-yne (**IV**) systems.

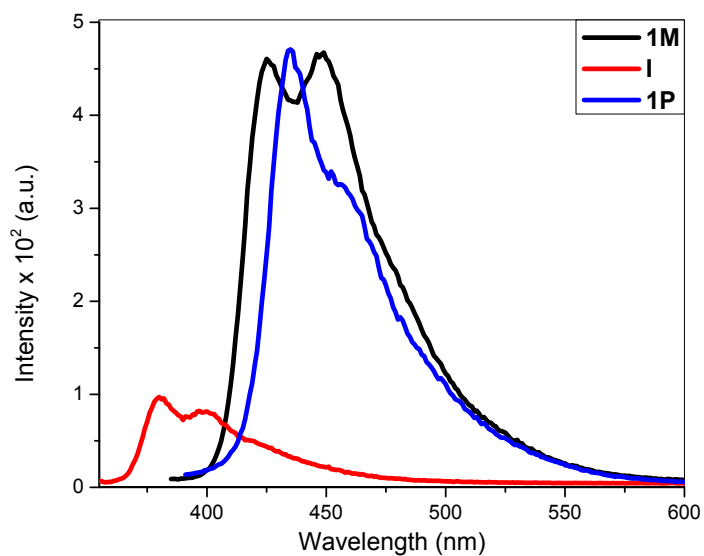


Figure S33. Emission spectra of azobenzene emitted species along with **1M** and **1P**. Excitation was at 380 nm for **1M** and **1P** and 340 nm for **I**.

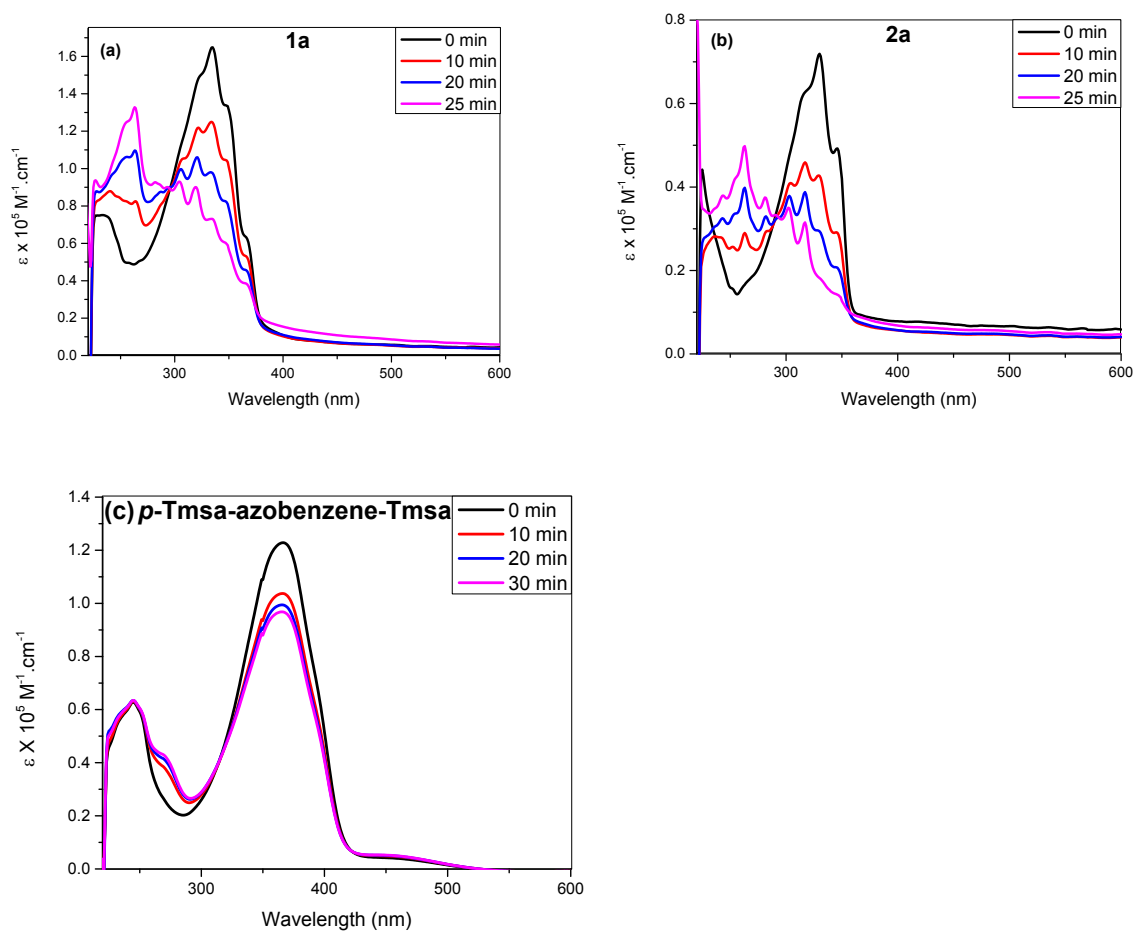


Figure S34. Changes in absorption spectral profile of compounds **1a** (a), **2a** (b) and *p*-Tmsa-azobenzene-Tmsa (c) upon irradiation with light at 254 nm at different exposure times.

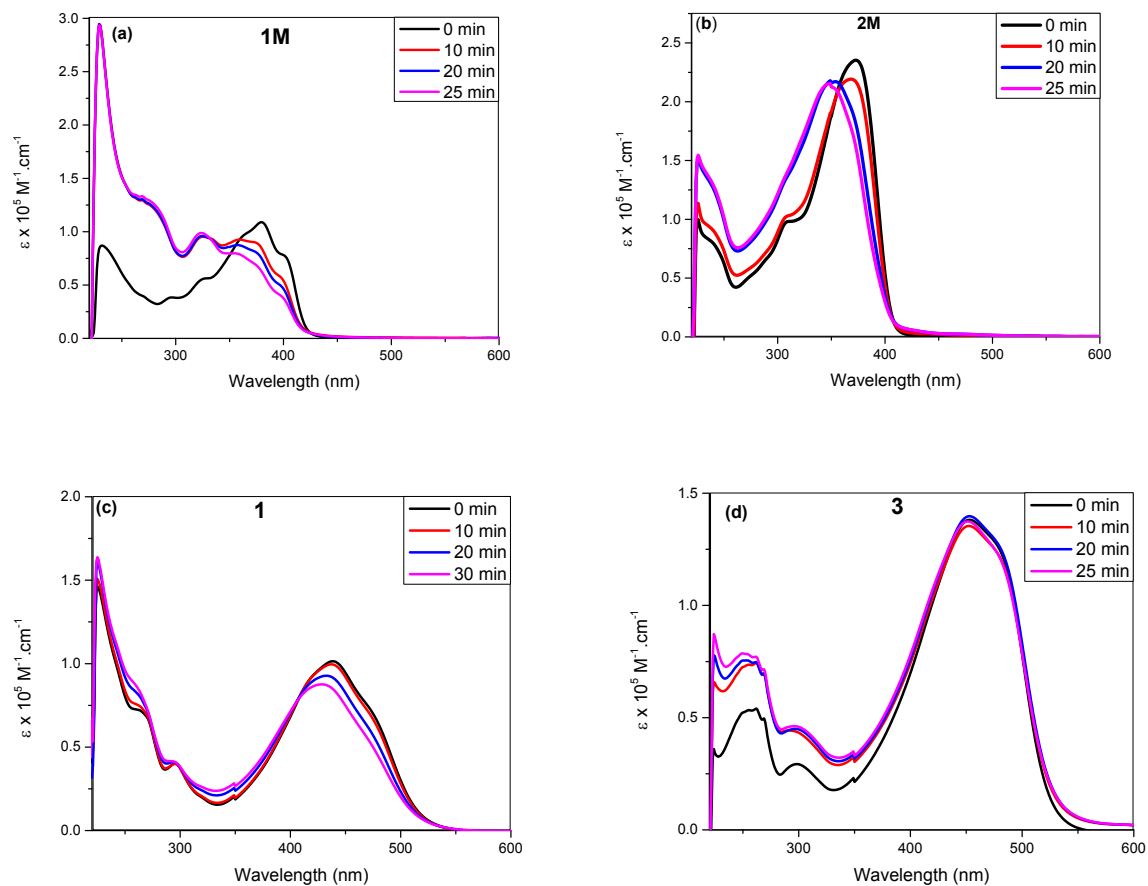


Figure S35. Changes in absorption spectral profile of compounds 1M (a), 2M (b) 1 (Pt(II) di-yne with para-azobenzene spacer group) (c) and 3 (Pt(II) poly-yne with para-azobenzene spacer group) upon irradiation with light at 254 nm at different exposure times.

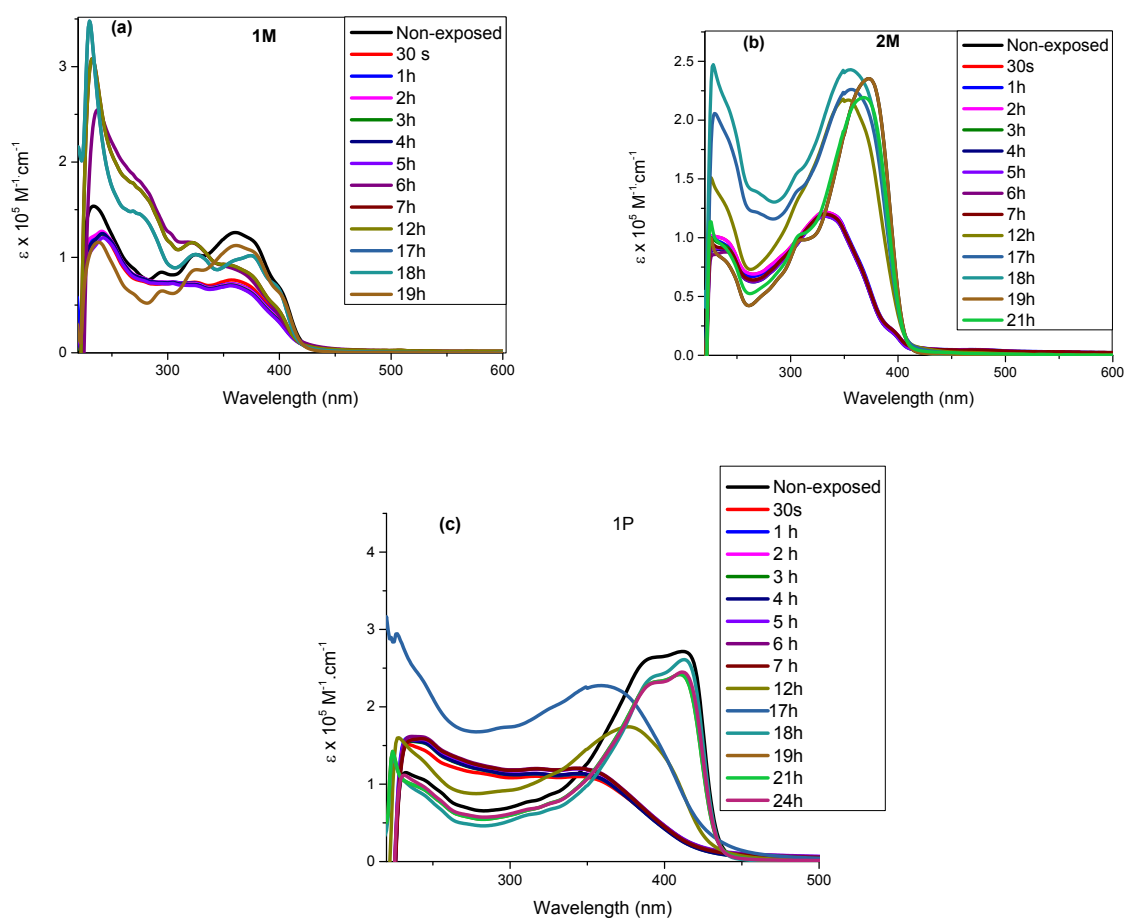


Figure S36. Absorption spectra profile with recovery time required for **1M (a)**, **2M (b)** and **1P (c)** after 25 minutes irradiation at 365 nm.

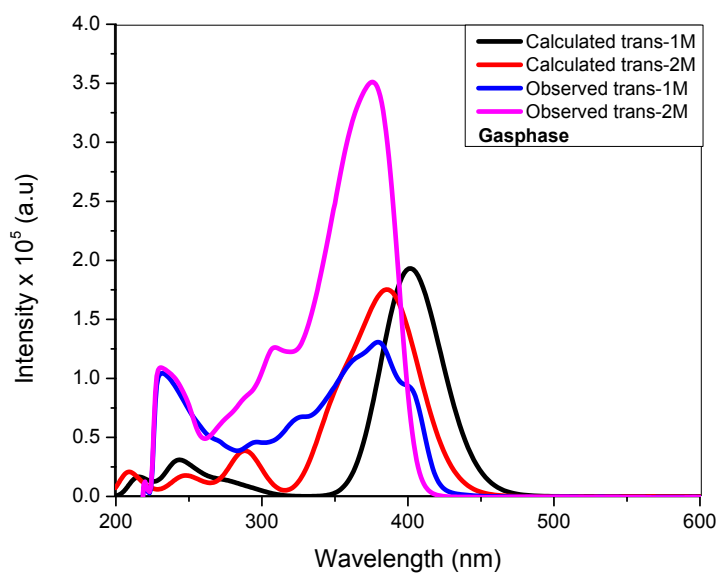


Figure S37. Comparison of simulated *trans*-configurations of **1M** and **2M** optical absorption spectra versus the experimental observed spectra of *trans*-**1M** and **2M**. The simulated spectra are in gas phase simulations.

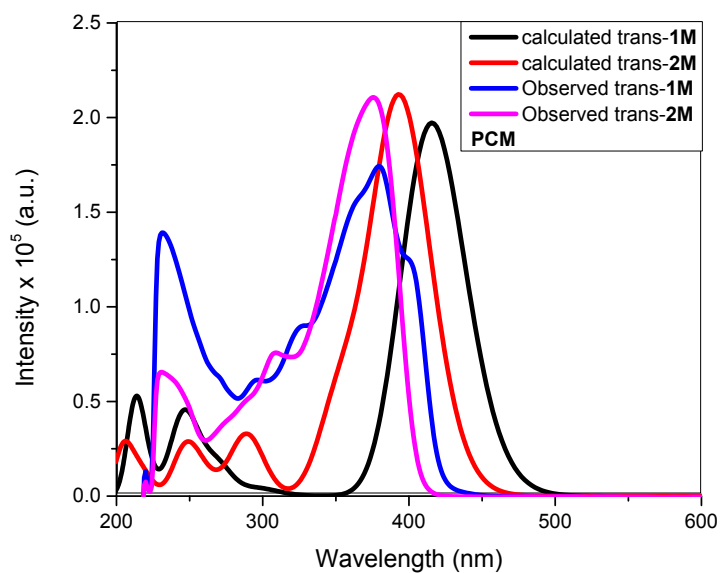


Figure S38. Comparison of simulated *trans*-configurations of **1M** and **2M** optical absorption spectra versus the experimental observed spectra of *trans*-**1M** and **2M**. The calculated spectra are in PCM simulations.

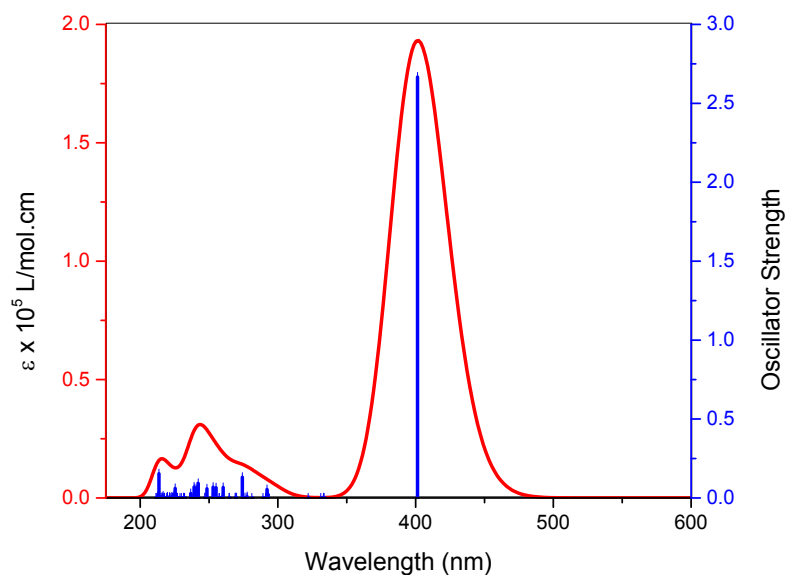


Figure S39. Convolved spectra and line spectra of *trans*-**1M** in gas phase showing the lowest 100 spin-allowed transitions. The convolved spectra were produced using Gaussian peaks with a FWHM of 3000 cm^{-1} .

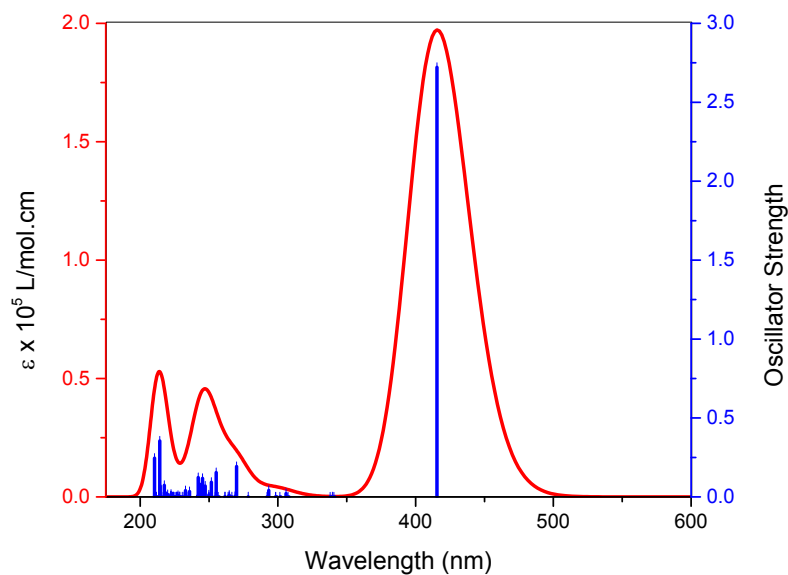


Figure S40. Convolved spectra and line spectra of *trans*-**1M** in PCM showing the lowest 100 spin-allowed transitions. The convolved spectra were produced using Gaussian peaks with a FWHM of 3000 cm^{-1} .

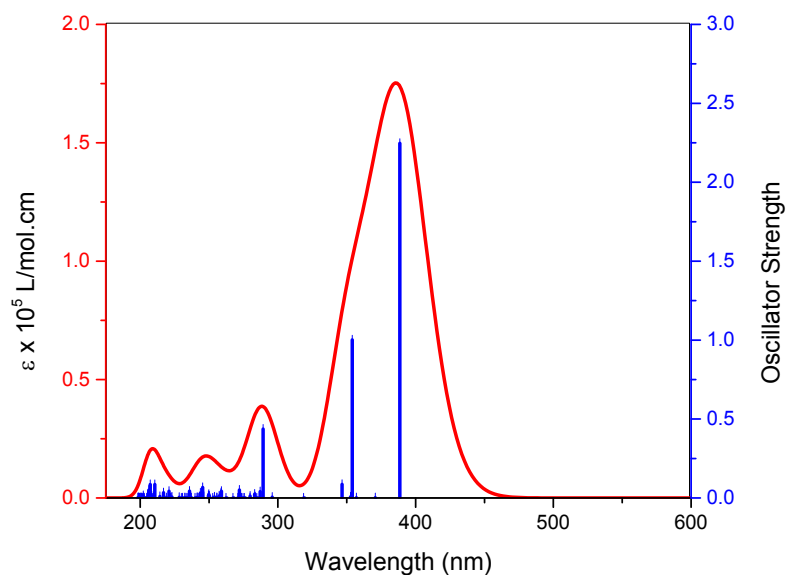


Figure S41. Convolution spectra and line spectra of *trans*-**2M** in gas phase showing the lowest 100 spin-allowed transitions. The convoluted spectra were produced using Gaussia peaks with a FWHM of 3000 cm⁻¹.

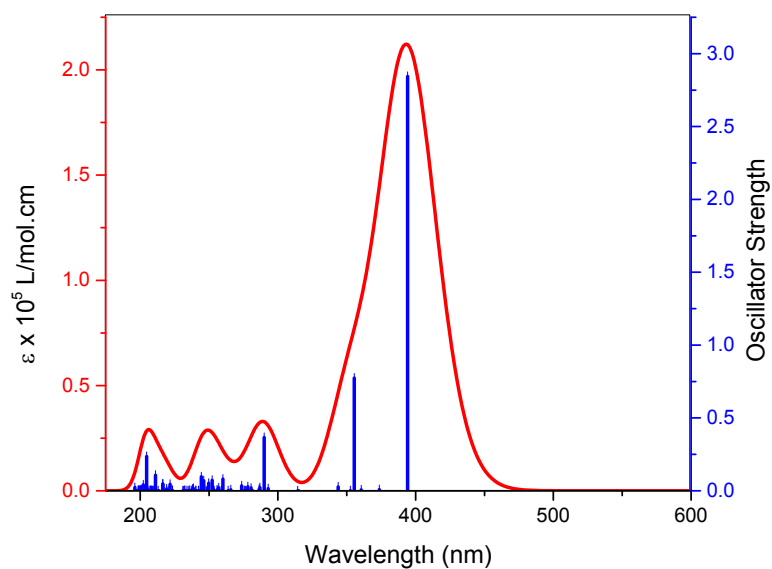


Figure S42. Convolution spectra and line spectra of *trans*-**2M** in PCM showing the lowest 100 spin-allowed transitions. The convoluted spectra were produced using Gaussian peaks with a FWHM of 3000 cm⁻¹.

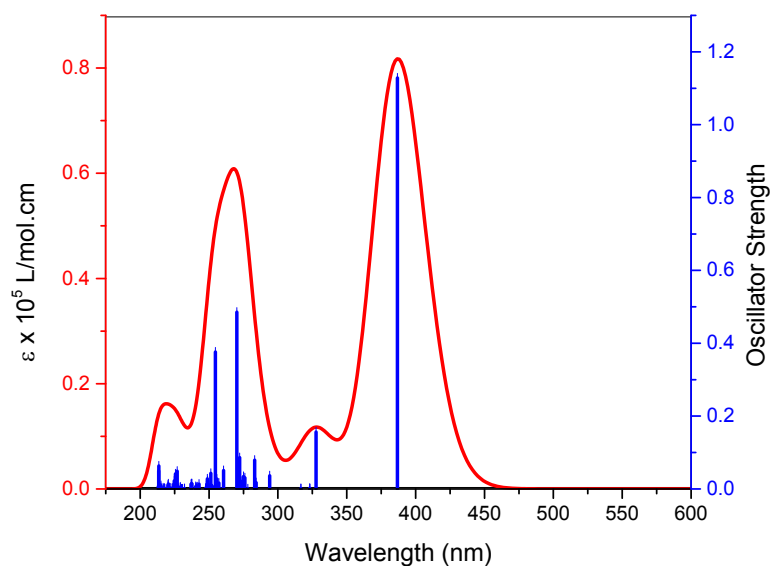


Figure S43. Convolved spectra and line spectra of *cis*-**1M** in gas phase showing the lowest 100 spin-allowed transitions. The convolved spectra were produced using Gaussian peaks with a FWHM of 3000 cm^{-1} .

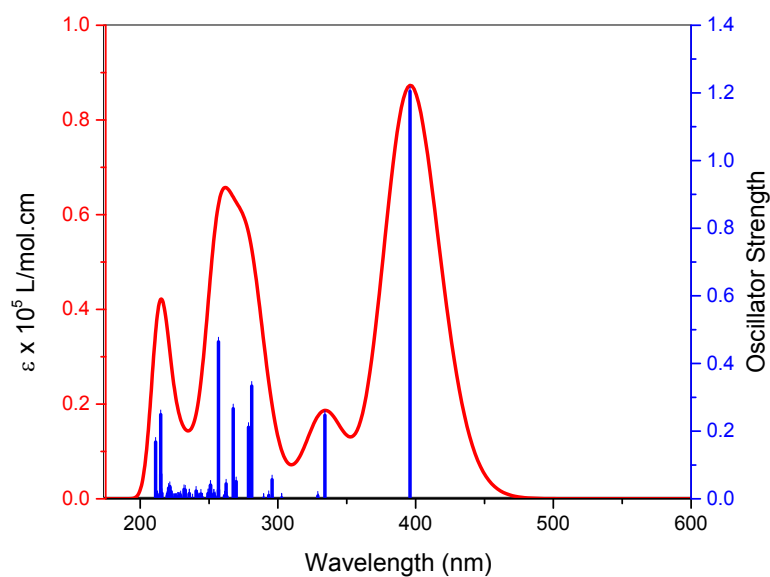


Figure S44. Convolved spectra and line spectra of *cis*-**1M** in PCM showing the lowest 100 spin-allowed transitions. The convolved spectra were produced using Gaussian peaks with a FWHM of 3000 cm^{-1} .

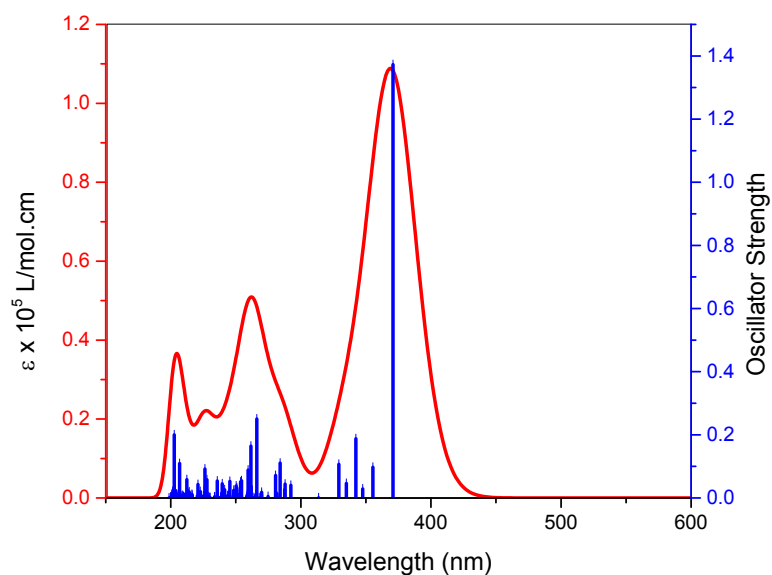


Figure S45. Convolved spectra and line spectra of *cis*-**2M** in gas phase showing the lowest 100 spin-allowed transitions. The convolved spectra were produced using Gaussian peaks with a FWHM of 3000 cm^{-1} .

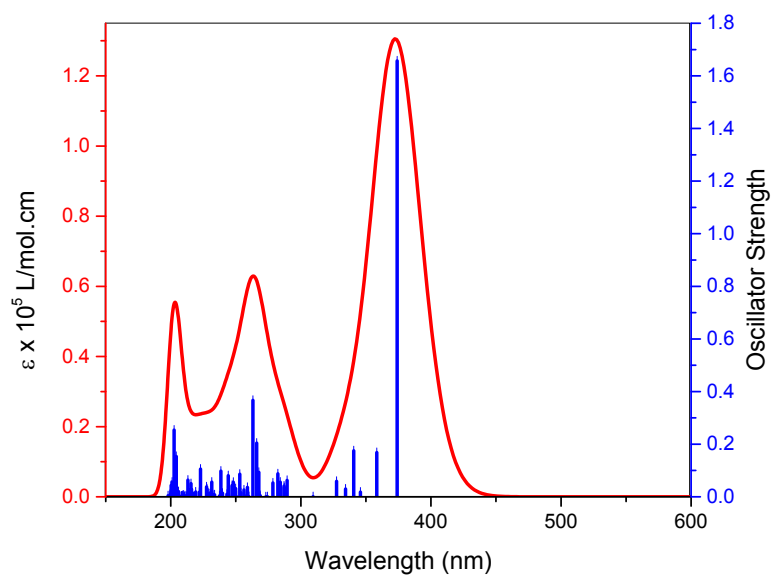


Figure S46. Convolved spectra and line spectra of *cis*-**2M** in PCM showing the lowest 100 spin-allowed transitions. The convolved spectra were produced using Gaussian peaks with a FWHM of 3000 cm^{-1} .

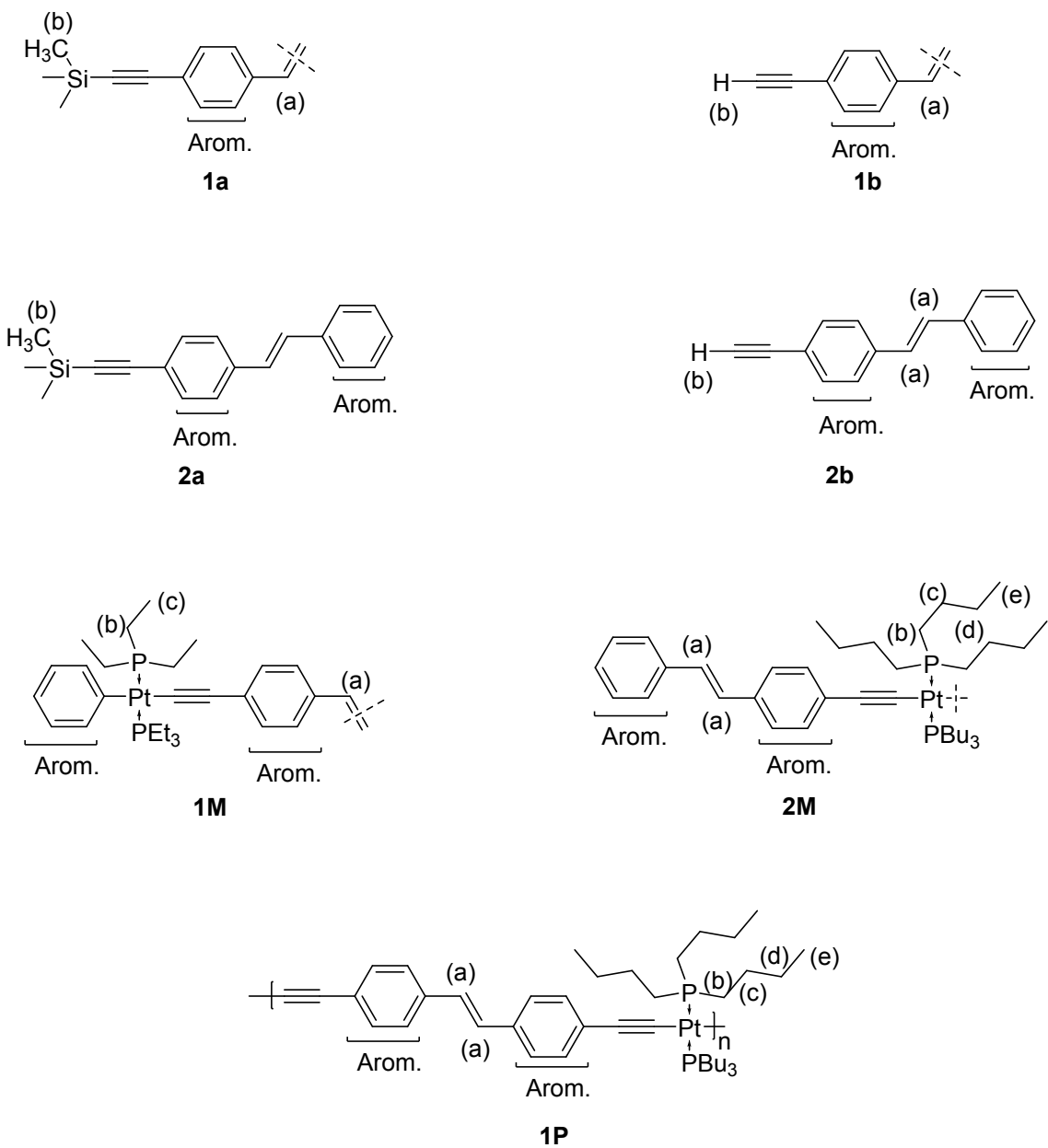


Figure S47. Labelled (numbered) chemical structures used in proton-resonance assignments.

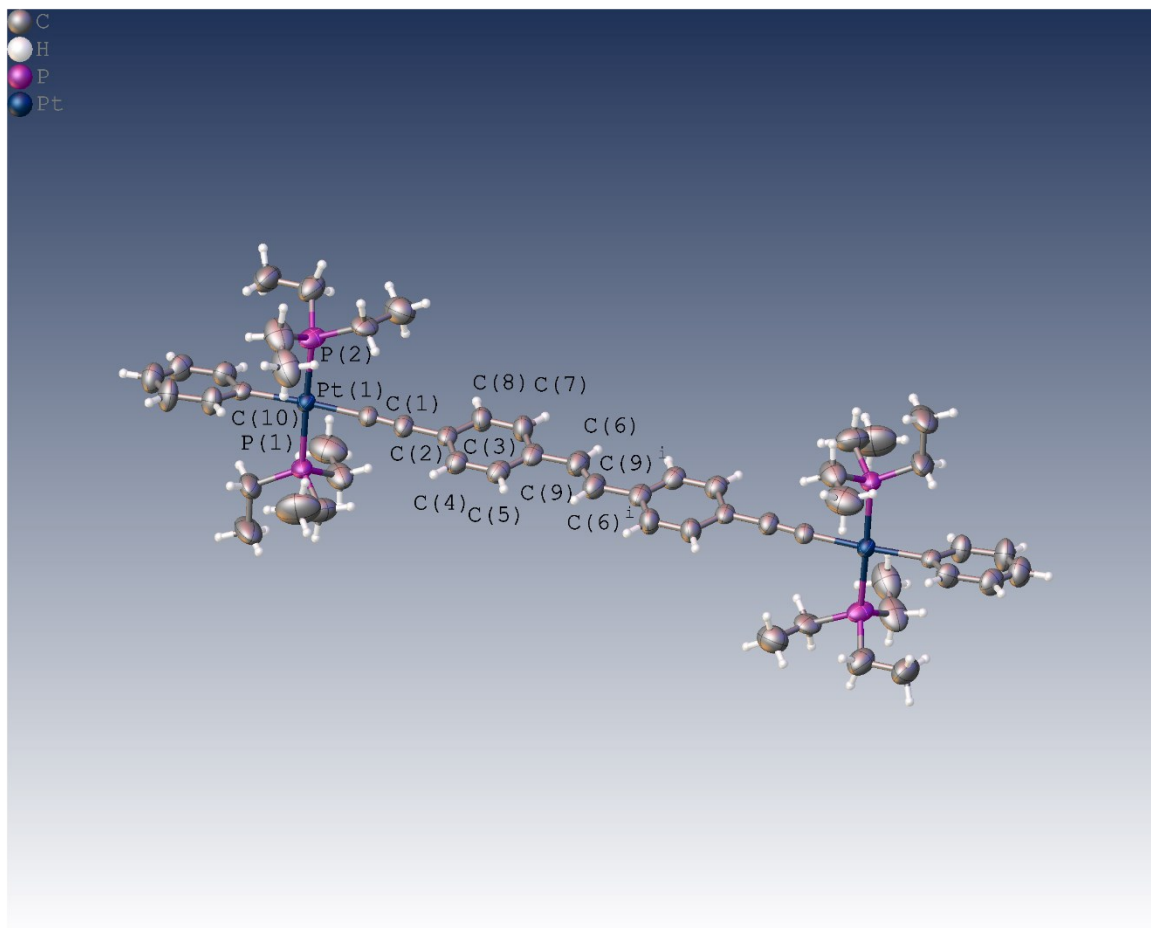


Table ST1. Crystal data and structure refinement for PtStil1.

CCDC ID number	2035894
Identification code	shelx
Empirical formula	C ₂₇ H ₄₀ P ₂ Pt
Formula weight	621.62
Temperature/K	296.15
Crystal system	monoclinic
Space group	P2 ₁ /c
a/Å	9.7774(5)
b/Å	9.2190(5)
c/Å	30.2215(16)
α /°	90
β /°	91.884(2)
γ /°	90
Volume/Å ³	2722.6(2)
Z	4

$\rho_{\text{calc}}/\text{cm}^3$ 1.517
 μ/mm^{-1} 5.282
 $F(000)$ 1240.0
 Crystal size/ mm^3 $0.2 \times 0.2 \times 0.2$
 Radiation $\text{MoK}\alpha$ ($\lambda = 0.71073$)
 2θ range for data collection/ $^\circ$ 5.992 to 50.998
 Index ranges $-11 \leq h \leq 11$, $-11 \leq k \leq 11$, $-36 \leq l \leq 36$
 Reflections collected 38985
 Independent reflections 5056 [Rint = 0.0501, Rsigma = 0.0278]
 Data/restraints/parameters 5056/285/344
 Goodness-of-fit on F^2 1.372
 Final R indexes [$I \geq 2\sigma(I)$] $R_1 = 0.0513$, $wR_2 = 0.0999$
 Final R indexes [all data] $R_1 = 0.0569$, $wR_2 = 0.1018$
 Largest diff. peak/hole / $e \text{ \AA}^{-3}$ 1.28/-2.88

Table ST2 Fractional Atomic Coordinates ($\times 10^4$) and Equivalent Isotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for shelx. Ueq is defined as 1/3 of the trace of the orthogonalised Uij tensor.

Atom	x	y	z	U(eq)
Pt1	3503.1(3)		3914.7(3)	6260.5(2) 41.22(14)
P1	5271(2)	2386(3)	6438.6(8)	51.6(6)
P2	1816(11)		5588(12)	6106(5) 61.3(8)
C1	4886(8)	5113(9)	5939(3)	45.6(18)
C2	5727(9)	5864(9)	5779(3)	53(2)
C3	6699(8)	6813(9)	5574(3)	47.2(15)
C4	7566(9)	7701(10)		5824(3) 56.6(18)
C5	8492(9)	8620(9)	5624(3)	58.4(18)
C6	8581(9)	8644(9)	5165(3)	58.5(18)
C7	7722(10)		7774(11)	4926(3) 65(2)
C8	6804(10)		6875(11)	5120(3) 60(2)
C9	9562(10)		9575(11)	4927(4) 72(3)
C10	2157(7)	2705(8)	6626(3)	41.1(15)
C11	1926(10)		3019(11)	7062(3) 63(2)
C12	1062(11)		2220(12)	7322(4) 79(3)
C13	362(12)	1067(12)		7146(4) 80(3)
C14	550(10)	716(10)	6721(4)	70(2)

C15	1431(10)	1510(10)	6460(3)	61(2)
C16	6702(12)	3404(15)	6714(5)	98(3)
C17	6289(17)	4260(20)	7103(6)	146(7)
C18	6103(12)	1651(13)	5953(4)	83(3)
C19	5201(15)	614(16)	5697(5)	116(5)
C20	4929(12)	869(13)	6796(4)	91(3)
C21	6136(14)	-169(16)	6903(5)	118(5)
C23	2190(20)	7129(19)	6873(5)	109(5)
C25	2480(40)	6960(40)	5284(7)	103(9)
C22	1194(19)	6409(16)	6594(5)	100(5)
C24	-530(30)	4060(40)	5895(10)	101(7)
C29	2320(17)	7236(15)	5802(5)	67(4)
C27	439(17)	5100(20)	5696(6)	86(5)
P2A	1858(15)	5626(16)	6092(6)	61.2(8)
C23A	2990(70)	7250(80)	6776(17)	109(19)
C25A	2860(50)	7390(60)	5365(12)	104(10)
C22A	2220(50)	7310(30)	6362(15)	107(13)
C24A	-400(40)	4130(50)	5731(13)	101(7)
C29A	1720(30)	6300(30)	5517(8)	99(8)
C27A	61(18)	5010(30)	6126(9)	86(7)

Table ST3 Anisotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for shelx. The Anisotropic displacement factor exponent takes the form: $-2\pi^2[h^2a^{*2}U_{11}+2hka^*b^*U_{12}+\dots]$.

Atom	U11	U22	U33	U23	U13	U12
Pt1	37.32(19)		40.06(19)		46.6(2)	3.53(15)
P1	41.3(11)	56.2(14)	57.8(13)	10.4(11)	6.6(10)	12.9(10)
P2	46.9(14)	39.8(14)	97(2)	0.5(14)	-1.9(14)	10.8(12)
C1	44(3)	40(4)	52(5)	5(3)	8(3)	7(3)
C2	50(3)	44(4)	67(5)	11(4)	12(3)	6(3)
C3	42(3)	37(3)	63(3)	11(3)	10(2)	11(2)
C4	51(4)	50(4)	70(4)	3(3)	9(3)	5(3)
C5	49(4)	44(4)	82(4)	3(3)	8(3)	5(3)
C6	50(4)	44(4)	83(4)	13(3)	13(3)	13(3)
C7	67(4)	62(5)	68(4)	15(3)	16(3)	-3(4)
C8	61(4)	56(5)	63(3)	11(3)	11(3)	-1(3)
C9	60(5)	57(5)	99(7)	19(5)	15(5)	4(4)

C10	35(3)	38(3)	51(3)	3(3)	4(3)	8(2)
C11	70(5)	63(5)	56(3)	-4(3)	19(3)	-9(4)
C12	90(6)	76(5)	72(5)	1(4)	36(4)	-12(5)
C13	82(6)	63(5)	96(5)	13(4)	24(4)	-8(4)
C14	68(5)	46(5)	96(5)	6(4)	7(4)	-10(4)
C15	64(5)	50(4)	69(4)	1(3)	-2(3)	-9(3)
C16	70(7)	97(7)	126(8)	2(6)	-21(6)	5(5)
C17	127(13)	162(15)	144(10)	-51(10)	-43(9)	9(11)
C18	85(7)	83(7)	83(6)	15(4)	22(5)	40(5)
C19	125(10)	105(8)	117(10)	-13(7)	-12(8)	48(7)
C20	77(6)	90(7)	106(8)	57(7)	19(6)	22(4)
C21	106(9)	125(11)	123(11)	69(9)	24(8)	53(8)
C23	160(15)	81(11)	89(10)	-2(8)	43(8)	30(9)
C25	100(20)	110(20)	91(8)	9(10)	-2(9)	19(15)
C22	141(13)	59(8)	103(7)	10(6)	36(7)	43(8)
C24	64(7)	103(9)	137(17)	33(11)	6(9)	-4(6)
C29	63(8)	47(7)	90(8)	16(7)	-6(7)	22(5)
C27	64(9)	75(10)	120(9)	11(8)	14(9)	5(7)
P2A	46.9(15)	39.8(14)	97(2)	0.6(14)	-1.7(14)	10.7(12)
C23A	90(20)	110(30)	129(19)	2(15)	6(14)	4(17)
C25A	84(16)	120(20)	110(18)	42(15)	-13(14)	-9(14)
C22A	100(20)	90(20)	135(18)	7(14)	1(19)	5(11)
C24A	64(8)	103(9)	137(17)	33(12)	6(9)	-4(6)
C29A	79(14)	107(19)	111(14)	38(10)	6(8)	6(14)
C27A	49(11)	83(13)	124(16)	43(11)	3(7)	19(7)

Table ST4 Bond Lengths for shelx.

Atom	Atom	Length/Å	Atom	Atom	Length/Å
Pt1	P1	2.281(2)	C9	C91	1.23(2)
Pt1	P2	2.295(8)	C10	C11	1.378(12)
Pt1	C1	2.019(8)	C10	C15	1.395(12)
Pt1	C10	2.070(8)	C11	C12	1.383(13)
Pt1	P2A	2.298(11)	C12	C13	1.362(15)
P1	C16	1.859(12)	C13	C14	1.345(15)
P1	C18	1.831(11)	C14	C15	1.393(13)
P1	C20	1.805(10)	C16	C17	1.484(19)

P2	C22	1.781(19)	C18	C19	1.499(18)
P2	C29	1.852(15)	C20	C21	1.546(14)
P2	C27	1.855(15)	C23	C22	1.43(2)
C1	C2	1.190(11)	C25	C29	1.60(2)
C2	C3	1.446(12)	C24	C27	1.49(2)
C3	C4	1.385(12)	P2A	C22A	1.78(2)
C3	C8	1.380(12)	P2A	C29A	1.846(16)
C4	C5	1.392(12)	P2A	C27A	1.852(17)
C5	C6	1.393(13)	C23A	C22A	1.44(3)
C6	C7	1.354(14)	C25A	C29A	1.57(3)
C6	C9	1.489(13)	C24A	C27A	1.50(3)
C7	C8	1.368(12)			

12-X,2-Y,1-Z

Table ST5 Bond Angles for shelx.

Atom	Atom	Atom	Angle/°	Atom	Atom	Atom	Angle/°
P1	Pt1	P2	175.8(3)	C7	C6	C5	117.7(9)
P1	Pt1	P2A	174.8(5)	C7	C6	C9	118.8(10)
C1	Pt1	P1	86.4(2)	C6	C7	C8	122.3(10)
C1	Pt1	P2	91.3(4)	C7	C8	C3	121.5(10)
C1	Pt1	C10	176.4(3)	C91	C9	C6	130.2(16)
C1	Pt1	P2A	89.6(5)	C11	C10	Pt1	121.7(6)
C10	Pt1	P1	91.8(2)	C11	C10	C15	114.3(8)
C10	Pt1	P2	90.3(4)	C15	C10	Pt1	124.0(6)
C10	Pt1	P2A	92.0(5)	C10	C11	C12	123.6(10)
C16	P1	Pt1	110.3(4)	C13	C12	C11	120.2(10)
C18	P1	Pt1	113.1(4)	C14	C13	C12	118.6(10)
C18	P1	C16	101.5(6)	C13	C14	C15	121.3(10)
C20	P1	Pt1	117.8(4)	C14	C15	C10	122.0(9)
C20	P1	C16	106.0(6)	C17	C16	P1	113.8(10)
C20	P1	C18	106.8(6)	C19	C18	P1	112.4(9)
C22	P2	Pt1	112.3(7)	C21	C20	P1	116.7(8)
C22	P2	C29	99.6(9)	C23	C22	P2	116.1(13)
C22	P2	C27	113.3(11)	C25	C29	P2	113.1(14)
C29	P2	Pt1	116.8(7)	C24	C27	P2	110.2(15)

C29	P2	C27	93.8(9)	C22A	P2A	Pt1	111.7(14)
C27	P2	Pt1	118.3(8)	C22A	P2A	C29A	98.4(16)
C2	C1	Pt1	175.3(8)	C22A	P2A	C27A	114.5(19)
C1	C2	C3	177.5(9)	C29A	P2A	Pt1	117.8(11)
C4	C3	C2	121.6(8)	C29A	P2A	C27A	96.7(12)
C8	C3	C2	121.4(9)	C27A	P2A	Pt1	115.9(10)
C8	C3	C4	117.0(8)	C23A	C22A	P2A	117(3)
C3	C4	C5	121.2(9)	C25A	C29A	P2A	117.4(19)
C4	C5	C6	120.3(9)	C24A	C27A	P2A	112.5(18)
C5	C6	C9	123.6(10)				

12-X,2-Y,1-Z

Table ST6 Torsion Angles for shelx.

A	B	C	D	Angle/°	A	B	C	D	Angle/°
Pt1	P1	C16	C17	53.7(13)	C11	C10	C15	C14	0.0(13)
Pt1	P1	C18	C19	-68.9(9)	C11	C12	C13	C14	1.2(18)
Pt1	P1	C20	C21	178.7(9)	C12	C13	C14	C15	-0.5(17)
Pt1	P2	C22	C23	57.5(15)	C13	C14	C15	C10	-0.1(16)
Pt1	P2	C29	C25	77.6(19)	C15	C10	C11	C12	0.7(14)
Pt1	P2	C27	C24	75(2)	C16	P1	C18	C19	172.9(9)
Pt1	C10	C11	C12	-178.4(8)	C16	P1	C20	C21	-57.4(13)
Pt1	C10	C15	C14	179.1(7)	C18	P1	C16	C17	173.9(12)
Pt1	P2A	C22A	C23A	-28(5)	C18	P1	C20	C21	50.2(13)
Pt1	P2A	C29A	C25A	-75(4)	C20	P1	C16	C17	-74.8(13)
Pt1	P2A	C27A	C24A	-78(3)	C20	P1	C18	C19	62.2(10)
C2	C3	C4	C5	-179.5(8)	C22	P2	C29	C25	-161.2(18)
C2	C3	C8	C7	179.9(8)	C22	P2	C27	C24	-59(2)
C3	C4	C5	C6	-1.2(13)	C29	P2	C22	C23	-66.9(14)
C4	C3	C8	C7	0.0(13)	C29	P2	C27	C24	-162(2)
C4	C5	C6	C7	1.4(13)	C27	P2	C22	C23	-165.3(13)
C4	C5	C6	C9	-178.8(8)	C27	P2	C29	C25	-46.8(18)
C5	C6	C7	C8	-1.0(14)	C22A	P2A	C29A	C25A	45(4)
C5	C6	C9	C91	1(2)	C22A	P2A	C27A	C24A	150(3)
C6	C7	C8	C3	0.3(15)	C29A	P2A	C22A	C23A	-153(5)
C7	C6	C9	C91	-179.0(15)	C29A	P2A	C27A	C24A	47(3)

C8	C3	C4	C5	0.4(12)	C27A	P2A	C22A	C23A	106(5)
C9	C6	C7	C8	179.2(9)	C27A	P2A	C29A	C25A	161(3)
C10	C11	C12	C13	-1.3(18)					

12-X,2-Y,1-Z

Table ST7 Hydrogen Atom Coordinates ($\text{\AA} \times 10^4$) and Isotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for shelx.

Atom	x	y	z	U(eq)
H4	7529.06	7683.18	6131.49	68
H5	9053.9	9220.35	5798.16	70
H7	7756.65	7786.89	4618.94	78
H8	6237.09	6292.73	4942.55	71
H9	9502.98	9502.6	4620.26	86
H11	2375.76	3811.59	7190.13	75
H12	959.02	2470.46	7616.85	94
H13	-233.14	535.31	7316.68	96
H14	83.23	-73.75	6597.61	84
H15	1537.69	1234.33	6167.17	73
H16A	7407.53	2721.52	6809.35	118
H16B	7096.38	4053.36	6500.45	118
H17A	5674.98	5022.15	7006.71	218
H17B	7086.52	4684.56	7245.68	218
H17C	5839.44	3643.56	7307.46	218
H18A	6353.69	2443.91	5761.07	100
H18B	6937.06	1154.89	6048.38	100
H19A	5690.27	226.4	5453.35	174
H19B	4399.42	1114.41	5586.18	174
H19C	4934.47	-163.52	5887.13	174
H20A	4603.87	1246.47	7072.9	109
H20B	4189.33	306.68	6659.61	109
H21A	6418.58	-624.08	6634.91	176
H21B	5852.35	-898.55	7106.96	176
H21C	6887.08	370.35	7032.47	176
H23A	2548.25	7950.06	6719.25	164
H23B	2921.43	6469.22	6947.85	164
H23C	1769.66	7449.42	7138.02	164

H25A	2685.78	7861.11	5140.89	154
H25B	1636.02	6578.96	5159.5	154
H25C	3202.95	6281.13	5240.3	154
H22A	494.32	7107.51	6505.89	120
H22B	757.7	5664.25	6766.56	120
H24A	-528.11	4198.63	6209.93	152
H24B	-261.44	3086.81	5829.86	152
H24C	-1438.61	4233	5773.28	152
H29A	1637.11	7983.46	5842.17	80
H29B	3182.97	7590.25	5926.76	80
H27A	-47.78	5966.3	5601	103
H27B	837.11	4660.13	5438.57	103
H23D	2632.65	6496.94	6958.6	164
H23E	3934.35	7046.74	6719.9	164
H23F	2926.35	8164.34	6925.1	164
H25D	2578.71	7807.65	5085.78	157
H25E	3704.1	6880.47	5335.79	157
H25F	2972.32	8143.44	5582.29	157
H22C	2713.08	7914.23	6160.21	129
H22D	1354.61	7782.9	6415.19	129
H24D	-730.71	4767.9	5500.44	152
H24E	-1119.44	3486.9	5813.45	152
H24F	356.57	3576.56	5627.79	152
H29C	1730.7	5465.41	5321.03	119
H29D	840.77	6766.29	5475.15	119
H27C	-30.67	4431.91	6391.16	103
H27D	-529.09	5850.64	6149.5	103

Table ST8 Atomic Occupancy for shelx.

Atom	Occupancy	Atom	Occupancy	Atom	Occupancy
P2	0.582(14)	C23	0.8	H23A	0.8
H23B	0.8	H23C	0.8	C25	0.582(14)
H25A	0.582(14)	H25B	0.582(14)	H25C	0.582(14)
C22	0.8	H22A	0.8	H22B	0.8
C24	0.582(14)	H24A	0.582(14)	H24B	0.582(14)
H24C	0.582(14)	C29	0.582(14)	H29A	0.582(14)

H29B	0.582(14)	C27	0.582(14)	H27A	0.582(14)
H27B	0.582(14)	P2A	0.418(14)	C23A	0.2
H23D	0.2	H23E	0.2	H23F	0.2
C25A	0.418(14)	H25D	0.418(14)	H25E	0.418(14)
H25F	0.418(14)	C22A	0.2	H22C	0.2
H22D	0.2	C24A	0.418(14)	H24D	0.418(14)
H24E	0.418(14)	H24F	0.418(14)	C29A	0.418(14)
H29C	0.418(14)	H29D	0.418(14)	C27A	0.418(14)
H27C	0.418(14)	H27D	0.418(14)		

Experimental

Single crystals of $(C_{27}H_{40}P_2Pt)_2$ were recrystallised from a hexane/ CH_2Cl_2 mixture. A suitable crystal was selected and glued on to a glass fibre on a diffractometer. The crystal was kept at 296 K during data collection. Using Olex2 [1] and WinGX [2], the structure was solved with the by direct methods using WinGX [2] and refined by full-matrix least-squares (SHELXL-2014) on F^2 [3].

1. O.V. Dolomanov, et al., J Appl Cryst, 2009, 42, 339-341
2. L. Farrugia, J. Appl. Crystallogr, 1999, 32, 837-838.
3. G. M. Sheldrick, Acta Crystallographica a-Foundation and Advances, 2015, 71, 3-8.

Crystal structure determination of $(C_{27}H_{40}P_2Pt)_2$

Crystal Data for $C_{27}H_{40}P_2Pt$ ($M=621.62$ g/mol): monoclinic, space group $P2_1/c$ (no. 14), $a = 9.7774(5)$ Å, $b = 9.2190(5)$ Å, $c = 30.2215(16)$ Å, $\beta = 91.884(2)^\circ$, $V = 2722.6(2)$ Å³, $Z = 4$, $T = 296.15$ K, $\mu(MoK\alpha) = 5.282$ mm⁻¹, $D_{calc} = 1.517$ g/cm³, 38985 reflections measured ($5.992^\circ \leq 2\theta \leq 50.998^\circ$), 5056 unique ($R_{int} = 0.0501$, $R_{sigma} = 0.0278$) which were used in all calculations. The final R_1 was 0.0513 ($I > 2\sigma(I)$) and wR_2 was 0.1018 (all data).

Refinement model description

Number of restraints - 285.

Details:

1. Fixed Uiso
At 1.2 times of:
All C(H) groups, All C(H,H) groups
At 1.5 times of:
All C(H,H,H) groups
2. Restrained distances
P2-C27 $\approx$ P2-C29
with sigma of 0.02
Pt1-P2 $\approx$ Pt1-P2A
with sigma of 0.02
P2-C22 $\approx$ P2A-C22A
with sigma of 0.02
P2-C29 $\approx$ P2A-C29A
with sigma of 0.02
P2-C27 $\approx$ P2A-C27A
with sigma of 0.02
C29-C27 $\approx$ C29A-C27A
with sigma of 0.04
C23-C22 $\approx$ C23A-C22A
with sigma of 0.02
C22-C27 $\approx$ C22A-C27A
with sigma of 0.04
C25-C29 $\approx$ C25A-C29A
with sigma of 0.02
C24-C27 $\approx$ C24A-C27A

with sigma of 0.02
 C22-C29 $\approx$ C22A-C29A
 with sigma of 0.04
 Pt1-C22 $\approx$ Pt1-C22A
 with sigma of 0.04
 Pt1-C29 $\approx$ Pt1-C29A
 with sigma of 0.04
 Pt1-C27 $\approx$ Pt1-C27A
 with sigma of 0.04
 P2-C23 $\approx$ P2A-C23A
 with sigma of 0.04
 P2-C25 $\approx$ P2A-C25A
 with sigma of 0.04
 P2-C24 $\approx$ P2A-C24A
 with sigma of 0.04
 3. Uiso/Uanis restraints and constraints
 C24A $\approx$ C24: within 2A with sigma of 0.001 and sigma for terminal atoms of 0.002 within 2A
 C25 $\approx$ C25A: within 2A with sigma of 0.01 and sigma for terminal atoms of 0.02 within 2A
 P2A $\approx$ P2: within 2A with sigma of 0.001 and sigma for terminal atoms of 0.002 within 2A
 Uanis(C23A) $\approx$ Ueq: with sigma of 0.01 and sigma for terminal atoms of 0.02
 4. Rigid body (RIGU) restrains
 Pt1, P1, P2, C1, C10, P2A, C16, C18, C20, C22, C29, C27, C2, C11, C15, C22A, C29A, C27A, C17, C19, C21, C23, C25, C24, C3, C12, C14, C23A, C25A, C24A, C4, C8, C13, C5, C7, C6, C9
 with sigma for 1-2 distances of 0.001 and sigma for 1-3 distances of 0.001
 5. Others
 Sof (P2A)=Sof (C25A)=Sof (H25D)=Sof (H25E)=Sof (H25F)=Sof (C24A)=Sof (H24D)=
 Sof (H24E)=Sof (H24F)=Sof (C29A)=Sof (H29C)=Sof (H29D)=Sof (C27A)=Sof (H27C)=
 Sof (H27D)=1-FVAR(1)
 Sof (P2)=Sof (C25)=Sof (H25A)=Sof (H25B)=Sof (H25C)=Sof (C24)=Sof (H24A)=Sof (H24B)=
 Sof (H24C)=Sof (C29)=Sof (H29A)=Sof (H29B)=Sof (C27)=Sof (H27A)=Sof (H27B)=FVAR(1)
 Fixed Sof: C23(0.8) H23A(0.8) H23B(0.8) H23C(0.8) C22(0.8) H22A(0.8)
 H22B(0.8) C23A(0.2) H23D(0.2) H23E(0.2) H23F(0.2) C22A(0.2) H22C(0.2) H22D(0.2)
 6.a Riding coordinates:
 C23A(H23D,H23E,H23F)
 6.b Secondary CH2 refined with riding coordinates:
 C16(H16A,H16B), C18(H18A,H18B), C20(H20A,H20B), C22(H22A,H22B), C29(H29A,H29B), C27(H27A,H27B), C22A(H22C,H22D), C29A(H29C,H29D), C27A(H27C,H27D)
 6.c Aromatic/amide H refined with riding coordinates:
 C4(H4), C5(H5), C7(H7), C8(H8), C9(H9), C11(H11), C12(H12), C13(H13), C14(H14), C15(H15)
 6.d Idealised Me refined as rotating group:
 C17(H17A,H17B,H17C), C19(H19A,H19B,H19C), C21(H21A,H21B,H21C), C23(H23A,H23B,H23C), C25(H25A,H25B,H25C), C24(H24A,H24B,H24C), C25A(H25D,H25E,H25F), C24A(H24D,H24E,H24F)

Table ST9. The quantitative analysis of reduction and enhancement in A and B bands, respectively for **1a**.

Irradiation time (min)	A-band ($\epsilon \times 10^5$ $M^{-1}.cm^{-1}$) (λ (nm))	B-band ($\epsilon \times 10^5$ $M^{-1}.cm^{-1}$) (λ (nm))	% reduction in A- band	% enhancement in B-band
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0 (non-exposed)	1.65 (334)	0.50 (257)	-	-
10	1.25 (333)	0.83 (263)	24	66
20	1.06 (320)	1.10 (263)	36	120
25	0.90 (319)	1.32 (263)	45	164

Table ST10. The quantitative analysis of reduction and enhancement in A and B bands, respectively for **2a**.

Irradiation time (min)	A-band ($\epsilon \times 10^5$ M ⁻¹ .cm ⁻¹) (λ)	B-band ($\epsilon \times 10^5$ M ⁻¹ .cm ⁻¹)(λ)	% reduction in A-band	% enhancement in B-band
0 (non-exposed)	0.72 (328)	0.17 (263)	-	-
10	0.46 (317)	0.28 (262)	36	64
20	0.39 (316)	0.40 (262)	45	135
25	0.31 (315)	0.49 (262)	56	188

Table ST11. The quantitative analysis of reduction and enhancement in A and B bands, respectively for **1M**.

Irradiation time (min)	A-band ($\epsilon \times 10^5$ M ⁻¹ .cm ⁻¹)	B-band ($\epsilon \times 10^5$ M ⁻¹ .cm ⁻¹)	% reduction in A-band	% enhancement in B-band
0 (non-exposed)	1.07 (380)	0.38 (292)	-	-
10	0.90 (372)	1.21 (280)	15	218
20	0.83 (372)	1.22 (280)	22	221
25	0.72 (372)	1.28 (280)	32	280

Table ST12. The quantitative analysis of reduction and enhancement in A and B bands, respectively for **2M**.

Irradiation time (min)	A-band ($\epsilon \times 10^5$ M ⁻¹ .cm ⁻¹)	B-band ($\epsilon \times 10^5$ M ⁻¹ .cm ⁻¹)	% reduction in A-band	% enhancement in B-band
0 (non-exposed)	2.35 (376)	1.00 (225)	-	-
10	2.19 (366)	1.12 (224)	6	12
20	2.17 (353)	1.49 (226)	7	49
25	2.14 (348)	1.53 (225)	9	53

Table ST13. The quantitative analysis of reduction and enhancement in A and B bands, respectively for **1P**.

Irradiation time (min)	A-band ($\epsilon \times 10^5$ M ⁻¹ .cm ⁻¹)	B-band ($\epsilon \times 10^5$ M ⁻¹ .cm ⁻¹)	% reduction in A-band	% enhancement in B-band
0 (non-exposed)	2.62 (413)	0.95 (230)	-	-
10	2.54 (378)	2.43 (231)	3	155
20	2.09 (352)	2.52 (231)	20	165
25	1.81 (342)	2.61 (231)	30	174

Table ST14. Electronic transitions in the *cis*- and *trans*-stilbene isomers of the Pt(II) complexes **1M** and **2M** in gas phase.

System (Gas phase)	E[eV] (nm)	<i>f</i>	Assignment
Trans-1M	3.09(401)	2.70	HOMO--> LUMO +HOMO-1---> LUMO+1
	4.53(274)	0.13	HOMO--> LUMO
	5.12(242)	0.092	HOMO--> LUMO
	5.82(213)	0.15	HOMO--> LUMO
	3.20(388)	2.25	HOMO--> LUMO +HOMO-1---> LUMO+1
Trans-2M	3.51(353)	1.00	HOMO--> LUMO +HOMO-1---> LUMO+1
	4.29(289)	0.44	HOMO--> LUMO
	4.98(249)	0.023	HOMO--> LUMO
	5.90(210)	0.085	HOMO--> LUMO
	3.21(386)	1.13	HOMO--> LUMO +HOMO-1---> LUMO+1
Cis-1M	3.79(327)	0.16	HOMO--> LUMO +HOMO-1---> LUMO+1
	4.59(270)	0.49	HOMO--> LUMO
	4.88(254)	0.38	HOMO--> LUMO

	5.82(213)	0.050	HOMO--> LUMO
Cis-2M	3.35(370)	1.37	HOMO--> LUMO +HOMO-1---> LUMO+1
	4.66(266)	0.25	HOMO--> LUMO
	5.46(227)	0.058	HOMO--> LUMO
	6.14(202)	0.20	HOMO--> LUMO

Table ST15. Electronic transitions in the *cis*- and *trans*-stilbene isomers of the Pt(II) complexes **1M** and **2M** in PCM.

System (PCM)	E[eV] (nm)	<i>f</i>	Assignment
Trans-1M	2.99(415)	2.72	HOMO--> LUMO +(HOMO-1---> LUMO+1)
	4.49(270)	0.19	HOMO--> LUMO
	4.94(251)	0.093	HOMO--> LUMO
	5.79(214)	0.35	HOMO--> LUMO
	3.15(394)	2.84	HOMO--> LUMO +(HOMO-1---> LUMO+1)
Trans-2M	3.49(334)	0.77	HOMO--> LUMO +(HOMO-1---> LUMO+1)
	4.29(290)	0.37	HOMO--> LUMO
	5.04(246)	0.069	HOMO--> LUMO
	6.01(204)	0.24	HOMO--> LUMO
	3.14(395)	1.20	HOMO--> LUMO +HOMO-1---> LUMO+1
Cis-1M	3.79(327)	0.25	HOMO--> LUMO + HOMO-1---> LUMO+1
	4.41(281)	0.33	HOMO--> LUMO
	4.64(267)	0.27	HOMO--> LUMO
	5.79(214)	0.25	HOMO--> LUMO

Cis-2M	3.35(374)	1.66	MLCT + $\pi \rightarrow \pi^*$
	4.66(263)	0.37	HOMO $\rightarrow$ LUMO
	6.14(202)	0.25	HOMO $\rightarrow$ LUMO