

Electronic Supplementary Information

Air Stable Organometallic Polymer Containing Titanafluorene Moieties by Sonogashira-Hagihara Cross-coupling Polycondensation

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reactions; Organometallic polymers; Titanafluorenes

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1. NMR Spectra

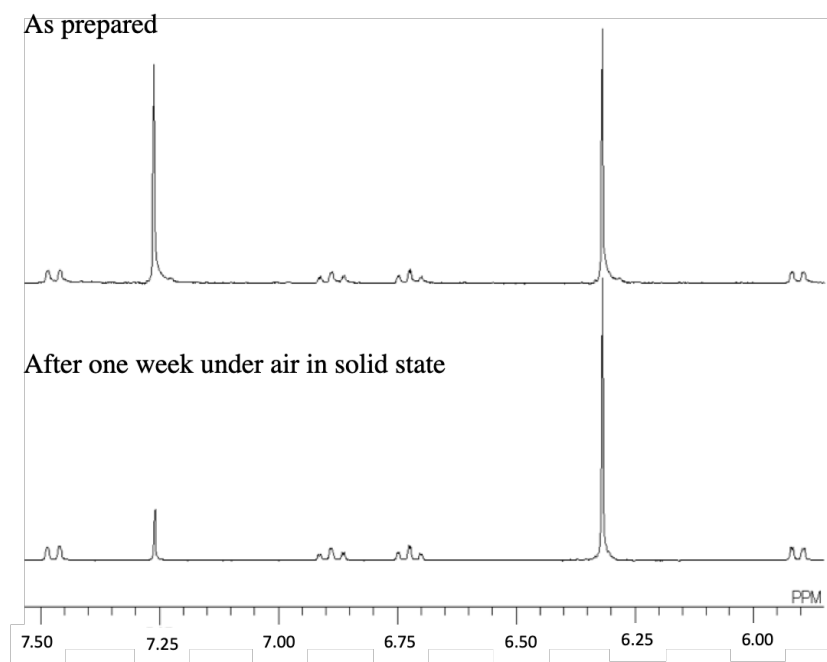


Figure S1. ^1H NMR spectra of a titanafluorene derivative (**2**) before and after kept under air at ambient temperature for 1 week.

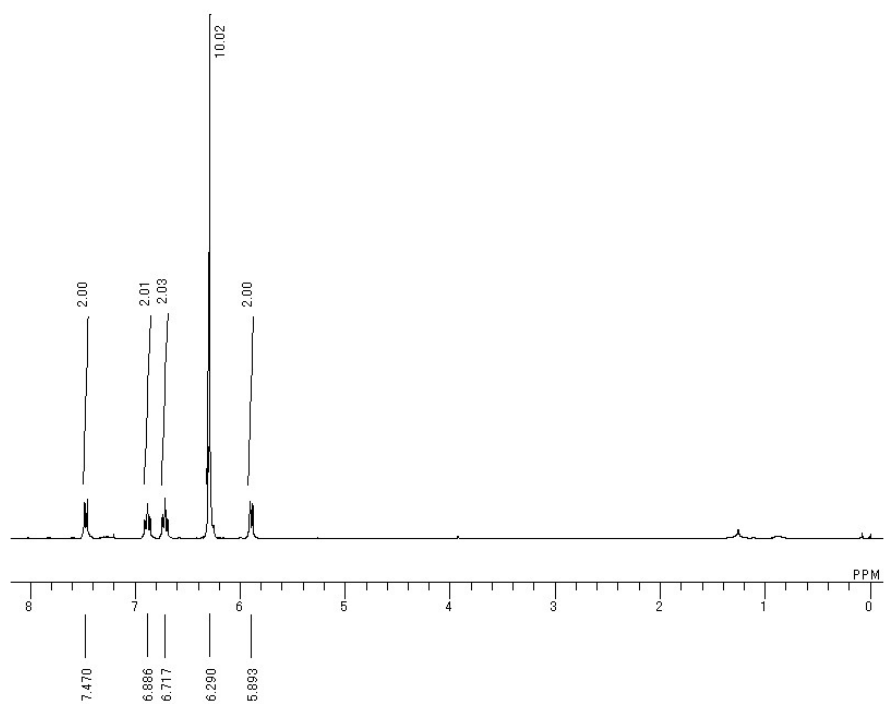


Figure S2. ^1H NMR spectrum of **2**.

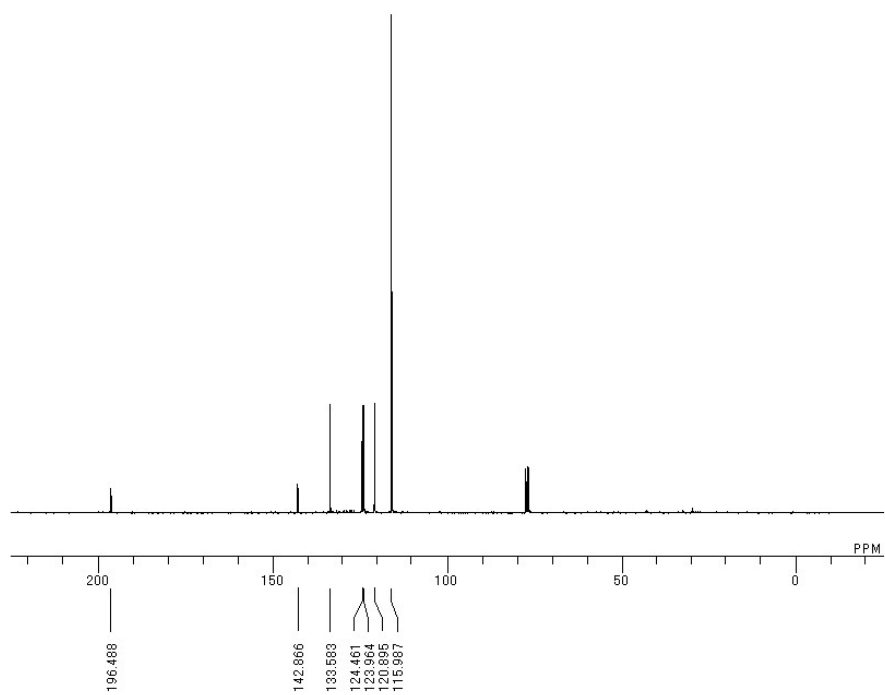


Figure S3. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of 2.

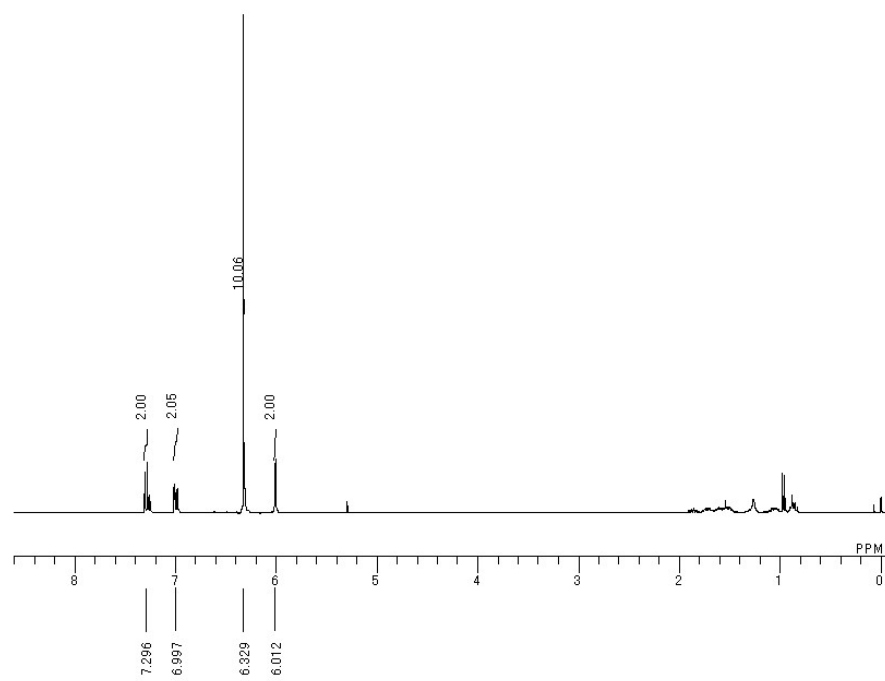


Figure S4. ^1H NMR spectrum of 4.

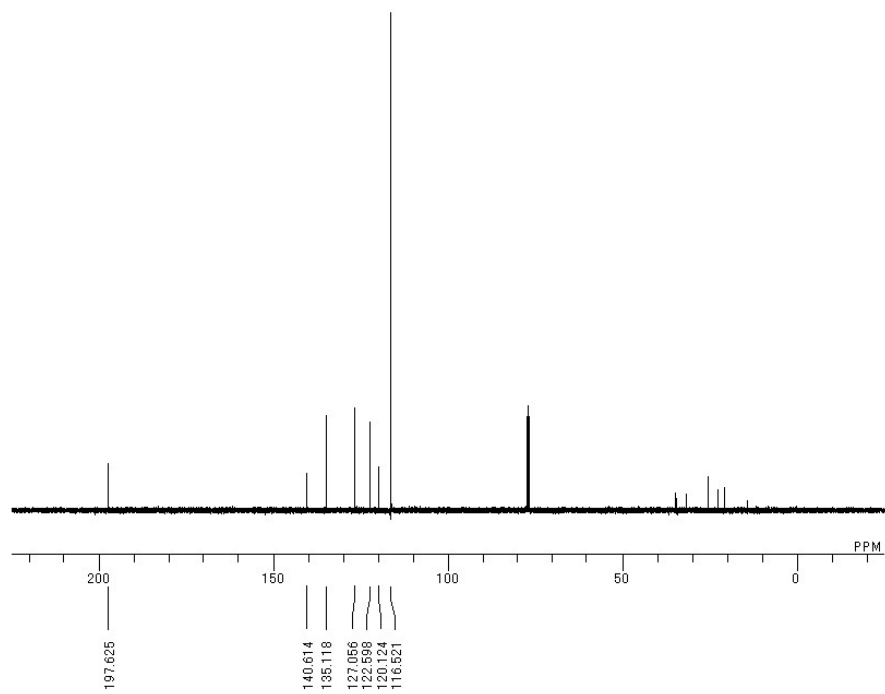


Figure S5. $^{13}\text{C}\{^1\text{H}\}$ NMR of spectrum of **4**.

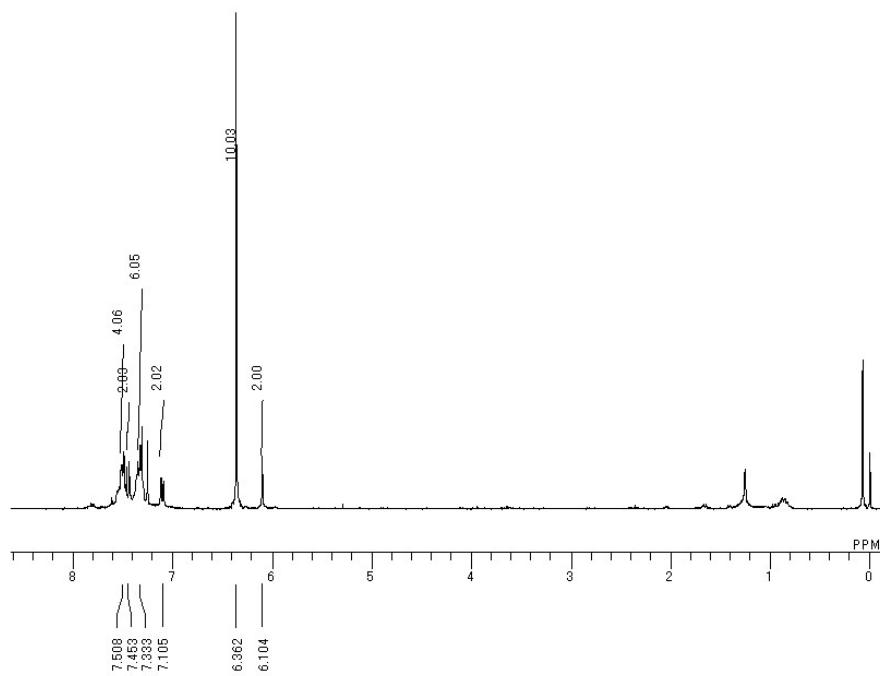


Figure S6. ^1H NMR of spectrum of **6**.

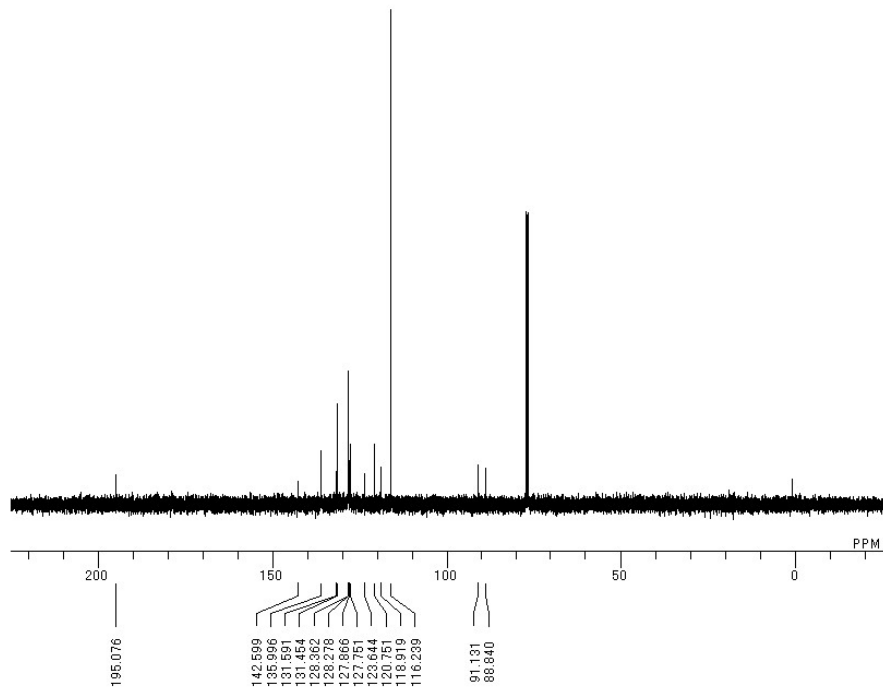


Figure S7. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of **6**.

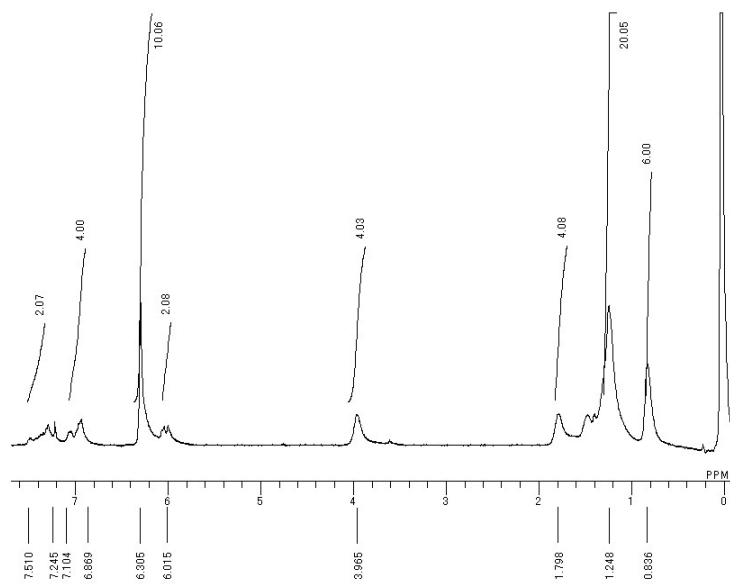


Figure S8. ^1H NMR spectrum of **8**.

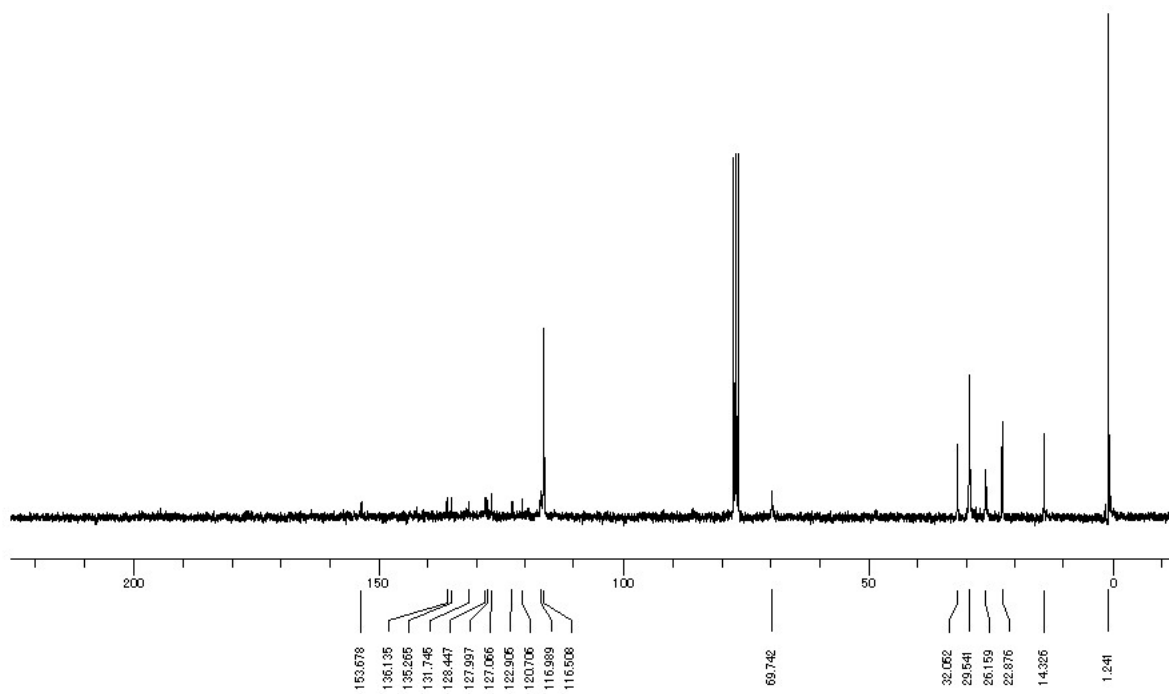


Figure S9. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of **8**.

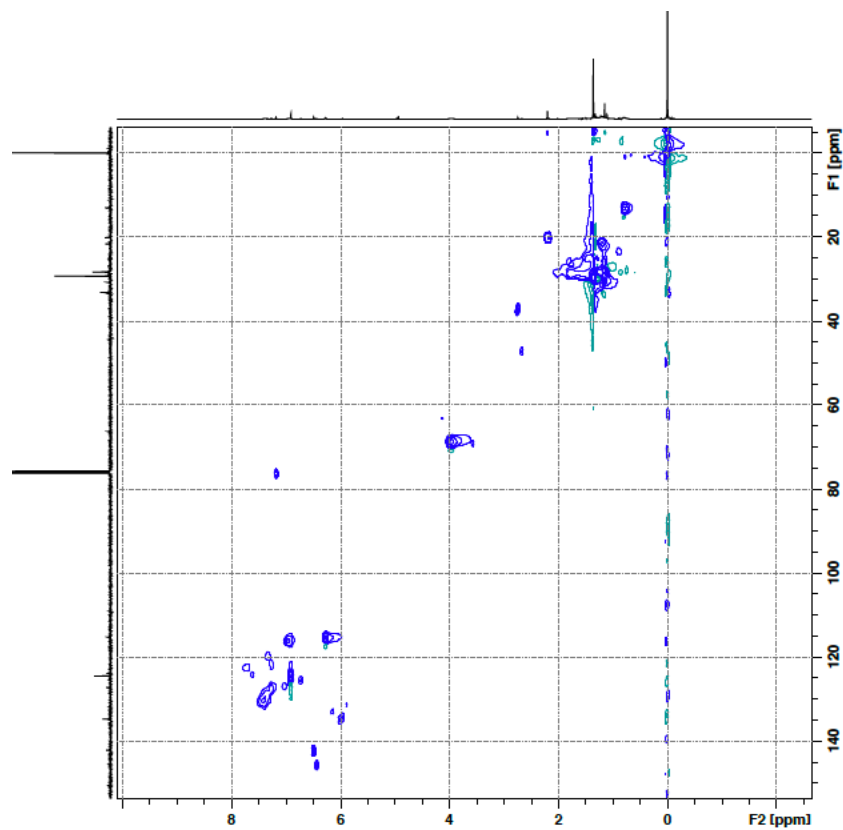


Figure S10. 2D ^1H - ^{13}C correlation NMR spectrum of **8**.

2. IR Spectra

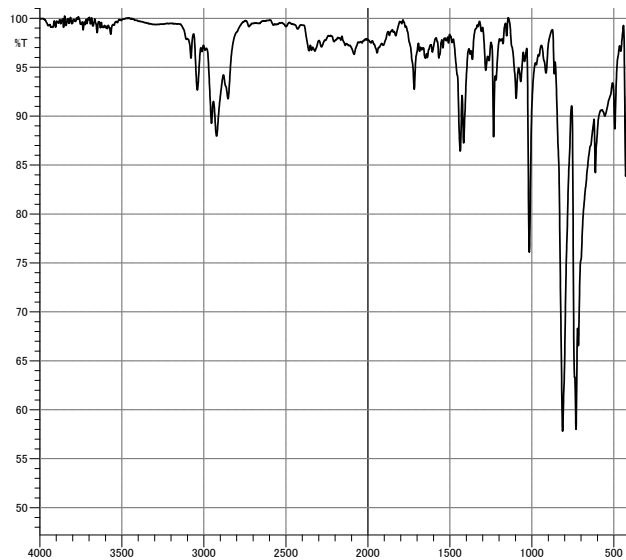


Figure S11. IR spectrum of **2**.

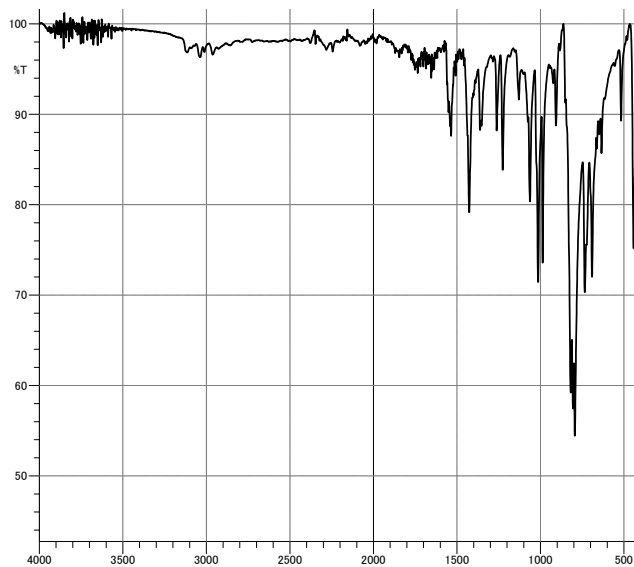


Figure S12. IR spectrum of **4**.

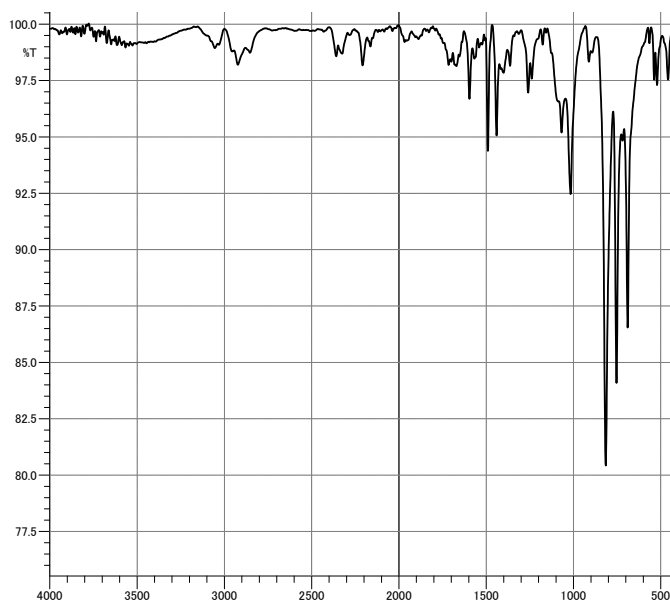


Figure S13. IR spectrum of **6**.

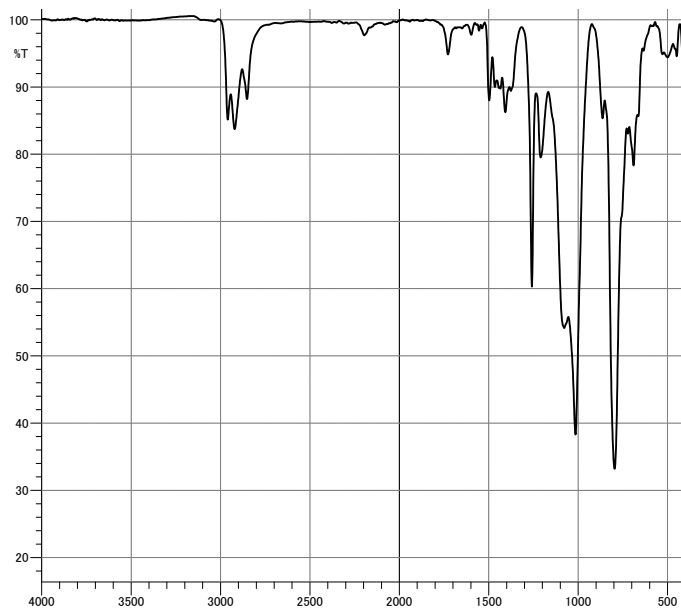


Figure S14. IR spectrum of **8**.

3. Mass Spectra

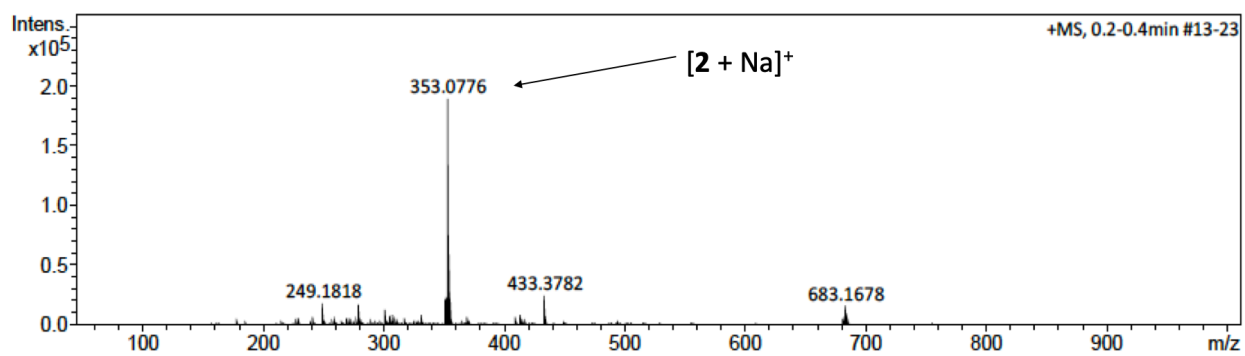


Figure S15. ESI-TOF mass spectrum of **2**

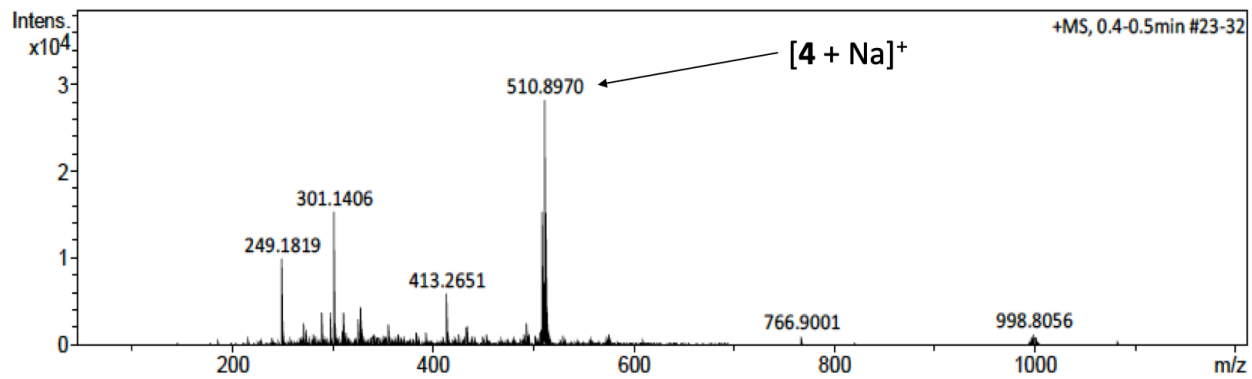


Figure S16. ESI-TOF mass spectrum of 4

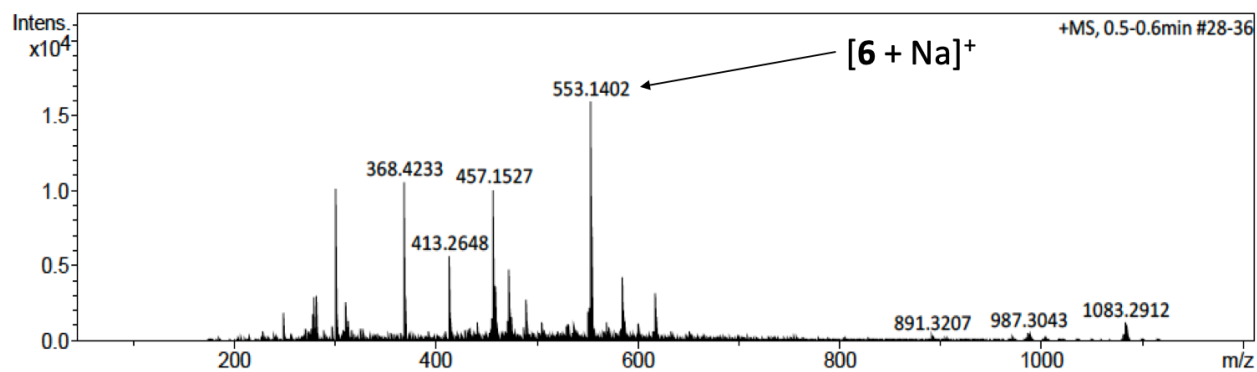
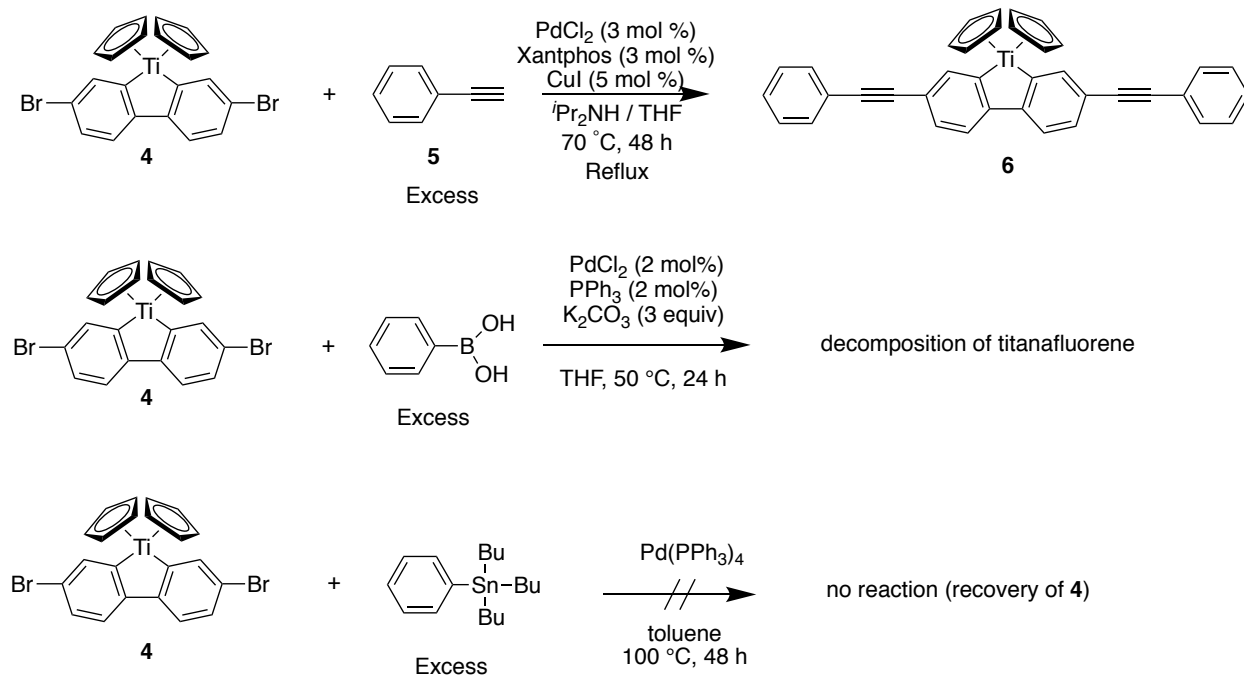


Figure S17. ESI-TOF mass spectrum of 6

3. Synthetic Attempts of π -Extended Titanafluorene Derivatives



Scheme S1. Coupling reactions of arylene dihalide containing titanafluorene unit (4).

4. Crystallographic Data

Table S1. Crystallographic data of $[\text{C}_{22}\text{H}_{16}\text{TiBr}_2]_2 \cdot \text{CH}_2\text{Cl}_2$

Crystal data	$[\text{C}_{22}\text{H}_{16}\text{TiBr}_2]_2 \cdot \text{CH}_2\text{Cl}_2$
CCDC	2027344
Empirical formula	$\text{C}_{45}\text{H}_{34}\text{Br}_4\text{Cl}_2\text{Ti}_2$
Formula weight	1061.06
Crystal size/ mm^3	0.308 x 0.106 x 0.07
Crystal system	monoclinic
Space group	P2_1
a, Å	9.2921 (2)
b, Å	14.6419 (4)
c, Å	14.6728 (3)
α , deg	90
β , deg	99.445 (2)
γ , deg	90
Volume, Å ³	1969.23 (8)
Density _{calcd} , g cm^{-3}	1.789
Z	2
F(000)	1044.0
Temperature, deg	-183.15
2 θ _{max} , deg	149.374
T _{min} /T _{max}	0.251/1.000
Absorption correction	Multi-scan
No. of Reflection	7706
No. of Parameters	478
Goodness-of-fit on F ²	1.059
Final R indexes [$I \geq 2\sigma(I)$]	R1 = 0.0283, wR2 = 0.0734
Final R indexes [all data]	R1 = 0.0287, wR2 = 0.0737
Largest diff. peak/hole /	1.18 and -0.64 $\text{e} \text{Å}^{-3}$

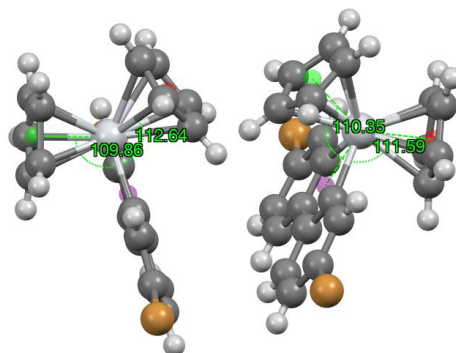


Fig S18. Angle of Cp(centroid) to fluorene(centroid)

5. Size Exclusion Chromatographic Analysis

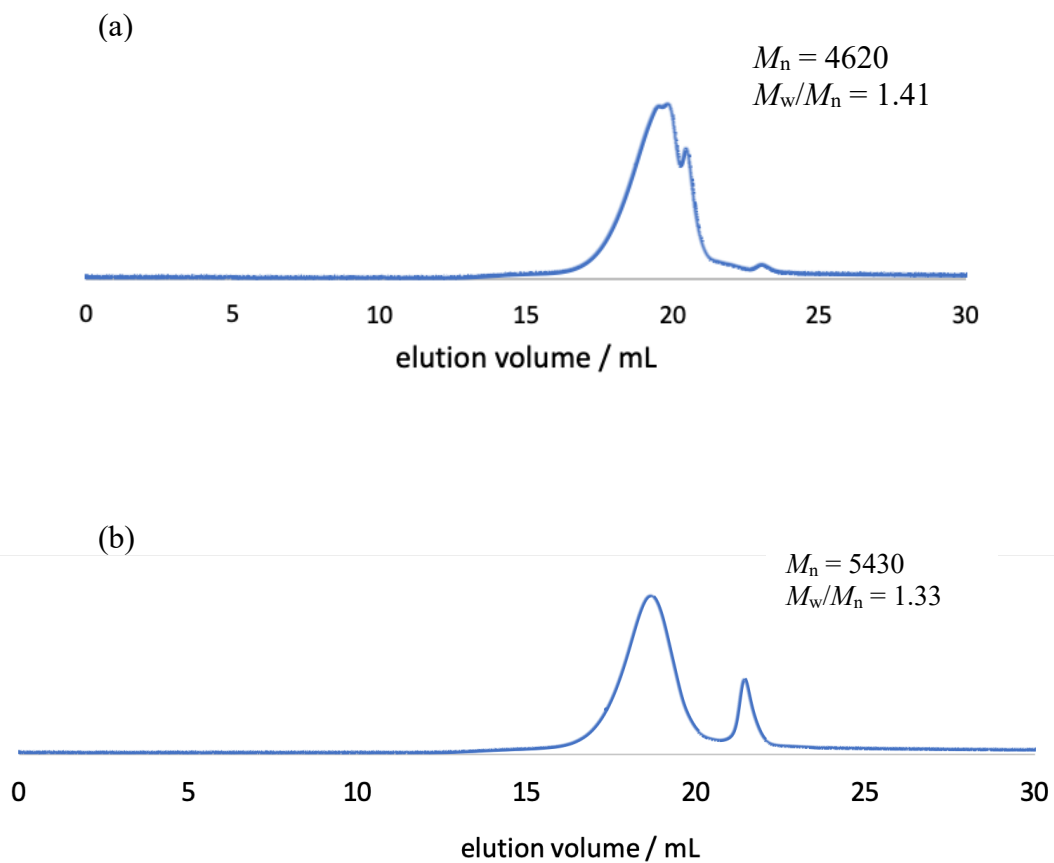


Figure S19. SEC profiles of **8** after isolation by precipitation into ethanol (a) and by precipitation into ethanol/ethyl acetate (v/v = 9/1).

6. DFT and TD-DFT Optimized Molecular Diagrams

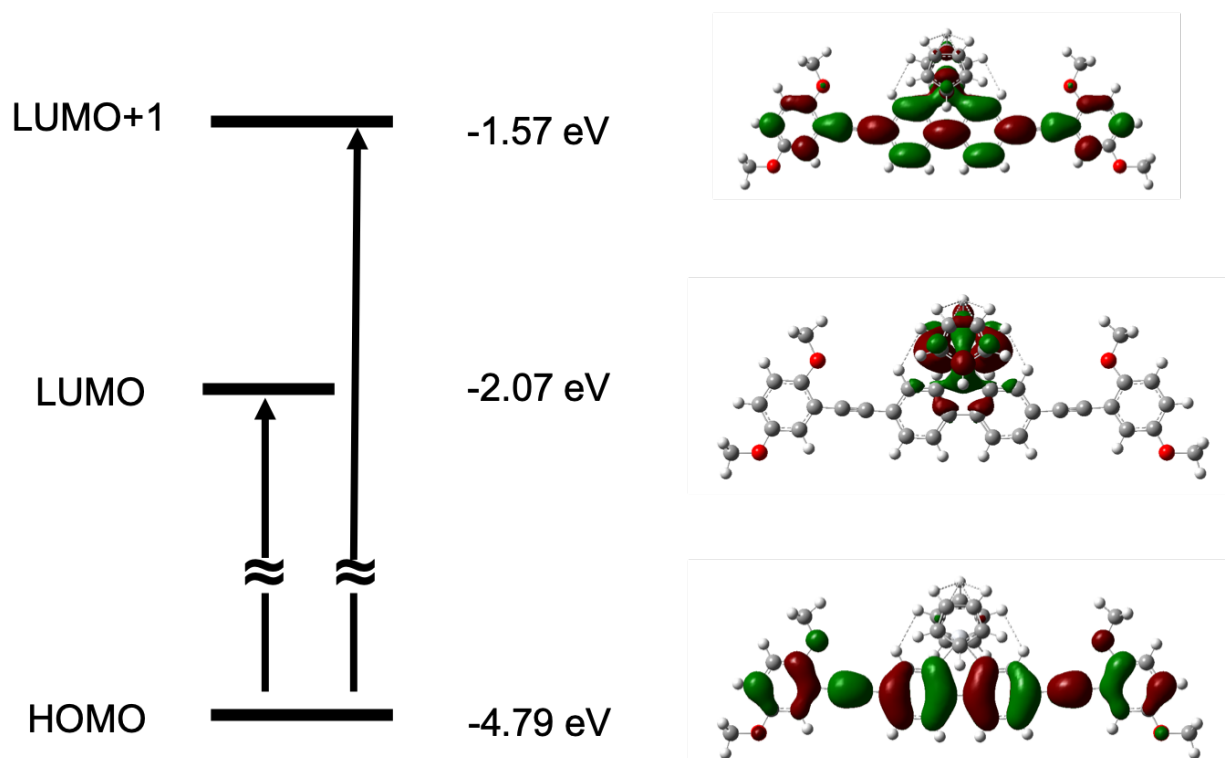


Fig S20. Energy profiles and molecular orbital diagrams of HOMO, LUMO, and LUMO+1 of **9**.

Table S2. HOMO to LUMO and HOMO to LUMO+1 transitions by TD-DFT calculations.

Assignment	HOMO to LUMO	HOMO to LUMO+1
λ_{\max} (nm) ^{a)}	596.86	416.03
f ^{b)}	0.002	0.688
Excitation energy (eV)	2.08	2.98

a) Calculated absorption maximum in the UV-vis spectrum. b) Oscillator strength.