

SUPPORTING INFORMATION

The unexpected reactivity of 9-iodo-*nido*-carborane: from nucleophilic substitution reactions to the synthesis of tricobalt tris(dicarbollide) Na[4,4',4''- (MeOCH₂CH₂O)₃-3,3',3''-Co₃(μ³-O)(μ³-S)(1,2-C₂B₉H₁₀)

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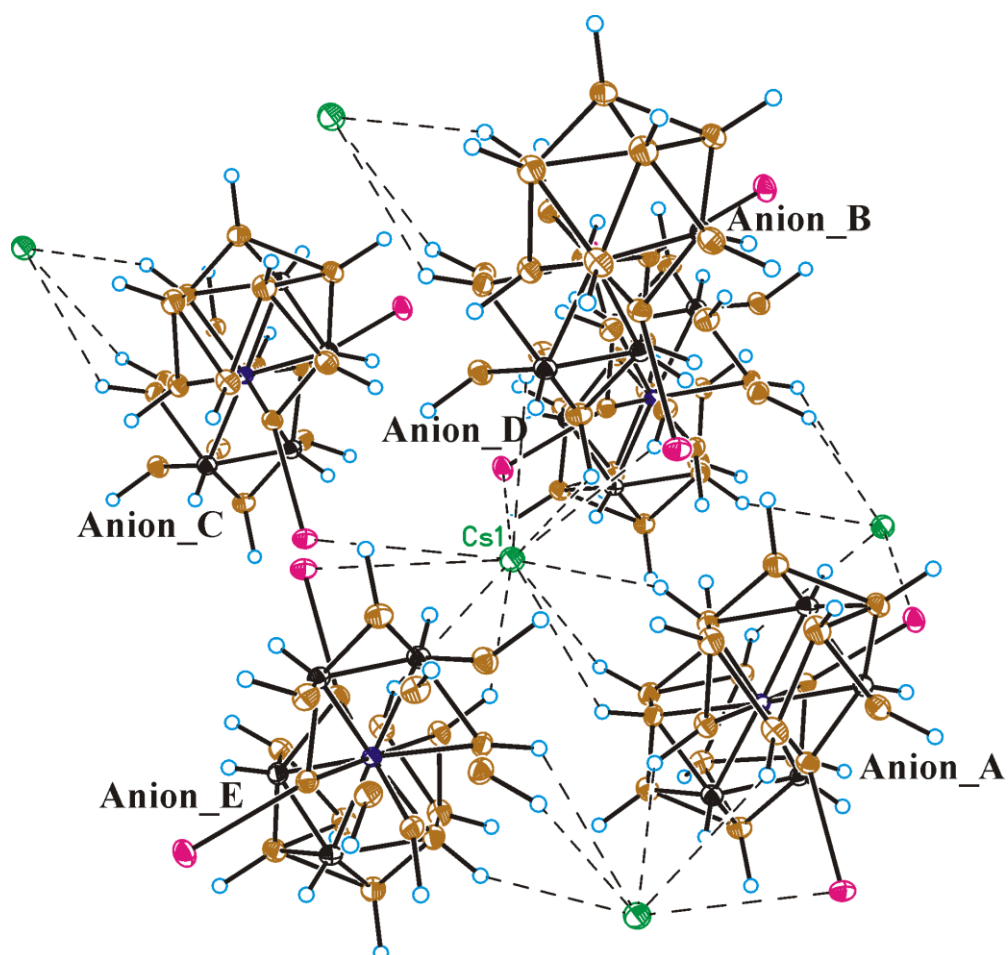


Figure 1S. Crystal structure fragment of Si18F showing close contacts formed by Cs atom which links 5 dicarbollide anions

Cs...Anion_A : Cs1...H7, 3.03(2)Å; Cs1...H7', 3.08(2)Å; Cs1...H12', 3.42(2)Å

Cs...Anion_B : Cs1...H6', 3.25(2)Å; Cs1...H10', 3.21(2)Å

Cs...Anion_C : Cs1...I1, 4.0869(11)Å

Cs...Anion_D : Cs1...I2, 3.7708(12)Å; Cs1...H9', 3.39(2)Å

Cs...Anion_E : Cs1...I2, 4.1904(12)Å; Cs1...H8, 3.24(2)Å; Cs1...H9, 3.04(2)Å

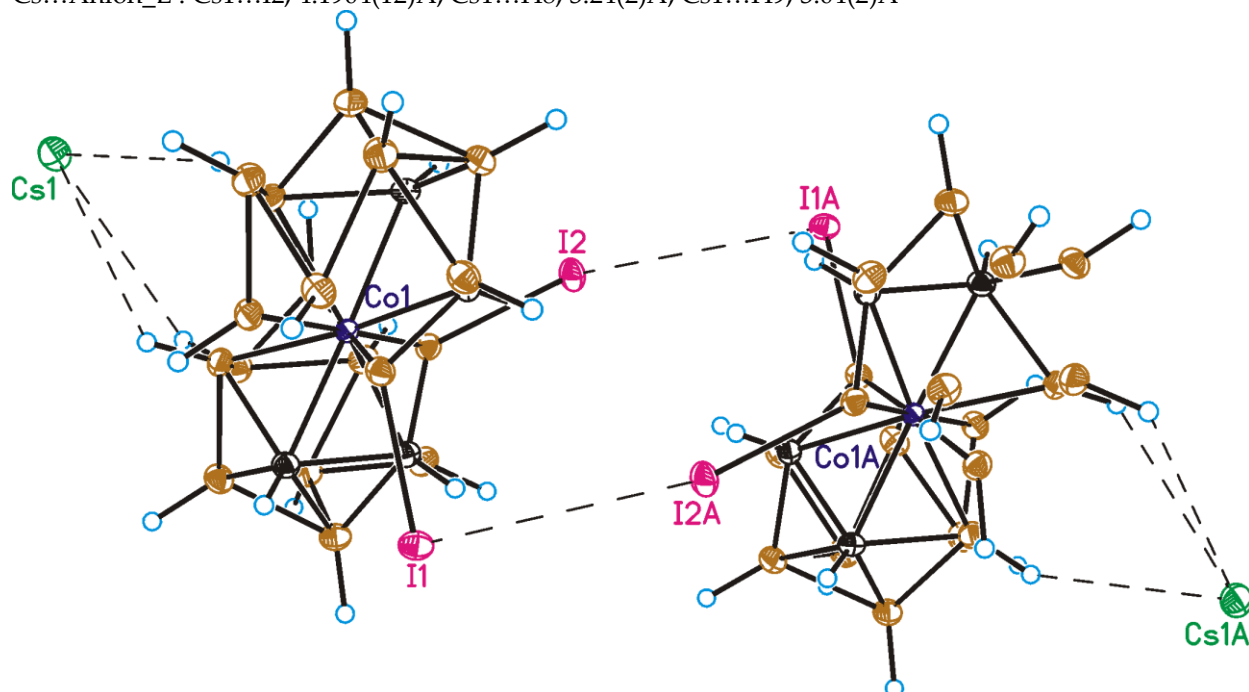


Figure 2S. Centrosymmetric dimer formed by I...I shortened contacts (I1...I2A, 4.0803(7)Å)

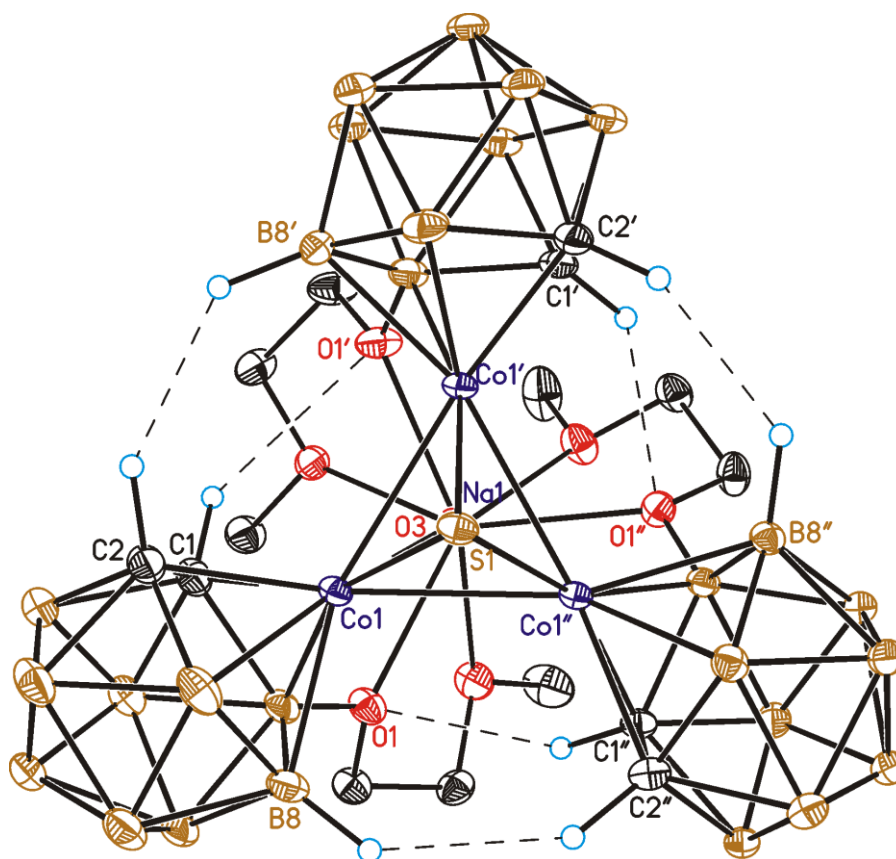


Figure 3S. Intramolecular noncovalent bonding in Na[12].

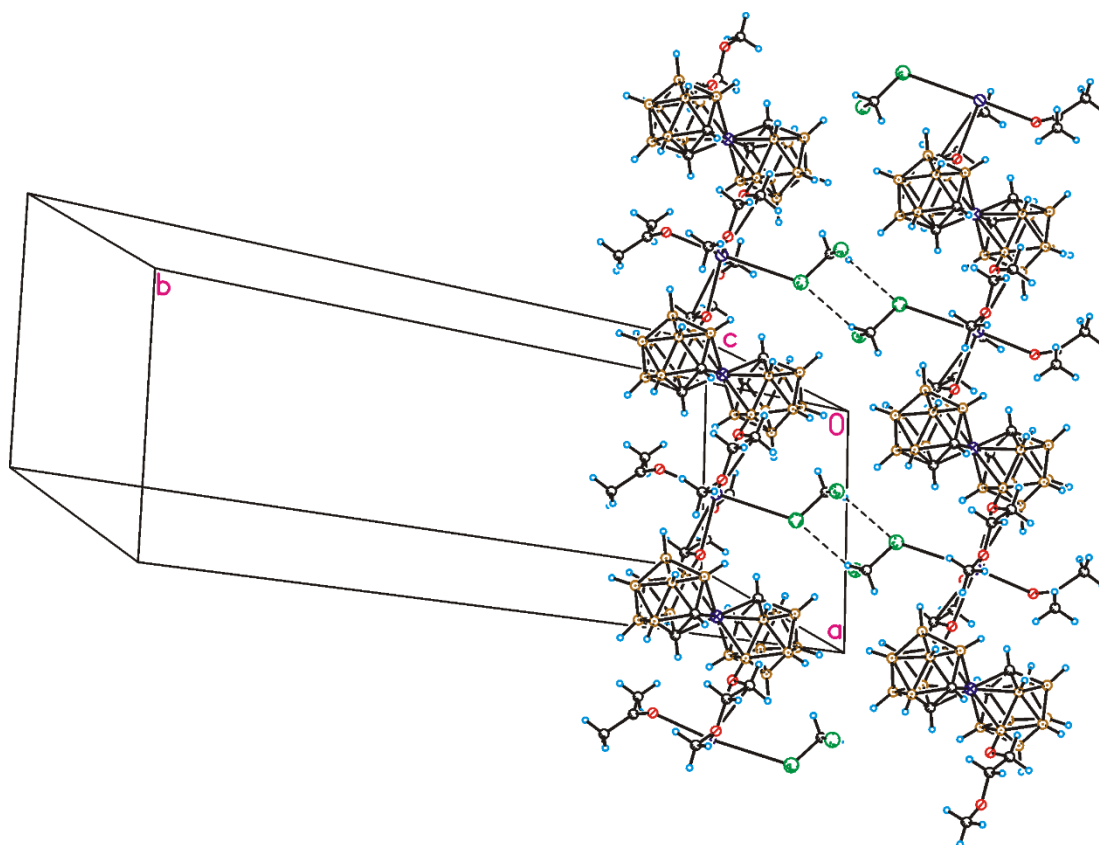


Figure 4S. Interchain bonding in crystal structure of $\{K(Me_2CO)(CH_2Cl_2)[14]\}_n$.

Table 1S. Selected characteristics of all optimized (DFT/BP86/cc-pvDZ) isomers of diiodo cobalt bis(dicarbollide), and key properties for all found intercage bond critical points (BCP').

	4-4'- isomer					4-7'- isomer			8-8' isomer		
rotamer	<i>cisoid</i>	<i>gauche</i>	<i>transoid</i>	<i>gauche-2</i>	<i>cisoid-2</i>	<i>cisoid</i>	<i>gauche</i>	<i>transoid</i>	<i>cisoid</i>	<i>gauche</i>	<i>transoid</i>
B(8'')-B(10'')-B(10)-B(8) dihedral ^a	36	93	161	116	45	46	107	180	42	113	180
Deviation from ideal ^b	0	15	19	8	9	10	1	0	6	5	0
B(I'')-B(10'')-B(10)-B(I) dihedral	96	34	63	104	173	45	103	180	42	113	180
Deviation from ideal	12	2	27	4	7	9	5	0	6	5	0
ΔE, kcal·mol ⁻¹ ^c	17.3	17.4	12.8	5.4	11.8	16.5	10.4	12.3	13.8	4.2	0.0
ΔG, kcal·mol ⁻¹ ^c	16.7	17.1	11.8	5.2	11.5	16.1	10.2	12.0	13.5	4.1	0.0
CH···I BCP											
r, Å ^d	2.882								2.935		
	3.001								2.849		
	2.646					2.785			2.850		
	2.646					3.059			2.936		
ρ _{BCP} , a.u.	3.001								2.934		
	0.010								0.009		
	0.017					0.012			0.011		
	0.017					0.007			0.009		
E _{BCP} , kcal·mol ⁻¹ ^e	0.008								0.009		
	-1.8								-1.6		
	-3.3					-2.3			-2.0		
	-3.3					-1.3			-1.6		
DI ^f	-1.4								-1.6		
	0.037								0.031		
	0.075					0.052			0.040		
	0.075					0.021			0.031		
I···I BCP											
r, Å ^d	4.225					4.013			4.002		
ρ _{BCP} , a.u.	0.003					0.005			0.006		
E _{BCP} , kcal·mol ⁻¹ ^e	-0.6					-1.0			-1.0		
DI ^f	0.017					0.030			0.036		
BH···I BCP											
r, Å ^d	3.069					2.970			3.061		
	3.069					3.122			2.940		
ρ _{BCP} , a.u.	0.008					0.009			0.010		
	0.008					0.007			0.010		
E _{BCP} , kcal·mol ⁻¹ ^e	-1.3					-1.7			-1.9		
	-1.3					-1.3			-1.9		
DI ^f	0.029					0.036			0.038		
	0.029					0.025			0.029		

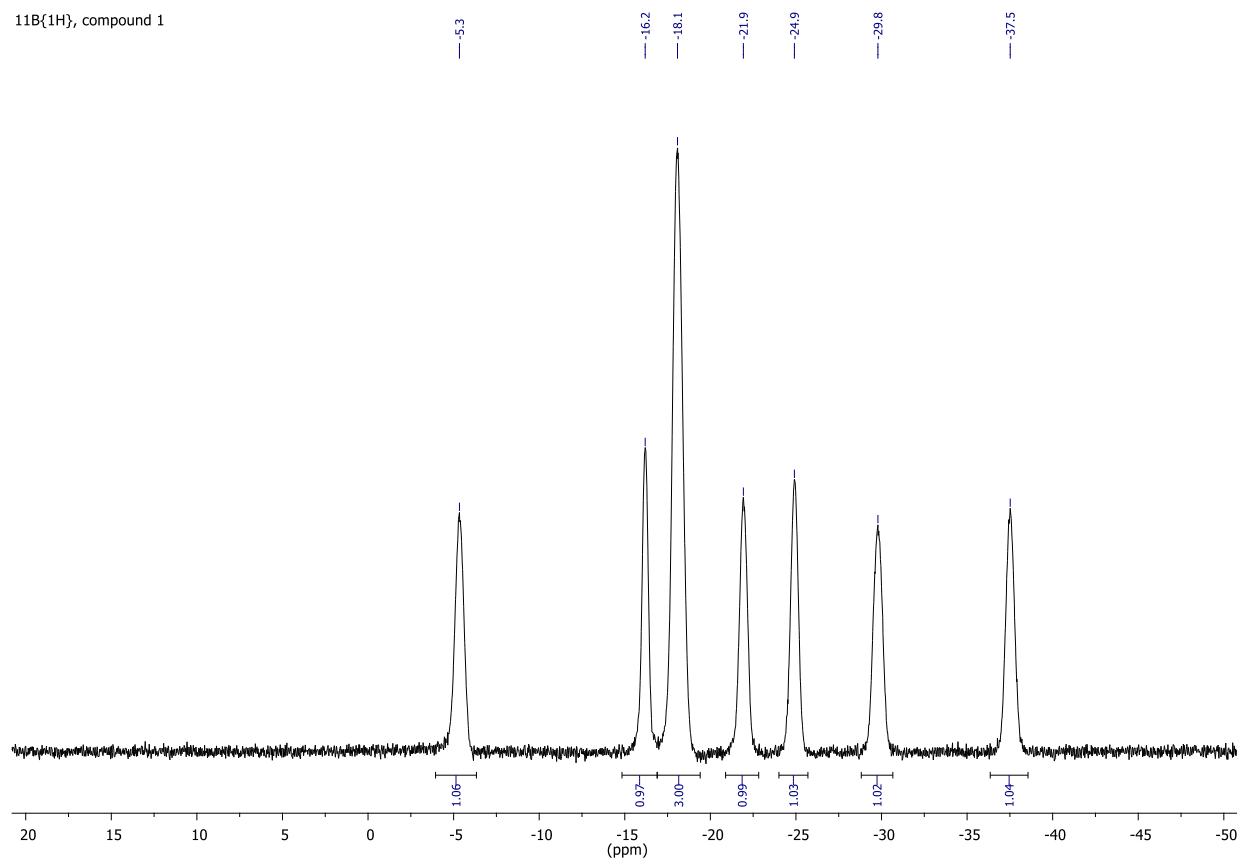
^{a,b} ideal angles are 36° for *cisoid*, 108° for *gauche* and 180° for *transoid* rotamers ^c all energies are respective to most favored 8,8'-*transoid* isomer.

^d H...I and I...I distances respectively ^e energy of interaction calculated as $E_{BCP}=0.5 \cdot V_{BCP}$ ^f QTAIM electron delocalization index (formal bond order)

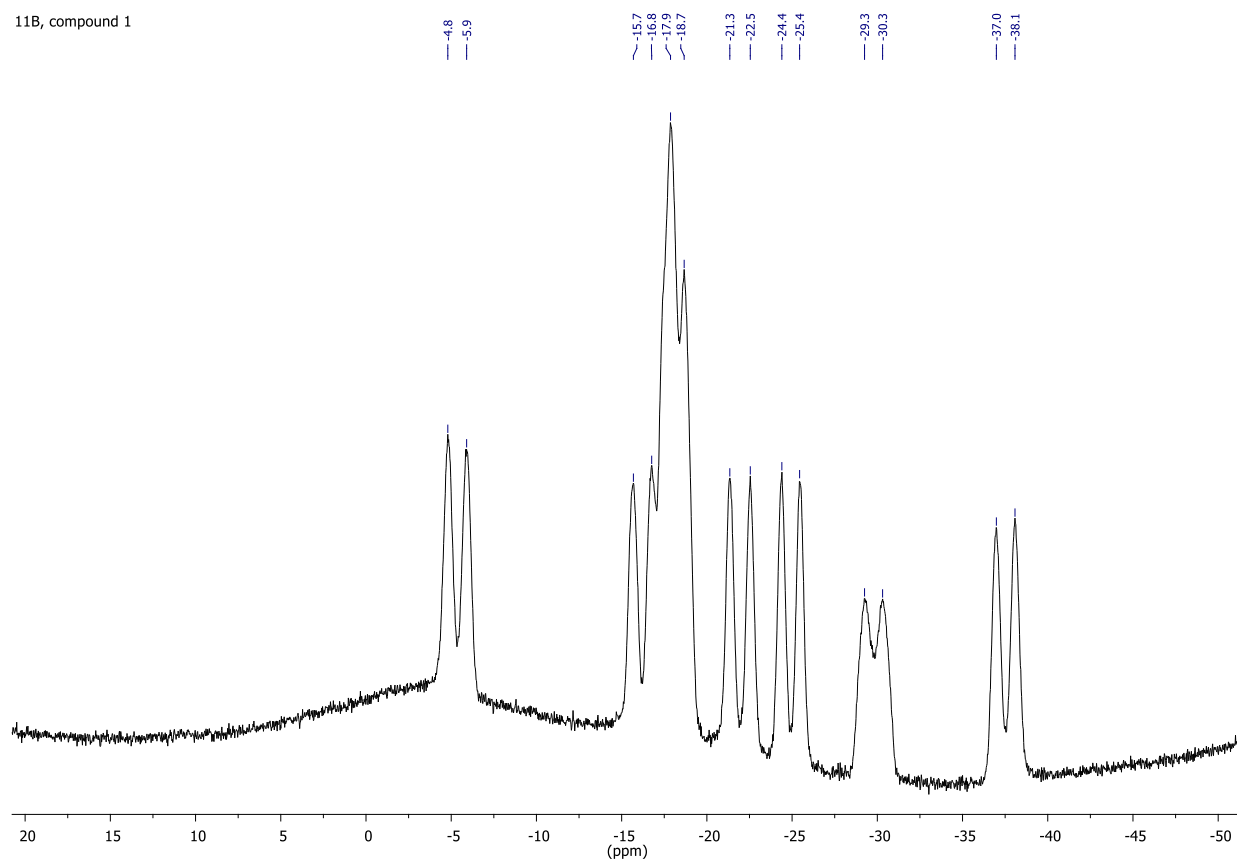
Table 2. Selected geometry parameters (distances in Å, angles in deg.) for Na[12].
Comparison of experiment and theory.

Bond	X-ray	Calc	Torsion angle or noncovalent bond	X-ray	Calc
Co1-Co1'	2.5401(6)	2.557	B8-Co1-Co1''-C2	-6.6(3)	-10.6
Co1-Co1''	2.5436(6)	2.557	C2-Co1-Co1'-B8'	-12.1(3)	-10.6
Co1'-Co1''	2.5339(6)	2.557	C2'-Co1'-Co1''-B8''	-14.7(3)	-10.5
Co1-S1	2.1637(8)	2.161	B8-B4-O1-C3	97.4(4)	92.5
Co1'-S1	2.1614(9)	2.161	B8'-B4'-O1'-C3'	99.3(4)	92.5
Co1''-S1	2.1636(8)	2.162	B8''-B4''-O1''-C3''	93.4(4)	92.4
Co1-O3	1.835(2)	1.823	O1...H1''	2.33	2.45
Co1'-O3	1.8408(19)	1.823	O1'...H1	2.68	2.46
Co1''-O3	1.844(2)	1.822	O1''...H1'	2.40	2.46
Co1-C1	2.015(3)	1.984	H8...H2''	2.13	2.07
Co1-C2	2.021(3)	1.982	H8''...H2'	2.22	2.06
Co1-B4	2.096(3)	2.083	H8'...H2	2.19	2.06
Co1-B8	2.104(4)	2.075			
Co1-B7	2.112(4)	2.097			
Co1'-C1'	2.014(3)	1.984			
Co1'-C2'	2.012(3)	1.982			
Co1'-B4'	2.109(4)	2.083			
Co1'-B8'	2.107(4)	2.076			
Co1'-B7'	2.112(3)	2.097			
Co1''-C1''	2.029(3)	1.983			
Co1''-C2''	2.029(3)	1.982			
Co1''-B4''	2.101(3)	2.084			
Co1''-B8''	2.110(3)	2.076			
Co1''-B7''	2.111(4)	2.098			

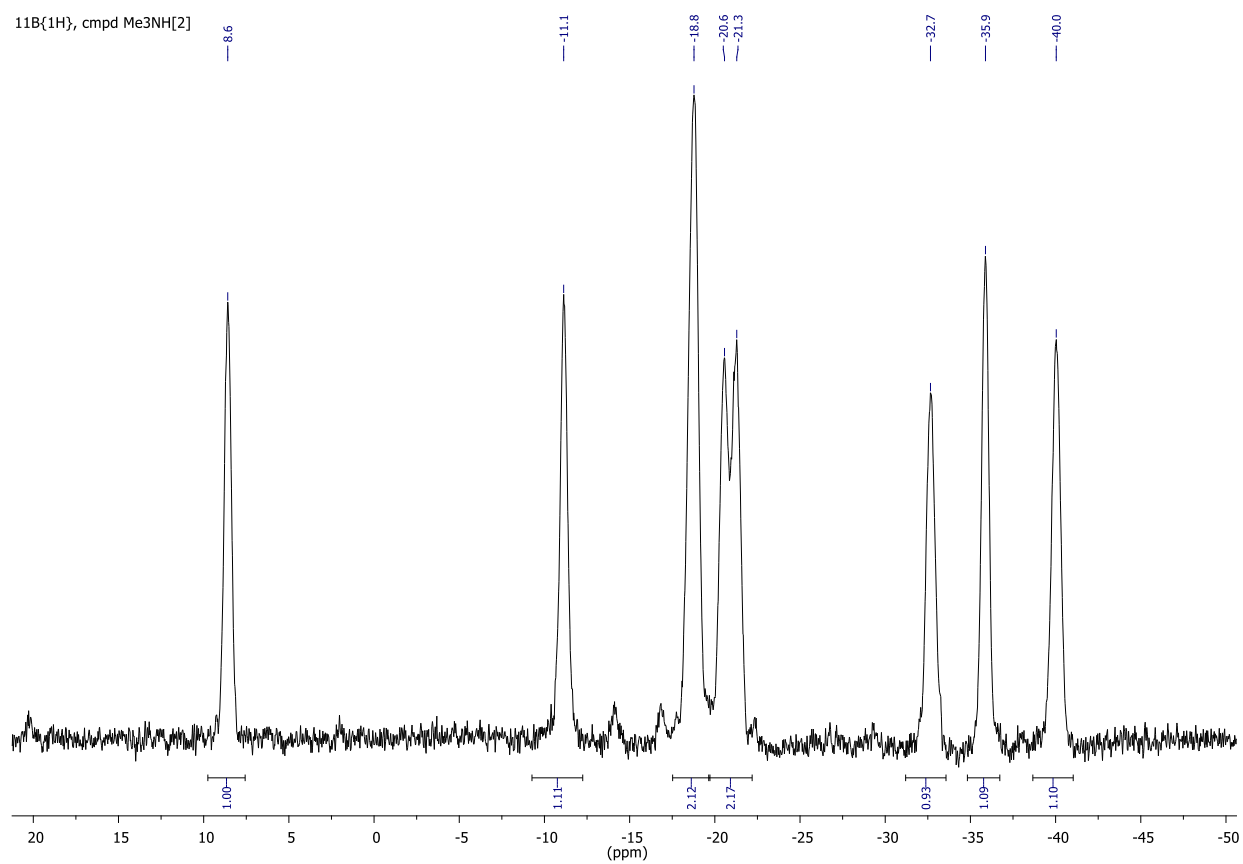
$^{11}\text{B}\{^1\text{H}\}$, compound 1



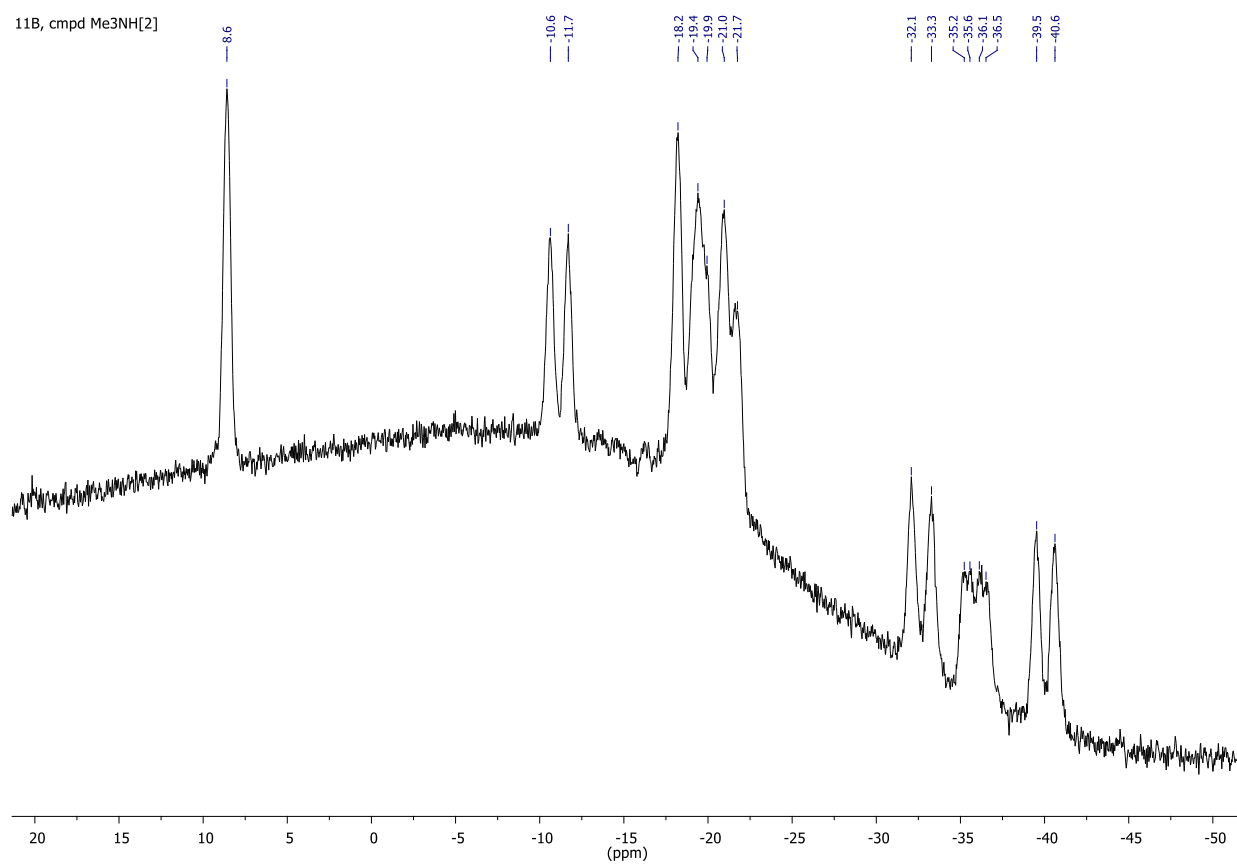
11B, compound 1

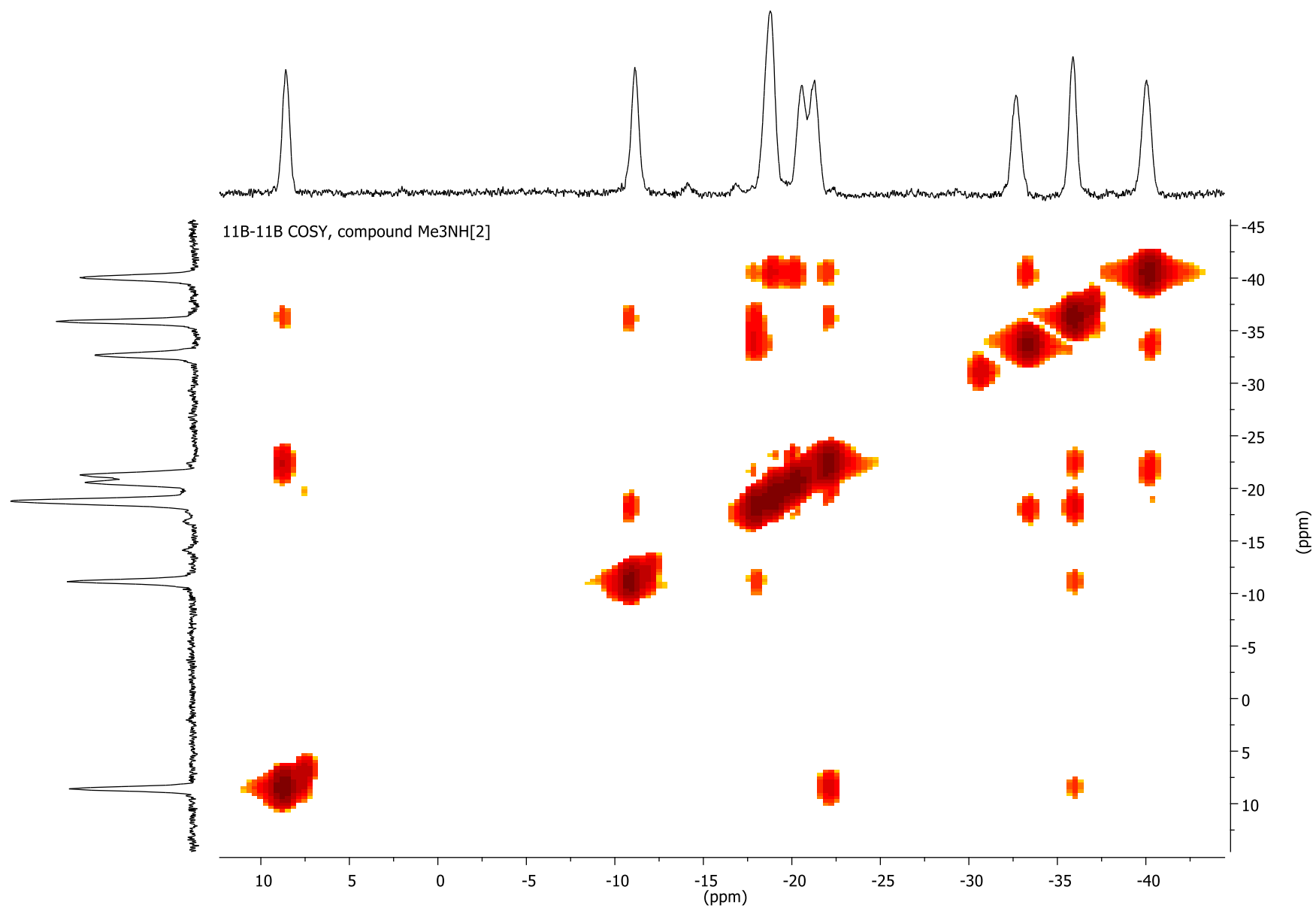


11B{1H}, compd Me3NH[2]

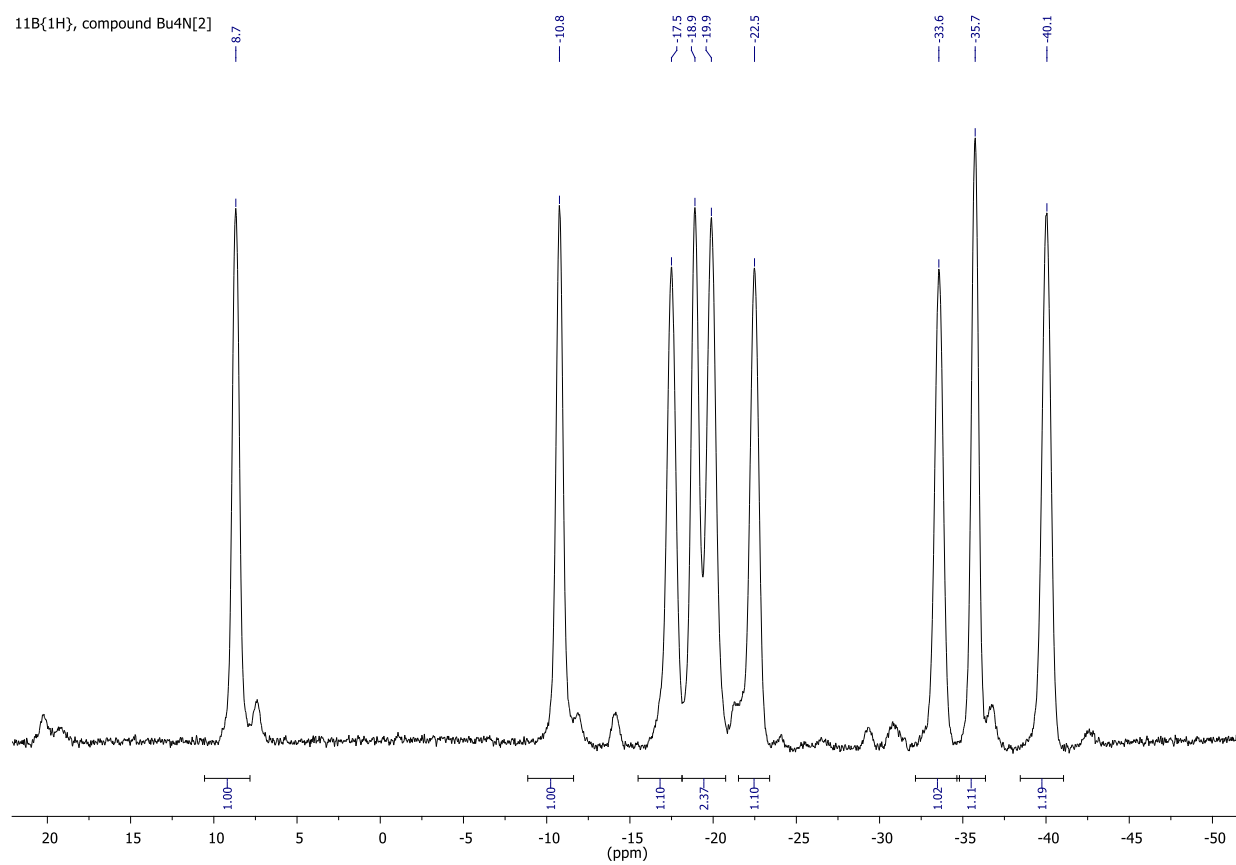


11B, cmpd Me3NH[2]

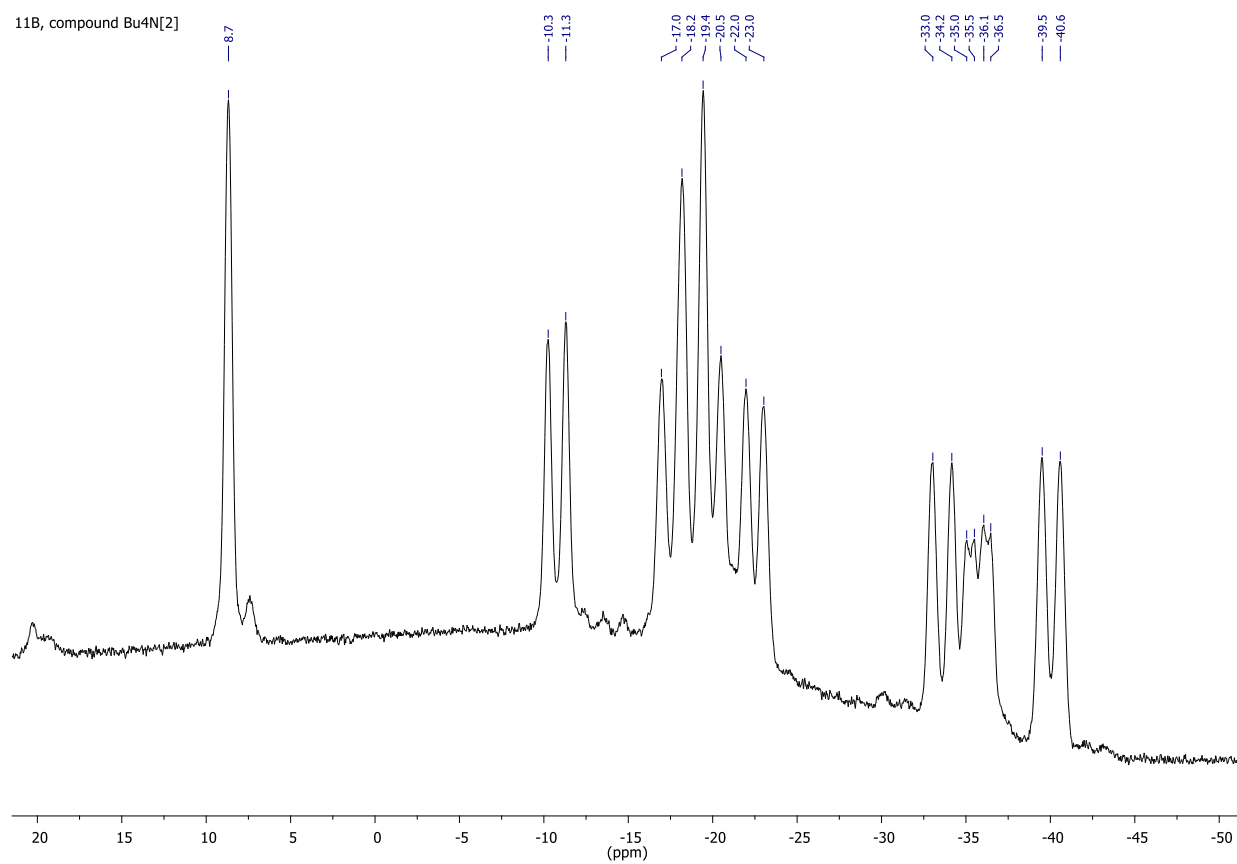




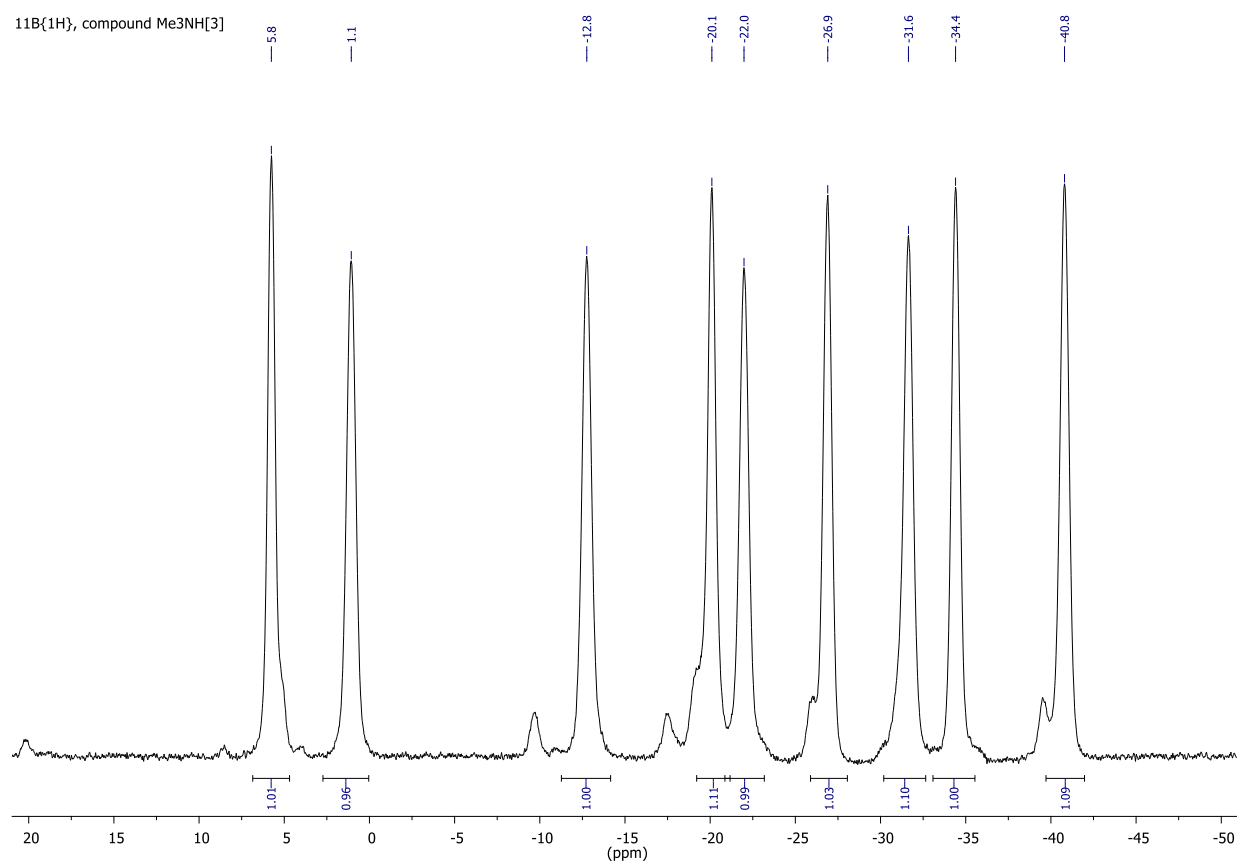
$^{11}\text{B}\{^1\text{H}\}$, compound Bu₄N[2]



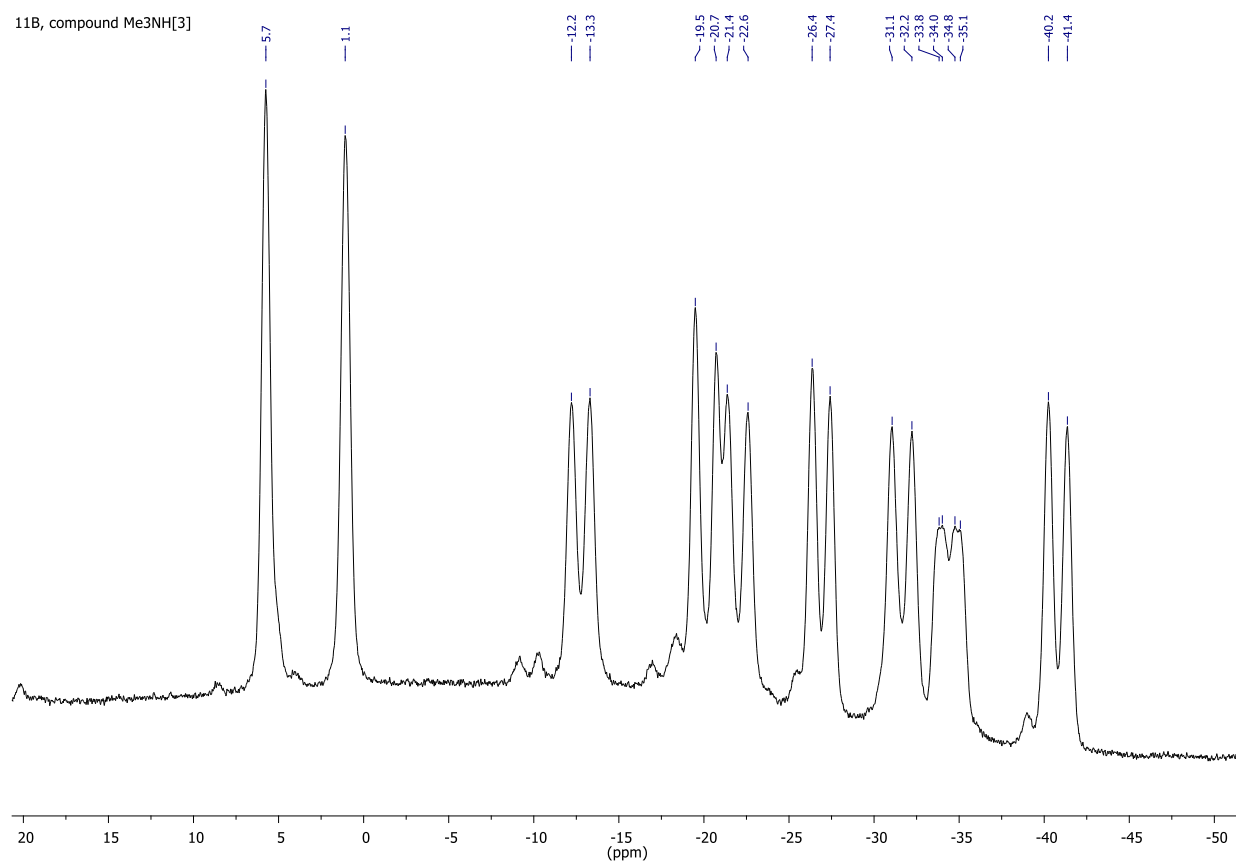
^{11}B , compound Bu₄N[2]

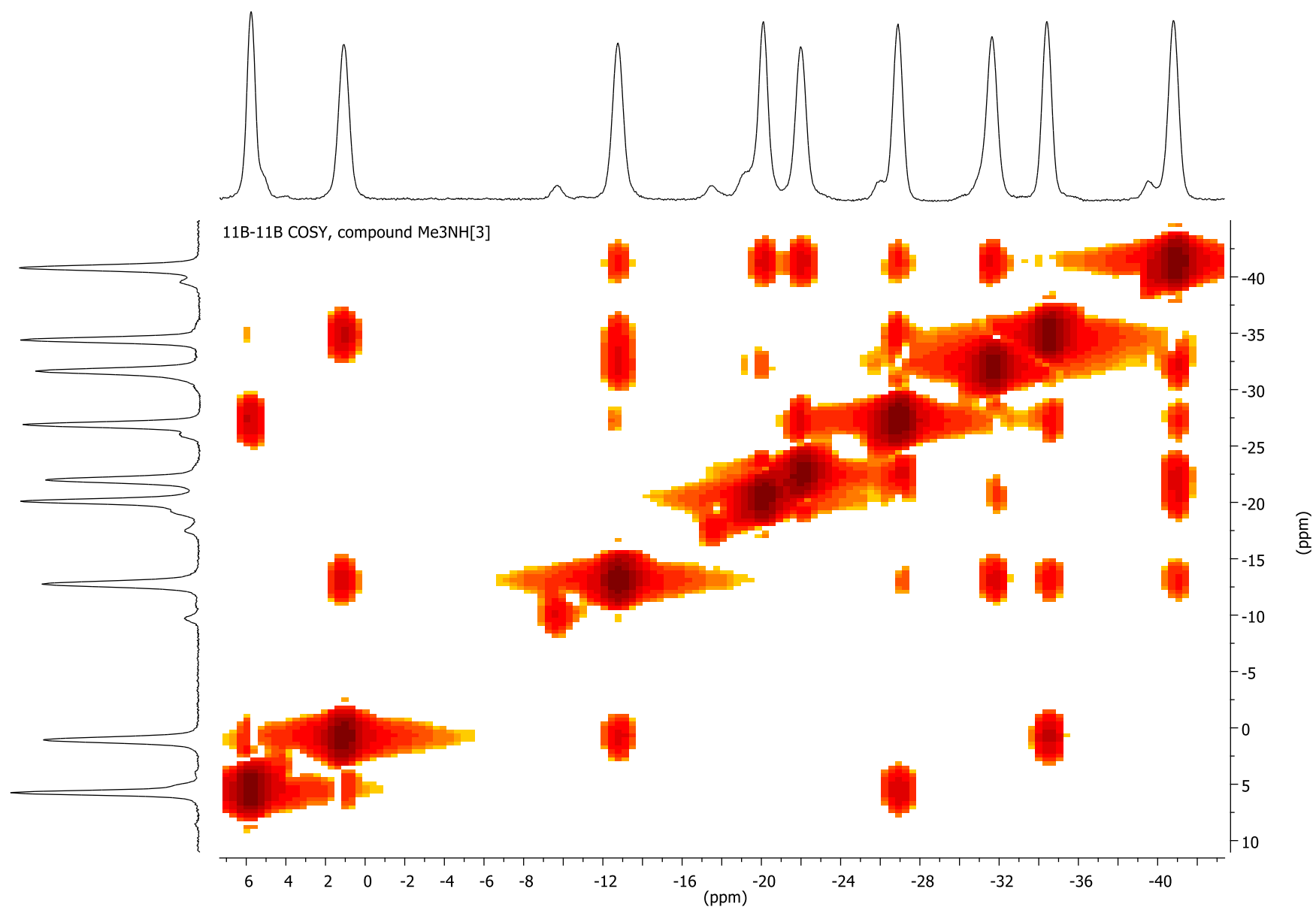


11B{1H}, compound Me3NH[3]

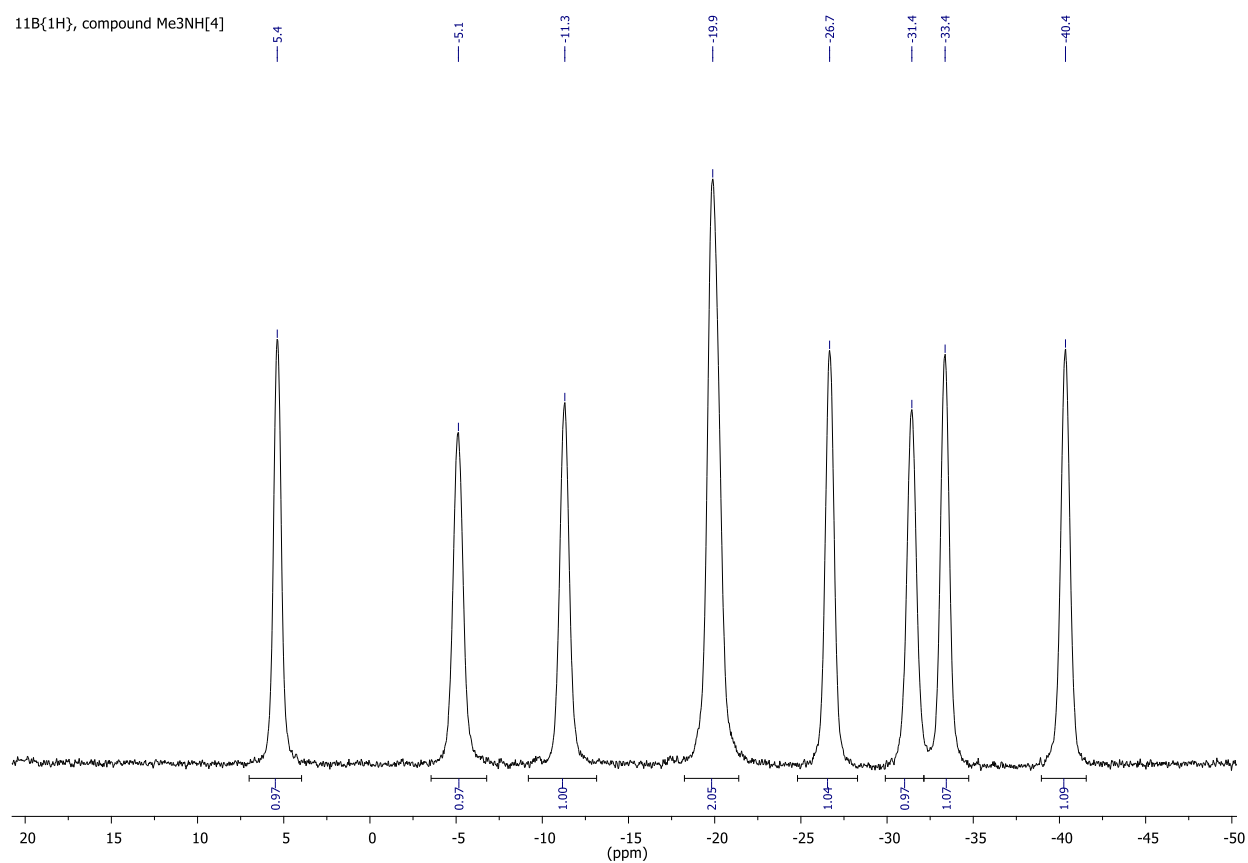


11B, compound Me3NH[3]

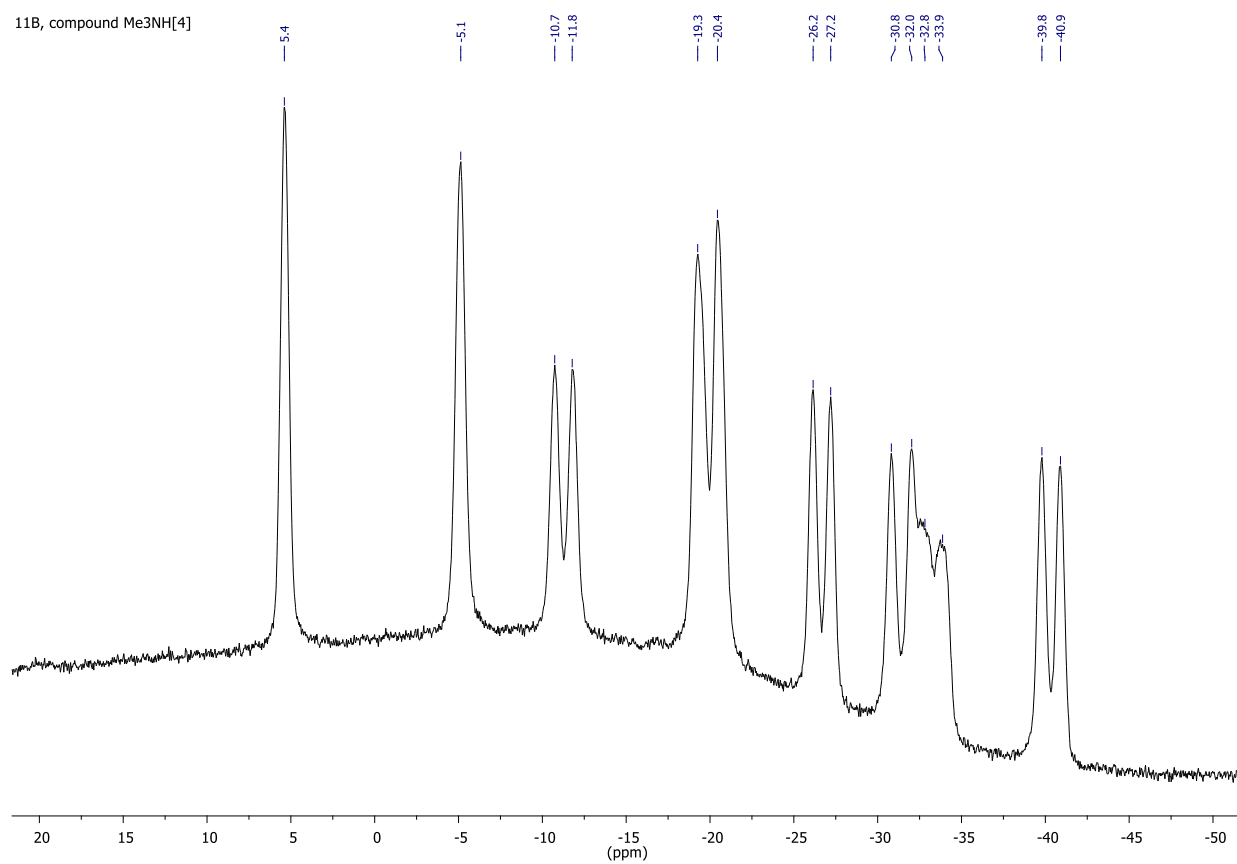


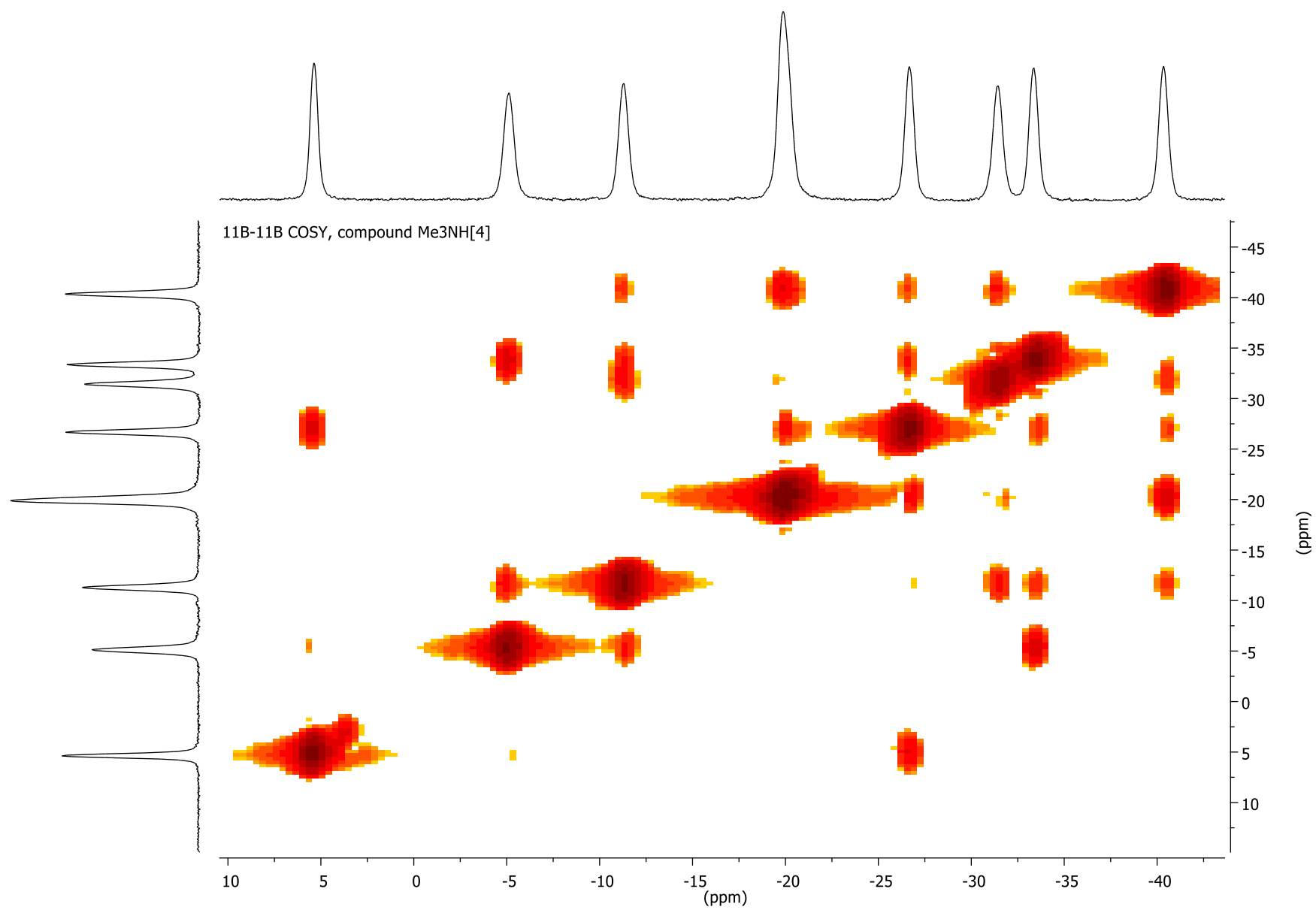


11B{1H}, compound Me3NH[4]

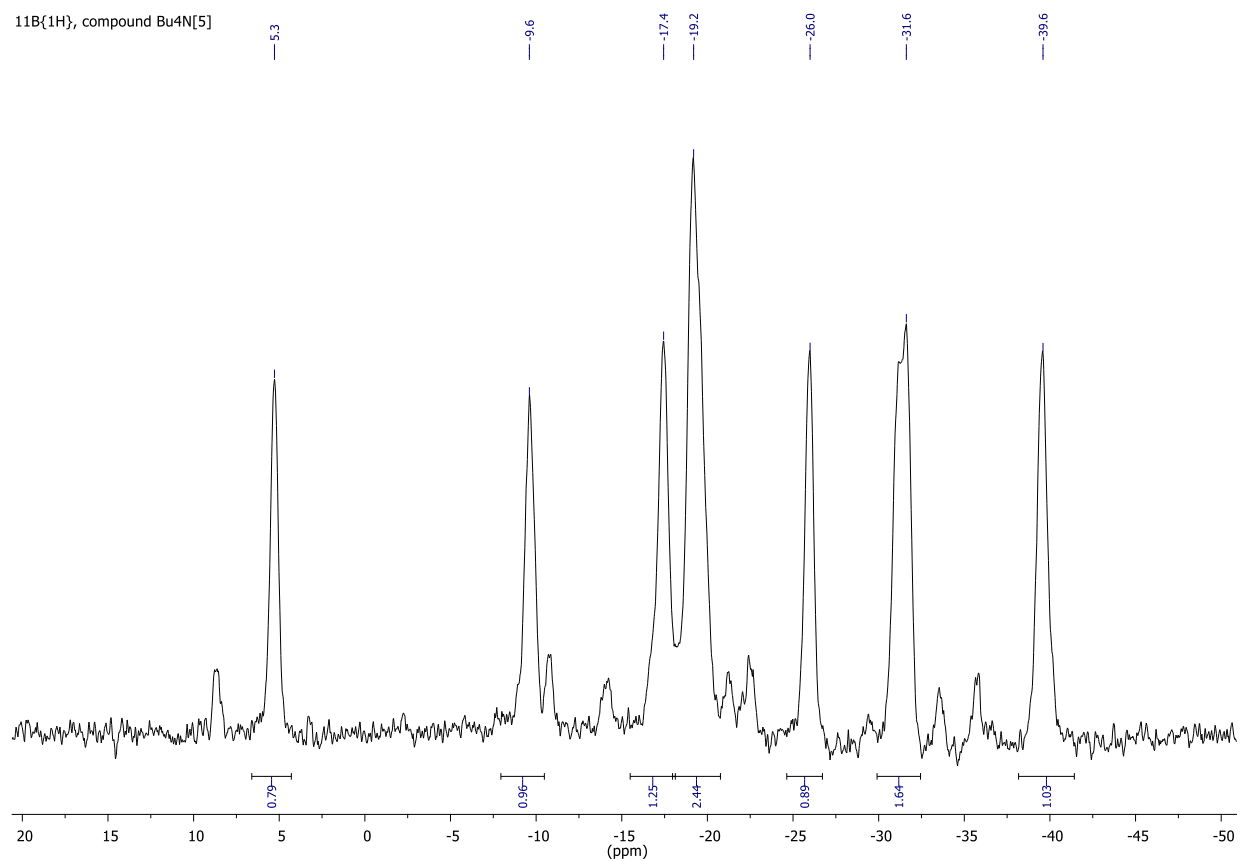


11B, compound Me3NH[4]

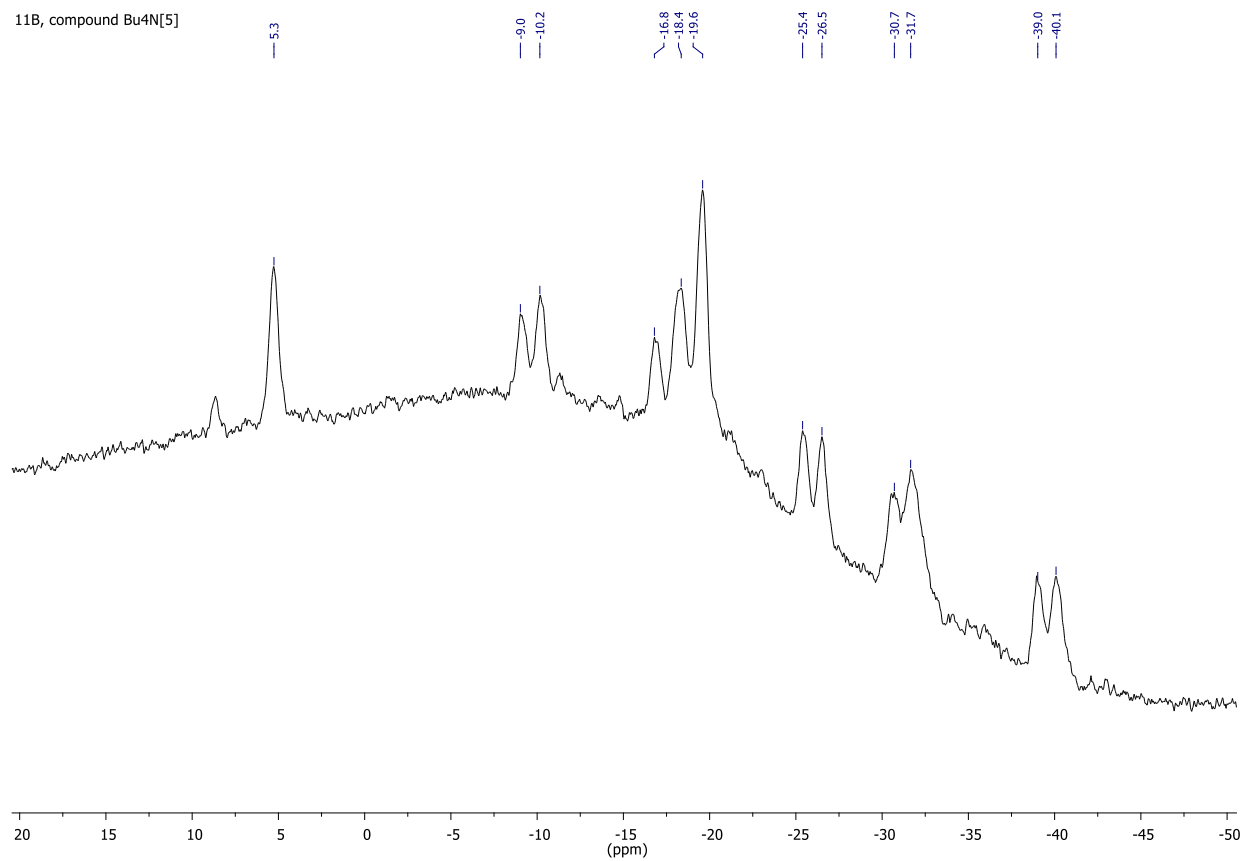


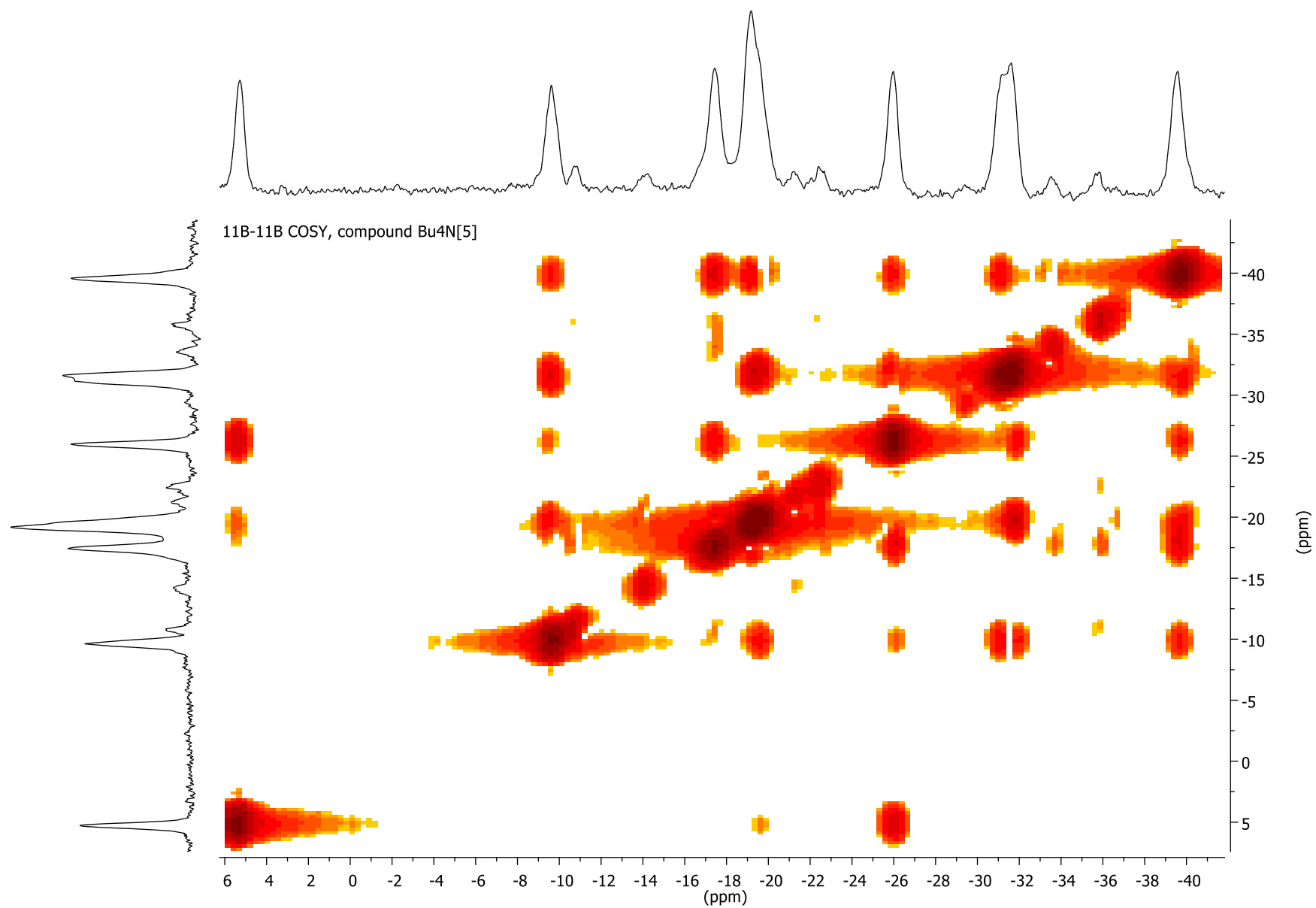


$^{11}\text{B}\{^1\text{H}\}$, compound Bu₄N[5]

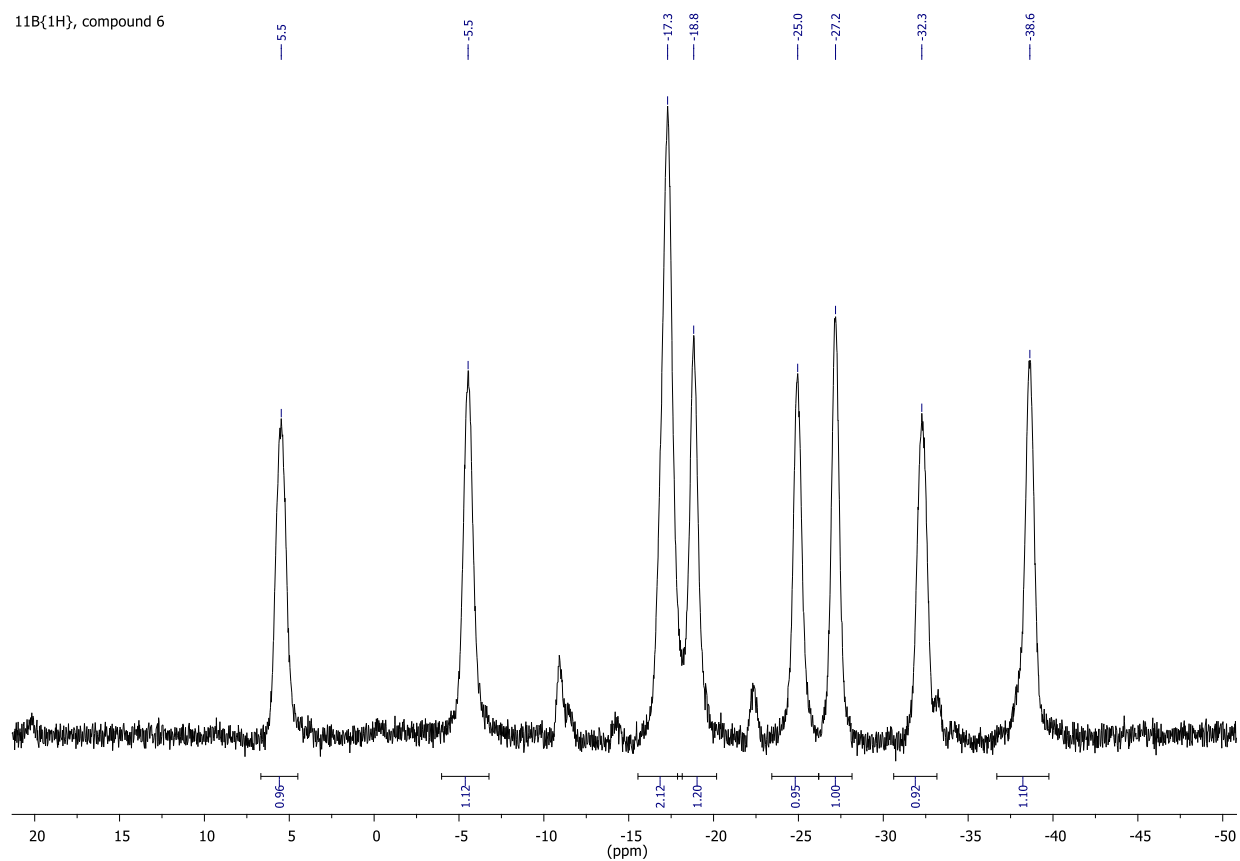


^{11}B , compound Bu₄N[5]

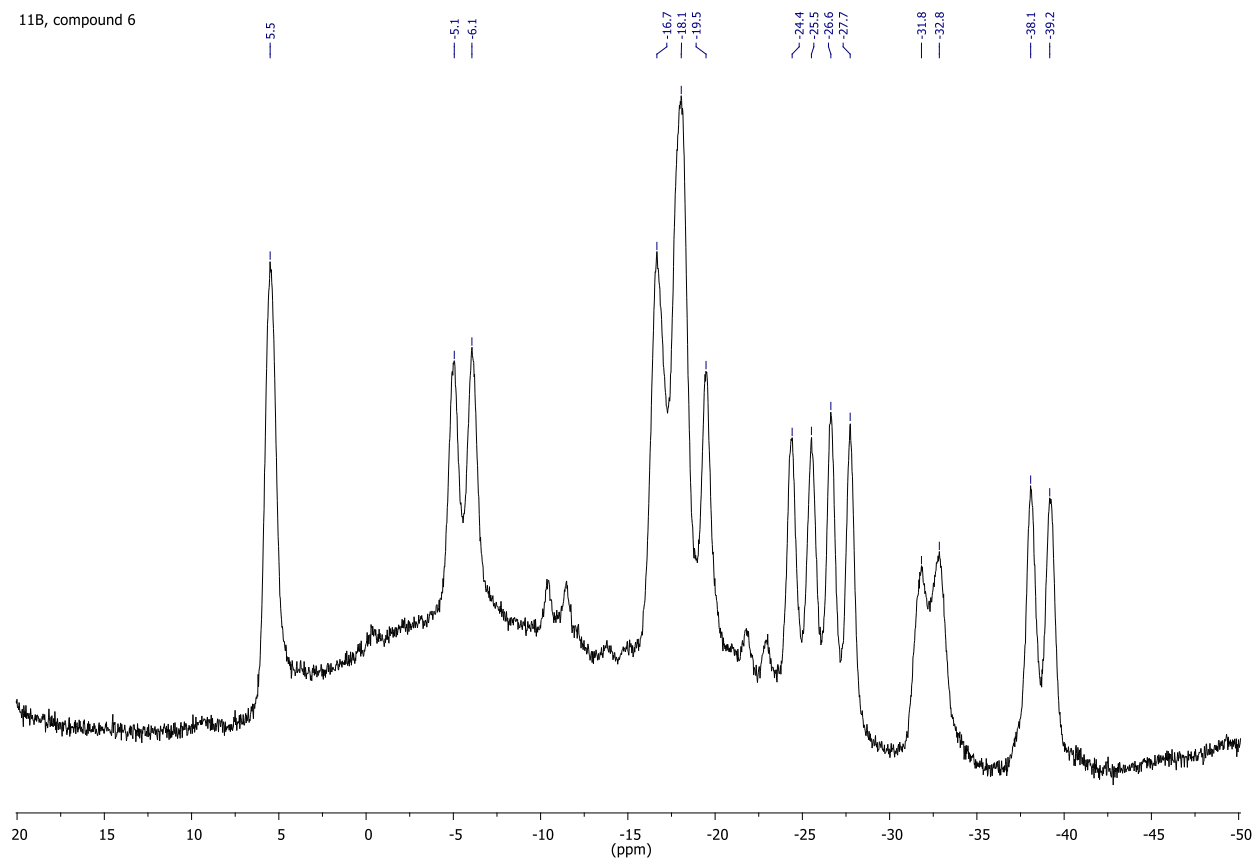




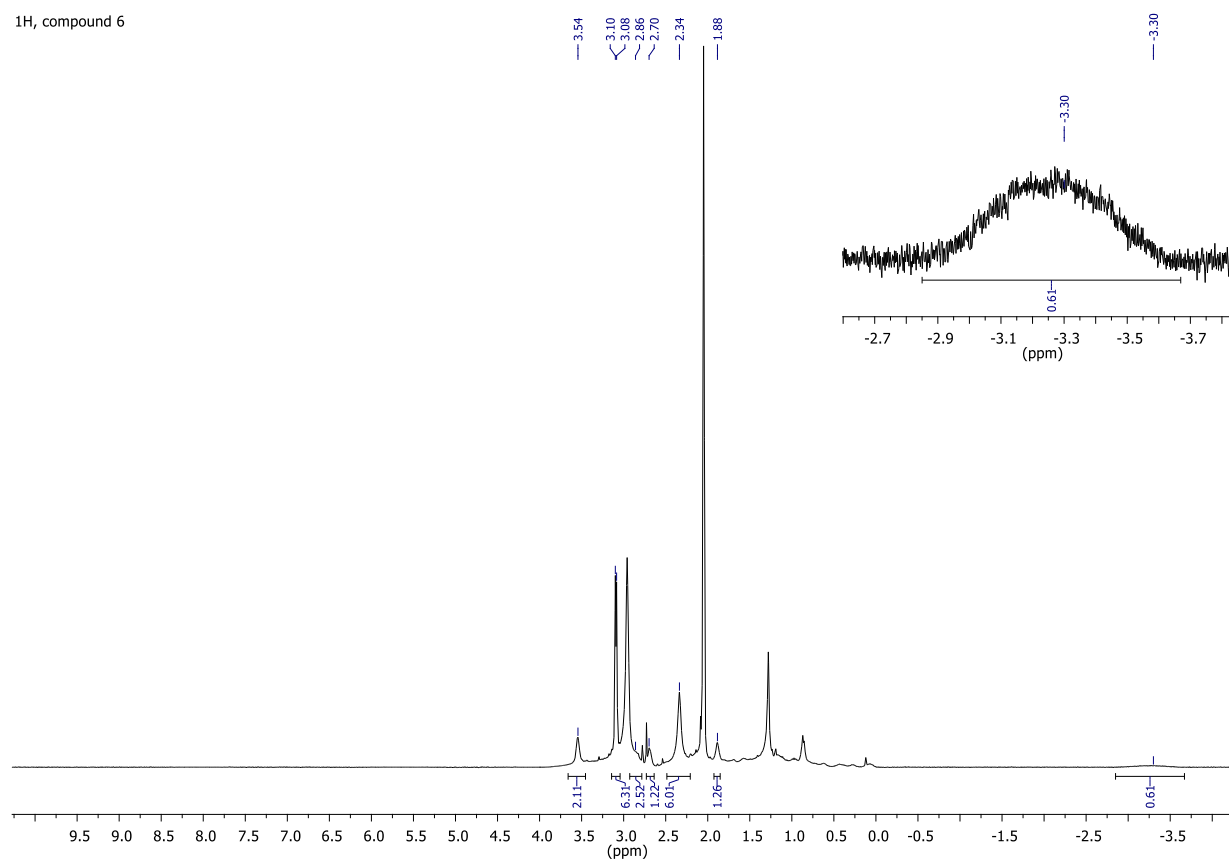
$^{11}\text{B}\{^1\text{H}\}$, compound 6



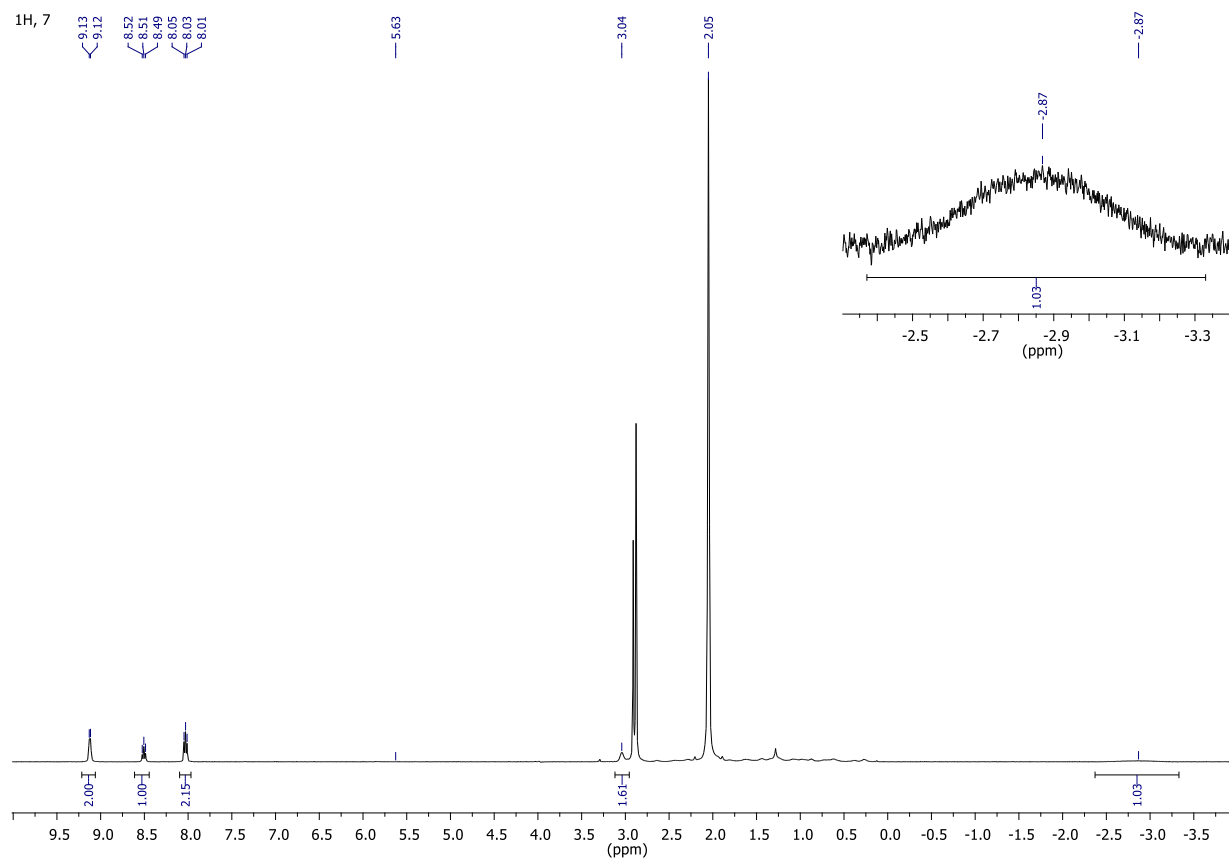
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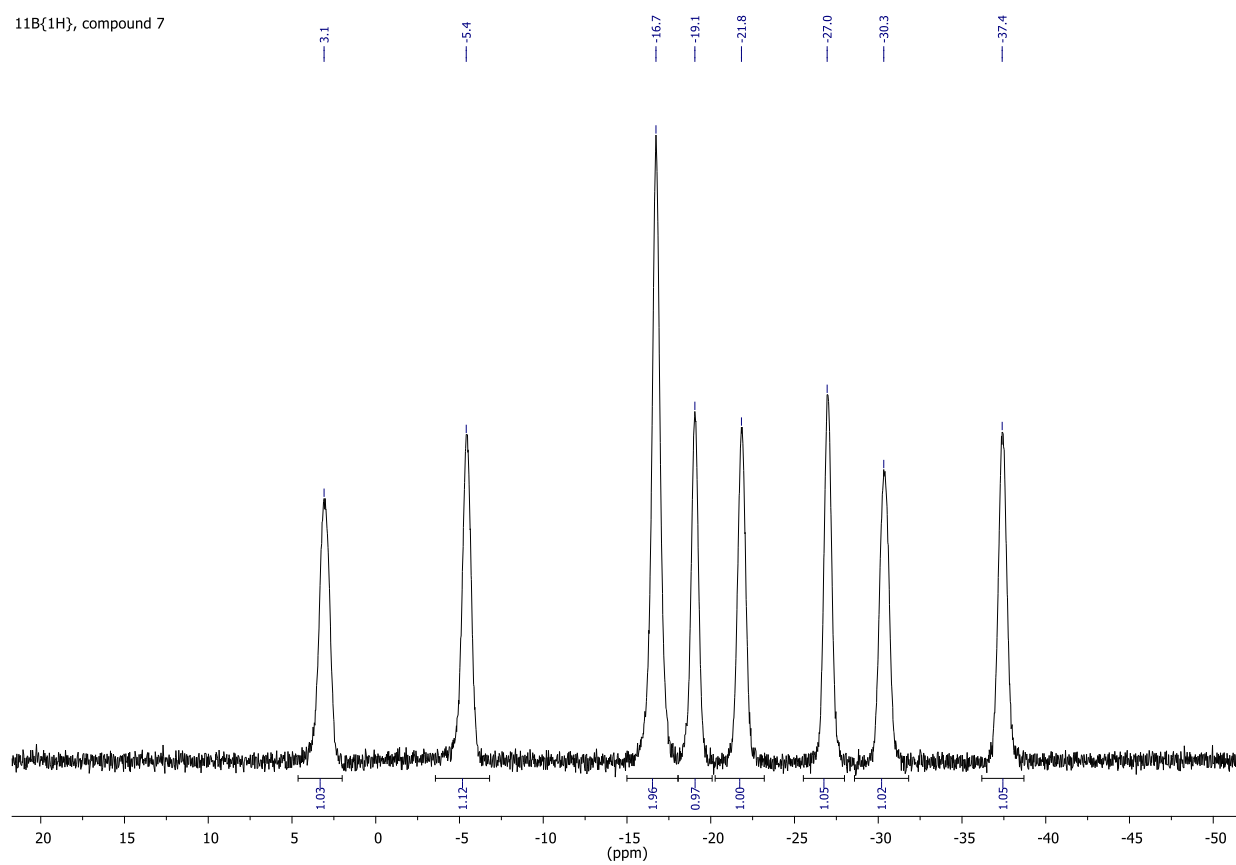
¹H, compound 6



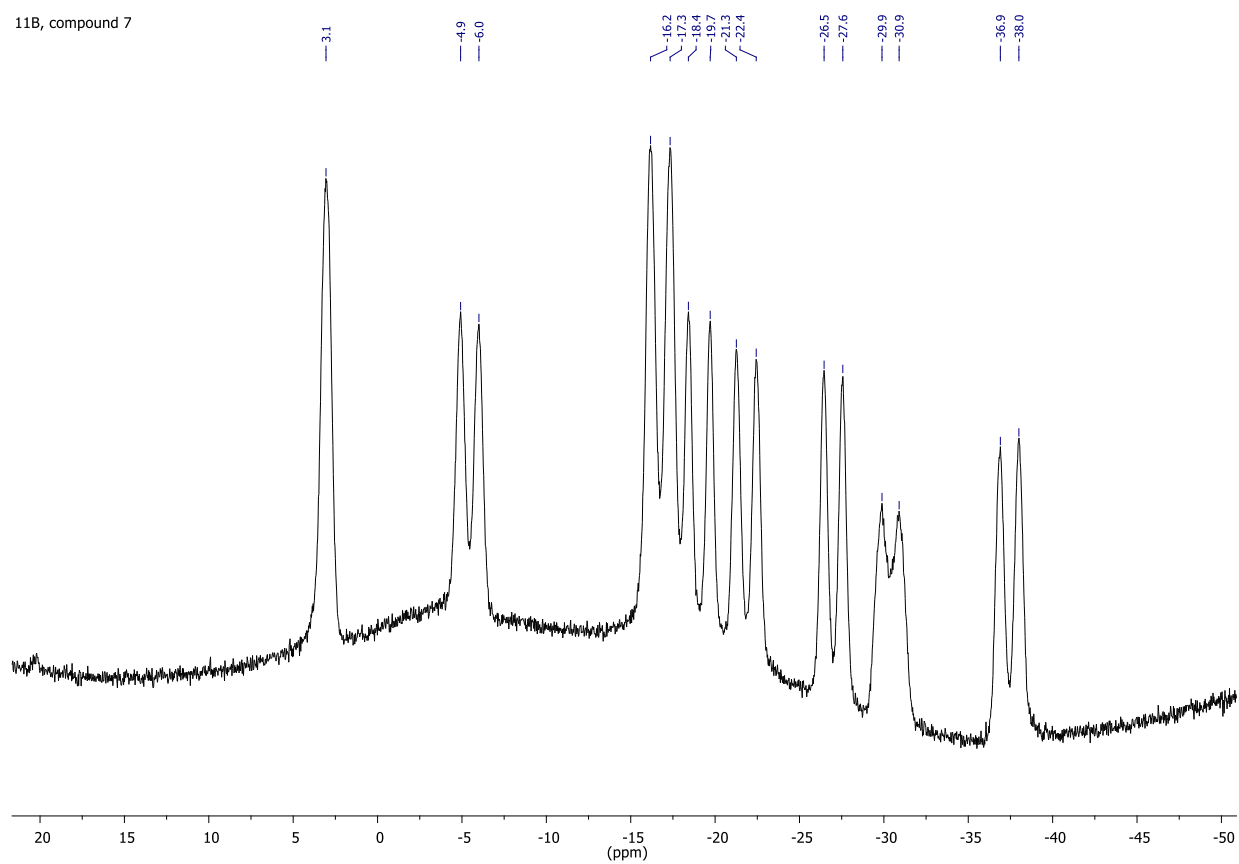
¹H, 7



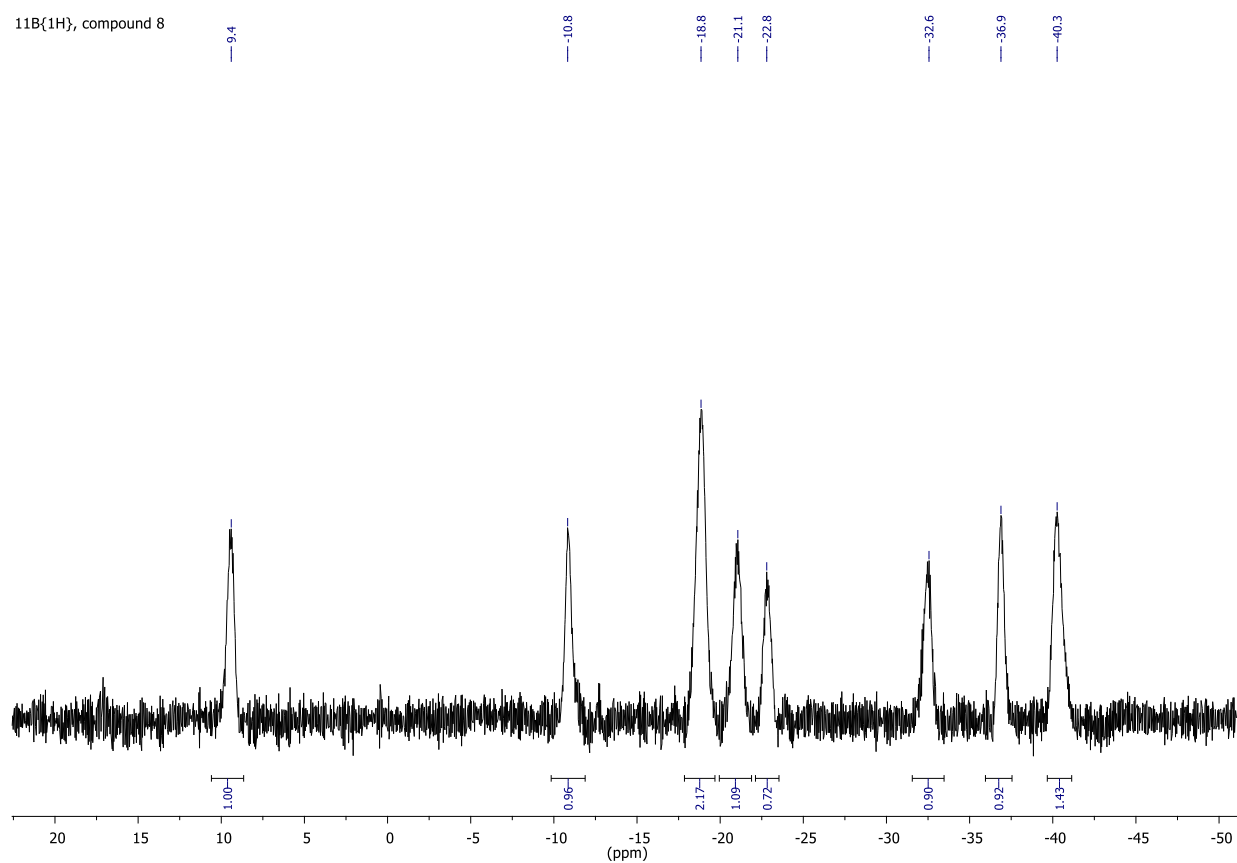
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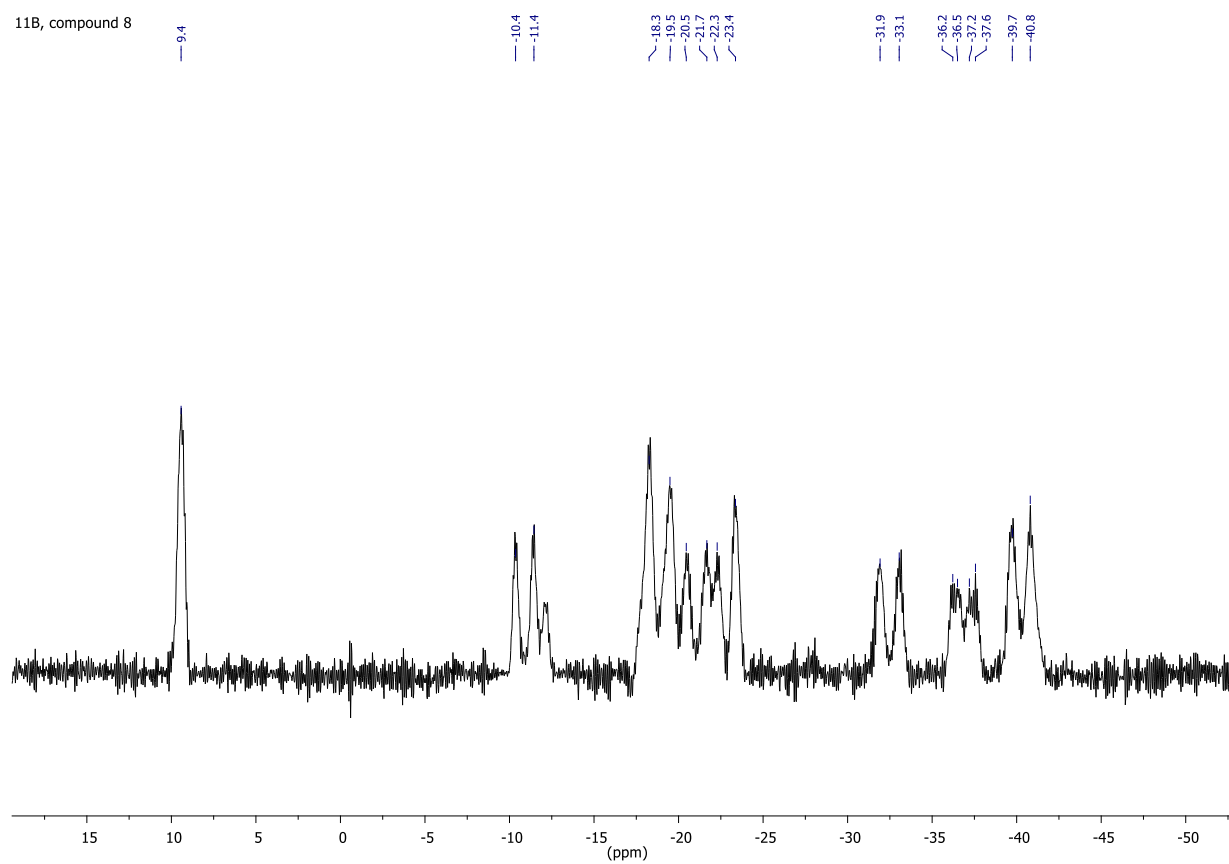
^{11}B , compound 7



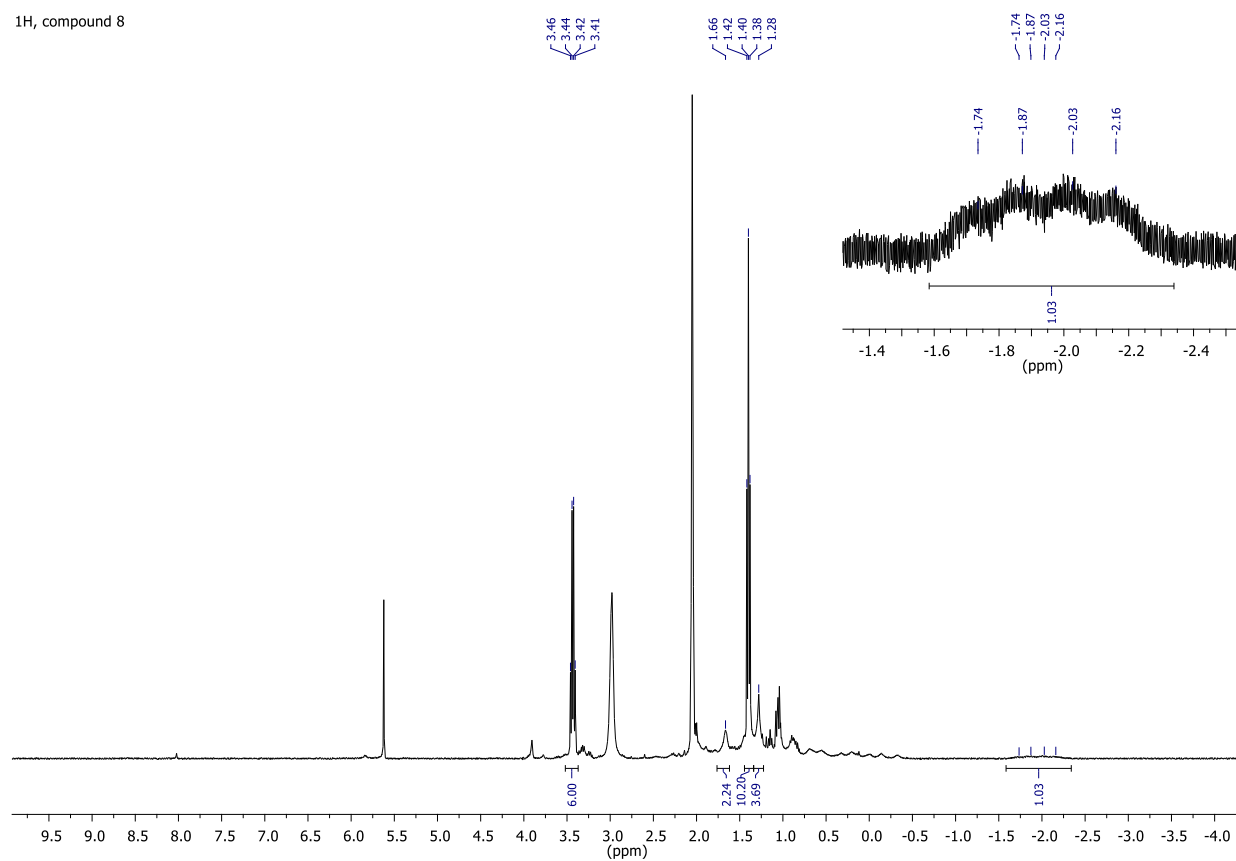
$^{11}\text{B}\{^1\text{H}\}$, compound 8



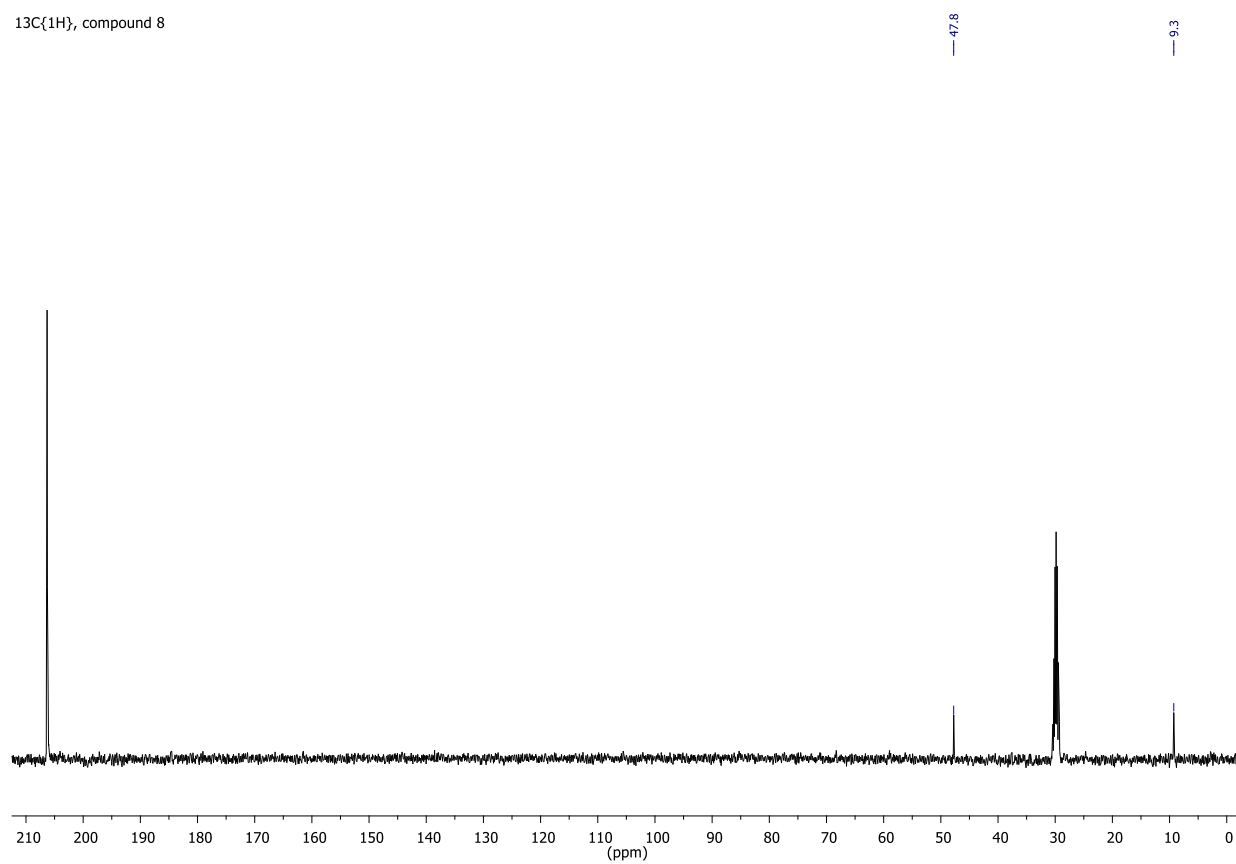
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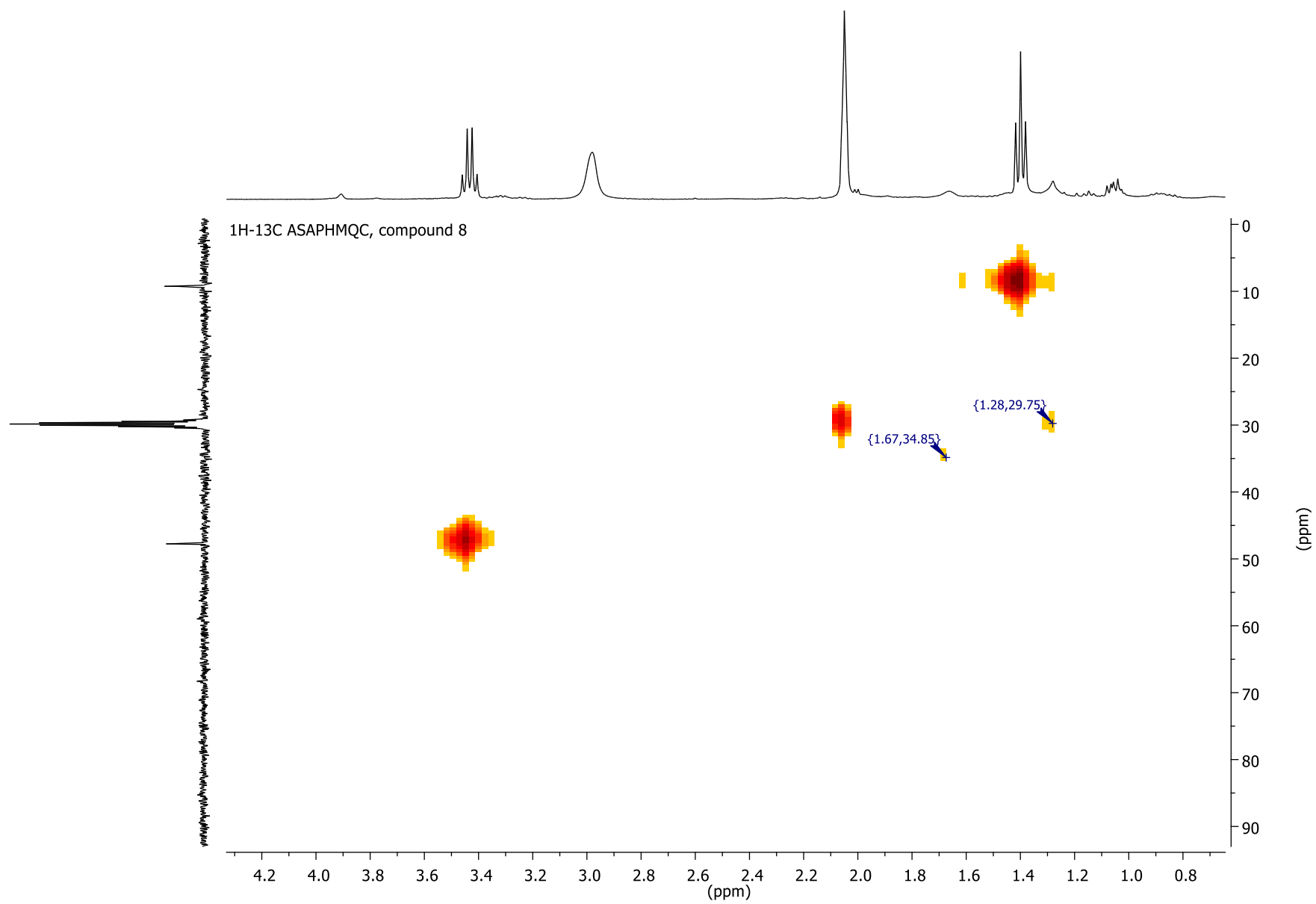


^1H , compound 8

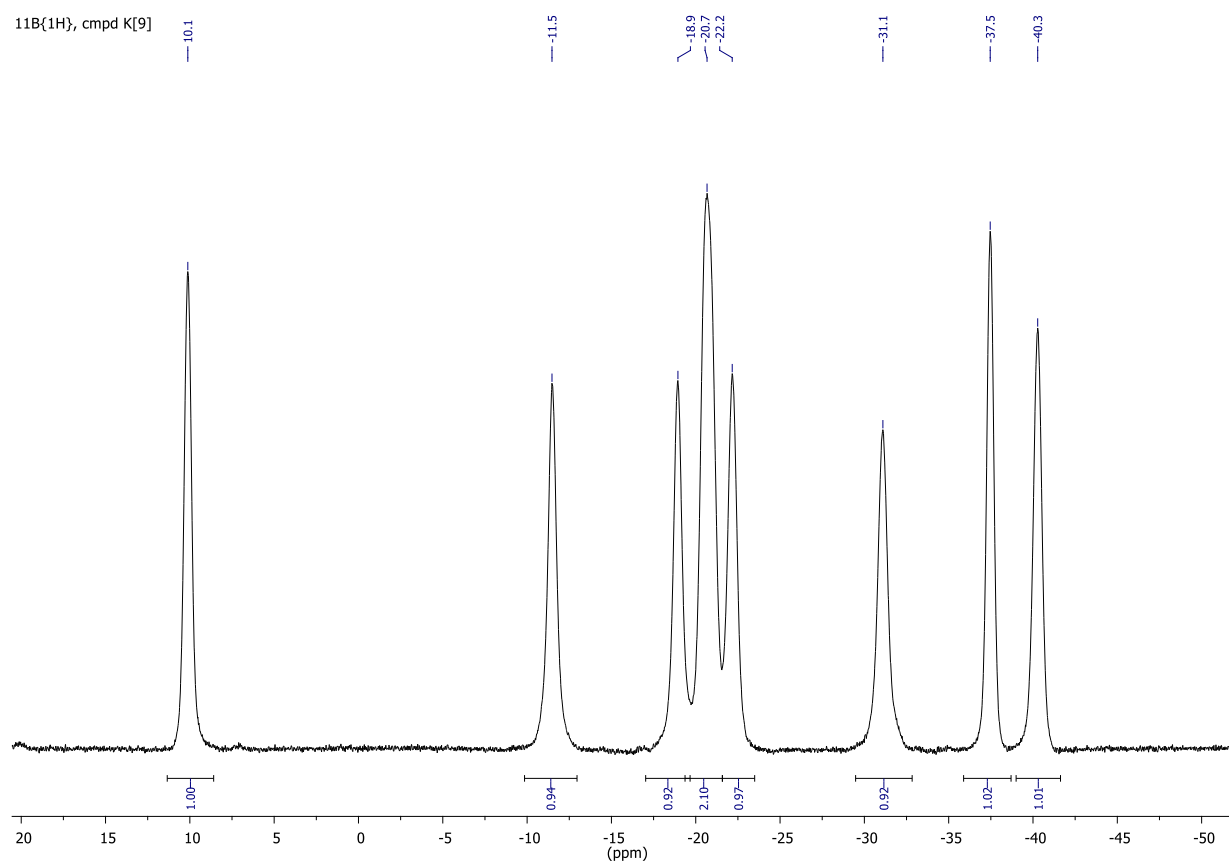


$^{13}\text{C}\{^1\text{H}\}$, compound 8

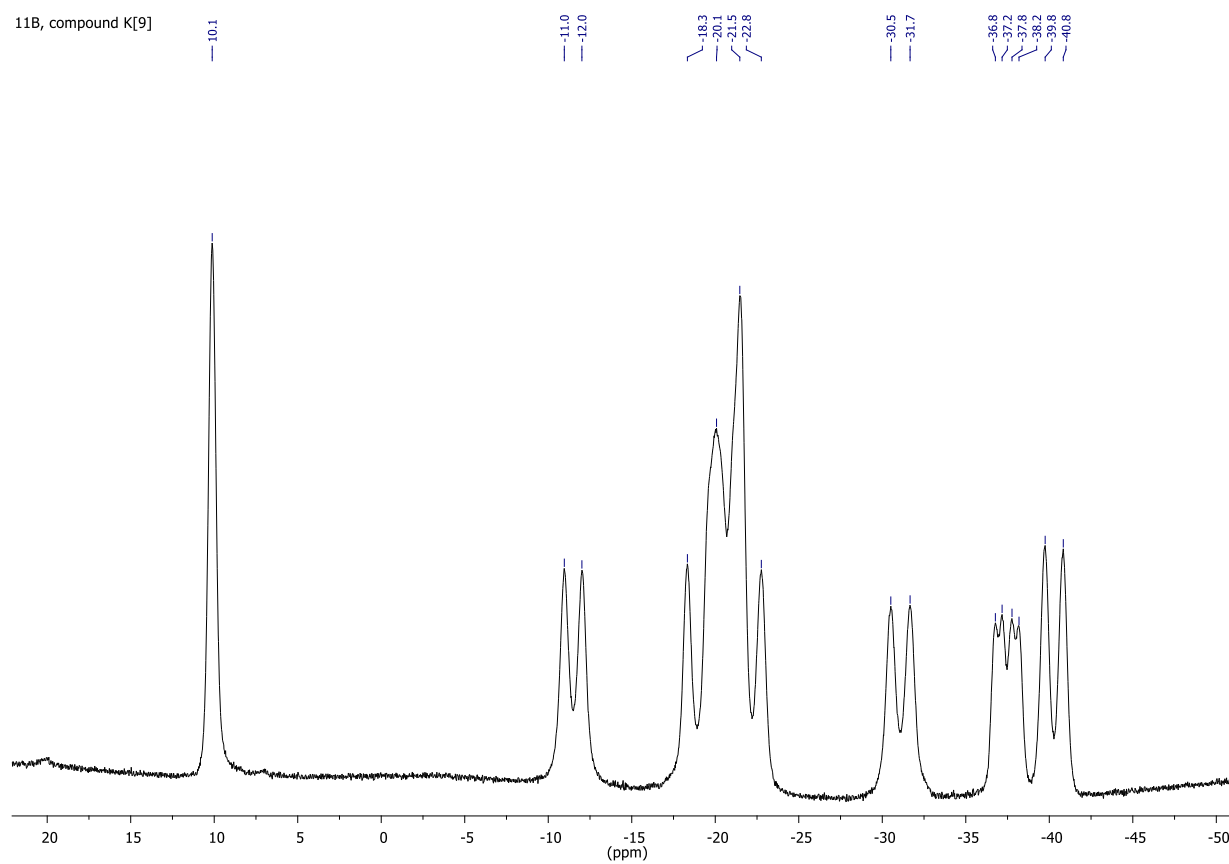




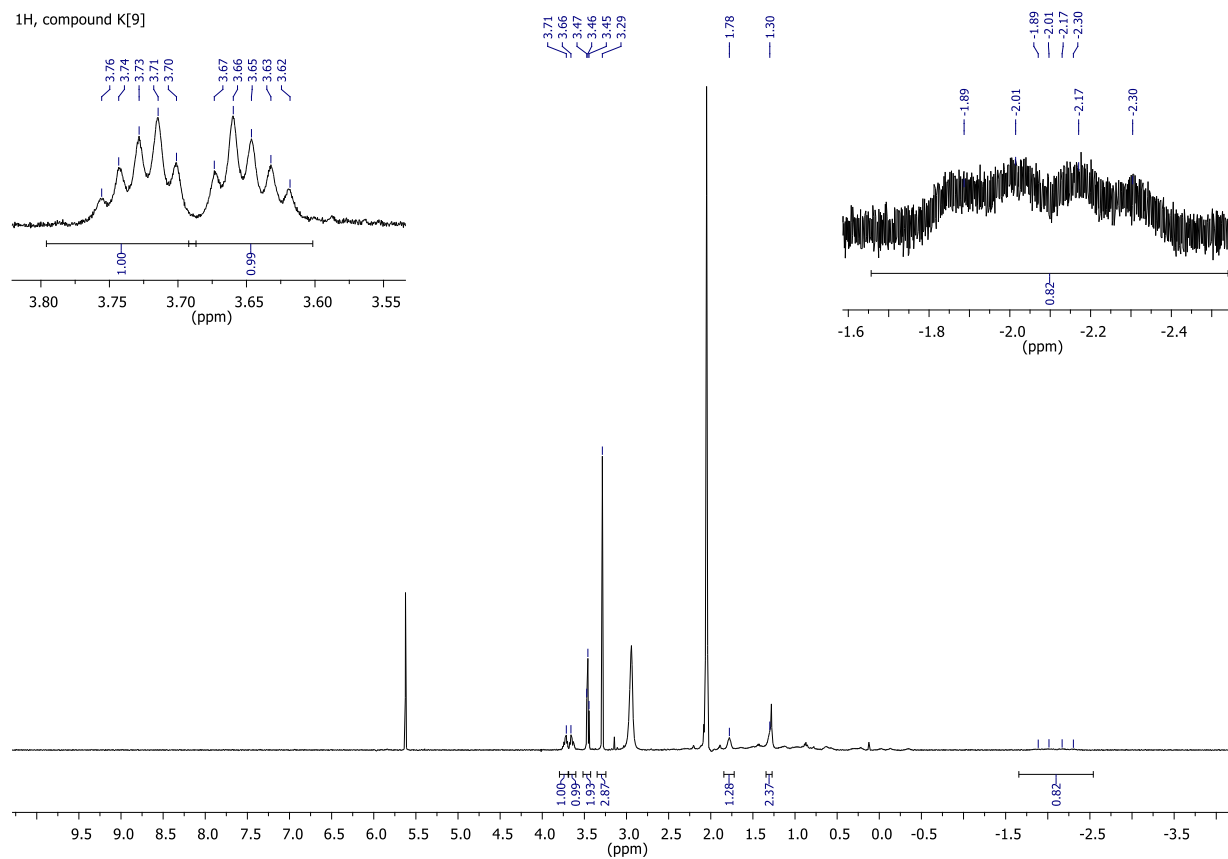
11B{1H}, compd K[9]



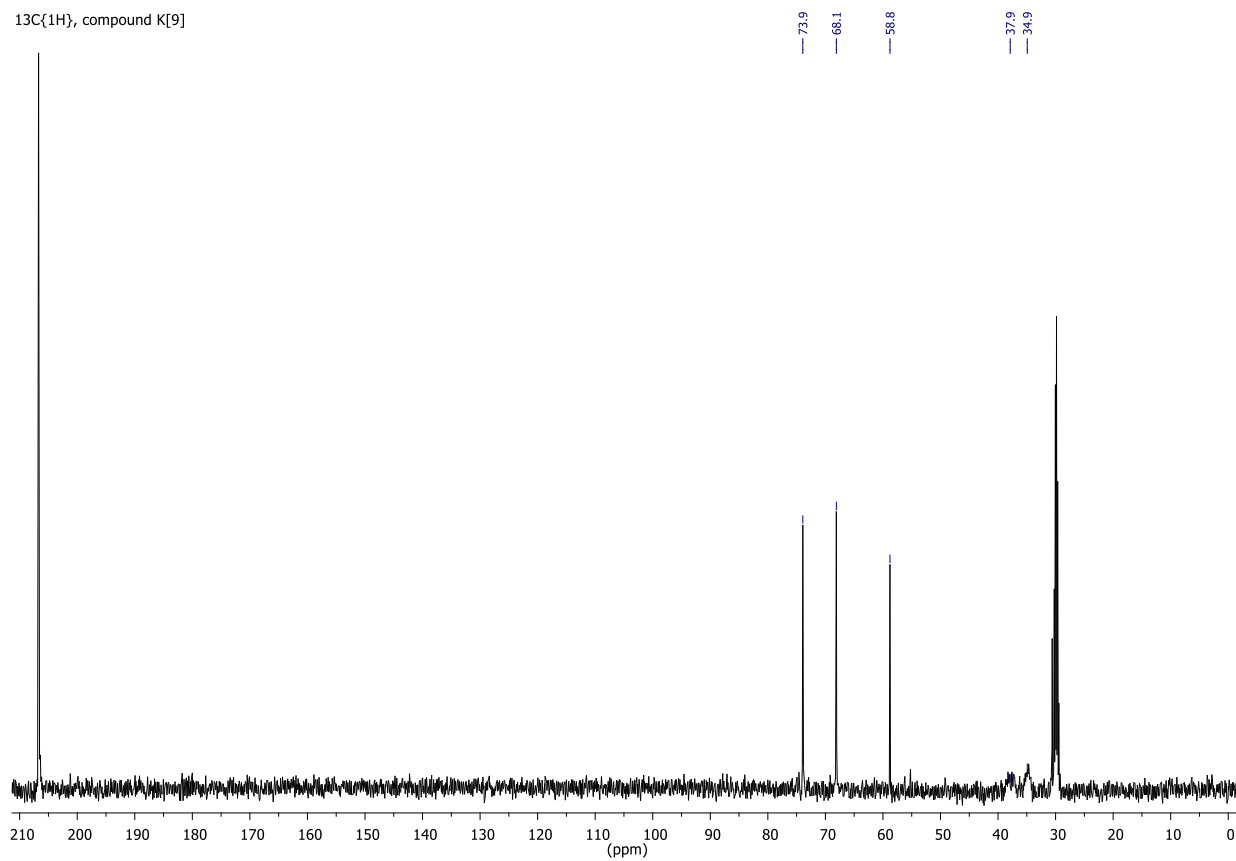
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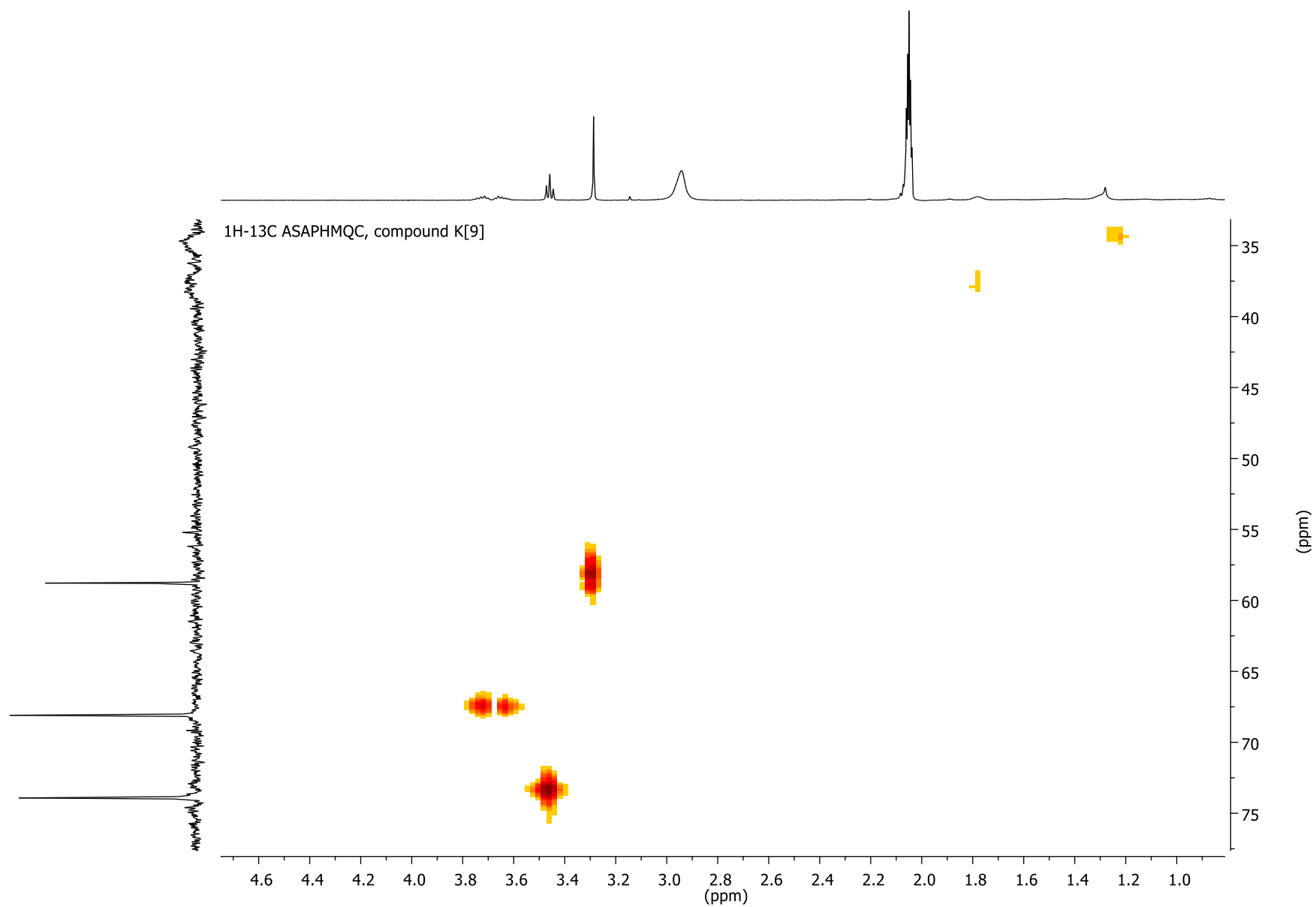


¹H, compound K[9]

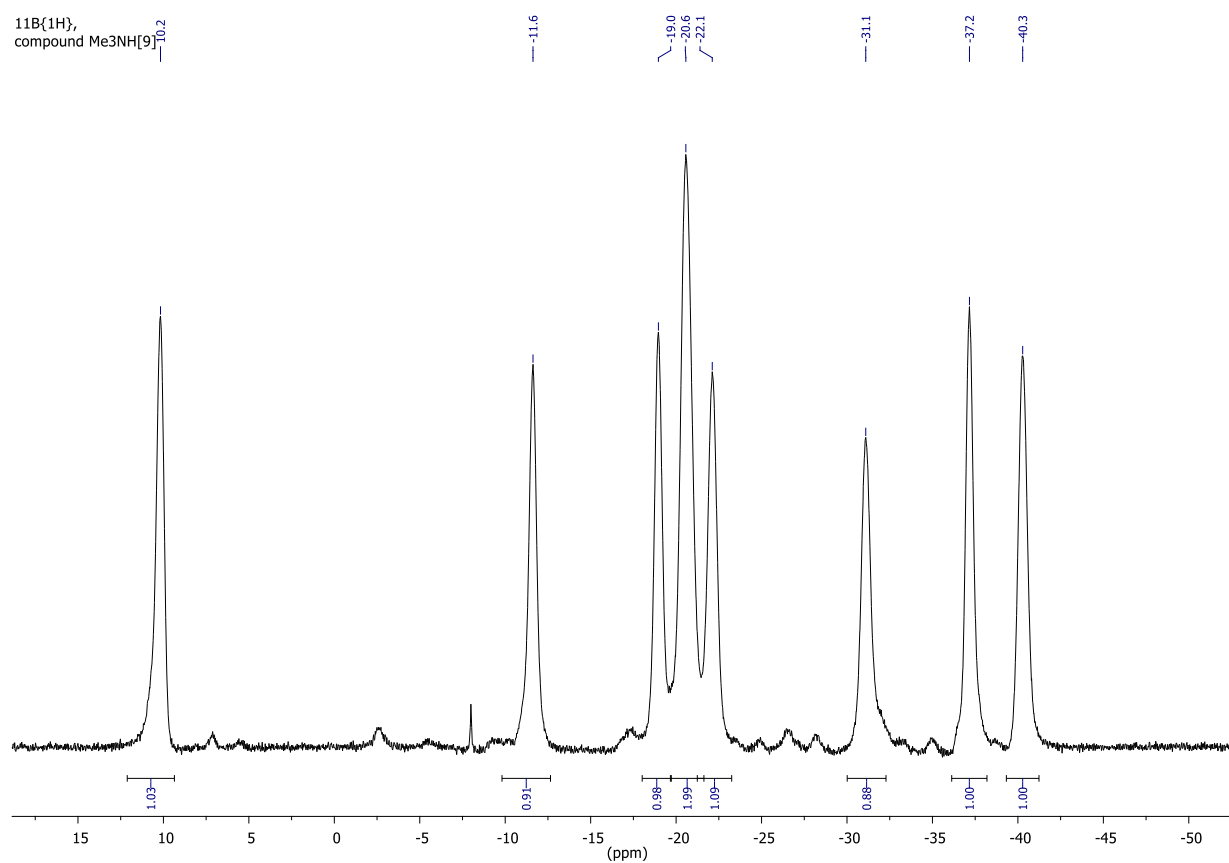


¹³C{¹H}, compound K[9]

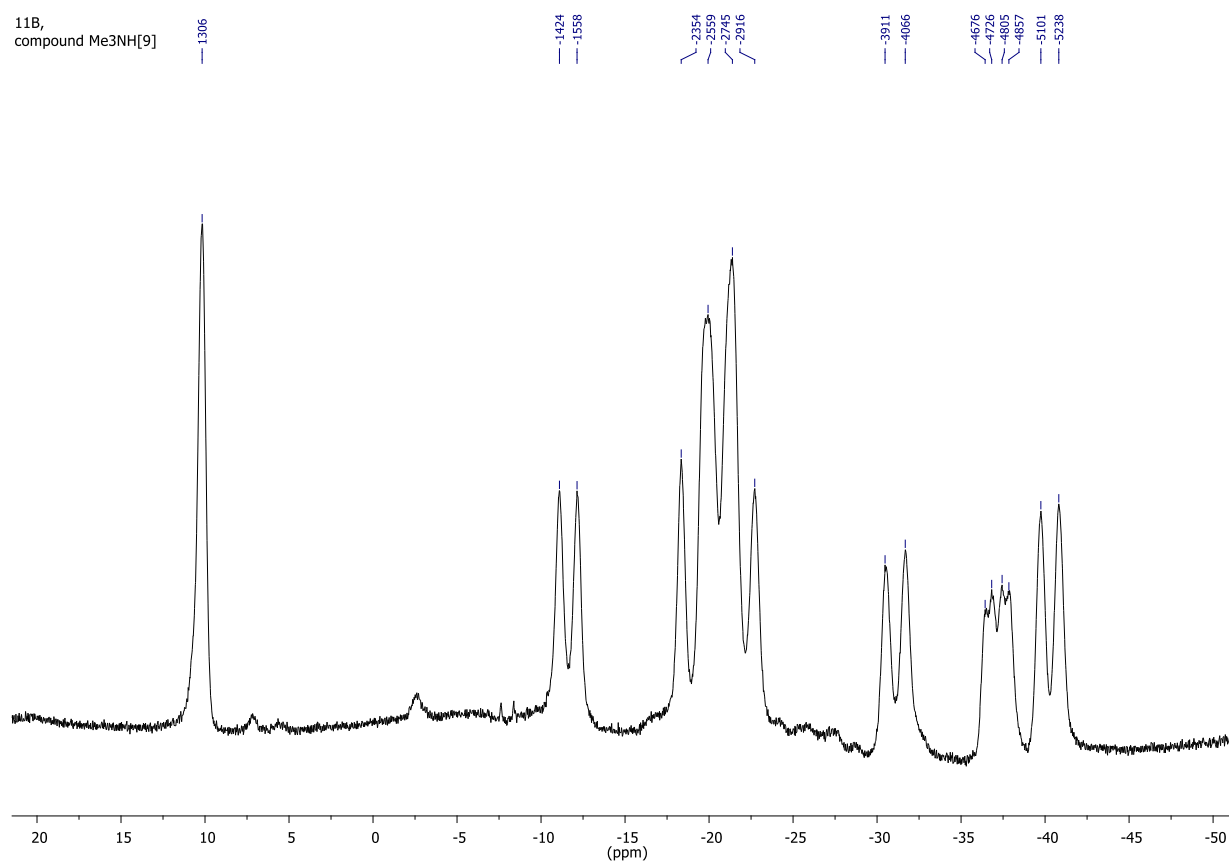




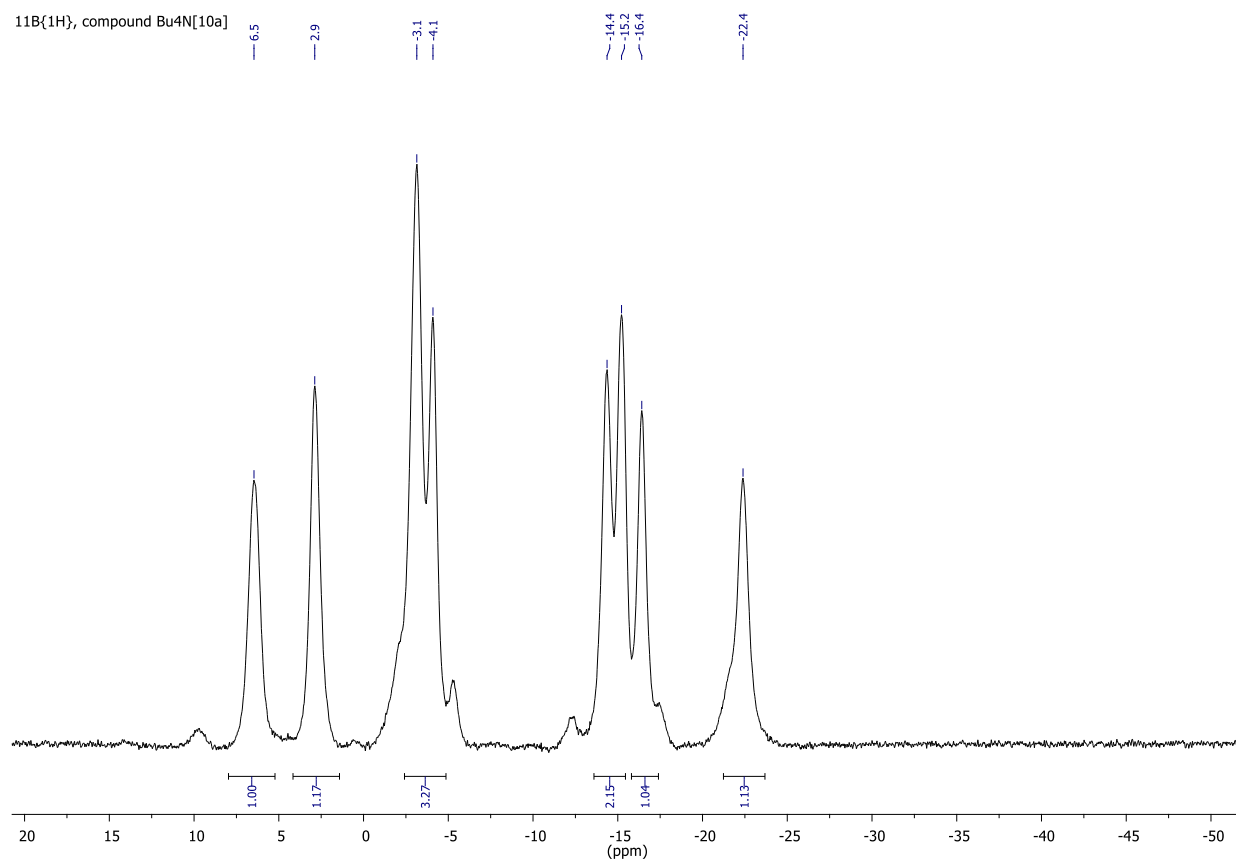
11B(1H),
compound Me3NH[9]



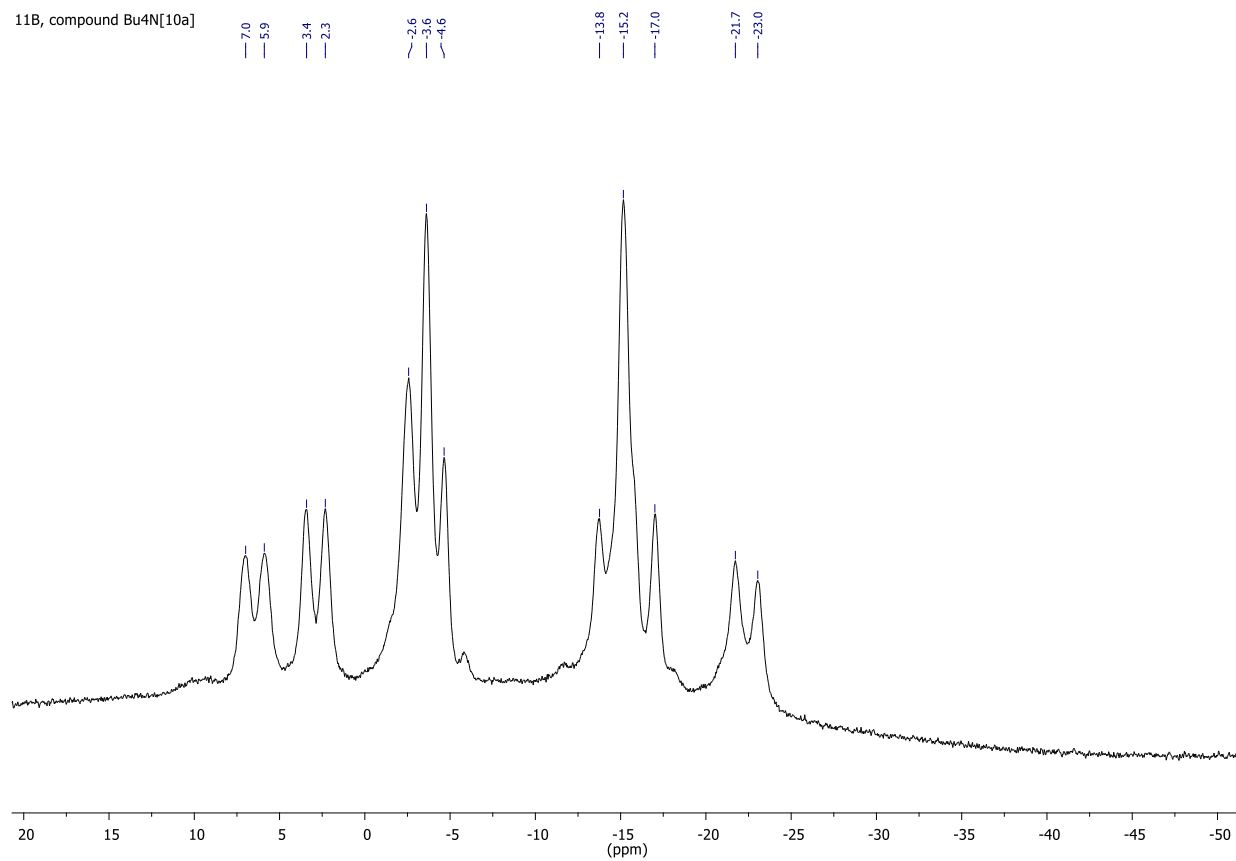
11B,
compound Me3NH[9]



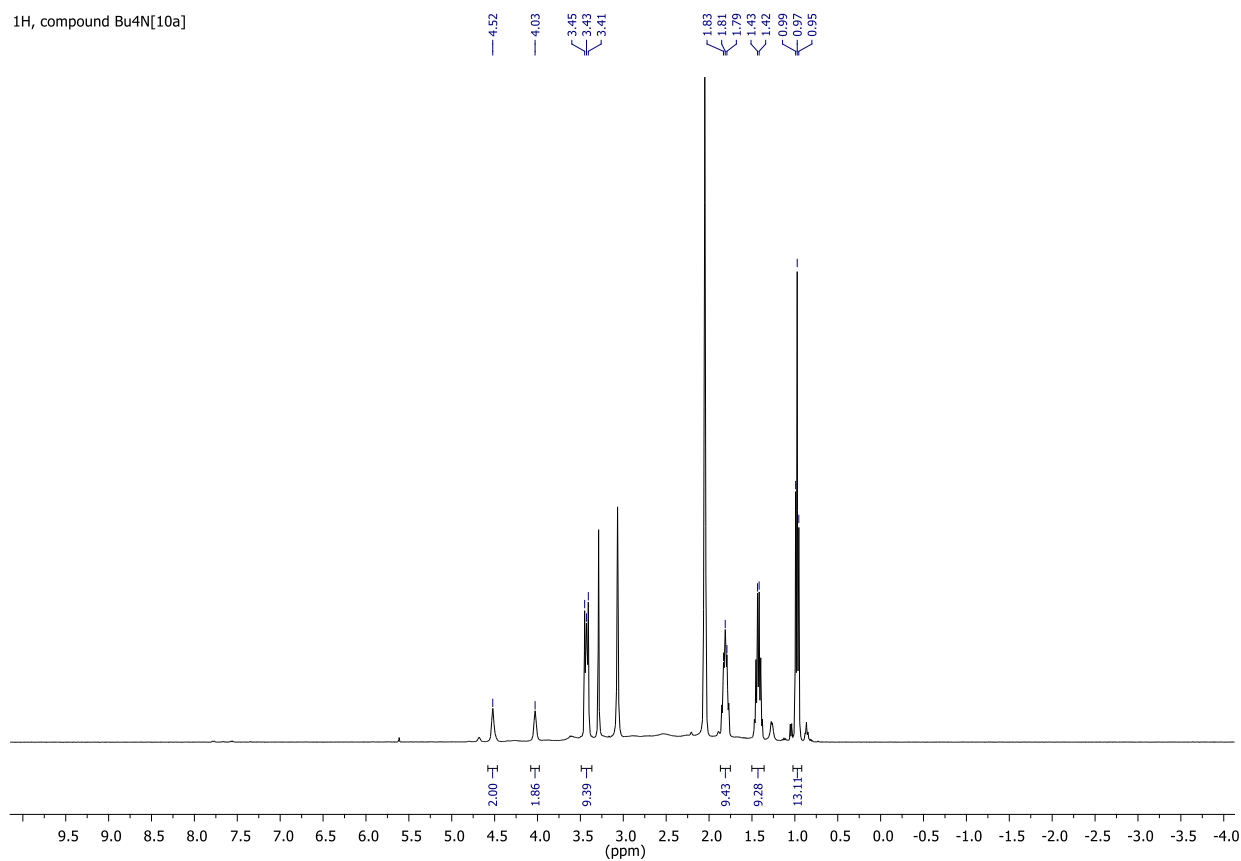
$^{11}\text{B}\{^1\text{H}\}$, compound Bu₄N[10a]



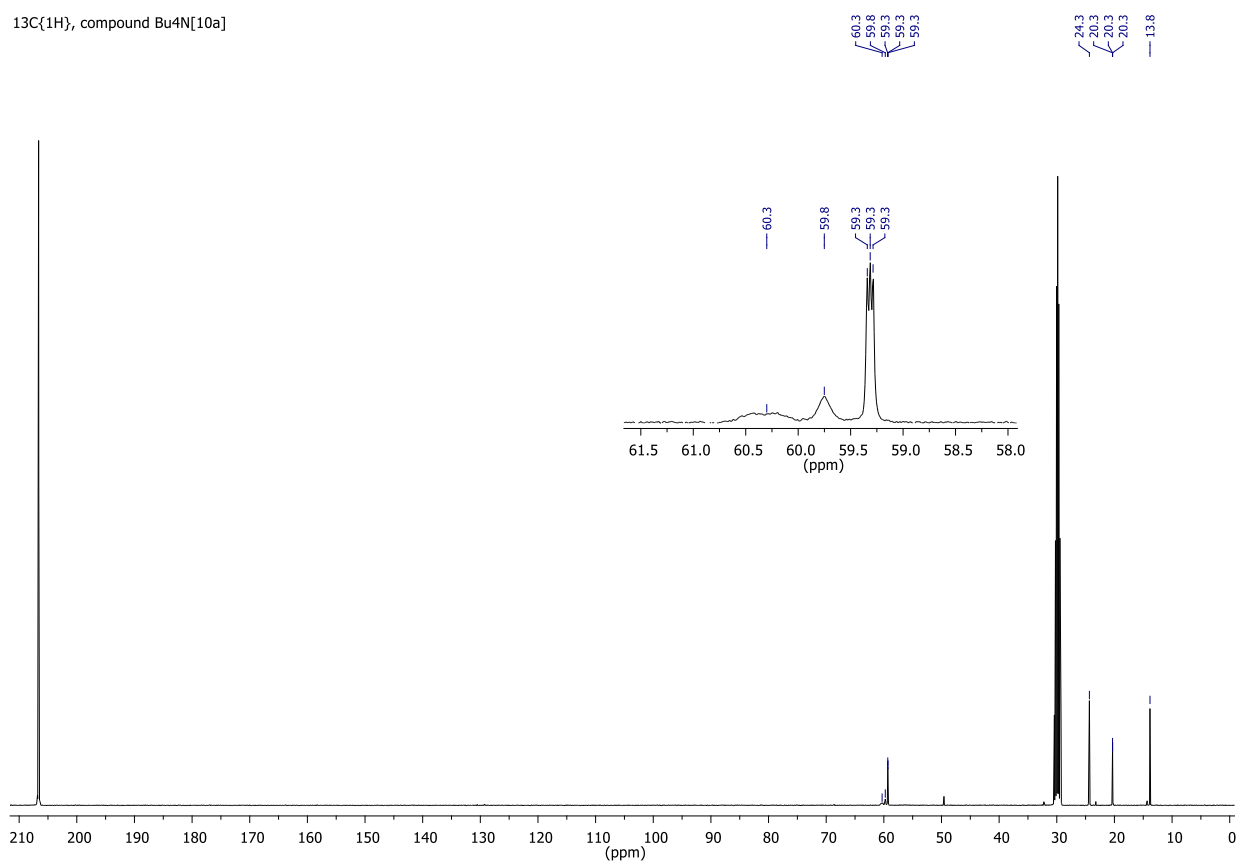
^{11}B , compound Bu₄N[10a]

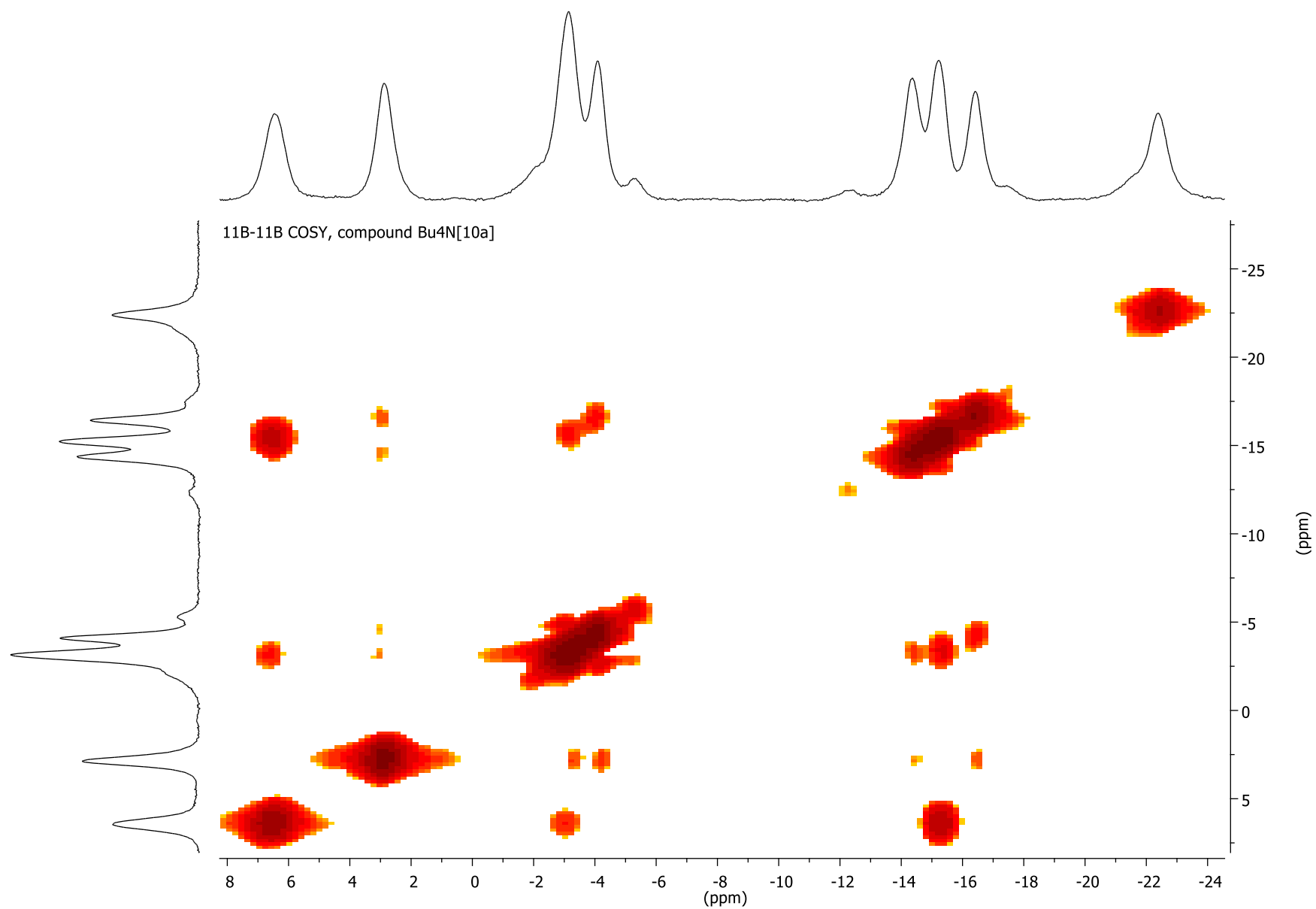


¹H, compound Bu4N[10a]

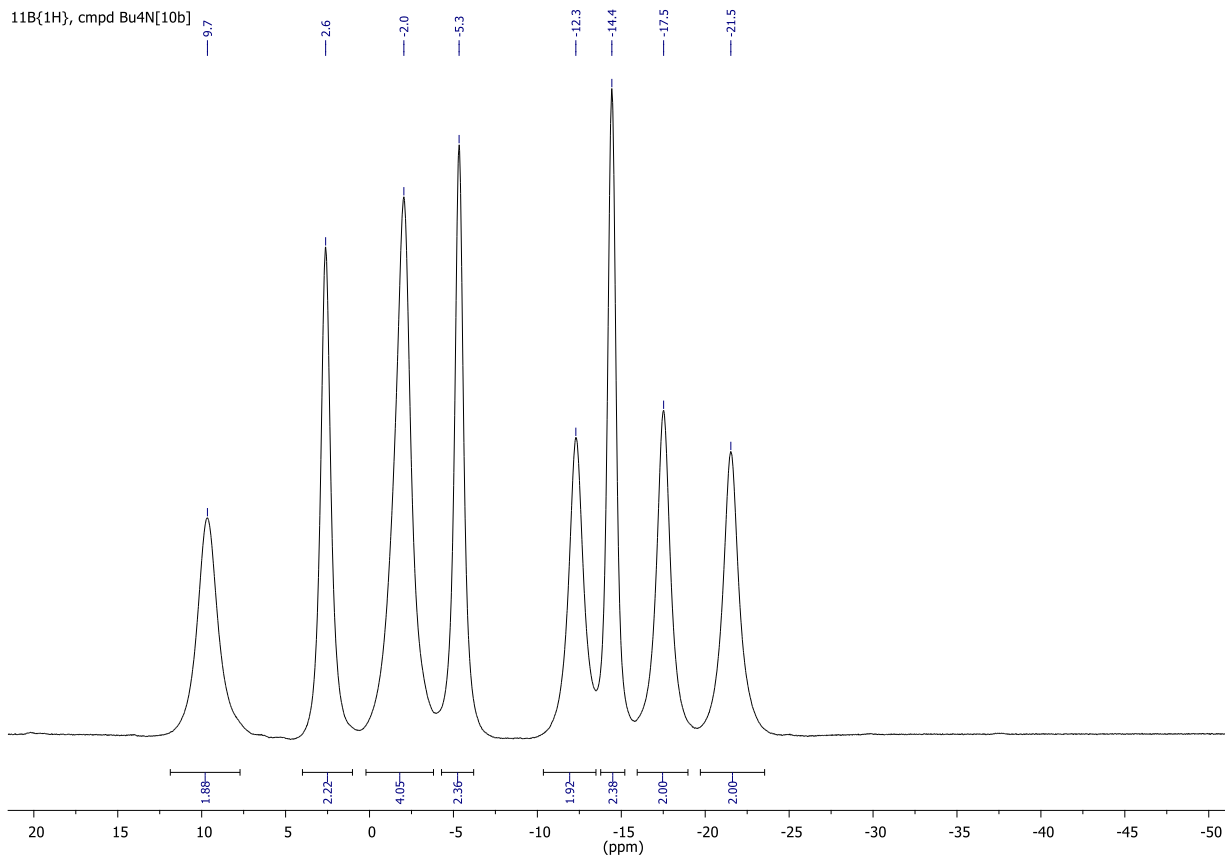


¹³C{¹H}, compound Bu4N[10a]

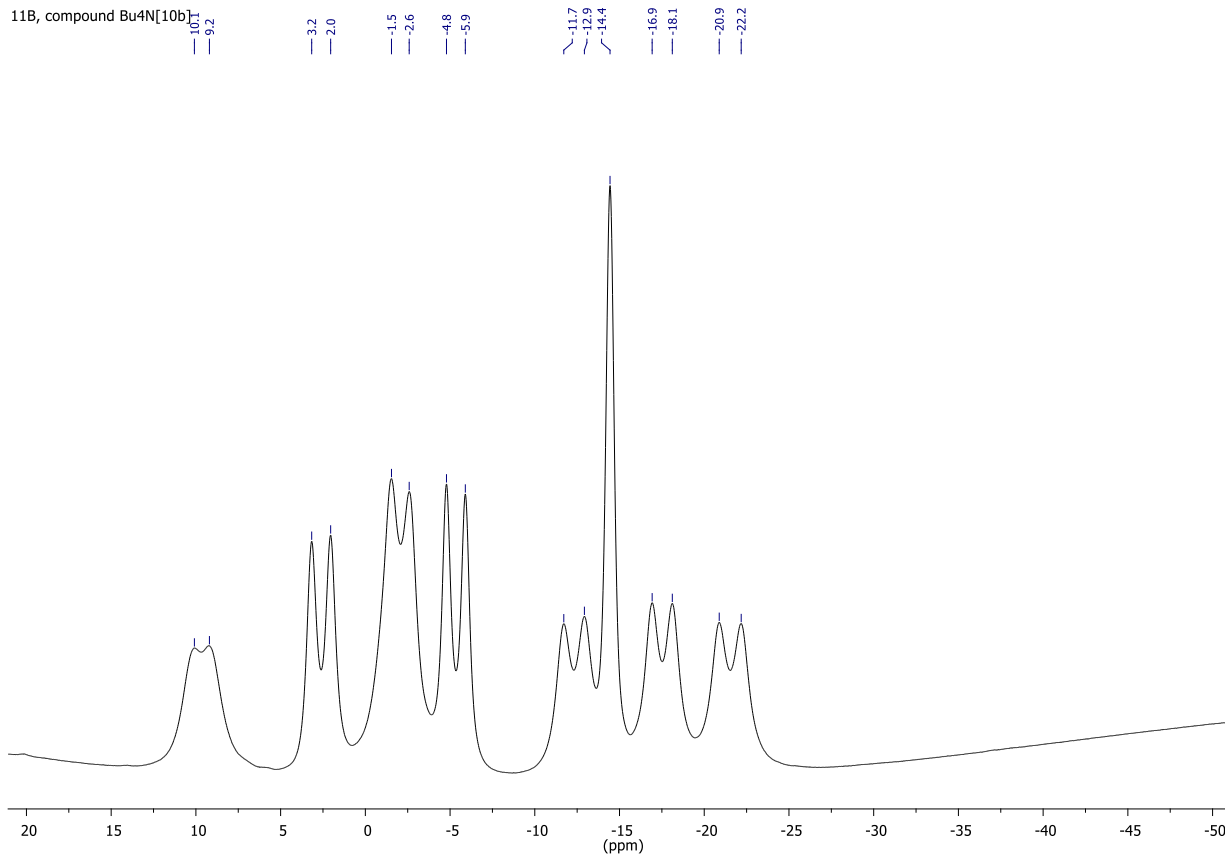




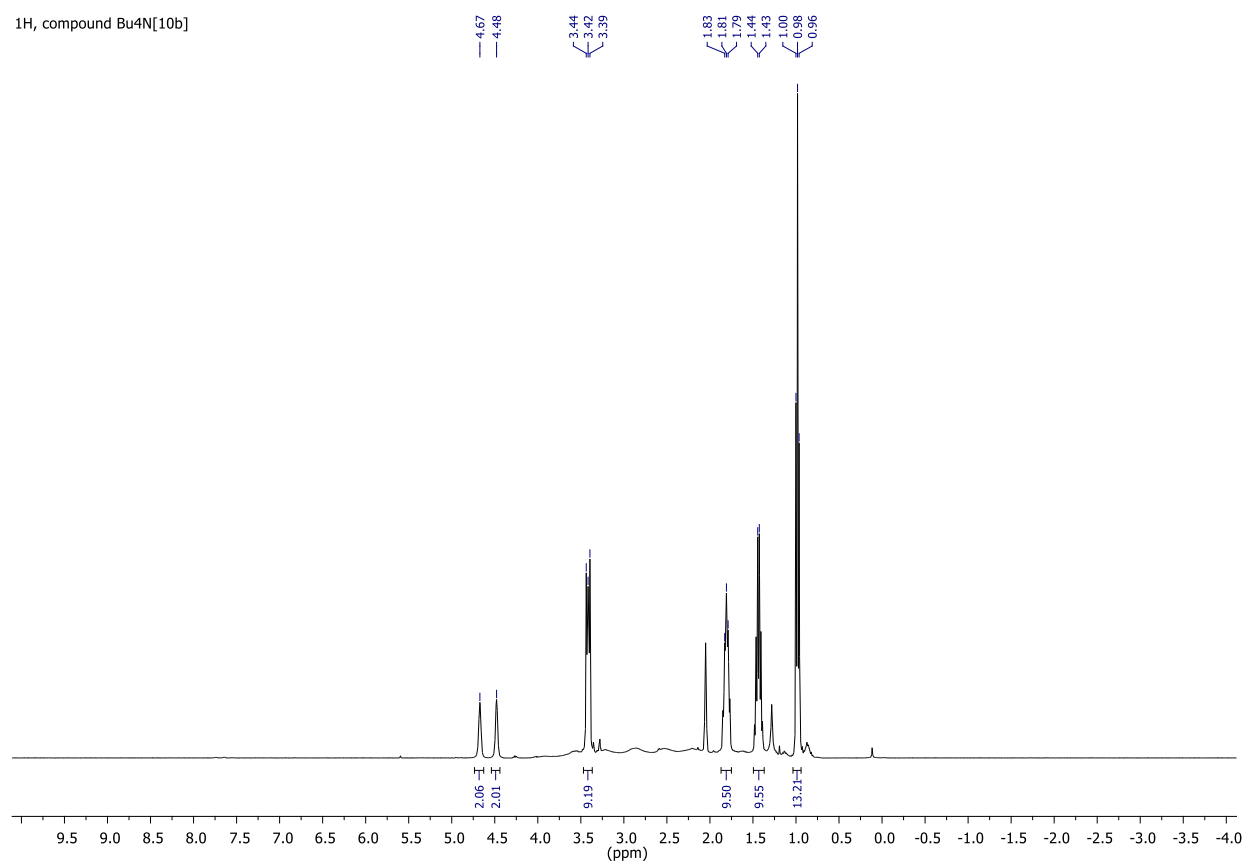
11B{1H}, compd Bu4N[10b]



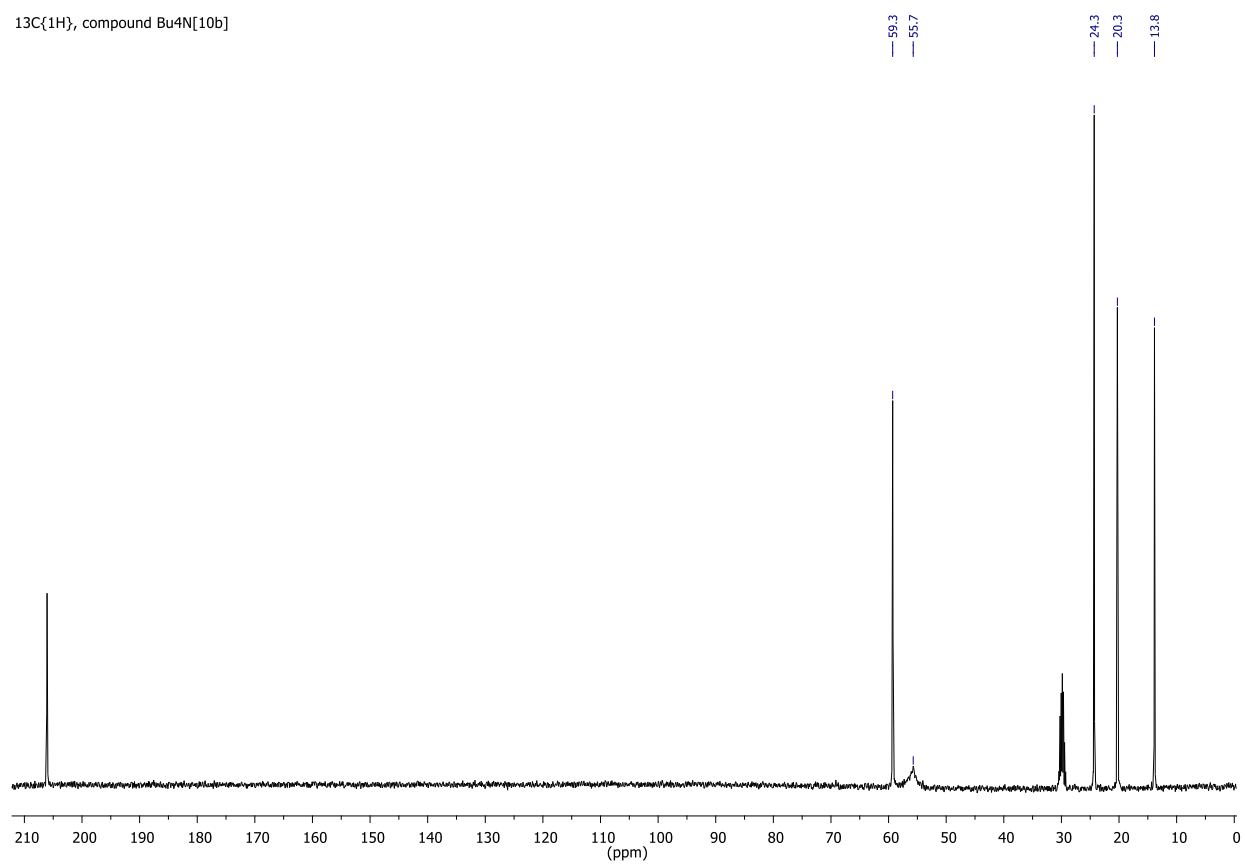
11B, compound Bu4N[10b]

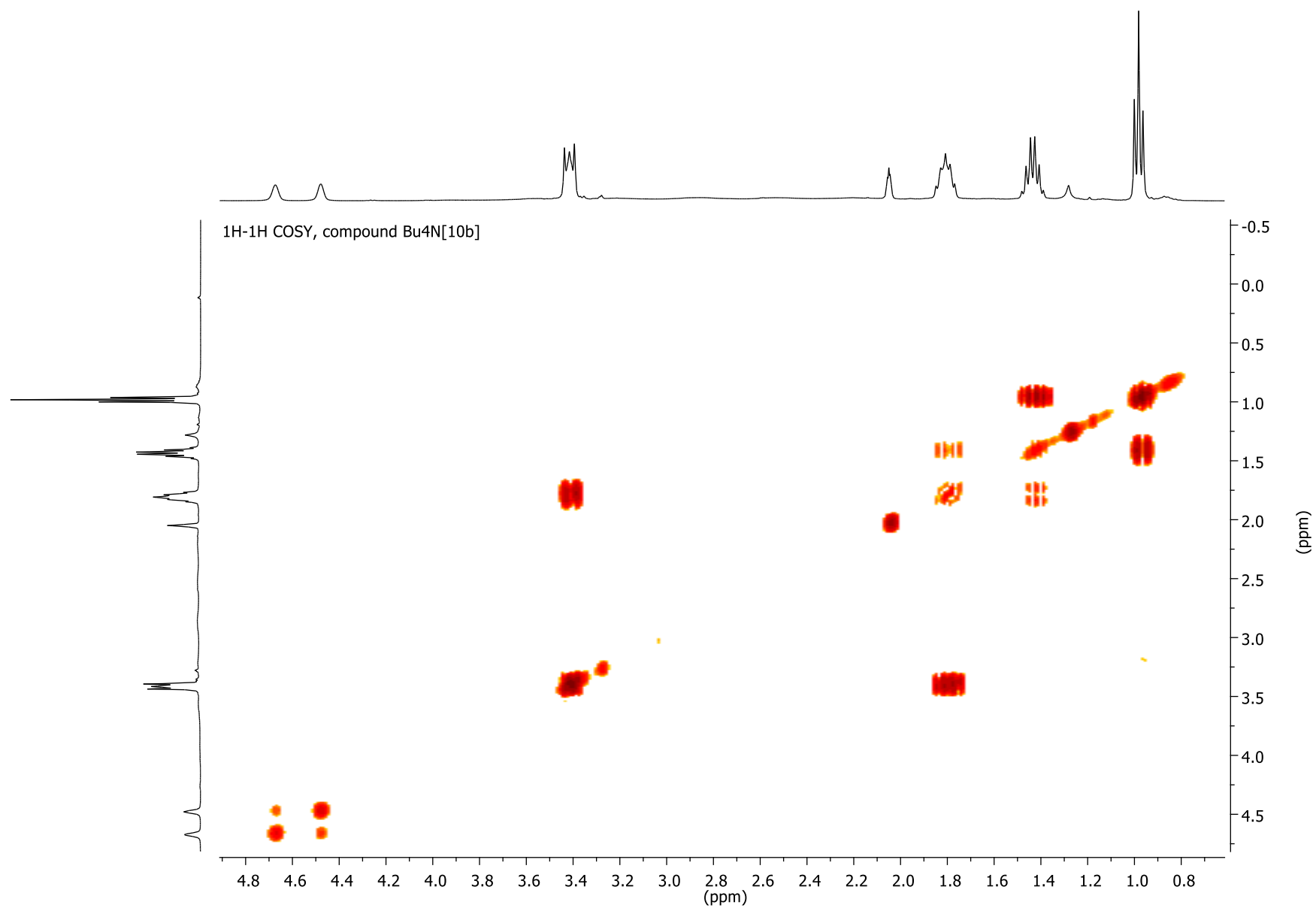


^1H , compound Bu4N[10b]

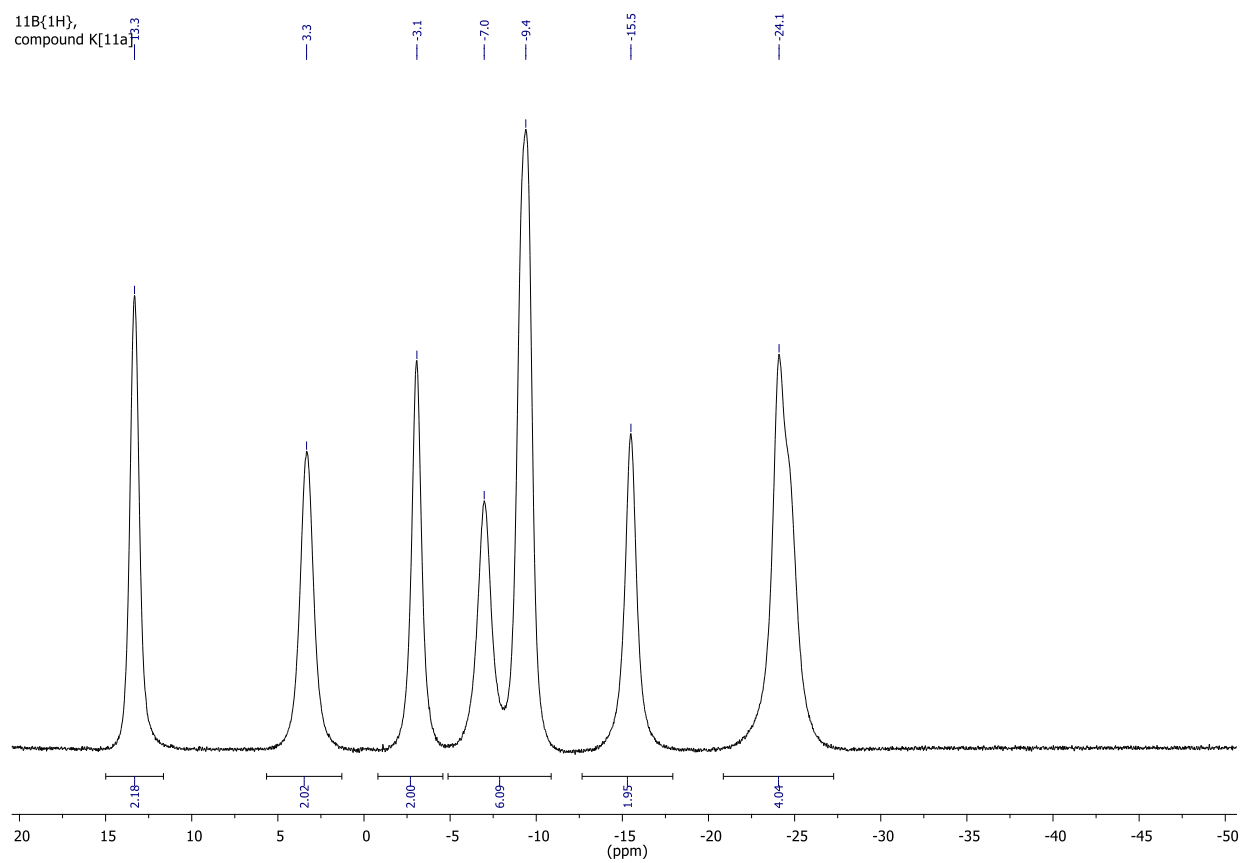


$^{13}\text{C}\{^1\text{H}\}$, compound Bu4N[10b]

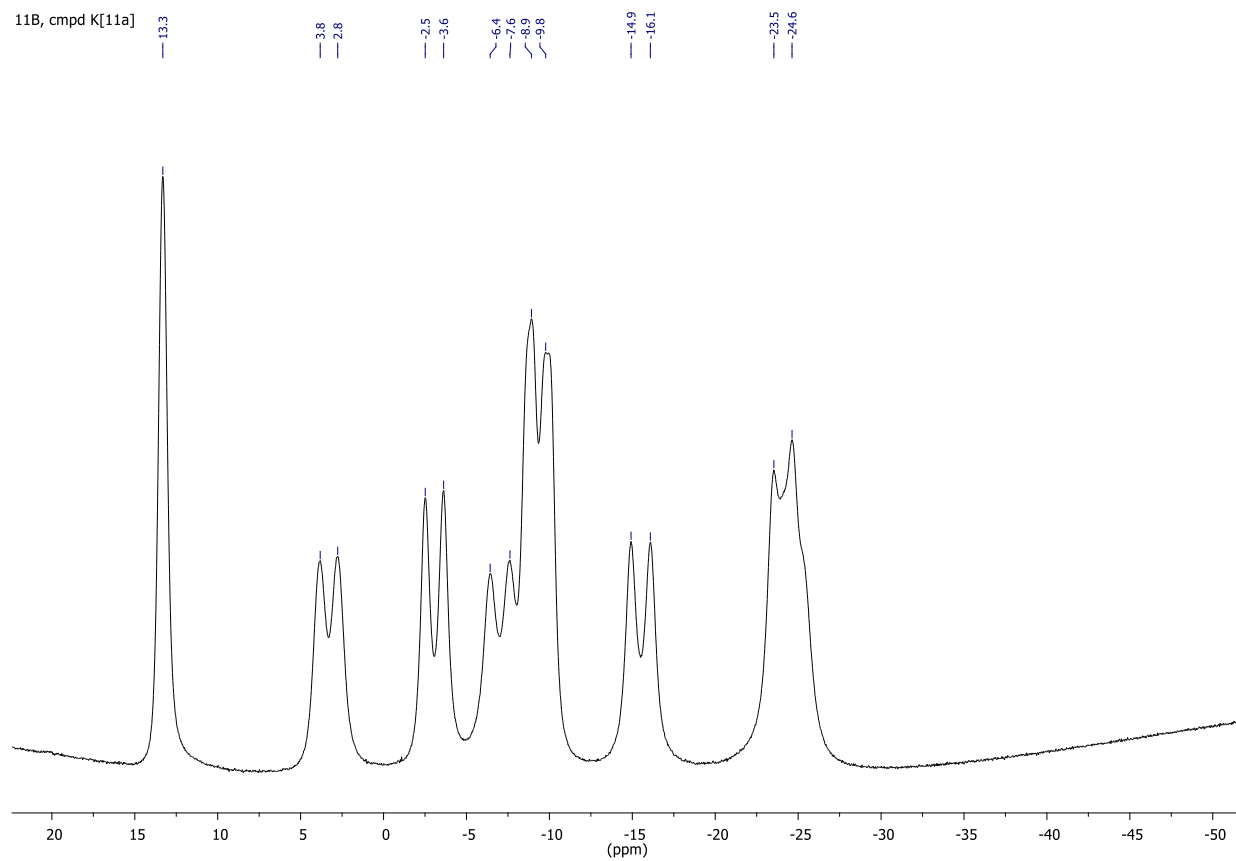




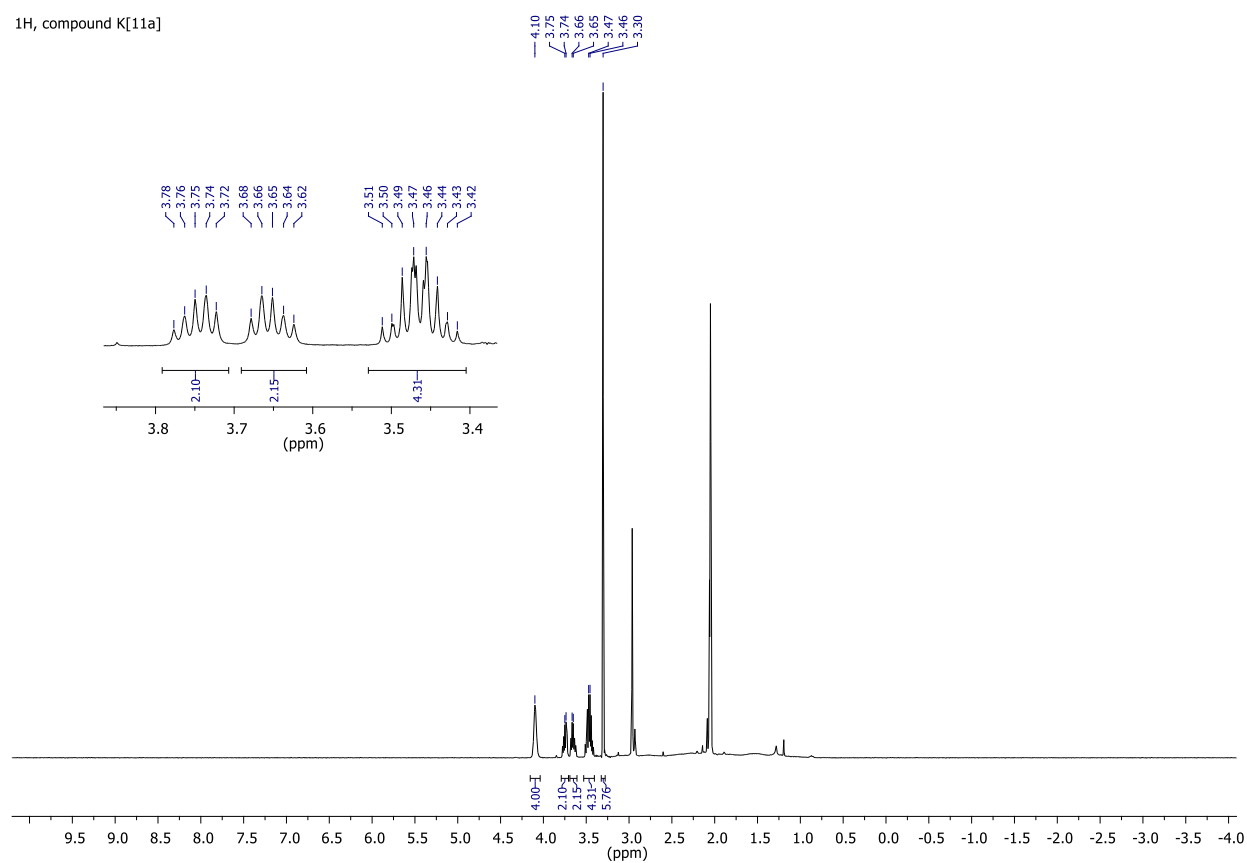
11B(1H),
compound K[11a]



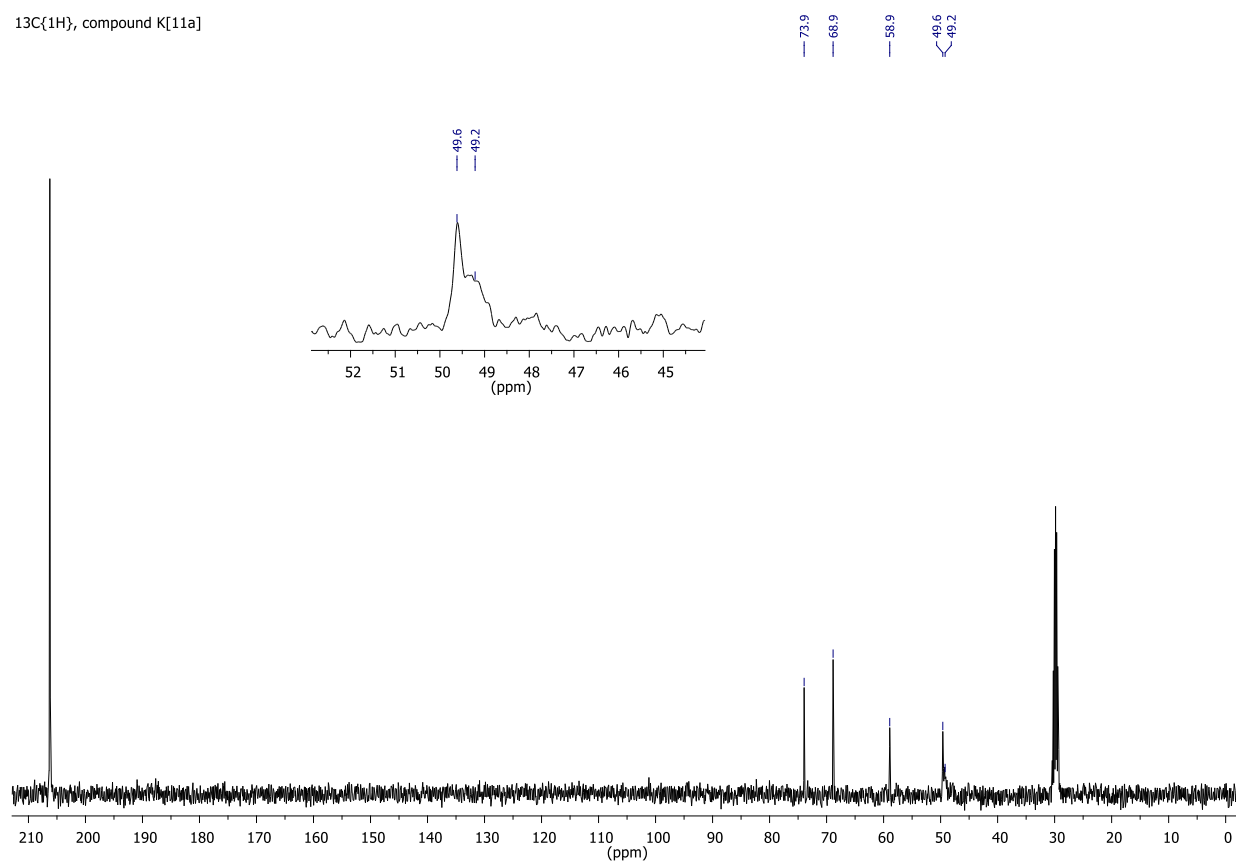
11B, cmpd K[11a]

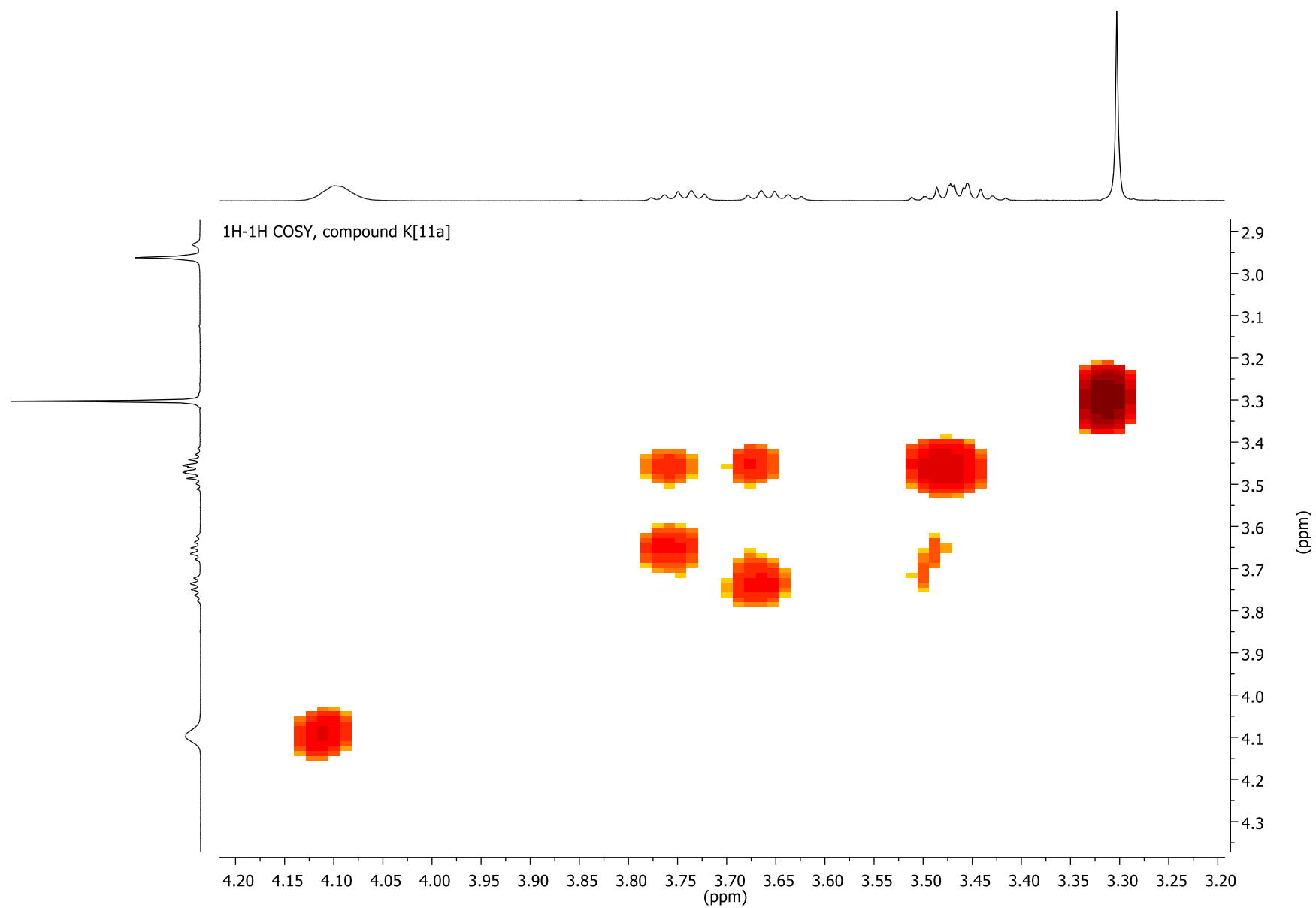


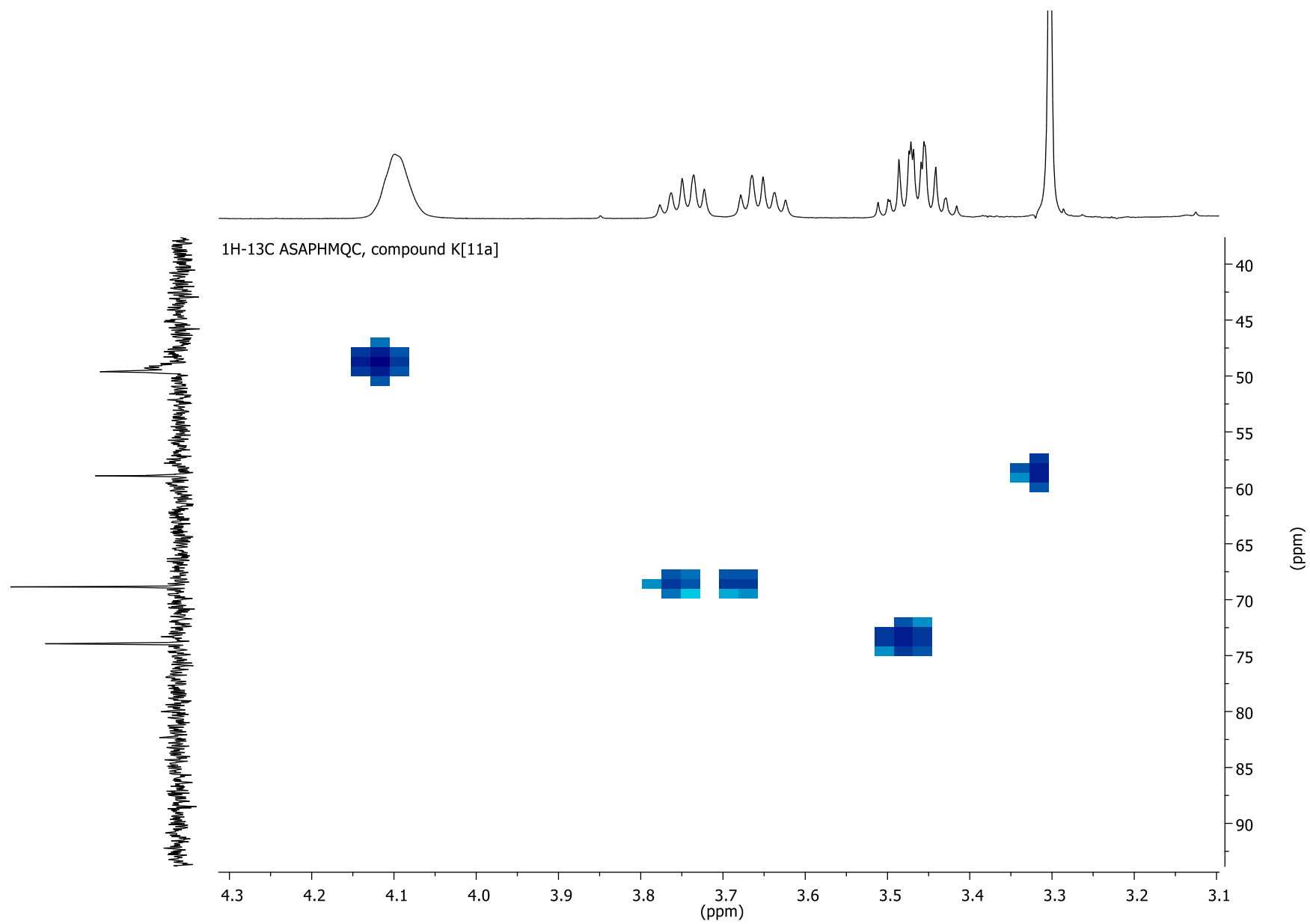
¹H, compound K[11a]

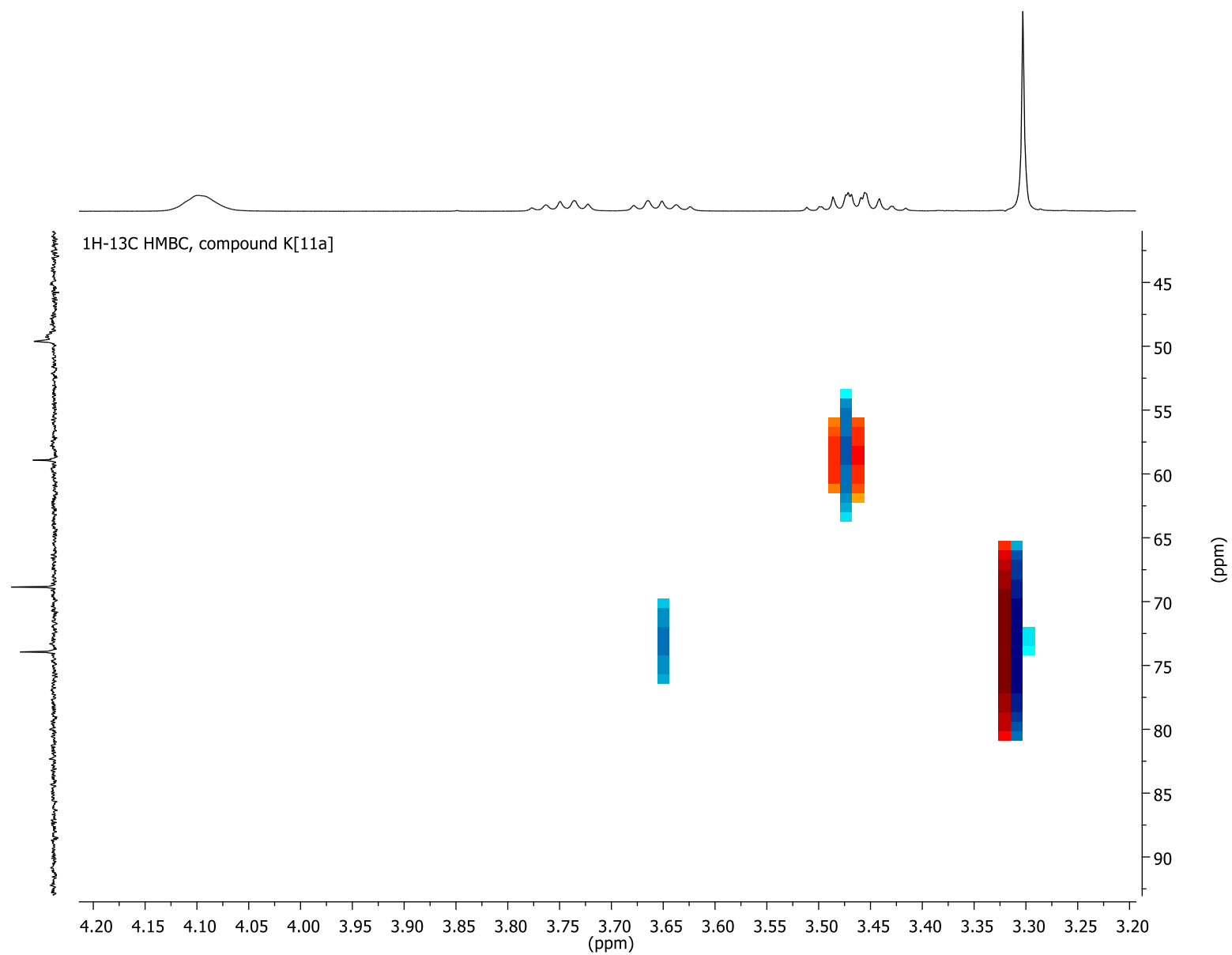


¹³C{¹H}, compound K[11a]

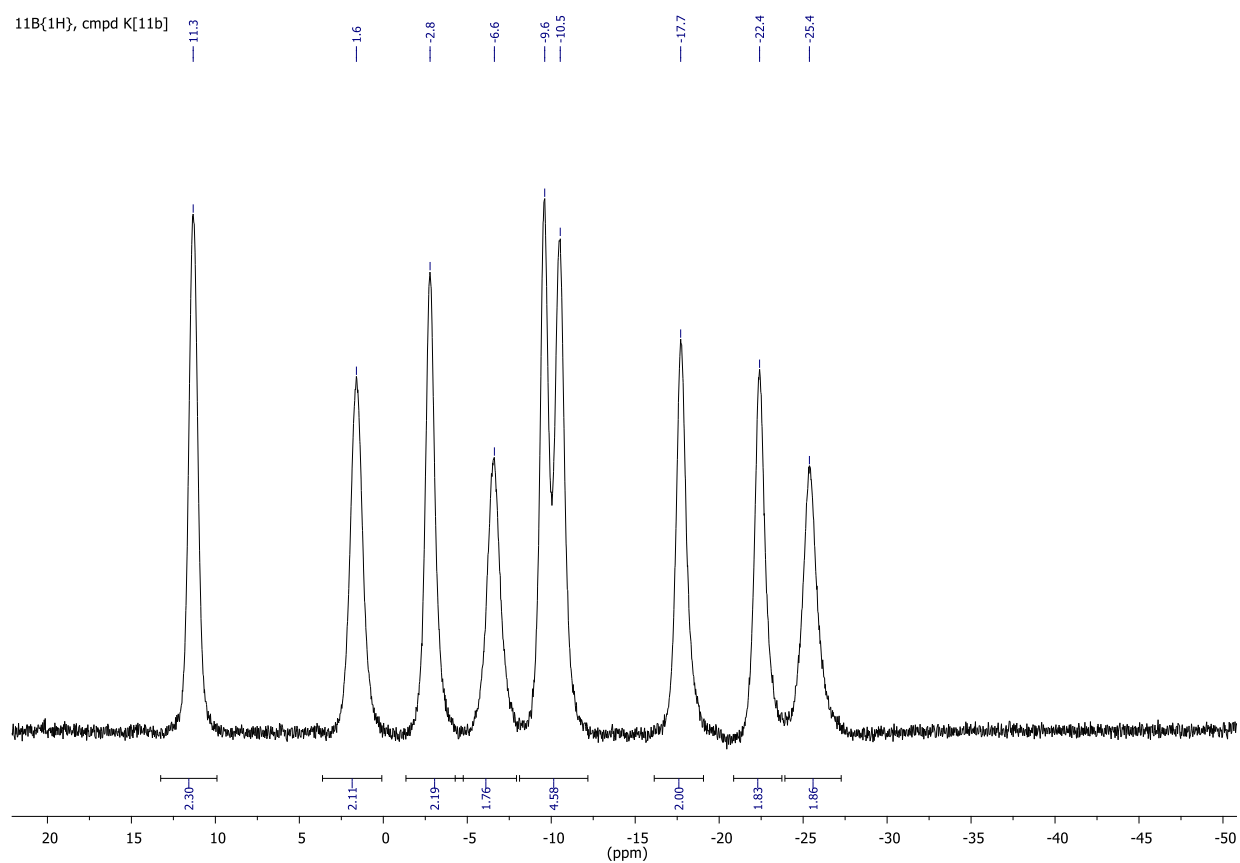




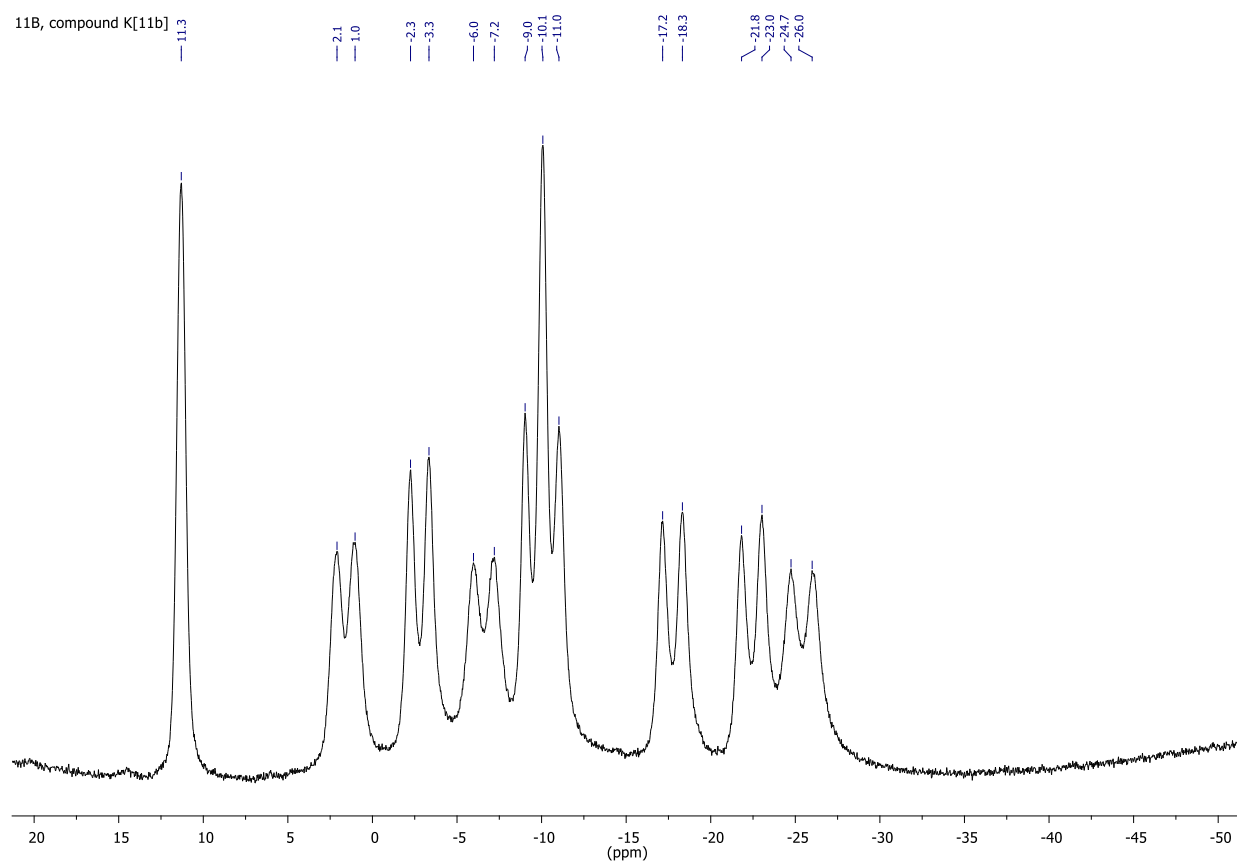




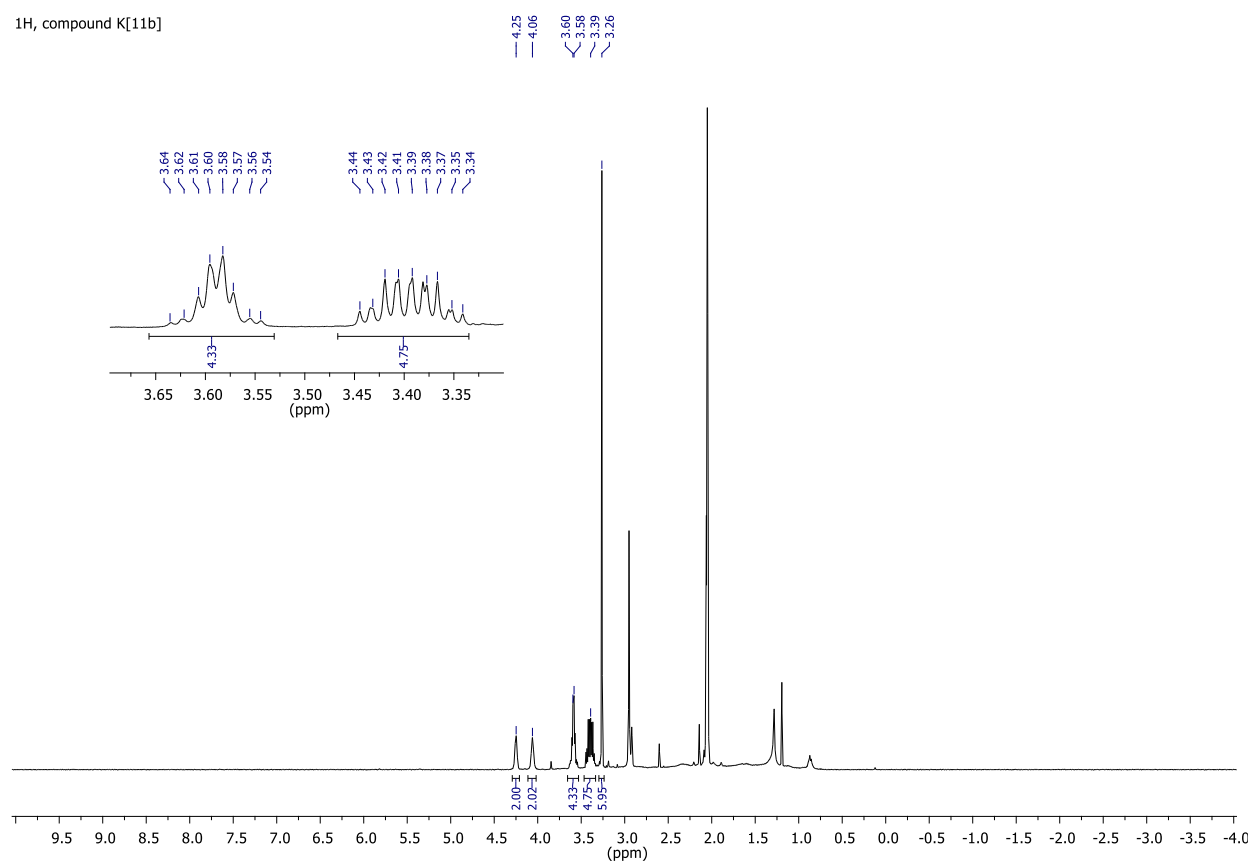
11B{1H}, compd K[11b]



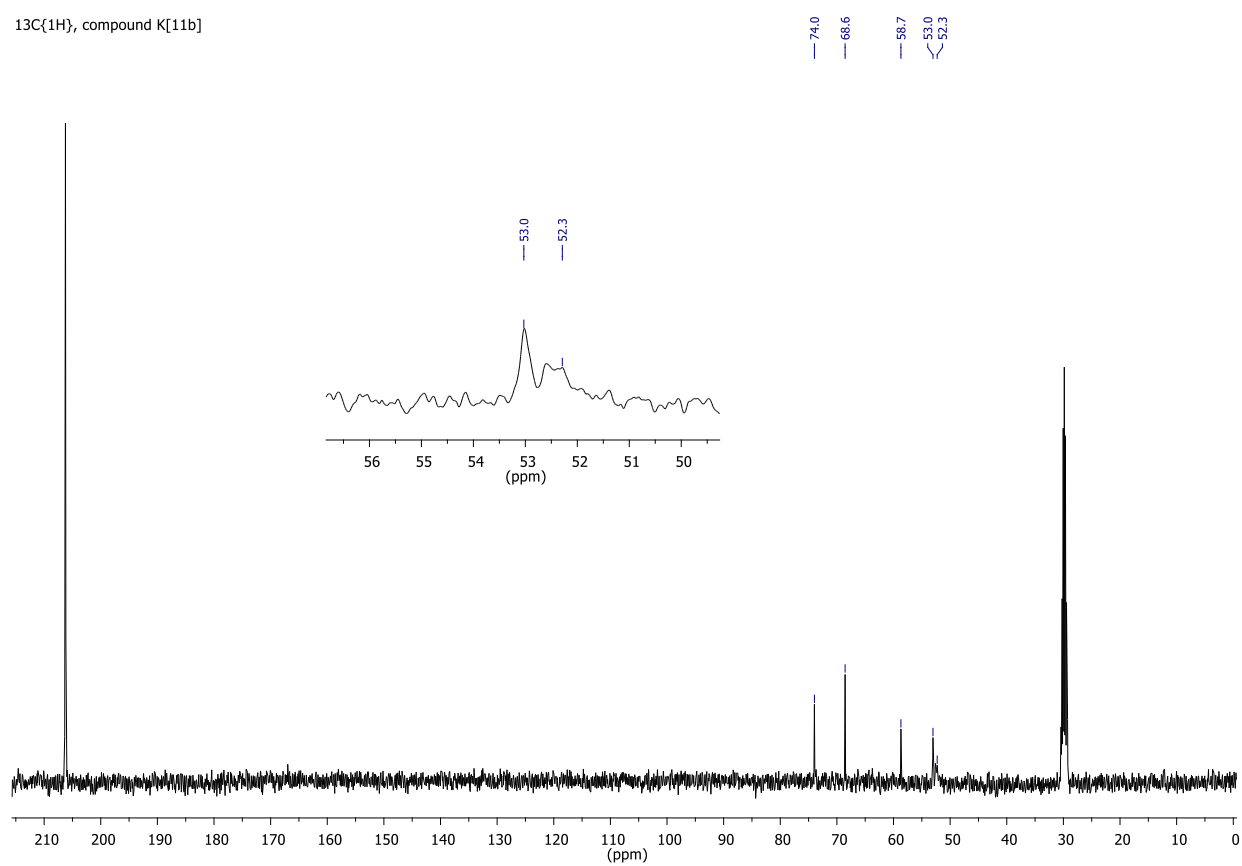
11B, compound K[11b]

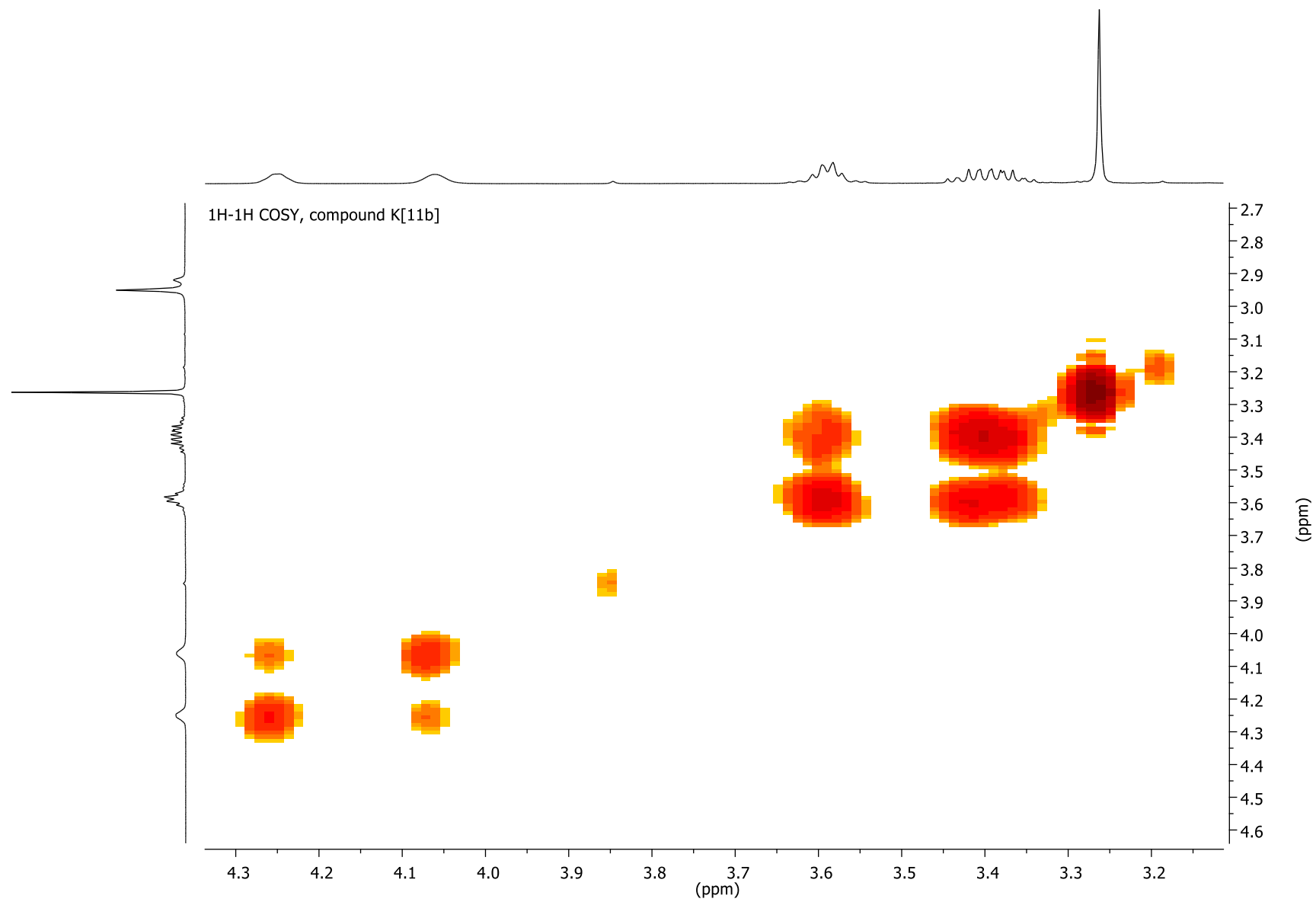


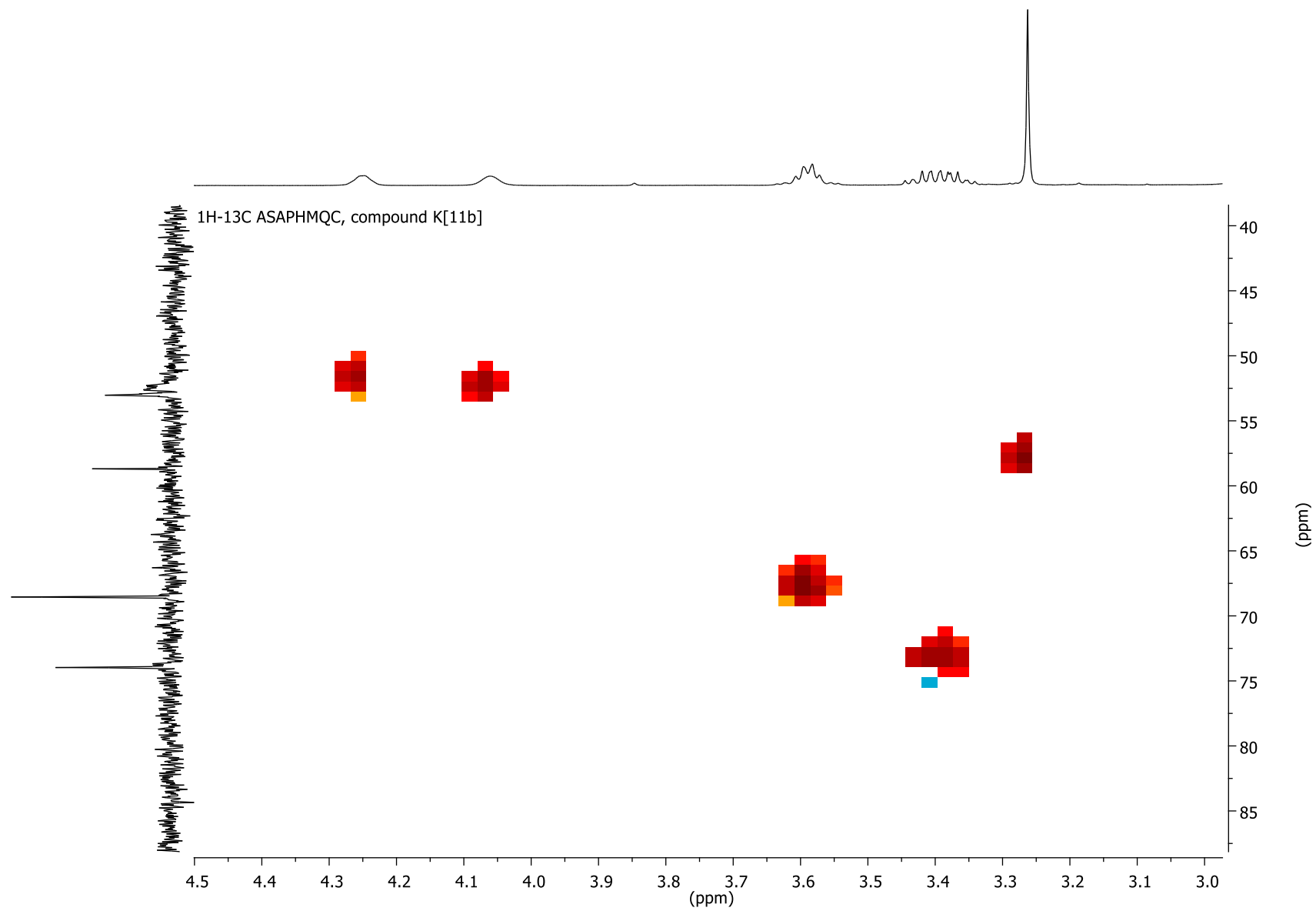
¹H, compound K[11b]

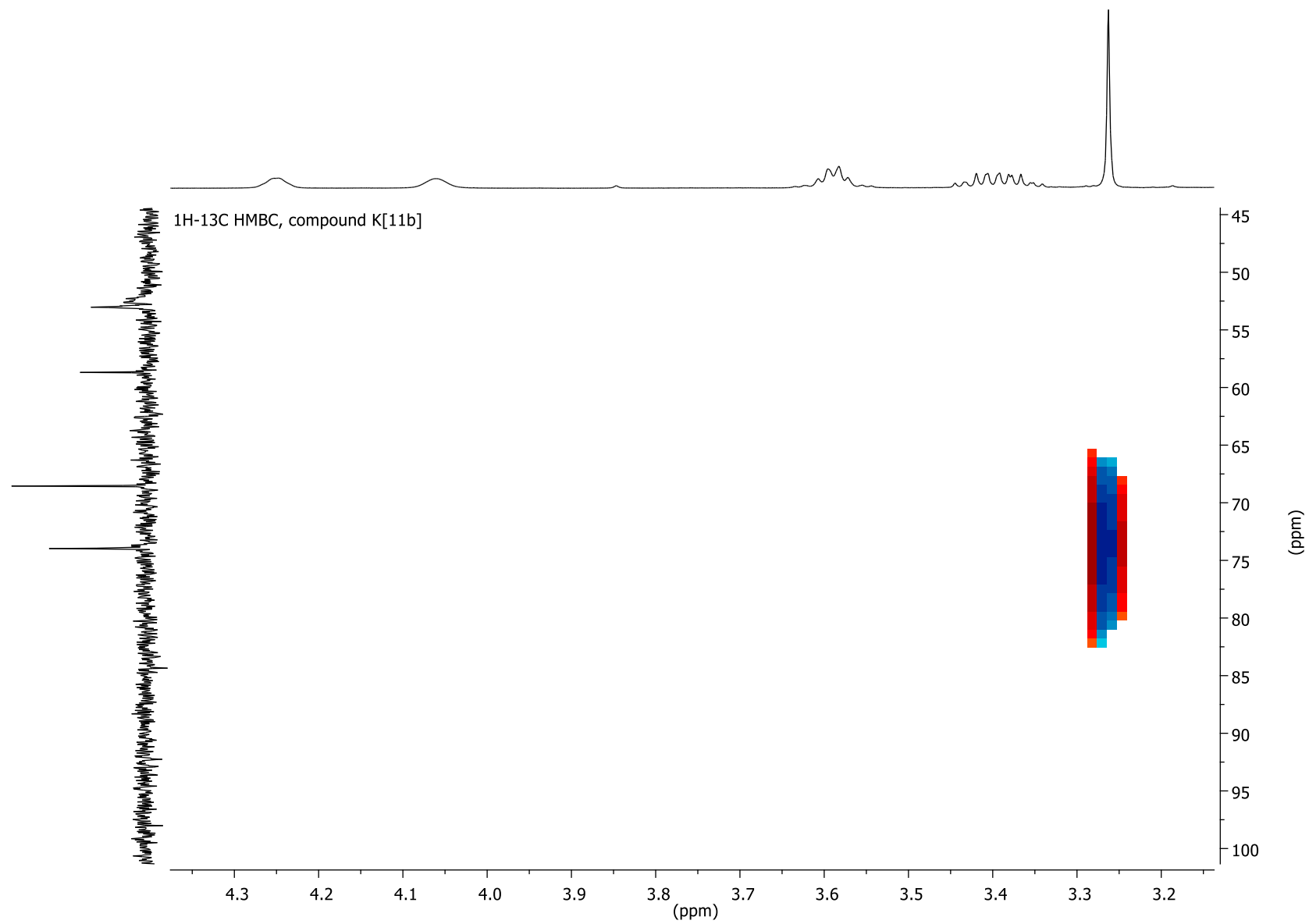


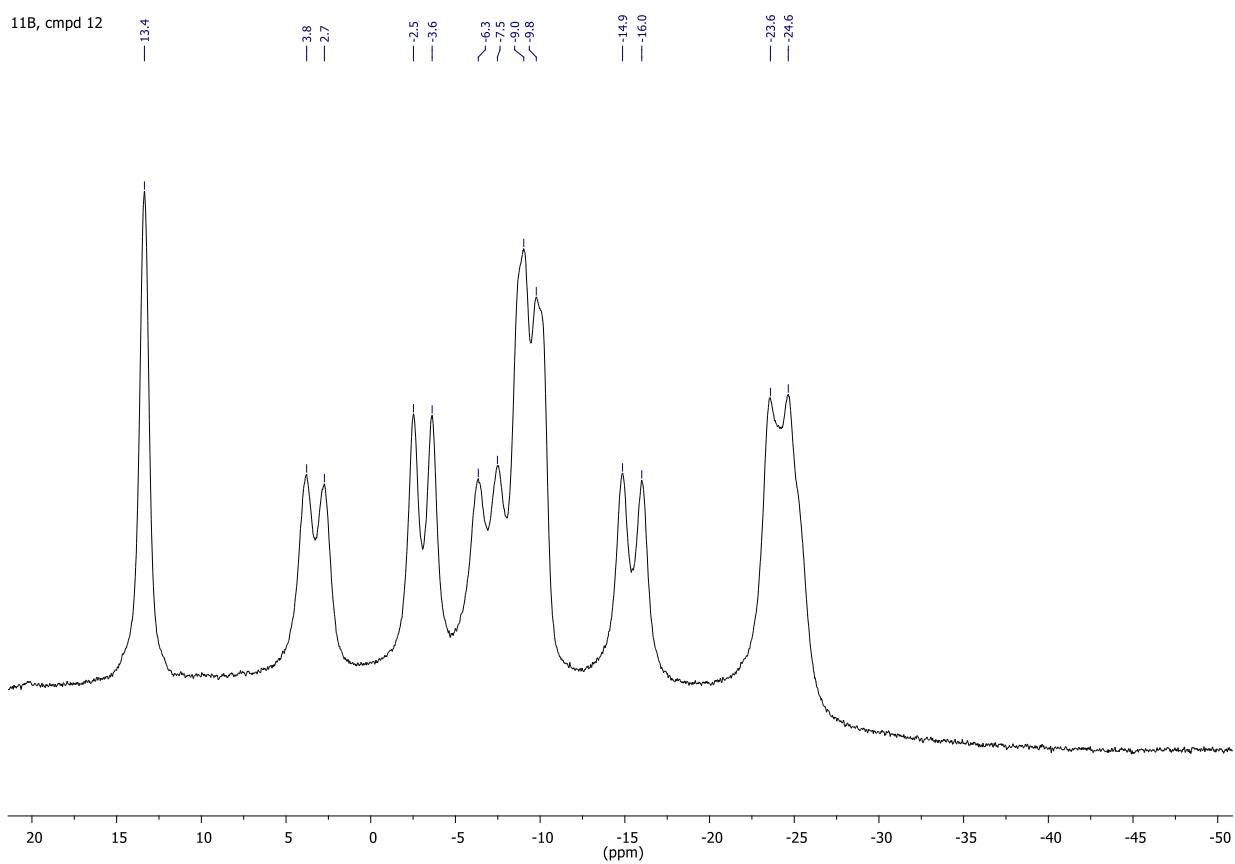
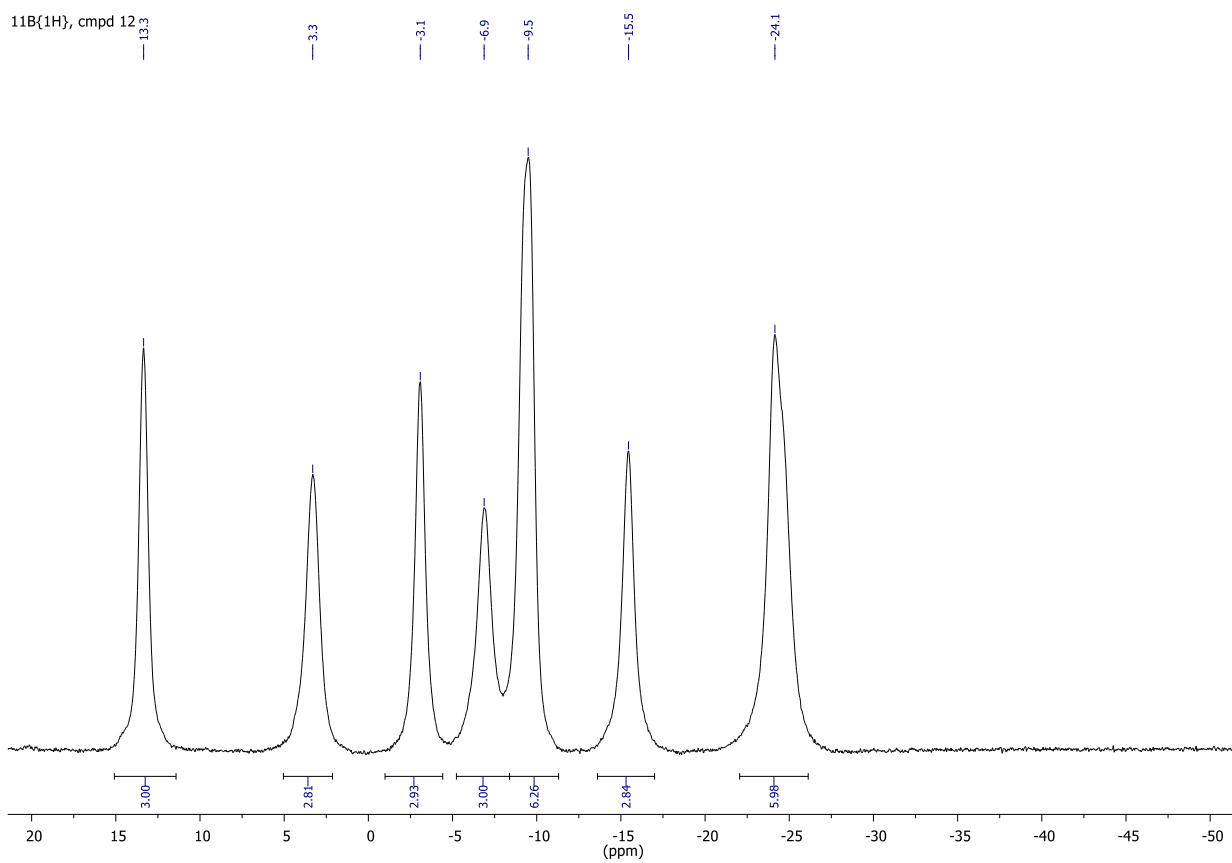
¹³C{¹H}, compound K[11b]



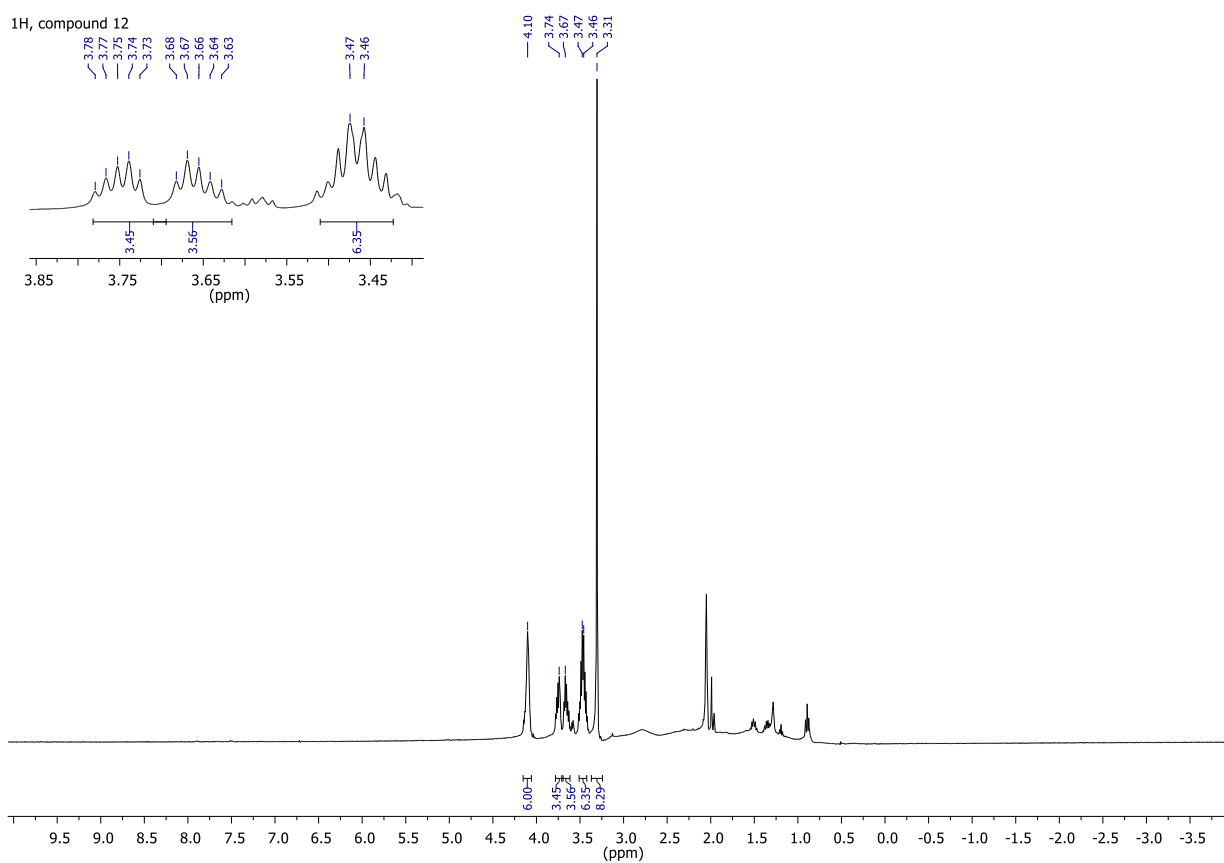




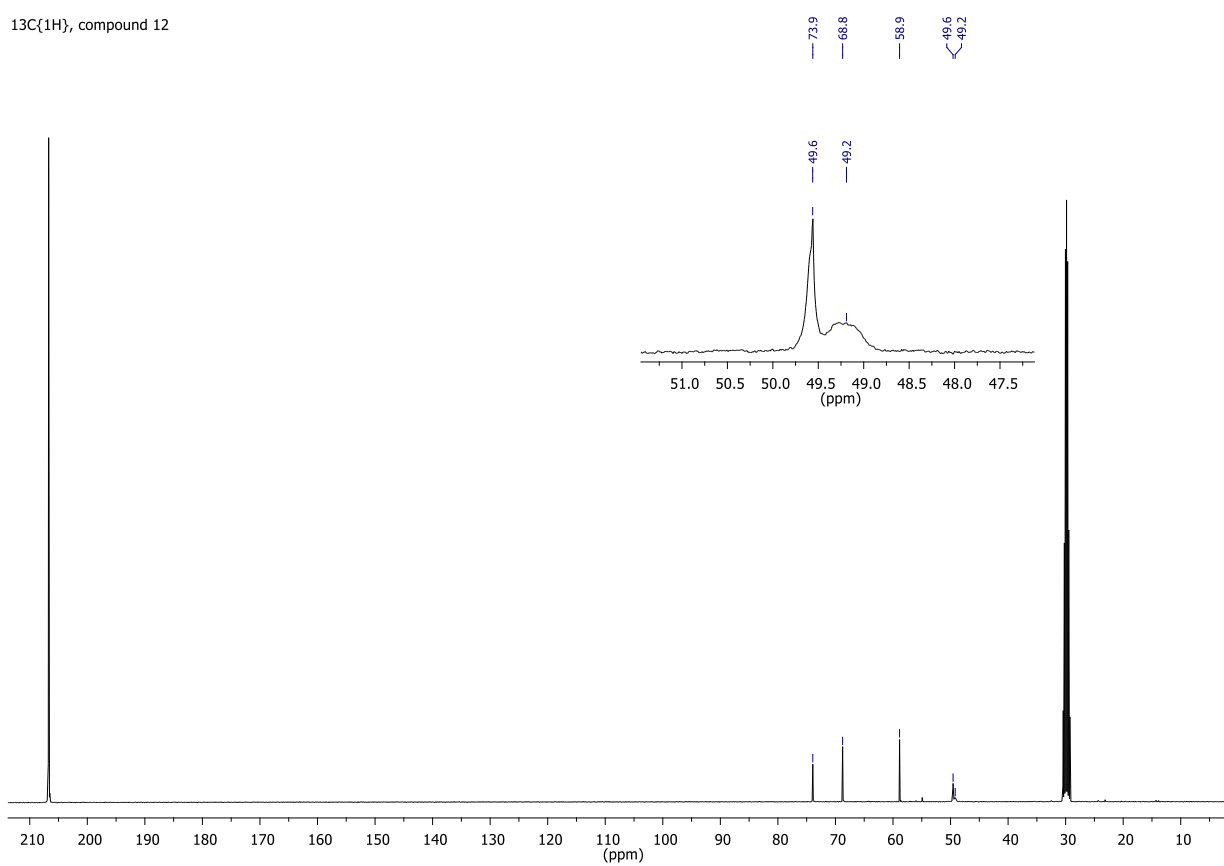


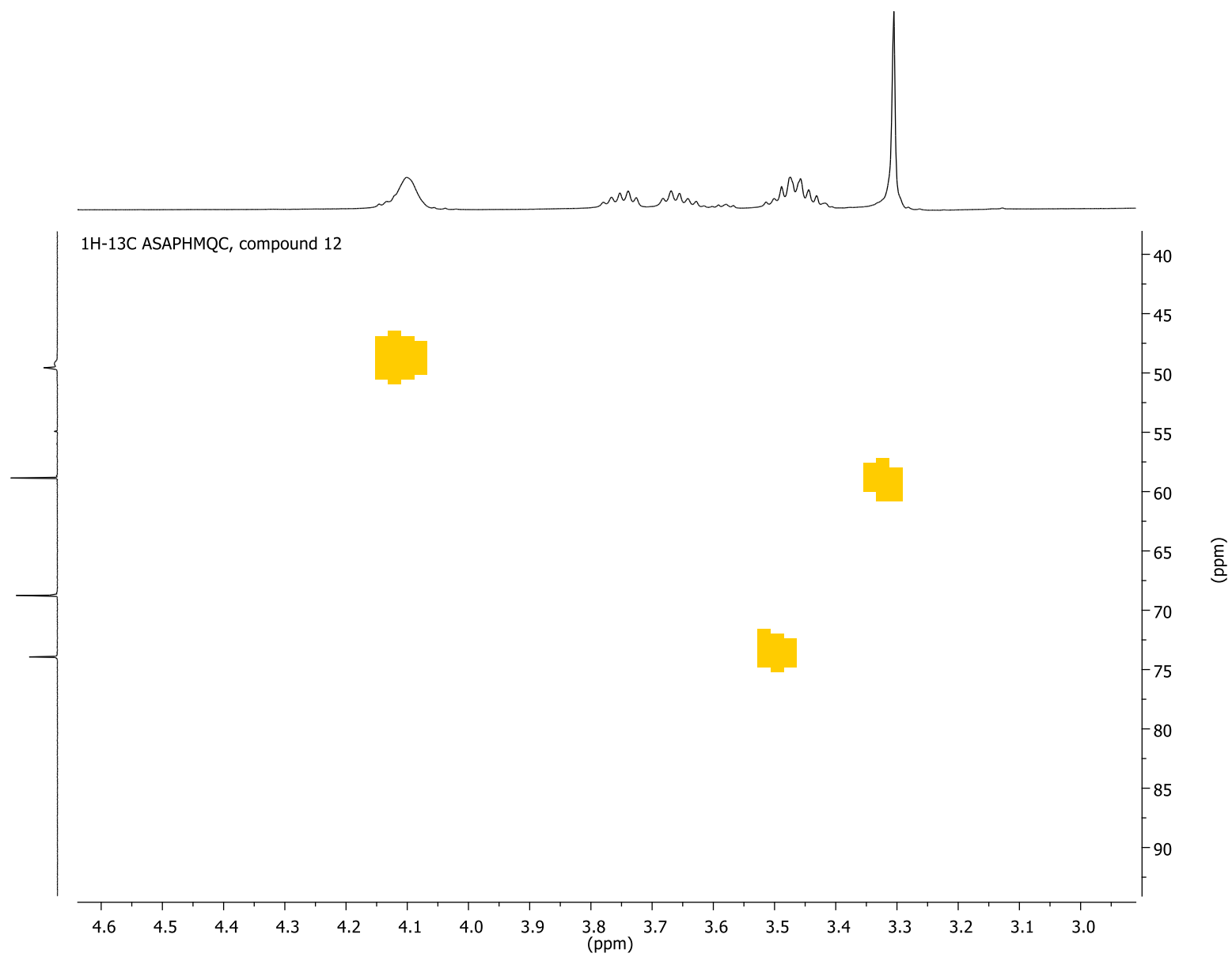


¹H, compound 12

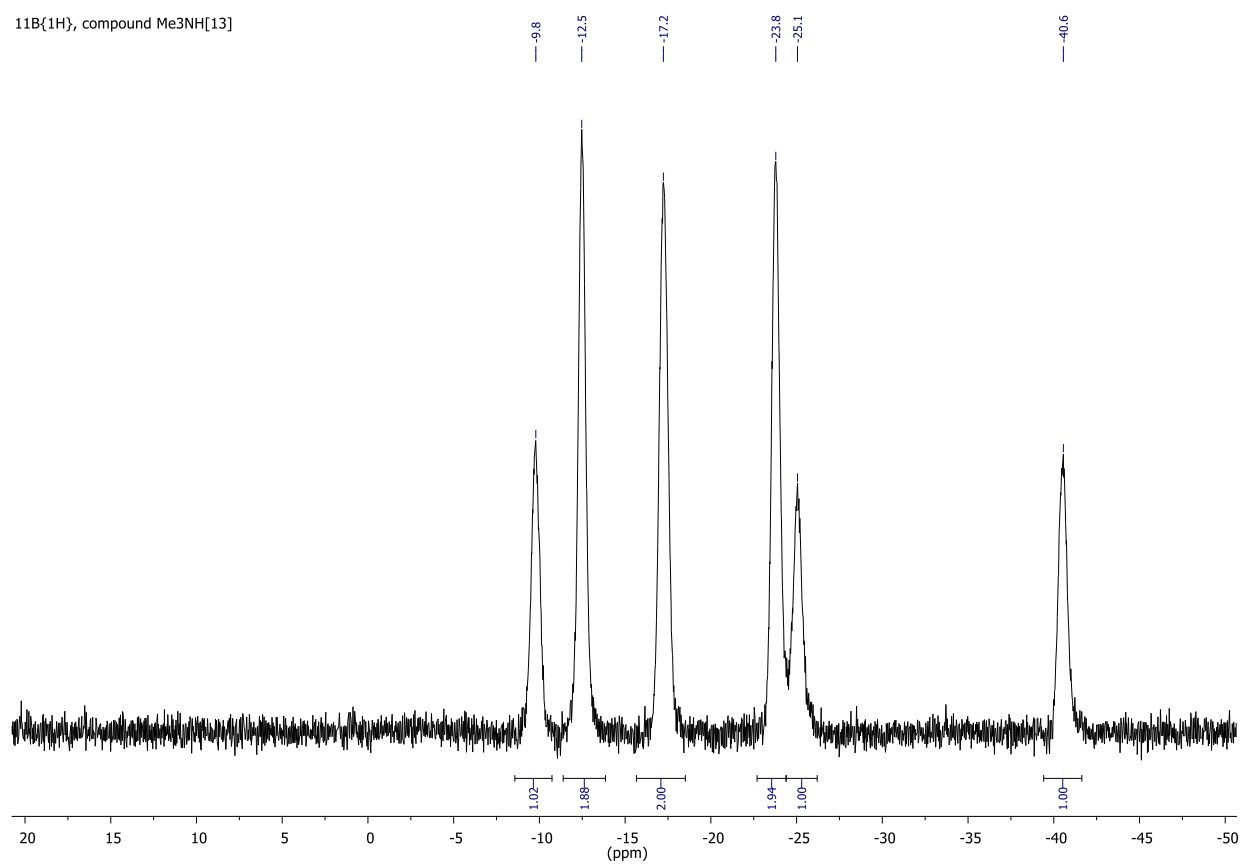


¹³C{¹H}, compound 12

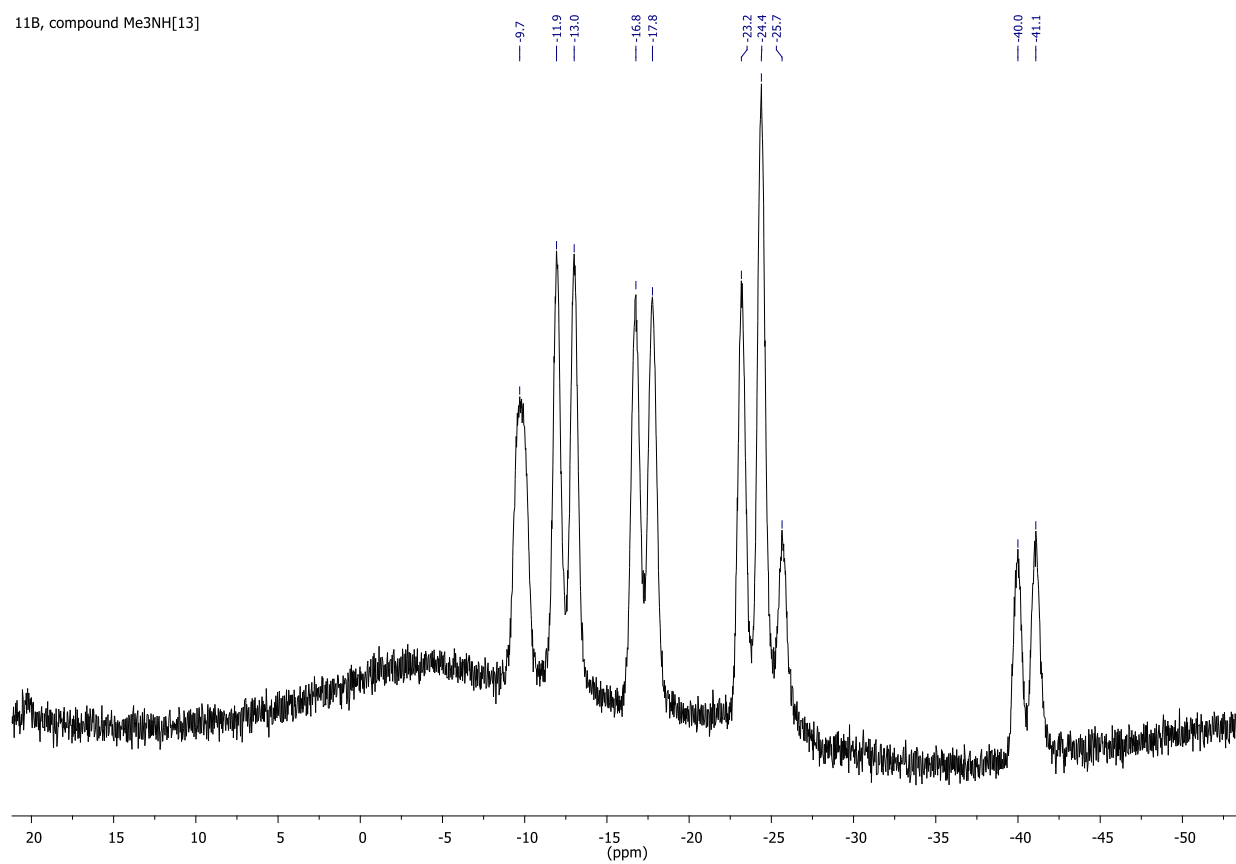




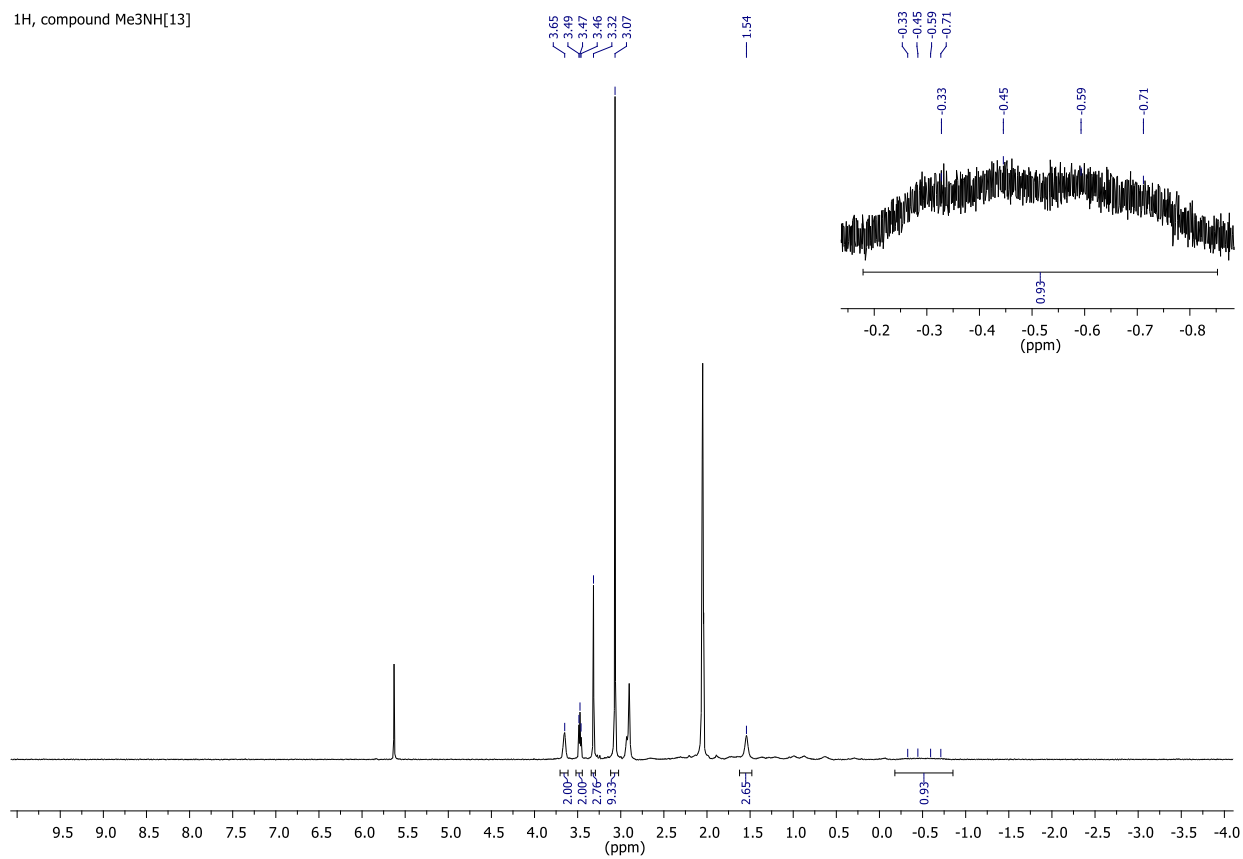
$^{11}\text{B}\{^1\text{H}\}$, compound $\text{Me}_3\text{NH}[13]$



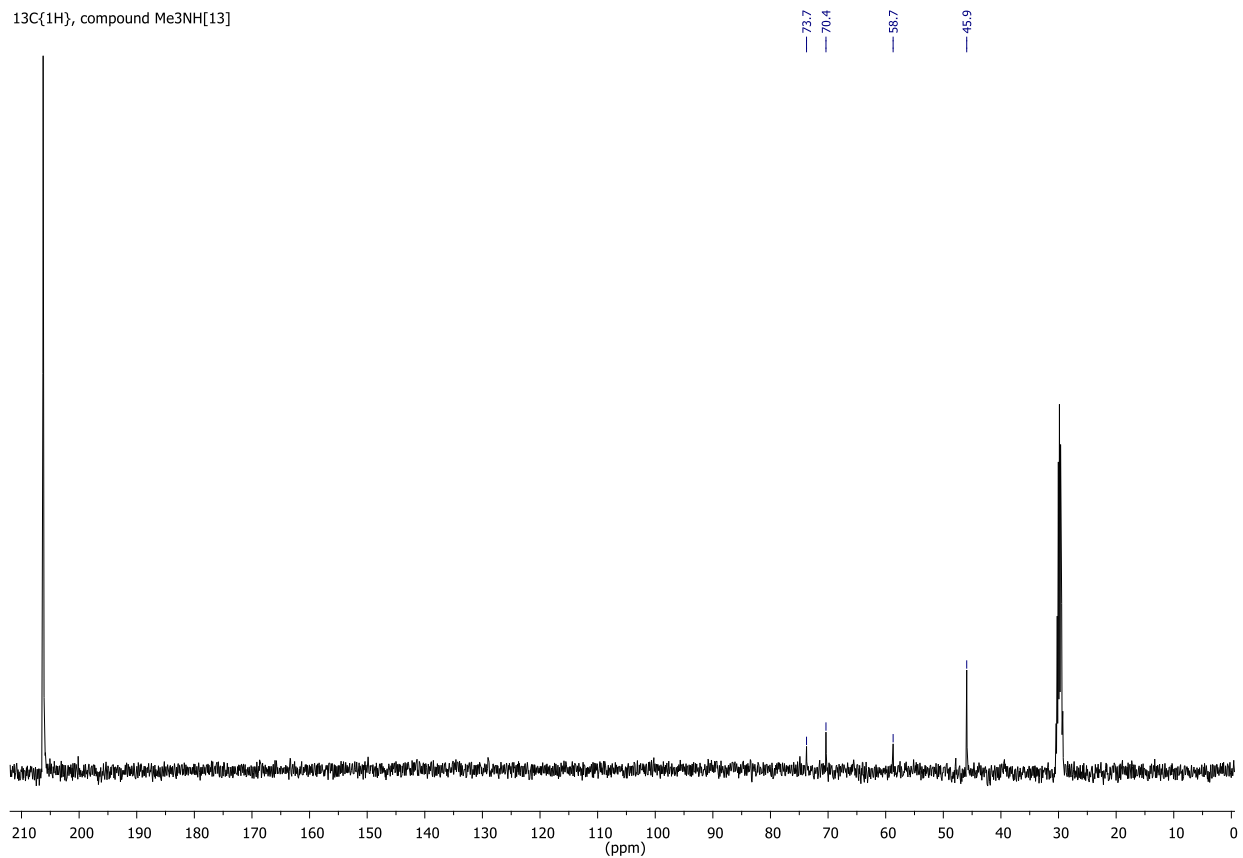
^{11}B , compound $\text{Me}_3\text{NH}[13]$

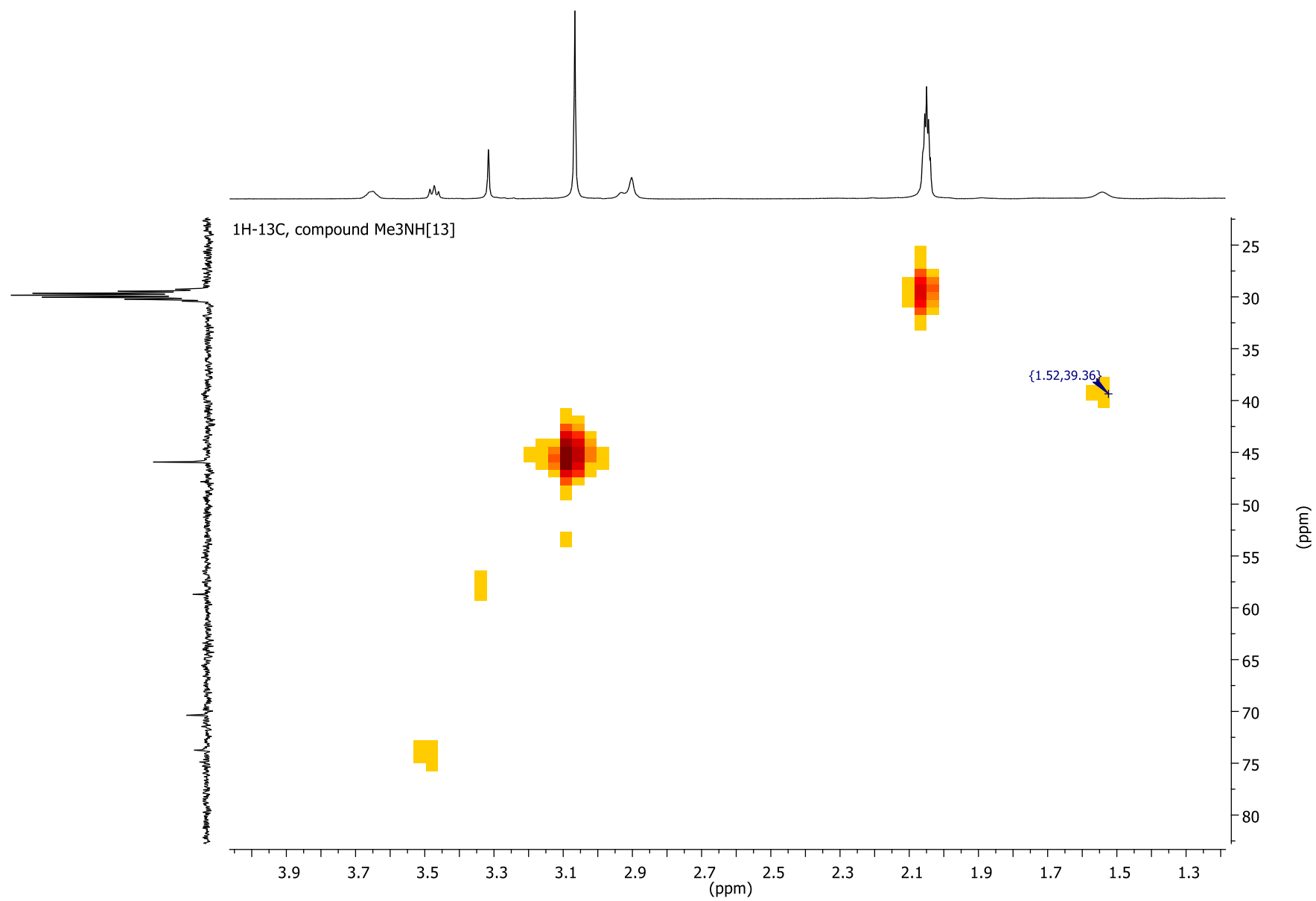


¹H, compound Me3NH[13]

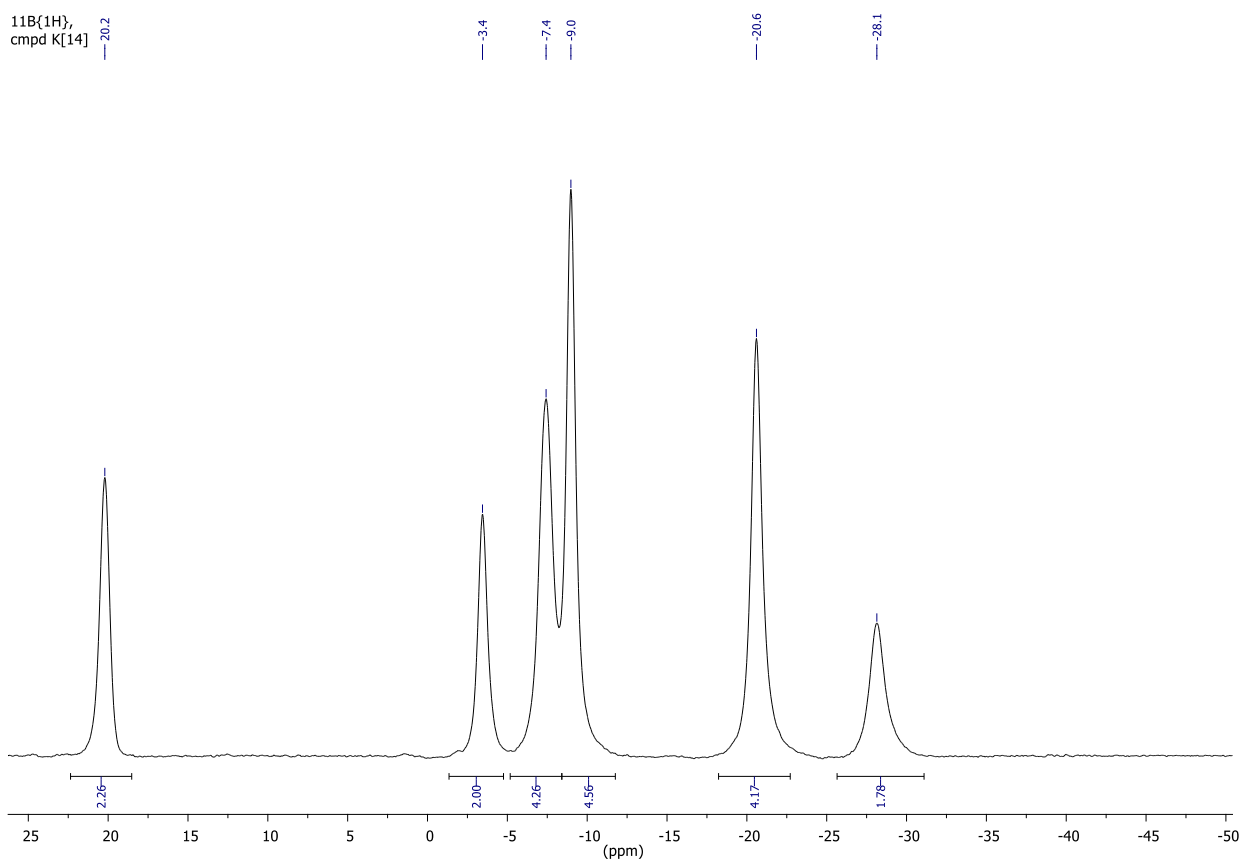


¹³C{¹H}, compound Me3NH[13]

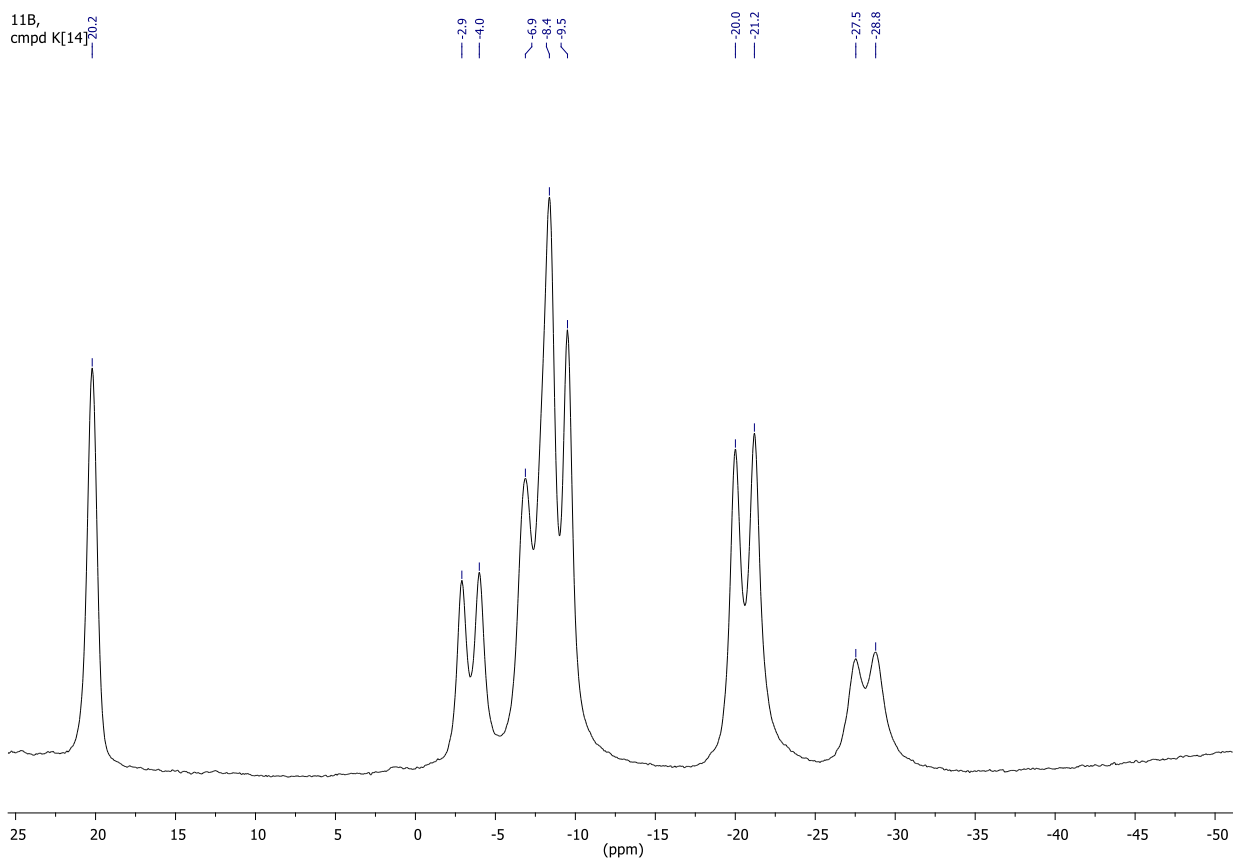




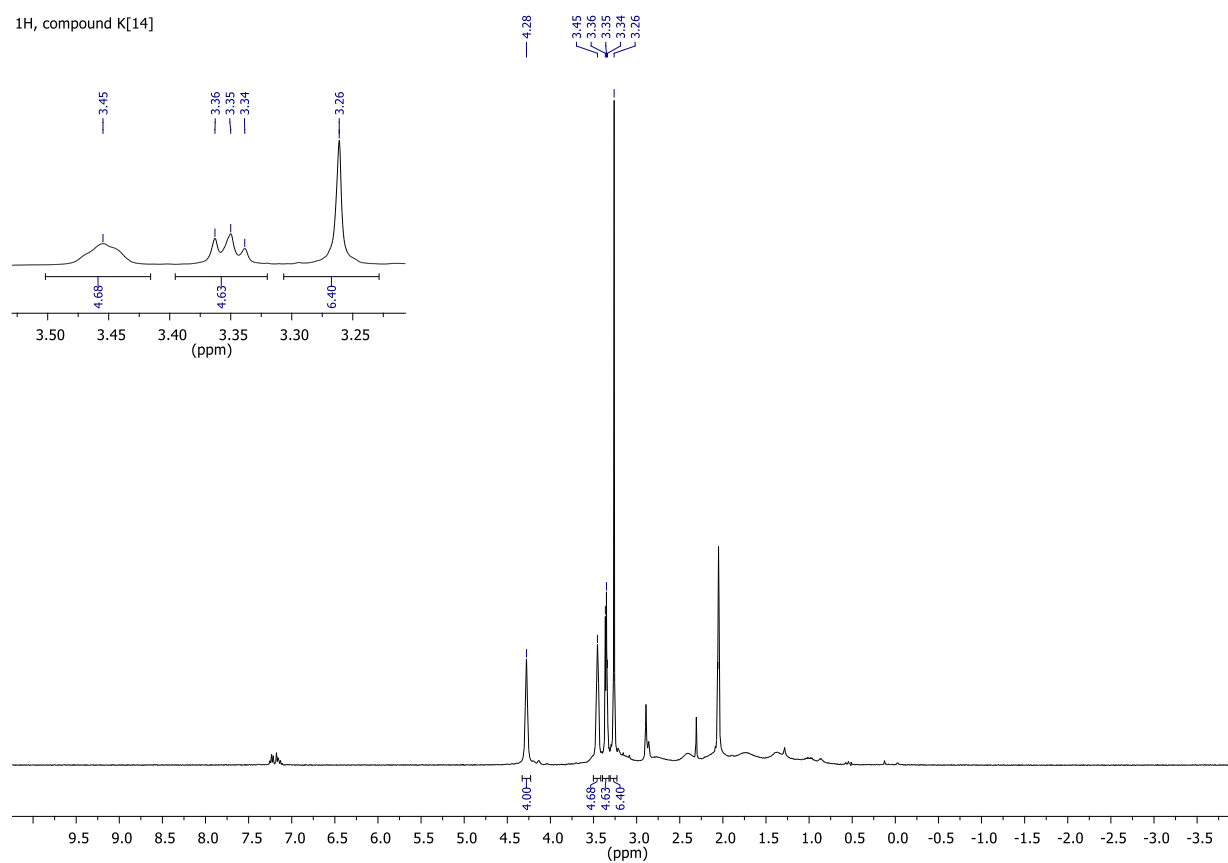
11B{1H},
compd K[14]



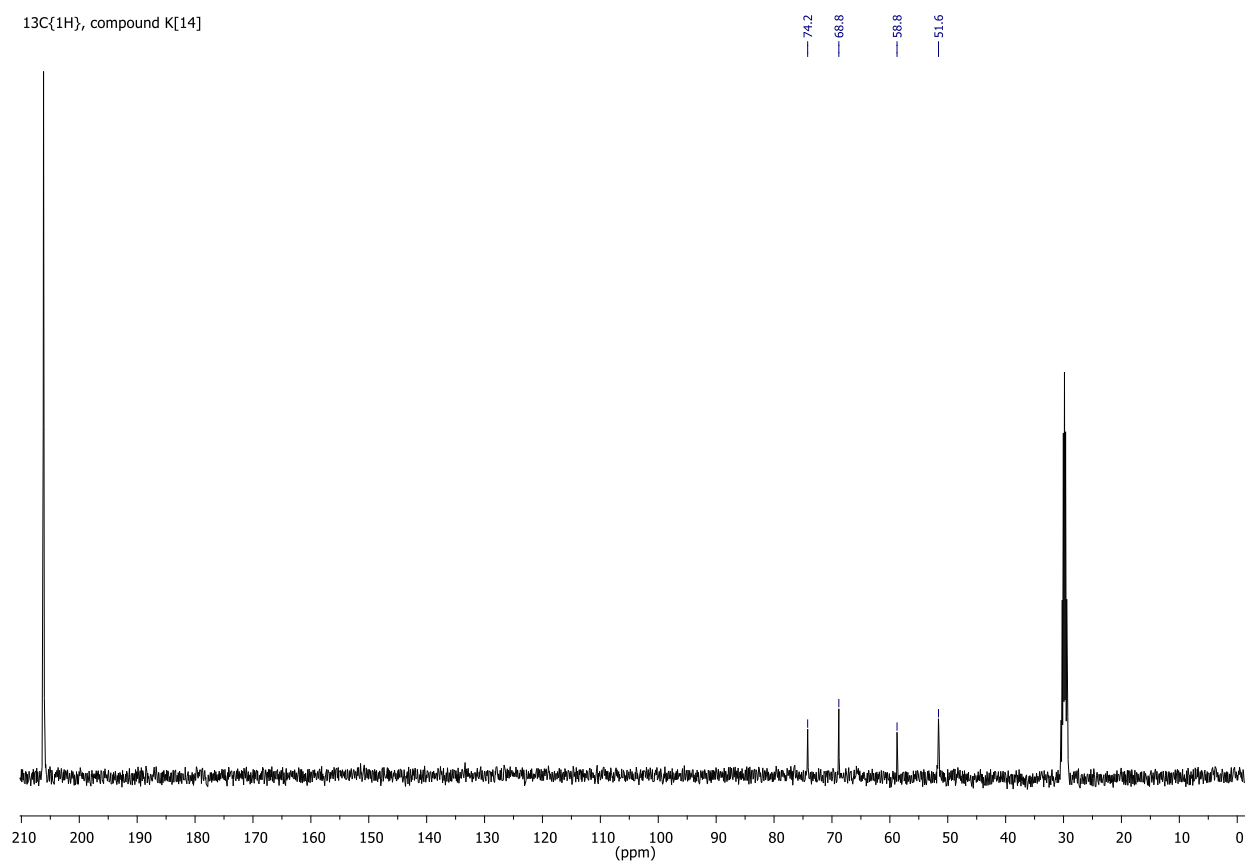
11B,
compd K[14]

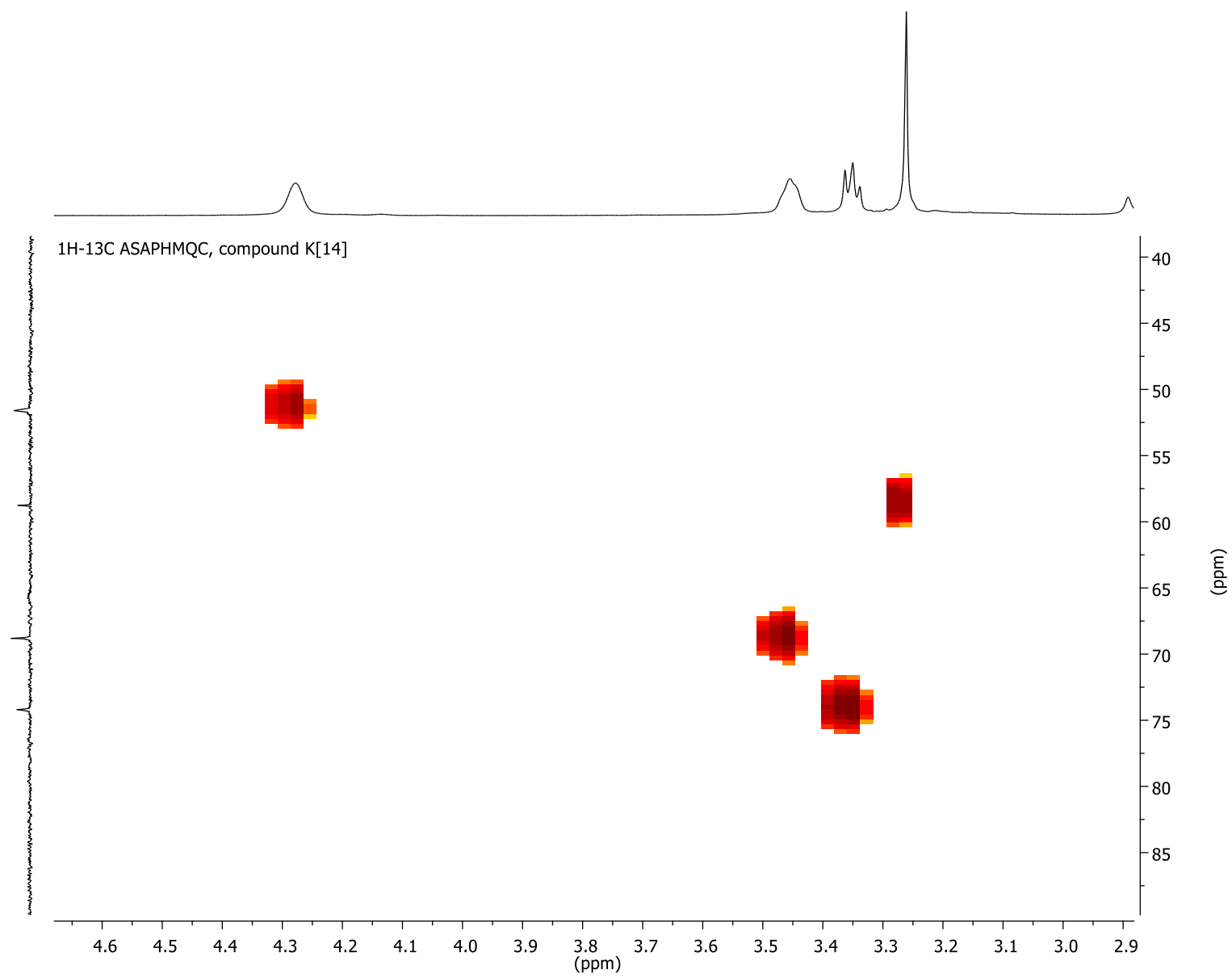


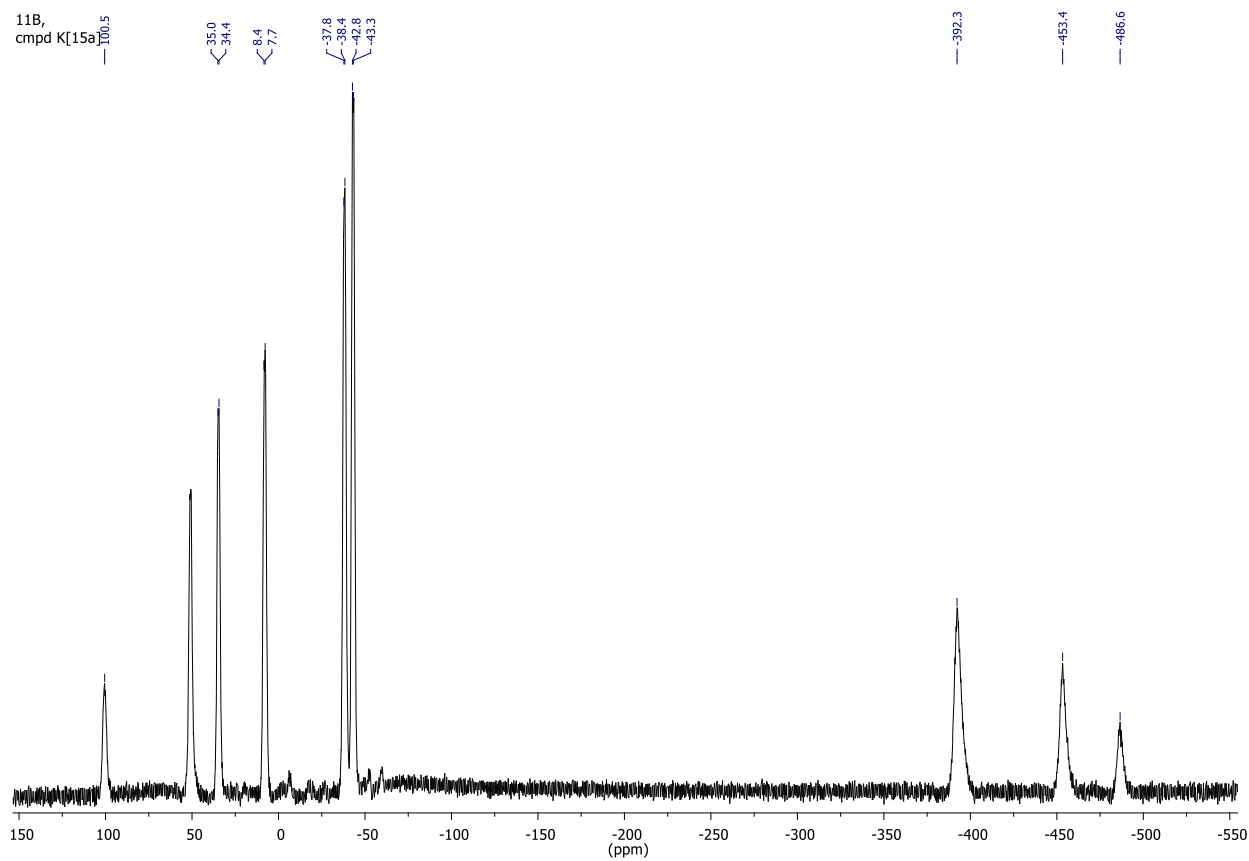
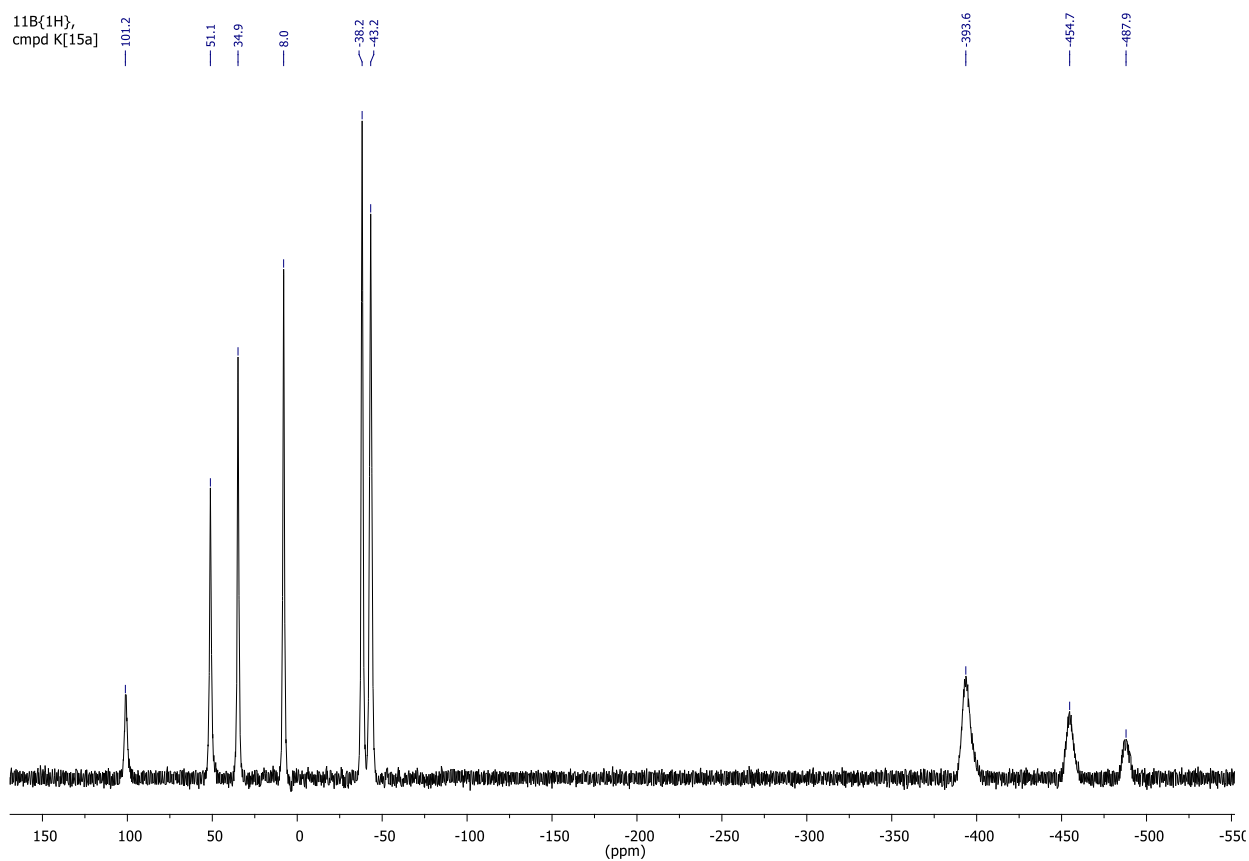
¹H, compound K[14]

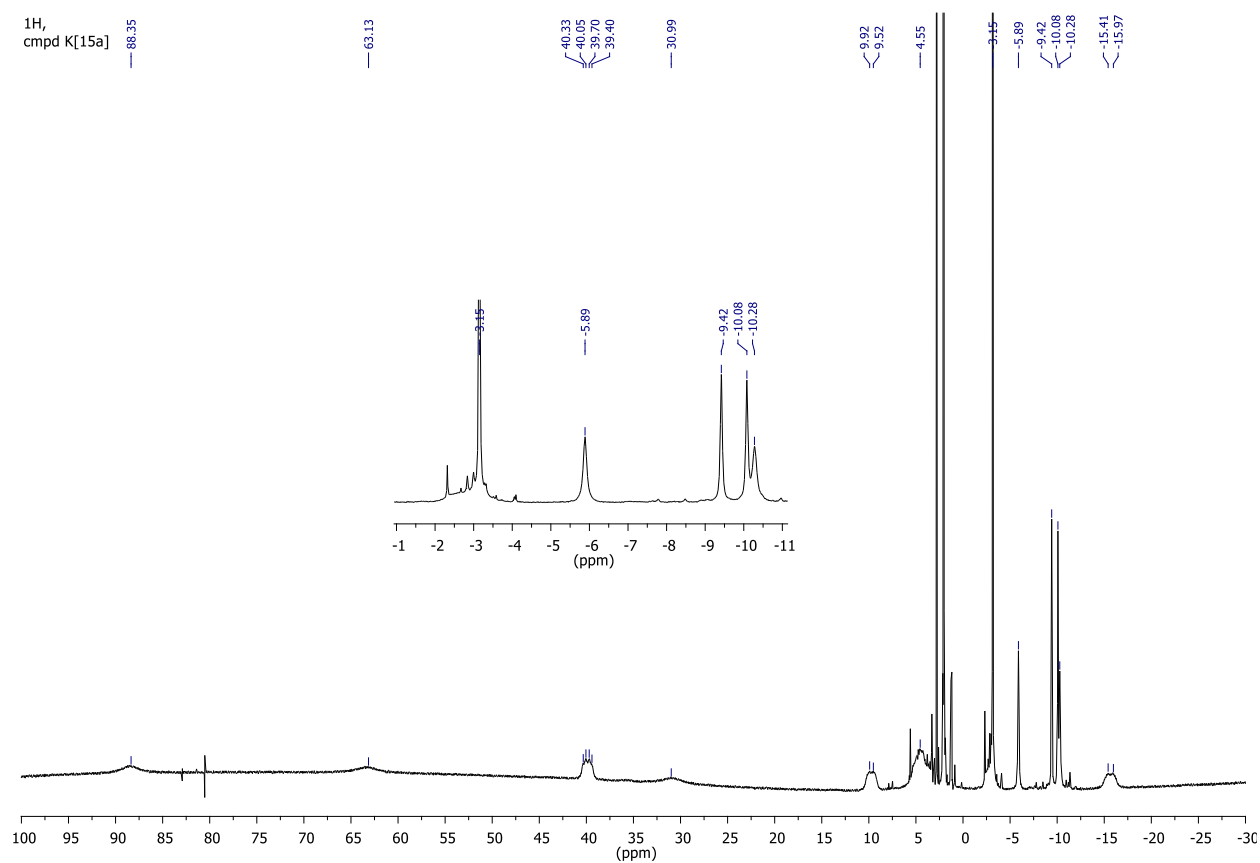


¹³C{¹H}, compound K[14]



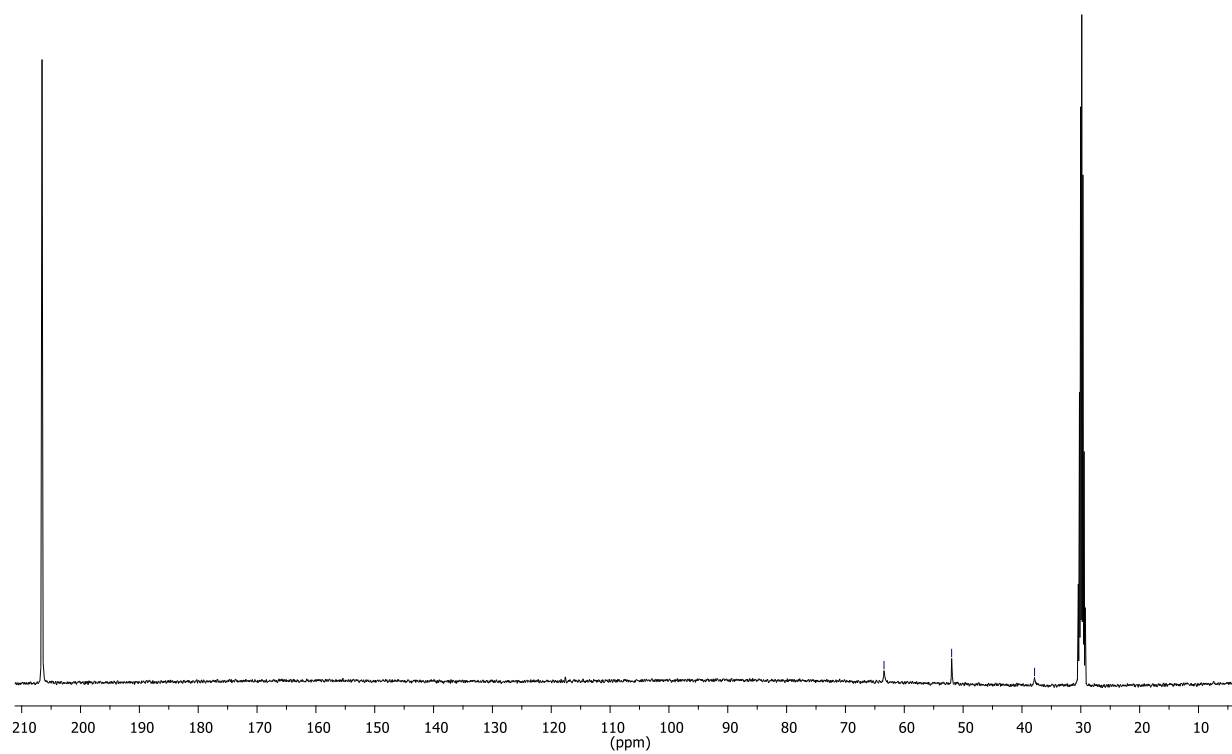


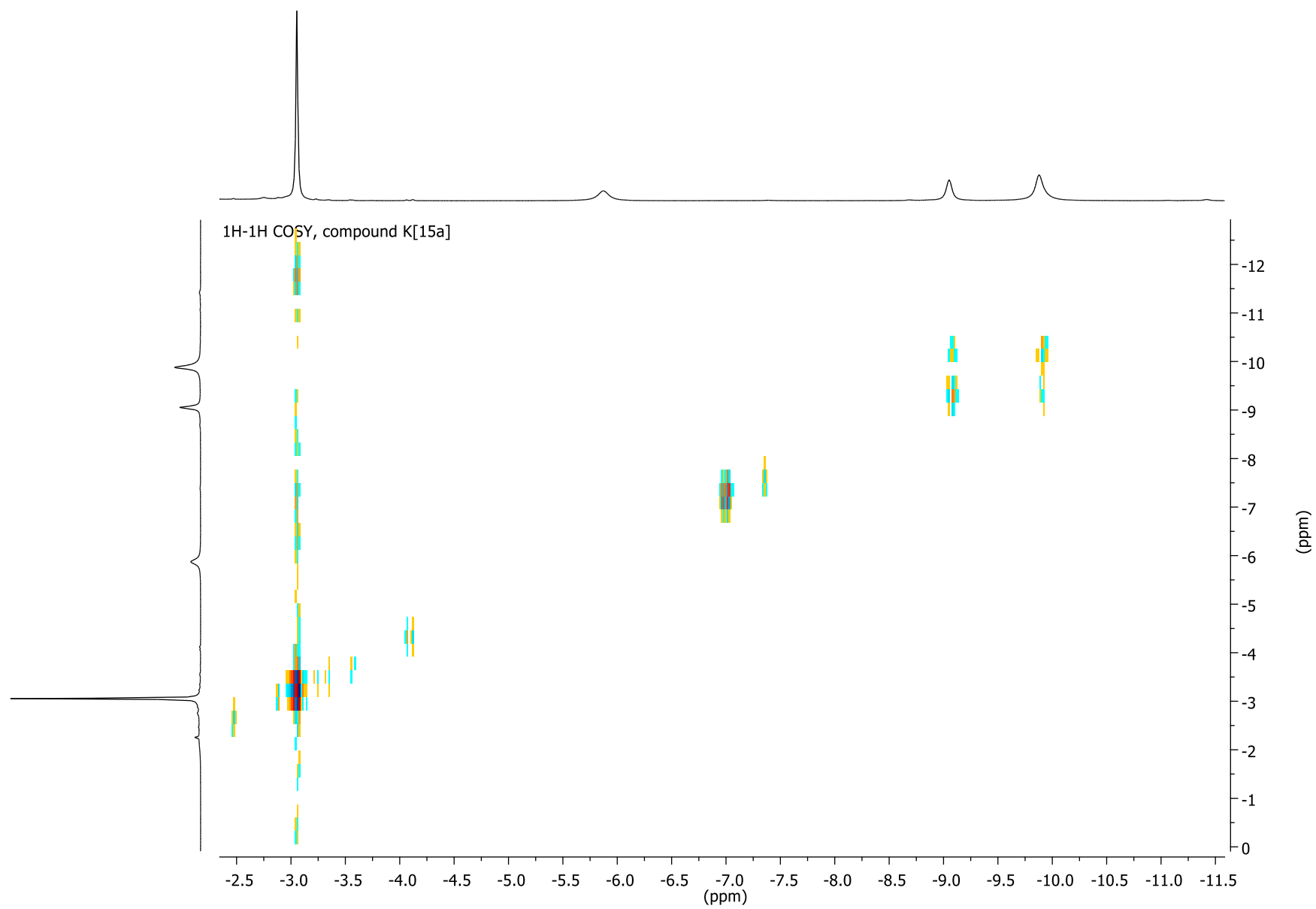


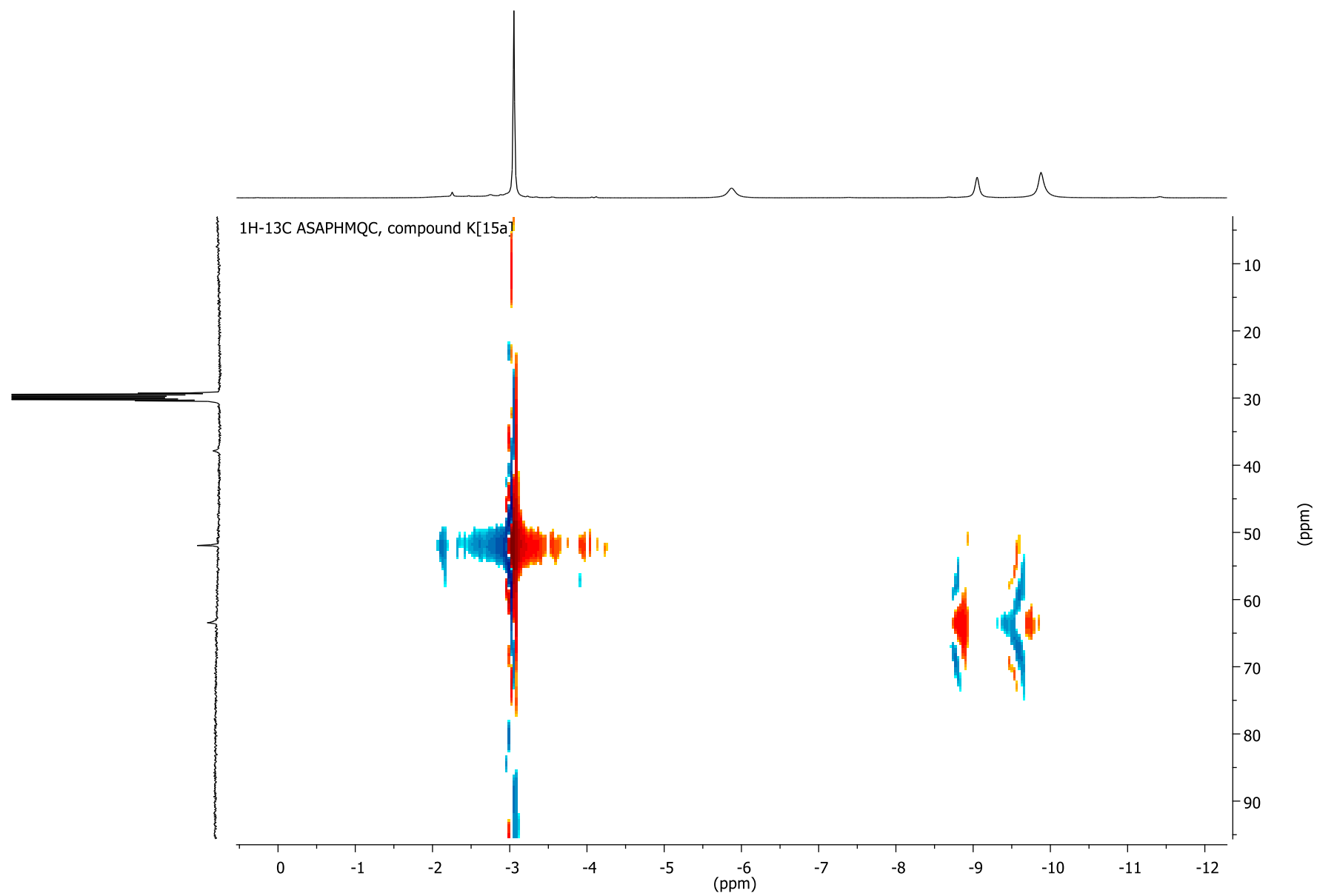


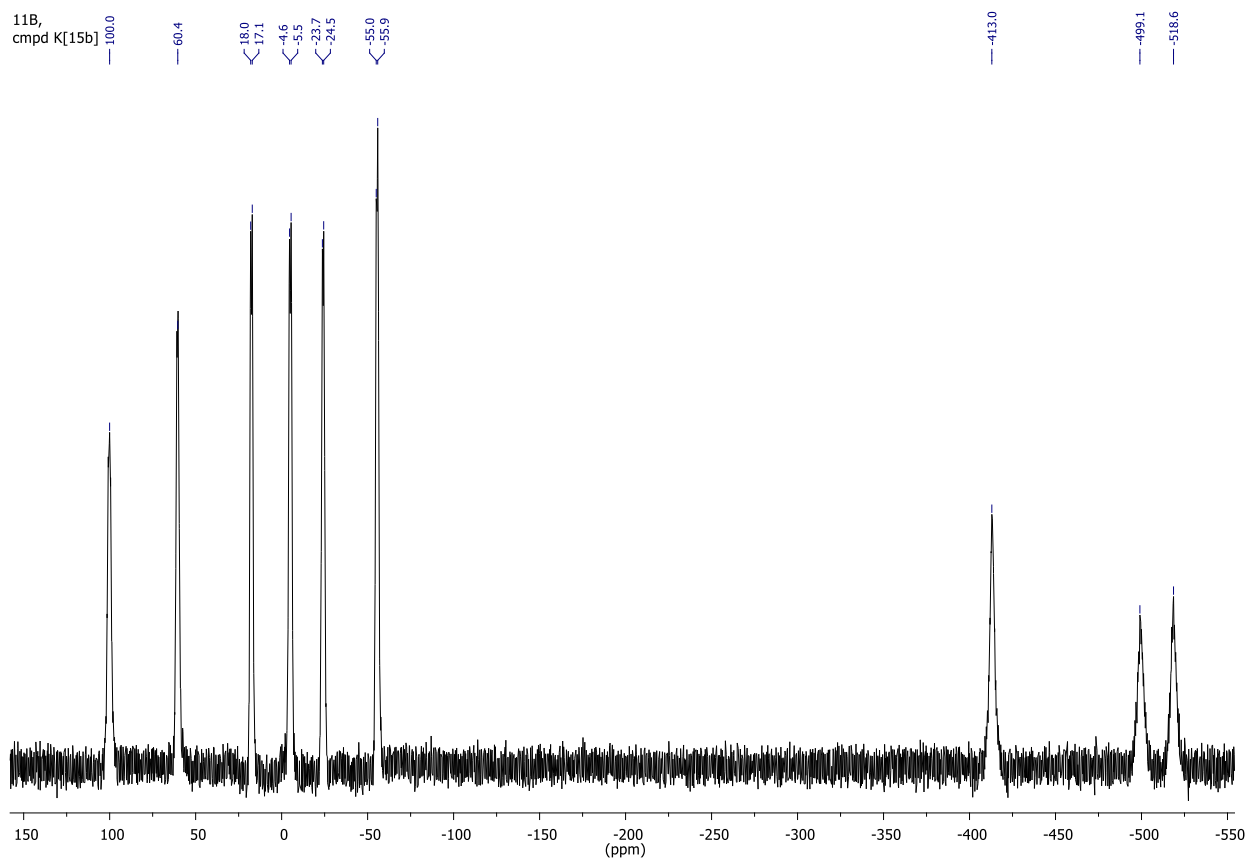
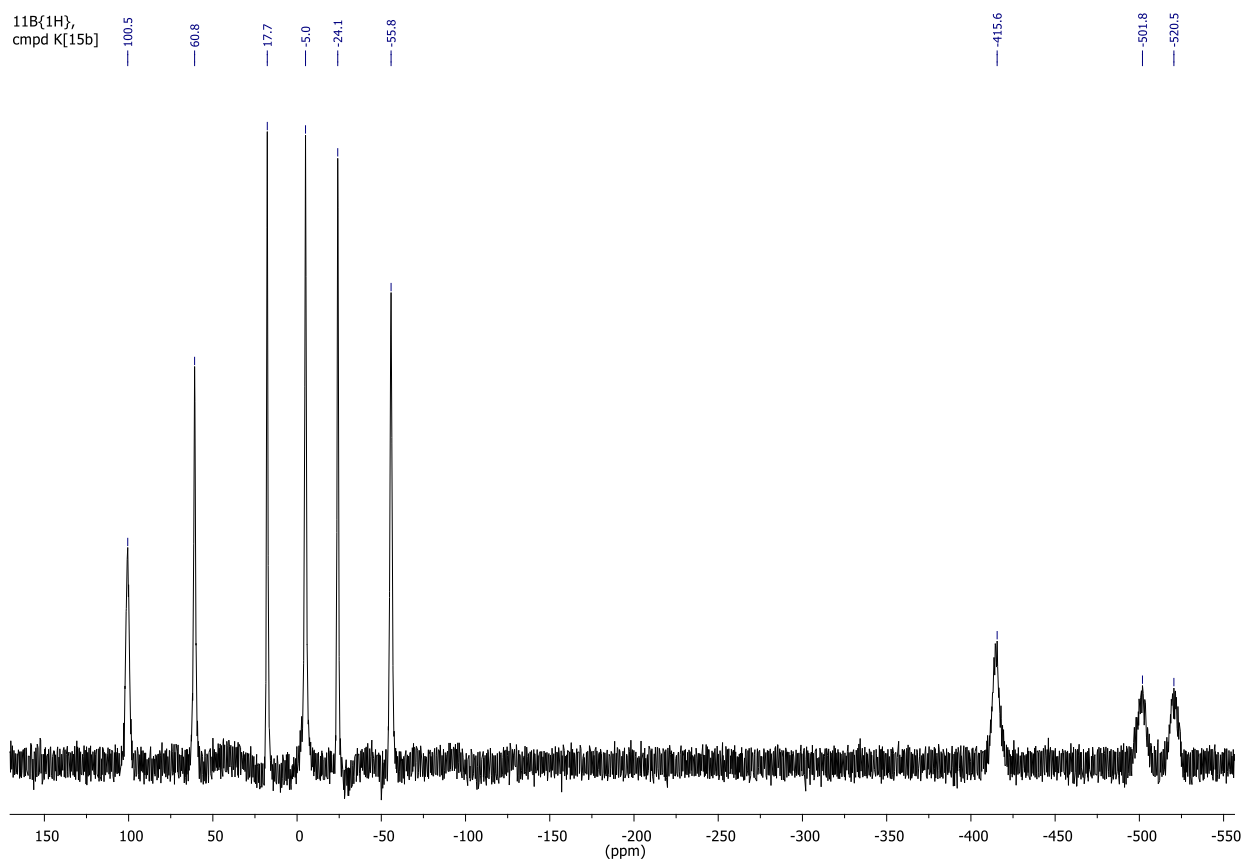
¹³C{¹H}, compound K[15a]

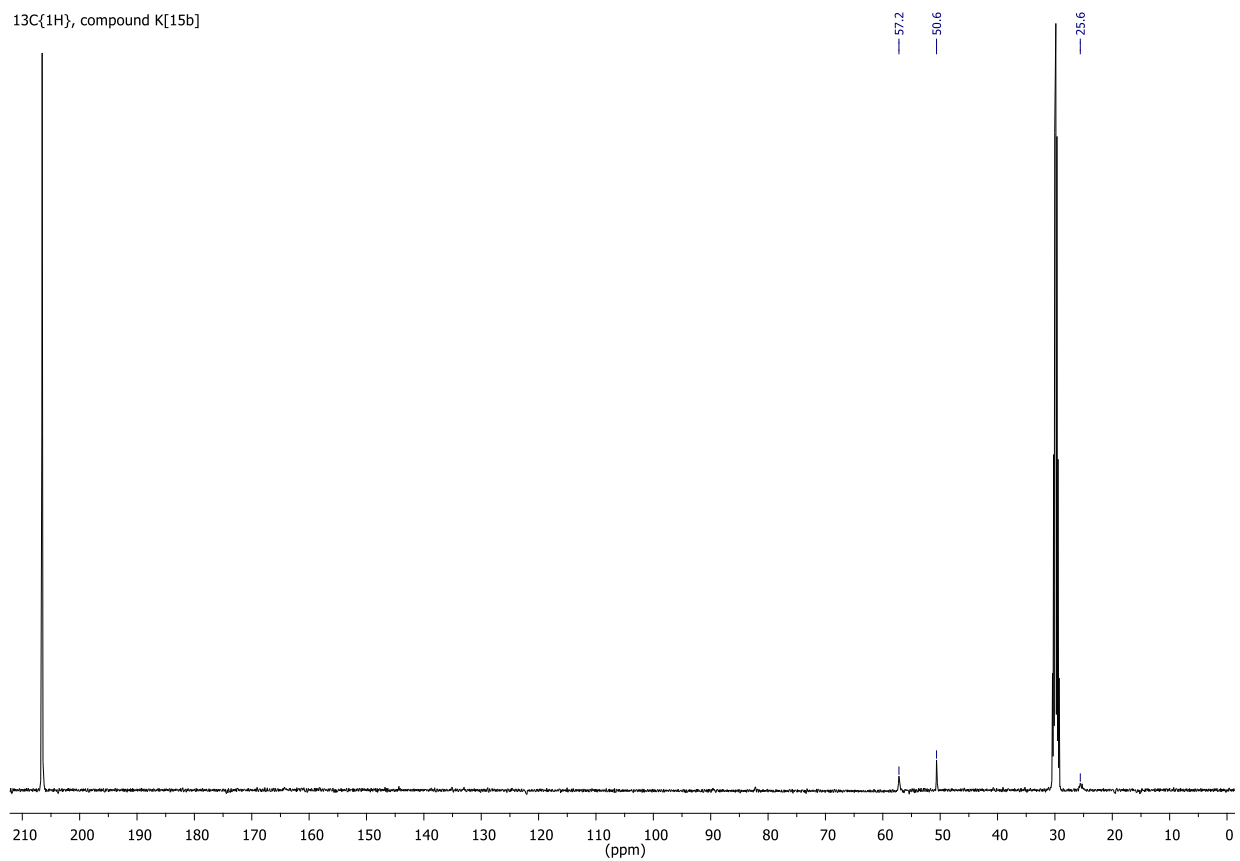
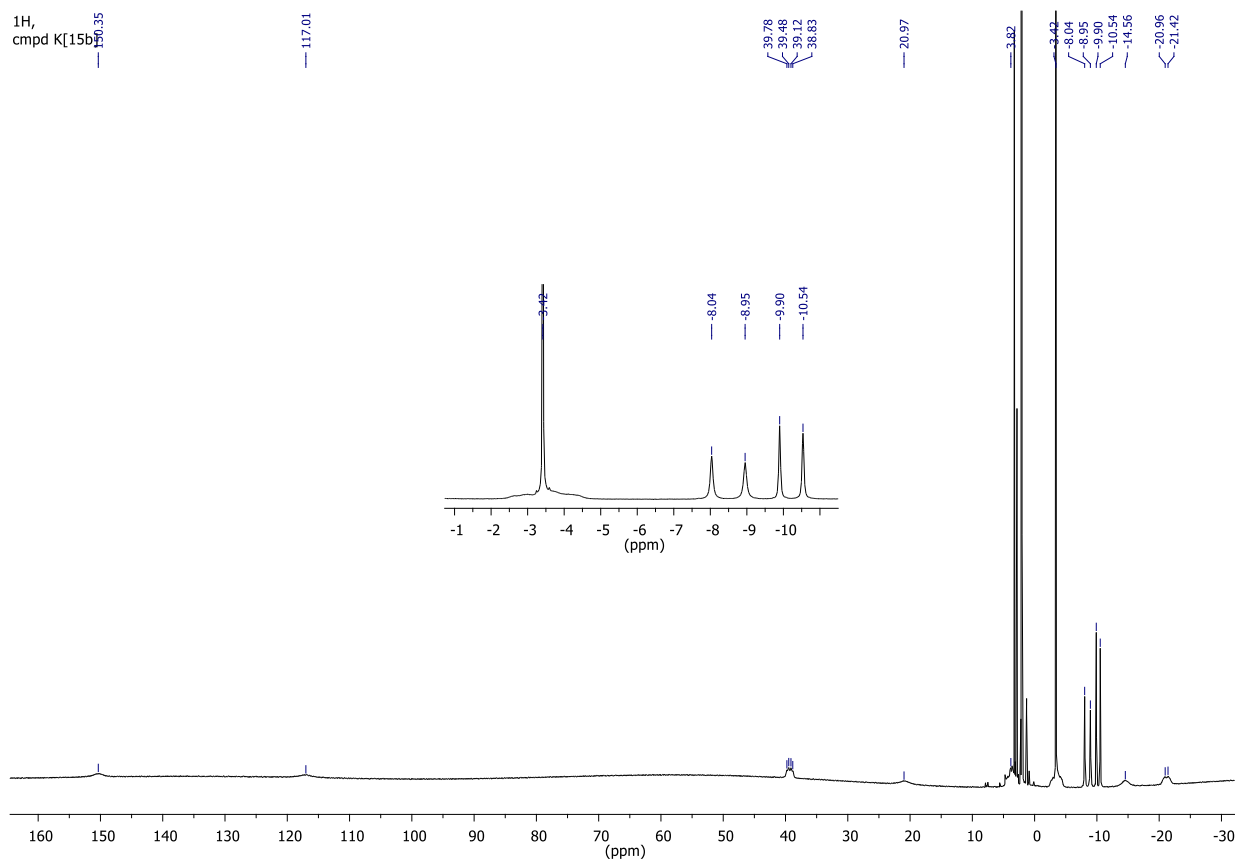
63.4, 52.0, 37.9

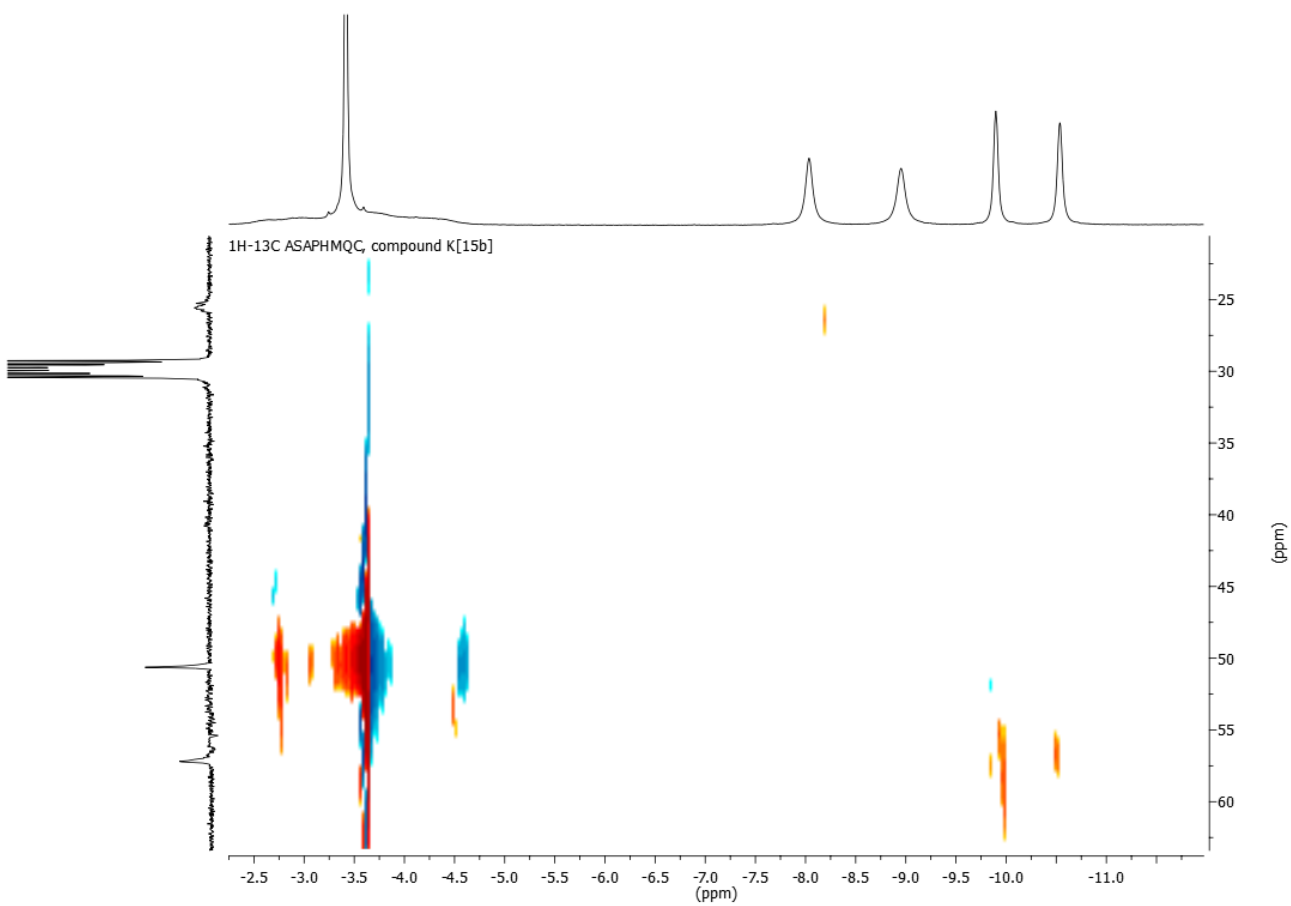
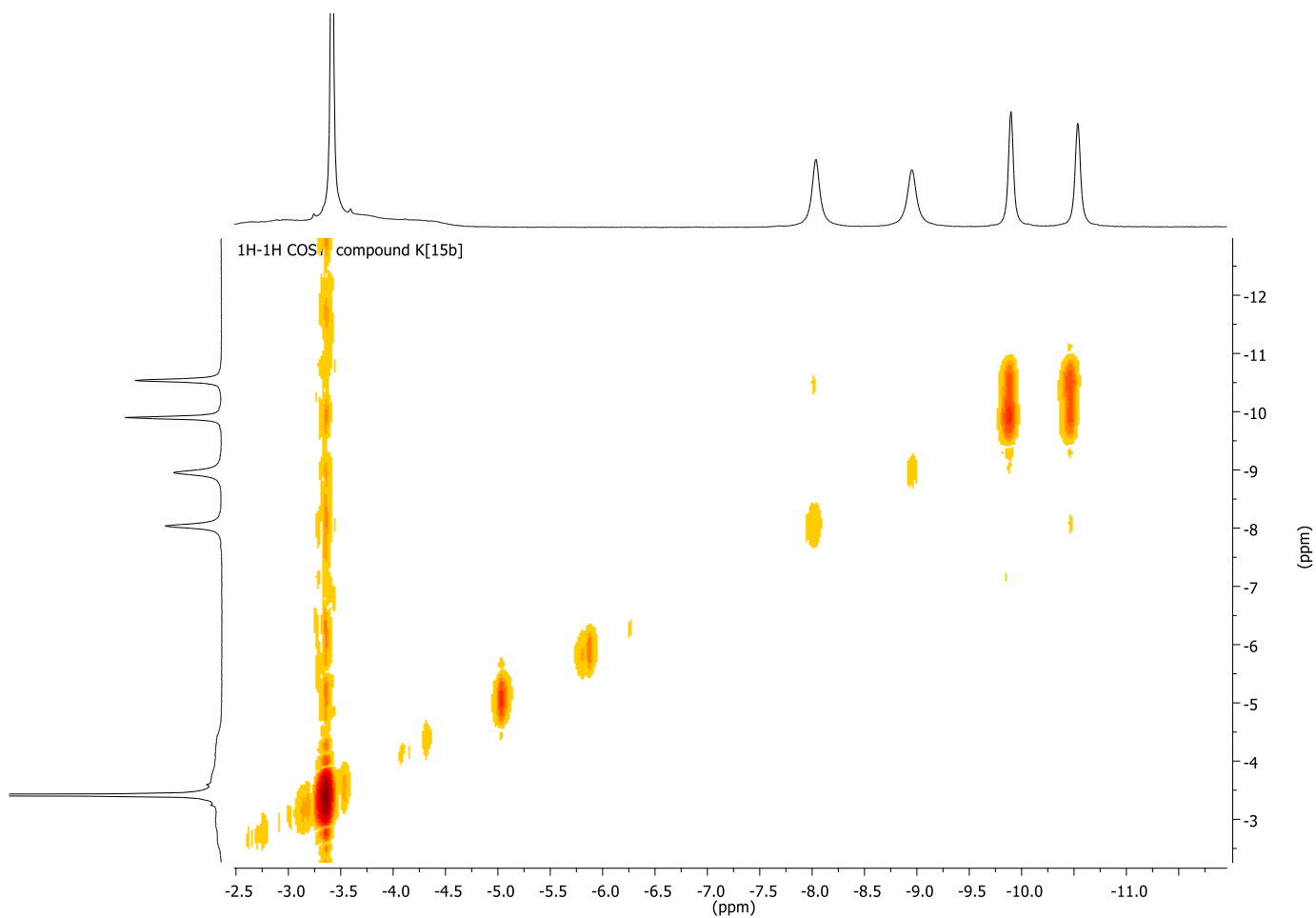












^{11}B NMR study of reaction of $[\text{9-I-7,8-C}_2\text{B}_9\text{H}_{11}]^-$ with 1,2-dimethoxyethane.

