

Supporting for

Several carbon-coated Ga₂O₃ anodes: Efficient coating of reduced Graphene Oxide enhanced electrochemical performance for lithium ion batteries

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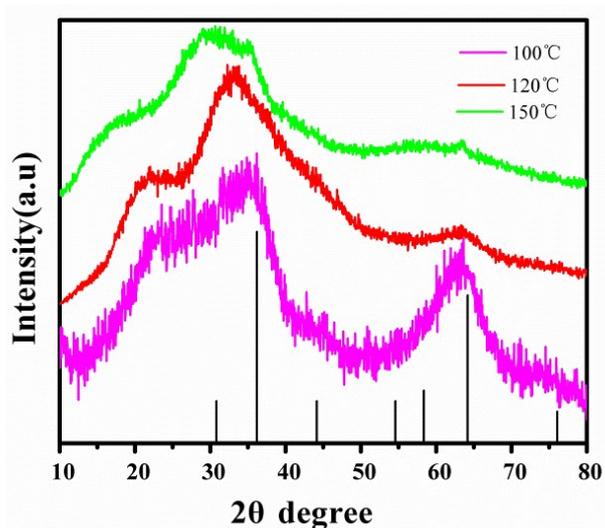


Fig.S1 The XRD pattern of Ga₂O₃ at 100 °C, 120 °C and 150 °C

The XRD pattern of Ga₂O₃ at different temperature was shown in Fig.S1. It can be observed that the XRD pattern of Ga₂O₃ in 100 °C was indexed to the space group (JCPDS#20-0426), which corresponded to the face-centered cubic structure of γ -Ga₂O₃. It is worth noting that with the increase of temperature, the diffraction peak of Ga₂O₃ will shift slightly, but the XRD diffraction peak at 64° of Ga₂O₃ in 120 and 150 °C was still assigned to the Ga₂O₃ (JCPDS#20-0426).

The average particle diameters (d) were calculated based on the peak broadening of the

(311) reflection of γ -Ga₂O₃ in 100°C by using the Scherrer formula: $D = \frac{0.89\gamma}{\beta \cos\theta}$,

$$\beta = \frac{X1 - X2}{180} * \pi = 0.0488, \gamma = 0.15405, D = 3.87nm$$

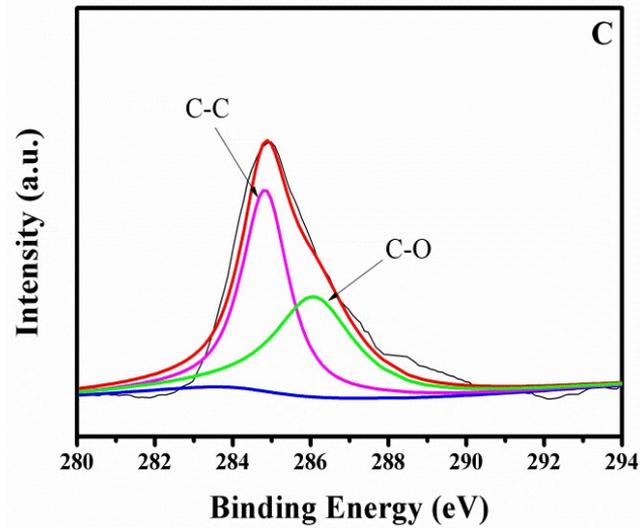


Fig.S2 High-resolution XPS spectrum of C1s.

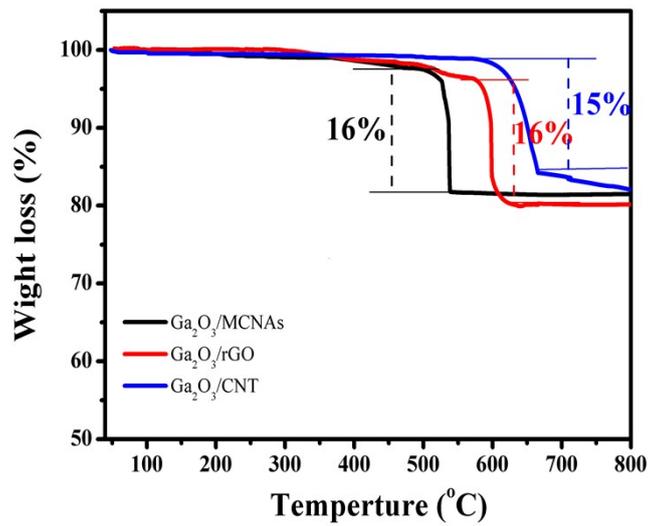


Fig.S3 The TG curves of Ga₂O₃/MCNAs(black line), Ga₂O₃/rGO(red line), and Ga₂O₃/CNT(blue line)

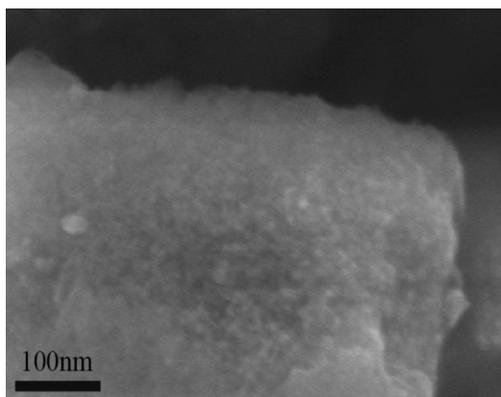


Fig.S4 The SEM image of Ga₂O₃ nanoparticles

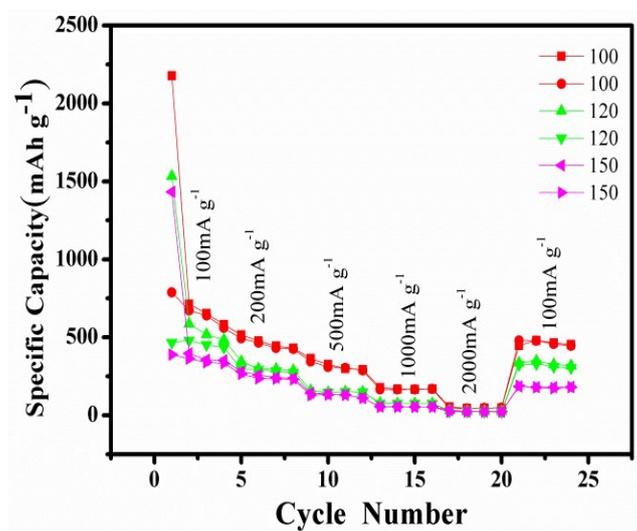


Fig.S5 Rate performance of Ga₂O₃ at three temperatures

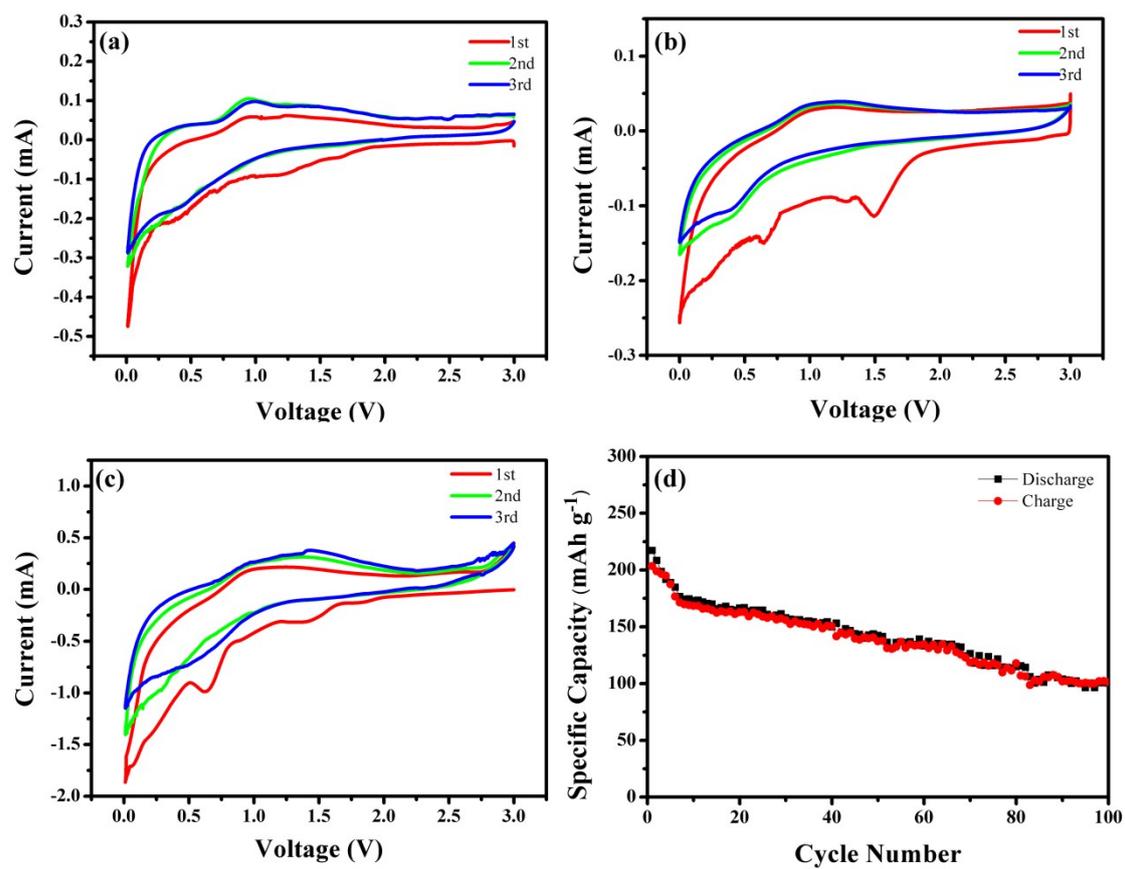


Fig. S6 CV curves of the first three cycles of Ga₂O₃/MCNAs(a), Ga₂O₃/CNT(b) and Ga₂O₃(c) electrodes. (d) Cycling performance of the Ga₂O₃ electrodes at the current density of 1000 mAh g⁻¹

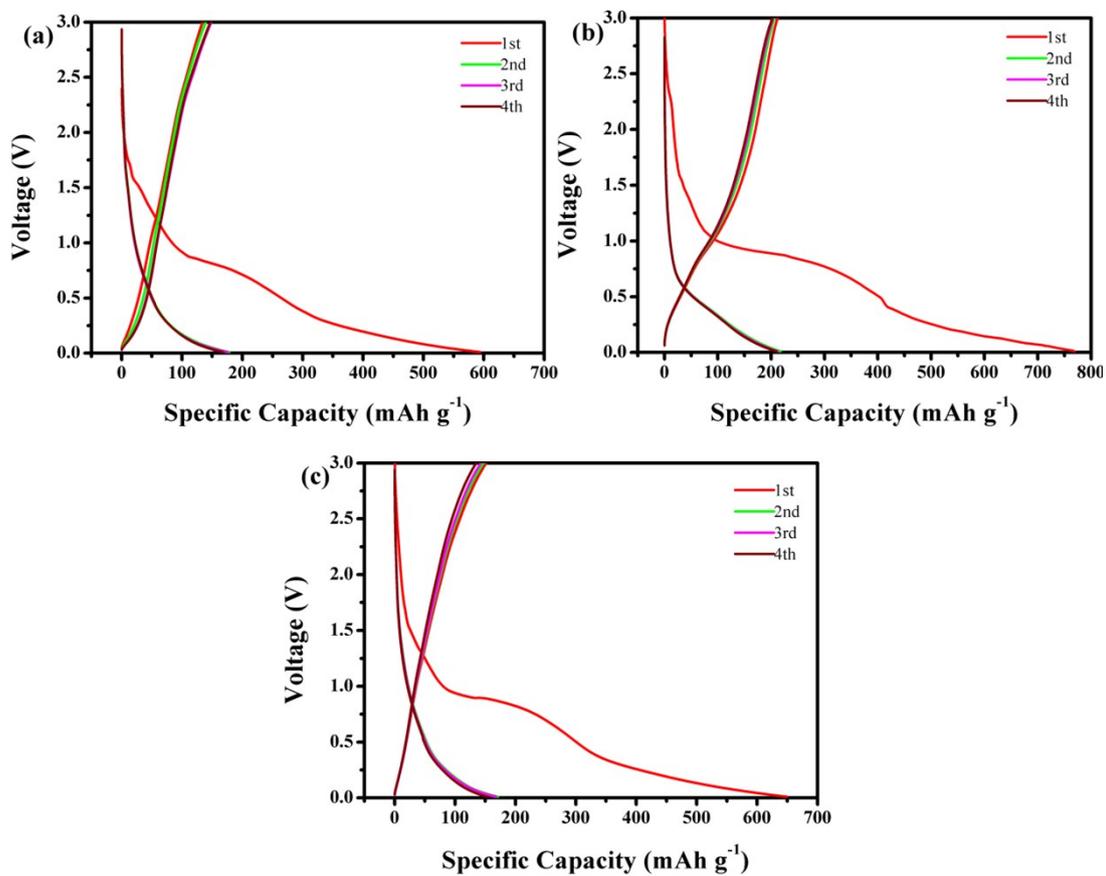


Fig. S7 the galvanostatic charge/discharge curves at 100 mA g^{-1} of rGO (a), MCNAs (b), and CNT(c) electrodes.

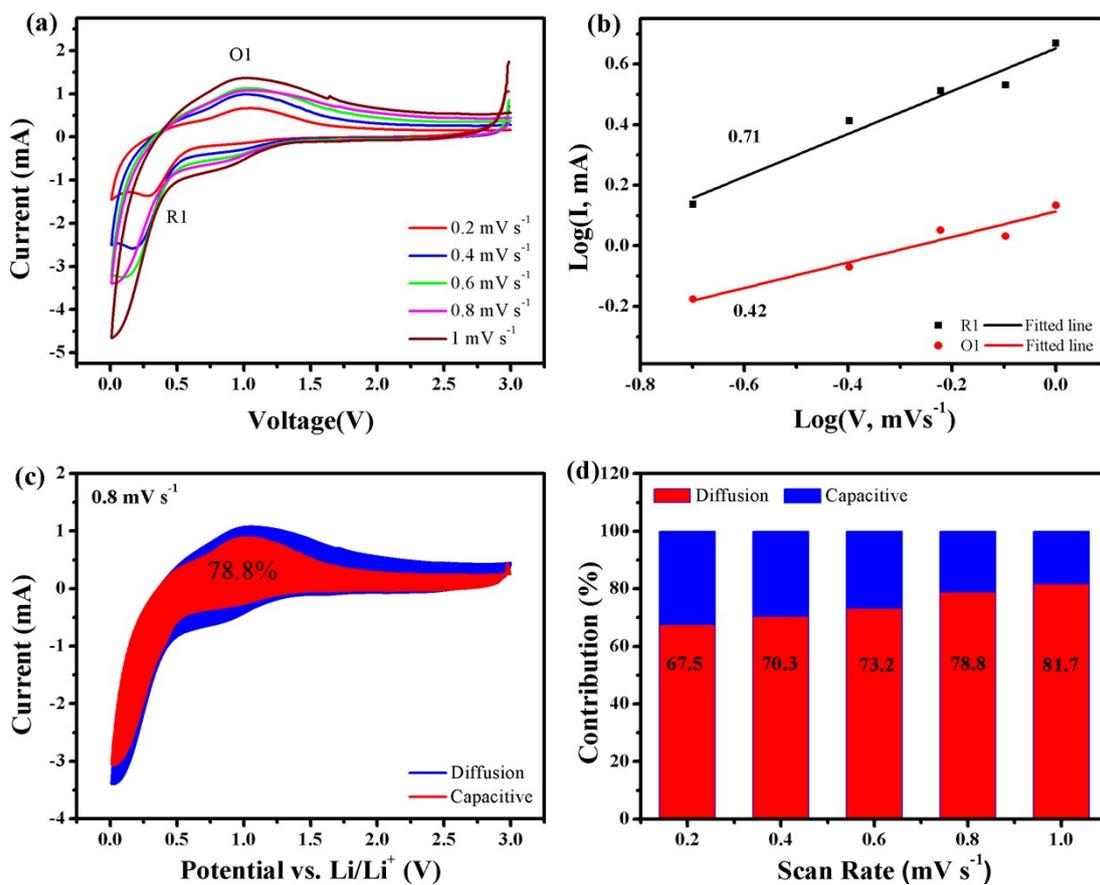


Fig. S8 (a) CV curves at different scan rates ranging from 0.01 V to 3.00 V, (b) relationship of Log *i* vs. Log *v* at oxidation and reduction states, (c) capacitive and diffusion contribution to charge storage at 0.6 mV s⁻¹, (d) contribution ratio of the capacitive and diffusion controlled capacities at different scan rates of Ga₂O₃

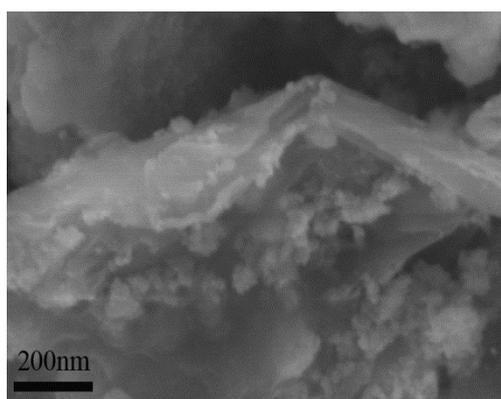


Fig. S9 The SEM image of the Ga₂O₃/rGO electrode after cycling test.

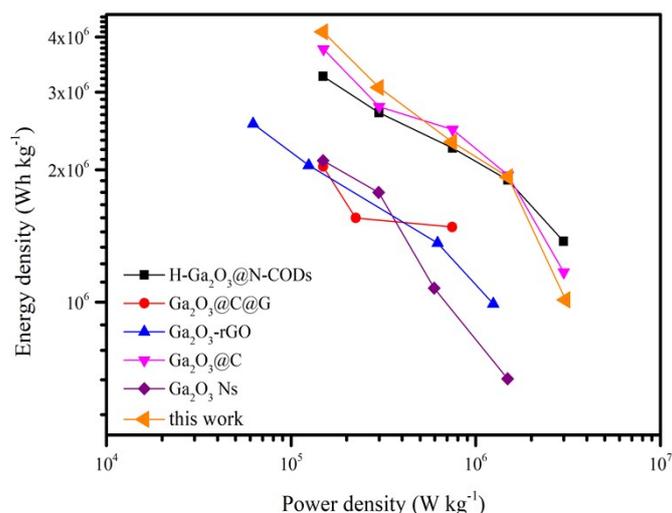


Fig. S10 Ragone plots of different Ga₂O₃ composite materials.

Table.S1 The N₂ adsorption-desorption test results of Ga₂O₃/C and three types of carbon materials.

samples	specific surface area (m ² g ⁻¹)	pore volume (cm ³ g ⁻¹)	pore size (nm)
Ga ₂ O ₃ /rGO	253	0.18	2.8
Ga ₂ O ₃ /MCNAs	307	0.24	3.4
Ga ₂ O ₃ /CNT	293	0.23	3.1
rGO	167	0.16	2.4
MCNAs	1290	1.59	4.2
CNT	187	0.97	3.7

References

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