Electronic Supplementary Information

Tailoring copper(II) complexes with pyridine-4,5-dicarboxylate esters for

anti-Candida activity

Tina P. Andrejević,^a Ivana Aleksic,^b Marta Počkaj,^c Jakob Kljun,^c Dusan Milivojevic,^b

Nevena Lj. Stevanović,^a Jasmina Nikodinovic-Runic,^{*b} Iztok Turel,^{*c} Miloš I. Djuran,^{*d} and

Biljana Đ. Glišić*a

^aUniversity of Kragujevac, Faculty of Science, Department of Chemistry, R. Domanovića 12, 34000 Kragujevac, Serbia

^bInstitute of Molecular Genetics and Genetic Engineering, University of Belgrade, Vojvode Stepe 444a, 11042 Belgrade, Serbia

^cUniversity of Ljubljana, Faculty of Chemistry and Chemical Technology, Večna pot 113, SI-1000 Ljubljana, Slovenia

eSerbian Academy of Sciences and Arts, Knez Mihailova 35, 11000 Belgrade, Serbia

*Corresponding authors. Tel.: +381 11 397 6034 (J. Nikodinovic-Runic); Tel.: +386 1 47 98 525 (I. Turel); Tel.: +381 34 300 251 (M. I. Djuran); Tel.: +381 34 336 223 (B. Đ. Glišić).

E-mail addresses: jasmina.nikodinovic@imgge.bg.ac.rs (J. Nikodinovic-Runic); <u>Iztok.Turel@fkkt.uni-lj.si</u> (I. Turel); <u>milos.djuran@pmf.kg.ac.rs</u> (M. I. Djuran); <u>biljana.glisic@pmf.kg.ac.rs</u> (B. Đ. Glišić)

TABLE OF CONTENTS

Fig. S1 Time stability of complex 1 followed by UV-Vis spectrophotometry at roomS3temperature in DMSO.

Fig. S2 Cyclic voltammograms of the copper(II) complexes 2 and 3 (A) and 4 and S4 5 (B), alongside the corresponding CuX₂ salt (X = NO₃⁻ and Cl⁻) at GC electrode in DMSO and 0.1 M TBAHP as a supporting electrolyte with a scan rate of 50 mV s⁻¹. The conditions were the following: $E_{\text{begin}} = -2.0 \text{ V}$, $E_{\text{end}} = 2.0 \text{ V}$ and $E_{\text{step}} = 0.002 \text{ V}$.

Fig. S3 Filamentation of C. albicans ATCC 10231 in presence of subinhibitoryS5concentrations $(0.5 \times MIC)$ of complexes 1 and 4 in liquid RPMI medium (OlympusBX51, Applied Imaging Corp., San Jose, CA, United States, under 20 \times magnification).

Table S1 Values of the binding constants of copper(II) complexes 1 – 5 withS6ct-DNA.

Table S2 Details of the crystal structure determination for copper(II) complexesS71-5.



Fig. S1 Time stability of complex 1 followed by UV-Vis spectrophotometry at room temperature in DMSO.



Fig. S2 Cyclic voltammograms of the copper(II) complexes 2 and 3 (A) and 4 and 5 (B), alongside the corresponding CuX₂ salt (X = NO₃⁻ and Cl⁻) at GC electrode in DMSO and 0.1 M TBAHP as a supporting electrolyte with a scan rate of 50 mV s⁻¹. The conditions were the following: $E_{\text{begin}} = -2.0 \text{ V}$, $E_{\text{end}} = 2.0 \text{ V}$ and $E_{\text{step}} = 0.002 \text{ V}$.



Fig. S3 Filamentation of *C. albicans* ATCC 10231 in presence of subinhibitory concentrations (0.5 × MIC) of complexes 1 and 4 in liquid RPMI medium (Olympus BX51, Applied Imaging Corp., San Jose, CA, United States, under 20 × magnification).

Complex	$K_{sv}(\mathrm{M}^{-1})$	Hypochromism (%)	K_q (M ⁻¹ s ⁻¹)	$K_A(\mathrm{M}^{-1})$	п
1	$(1.06\pm0.01)\cdot10^3$	15.6	$1.06 \cdot 10^{11}$	$8.75 \cdot 10^2$	0.97
2	(8.15±0.40)·10 ²	12.0	8.15.1010	8.51.102	0.99
3	(9.66±0.15)·10 ²	15.4	9.66.1010	$1.12 \cdot 10^{2}$	0.75
4	$(1.22\pm0.03)\cdot10^{3}$	17.1	$1.22 \cdot 10^{11}$	7.09.102	0.93
5	$(1.01\pm0.04)\cdot10^{3}$	15.5	$1.01 \cdot 10^{11}$	3.55.102	0.86

Table S1 Values of the binding constants of copper(II) complexes 1-5 with ct-DNA.

Complex	1	2	3	4	5
CCDC No.	2041822	2041823	2041819	2041821	2041820
Empirical formula	C ₁₂ H ₁₆ CuN ₄ O ₁₃ S	C ₁₃ H ₁₄ Cu ₁ N ₄ O ₁₁ S	C ₁₄ H ₁₆ Cu ₁ N ₄ O ₁₂	$C_{24}H_{20}Cl_4Cu_2N_4O_8S_2$	$C_{26}H_{24}Cl_4Cu_2N_4O_8S_2$
Formula weight	519.89	497.88	495.85	825.44	853.49
Temperature/K	150(2)	150(2)	150(2)	150(2)	150(2)
Crystal system	triclinic	monoclinic	monoclinic	triclinic	monoclinic
Space group	Pī	$P2_1/c$	Cc	Pī	P21/c
a/Å	7.8810(3)	9.1313(6)	10.3255(6)	7.7848(5)	12.0032(7)
b/Å	12.4005(4)	24.6456(11)	22.7235(11)	9.0261(6)	17.3134(8)
c/Å	20.9376(6)	8.5875(5)	8.0546(5)	11.9040(5)	7.6901(4)
α/°	79.593(3)	90	90	101.778(4)	90
	88.991(2)	106.946(7)	101.849(6)	100.435(4)	99.719(5)
γ/°	88.486(3)	90	90	109.621(6)	90
Volume/Å ³	2011.66(12)	1848.67(19)	1849.59(19)	742.48(8)	1575.19(14)
Z	4	4	4	1	2
$\rho_{calc}/g \text{ cm}^{-3}$	1.717	1.789	1.781	1.846	1.799
μ/mm ⁻¹	1.265	1.364	1.259	6.887	1.878
F(000)	1060	1012	1012	414	860
Crystal size/mm ³	$0.5 \times 0.3 \times 0.05$	$0.35 \times 0.10 \times 0.10$	$0.30 \times 0.05 \times 0.05$	$0.15 \times 0.10 \times 0.02$	$0.1 \times 0.1 \times 0.05$
Padiation	Μο Κα	Μο Κα	Μο Κα	Cu Ka	Μο Κα
Radiation	$(\lambda = 0.71073)$	$(\lambda = 0.71073)$	$(\lambda = 0.71073)$	$(\lambda = 1.54184)$	$(\lambda = 0.71073)$
2⊖ range for data collection/°	4.696 to 60.912	5.228 to 60.76	6.134 to 60.586	7.882 to 150.492	4.706 to 54.962
	$-11 \le h \le 11$,	$-6 \le h \le 12$,	$-14 \le h \le 9$,	$-9 \le h \le 9,$	$-15 \le h \le 11$,
Index ranges	$-17 \le k \le 17$,	$-34 \le k \le 21$,	$-32 \le k \le 21$,	$-11 \le k \le 10$,	$-22 \le k \le 22$,
	$-29 \le l \le 29$	$-12 \le l \le 9$	-11 ≤ l ≤ 11	$-14 \le l \le 12$	$-9 \le l \le 9$
Reflections collected	25294	10122	5050	7240	8104
Indonandant	10742	4849	3279	3047	3608
raflactions	$[R_{int} = 0.0300,$	$[R_{int} = 0.0299],$	$[R_{int} = 0.0284,$	$[R_{int} = 0.0347,$	$[R_{int} = 0.0277,$
Teffections	$R_{sigma} = 0.0389$]	$R_{sigma} = 0.0521$]	$R_{sigma} = 0.0494$]	$R_{sigma} = 0.0420$]	$R_{sigma} = 0.0374$]
Data/restraints/ parameters	10742/6/594	4849/2/282	3279/6/298	3047/0/201	3608/0/211
Goodness-of-fit on F ²	1.057	1.031	1.053	1.074	1.048
Final R indexes	$R_1 = 0.0378,$	$R_1 = 0.0382,$	$R_1 = 0.0336$,	$R_1 = 0.0400,$	$R_1 = 0.0315,$
$[I \ge 2\sigma(I)]$	$wR_2 = 0.0961$	$wR_2 = 0.0769$	$wR_2 = 0.0714$	$wR_2 = 0.1044$	$wR_2 = 0.0732$
Final R indexes	$R_1 = 0.0473,$	$R_1 = 0.0556,$	$R_1 = 0.0391$,	$R_1 = 0.0465,$	$R_1 = 0.0400,$
[all data]	$wR_2 = 0.1029$	$wR_2 = 0.0846$	$wR_2 = 0.0745$	$wR_2 = 0.1108$	$wR_2 = 0.0787$
Largest diff. peak/hole /e Å ⁻³	+1.135/-0.872	+0.484/-0.463	+0.501/-0.369	+0.772/-0.835	+0.55/-0.46

Table S2 Details of the crystal structure determination for copper(II) complexes 1-5.