

Chemosensing of Neurotransmitters with Selectivity and Naked Eye Detection of L-DOPA Based on Fluorescent Zn(II)-Terpyridine Bearing Boronic Acid Complexes

Iván J. Bazany-Rodríguez,^a María K. Salomón-Flores,^a Alejandro O. Viviano-Posadas,^a Marco A. García-Eleno,^b Joaquín Barroso-Flores^{c*} Diego Martínez-Otero,^c and Alejandro Dorazco-González^{a*}

^aInstituto de Química, Universidad Nacional Autónoma de México, Circuito Exterior, Ciudad Universitaria, México, 04510, D.F., México.

^bCentro Conjunto de Investigación en Química Sustentable, UAEM-UNAM, Carretera Toluca-Atlaconulco Km 14.5, C. P. 50200, Toluca, Estado de México, México. Facultad de Química, Universidad Autónoma del Estado de México.

^cCentro Conjunto de Investigación en Química Sustentable, UAEM-UNAM, Carretera Toluca-Atlaconulco Km 14.5, C. P. 50200, Toluca, Estado de México, México. Instituto de Química, Universidad Nacional Autónoma de México.

Corresponding author: E-mail: adg@unam.mx

Electronic Supporting Information

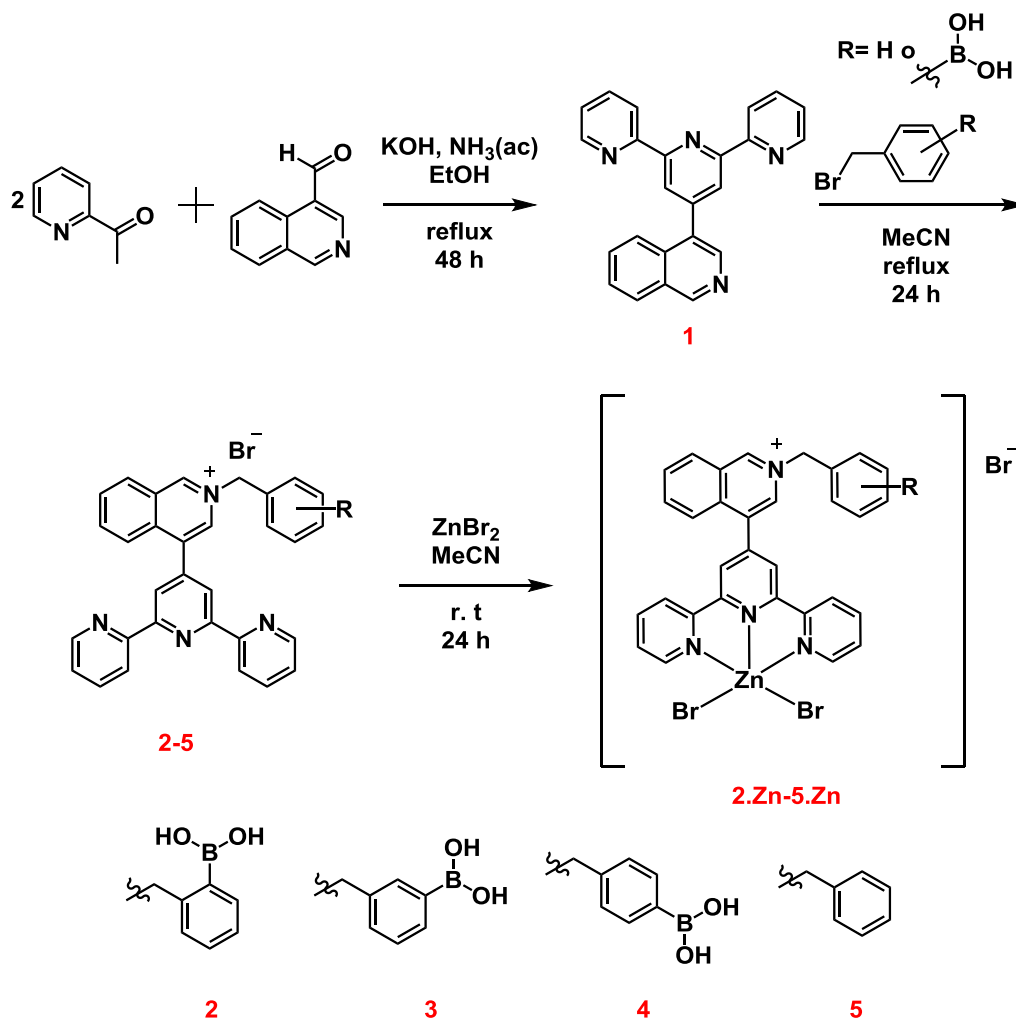


Table S1	Crystallographic data for 4.Zn and 5.Zn
Table S2	Hydrogen bonds for 4.Zn [Å and °].
Fig. S1	¹ H NMR spectrum of 1 in CDCl ₃ .
Fig. S2	¹³ C NMR spectrum of 1 in CDCl ₃ .
Fig. S3	Positive scan MS-EI spectrum of 1 .
Fig. S4	¹ H NMR spectrum of 2 in DMSO- <i>d</i> ₆ .
Fig. S5	¹³ C NMR spectrum of 2 in DMSO- <i>d</i> ₆ .
Fig. S6	¹¹ B NMR spectrum of 2 in DMSO- <i>d</i> ₆ .
Fig. S7	Positive scan MS-ESI spectrum of 2 .
Fig. S8	IR (ATR) spectrum of 2 .
Fig. S9	¹ H NMR spectrum of 3 in DMSO- <i>d</i> ₆ .
Fig. S10	¹³ C NMR spectrum of 3 in DMSO- <i>d</i> ₆ .
Fig. S11	¹¹ B NMR spectrum of 3 in DMSO- <i>d</i> ₆ .
Fig. S12	Positive scan MS-ESI spectrum of 3 .
Fig. S13	IR (ATR) spectrum of 3 .
Fig. S14	¹ H NMR spectrum of 4 in DMSO- <i>d</i> ₆ .
Fig. S15	¹³ C NMR spectrum of 4 in DMSO- <i>d</i> ₆ .
Fig. S16	¹¹ B NMR spectrum of 4 in DMSO- <i>d</i> ₆ .
Fig. S17	Positive scan MS-ESI spectrum of 4 .
Fig. S18	IR (ATR) spectrum of 4 .
Fig. S19	¹ H NMR spectrum of 5 in DMSO- <i>d</i> ₆ .
Fig. S20	¹³ C NMR spectrum of 5 in DMSO- <i>d</i> ₆ .
Fig. S21	Positive scan APCI-ESI spectrum of 5 .
Fig. S22	IR (ATR) spectrum of 5 .
Fig. S23	¹ H NMR spectrum of 2.Zn in DMSO- <i>d</i> ₆ .
Fig. S24	¹³ C NMR spectrum of 2.Zn in DMSO- <i>d</i> ₆ .
Fig. S25	¹¹ B NMR spectrum of 2.Zn in DMSO- <i>d</i> ₆ .
Fig. S26	Positive scan MS-ESI spectrum of 2.Zn .
Fig. S27	¹ H NMR spectrum of 3.Zn in DMSO- <i>d</i> ₆ .
Fig. S28	¹³ C NMR spectrum of 3.Zn in DMSO- <i>d</i> ₆ .
Fig. S29	¹¹ B NMR spectrum of 3.Zn in DMSO- <i>d</i> ₆ .
Fig. S30	Positive scan MS-ESI spectrum of 3.Zn .
Fig. S31	¹ H NMR spectrum of 4.Zn in DMSO- <i>d</i> ₆ .
Fig. S32	¹³ C NMR spectrum of 4.Zn in DMSO- <i>d</i> ₆ .
Fig. S33	¹¹ B NMR spectrum of 4.Zn in DMSO- <i>d</i> ₆ .
Fig. S34	Positive scan MS-ESI spectrum of 4.Zn .
Fig. S35	¹ H NMR spectrum of 5.Zn in DMSO- <i>d</i> ₆ .
Fig. S36	¹³ C NMR spectrum of 5.Zn in DMSO- <i>d</i> ₆ .
Fig. S37	Fluorescence pH-titration of buffered aqueous solution of 3.Zn (20 μM).
Fig. S38	Fluorescence pH-titration of buffered aqueous solution of 4.Zn (20 μM).
Fig. S39	Changes of emission spectra (λ _{ex} = 330 nm) of buffered aqueous solutions at pH 7.4 of 4.Zn (10 μM) upon addition of increasing amounts of L-DOPA.
Fig. S40	Stoichiometric analysis of 2.Zn by Job plot with L-DOPA at 305 nm.
Fig. S41	Stern–Volmer plot of 2.Zn upon addition of L-DOPA at 409 nm (λ _{ex} = 330 nm) at pH= 7.4.

Table S1 Crystallographic data for **4.Zn** and **5.Zn**.

Crystal data ^[a]	4.Zn	5.Zn
Formula	C ₃₁ H _{23.81} Br _{0.81} N ₄ O _{1.63} Zn	C ₃₃ H ₂₉ Br ₅ N ₄ OSZn ₂
MW (g mol ⁻¹)	792.27	1059.95
Temperature (K)	100(2)	100(2)
Crystal system	Monoclinic	Monoclinic
Space group	P2 ₁ /c	P2 ₁ /n
<i>a</i> (Å)	14.9965(3)	11.8967(3)
<i>b</i> (Å)	13.1054(2)	13.6816(3)
<i>c</i> (Å)	17.7644(3)	22.4595(5)
α (°)	90°	90°
β (°)	109.7882(7)°	98.2926(10)°
γ (°)	$\gamma = 90^\circ$	90°
<i>V</i> (Å ³)	3285.17(10)	3617.42(15)
<i>Z</i>	4	4
<i>P</i> _{calcd} (g cm ⁻³)	1.602	1.946
μ (mm ⁻¹)	5.566	8.905
<i>R</i> [<i>I</i> > 2 σ (<i>I</i>)] ^[b]	0.0747	0.0446
<i>R</i> _w ^[d]	0.2299	0.1175

[a] $\lambda_{\text{CuK}\alpha} = 1.54178$ Å; [b] $F_o > 4\sigma(F_o)$. [c] $R = \Sigma |F_o| - |F_c| / \Sigma |F_o|$ [d] all data.

Table S2. Hydrogen bonds for **4.Zn** [Å and °].

D-H...A	d(D-H)	d(H...A)	d(D...A)	<(DHA)
O(1)-H(1)...Br(3)	0.84	2.61	3.269(8)	136.7
O(2)-H(2A)...Br(3)	0.84	2.45	3.230(8)	153.9

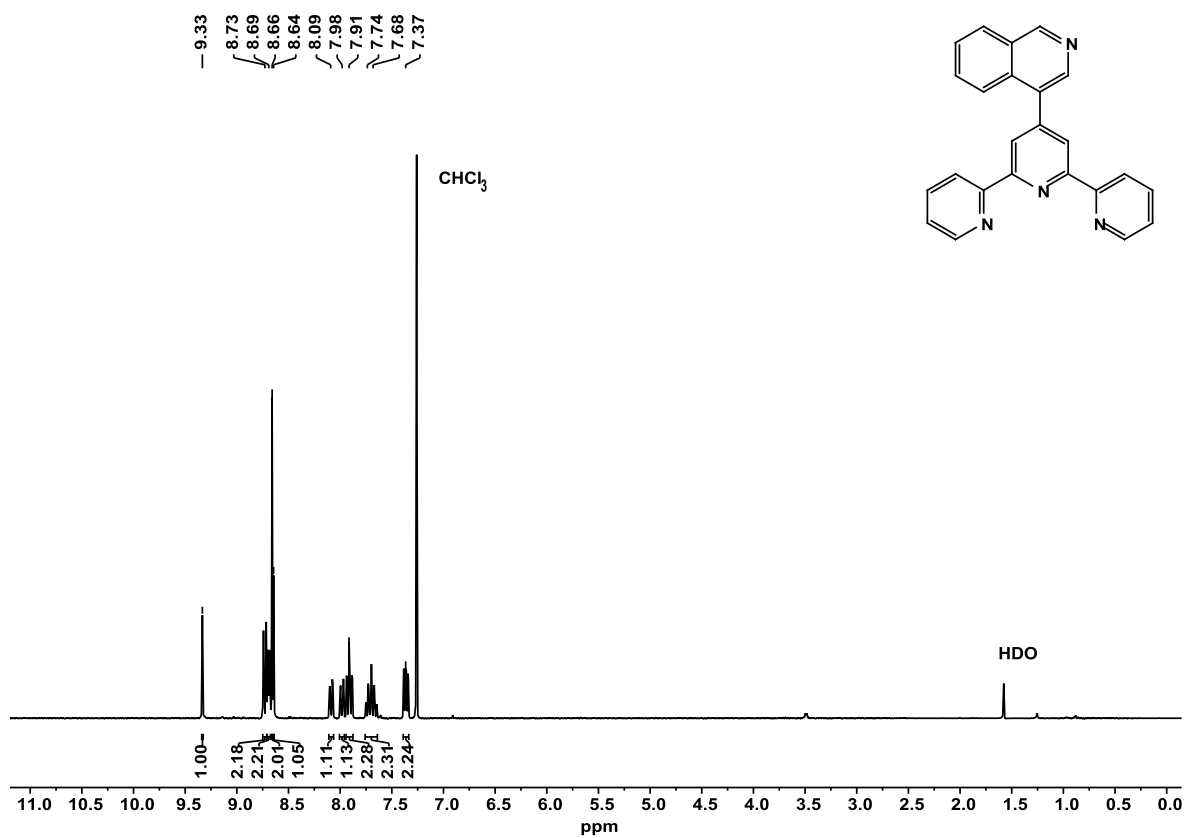


Fig. S1 ¹H NMR spectrum of **1** in CDCl₃.

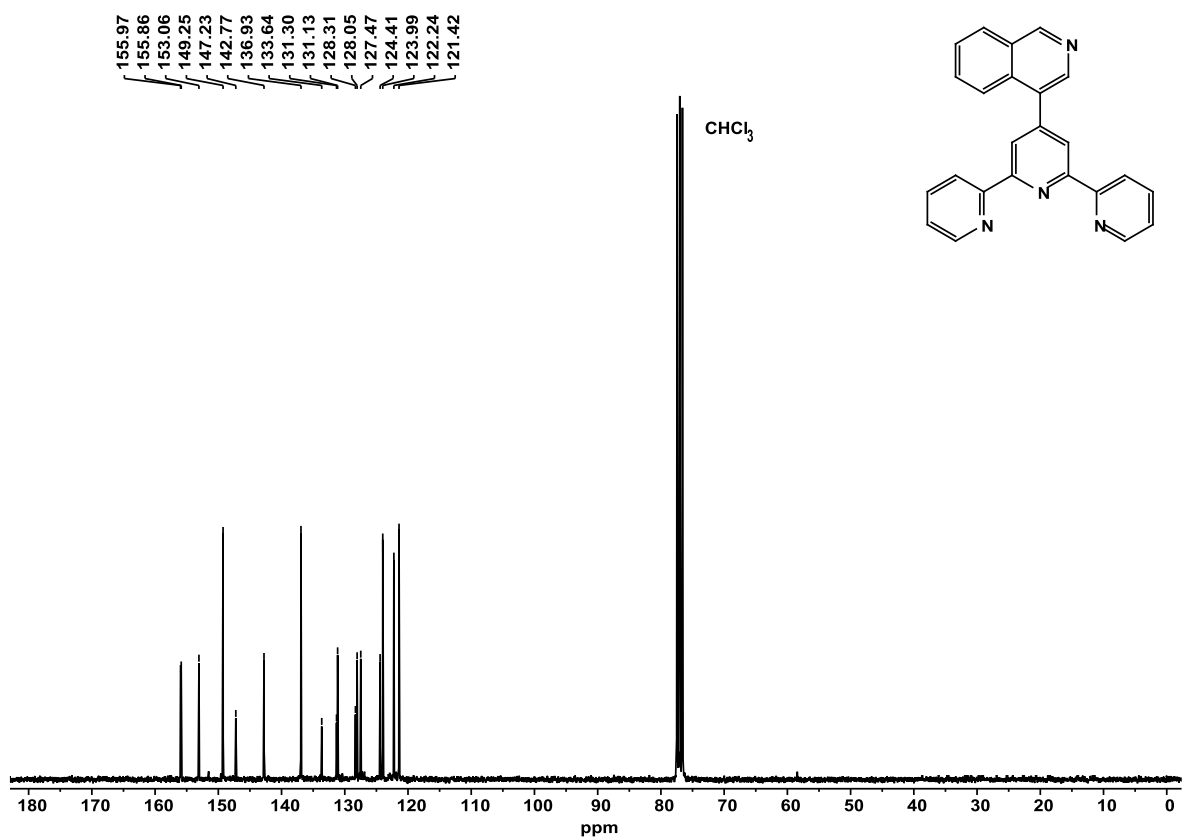


Fig. S2 ¹³C NMR spectrum of **1** in CDCl₃.

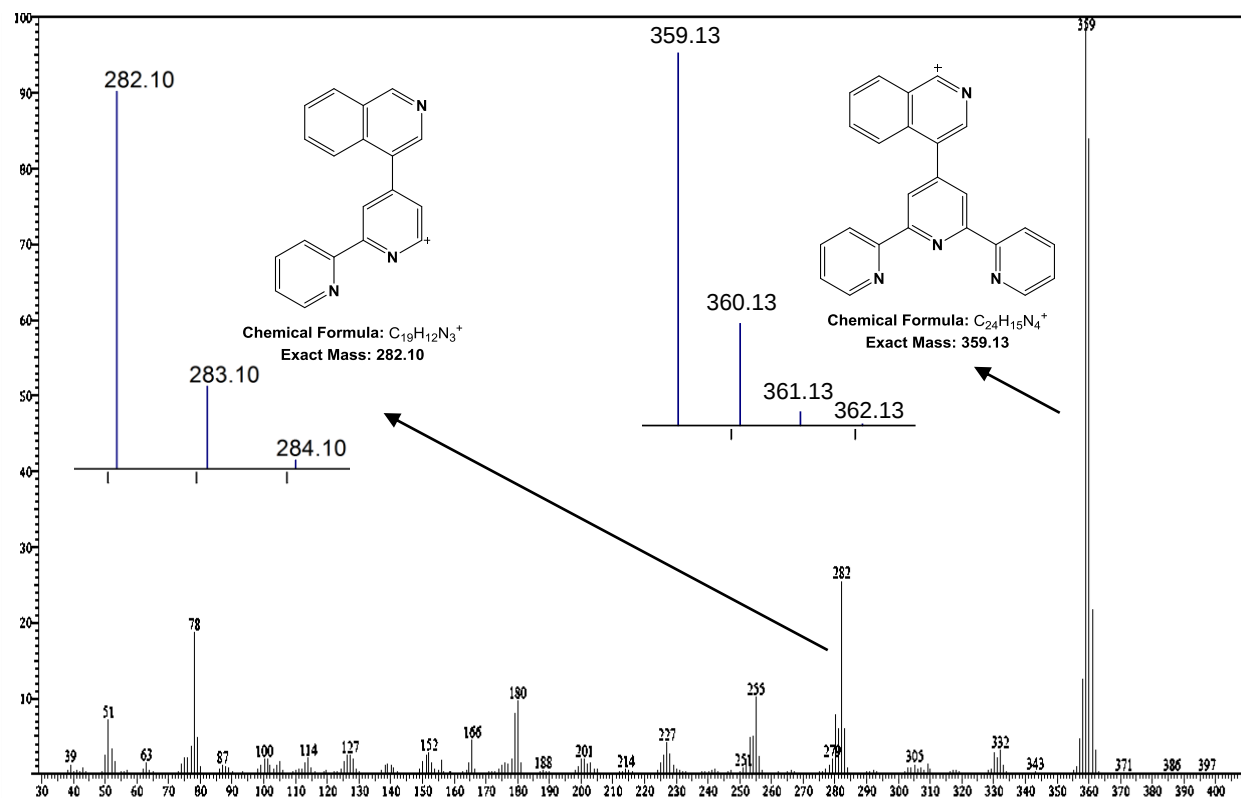


Fig. S3 Positive scan MS-EI spectrum of **1**. Inset: theoretically calculated MS isotopic patterns.

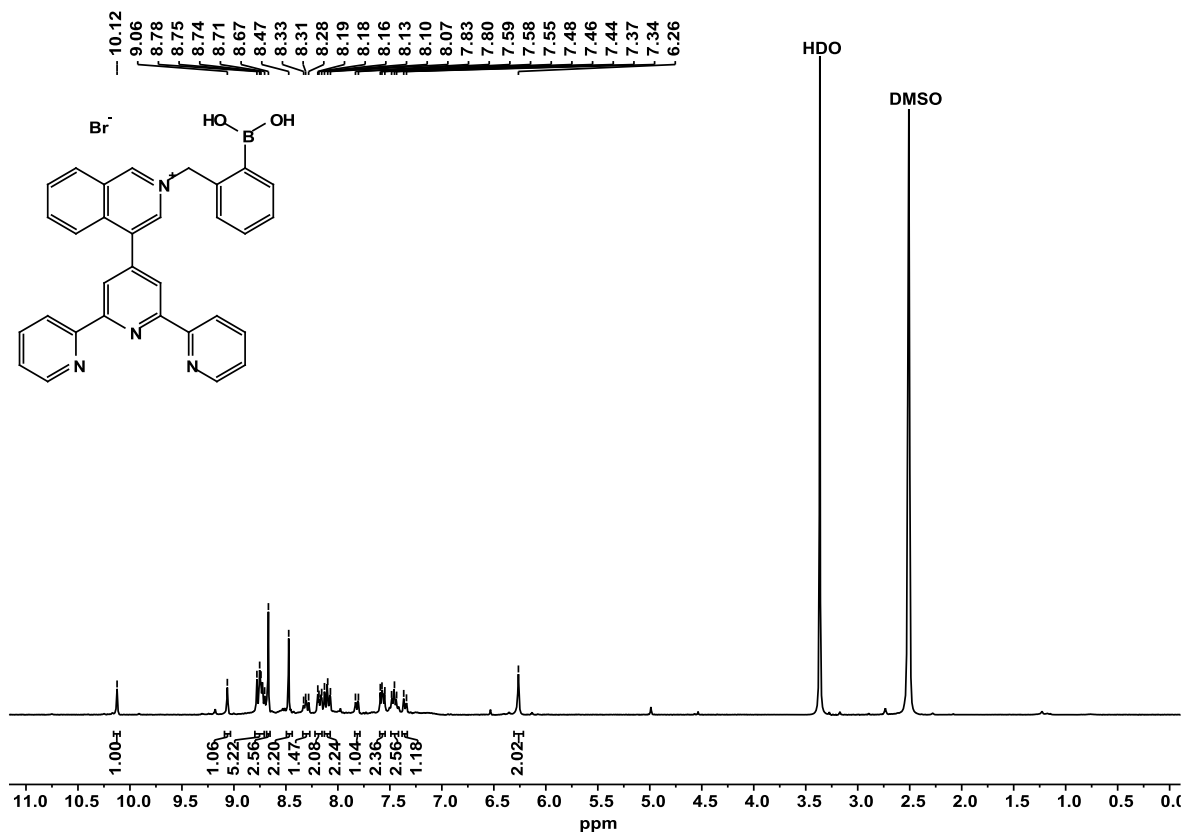


Fig. S4 1H NMR spectrum of **2** in $DMSO-d_6$.

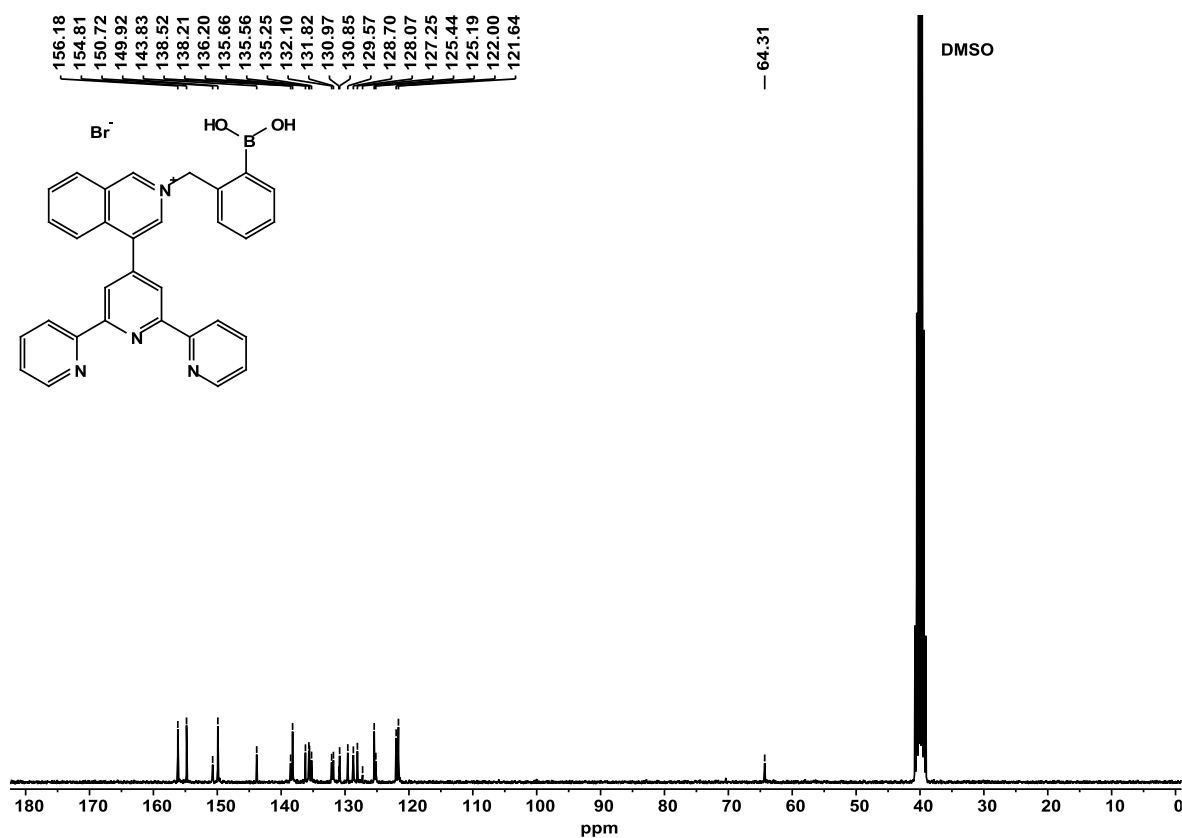


Fig. S5 ¹³C NMR spectrum of 2 in DMSO-*d*₆.

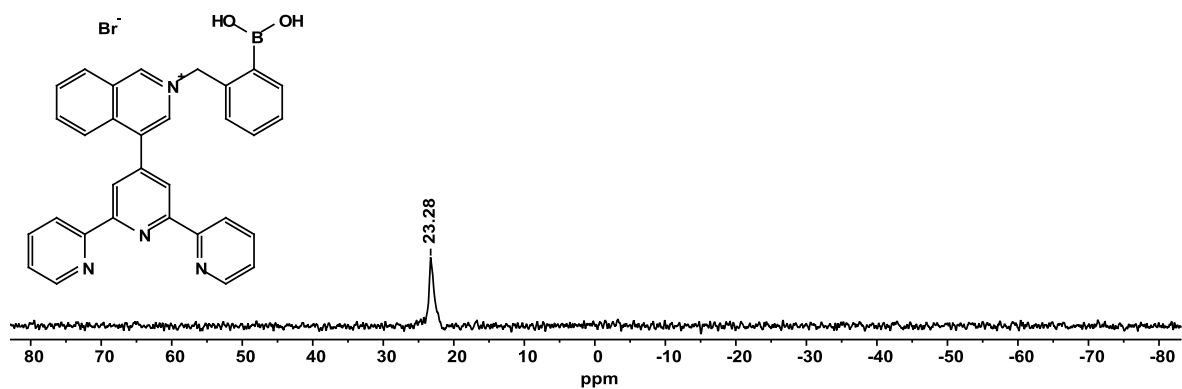


Fig. S6 ¹¹B NMR spectrum of 2 in DMSO-*d*₆.

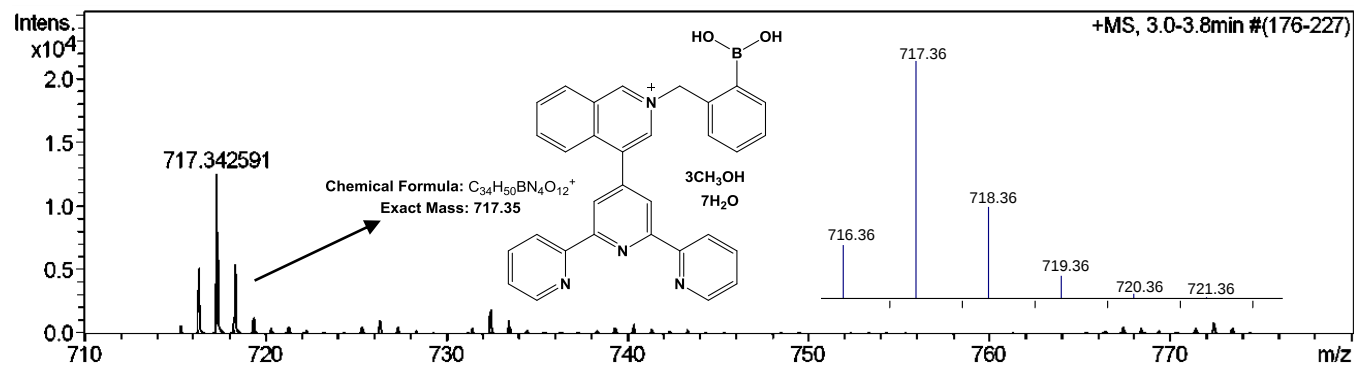


Fig. S7 Positive scan MS-ESI spectrum of 2. Inset: theoretically calculated MS isotopic patterns.

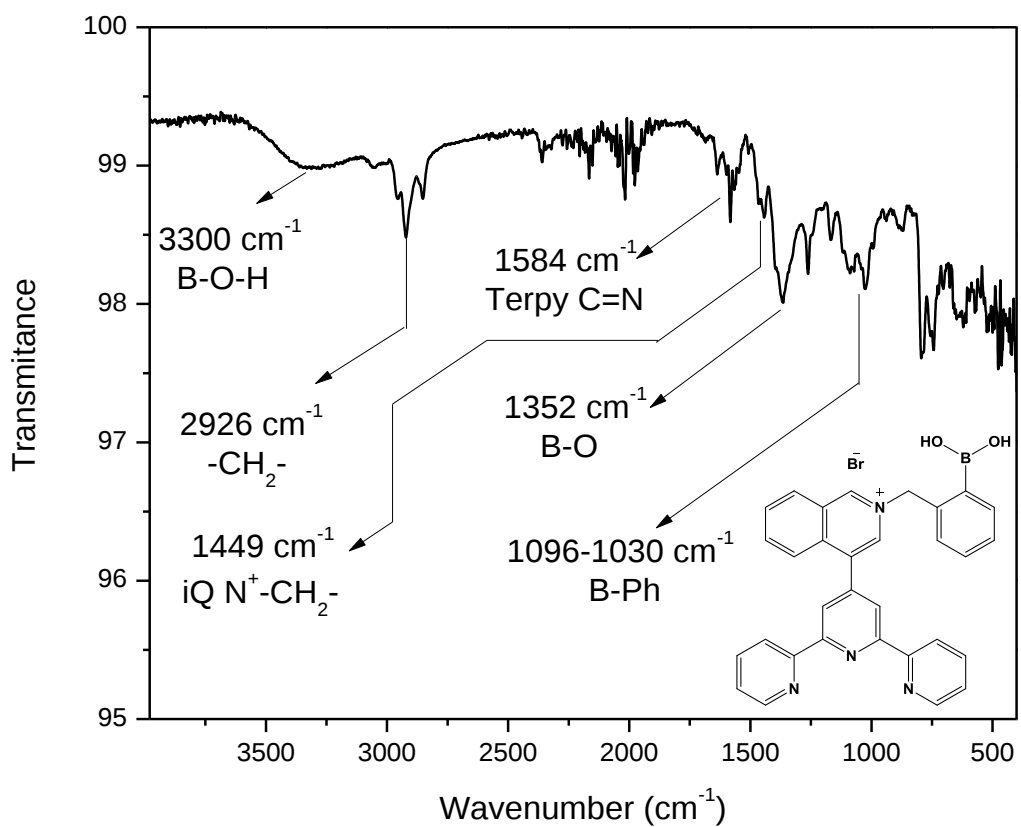


Fig. S8 IR (ATR) spectrum of 2.

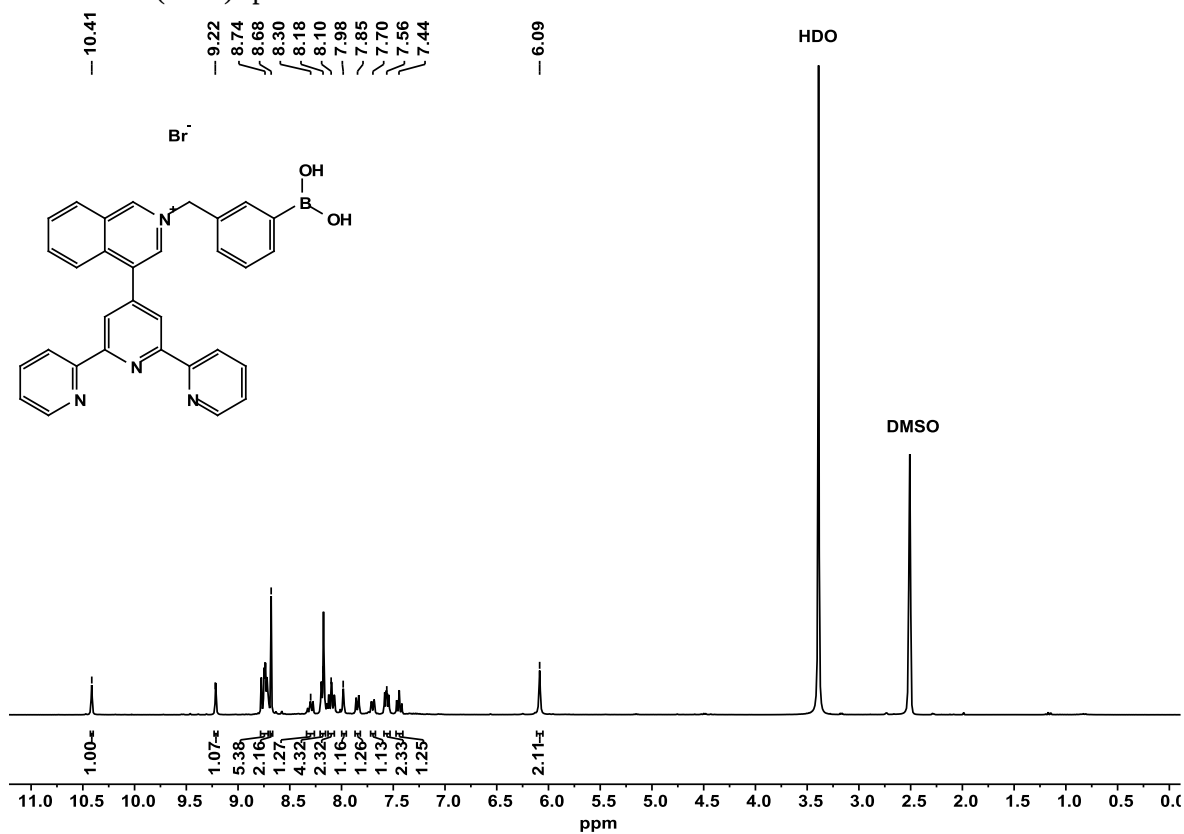


Fig. S9 ^1H NMR spectrum of 3 in $\text{DMSO}-d_6$.

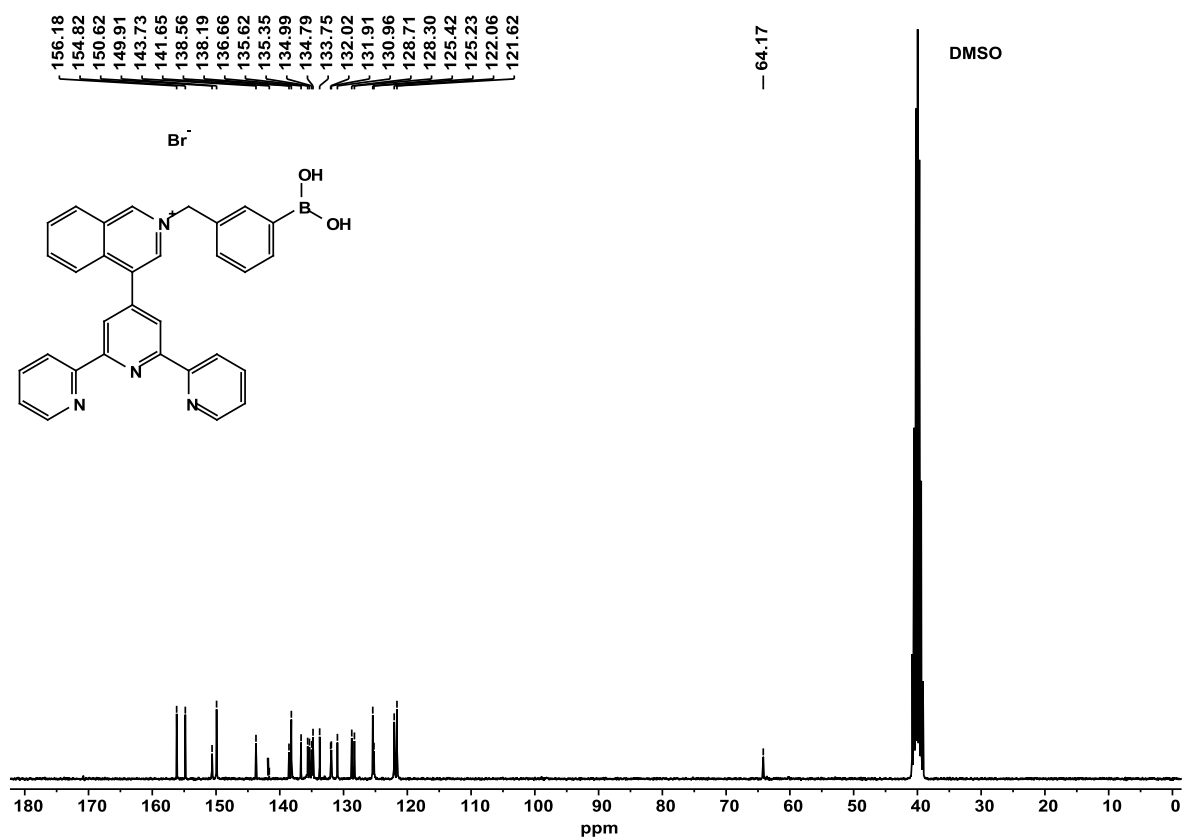


Fig. S10 ¹³C NMR spectrum of 3 in DMSO-*d*₆.

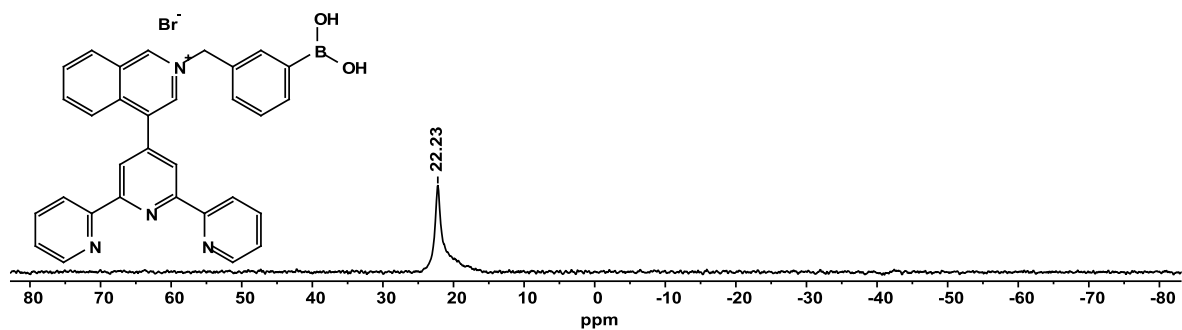
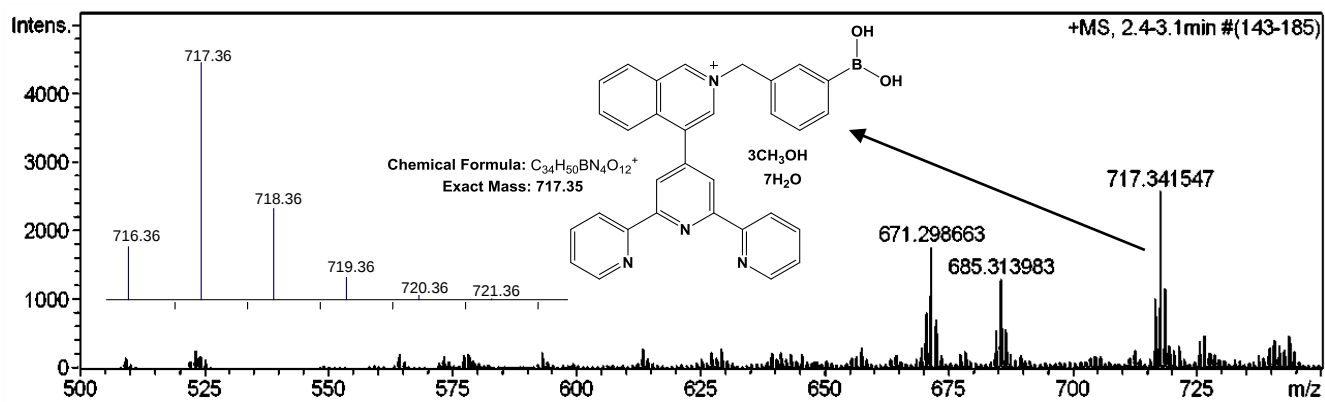


Fig. S11 ¹¹B NMR spectrum of 3 in DMSO-*d*₆.



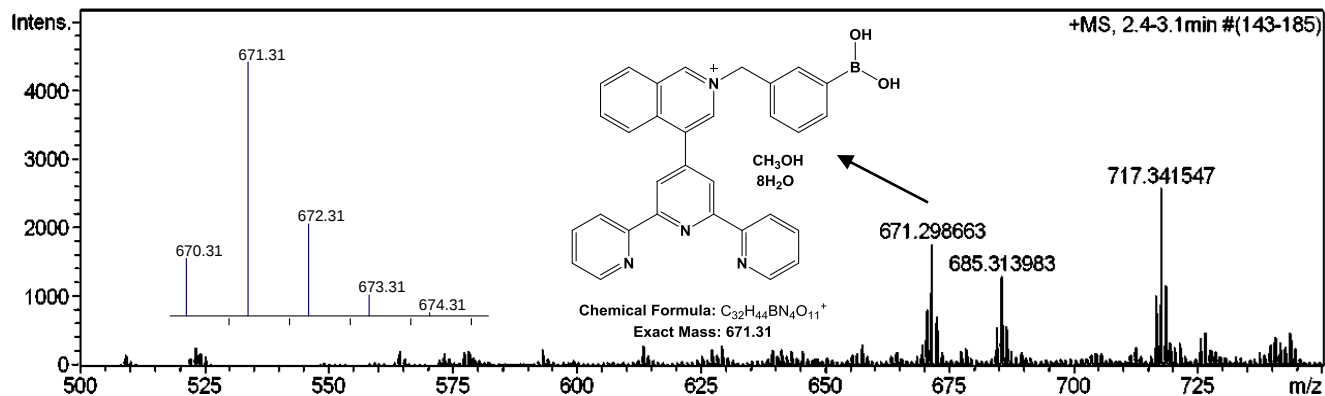
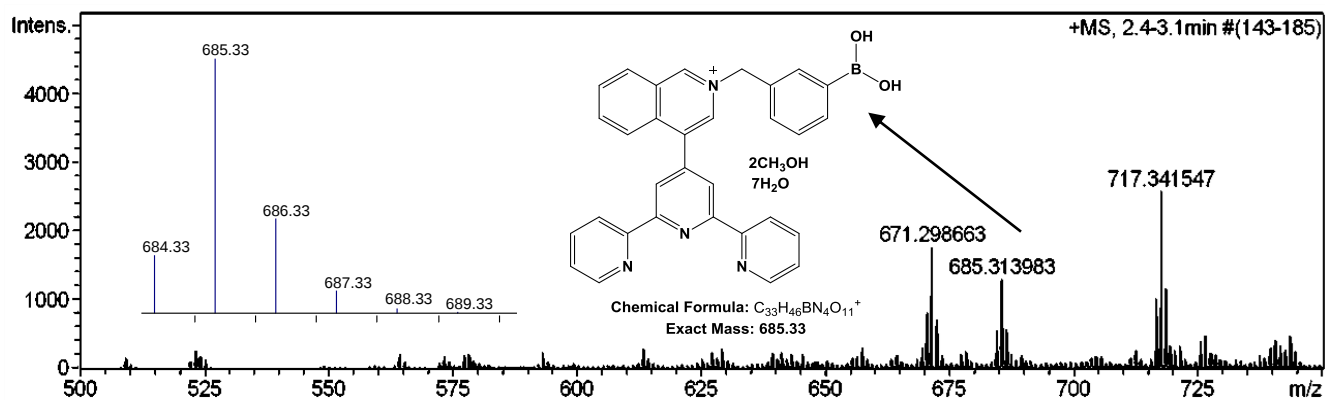


Fig. S12 Positive scan MS-ESI spectrum of **3**. Inset: theoretically calculated MS isotopic patterns.

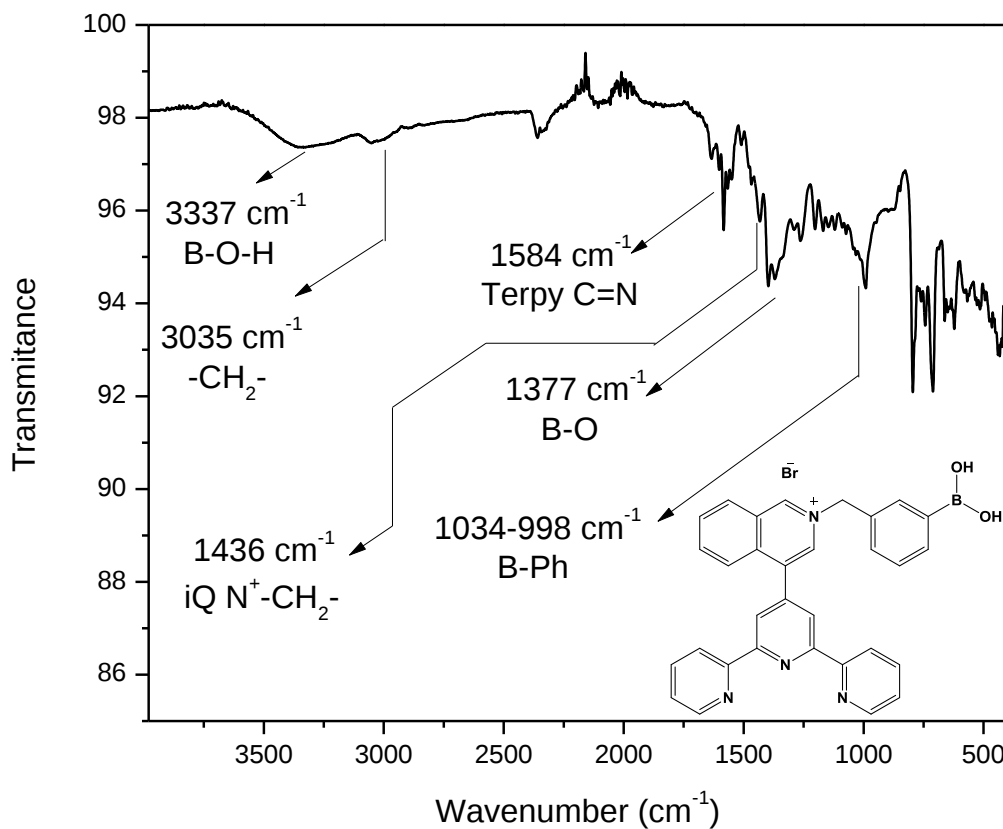


Fig. S13 IR (ATR) spectrum of **3**.

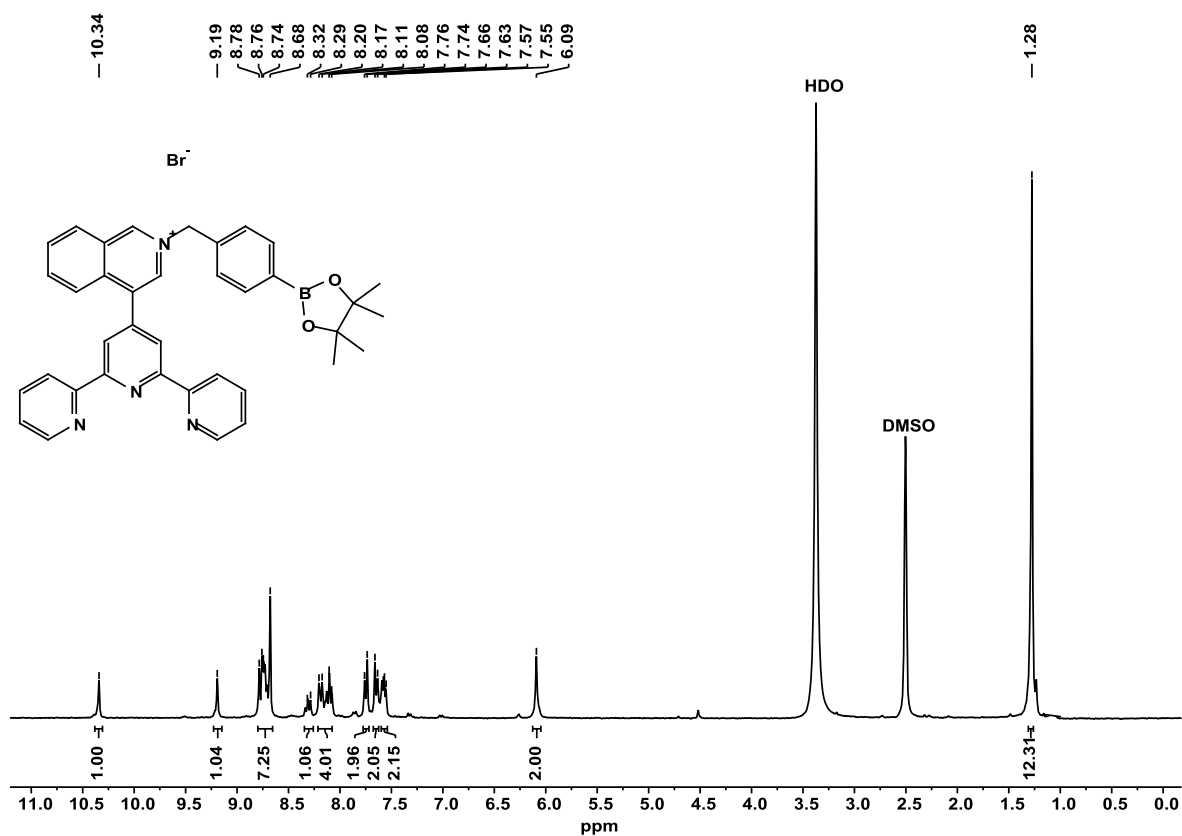


Fig. S14 ¹H NMR spectrum of 4 in DMSO-*d*₆.

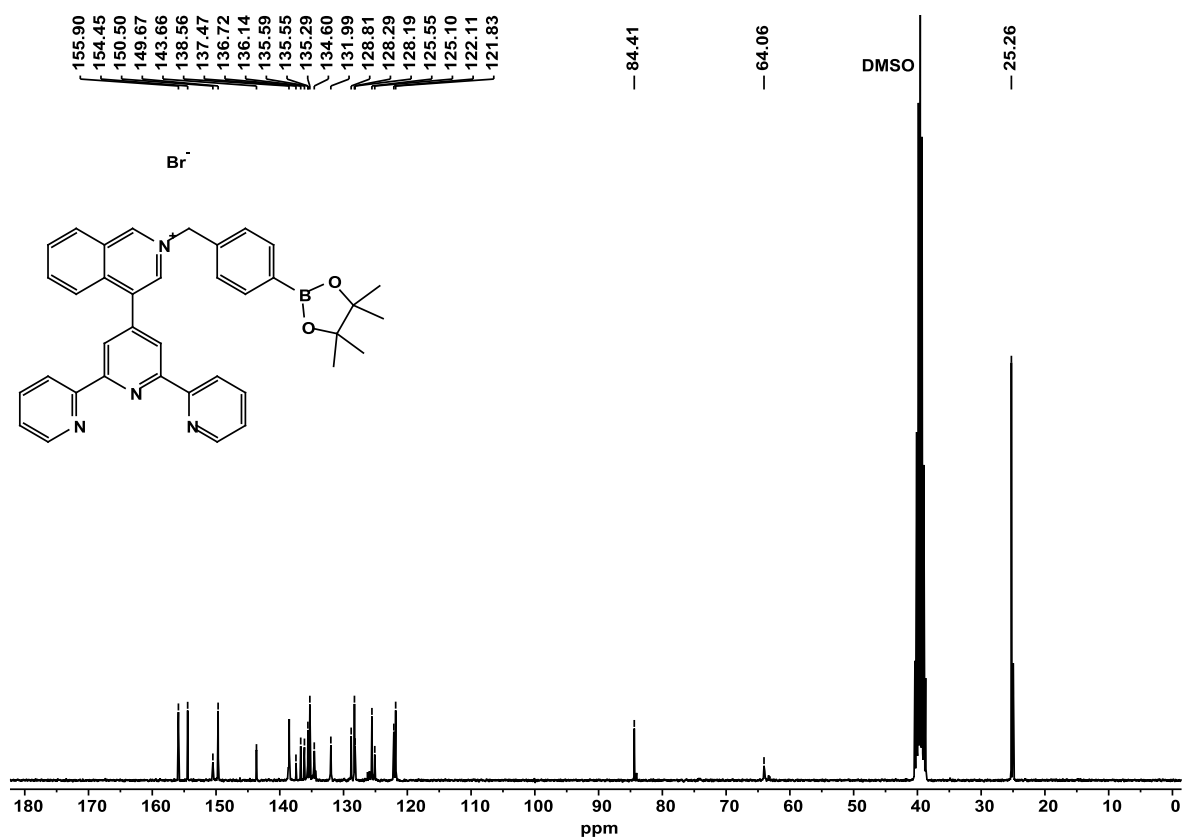


Fig. S15 ¹³C NMR spectrum of 4 in DMSO-*d*₆.

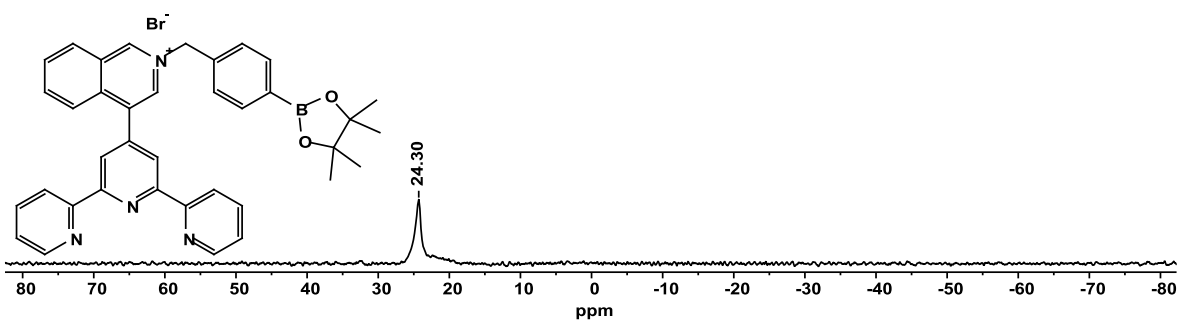


Fig. S16 ^{11}B NMR spectrum of **4** in $\text{DMSO-}d_6$.

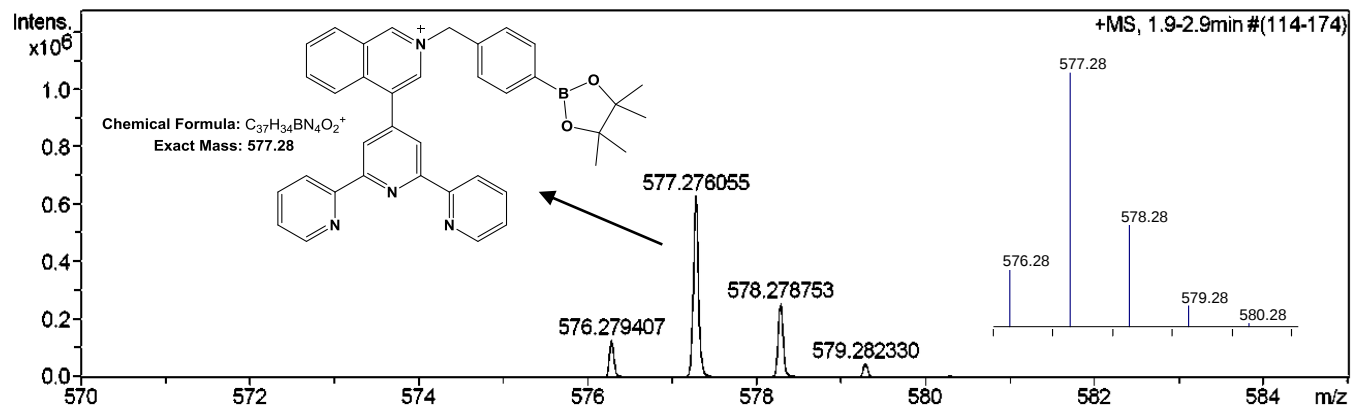


Fig. S17 Positive scan MS-ESI spectrum of **4**. Inset: theoretically calculated MS isotopic patterns.

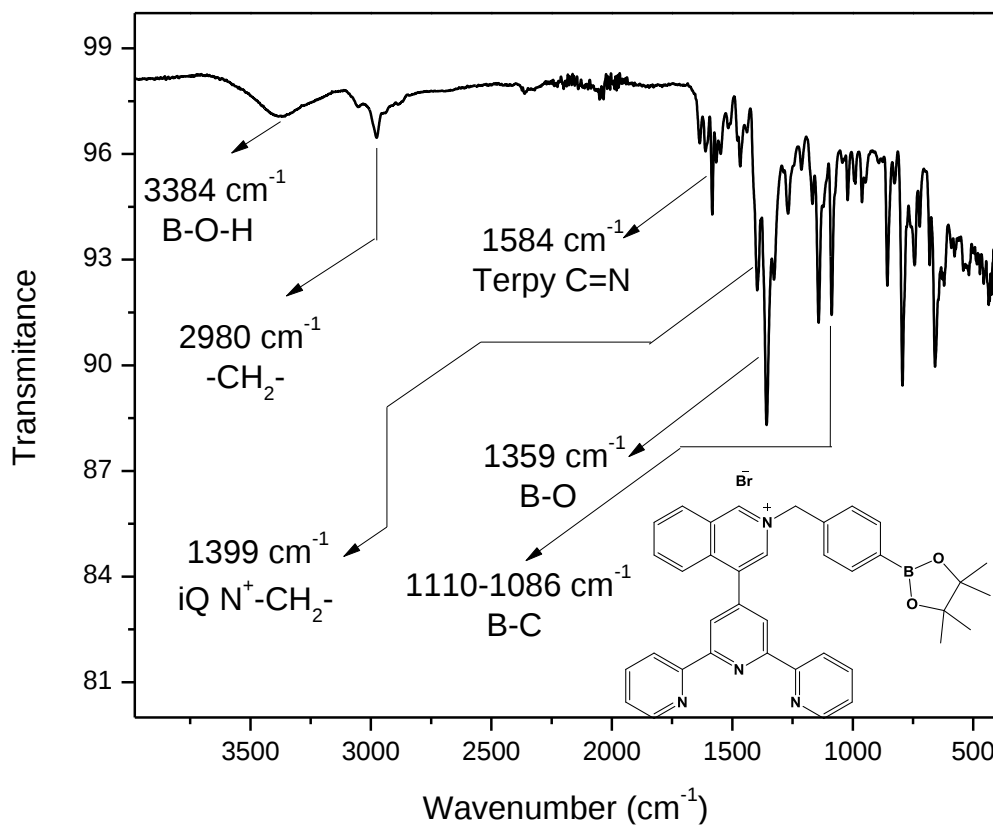


Fig. S18 IR (ATR) spectrum of **4**.

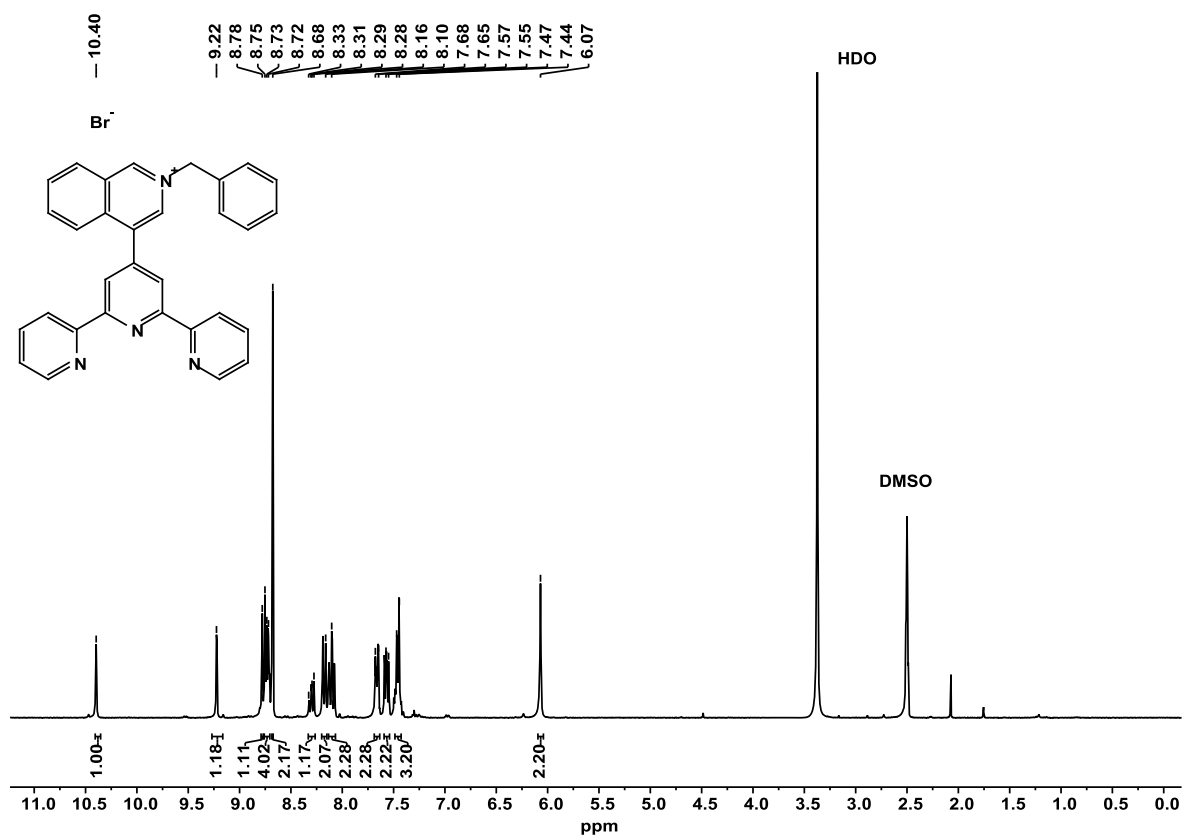


Fig. S19 ¹H NMR spectrum of 5 in DMSO-*d*₆.

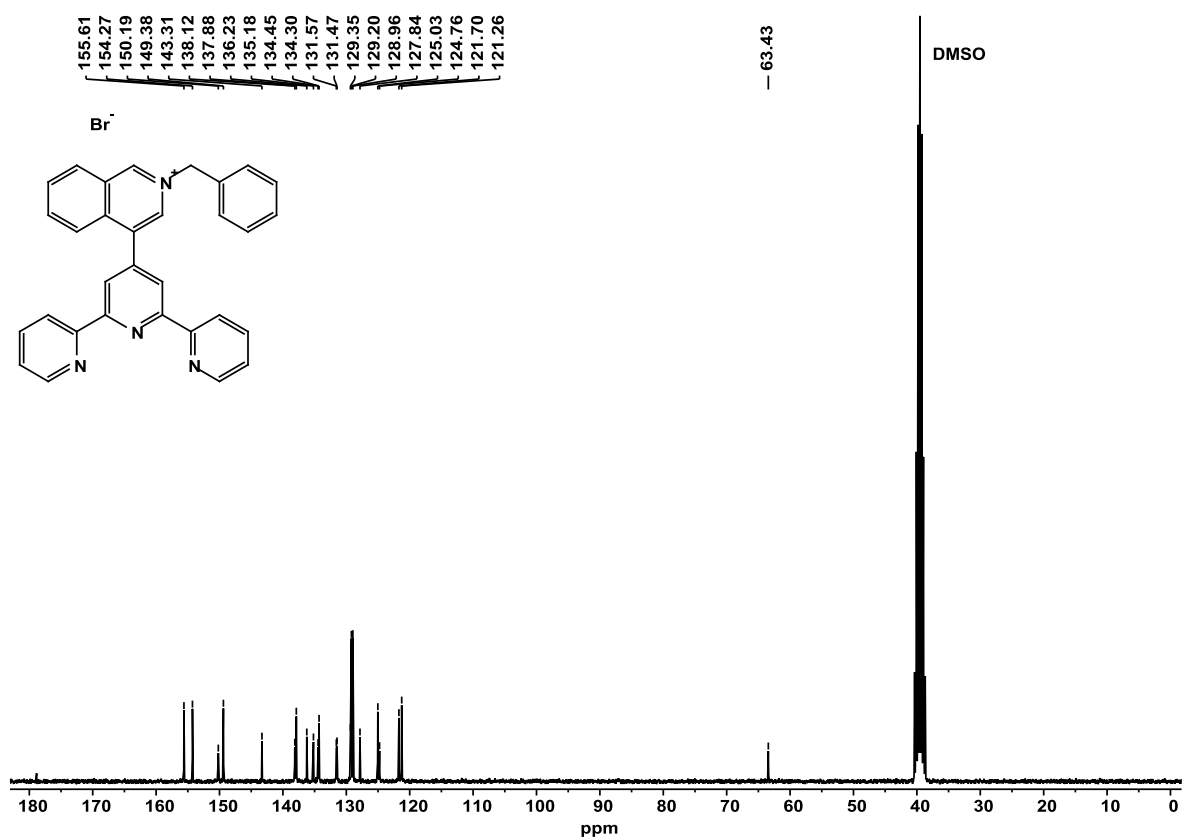


Fig. S20 ¹³C NMR spectrum of 5 in DMSO-*d*₆.

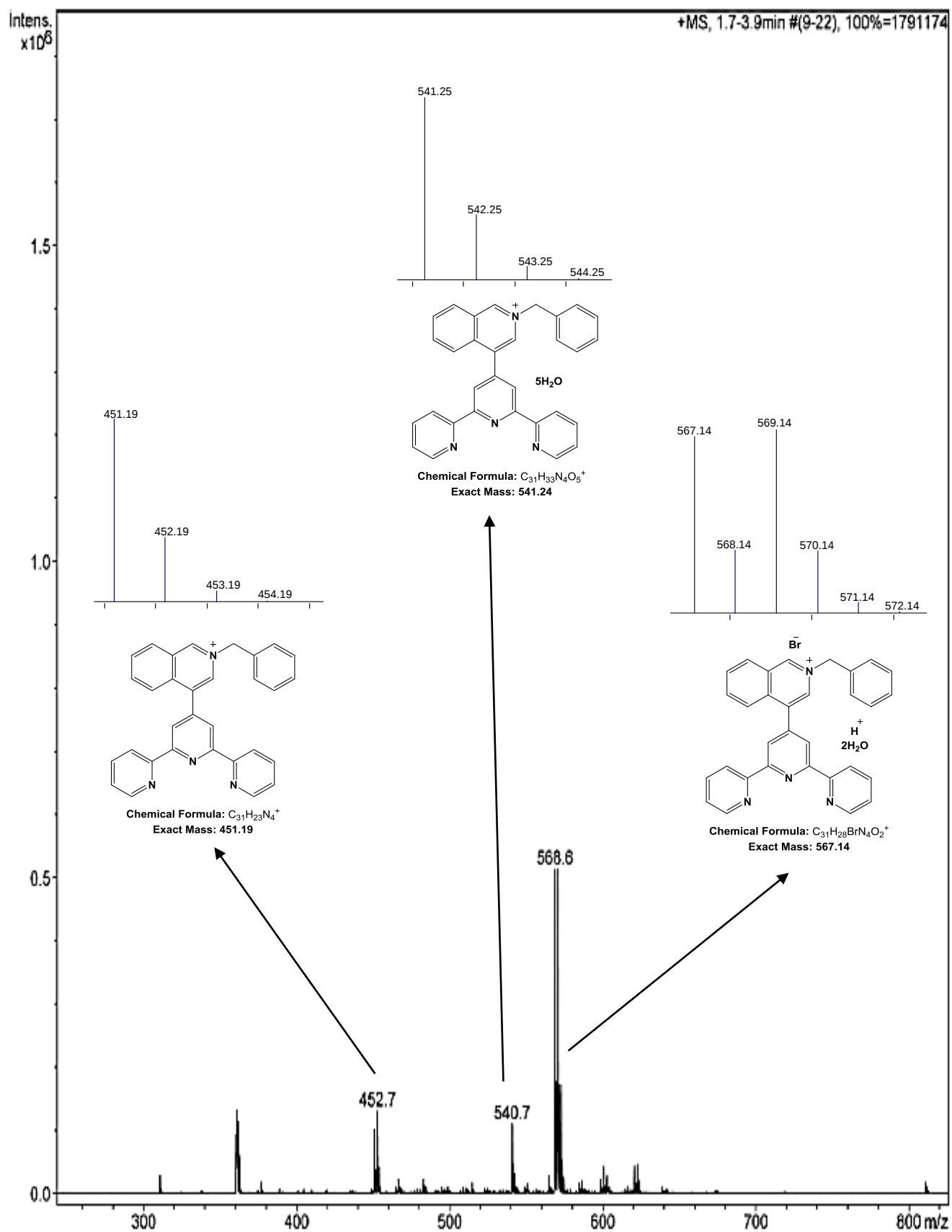


Fig. S21 Positive scan MS-APCI spectrum of 5. Inset: theoretically calculated MS isotopic patterns.

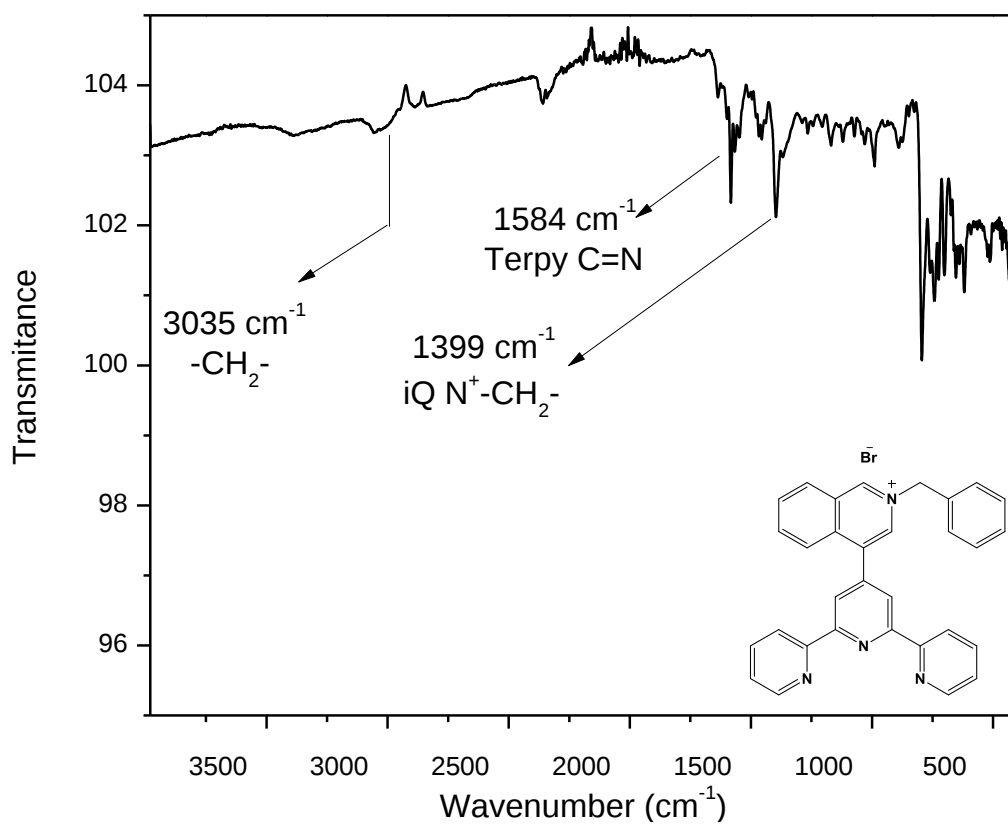


Fig. S22 IR (ATR) spectrum of 5.

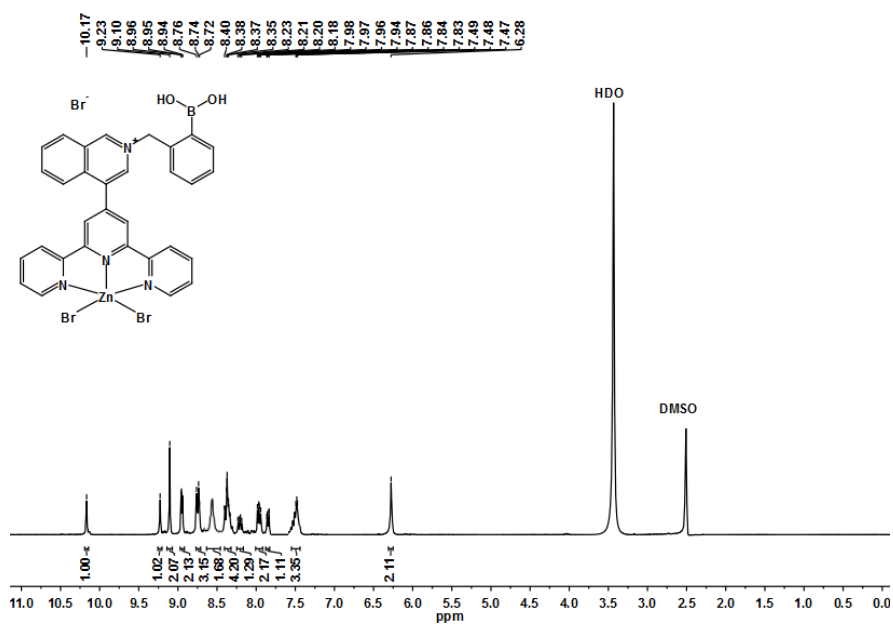


Fig. S23 ^1H NMR spectrum of 2.Zn in $\text{DMSO}-d_6$.

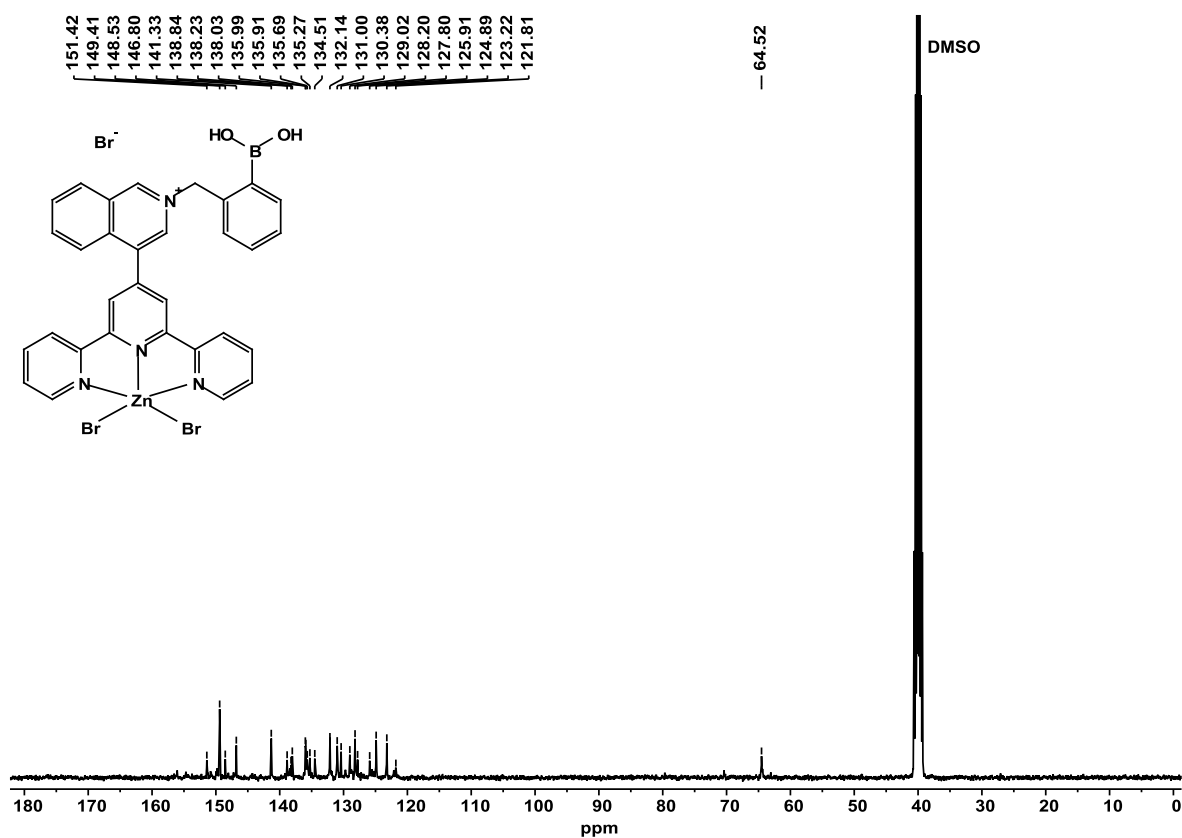


Fig. S24 ¹³C NMR spectrum of **2.Zn** in DMSO-*d*₆.

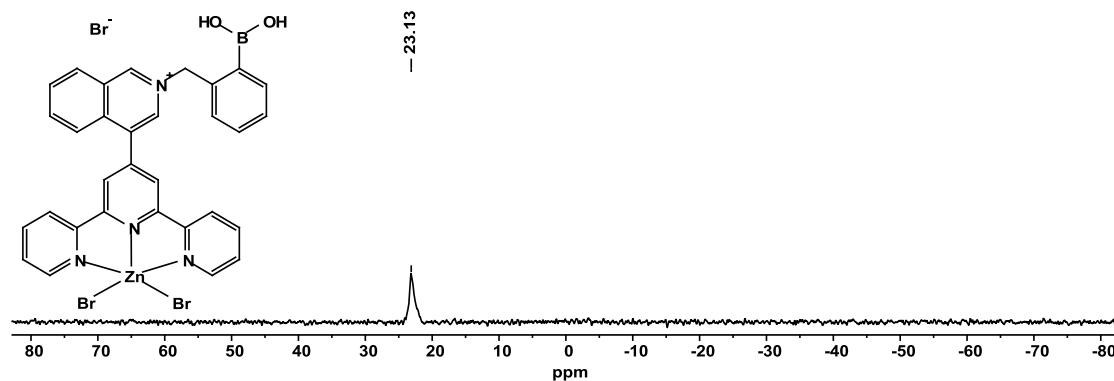


Fig. S25 ¹¹B NMR spectrum of **2.Zn** in DMSO-*d*₆.

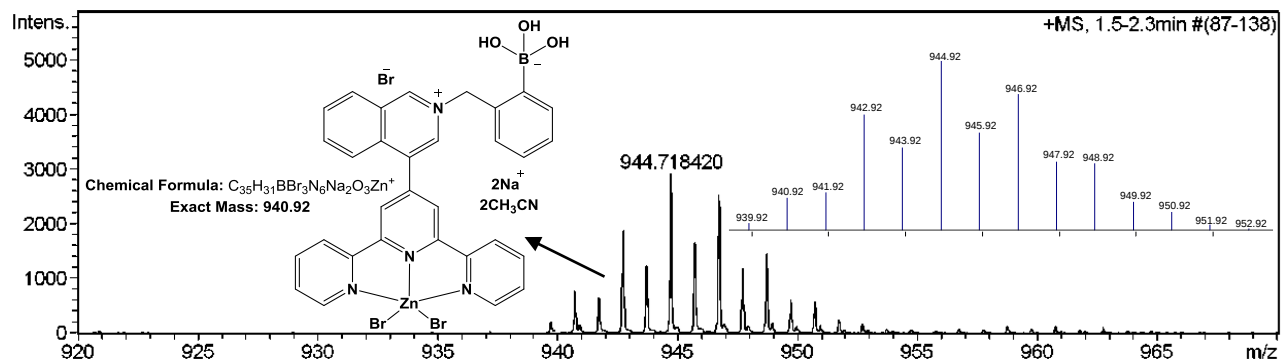


Fig. S26 Positive scan MS-ESI spectrum of **2.Zn**. Inset: theoretically calculated MS isotopic patterns.

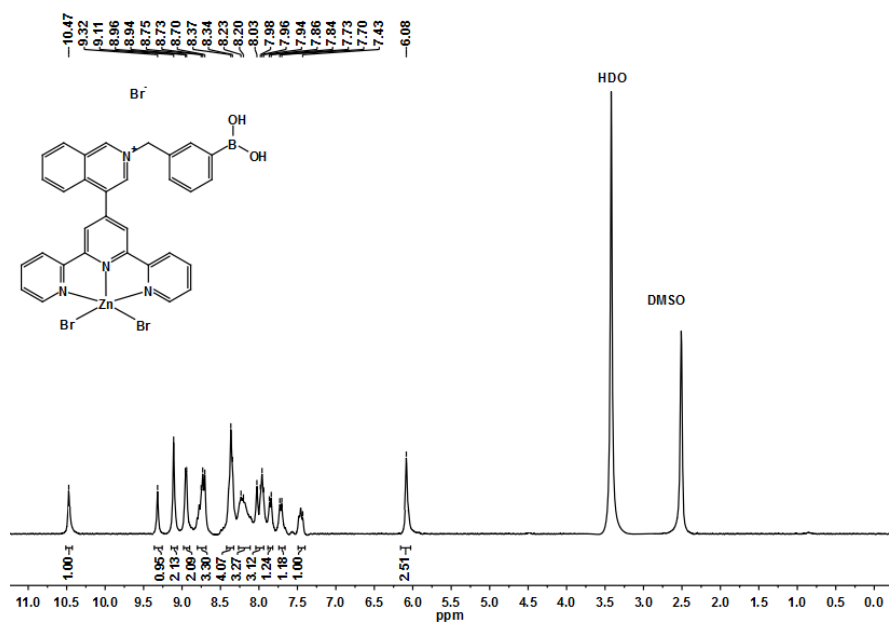


Fig. S27 ¹H NMR spectrum of 3.Zn in DMSO-*d*₆.

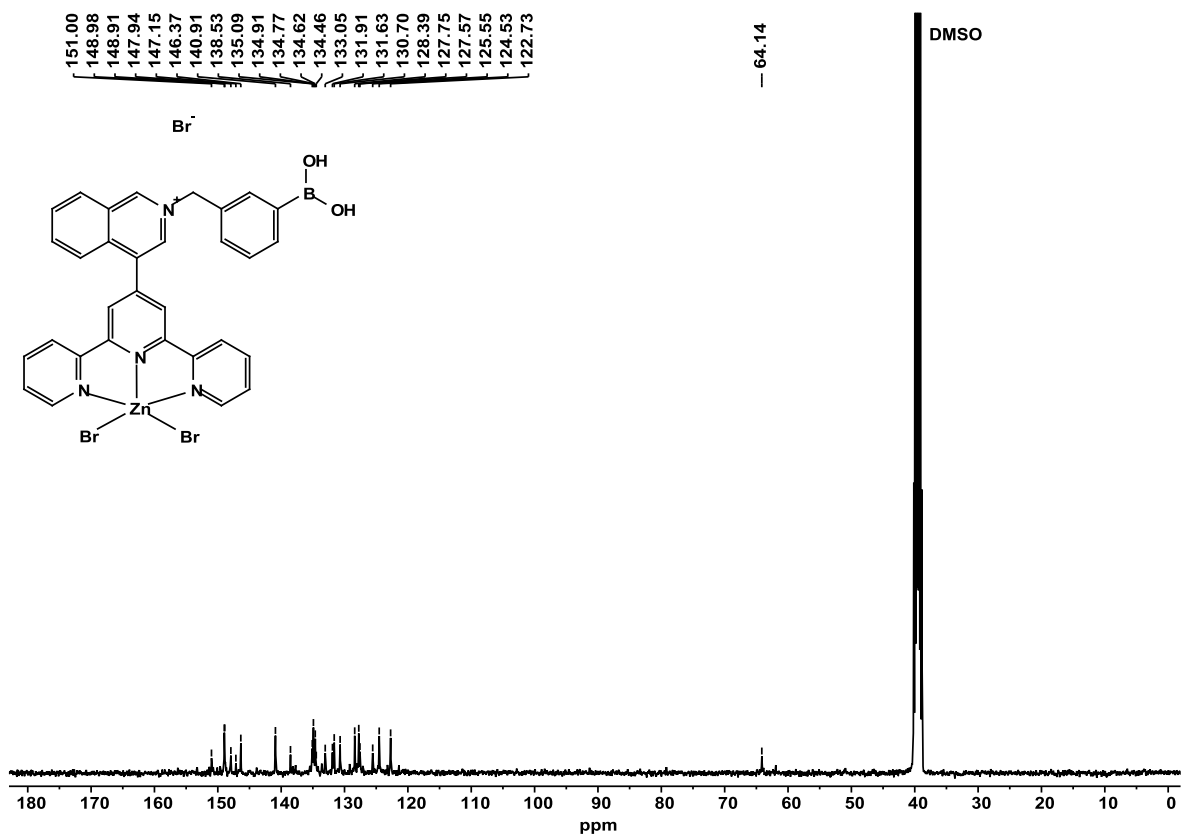


Fig. S28 ¹³C NMR spectrum of 3.Zn in DMSO-*d*₆.

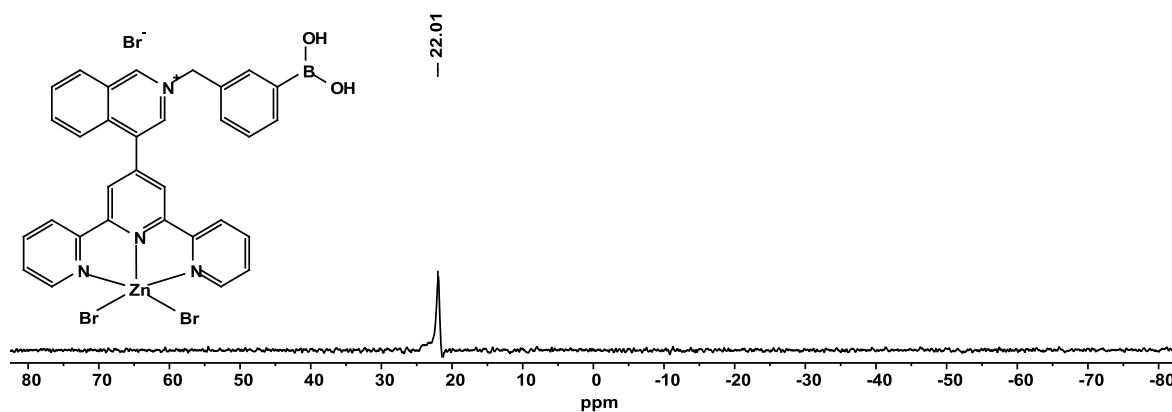


Fig. S29 ^{11}B NMR spectrum of **3.Zn** in $\text{DMSO-}d_6$.

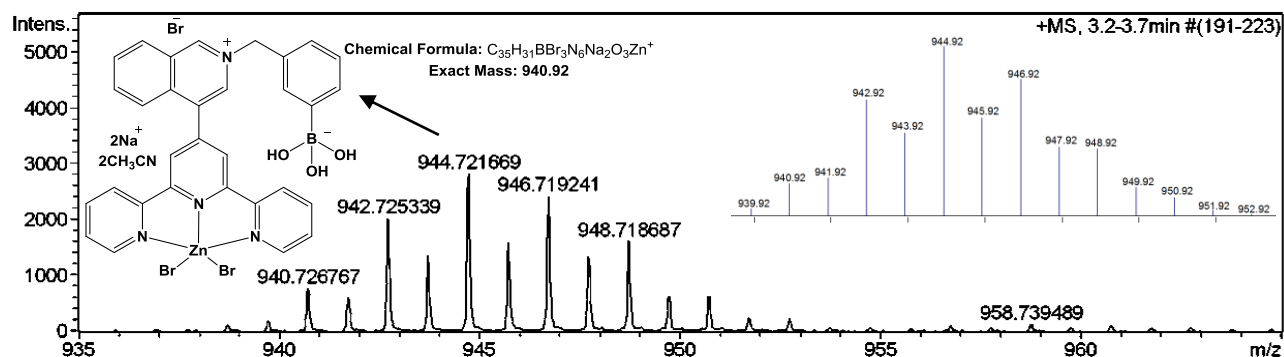


Fig. S30 Positive scan MS-ESI spectrum of **3.Zn**. Inset: theoretically calculated MS isotopic patterns.

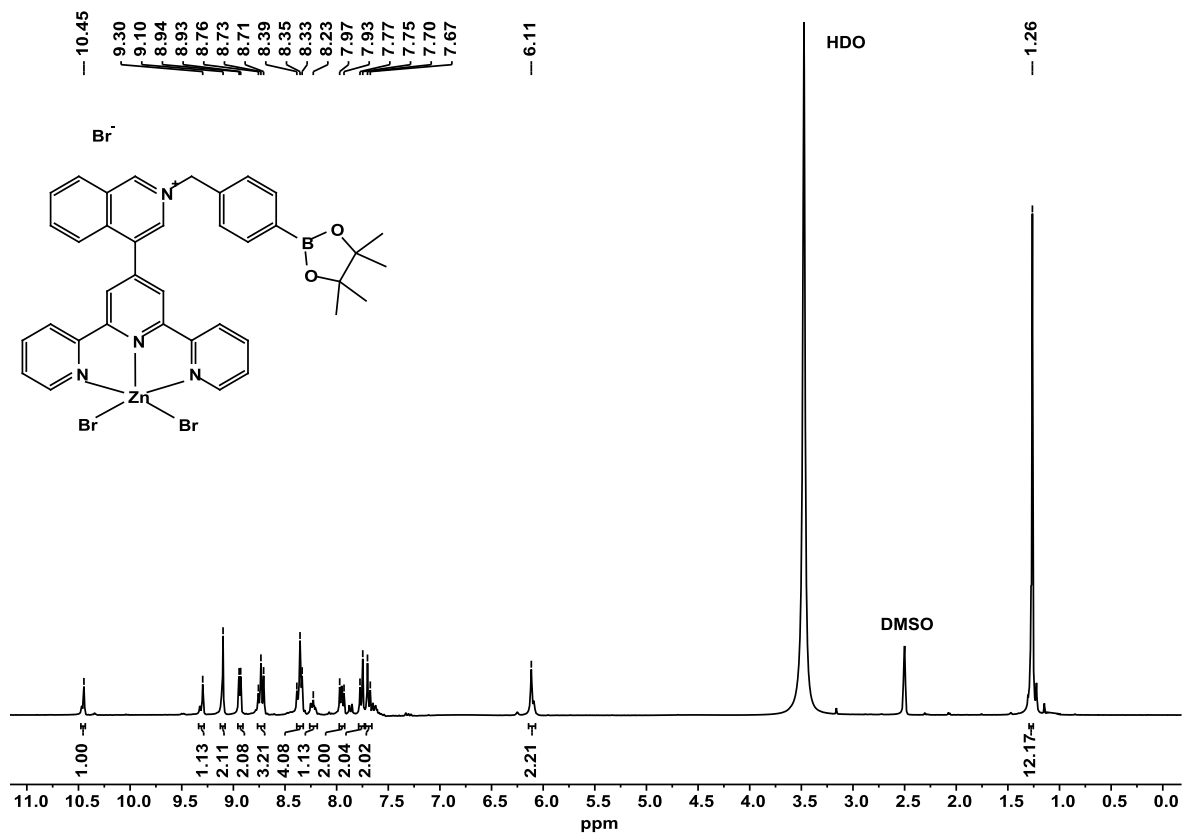


Fig. S31 ^1H NMR spectrum of **4.Zn** in $\text{DMSO-}d_6$.

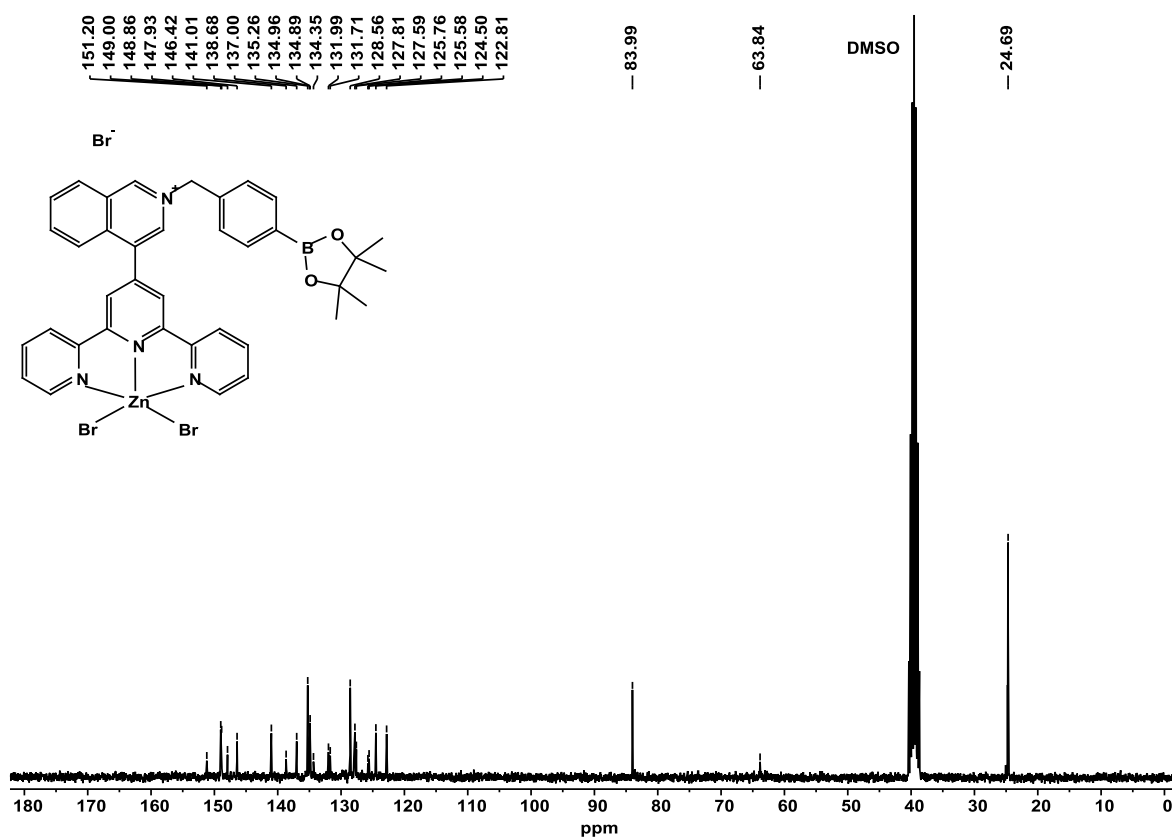


Fig. S32 ^{13}C NMR spectrum of **4.Zn** in $\text{DMSO-}d_6$.

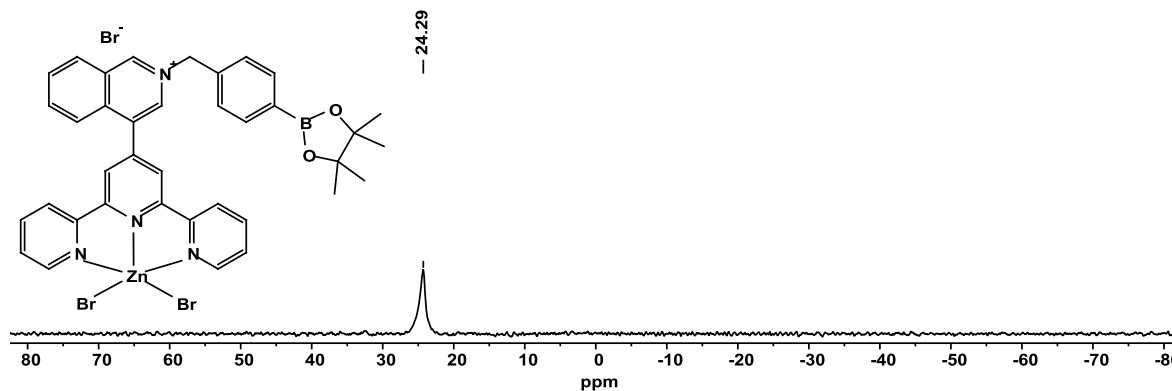


Fig. S33 ^{11}B NMR spectrum of **4.Zn** in $\text{DMSO-}d_6$.

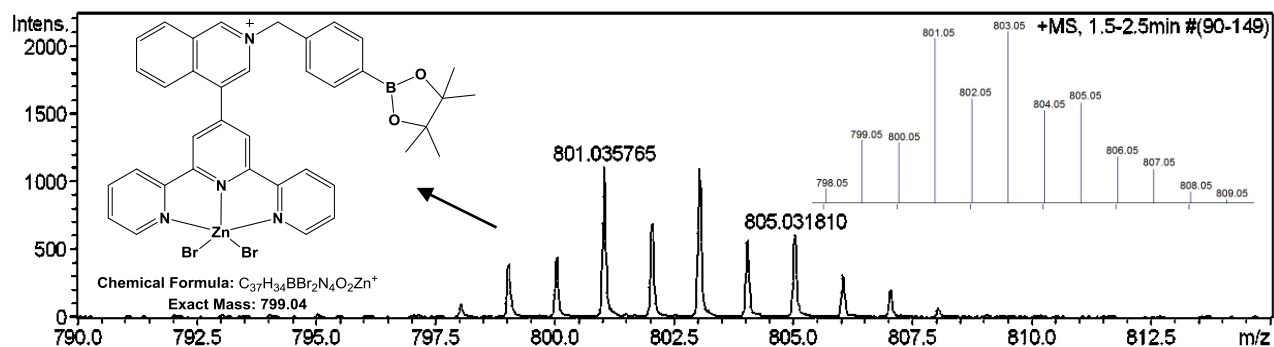


Fig. S34 Positive scan MS-ESI spectrum of **4.Zn**. Inset: theoretically calculated MS isotopic patterns.

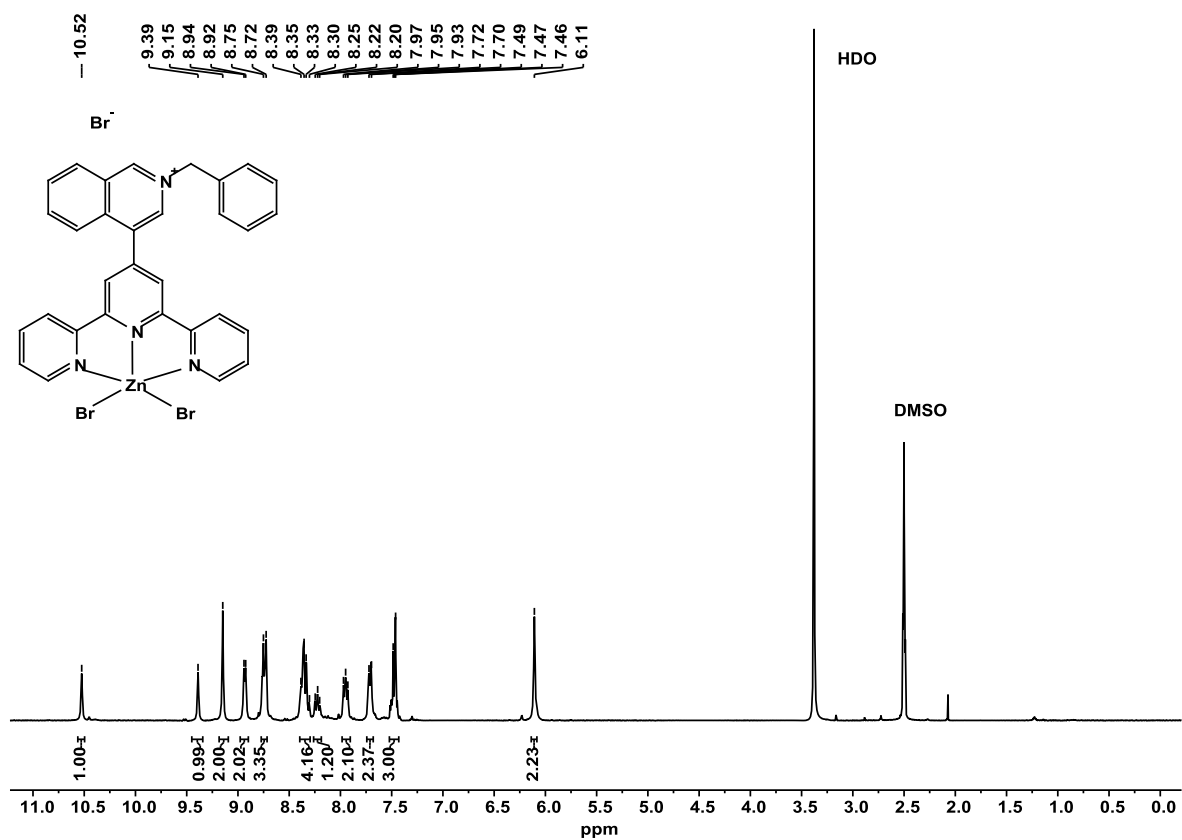


Fig. S35 ^1H NMR spectrum of **5.Zn** in $\text{DMSO-}d_6$.

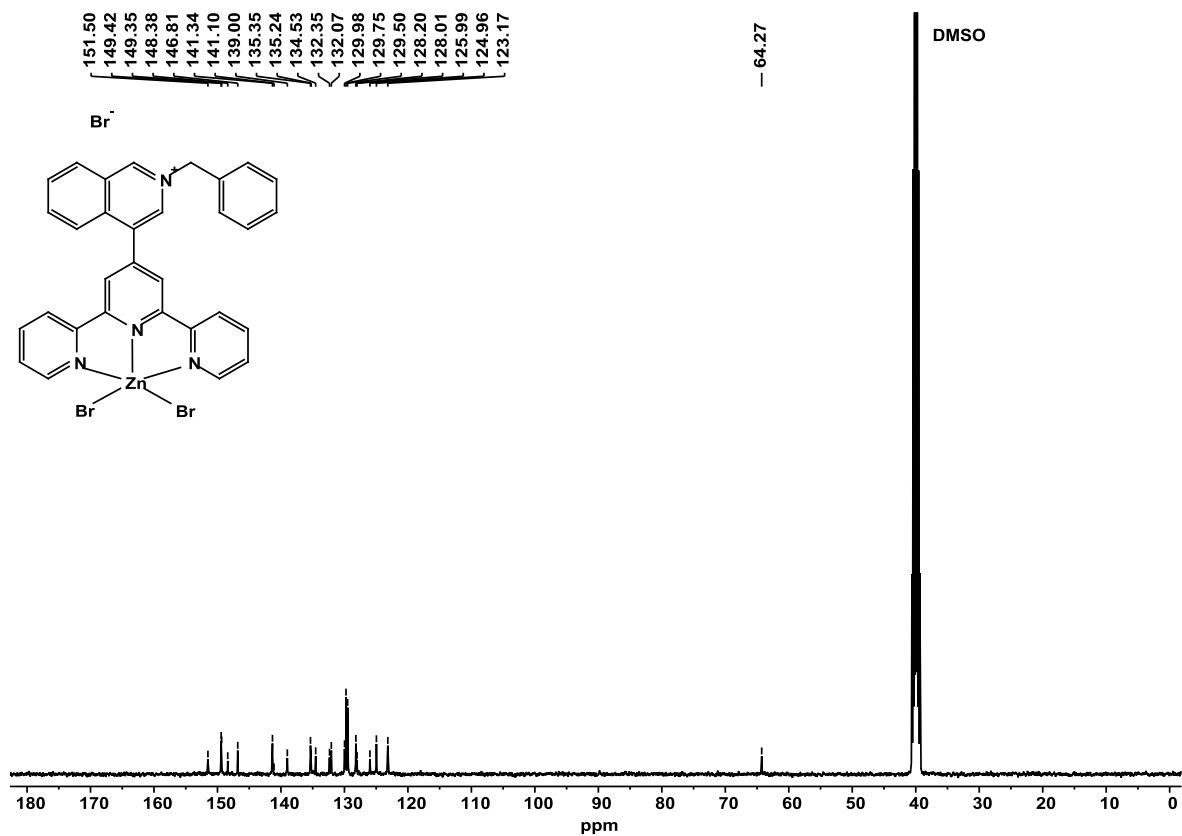


Fig. S36 ^{13}C NMR spectrum of **5.Zn** in $\text{DMSO-}d_6$.

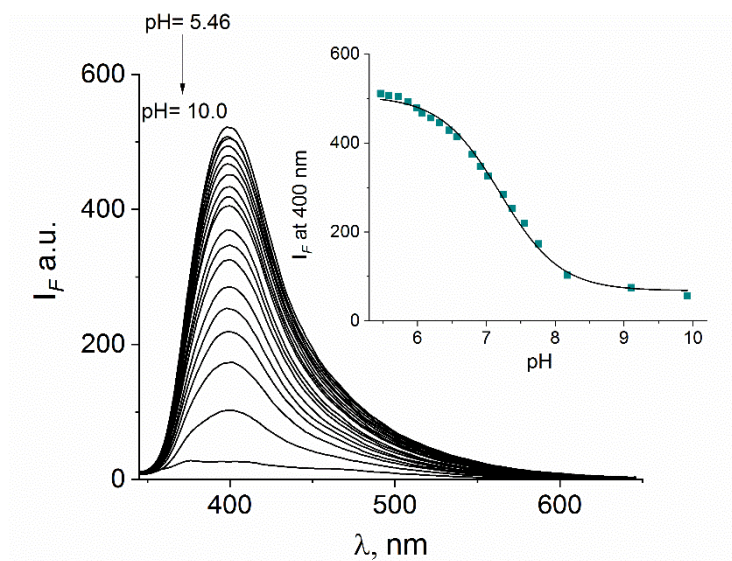


Fig. S37 Fluorescence pH-titration of buffered aqueous solution of **3.Zn** (20 μ M). The inset show pH-titration profile observed at emission maxima.

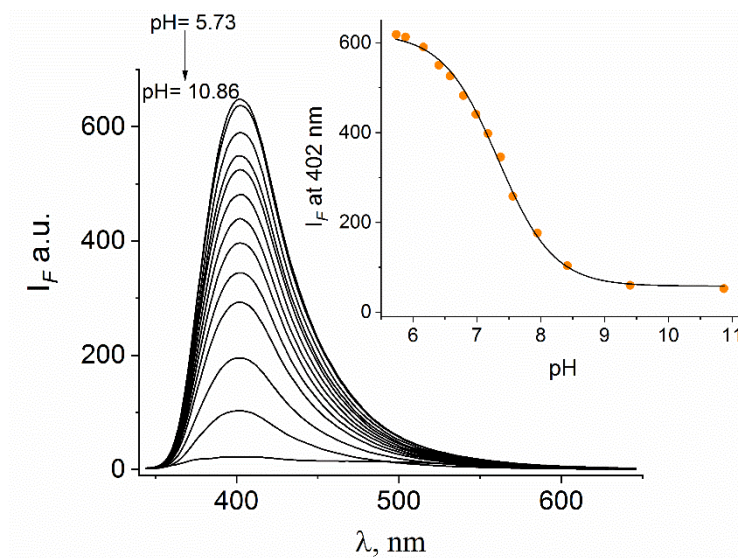


Fig. S38 Fluorescence pH-titration of buffered aqueous solution of **4.Zn** (20 μ M). The inset show pH-titration profile observed at emission maxima.

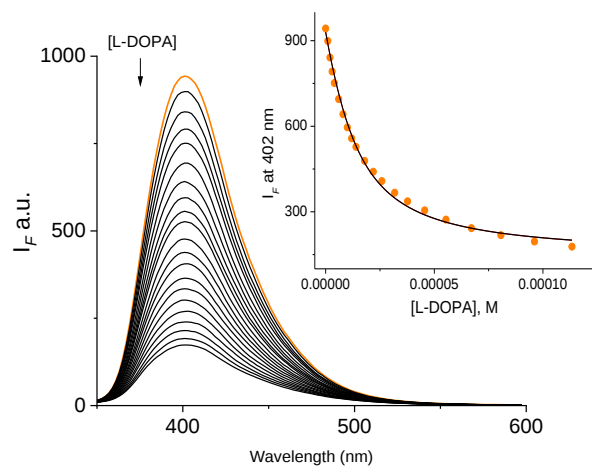


Fig. S39 Changes of emission spectra ($\lambda_{\text{ex}} = 330$ nm) of buffered aqueous solutions at pH 7.4 of **4.Zn** (10 μM) upon addition of increasing amounts of L-DOPA. The inset shows the profile at 402 nm. The solid line was obtained by fitting to Eq. (1).

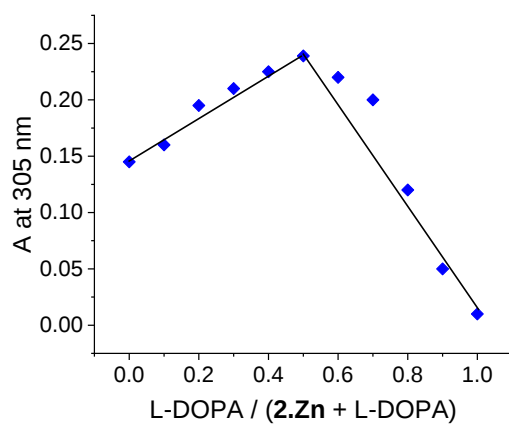


Fig. S40 Stoichiometric analysis of **2.Zn** by Job plot with L-DOPA at 305 nm

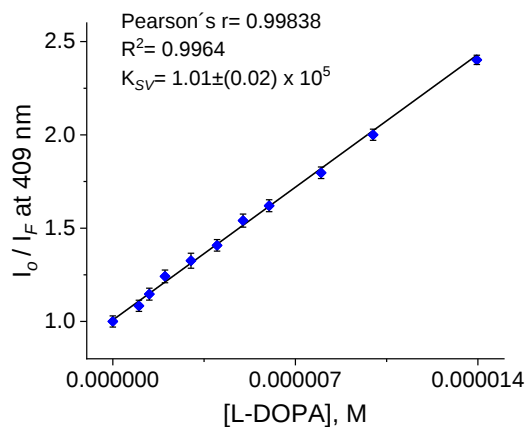


Fig. S41 Stern-Volmer plot of **2.Zn** upon addition of L-DOPA at 409 nm ($\lambda_{\text{ex}} = 330$ nm) at pH= 7.4.