

Supporting Information

**The Structural Manipulation of a Series of Ni<sub>4</sub> Defective  
Dicubanes: Synthesis, X-ray Structures, Magnetic, and  
Computational Analyses.**

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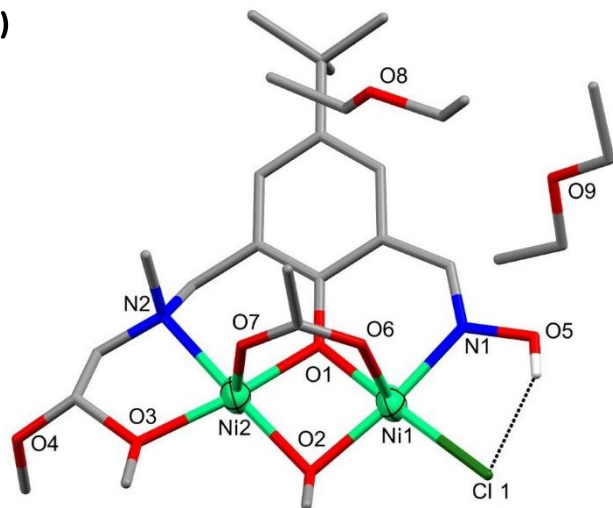
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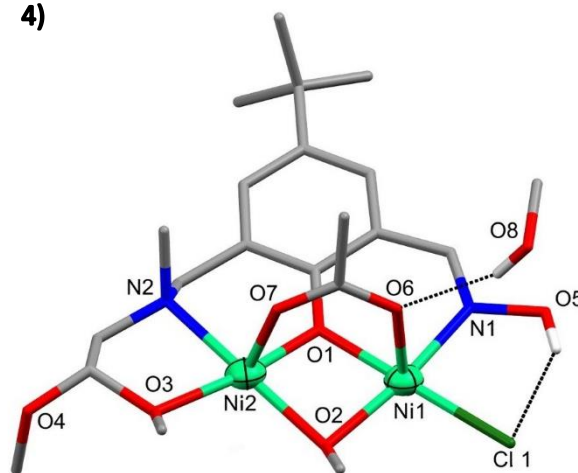
## Content

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1)



4)



**Figure S1:** The asymmetric units of complexes **1** (left) and **4** (right) showing the free solvent molecules. Non-interacting hydrogen atoms removed for clarity. Colour code: Ni = light green, N = blue, O = red, Cl = dark green, C = grey, and H = white. Hydrogen bonds represented as black dotted lines.

**Table S1:** Non-averaged bond lengths (Å) and bond angles (°) of **2** and **3**.

2					
Ni1-Cl1	2.412(2)	Ni2-O2	2.080(5)	Ni3-O11	2.094(6)
Ni1-O2	2.091(5)	Ni2-O5	2.174(6)	Ni3-N3	2.046(7)
Ni1-O3	2.055(6)	Ni2-Cl3	2.377(3)	Ni4-Cl2	2.479(3)
Ni1-O4	2.015(5)	Ni2-N2	2.086(7)	Ni4-Cl4	2.371(3)
Ni1-O12	2.085(6)	Ni3-Cl2	2.401(2)	Ni4-O3	2.080(5)
Ni1-N1	2.054(7)	Ni3-O1	2.000(5)	Ni4-O4	2.037(6)
Ni2-Cl1	2.473(3)	Ni3-O2	2.073(6)	Ni4-O8	2.182(6)
Ni2-O1	2.020(6)	Ni3-O3	2.100(5)	Ni4-N4	2.104(7)
Ni1-O2-Ni3	95.51(7)	Ni1-O3-Ni3	95.77(7)	Ni1-Cl1-Ni2	86.68(8)
Ni3-Cl2-Ni4	86.91(8)	Ni2-O1-Ni3	100.75(6)	Ni1-O4-Ni4	99.96(6)
Ni1-O2-Ni2	106.95(5)	Ni3-O3-Ni4	106.84(5)	Ni2-O2-Ni3	96.38(7)
Ni1-O3-Ni4	97.25(7)				
3					
Ni1-Cl1	2.283(6)	Ni2-O5	2.173(3)	Ni3-N3	2.014(4)
Ni1-F1	2.092(7)	Ni2-Cl3	2.369(2)	Ni4-Cl2	2.464(6)
Ni1-O2	2.058(3)	Ni2-F1	2.017(6)	Ni4-Cl4	2.351(2)
Ni1-O3	2.069(3)	Ni2-N2	2.089(3)	Ni4-F2	2.069(9)
Ni1-O4	2.005(3)	Ni3-Cl2	2.309(7)	Ni4-O3	2.092(3)
Ni1-O12	2.088(3)	Ni3-F2	2.122(10)	Ni4-O4	2.0214(3)
Ni1-N1	2.013(4)	Ni3-O1	2.019(3)	Ni4-O8	2.175(3)
Ni2-Cl1	2.506(5)	Ni3-O2	2.057(3)	Ni4-N4	2.090(4)
Ni2-O1	2.036(3)	Ni3-O3	2.051(3)	Ni1-O2-Ni3	95.02(11)
Ni2-O2	2.090(3)	Ni3-O11	2.117(3)	Ni1-O3-Ni3	94.90(11)
Ni1-Cl1-Ni2	84.29(12)	Ni3-Cl2-Ni4	85.35(16)	Ni1-F1-Ni2	103.06(5)
Ni3-F2-Ni4	101.13(5)	Ni2-O1-Ni3	99.99(13)	Ni1-O4-Ni4	99.29(12)
Ni1-O2-Ni2	101.73(10)	Ni3-O3-Ni4	102.78(11)	Ni2-O2-Ni3	96.34(11)
Ni1-O3-Ni4	95.73(12)				

**Table S2:** Crystal data and structural refinement details for **1-4**.

	<b>1</b>	<b>2</b>	<b>3</b>	<b>4</b>
<b>Formula</b>	C <sub>52</sub> H <sub>96</sub> N <sub>4</sub> O <sub>17</sub> Cl <sub>2</sub> Ni <sub>4</sub>	C <sub>42</sub> H <sub>78</sub> N <sub>4</sub> O <sub>13</sub> Cl <sub>4</sub> Ni <sub>4</sub>	C <sub>42</sub> H <sub>78</sub> N <sub>4</sub> O <sub>13</sub> F <sub>1.2</sub> Cl <sub>2.8</sub> Ni <sub>4</sub>	C <sub>42</sub> H <sub>74</sub> N <sub>4</sub> O <sub>16</sub> Cl <sub>2</sub> Ni <sub>4</sub>
<b>M<sub>r</sub> (g mol<sup>-1</sup>)</b>	1355.06	1223.72	1203.98	1196.74
<b>T (K)</b>	120	123	183	123
<b>Crystal System</b>	Monoclinic	Monoclinic	Monoclinic	Monoclinic
<b>Space Group</b>	P2 <sub>1</sub> /c	P2 <sub>1</sub> /c	P2 <sub>1</sub> /c	P2 <sub>1</sub> /c
<b>a (Å)</b>	13.33(6)	10.16(11)	9.56(8)	10.94(8)
<b>b (Å)</b>	13.86(7)	20.49(2)	21.05(14)	24.80(6)
<b>c (Å)</b>	17.33(12)	25.79(3)	26.07(2)	10.14(3)
<b>α (°)</b>	90	90	90	90
<b>β (°)</b>	90.34	91.18(6)	90.15(6)	109.74(8)
<b>γ (°)</b>	90	90	90	90
<b>V (Å<sup>3</sup>)</b>	3205.7(3)	5371.6(11)	5253.1(7)	2593.7(2)
<b>Z</b>	2	1	1	2
<b>ρ<sub>calc</sub> (g cm<sup>-3</sup>)</b>	1.404	1.513	1.522	1.532
<b>μ (mm<sup>-1</sup>)</b>	2.628	3.907	3.471	3.154
<b>F (000)</b>	1436.0	2568.0	2530.0	1256.0
<b>2θ Range for Data Collection/°</b>	12.042 to 140.134	5.508 to 140.158	9.06 to 130.178	7.126 to 144.138
<b>Index Ranges</b>	-12 ≤ h ≤ 16, -15 ≤ k ≤ 16, -20 ≤ l ≤ 21	-11 ≤ h ≤ 12, -19 ≤ k ≤ 23, -31 ≤ l ≤ 30	-10 ≤ h ≤ 11, -23 ≤ k ≤ 24, -30 ≤ l ≤ 30	-13 ≤ h ≤ 13, -30 ≤ k ≤ 30, -10 ≤ l ≤ 12
<b>Reflections Collected</b>	45862	42778	37702	29026
<b>Independent Reflections</b>	5919 [R <sub>int</sub> = 0.0852, R <sub>sigma</sub> = 0.0431]	9809 [R <sub>int</sub> = 0.2394, R <sub>sigma</sub> = 0.4683]	8926 [R <sub>int</sub> = 0.0727, R <sub>sigma</sub> = 0.1558]	5013 [R <sub>int</sub> = 0.1022, R <sub>sigma</sub> = 0.0845]
<b>Data/Restraints/Parameters</b>	5919/51/386	9809/0/635	8926/3/646	5013/0/318
<b>Goodness-of-Fit on F<sup>2</sup></b>	1.044	0.863	1.151	1.125
<b>Final R Indexes [<i>I</i> &gt; 2σ(<i>I</i>)]</b>	R <sub>1</sub> = 0.0580, wR <sub>2</sub> = 0.1593	R <sub>1</sub> = 0.0941, wR <sub>2</sub> = 0.1952	R <sub>1</sub> = 0.0581, wR <sub>2</sub> = 0.1645	R <sub>1</sub> = 0.0677, wR <sub>2</sub> = 0.1775
<b>Final R Indexes [All Data]</b>	R <sub>1</sub> = 0.0598, wR <sub>2</sub> = 0.1617	R <sub>1</sub> = 0.2708, wR <sub>2</sub> = 0.2526	R <sub>1</sub> = 0.1109, wR <sub>2</sub> = 0.1904	R <sub>1</sub> = 0.0853, wR <sub>2</sub> = 0.2100
<b>Largest Diff. Peak/Hole (e<sup>-</sup>/Å<sup>3</sup>)</b>	0.88/-0.68	0.86/-0.96	1.02/-0.60	1.06/-1.09

**Table S3:** Ni-X-Ni angles ( $^{\circ}$ ) mediating the exchange interactions  $J_1$ ,  $J_2$ , and  $J_3$  in compounds **1-4**.

	Ni- $\mu_3$ O-Ni (Ni- $\mu_2$ O-Ni) / $^{\circ}$ $J_1$	Ni- $\mu$ Cl-Ni (Ni- $\mu_3$ O-Ni) / $^{\circ}$ $J_2$	Ni- $\mu_3$ O-Ni (Ni- $\mu_3$ O-Ni) / $^{\circ}$ $J_3$
<b>1</b>	92.1 (95.4)	86.3 (111.3)	98.8 (98.8)
<b>2</b>	97.2 (99.9)	86.7 (107.0)	95.5 (95.8)
<b>3</b>	96.3 (99.3)	85.4* (101.8)	95.0 (94.9)
<b>4</b>	91.1 (94.8)	85.6 (107.8)	98.2 (98.2)

\* Disordered Ni-F-Ni angle, 108.1 $^{\circ}$ **Table S4.** Spin configurations for complexes **1-4** used for estimating the magnetic exchange parameters.

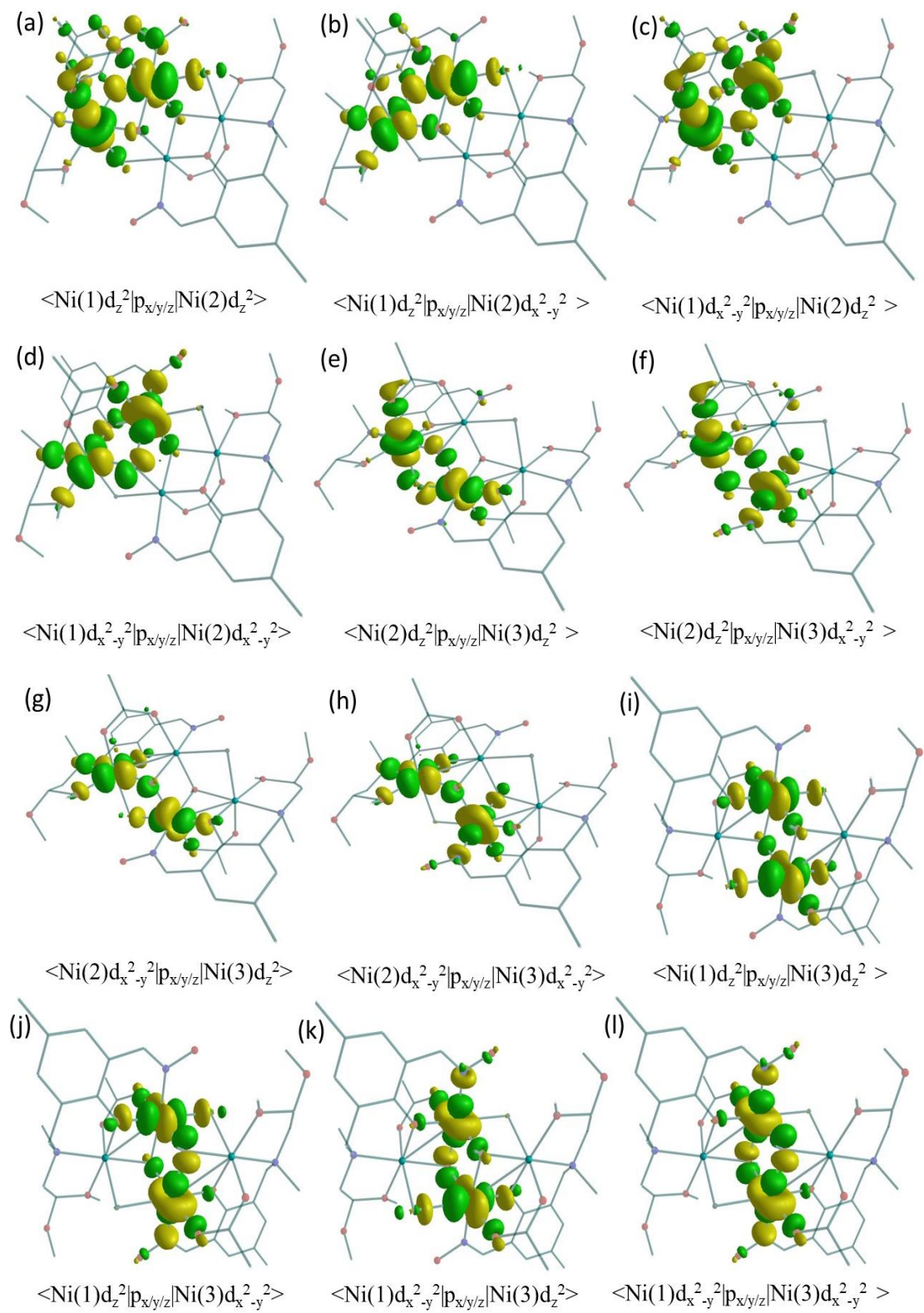
Serial Number	$S$	Spin configurations			
		Ni1	Ni2	Ni3	Ni4
1	4	$\alpha$	$\alpha$	$\alpha$	$\alpha$
2	2	$\beta$	$\alpha$	$\alpha$	$\alpha$
3	2	$\alpha$	$\beta$	$\alpha$	$\alpha$
4	2	$\alpha$	$\alpha$	$\beta$	$\alpha$
5	2	$\alpha$	$\alpha$	$\alpha$	$\beta$
6	0	$\beta$	$\beta$	$\alpha$	$\alpha$
7	0	$\alpha$	$\beta$	$\beta$	$\alpha$
8	0	$\beta$	$\alpha$	$\beta$	$\alpha$

**Table S5.** Pertinent structural parameters for **1-4**.

Complex	Exchange	avg Ni- $\mu_{2/3}$ O/F/Cl-Ni angle ( $^{\circ}$ )	avg Ni- $\mu_{2/3}$ O/F/Cl distance ( $\text{\AA}$ )
<b>1</b>	$J_1$	93.7	2.019
	$J_2$	98.8	2.320
	$J_3$	98.8	2.077
<b>2</b>	$J_1$	98.6	2.043
	$J_2$	96.8	2.264
	$J_3$	95.7	2.080
<b>3</b>	$J_1$	97.8	2.051
	$J_2$	102.4	2.064
	$J_3$	95.0	2.059
<b>4</b>	$J_1$	93.0	2.037
	$J_2$	96.7	2.256
	$J_3$	98.2	2.065

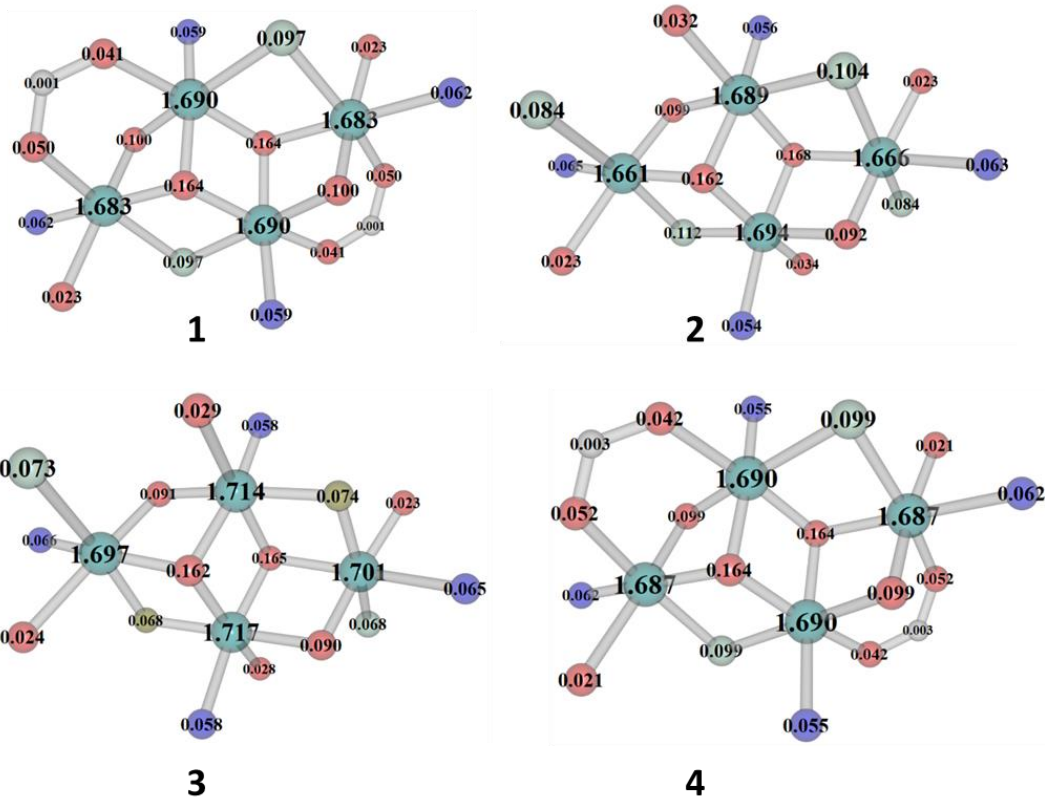
**Table S6.** DFT computed overlap integrals for **1-4** on dimeric model complexes. The values highlighted in red suggest large magnetic orbital overlap and the values highlighted in yellow highlighted suggest moderate magnetic orbitals overlap. Both contribute to an antiferromagnetic interaction. The non-highlighted values represent orthogonal/weak magnetic orbital overlap which contribute to a ferromagnetic interaction.

Complex	$J_1$			$J_2$			$J_3$		
	Ni(1) $\alpha$ /Ni(2) $\beta$	$d_z^2$	$d_{x^2-y^2}$	Ni(2) $\alpha$ /Ni(3) $\beta$	$d_z^2$	$d_{x^2-y^2}$	Ni(1) $\alpha$ /Ni(3) $\beta$	$d_z^2$	$d_{x^2-y^2}$
<b>1</b>	$d_z^2$	0.041	0.015	$d_z^2$	0.022	0.034	$d_z^2$	0.005	0.022
	$d_{x^2-y^2}$	0.005	0.006	$d_{x^2-y^2}$	0.028	0.036	$d_{x^2-y^2}$	0.022	0.043
	Ni(1) $\alpha$ /Ni(2) $\beta$	$d_z^2$	$d_{x^2-y^2}$	Ni(2) $\alpha$ /Ni(3) $\beta$	$d_z^2$	$d_{x^2-y^2}$	Ni(1) $\alpha$ /Ni(3) $\beta$	$d_z^2$	$d_{x^2-y^2}$
<b>2</b>	$d_z^2$	0.011	0.028	$d_z^2$	0.001	0.022	$d_z^2$	0.007	0.009
	$d_{x^2-y^2}$	0.061	0.011	$d_{x^2-y^2}$	0.024	0.035	$d_{x^2-y^2}$	0.013	0.019
	Ni(1) $\alpha$ /Ni(2) $\beta$	$d_z^2$	$d_{x^2-y^2}$	Ni(2) $\alpha$ /Ni(3) $\beta$	$d_z^2$	$d_{x^2-y^2}$	Ni(1) $\alpha$ /Ni(3) $\beta$	$d_z^2$	$d_{x^2-y^2}$
<b>3</b>	$d_z^2$	0.024	0.016	$d_z^2$	0.036	0.061	$d_z^2$	0.001	0.016
	$d_{x^2-y^2}$	0.012	0.050	$d_{x^2-y^2}$	0.006	0.006	$d_{x^2-y^2}$	0.012	0.018
	Ni(1) $\alpha$ /Ni(2) $\beta$	$d_z^2$	$d_{x^2-y^2}$	Ni(2) $\alpha$ /Ni(3) $\beta$	$d_z^2$	$d_{x^2-y^2}$	Ni(1) $\alpha$ /Ni(3) $\beta$	$d_z^2$	$d_{x^2-y^2}$
<b>4</b>	$d_z^2$	0.041	0.027	$d_z^2$	0.004	0.023	$d_z^2$	0.011	0.009
	$d_{x^2-y^2}$	0.004	0.004	$d_{x^2-y^2}$	0.019	0.055	$d_{x^2-y^2}$	0.009	0.044
	Ni(1) $\alpha$ /Ni(2) $\beta$	$d_z^2$	$d_{x^2-y^2}$	Ni(2) $\alpha$ /Ni(3) $\beta$	$d_z^2$	$d_{x^2-y^2}$	Ni(1) $\alpha$ /Ni(3) $\beta$	$d_z^2$	$d_{x^2-y^2}$



**Figure S2.** Representative figures showing SOMO(s)-SOMO(s) interactions.





**Figure S3.** DFT computed spin density values on selected atoms for **1-4**.