

Bis(pyridineylidene)isoindoline, an isoindoline chelate with a delocalized electronic structure

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General Information

Experimental

All reagents and starting materials were purchased from commercial vendors and used without further purification. 1,3-diiminoisoindoline (DII) was synthesized according to previously published procedures.¹ Deuterated solvents were purchased from Cambridge Isotope Laboratories and used as received.

NMR spectra were recorded on 300 MHz and 500 MHz spectrometers and chemical shifts were given in ppm relative to residual solvent resonances (¹H NMR and ¹³C NMR spectra). High-resolution mass spectrometry experiments were performed on a Bruker MicroTOF-III and MicroTOF-qIII instruments. Infrared spectra were collected on Thermo Scientific Nicolet iS5 that was equipped with an iD5 ATR. UV-visible spectra were recorded on a Shimadzu UV-2600 UV-Visible spectrophotometer.

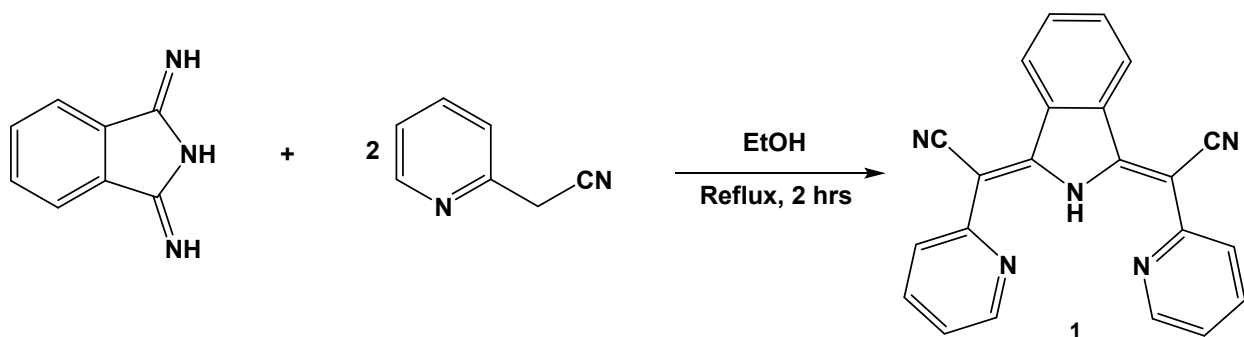
X-ray intensity data were measured on a Bruker CCD-based and PHOTON II CPAD-based diffractometer with dual Cu/Mo ImuS microfocus optics (Cu K α radiation, $\lambda = 1.54178 \text{ \AA}$, Mo K α radiation, $\lambda = 0.71073 \text{ \AA}$). Crystals were mounted on a cryoloop using Paratone oil and placed under a stream of nitrogen at 100 K (Oxford Cryosystems). The detector was placed at a distance of 5.00 cm from the crystal. The data were corrected for absorption with the SADABS program. The structures were refined using the Bruker SHELXTL Software Package (Version 6.1),² and were solved using direct methods until the final anisotropic full-matrix, least squares refinement of F² converged.

Electrochemistry measurements were conducted using a CHI 820D potentiostat in a standard three-electrode configuration. Platinum wire was used as the counter electrode. The working electrode used was a 2 mm diameter platinum disk. A nonaqueous Ag/Ag⁺ reference

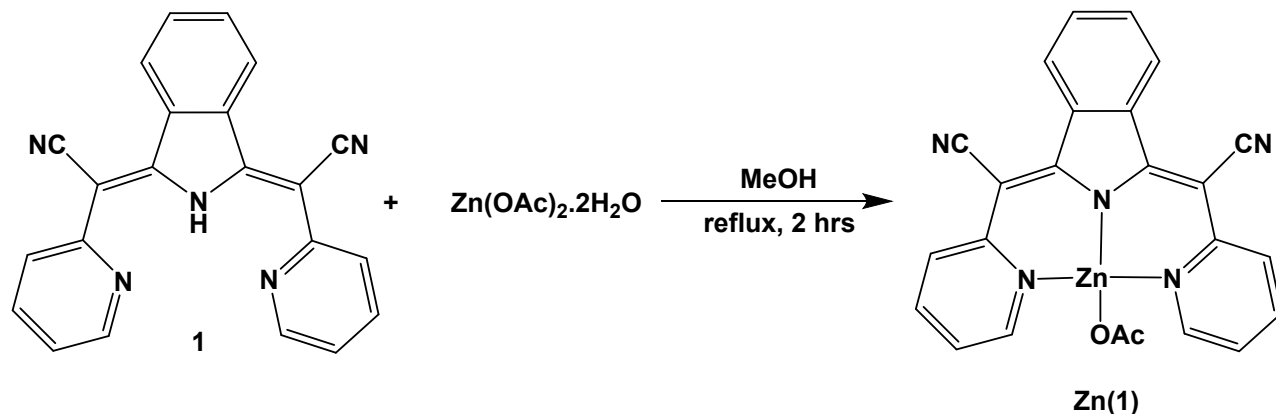
electrode was used by immersing silver wire in a degassed dimethylformamide (DMF) solution of 0.01 M AgNO_3 /0.1 M tetrabutylammonium hexafluorophosphate (TBAPF_6). All potentials were referenced to the ferrocene/ferrocenium couple. The concentration of analyte was 1.0 mM, and the supporting electrolyte was 0.1 M TBAPF_6 dissolved in DMF.

Computational Details

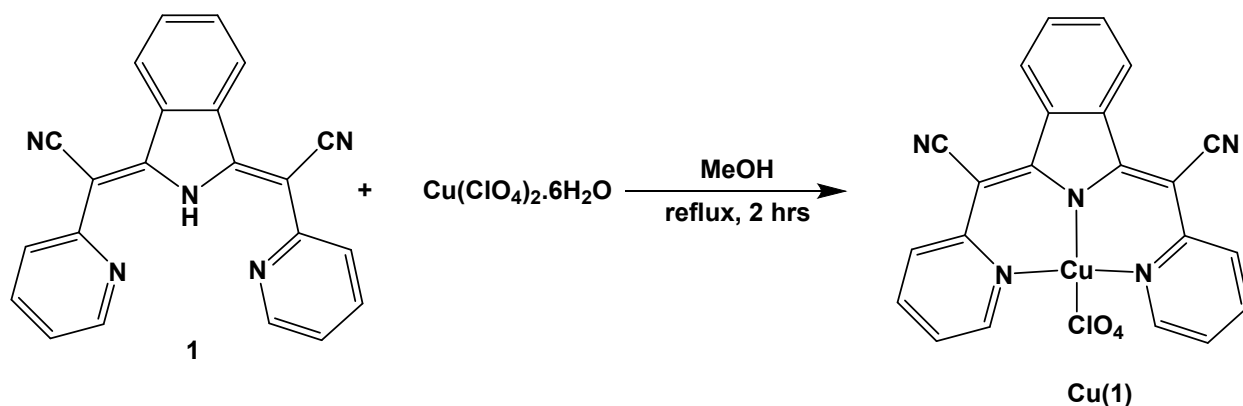
The starting geometries of compounds **BPI**, **1**, **Zn(1)**, **Cu(1)**, and **Ni(1)** were optimized using a B3LYP exchange–correlation functional.³ Energy minima in optimized geometry were confirmed by the frequency calculations (absence of the imaginary frequencies). The solvent effect was modeled using the polarized continuum model (PCM).⁴ In all calculations, DMF was used as the solvent (CHCl₃ for BPI). In PCM-TDDFT calculation, the first 30 states were calculated. All atoms were modeled using the 6-311G(d)⁵ basis set. Gaussian 09 software was used in all calculations.⁶ The QM Forge program was used for molecular orbital analysis in all cases.⁷



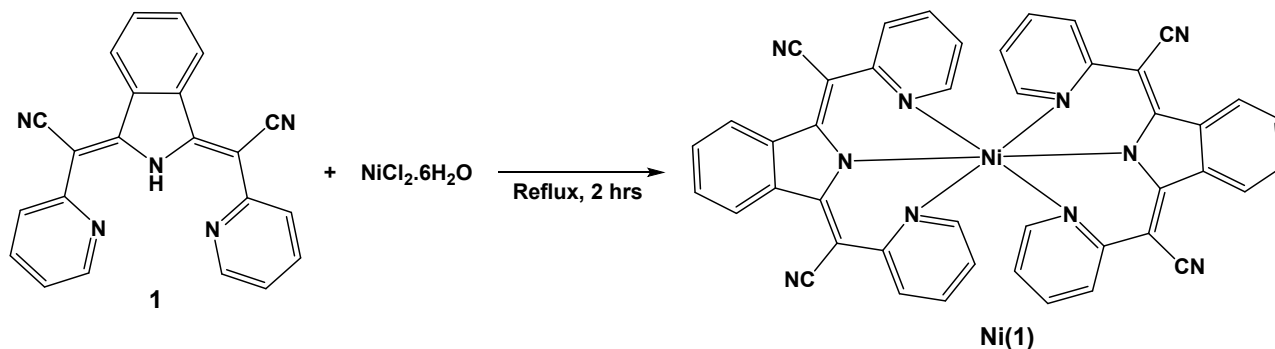
Synthesis of **1**: 1.00 g of DII (7 mmol), and 1.6 mL of 2-pyridylacetonitrile (14 mmol) were mixed in 30 mL of EtOH and refluxed for 2 hours. After cooling to room temperature, the precipitate was filtered, washed with cold EtOH, and dried in a vacuum oven, yielding an orange solid. Yield: 1.75 g (72%). MP: (decomp) 280°C. ESI MS (positive mode) $[C_{22}H_{13}N_5Na]^+$ calculated $[M+Na]^+$ 370.1063 m/z, found 370.1021 m/z. 1H NMR (300 MHz, $CDCl_3$): δ = 7.25 (m, 2H), 7.67 (m, 2H), 7.82 (dd, 4H), 8.73 (m, 2H) and 8.81 (dd, 2H). ^{13}C NMR $\{^1H\}$: due to limited solubility we were unable to get ^{13}C NMR. Crystals suitable for X-ray diffraction were grown in DMF.



Synthesis of **Zn(1)**: 50.80 mg of **1** (146.24 μ mol), and 32.80 mg (146.24 μ mol) of $Zn(OAc)_2 \cdot 2H_2O$ were dissolved in MeOH (20 mL) and refluxed for 2 hours. The resulting solution was cooled down to room temperature, filtered, washed with cold MeOH and dried in vacuum oven to get reddish brown powder of pure compound **2**. Yield: 68 mg (87%). ESI MS (positive mode) $[C_{24}H_{15}N_5O_2ZnNa]^+$ calculated $[M+Na]^+$ 492.0420 m/z, found 492.0421 m/z. 1H NMR (300 MHz, $CDCl_3$): δ = 1.99 (s, 3H), 7.35 (m, 2H), 7.65 (m, 2H), 7.91 (m, 2H), 8.13 (m, 2H), 8.75 (m, 2H), 8.91 (m, 2H). ^{13}C NMR $\{^1H\}$: due to limited solubility we were unable to get ^{13}C NMR. Crystals suitable for X-ray diffraction were grown in chloroform.



Synthesis of **Cu(1)**: 50 mg of **1** (143.93 μmol) and 53.33 mg (143.93 μmol) of $\text{Cu}(\text{ClO}_4)_2 \cdot 6\text{H}_2\text{O}$ were dissolved in MeOH (20 mL) and refluxed for 2 hours. The resulting solution was cooled down to room temperature, filtered, washed with cold MeOH and dried in a vacuum oven to get brown powder of pure compound **Cu(1)**. Yield: 41 mg (56%). ESI MS (negative mode) $[\text{C}_{22}\text{H}_{12}\text{Cl}_2\text{CuN}_5\text{O}_8]^-$ calculated $[\text{M}]^-$ 606.936 m/z, found 606.935 m/z. ^1H NMR: due to paramagnetic nature we were unable to get ^1H NMR. Crystals suitable for X-ray diffraction were grown in DMF.



Synthesis of **Ni(1)**: Two equivalents of **1** (75 mg, 215.90 μmol) and one equivalent of $\text{NiCl}_2 \cdot 6\text{H}_2\text{O}$ (107.95 μmol , 25.66 mg) were dissolved in DMF (20 mL) and refluxed for 2 hours. The resulting solution was cooled down to room temperature left to evaporate to form dark red crystals of compound **Ni(1)**. Yield: 64 mg (79%). ESI MS (negative mode) $[\text{C}_{44}\text{H}_{23}\text{N}_{10}\text{NiNa}]^-$ calculated $[\text{M}+\text{Na}-\text{H}]^-$ 772.1364 m/z, found 772.1430 m/z. ^1H NMR: due to paramagnetic nature we were unable to get ^1H NMR. Crystals suitable for X-ray diffraction were grown in DMF.

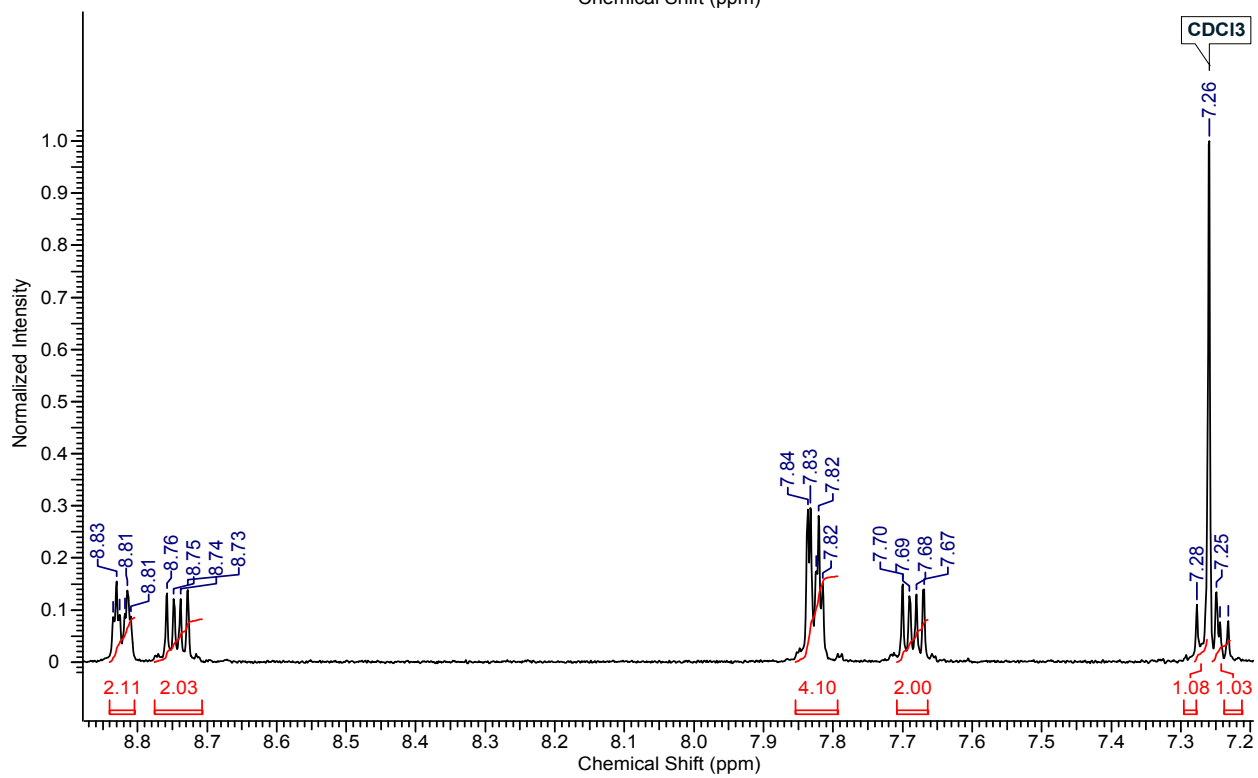
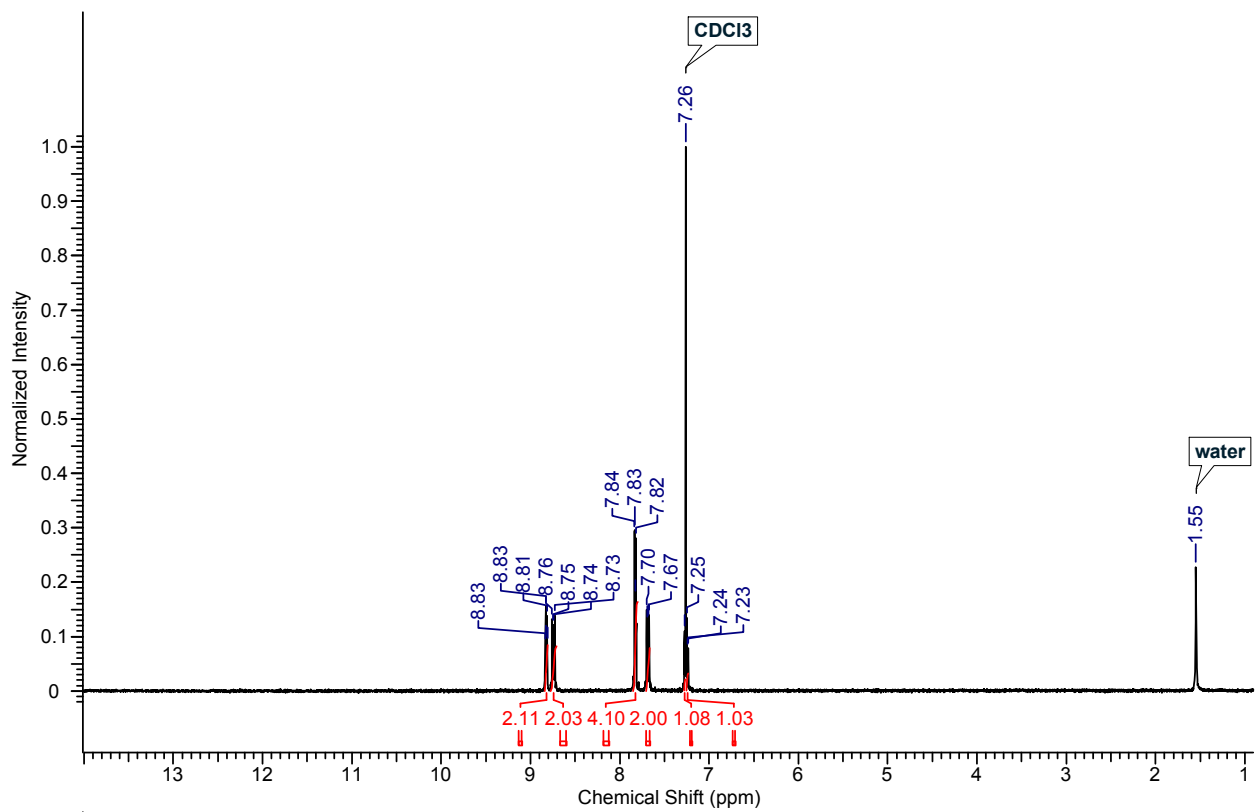


Figure S1: ¹H NMR (300 MHz) of **1** in CDCl₃

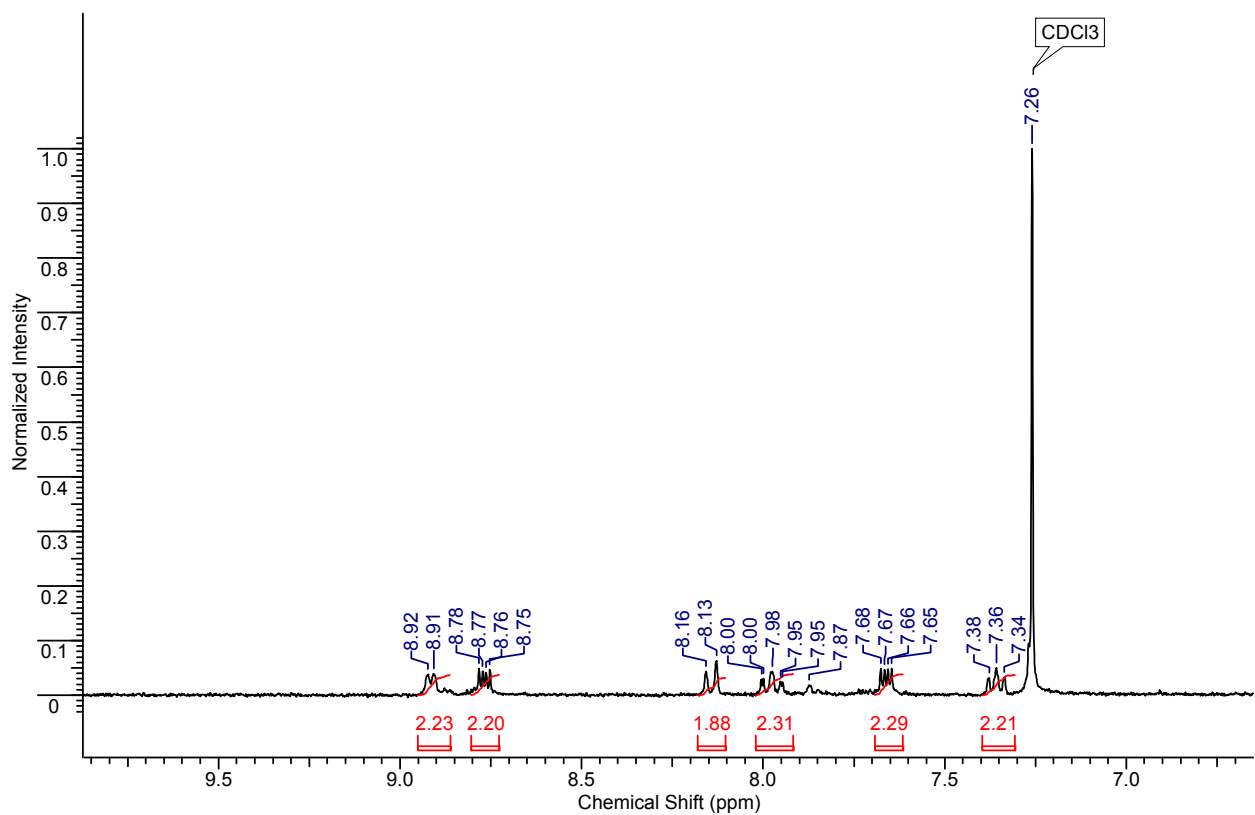
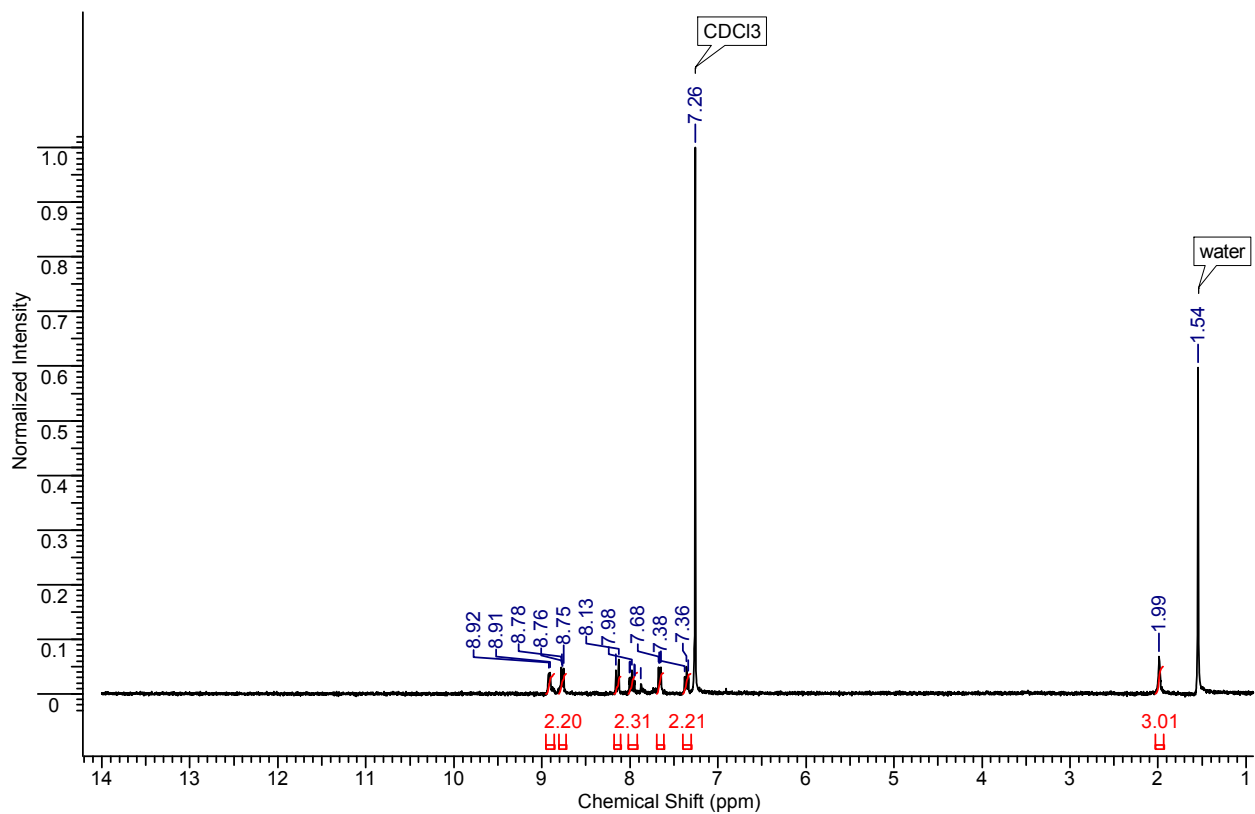


Figure S2: ¹H NMR (300 MHz) of Zn(1) in CDCl₃

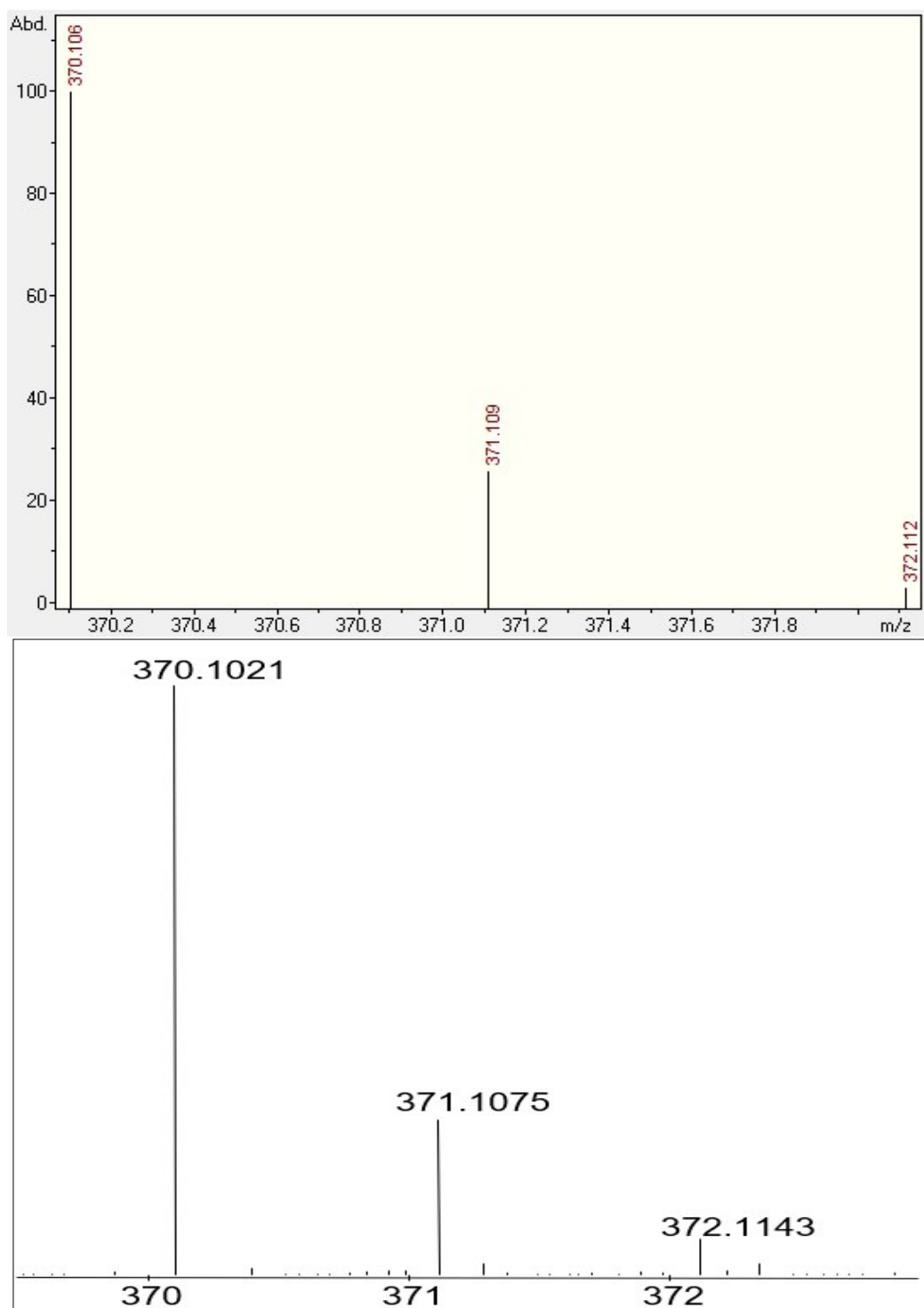


Figure S3: ESI mass spectra of **1**. Top: calculated spectrum. Bottom: experimental spectrum.

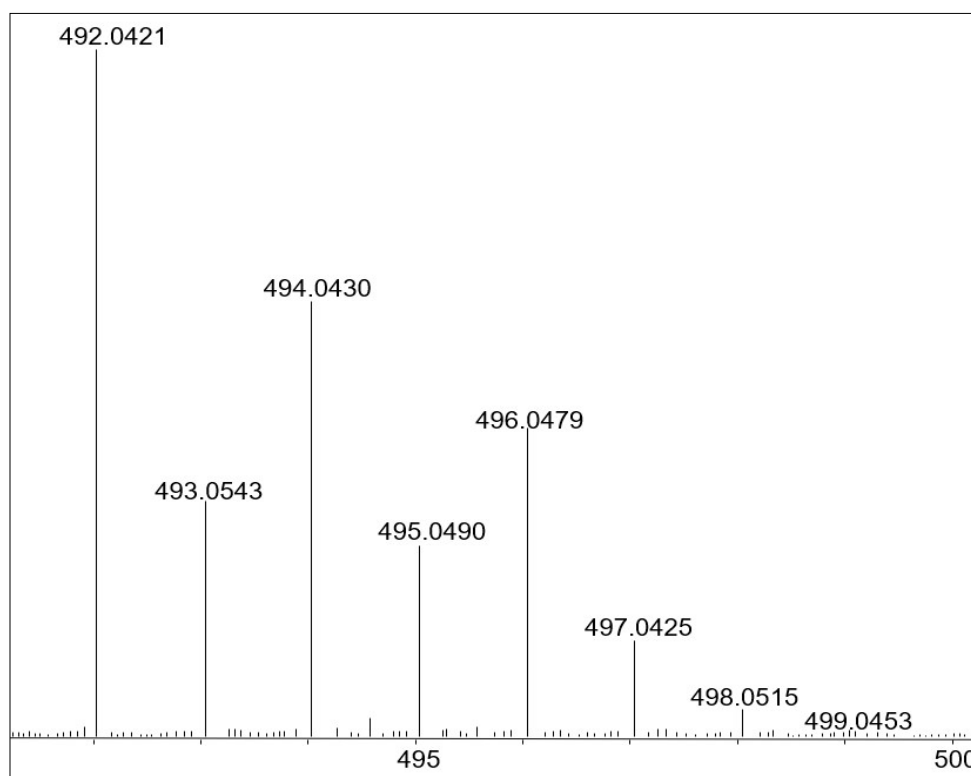
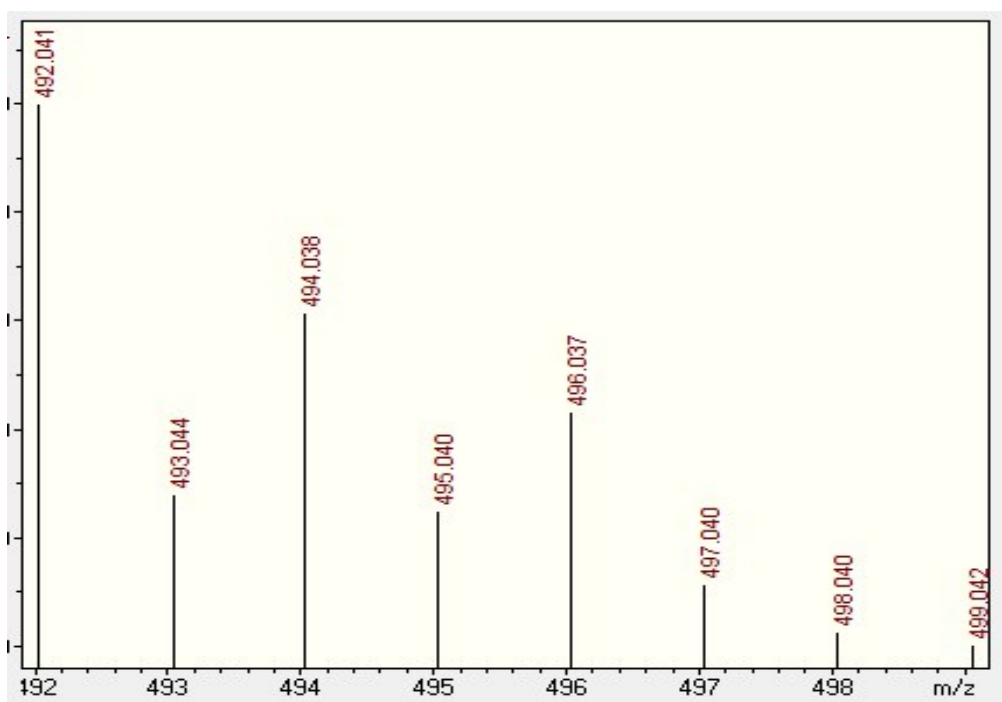


Figure S4: ESI mass spectra of **Zn(1)**. Top: calculated spectrum. Bottom: experimental spectrum.

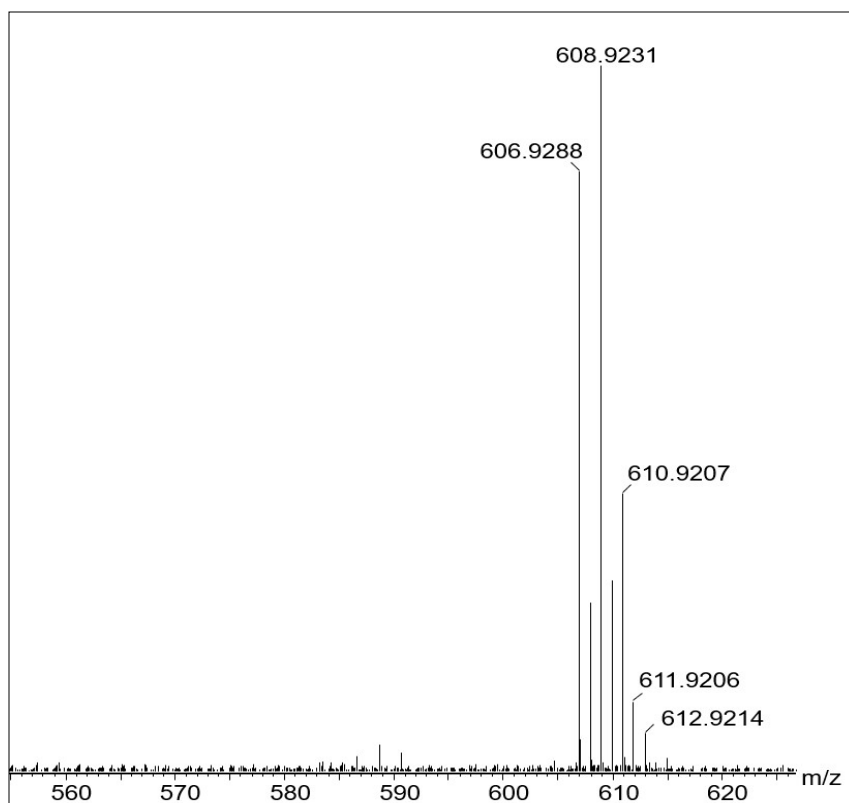
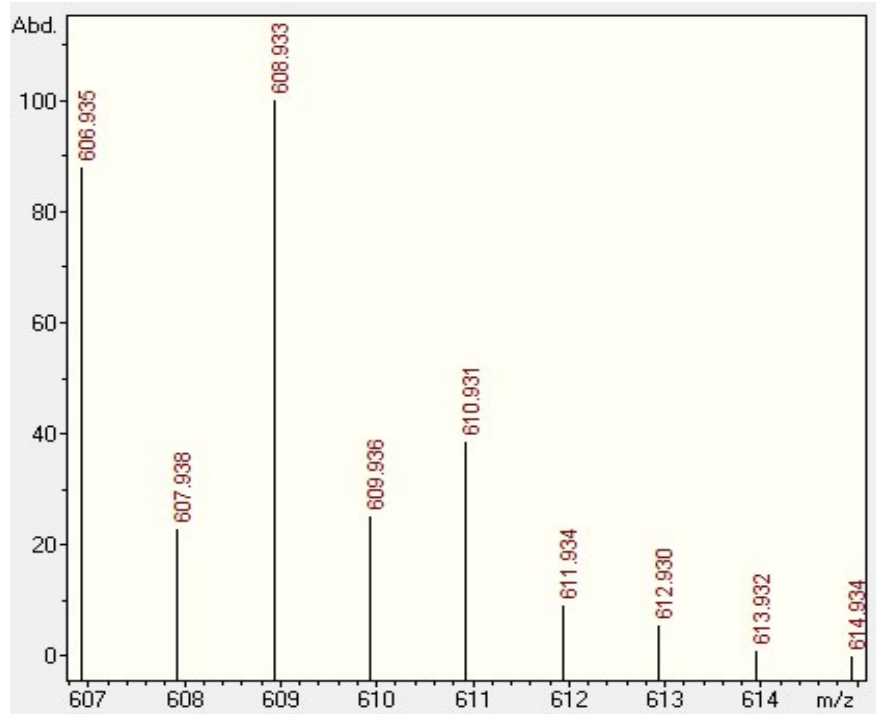


Figure S5: ESI mass spectra of **Cu(1)**. Top: calculated spectrum. Bottom: experimental spectrum.

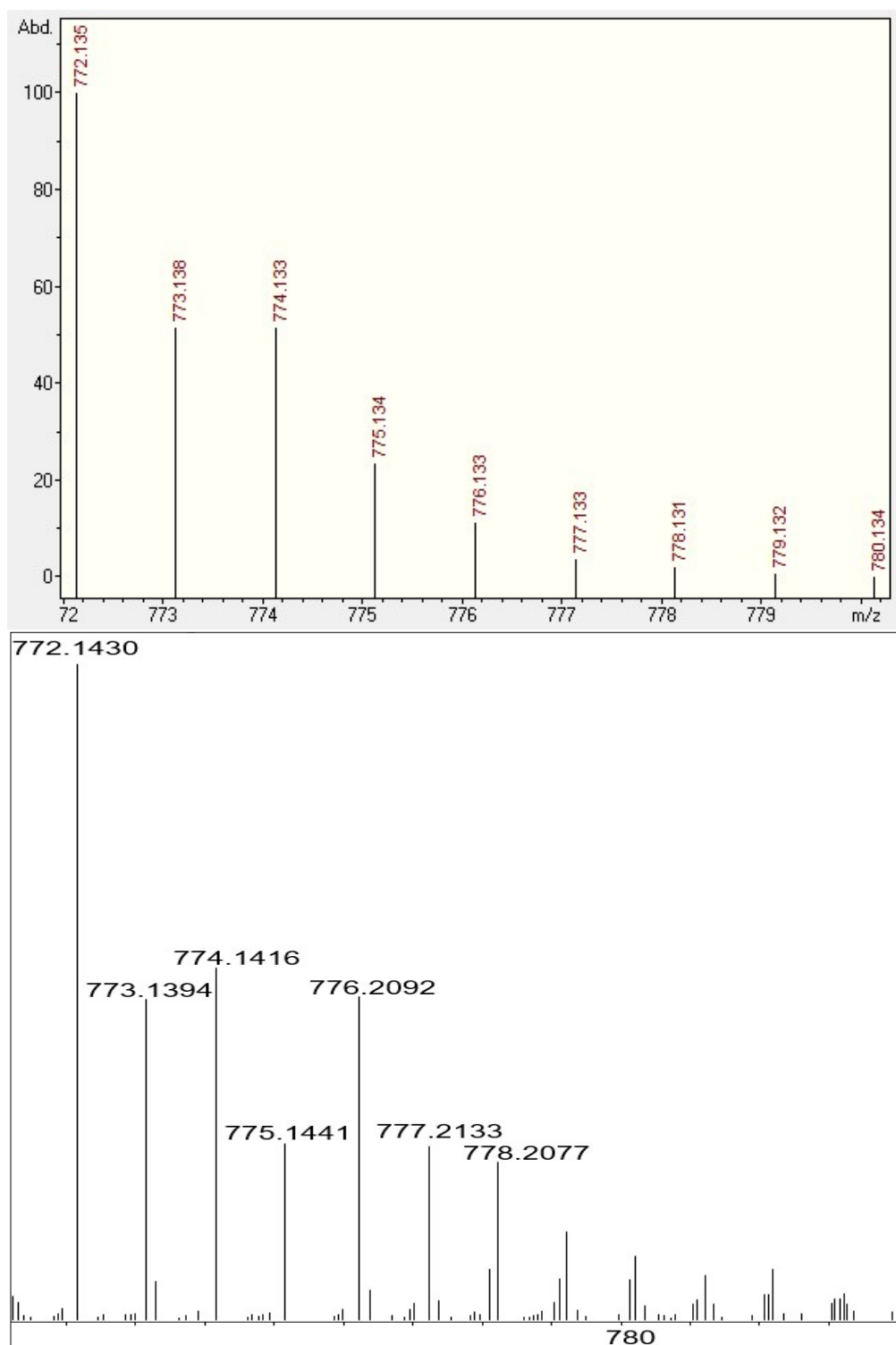


Figure S6: ESI mass spectra of Ni(1). Top: calculated spectrum. Bottom: experimental spectrum.

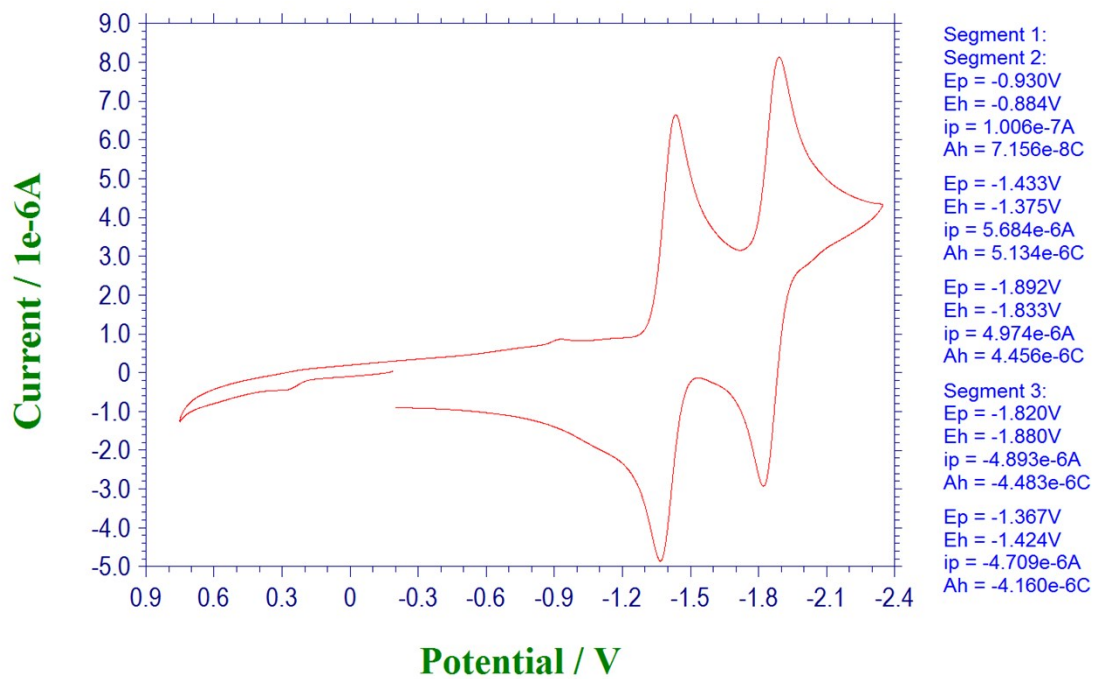


Figure S7: Cyclic voltammogram of compound **1** in DMF/0.1 TBAPF₆ (scan toward positive potentials).

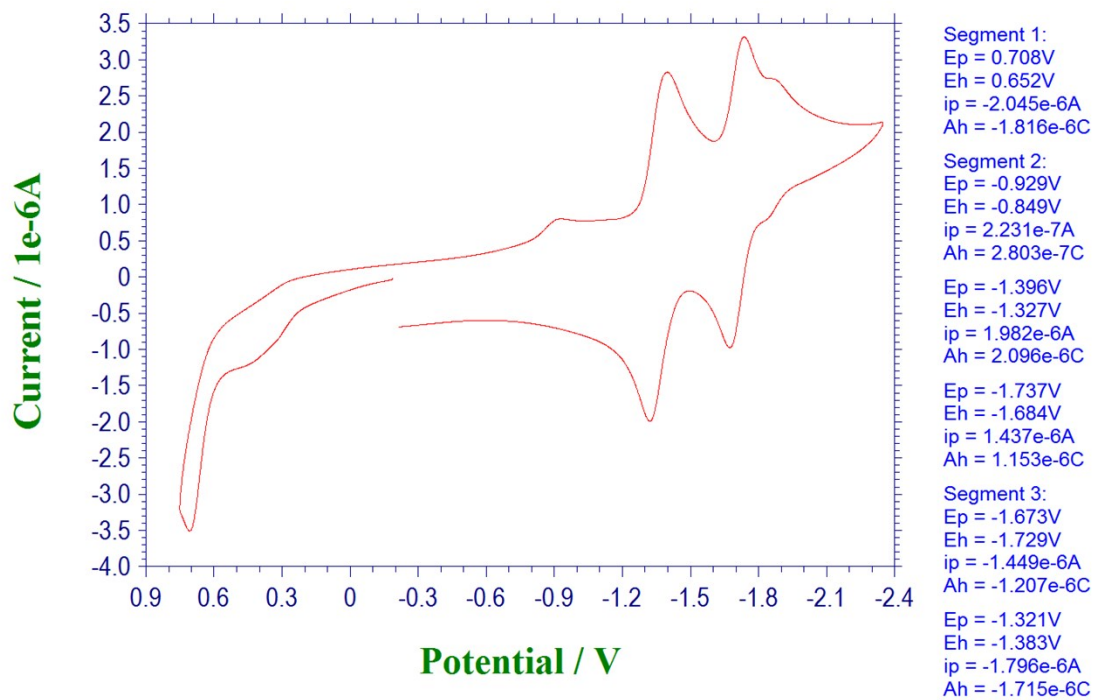


Figure S8: Cyclic voltammogram of compound **Zn(1)** in DMF/0.1 TBAPF₆ (scan toward positive potentials).

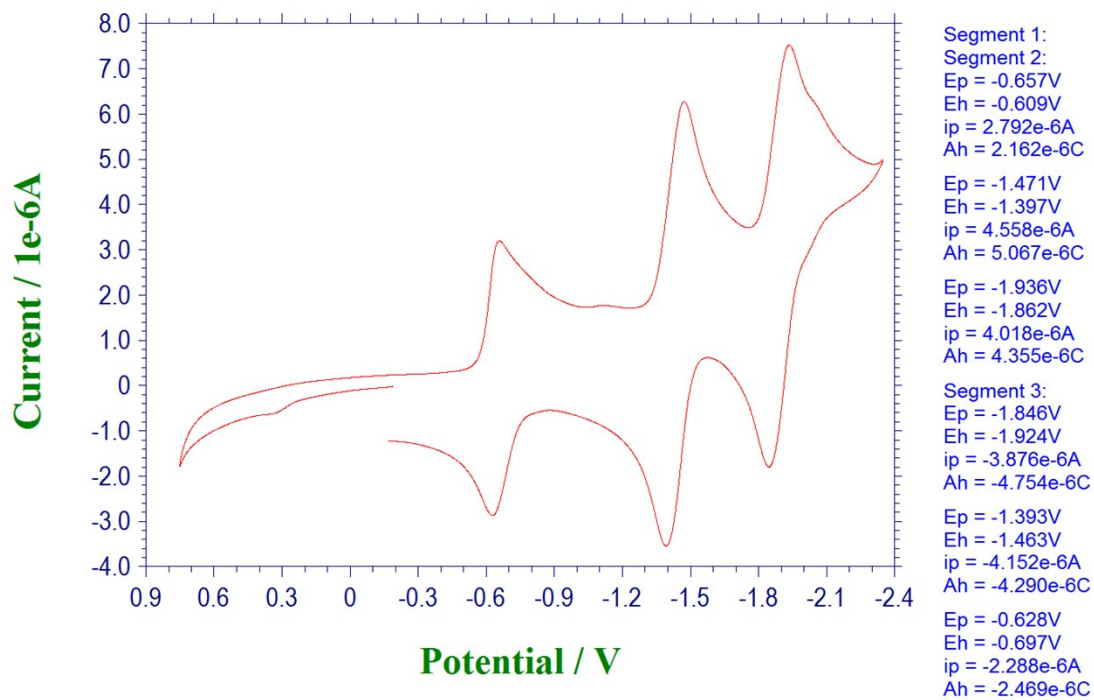


Figure S9: Cyclic voltammogram of compound **Cu(1)** in DMF/0.1 TBAPF₆. (scan toward positive potentials).

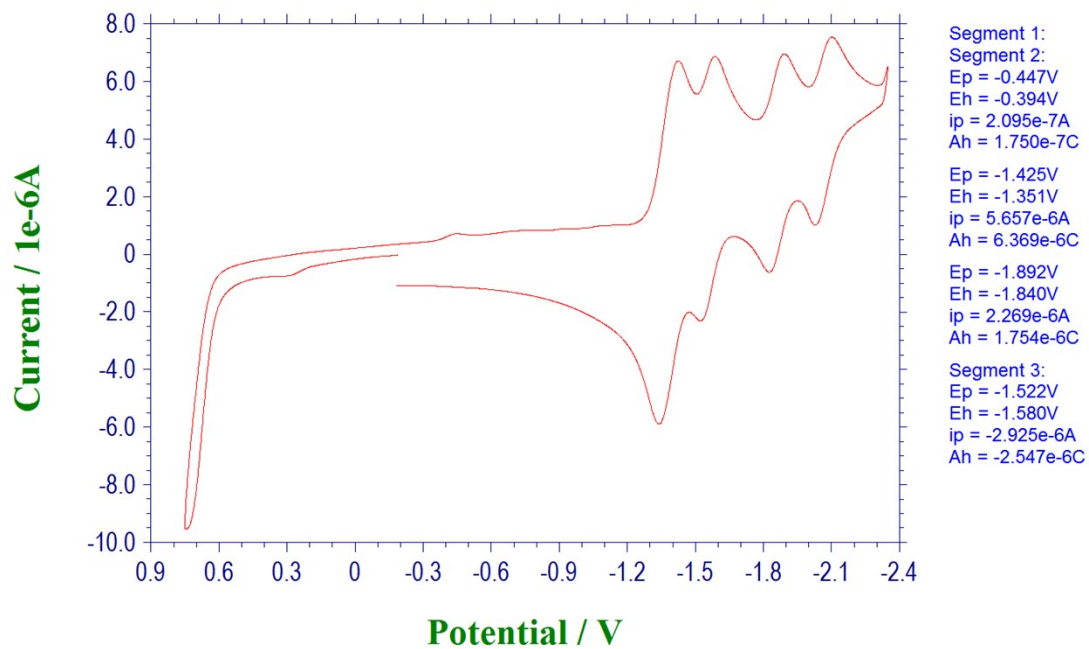


Figure S10: Cyclic voltammogram of compound **Ni(1)** in DMF/0.1 TBAPF₆ (scan toward positive potentials).

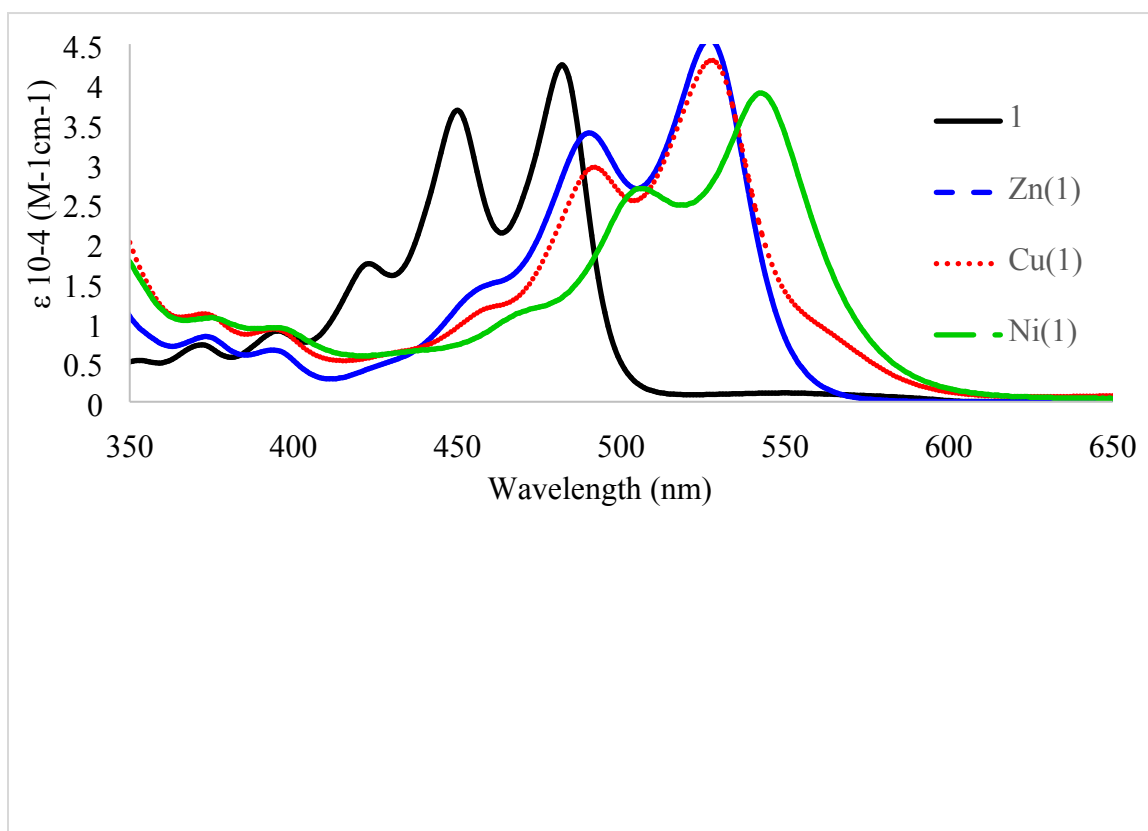


Figure S11: The UV-visible spectra of BPYI and its metal complexes in DMF solution.

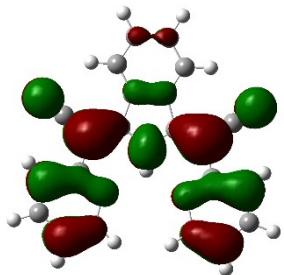
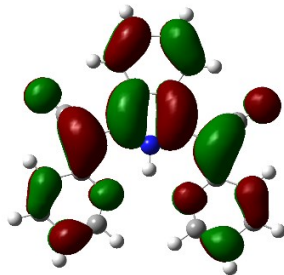
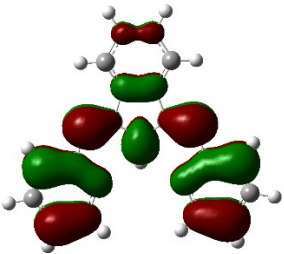
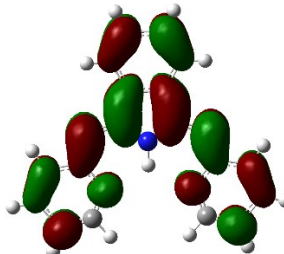
Compound		
	HOMO	LUMO
BPYI (1)		
BPI		

Figure S12. DFT-predicted frontier orbitals for BPYI (1) and BPI.

Compound	

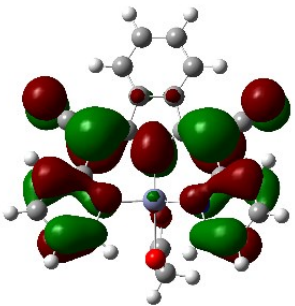
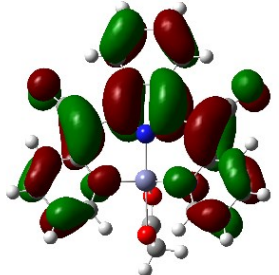
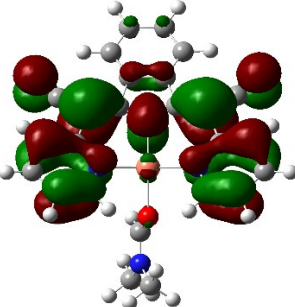
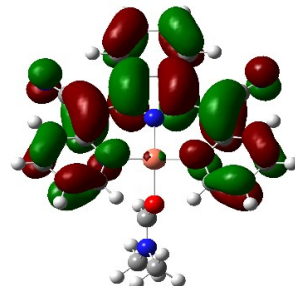
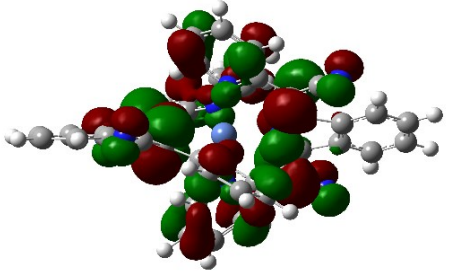
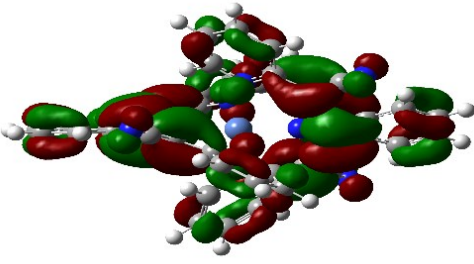
	HOMO	LUMO
Zn(1)		
Cu(1)		
Ni(1)		

Figure S13. DFT-predicted frontier orbitals for the metal complexes of (1).

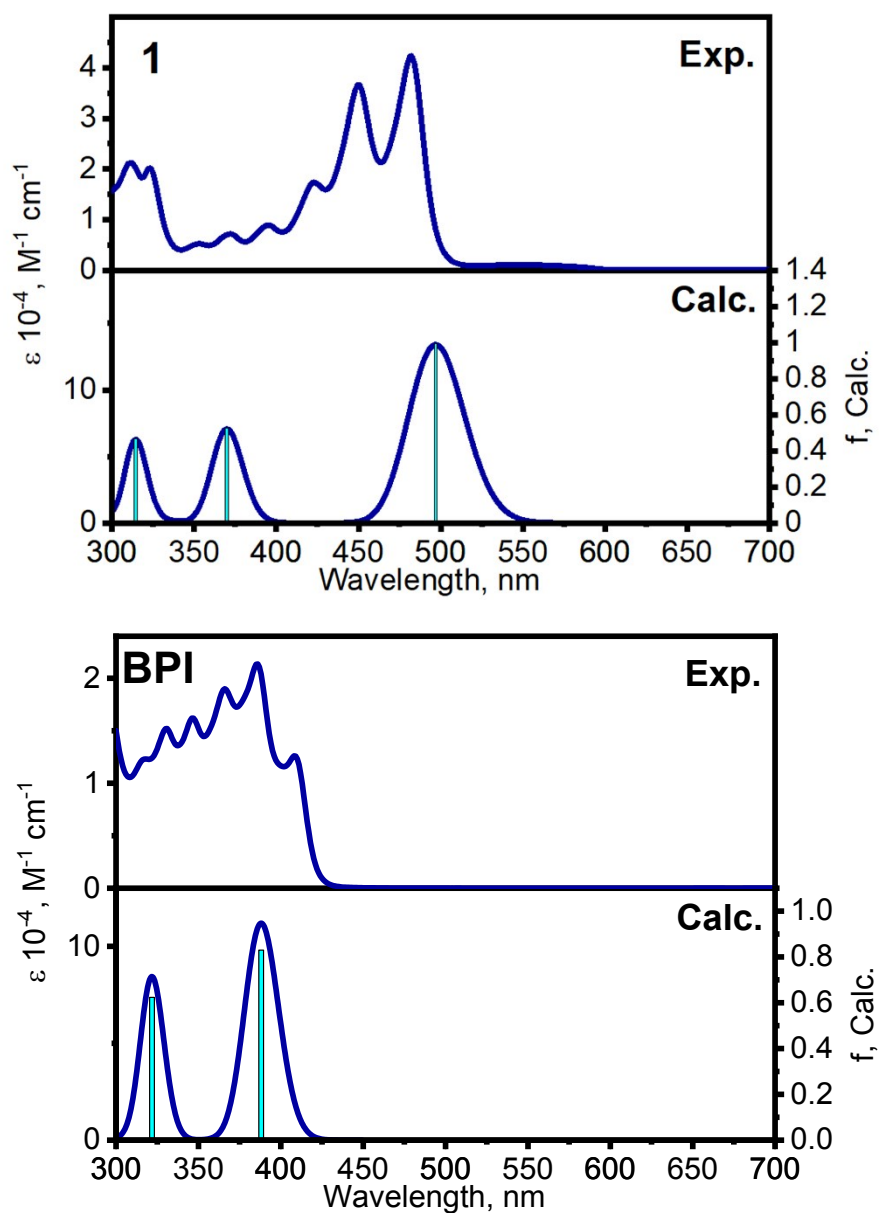


Figure S14: Experimental and TDDFT-predicted spectra for compounds BPYI (1) and BPI.

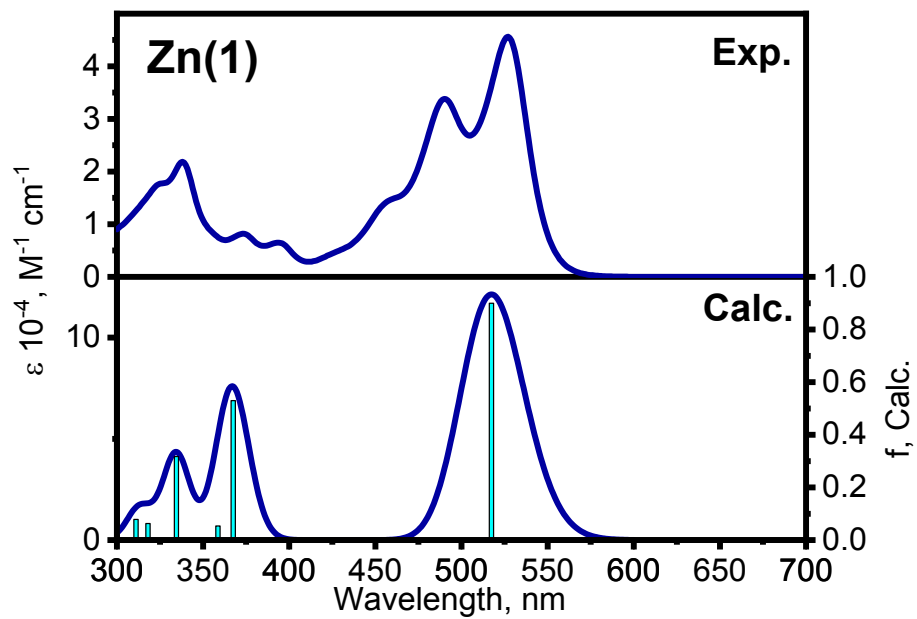


Figure S15: Experimental and TDDFT-predicted spectra for **Zn(1)**.

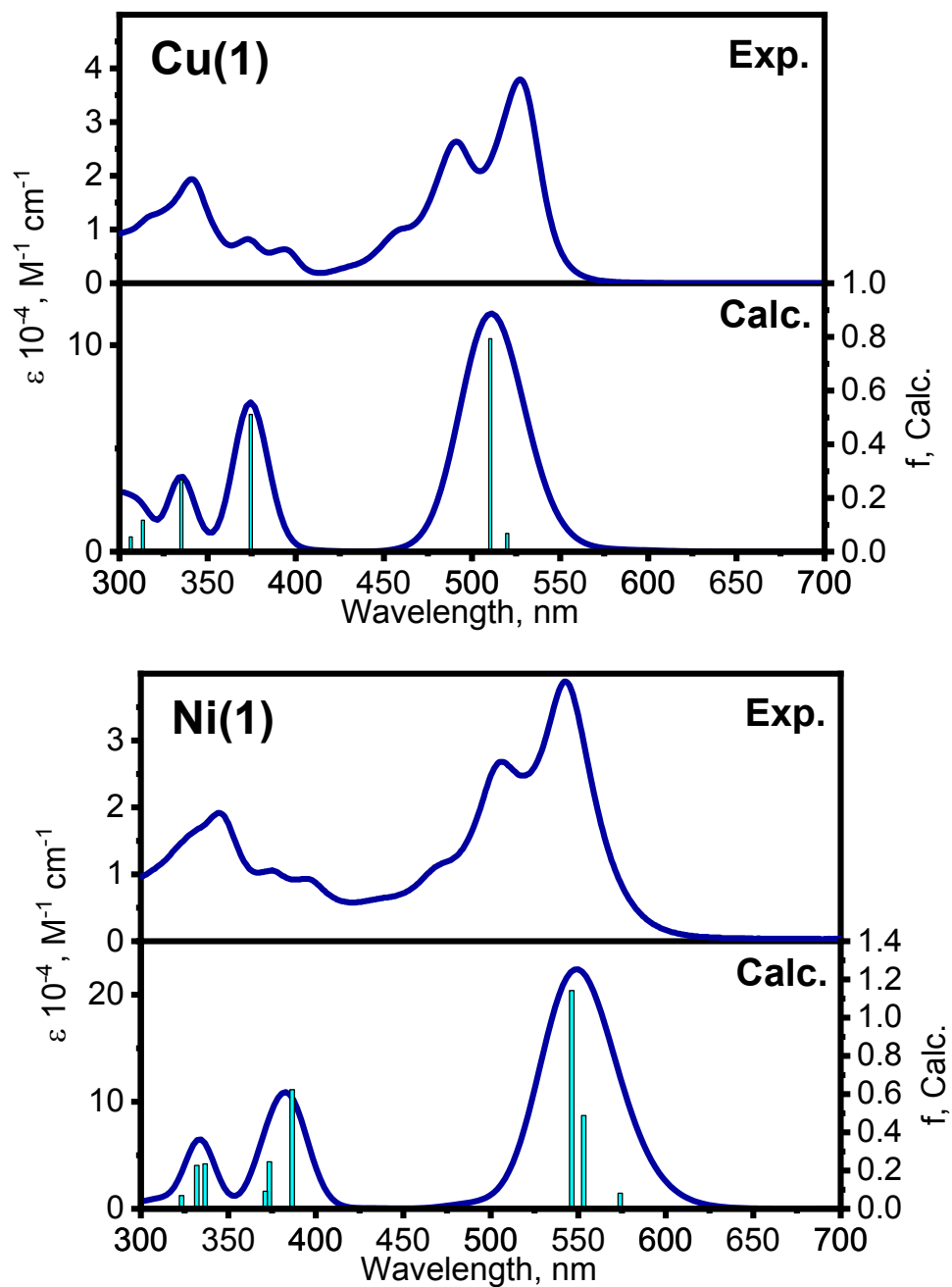


Figure S15: Experimental and TDDFT-predicted spectra for compounds **Cu(1)** and **Ni(1)**.

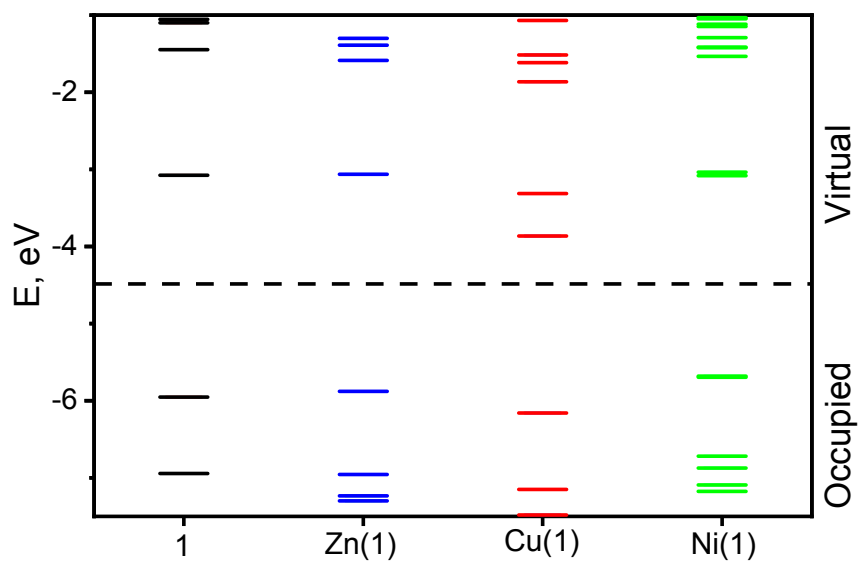
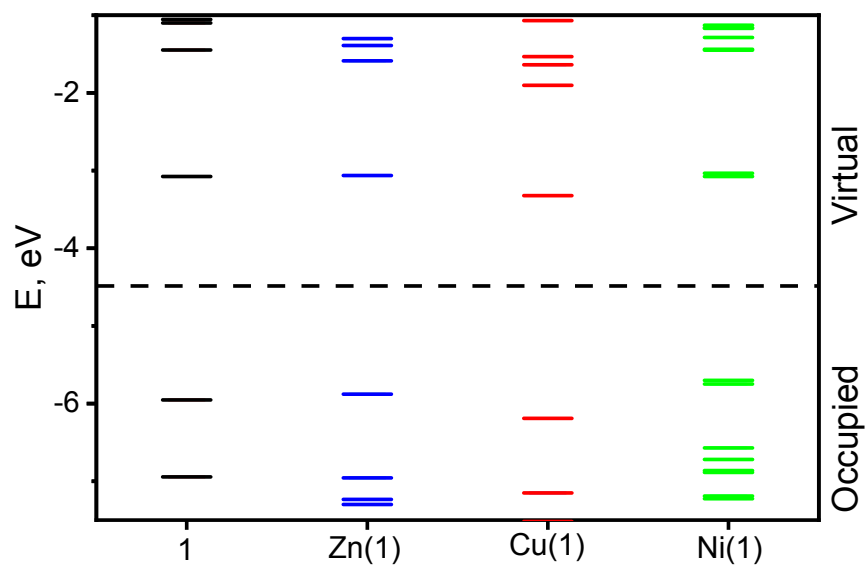


Figure S16: B3LYP α -spin relative energies of the frontier orbitals for compounds.

Figure S17: B3LYP β -spin relative energies of the frontier orbitals for compounds.

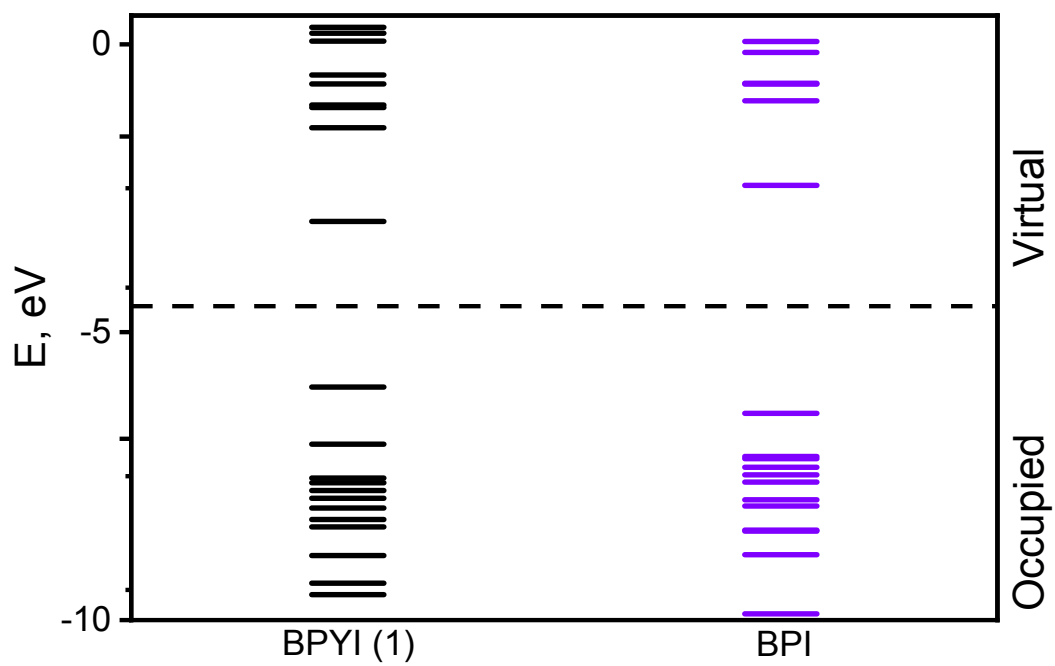


Figure S17: B3LYP relative energies of the frontier orbitals for BPYI (1) and BPI.

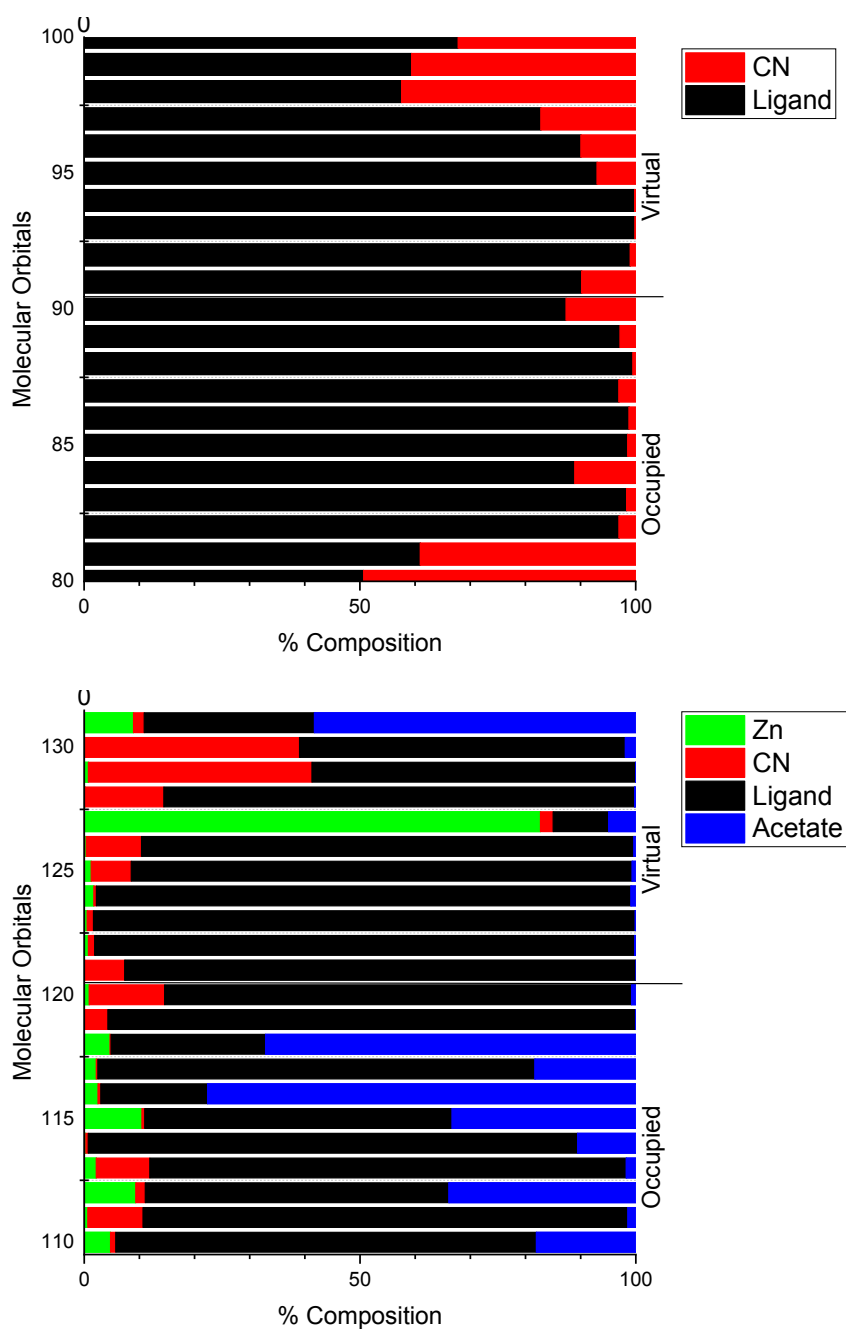


Figure S18: Orbital compositions for **1** and **Zn(1)**.

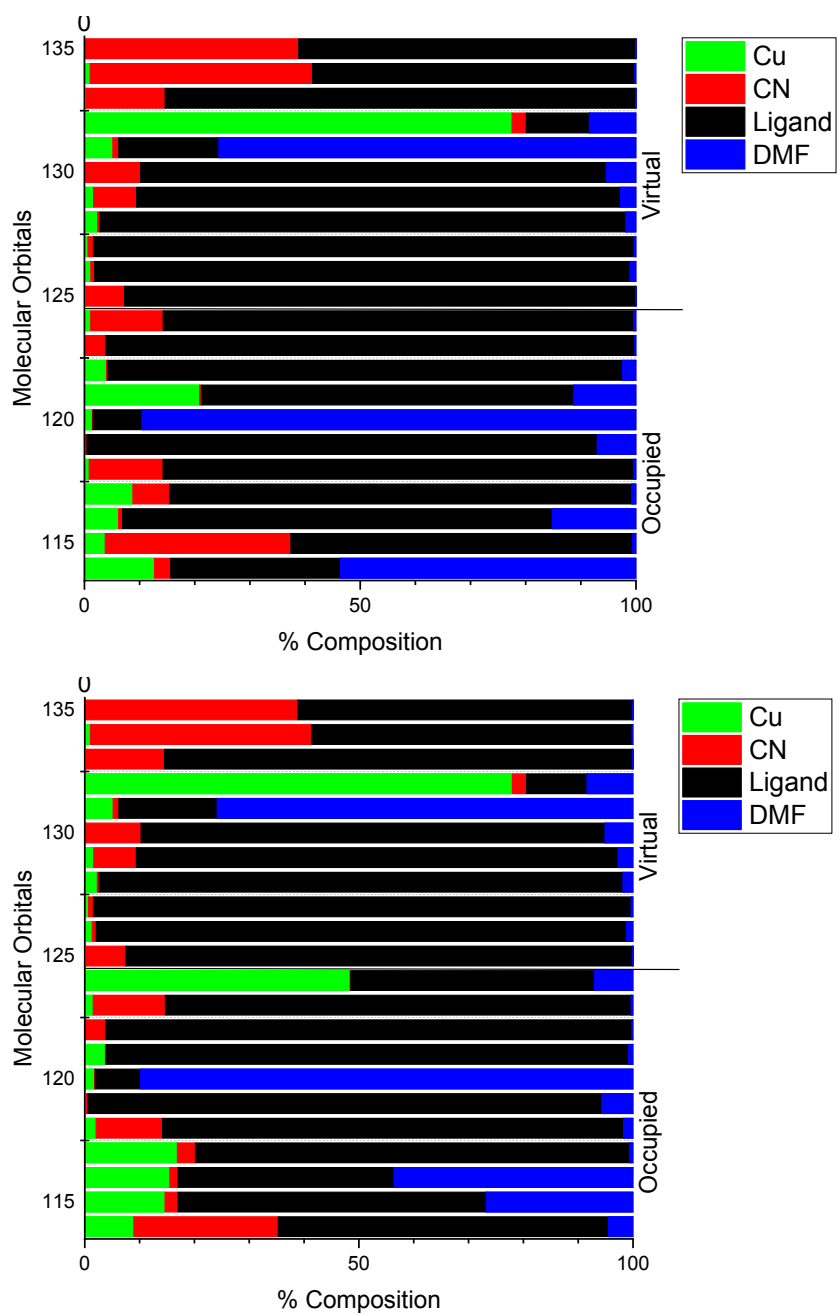


Figure S19: Orbital compositions for Cu(1), α -spin on top and β -spin on bottom.

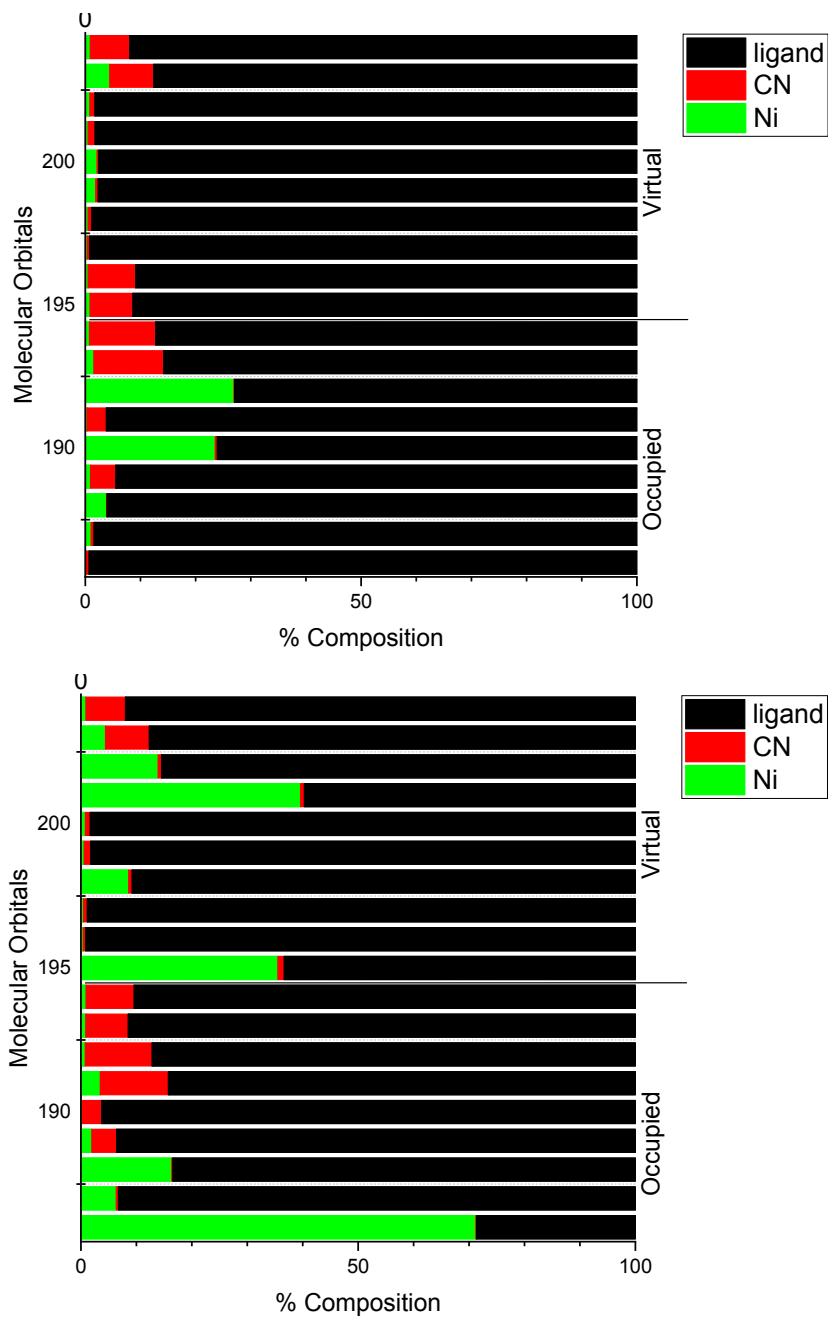


Figure S20: Orbital compositions for Ni(1), α -spin on top and β -spin on bottom.

Table S1: X-ray crystal data and structure parameters for compounds **1**, **Zn(1)**, **Cu(1)**, and **Ni(1)**.

Compound	1 BPYI	Zn(1)	Cu(1)	Ni(1)
CCDC	2043926	2043927	2043928	2043929
Empirical formula	C ₂₅ H ₂₀ N ₆ O	C ₂₄ H ₁₅ N ₅ O ₂ Zn	C ₂₅ H ₁₉ ClCuN ₆ O ₅	C ₄₄ H ₂₄ N ₁₀ Ni
Formula weight	420.47	470.78	582.45	751.44
Crystal system	Tetragonal	Triclinic	Monoclinic	Tetragonal
Space group	P4 ₁	P-1	P2 ₁ /n	I4 ₁ /acd
a/ Å	12.557(3)	7.9525(11)	8.025	17.9413(11)
b/ Å	12.557	10.8324(15)	28.235	17.9413(11)
c/ Å	25.933(6)	12.7721(19)	10.565	42.212(2)
α(°)	90	69.282(8)	90	90
β(°)	90	75.267(9)	101.66	90
γ(°)	90	83.931(9)	90	90
Volume (Å ³)	4089.2(19)	995.1(3)	2344.4	13587.6(18)
Z	8	2	4	16
Dc (Mg/m ³)	1.366	1.571	1.650	1.469
μ (mm ⁻¹)	0.088	1.268	1.098	0.623
F(000)	1760	480	1188	6176
reflns collected	5255	15222	51946	15457
indep. reflns	5255	3577	5789	3006
GOF on F ²	1.033	1.050	1.072	0.989
R1 (on F _o ² , I > 2σ(I))	0.1284	0.0922	0.0988	0.0531
wR2 (on F _o ² , I > 2σ(I))	0.2852	0.2340	0.2303	0.1009
R1 (all data)	0.2387	0.1511	0.1248	0.1089
wR2 (all data)	0.3539	0.2727	0.2402	0.1154

Equation 1: Evans NMR method calculation for effective magnetic moment

$$\mu_{eff} = \sqrt{\frac{3 k_B}{N_A \beta^2} (\chi_M^P T)}$$

k_B = Boltzmann constant

N_A = Avogadro's number

β = Bohr magneton

χ_p = molar susceptibility

T = absolute temperature (K)

Equation 2: Evans NMR method calculation for mass susceptibility

$$\chi = \frac{\delta\nu^p}{\nu_o S_f m^p} + \chi_o$$

$\delta\nu^p$ = Shift in frequency for an internal inert reference; regular methanol (Hz)

ν_o = frequency of the NMR spectrometer (Hz)

S_f = Shape factor of the NMR spectrometer; $4\pi/3$ sample axis parallel to magnetic field

m^p = concentration of solute (g/mL)

χ_o = mass susceptibility of the deuterated solvent (mL/g)

Equation 3: Evans NMR method calculation for molar susceptibility

$$\chi_M^P = \frac{\delta\nu^p M^p}{\nu_o S_f m^p} - \chi_M^{dia}$$

$\delta\nu^p$ = Shift in frequency for an internal inert reference; regular methanol (Hz)

M^p = Molecular weight of solute (g/mol)

ν_o = frequency of the NMR spectrometer (Hz)

S_f = Shape factor of the NMR spectrometer; $4\pi/3$ sample axis parallel to magnetic field

m^p = concentration of solute (g/mL)

χ_M^{dia} = diamagnetic constant (mL/mol)

Table S2: Evans NMR method data and parameters.

Compound	Ni	Cu
Mass (mg)	1.5	1.6
Volume (mL)	1	1
Deuterated Solvent	Methanol	Methanol
Mass Susceptibility of solvent (mL/g)	5.3×10^{-7}	5.3×10^{-7}
Shift in frequency (Hz)	14.02	15.67
Temperature (K)	303.15	303.15
Molecular weight of solute (g/mol)	751.44	509.36
Frequency of NMR spectrometer (Hz)	4×10^8	4×10^8
Mass Susceptibility (mL/g)	6.1×10^{-6}	6.37×10^{-6}
Molar Susceptibility (mL/mol)	4.59×10^{-3}	3.24×10^{-3}
Diamagnetic constant (mL/mol)	-3.87×10^{-4}	-2.30×10^{-4}
μ_{eff} (μ_{β})	3.4	2.8

Table S3: TDDFT-predicted energies and expansion coefficients for compound **1** (only excited states with $f > 0.05$ and $\lambda > 270$ nm are listed).

Excited State	1:	Singlet-A	2.4950 eV	496.93 nm	$f=0.9941$	$\langle S^{**2} \rangle = 0.000$
	90 -> 91	0.70623				
Excited State	2:	Singlet-A	3.3510 eV	369.99 nm	$f=0.5238$	$\langle S^{**2} \rangle = 0.000$
	89 -> 91	0.69523				
	90 -> 92	-0.10532				
Excited State	6:	Singlet-A	3.9424 eV	314.49 nm	$f=0.4650$	$\langle S^{**2} \rangle = 0.000$
	89 -> 91	0.10142				
	90 -> 92	0.69331				

Table S4: TDDFT-predicted energies and expansion coefficients for compound **BPI** (only excited states with $f > 0.05$ and $\lambda > 270$ nm are listed).

Excited State	1:	Singlet-B2	3.1945 eV	388.12 nm	$f=0.8299$	$\langle S^{**2} \rangle = 0.000$
	78 -> 79	0.70141				
Excited State	5:	Singlet-A1	3.8525 eV	321.83 nm	$f=0.6237$	$\langle S^{**2} \rangle = 0.000$
	77 -> 79	0.69420				
Excited State	8:	Singlet-A1	4.5389 eV	273.16 nm	$f=0.4748$	$\langle S^{**2} \rangle = 0.000$
	69 -> 79	0.11265				
	71 -> 79	-0.12727				
	78 -> 80	0.64409				
	78 -> 81	0.15667				

Table S5: TDDFT-predicted energies and expansion coefficients for compound **Zn(1)** (only excited states with $f > 0.05$ and $\lambda > 270$ nm are listed).

Excited State	1:	Singlet-A	2.3962 eV	517.42 nm	$f=0.8999$	$\langle S^{**2} \rangle = 0.000$
	120 -> 121	0.70374				
Excited State	2:	Singlet-A	3.3738 eV	367.49 nm	$f=0.5297$	$\langle S^{**2} \rangle = 0.000$
	119 -> 121	0.69910				
Excited State	3:	Singlet-A	3.4570 eV	358.65 nm	$f=0.0535$	$\langle S^{**2} \rangle = 0.000$
	117 -> 121	0.60845				
	118 -> 121	-0.34795				
Excited State	5:	Singlet-A	3.7062 eV	334.53 nm	$f=0.3172$	$\langle S^{**2} \rangle = 0.000$
	120 -> 122	0.69836				
Excited State	6:	Singlet-A	3.8983 eV	318.05 nm	$f=0.0631$	$\langle S^{**2} \rangle = 0.000$
	120 -> 123	0.69942				
Excited State	7:	Singlet-A	3.9854 eV	311.10 nm	$f=0.0785$	$\langle S^{**2} \rangle = 0.000$
	120 -> 124	0.69731				
Excited State	12:	Singlet-A	4.4742 eV	277.11 nm	$f=0.3183$	$\langle S^{**2} \rangle = 0.000$
	113 -> 121	0.12669				
	114 -> 121	0.29144				
	120 -> 125	0.60068				
	120 -> 126	-0.12703				

Table S6: TDDFT-predicted energies and expansion coefficients for compound **Cu(1)** (only excited states with $f > 0.05$ and $\lambda > 270$ nm are listed).

Excited State 6:	2.003-A	2.3843 eV	519.99 nm	$f=0.0683$	$\langle S^{*2} \rangle=0.753$
	124A ->125A	-0.21501			
	97B ->124B	0.26046			
	100B ->124B	0.54514			
	101B ->124B	0.25681			
	105B ->124B	-0.16894			
	107B ->124B	0.37628			
	113B ->124B	-0.18799			
	114B ->124B	0.18689			
	120B ->124B	-0.29054			
	122B ->124B	-0.27741			
	123B ->125B	-0.21708			
Excited State 7:	2.004-A	2.4294 eV	510.35 nm	$f=0.7930$	$\langle S^{*2} \rangle=0.754$
	124A ->125A	0.66736			
	100B ->124B	0.19036			
	107B ->124B	0.19517			
	120B ->124B	-0.10645			
	123B ->125B	0.65482			
Excited State 14:	2.010-A	3.3106 eV	374.51 nm	$f=0.5113$	$\langle S^{*2} \rangle=0.760$
	123A ->125A	0.69976			
	122B ->125B	0.69403			
Excited State 20:	2.006-A	3.6993 eV	335.15 nm	$f=0.2676$	$\langle S^{*2} \rangle=0.756$
	124A ->126A	0.69145			
	123B ->126B	0.69855			
Excited State 25:	2.037-A	3.9574 eV	313.30 nm	$f=0.1175$	$\langle S^{*2} \rangle=0.787$
	124A ->127A	0.68834			
	123B ->127B	0.68525			
Excited State 27:	2.005-A	4.0462 eV	306.42 nm	$f=0.0552$	$\langle S^{*2} \rangle=0.755$
	120A ->125A	-0.12475			
	124A ->128A	0.66939			
	109B ->124B	-0.12525			
	116B ->124B	0.15483			
	120B ->125B	-0.13076			
	123B ->128B	0.66380			
Excited State 30:	2.019-A	4.1321 eV	300.05 nm	$f=0.0805$	$\langle S^{*2} \rangle=0.769$
	124A ->128A	0.10073			
	101B ->124B	0.14283			
	105B ->124B	-0.13620			
	109B ->124B	0.18458			
	110B ->124B	0.11946			
	113B ->124B	0.34615			
	114B ->124B	-0.50127			
	115B ->124B	-0.16911			
	116B ->124B	-0.18618			
	117B ->124B	0.61714			
	120B ->124B	0.10620			
Excited State 31:	2.061-A	4.1742 eV	297.02 nm	$f=0.0704$	$\langle S^{*2} \rangle=0.812$
	119A ->125A	0.17589			

124A ->128A	-0.12568				
99B ->124B	0.10553				
100B ->124B	0.13426				
101B ->124B	-0.10506				
109B ->124B	-0.27997				
111B ->124B	0.24661				
113B ->124B	0.10912				
115B ->124B	-0.44315				
116B ->124B	0.65767				
117B ->124B	0.12566				
119B ->125B	0.12378				
123B ->128B	-0.11082				
Excited State 35:	2.049-A	4.3439 eV	285.42 nm	f=0.1382	<S**2>=0.800
119A ->125A	-0.18548				
124A ->129A	0.10594				
100B ->124B	0.21375				
101B ->124B	-0.23817				
109B ->124B	-0.40353				
111B ->124B	-0.18453				
114B ->124B	-0.31031				
115B ->124B	0.59600				
116B ->124B	0.15164				
119B ->125B	-0.23175				
Excited State 36:	2.165-A	4.4056 eV	281.42 nm	f=0.1936	<S**2>=0.921
105B ->124B	0.20757				
107B ->124B	0.31356				
113B ->124B	0.76446				
114B ->124B	0.19205				
117B ->124B	-0.25556				
118B ->125B	-0.23091				
123B ->130B	-0.21268				
Excited State 40:	2.240-A	4.5501 eV	272.49 nm	f=0.2571	<S**2>=1.005
115A ->125A	-0.12448				
119A ->125A	-0.20544				
124A ->129A	-0.52422				
114B ->125B	0.19495				
119B ->125B	-0.17010				
121B ->128B	-0.12306				
123B ->129B	0.70334				
Excited State 41:	2.072-A	4.5657 eV	271.56 nm	f=0.1223	<S**2>=0.823
118A ->125A	-0.35625				
124A ->130A	-0.22758				
105B ->124B	-0.28012				
107B ->124B	-0.34218				
110B ->124B	0.11848				
114B ->124B	0.46756				
115B ->124B	0.15618				
116B ->125B	-0.11215				
117B ->124B	0.26516				
118B ->125B	-0.32042				
123B ->130B	-0.31323				

Table S7: TDDFT-predicted energies and expansion coefficients for compound **Ni(1)** (only excited states with $f > 0.05$ and $\lambda > 270$ nm are listed).

Excited State 8:	3.699-A	2.1601 eV	573.96 nm	$f=0.0814$	$\langle S^{**2} \rangle = 3.171$
	193A -> 196A	-0.56232			
	194A -> 195A	0.78738			
	191B -> 194B	0.23828			
Excited State 10:	3.019-A	2.2415 eV	553.12 nm	$f=0.4881$	$\langle S^{**2} \rangle = 2.028$
	193A -> 195A	0.48975			
	194A -> 196A	0.50885			
	191B -> 193B	0.37794			
	192B -> 194B	0.58831			
Excited State 11:	3.031-A	2.2695 eV	546.29 nm	$f=1.1416$	$\langle S^{**2} \rangle = 2.046$
	193A -> 196A	0.67808			
	194A -> 195A	0.28707			
	182B -> 195B	0.10028			
	191B -> 194B	0.52656			
	192B -> 193B	0.38981			
Excited State 24:	3.005-A	3.2080 eV	386.49 nm	$f=0.6230$	$\langle S^{**2} \rangle = 2.007$
	189A -> 195A	0.15229			
	191A -> 196A	0.68671			
	189B -> 193B	0.16214			
	190B -> 194B	0.67533			
Excited State 29:	3.072-A	3.3198 eV	373.46 nm	$f=0.2461$	$\langle S^{**2} \rangle = 2.109$
	189A -> 195A	0.76278			
	191A -> 196A	-0.24291			
	189B -> 193B	0.56773			
Excited State 32:	3.151-A	3.3410 eV	371.10 nm	$f=0.0913$	$\langle S^{**2} \rangle = 2.232$
	187A -> 195A	-0.46255			
	188A -> 196A	-0.43393			
	193A -> 199A	-0.12498			
	194A -> 200A	-0.10206			
	187B -> 193B	-0.35859			
	188B -> 194B	-0.42407			
	192B -> 195B	0.46508			
Excited State 43:	3.094-A	3.6822 eV	336.71 nm	$f=0.2352$	$\langle S^{**2} \rangle = 2.144$
	194A -> 198A	0.69486			
	183B -> 193B	-0.11959			
	184B -> 194B	-0.12309			
	186B -> 193B	0.12277			
	191B -> 196B	-0.19786			
	192B -> 197B	0.62753			
Excited State 48:	3.071-A	3.7349 eV	331.96 nm	$f=0.2271$	$\langle S^{**2} \rangle = 2.108$
	185A -> 195A	-0.10336			
	186A -> 196A	-0.10447			
	193A -> 197A	0.69758			
	194A -> 201A	-0.17693			

191B -> 196B	0.63797				
191B -> 200B	-0.12999				
192B -> 197B	0.11851				
Excited State 54:	3.119-A	3.8356 eV	323.25 nm	f=0.0690	<S**2>=2.182
193A -> 199A	0.53547				
194A -> 200A	0.23341				
185B -> 201B	-0.10268				
191B -> 198B	0.75996				
Excited State 99:	3.500-A	4.4536 eV	278.39 nm	f=0.1505	<S**2>=2.813
181A -> 196A	0.10872				
182A -> 195A	-0.17776				
185A -> 195A	-0.11147				
187A -> 201A	0.16936				
188A -> 197A	-0.16198				
188A -> 202A	-0.10812				
190A -> 199A	0.20182				
192A -> 200A	-0.21922				
193A -> 203A	0.18160				
194A -> 204A	0.17369				
172B -> 193B	0.10928				
175B -> 194B	0.16361				
179B -> 194B	-0.34614				
184B -> 194B	0.12231				
187B -> 199B	-0.11060				
188B -> 196B	0.14438				
191B -> 203B	0.64180				
192B -> 199B	0.10345				
192B -> 204B	0.16474				
Excited State 104:	3.114-A	4.4757 eV	277.02 nm	f=0.5376	<S**2>=2.175
185A -> 195A	0.19268				
186A -> 196A	0.10475				
193A -> 203A	-0.25341				
194A -> 204A	-0.48353				
183B -> 193B	-0.17939				
192B -> 204B	0.74358				

Table S8: DFT optimized geometry of compound **1**.

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-2.038857	-3.258985	-0.057884
2	1	0	-1.187090	-3.931563	-0.089187
3	6	0	-3.334263	-3.765451	-0.041210
4	1	0	-3.507123	-4.834732	-0.057735
5	6	0	-4.387492	-2.855015	-0.002507
6	1	0	-5.416348	-3.197406	0.012907
7	6	0	-4.107627	-1.496342	0.015558
8	1	0	-4.915437	-0.776281	0.044625
9	6	0	-2.766961	-1.070382	-0.004386
10	6	0	-2.426885	0.368090	0.008097
11	6	0	-3.528374	1.264493	0.021816
12	6	0	-1.148478	0.895988	0.004641
13	6	0	-0.707241	2.299105	0.004012
14	6	0	-1.413382	3.502378	0.007831
15	1	0	-2.493203	3.526537	0.014070
16	6	0	-0.698407	4.698157	0.003754
17	1	0	-1.236949	5.639202	0.006774
18	6	0	0.697746	4.698244	-0.004113
19	1	0	1.236168	5.639355	-0.007281
20	6	0	1.412865	3.502533	-0.008025
21	1	0	2.492683	3.526818	-0.014247
22	6	0	0.706863	2.299207	-0.004037
23	6	0	1.148302	0.896092	-0.004609
24	6	0	2.426790	0.368404	-0.007963
25	6	0	3.528161	1.264944	-0.021591
26	6	0	2.767112	-1.070041	0.004505
27	6	0	4.107835	-1.495743	-0.015095
28	1	0	4.915534	-0.775537	-0.043682
29	6	0	4.387935	-2.854378	0.002760
30	1	0	5.416849	-3.196579	-0.012417
31	6	0	3.334874	-3.765011	0.040984
32	1	0	3.507925	-4.834268	0.057327
33	6	0	2.039371	-3.258781	0.057441
34	1	0	1.187728	-3.931525	0.088423
35	7	0	-0.000043	0.141702	0.000041
36	1	0	-0.000044	-0.880124	-0.000003
37	7	0	-1.752593	-1.955606	-0.039136
38	7	0	-4.456768	1.957363	0.033506
39	7	0	1.752880	-1.955447	0.038838
40	7	0	4.456524	1.957852	-0.033172

$E_h = -1120.173695$ Hartree

Table S9: DFT optimized geometry of compound **BPI**.

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	7	0	0.000000	1.946679	-1.751293
2	7	0	0.000000	0.000000	0.210966
3	7	0	0.000000	2.409879	0.618461
4	6	0	0.000000	0.704739	4.789695
5	1	0	0.000000	1.236565	5.734627
6	6	0	0.000000	4.231754	-0.943798
7	1	0	0.000000	4.898771	-0.092163
8	6	0	0.000000	1.161289	0.987110
9	6	0	0.000000	2.423456	-3.014758
10	1	0	0.000000	1.680330	-3.805118
11	6	0	0.000000	1.427955	3.583609
12	1	0	0.000000	2.511231	3.573232
13	6	0	0.000000	0.703512	2.391736
14	6	0	0.000000	2.837363	-0.717412
15	6	0	0.000000	3.791188	-3.321390
16	1	0	0.000000	4.120774	-4.352744
17	6	0	0.000000	4.709200	-2.256902
18	1	0	0.000000	5.775656	-2.449945
19	7	0	0.000000	-1.946679	-1.751293
20	7	0	0.000000	-2.409879	0.618461
21	6	0	0.000000	-0.704739	4.789695
22	1	0	0.000000	-1.236565	5.734627
23	6	0	0.000000	-4.231754	-0.943798
24	1	0	0.000000	-4.898771	-0.092163
25	6	0	0.000000	-1.161289	0.987110
26	6	0	0.000000	-2.423456	-3.014758
27	1	0	0.000000	-1.680330	-3.805118
28	6	0	0.000000	-1.427955	3.583609
29	1	0	0.000000	-2.511231	3.573232
30	6	0	0.000000	-0.703512	2.391736
31	6	0	0.000000	-2.837363	-0.717412
32	6	0	0.000000	-3.791188	-3.321390
33	1	0	0.000000	-4.120774	-4.352744
34	6	0	0.000000	-4.709200	-2.256902
35	1	0	0.000000	-5.775656	-2.449945
36	1	0	0.000000	0.000000	-0.807567

 $E_h = -966.678725$ Hartree

Table S10: DFT optimized geometry of compound **Zn(1)**.

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	2.683375	-2.232021	-0.894964
2	1	0	3.446142	-1.485468	-0.717084
3	6	0	2.992427	-3.460970	-1.449828
4	1	0	4.010392	-3.686023	-1.741546
5	6	0	1.959169	-4.381614	-1.608665
6	1	0	2.146102	-5.357817	-2.041646
7	6	0	0.680060	-4.037749	-1.207449
8	1	0	-0.128346	-4.745233	-1.332260
9	6	0	0.419957	-2.769152	-0.653189
10	6	0	-0.960981	-2.434195	-0.244807
11	6	0	-1.819947	-3.556546	-0.103803
12	6	0	-1.504519	-1.166324	-0.058718
13	6	0	-2.917508	-0.793959	0.173222
14	6	0	-4.084546	-1.536904	0.346096
15	1	0	-4.077763	-2.616838	0.338850
16	6	0	-5.288528	-0.859007	0.539468
17	1	0	-6.201097	-1.427008	0.682807
18	6	0	-5.330357	0.534982	0.550321
19	1	0	-6.275278	1.044853	0.702370
20	6	0	-4.169305	1.286688	0.367484
21	1	0	-4.227196	2.365074	0.377029
22	6	0	-2.959976	0.617765	0.183468
23	6	0	-1.572504	1.077502	-0.045872
24	6	0	-1.107613	2.378077	-0.221848
25	6	0	-2.029831	3.445317	-0.055259
26	6	0	0.244476	2.797804	-0.646985
27	6	0	0.424672	4.088691	-1.181217
28	1	0	-0.422219	4.754986	-1.272941
29	6	0	1.674306	4.506827	-1.604019
30	1	0	1.799911	5.500255	-2.019562
31	6	0	2.757668	3.638338	-1.489942
32	1	0	3.754778	3.922011	-1.801826
33	6	0	2.527254	2.385719	-0.950327
34	1	0	3.332129	1.677628	-0.804148
35	6	0	2.952418	0.083983	2.133468
36	6	0	3.812425	0.048525	3.375950
37	1	0	3.311797	0.537189	4.211915
38	1	0	4.780061	0.514242	3.191908
39	1	0	3.985750	-0.995290	3.655213
40	7	0	-0.767486	-0.020511	-0.132105
41	7	0	1.440648	-1.884622	-0.509218

42	7	0	1.314357	1.967686	-0.540818
43	7	0	-2.489478	-4.495390	0.013701
44	7	0	-2.752413	4.340787	0.084504
45	8	0	1.680309	0.038052	2.276589
46	8	0	3.474187	0.127764	0.992968
47	30	0	1.200464	0.044820	0.282884

$E_h = -3127.570282$ Hartree

Table S11: DFT optimized geometry of compound **Cu(1)**.

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-1.995164	2.589000	-1.127781
2	1	0	-2.830687	1.903864	-1.140055
3	6	0	-2.144840	3.901833	-1.529140
4	1	0	-3.106268	4.253779	-1.879965
5	6	0	-1.031330	4.737346	-1.461548
6	1	0	-1.096791	5.775793	-1.764866
7	6	0	0.168185	4.226156	-1.002952
8	1	0	1.039049	4.864457	-0.950897
9	6	0	0.270118	2.875747	-0.614492
10	6	0	1.555581	2.348781	-0.138837
11	6	0	2.530720	3.345659	0.137200
12	6	0	1.906443	1.016381	0.029025
13	6	0	3.218680	0.474043	0.431838
14	6	0	4.427149	1.075394	0.784473
15	1	0	4.547087	2.147537	0.806508
16	6	0	5.507605	0.261342	1.121243
17	1	0	6.449812	0.719068	1.401024
18	6	0	5.389154	-1.127963	1.101994
19	1	0	6.239614	-1.746037	1.367009
20	6	0	4.187262	-1.737577	0.745195
21	1	0	4.123501	-2.814775	0.737555
22	6	0	3.099086	-0.930541	0.412132
23	6	0	1.717109	-1.231758	-0.008758
24	6	0	1.151998	-2.480747	-0.229030
25	6	0	1.934741	-3.636745	0.037959
26	6	0	-0.184919	-2.764902	-0.764853
27	6	0	-0.496205	-4.060651	-1.222278
28	1	0	0.249915	-4.840938	-1.166132
29	6	0	-1.740464	-4.337327	-1.756391
30	1	0	-1.966961	-5.335686	-2.112351
31	6	0	-2.689231	-3.318854	-1.834882
32	1	0	-3.675735	-3.483679	-2.248182
33	6	0	-2.340114	-2.070678	-1.359066
34	1	0	-3.045411	-1.251306	-1.377777
35	6	0	-3.048591	-0.072296	1.631785
36	1	0	-2.300623	-0.453029	2.331269
37	6	0	-4.572331	-0.359081	3.496880
38	1	0	-3.668194	-0.706769	3.992745
39	1	0	-5.314915	-1.158788	3.506872
40	1	0	-4.968888	0.501222	4.038810
41	6	0	-5.389846	0.513535	1.310455

42	1	0	-5.795902	1.419900	1.763741
43	1	0	-6.173605	-0.245079	1.272447
44	1	0	-5.044283	0.731713	0.304931
45	29	0	-0.860584	0.116339	-0.302090
46	7	0	1.059779	-0.040989	-0.191993
47	7	0	-0.825852	2.080863	-0.684205
48	7	0	-1.126987	-1.792882	-0.835182
49	7	0	3.288920	4.193253	0.355913
50	7	0	2.529136	-4.607579	0.251073
51	7	0	-4.273293	0.020684	2.116395
52	8	0	-2.731309	0.244365	0.456613

 $E_h = -3008.566038$ Hartree

Table S12: DFT optimized geometry of compound Ni(1).

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-0.970011	-1.676388	2.316321
2	1	0	-1.696273	-0.878138	2.256601
3	6	0	-1.186933	-2.762418	3.147393
4	1	0	-2.082649	-2.813960	3.753560
5	6	0	-0.231151	-3.775430	3.152928
6	1	0	-0.361367	-4.662731	3.762434
7	6	0	0.907909	-3.624896	2.383029
8	1	0	1.675356	-4.386229	2.408649
9	6	0	1.089064	-2.468060	1.595557
10	6	0	2.387023	-2.240008	0.940737
11	6	0	3.287957	-3.335260	1.016880
12	6	0	2.857819	-1.049858	0.383700
13	6	0	4.277383	-0.674115	0.203923
14	6	0	5.481587	-1.349019	0.409764
15	1	0	5.510138	-2.383180	0.718028
16	6	0	6.679373	-0.666789	0.200888
17	1	0	7.619800	-1.184318	0.355634
18	6	0	6.679159	0.667881	-0.203995
19	1	0	7.619422	1.185467	-0.359527
20	6	0	5.481119	1.349983	-0.411886
21	1	0	5.509254	2.384137	-0.720239
22	6	0	4.277177	0.674937	-0.205016
23	6	0	2.857507	1.050589	-0.383877
24	6	0	2.386305	2.240849	-0.940338
25	6	0	3.286922	3.336358	-1.016221
26	6	0	1.088117	2.468725	-1.594780
27	6	0	0.906213	3.625985	-2.381445
28	1	0	1.673036	4.387978	-2.406309
29	6	0	-0.232803	3.776121	-3.151498
30	1	0	-0.363602	4.663723	-3.760440
31	6	0	-1.187780	2.762346	-3.146854
32	1	0	-2.083322	2.813510	-3.753314
33	6	0	-0.970301	1.676051	-2.316273
34	1	0	-1.696083	0.877331	-2.257029
35	7	0	2.069928	0.000309	0.000066
36	7	0	0.111824	-1.532620	1.532868
37	7	0	0.111474	1.532621	-1.532689
38	7	0	3.978618	-4.265278	1.065667
39	7	0	3.977328	4.266582	-1.064800
40	28	0	0.000004	0.000006	0.000247

41	6	0	0.969975	1.676428	2.316317
42	1	0	1.696245	0.878184	2.256608
43	6	0	1.186884	2.762470	3.147376
44	1	0	2.082598	2.814028	3.753545
45	6	0	0.231091	3.775472	3.152896
46	1	0	0.361298	4.662783	3.762389
47	6	0	-0.907965	3.624917	2.382997
48	1	0	-1.675417	4.386245	2.408605
49	6	0	-1.089108	2.468068	1.595539
50	6	0	-2.387061	2.239989	0.940715
51	6	0	-3.288043	3.335203	1.016870
52	6	0	-2.857825	1.049836	0.383667
53	6	0	-4.277380	0.674080	0.203878
54	6	0	-5.481588	1.348978	0.409709
55	1	0	-5.510145	2.383140	0.717974
56	6	0	-6.679367	0.666741	0.200821
57	1	0	-7.619799	1.184264	0.355559
58	6	0	-6.679139	-0.667928	-0.204064
59	1	0	-7.619398	-1.185521	-0.359604
60	6	0	-5.481093	-1.350021	-0.411950
61	1	0	-5.509219	-2.384174	-0.720308
62	6	0	-4.277155	-0.674969	-0.205069
63	6	0	-2.857478	-1.050604	-0.383927
64	6	0	-2.386255	-2.240853	-0.940399
65	6	0	-3.286843	-3.336386	-1.016283
66	6	0	-1.088060	-2.468701	-1.594834
67	6	0	-0.906135	-3.625942	-2.381524
68	1	0	-1.672950	-4.387941	-2.406413
69	6	0	0.232890	-3.776048	-3.151568
70	1	0	0.363705	-4.663634	-3.760529
71	6	0	1.187858	-2.762264	-3.146888
72	1	0	2.083409	-2.813407	-3.753337
73	6	0	0.970359	-1.675992	-2.316283
74	1	0	1.696135	-0.877269	-2.257008
75	7	0	-2.069917	-0.000314	0.000027
76	7	0	-0.111857	1.532638	1.532864
77	7	0	-0.111426	-1.532590	-1.532710
78	7	0	-3.978867	4.265096	1.065711
79	7	0	-3.977330	-4.266558	-1.064721

 $E_h = -3747.535260$ Hartree

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