

1 **Supplementary Information**

2 **Unique calibration of passive air sampling for field monitoring of PAHs with polyethylene**  
3 **thin films across seasons and locations**

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8 **S.1. Sampling sites**

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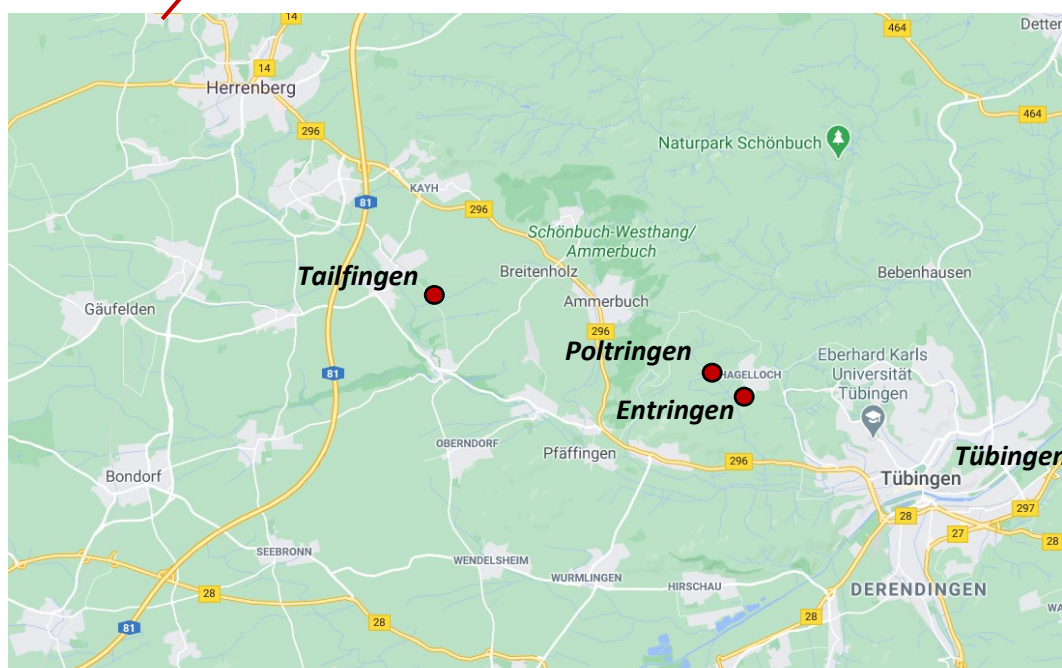


Figure.S 1: Top: Map of Germany with the study region marked in red. Bottom: Study sites Entringen, Poltringen and Tailfingen in red. Both maps were downloaded from Google.maps (11.12.20).

14 **S.2. Set**



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17 *Figure.S.2: Sampling set-up in the field. Left: Aluminium box with 55 x 55 x 10 cm (l, w, h), fixed in 1.2 m height above ground.*  
18 *Right: View into the aluminium box with the lid open, showing the PE sheets hanging within the box cover.*

19 **S.3. Test on photodegradation on the sampler**

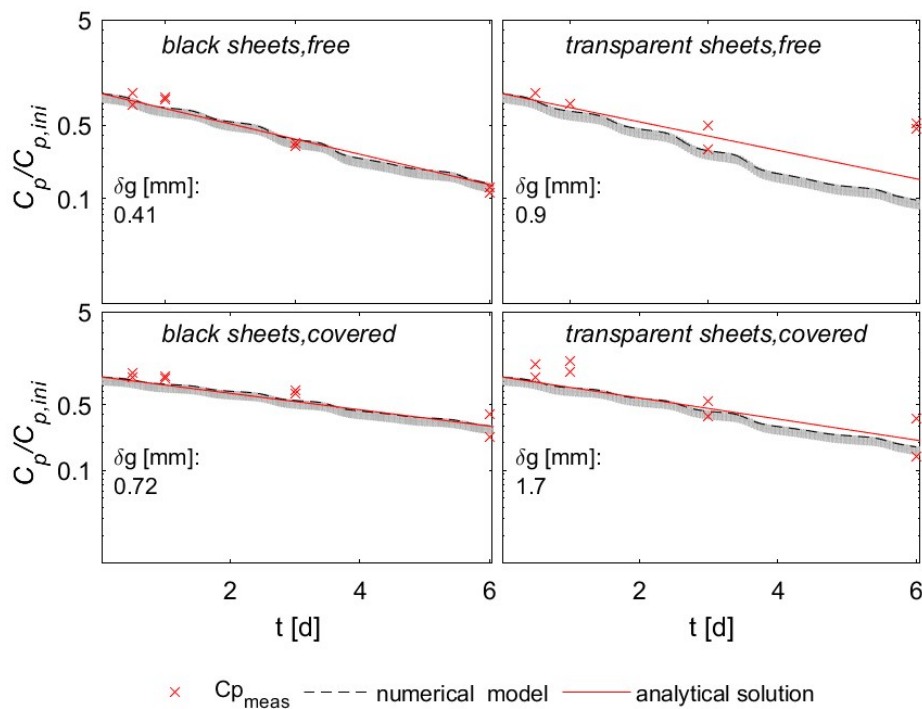
20 S.3.1. Preliminary test on the sampling design

21 The monitoring covered two weeks between 30<sup>th</sup> of September and 14<sup>th</sup> of October in 2015 with  
22 specific sampling times after 12 and 24 hours as well as 3, 6 and 13 days; samples were always taken  
23 in duplicate for each setting. Black and transparent sheets were spiked separately with Ant-D10 as  
24 PRC before deployment. In the field an aluminium cover at 1.2 m height was used to protect sheets  
25 against the influence of wind, rain and direct sunlight, while uncovered sheets were attached to a  
26 cord, spanned right beside the cover at about 1.6 m height. The different PE sheets varied additionally  
27 in their thickness with 30  $\mu\text{m}$  (transparent PE) and 80  $\mu\text{m}$  (black PE).



*Figure.S.3: Different sampling set-ups in the field (Entringen): Left: Transparent and black PE sheets hanging within the aluminium box cover. Right: Black (top) and transparent (bottom) sheets hanging freely without protection from wind, sunlight etc..*

28 Figure.S 4 shows a good fit for the measured and the modelled loss of Ant-D<sub>10</sub> from PE for each of the  
 29 individual settings and sheets. The release-curves of Ant-D10 from PE show a very comparable trend  
 30 for all four designs – only the initial concentrations on PE vary significantly (this was due to a spiking  
 31 solution without 20% organic solvent). Based on these loss curves, air side boundary layers ( $\delta_g$ ) were  
 32 fitted and respective rate constants for Ant-D10 were calculated based on the fitted  $\delta_g$ , as listed in  
 33 **Table.S 1**. Higher loss rates were measured for the uncovered samplers with 0.3-0.37 [ $\text{day}^{-1}$ ] compared  
 34 to 0.17-0.25 [ $\text{day}^{-1}$ ] for the covered sheets. As expected, the thickness of the air -side boundary layer  
 35 of the covered set-up is approx. two times larger than in the uncovered set-up. The resulting  
 36 atmospheric concentrations calculated for the four representative PAHs are also listed in **Table.S 1**. In  
 37 all settings similar concentration were determined for Phe indicating no major influence of  
 38 photodegradation of PRC or target compound in PE. For the other compounds, differences of a factor  
 39 of two or more were observed with no clear tendency of covered vs. open or black vs. transparent.



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41 *Figure.S 4: Measured (red crosses) and modelled (dashed line) loss of the PRC from PE sheets, normalized to the initial PRC-*  
 42 *concentration on PE. The different sampling set-ups in the field are compared with transparent versus black sheets as well as*  
 43 *covered and uncovered. The shaded grey area illustrates a standard deviation of 10 % and the red line shows the analytical*  
 44 *solution.*

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46 **Table.S 1: Loss rate constant  $k_e$  [ $\text{day}^{-1}$ ] (calculated with fitted  $\delta_g$ ) and fitted atmospheric concentrations**  
 47 **[ $\text{ng}/\text{m}^3$ ] of Fln, Phe, Fth and Pyr for the comparison of the different sampling set-ups in the field.**

field set-up	$k_e$ [ $\text{day}^{-1}$ ]	Fln [ $\text{ng}/\text{m}^3$ ]	Phe [ $\text{ng}/\text{m}^3$ ]	Fth [ $\text{ng}/\text{m}^3$ ]	Pyr [ $\text{ng}/\text{m}^3$ ]
black, uncovered	0.3	$1.7 \pm 0.3$	$2.2 \pm 0.1$	$0.3 \pm 0.03$	$0.2 \pm 0.04$
black, covered	0.17	$1.7 \pm 0.1$	$2.2 \pm 0.3$	$0.3 \pm 0.05$	$0.2 \pm 0.06$
transparent, uncovered	0.37	$3.4 \pm 0.5$	$2.6 \pm 0.4$	$0.1 \pm 0.07$	$0.1 \pm 0.02$

transparent, covered	0.25	3 ± 0.3	3.6 ± 0.2	0.9 ± 0.05	0.4 ± 0.01
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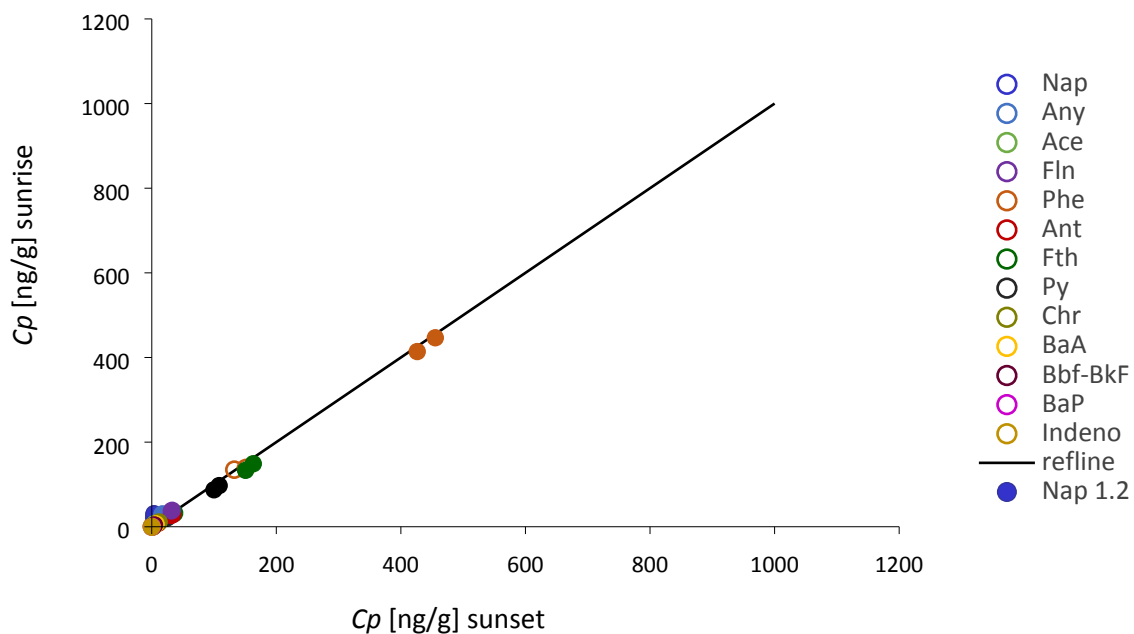
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50 S.3.2. Diurnal variations

51 In order to evaluate diurnal changes in concentration we conducted a day vs. night sampling in  
 52 Entringen at the end of the sampling campaign in November 2017, when LMW PAHs are expected to  
 53 have reached dynamic equilibrium between PE and air. During this sampling campaign we added a  
 54 second set of samplers at 10 cm above ground (also below a cover). After a very sunny day (with 7  
 55 hours of sunlight) samples were taken in duplicates in both heights (4 replicates in total) right before  
 56 sunset. The next set of samples was taken 15 hours later, during sunrise. For most of the PAHs sunset  
 57 and sunrise sampling results agreed very well as shown in Figure.S 5. Only the most volatile  
 58 compounds (naphthalene, acenaphthene and acenaphthylene) show higher concentrations on PE  
 59 sampled overnight, at both heights. This observation can be explained by the rapid equilibration of  
 60 these compounds (within hours), changes in gaseous concentrations and the temperature sensitivity  
 61 of the partition coefficient. The air temperature was higher when sampling at sunset (10°C) compared  
 62 to sampling in the morning (2°C) leading to a lower  $K_{pg}$  and lower  $C_p$ .

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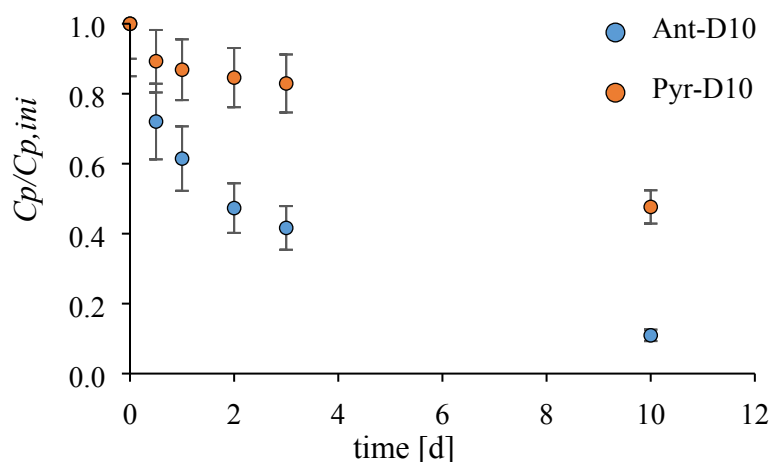
65 *Figure.S 5: Concentrations of 13 PAHs measured on PE after three weeks of sampling during November 2017 (here, all*  
 66 *compounds which exceed the limit of quantification are considered). Samples taken after a sunny (and warm) day (10 °C) are*  
 67 *compared to samples taken at the sunrise of the next day (2 °C); duplicate samples at 120 cm (filled symbols) and 10 cm (open*  
 68 *symbols) above ground, in total four replicate samples at both sampling times.*

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## S.3.3. Comparison of different deuterated compounds as PRC

70 During the first three sampling campaigns in our seasonal monitoring we spiked the PE sheets  
 71 additionally with Pyr-D10 (following [67]). But only during May 2016 we determined a reliable change  
 72 of concentrations of Pyr-D10 on PE, which is shown in Figure.S 6. As expected, the loss of Ant-D10 is  
 73 faster than for Pyr-D10, and the fitted air-side boundary layer was in close agreement for both  
 74 compounds with  $0.6 \pm 0.1$  mm based on Ant-D10 and  $0.3 \pm 0.1$  mm based on Pyr-D10.

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77 Figure.S 6: PRC-loss during May 2016, comparing Ant-D10 and Pyr-D10. Concentrations shown here are average values for  
 78 the three sites and 3 replicate samples per sampling time; error bars refer to a standard deviation of 10%.

## 79 S.4. Mass transfer resistances

80 To check on the potential influence of the polymer-side boundary layer on the overall exchange  
 81 kinetics, we use the double-film diffusion model and compare the relative thicknesses of air and  
 82 polymer-side boundary layers:

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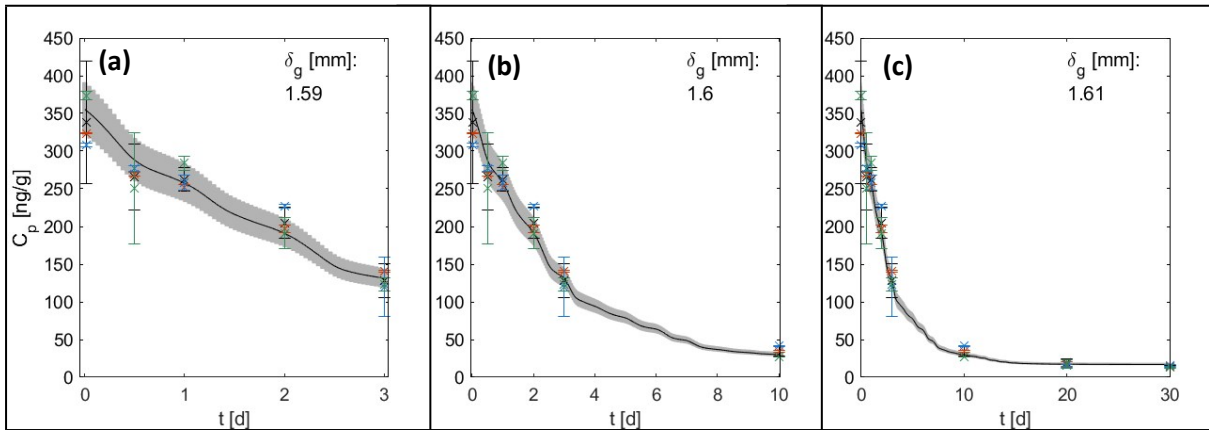
$$\frac{dC_p}{dt} = - \frac{2}{\frac{D_{poly}}{\delta_p} + \frac{1}{\delta_g K_{p/g}}} \frac{1}{D_g} (C_p - C_{p,eq}) \quad (1.7.b)$$

84 Here,  $D_p$  is the diffusion coefficient [ $\text{m}^2/\text{sec}$ ] within the polymer,  $\delta_p$  denotes the polymer-side  
 85 boundary layer, which was so set to a maximum value of  $d_p/2$ . Diffusion coefficients in polyethylene  
 86 ( $D_p$ ) are about  $10^6$  smaller than in the gas phase ( $D_g$ ) and  $\log K_{pg}$  is greater than  $10^6$  (compounds with  
 87 molecular weights larger than Naphthalene). Since  $\delta_p$  is at maximum half of the thickness of the PE  
 88 sheet ( $< 40 \mu\text{m}$ ),  $\delta_g$  with a thickness of 1 mm will control mass transfer.

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91 **S.5. Calibration of the air-side boundary layer comparing different time spans**



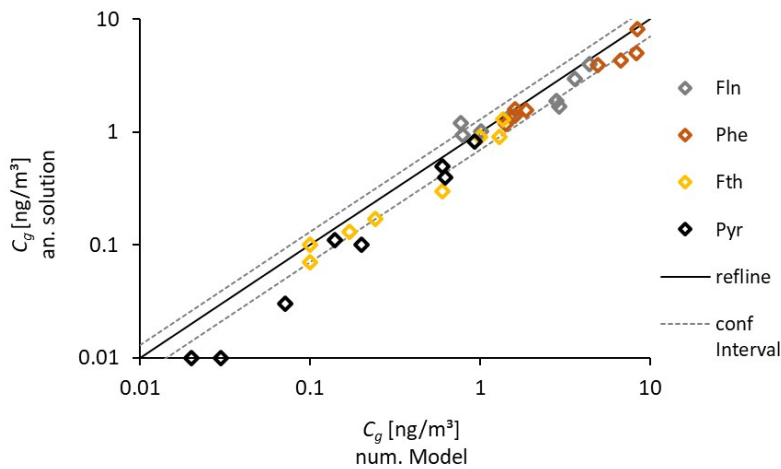
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93 Figure.S 7: PRC-loss during August 2016 measured at each location (crosses: red – Entringen, green – Poltringen, blue –  
94 Tailfingen) over different time periods; black solid line: numerical model fit based on measurements averaging over all  
95 locations. The air-side boundary layer thickness  $\delta_g$  was fitted for the average concentrations on PE, accounting for different  
96 time spans: (a) 3 days (b) 10 days (c) 30 days. Error bars refer to triplicate samples at each location and the grey area indicates  
97 the standard deviation of the predicted concentration.

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99 **S.6. Comparison of  $C_g$  estimated with the numerical model and the analytical solution**

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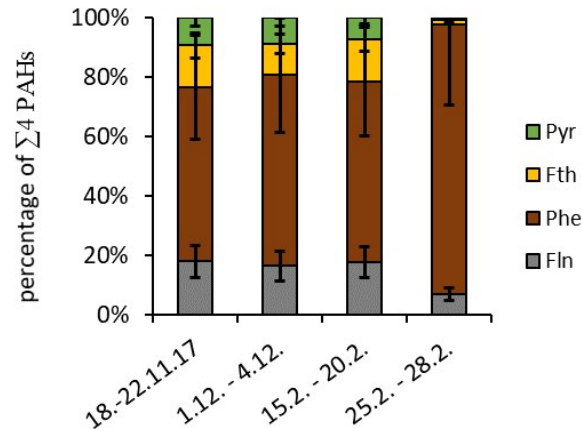


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102 Figure.S 8: Scatter plot comparing atmospheric concentrations of the four target PAHs determined with the analytical solution  
103 (y-axes) based on Eq. 1.4 to the numerical model (x-axes) based on Eq. 1.7. This comparison accounts for all seasonal  
104 samplings during the first year of monitoring. The solid line delineates the 1:1 reference line while dashed lines refer to a  
105 confidence interval of  $\pm 30\%$ .

106 **S.7. Distribution pattern of the representative PAHs determined with active samplers**

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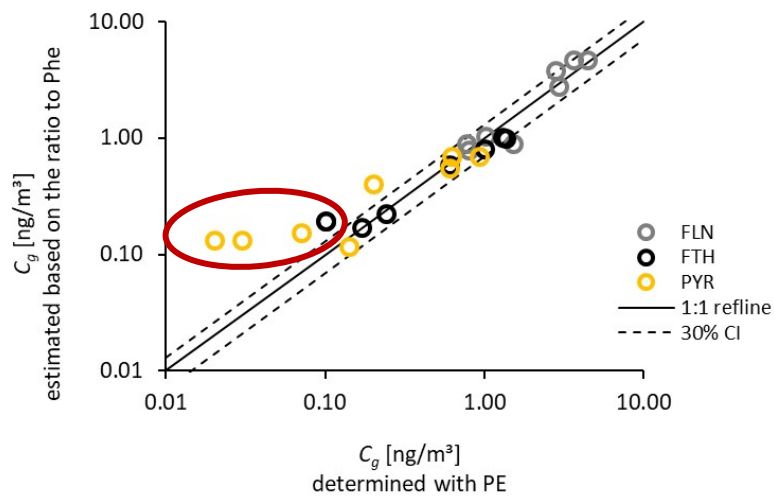
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109 *Figure.S 9: Percentage distribution of the four representative PAHs determined with active samplers. Error bars denote a*  
110 *standard deviation of 30%. The different pattern observed for the last sampling (25-28<sup>th</sup> of February 2018) is probably caused*  
111 *by the lack of particles sampled on glass fibre filters during this period.*

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113 **S.8.  $C_g$  of Fln, Fth and Pyr estimated based on their ratio to Phe**

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116 *Figure.S 10: Atmospheric concentrations of Fln, Fth and Pyr for all seasons and all locations calculated from Phe and the*  
117 *average distribution pattern. The red circle denotes outliers of Pyr during the warm periods as well as Fth during August 2017.*