

**Transition Metal Doped MgO Nanoparticles for Nutrient Recycling: An Alternate Mg
Source for Struvite Synthesis from Wastewater**

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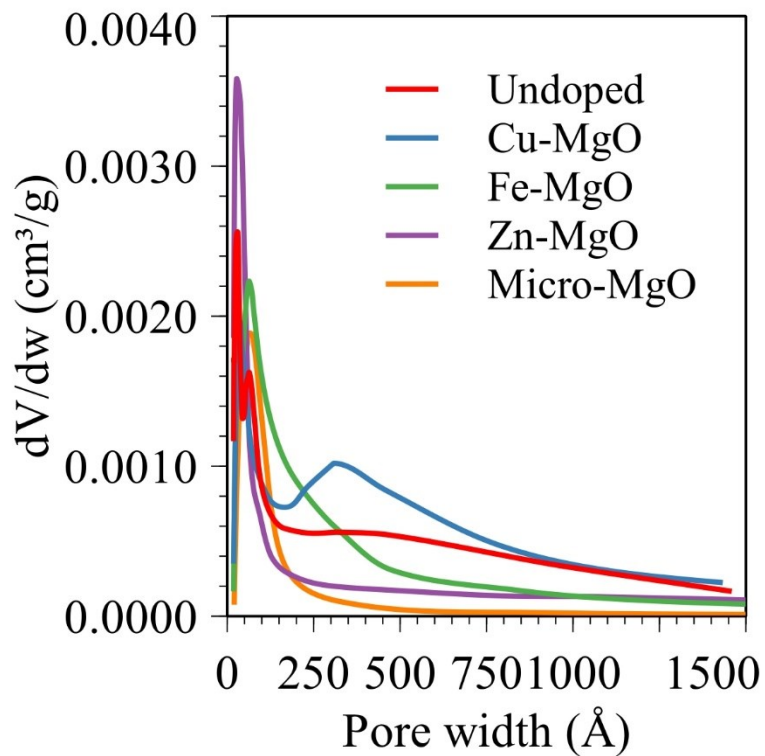


Figure S1. Pore size distributions for MgO NPs and microcrystalline MgO.

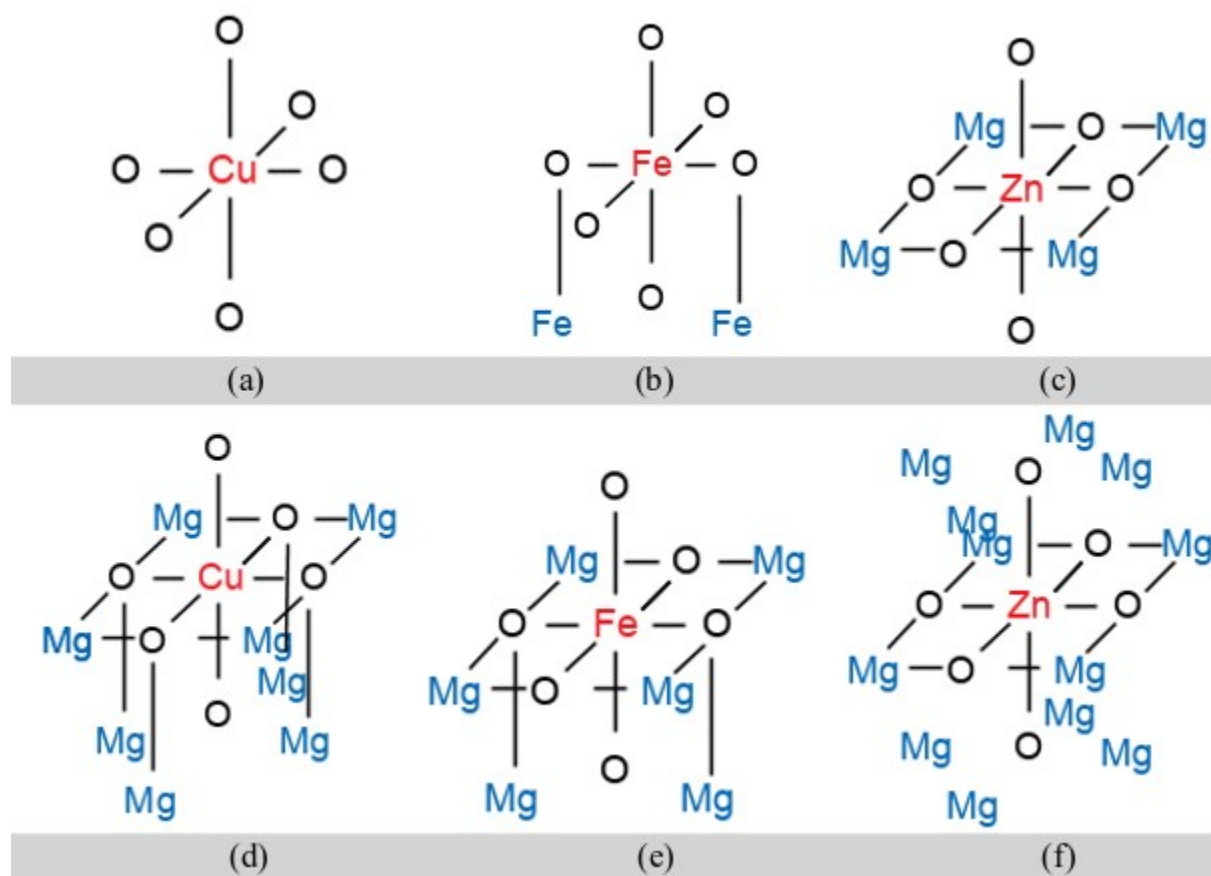


Figure S2. Models used for EXAFS fitting

Table S1. Adsorption kinetic model fits for undoped and doped MgO with varying temperatures and time ranges.

Model	Sample Name	Time Range (min)	Temperature (°C)	Parameters		R ²	
Pseudo-second Order	Undoped	0-35	25	q _e (mg g ⁻¹)	k ₂ (g mg ⁻¹ min ⁻¹)	0.992	
			35	1666.67	9.00E-04	0.998	
			45	1428.57	7.00E-04	0.999	
	Cu-MgO		25	1428.57	1.75E-04	0.973	
			35	1428.57	2.33E-04	0.983	
			45	1250.00	3.20E-04	0.989	
	Fe-MgO		25	1666.67	2.43E-05	0.289	
			35	3333.33	4.48E-06	0.046	
			45	1666.67	1.90E-04	0.301	
	Pseudo-second order		Zn-MgO	25	1111.11	1.45E-04	0.894
				35	5000.00	1.06E-06	0.010
				45	1250.00	3.20E-04	0.989
Undoped		25	1428.57	2.45E-03	0.971		
		35	1250.00	2.13E-03	0.964		
		45	1111.11	2.70E-03	0.971		
Cu-MgO	833.33	1.03E-03	0.806				
	Fe-MgO	77.52	6.66E-02	0.102			
		Zn-MgO	217.39	8.46E-03	0.246		
Intraparticle diffusion			0-5	25	k _p	C	
	Undoped				618.25	31.928	0.988
	Cu-MgO	354.72			7.064	0.998	
	Fe-MgO	134.40			30.435	0.815	
	Zn-MgO	142.53			6.908	0.990	
Microcrystalline MgO	177.43	64.766	0.884				

Table S2. Structural parameters (obtained from the fittings of K-edge EXAFS data) of the first two coordination shells around the dopant metal atom centers. The errors for R and σ^2 are within parentheses.

Sample name	Bond	N	R(Å)	$\sigma^2(\text{Å}^2)$
Cu-MgO	Cu-O (eq.)	4	2.03 (.01)	.0080 (.0013)
	Cu-O (ax.)	2	2.31 (.03)	.0081 (.0035)
	Cu-Mg(Cu)	8	2.96 (.02)	.0095 (.0011)
Fe-MgO	Fe-O	6	1.95 (.0008)	.0113 (.0006)
	Fe-Mg(Fe)	6	2.98 (.011)	.0087 (.0007)
Zn-MgO	Zn-O	6	2.12 (.015)	.0075 (.0018)
	Zn-Mg(Zn)	12	3.00 (.015)	.0066 (.0011)
Cu-Struvite	Cu-O (eq.)	4	1.96 (.010)	.0058 (.0005)
	Cu-O (ax.)	2	2.3 (0.1)	.035 (.016)
Fe-Struvite	Fe-O	6	1.97 (.004)	.0093 (.0002)
	Fe-Mg(Fe)	2	3.05 (.007)	.0127 (.0009)
Zn-Struvite	Zn-O	6	1.98 (.024)	.0123 (.0014)
	Zn-Mg(Zn)	4	2.95 (.030)	.0101 (.0027)

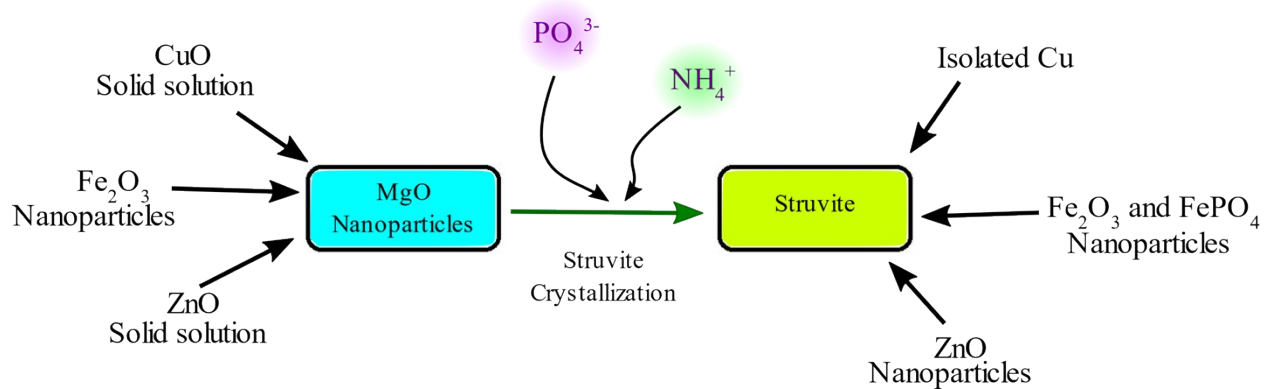


Figure S3. Summary of reaction process and fate of the dopants in MgO.