

Electronic Supplementary Information

Fabrication of 3D hierarchical flower-like δ -MnO₂@COF nanocomposites for the efficient and ultra-fast removal of UO₂²⁺ ion from aqueous solution

Xin Zhong^a, Zhipeng Lu^a, Wen Liang^a, Xiaojie Guo^{b*}, Baowei Hu^{a*}

^a School of Life Science, Shaoxing University, Huancheng West Road 508, Shaoxing 312000, P.R. China

^b College of Materials and Environmental Engineering, Hangzhou Dianzi University, Xiasha High Education Zone, Hangzhou, 310018, PR China

Corresponding author e-mail:

Baowei Hu: hbw@usx.edu.cn

Xiaojie Guo: xiaojie.guo@hdu.edu.cn

1. Chemicals

1, 3, 5-triformylphloroglucinol ($\geq 95\%$, Tp) was obtained from Chinese Academy of Science Jilin Yanshen Technology Company. Other chemicals were analytical reagents or better and supplied by Macklin Chemical Reagent Company. All reagents were used directly without further purification.

The crystallographic data of δ -MnO₂ acquired from American Mineralogist Crystal Structure Database.

2. Figures

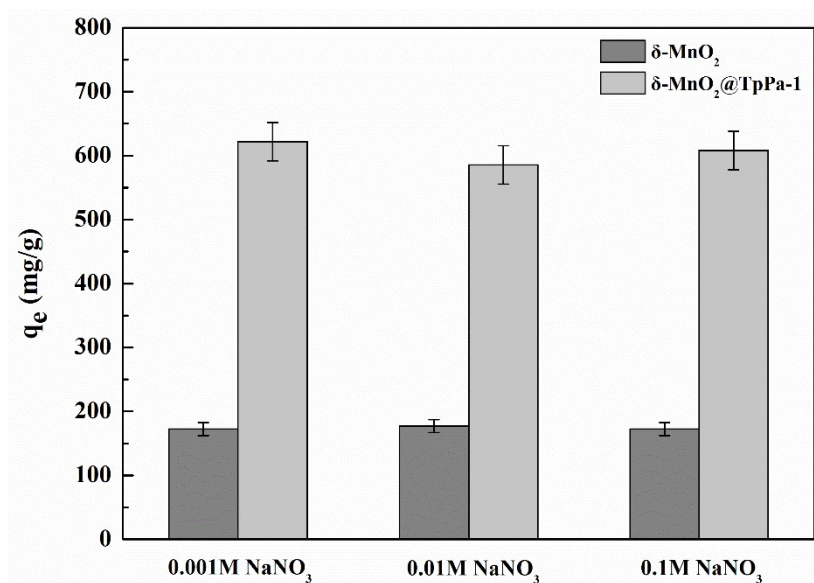


Fig. S1 The influence of ionic strength on the removal efficiency UO_2^{2+} ion adsorption onto δ -MnO₂ and δ -MnO₂@TpPa-1.

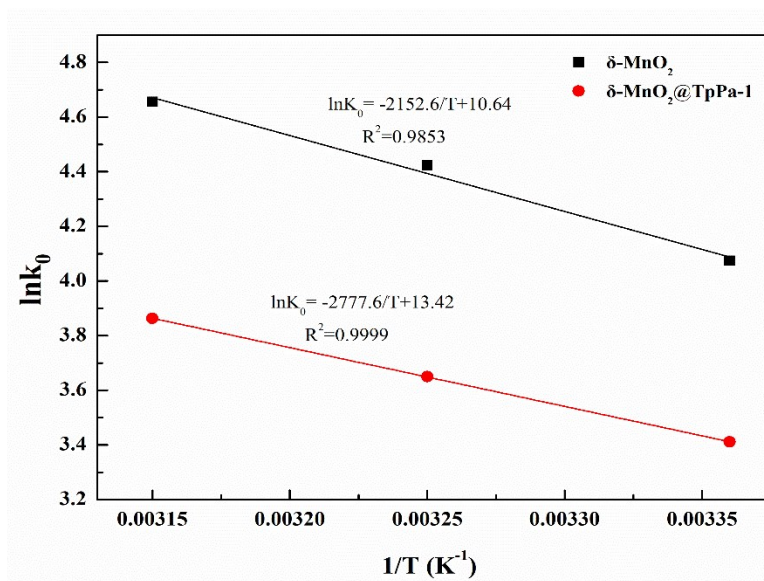


Fig. S2 The linear plots of $\ln K_0$ of UO_2^{2+} ion versus $1/T$ at three different temperatures ($T=298$ - 318 K).

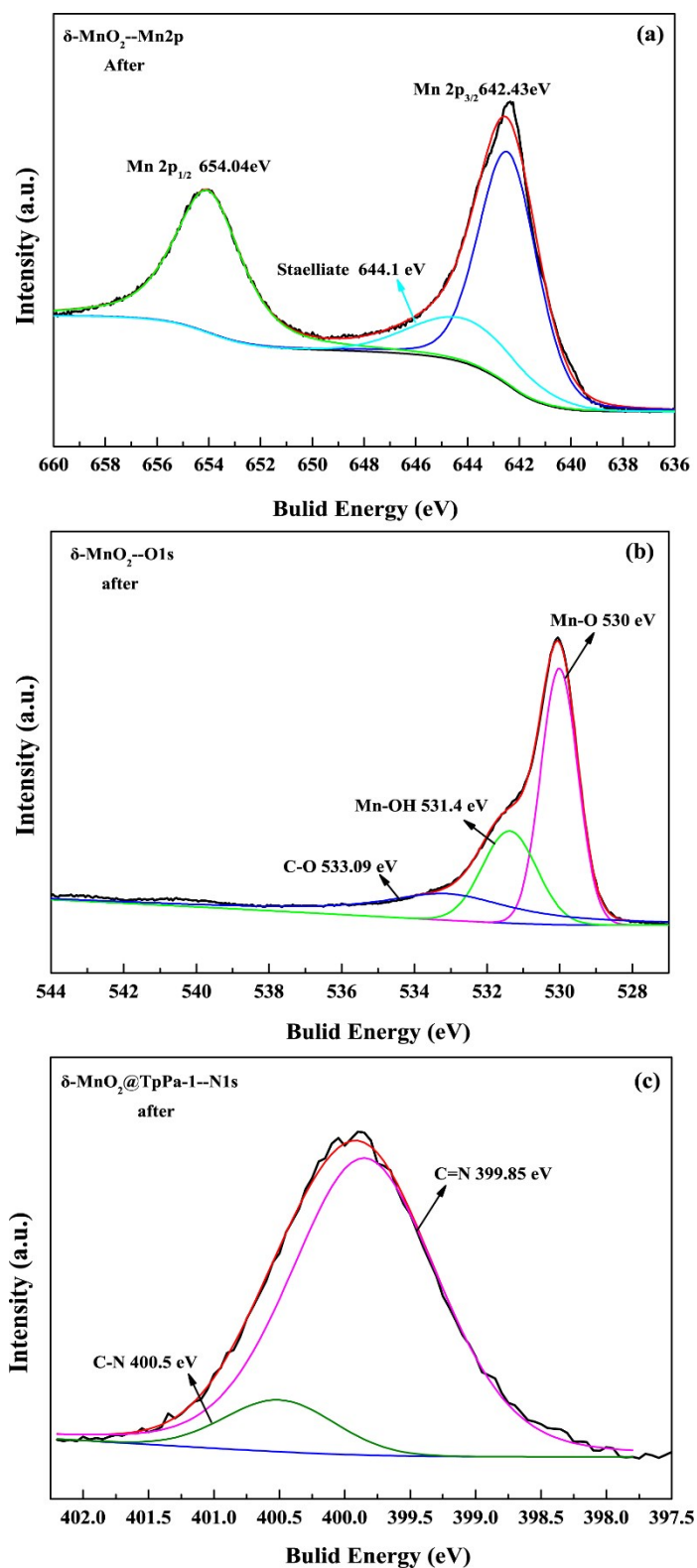


Fig. S3 XPS spectra of $\delta\text{-MnO}_2$ and $\delta\text{-MnO}_2\text{@TpPa-1}$: (a) $\delta\text{-MnO}_2\text{-Mn2p}$; (b) $\delta\text{-MnO}_2\text{-O1s}$; (c) $\delta\text{-MnO}_2\text{@TpPa-1}$ N 1s after UO_2^{2+} ion adsorption.