

Supplementary material

for

**Degradation of sulfachloropyridazine by UV-C/persulfate: kinetics, key factors,
degradation pathway**

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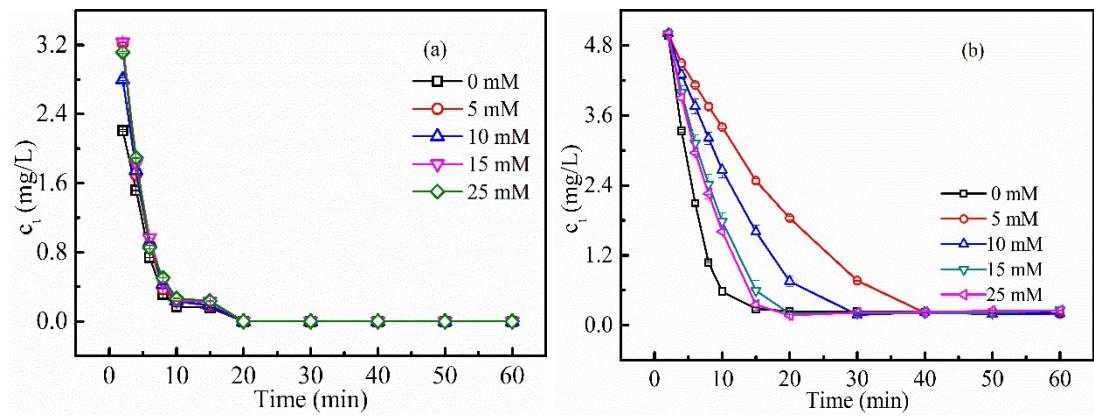


Fig. S1. Influences of SCP degradation in different concentrations of *TBA*(a)/*MeOH*(b). Condition: [SCP]=16.3 μM ; [BA]= 24.2 μM ; [PS]=1.0 mM; [H_2O_2]= 1.21 μM ; [TBA]= 10 mM; pH=6.3; 25 $^{\circ}\text{C}$; light intensity: 461 $\mu\text{W}/\text{cm}^2$.

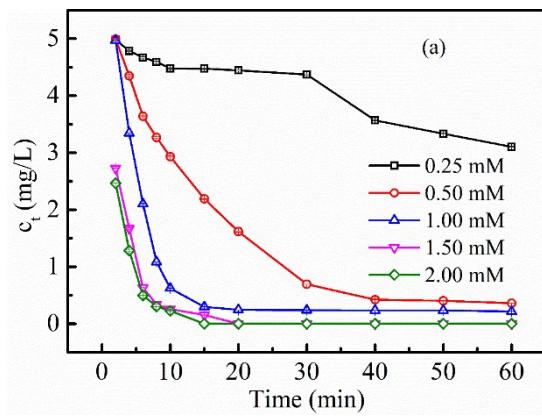


Fig. S2. Influences of SCP degradation in different initial PS dosages systems. Conditions: [SCP]=16.3 μM ; pH=6.3; 25 °C; light intensity: 461 $\mu\text{W}/\text{cm}^2$.

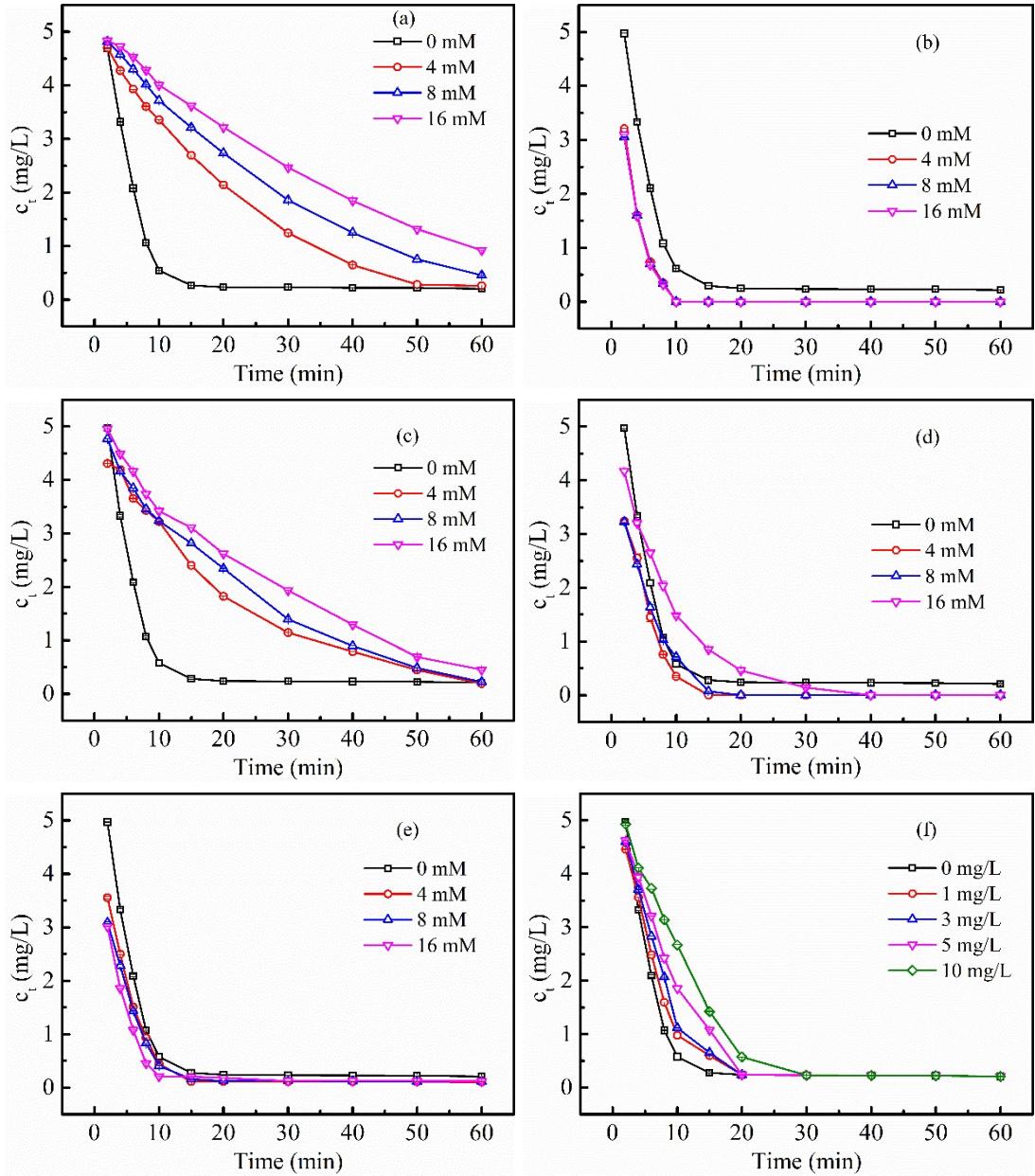
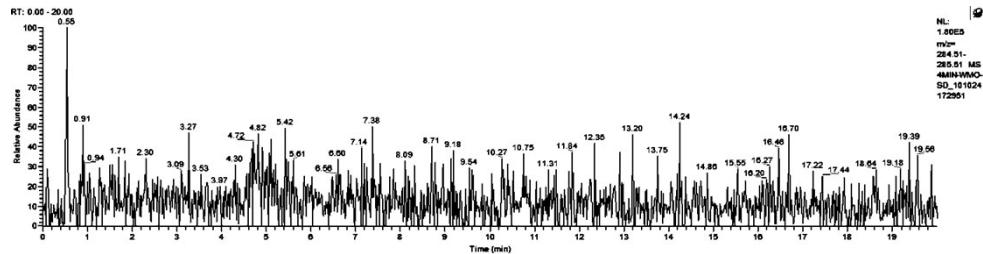
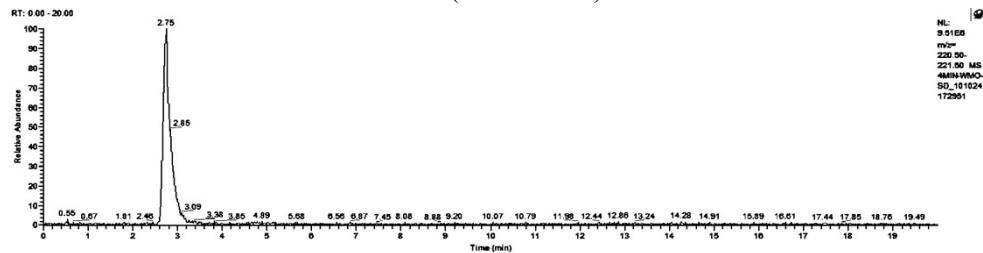


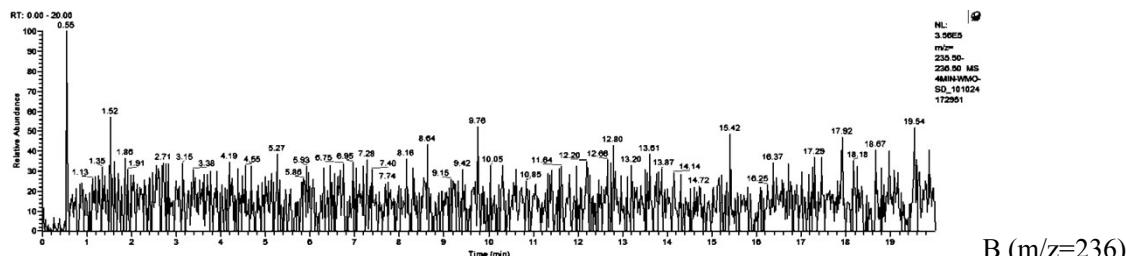
Fig. S3. Influences of HPO_4^{2-} (a), $H_2PO_4^-$ (b), HCO_3^- (c), NO_3^- (d), Cl^- (e) and NOM(f) on the degradation of SCP.



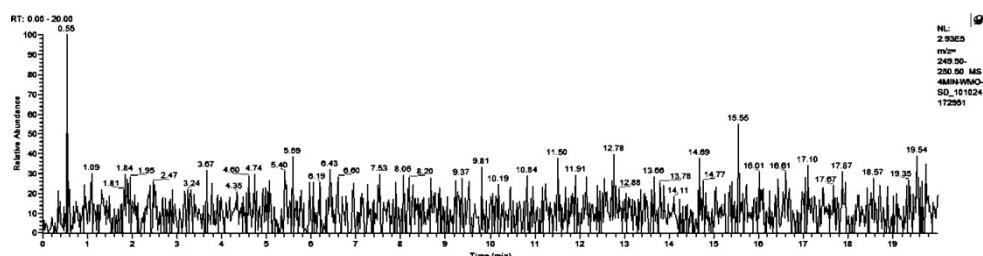
SCP (m/z=285.01)



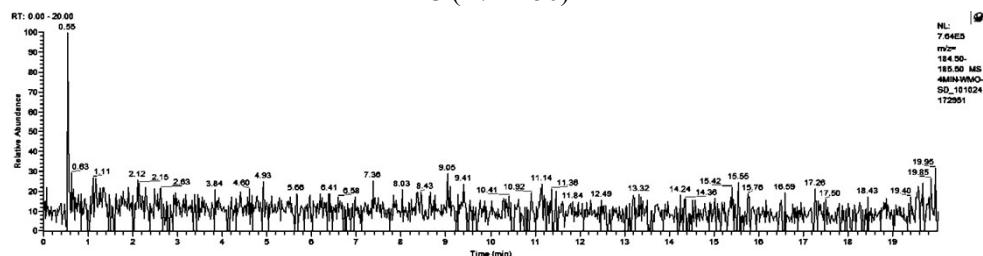
A (m/z=221)



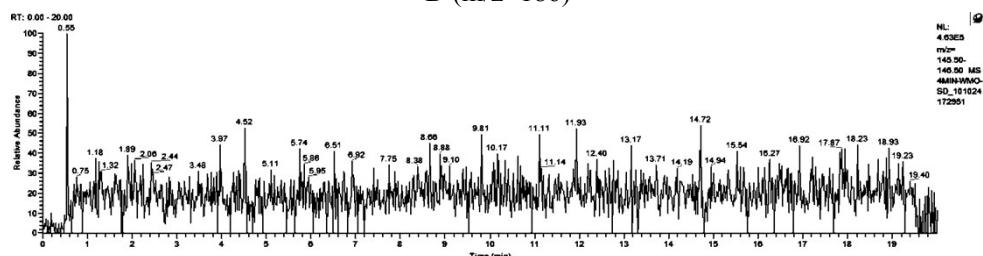
B (m/z=236)



C (m/z=250)



D (m/z=186)



E (m/z=146)

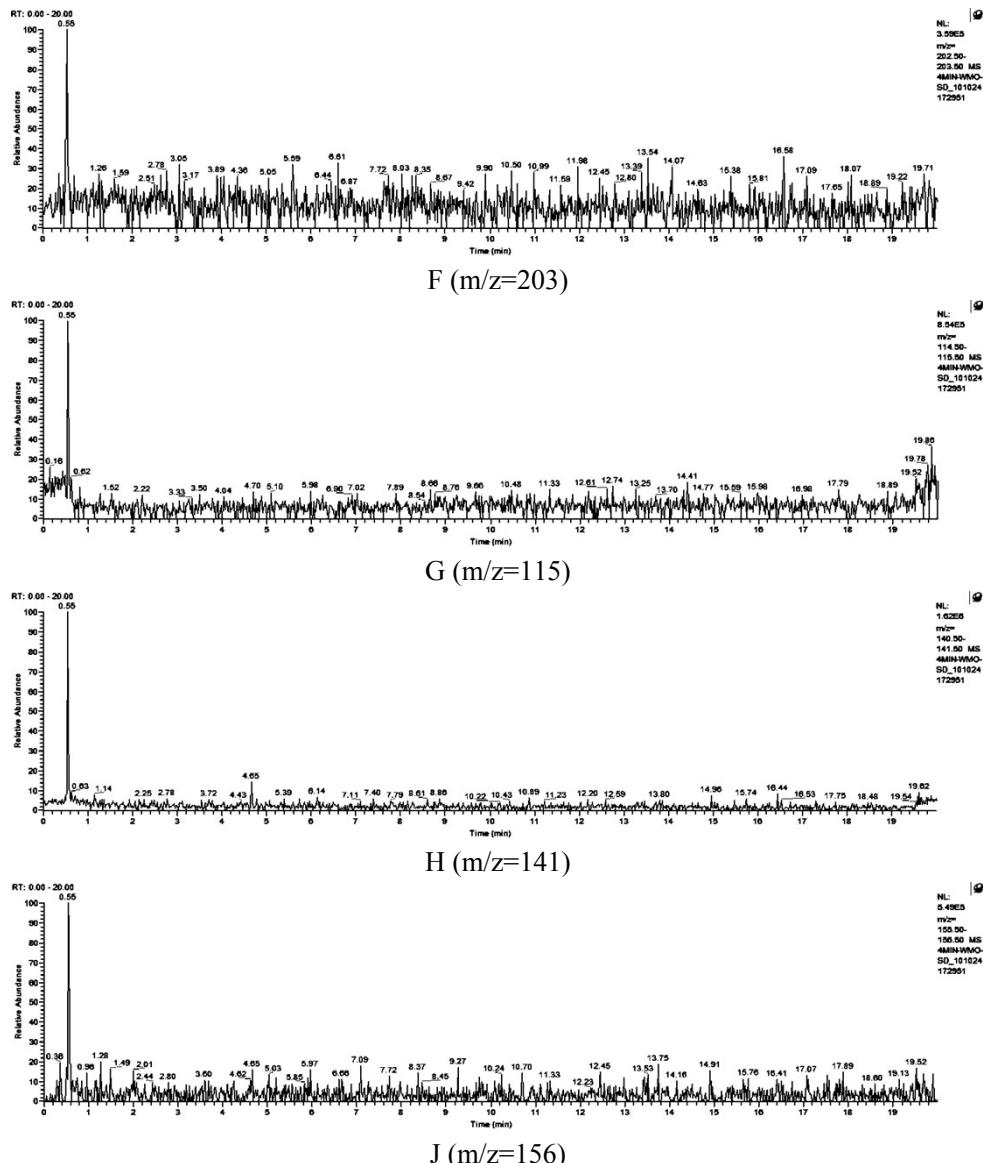


Fig. S4. Spectra of the parent ion and sub-ions

SCP $m/z=285.01$; A $m/z=221$; B $m/z=236$; C $m/z=250$; D $m/z=186$;

E $m/z=146$; F $m/z=203$; G $m/z=115$; H $m/z=141$; J $m/z=156$.