

Supporting Information

# A Tale of Two Metal Ions: Contrasting Behaviors of High Oxidation States of Cu and Mn in Bicarbonate-H<sub>2</sub>O<sub>2</sub> System

*Zhiwei Yang<sup>a,b</sup>, Xiaonan Tan<sup>a</sup>, Daojian Tang<sup>a</sup>, Jing Li<sup>\*b</sup>, Jiahai Ma<sup>\*a</sup>*

<sup>a</sup> School of Chemical Sciences, University of Chinese Academy of Sciences, Beijing 100049, P. R. China

<sup>b</sup> Chongqing Institute of Green and Intelligent Technology, Chinese Academy of Sciences, 100190, Beijing, P. R. China.

E-mail: [majia@ucas.ac.cn](mailto:majia@ucas.ac.cn)

**Table S1.** Comparison of the current AOPs for successful treatment for azo dyes in some recent literature.

Ref.	AOPs	100% Degradation Time	Azo dyes
This work	metal ions intermediated bicarbonate-H <sub>2</sub> O <sub>2</sub> system	10 min	congo red
1	Adsorption	15 min	methyl orange orange II congo red
2	Adsorption	60 min	reactive black 5 amaranth acid red 183
3	Synergistic Adsorption and Biodegradation	1400 min	orange II
4	Synergistic Adsorption and Photocatalysis	120 min	congo red
5	TiO <sub>2</sub> Photocatalysis	180 min	congo red
6	Fe–Cr Codoped BaTiO <sub>3</sub> Photocatalysis	100 min	methyl orange
7	Biodegradation	120 h	methyl orange Ponceau S Red
8	Bioelectrochemical System Integrated with a Membrane Biofilm Reactor	days	acid orange
9	Co Fenton-like system	50 min	sunset yellow
10	ozonation	30 min	methyl orange
11	Catalytic reduction by Ag and Au nanoparticles stabilized on graphene oxide functionalized with PAMAM dendrimers)	3 min	methyl orange congo red
12	Reduction by mediated Sulfide	50 min	methyl orange

## References.

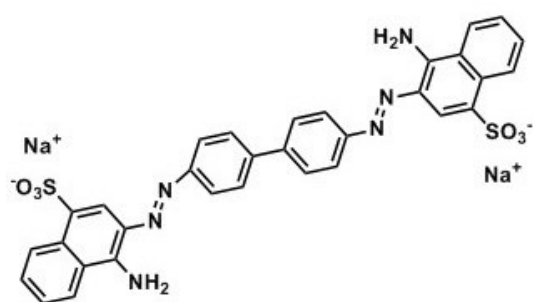
1. M. Chaaban, H. El-Rassy, Nickel–Aluminum Oxide Aerogels: Super-adsorbents for Azo Dyes for Water Remediation. *ACS Omega*. 5 (2020) 27401-27412.
2. M. Vall, M. Strømme, O. Cheung, Amine-Modified Mesoporous Magnesium Carbonate as an Effective Adsorbent for Azo Dyes. *ACS Omega*. 4 (2019) 2973-2979.
3. M. Ahmad, M. Yousaf, A. Nasir, I. Ah. Bhatti, A. Mahmood, X. Fang, X. Jian, K. Kalantar-Zadeh, N. Mahmood, Porous Eleocharis@MnPE Layered Hybrid for Synergistic Adsorption and Catalytic Biodegradation of Toxic Azo Dyes from Industrial Wastewater. *Environ. Sci. Technol.* 53 (2019) 2161–2170.
4. X. Liu, T. Zhu, Y. Gong, Efficient Removal of Azo-Dyes in Aqueous Solution by CeB<sub>6</sub> Nanocrystals. *ACS Appl. Nano Mater.* 2 (2019) 5704–5712.
5. J. A. Bumpus, J. Tricker, K. Andrzejewski, H. Rhoads, M. Tatarko, Remediation of Water Contaminated with an Azo Dye: An Undergraduate Laboratory Experiment Utilizing an Inexpensive Photocatalytic Reactor. *J. Chem. Edu.* 76 (1999) 1680-1683.
6. I. C. Amaechi, A. H. Youssef, D. Rawach, J. P. Claverie, S. Sun, A. Ruediger, Ferroelectric Fe–Cr Codoped BaTiO<sub>3</sub> Nanoparticles for the Photocatalytic Oxidation of Azo Dyes. *ACS Appl. Nano Mater.* 2 (2019) 2890–2901.
7. D. Baena-Baldiris, A. Montes-Robledo, R. Baldiris-Avila, *Franconibacter sp.*, IMS: A New Strain in Decolorization and Degradation of Azo Dyes Ponceau S Red and Methyl Orange. *ACS Omega*. 5 (2020) 28146-28157.
8. Y. Pan, T. Zhu, Z. He, Enhanced Removal of Azo Dye by a Bioelectrochemical System Integrated with a Membrane Biofilm Reactor. *Ind. Eng. Chem. Res.* 57 (2018) 16433–16441.
9. Z. Li, L. Wang, M. Tian, Z. Li, Z. Yuan, C. Lu. Tris–Co(II)–H<sub>2</sub>O<sub>2</sub> System-Mediated Durative Hydroxyl Radical Generation for Efficient Anionic Azo Dye Degradation by Integrating Electrostatic Attraction. *ACS Omega*. 4 (2019) 21704-21711.
10. C. Feng , P. Diao, Nickel foam supported NiFe<sub>2</sub>O<sub>4</sub>-NiO hybrid: A novel 3D porous catalyst for efficient heterogeneous catalytic ozonation of azo dye and nitrobenzene. *Applied Surface Science* 541 (2021) 148683.

11. R. Rajesh, S. S. Kumarb, R. Venkatesan, Efficient degradation of azo dyes using Ag and Au nanoparticles stabilized on graphene oxide functionalized with PAMAM dendrimers. *New J. Chem.* 38 (2014) 1551-1558.
12. H. Zhao, S. Huang, W. Xu, Y. Wang, Y. Wang, C. He, Y. Mu, Undiscovered Mechanism for Pyrogenic Carbonaceous Matter-Mediated Abiotic Transformation of Azo Dyes by Sulfide. *Environ. Sci. Technol.* 53 (2019) 4397–4405.

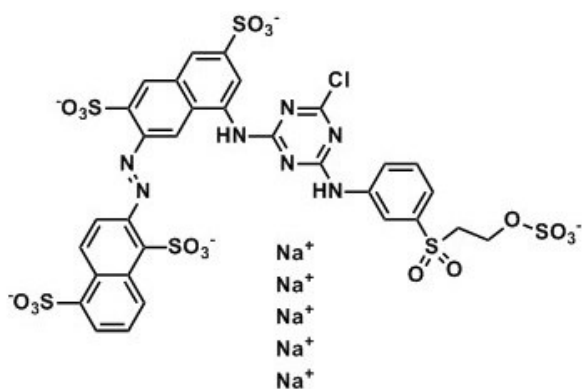
Table S2. Characteristic peak wavelength of selected dyes. Conditions: dyes 0.02 mM, metal ions

dyes	CR	MBH	MO	RB	RhB	RR
Characteristic peak (nm)	494	665	465	545	554	543

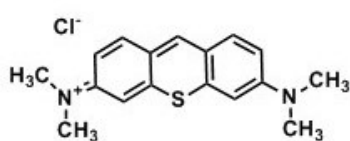
0.1 mM,  $\text{HCO}_3^-$  40 mM,  $\text{H}_2\text{O}_2$  10 mM.



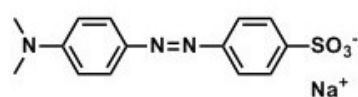
**CR**



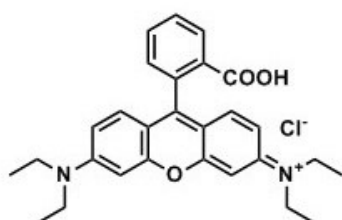
**RR**



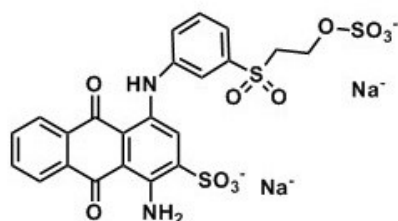
**MBH**



**MO**



**RhB**



**RB**

Scheme S1. Structural formula of selected six dyes.

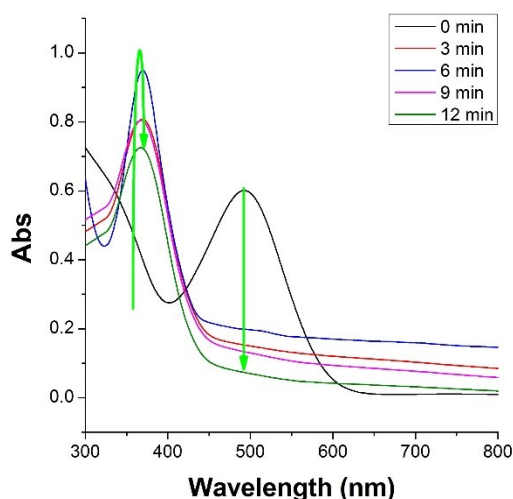


Figure S1. The Uv-Vis spectrum of CR during degradation. Conditions: CR 0.02 mM,  $\text{Cu}^{2+}$  0.1 mM,  $\text{HCO}_3^-$  40 mM,  $\text{H}_2\text{O}_2$  10 mM.

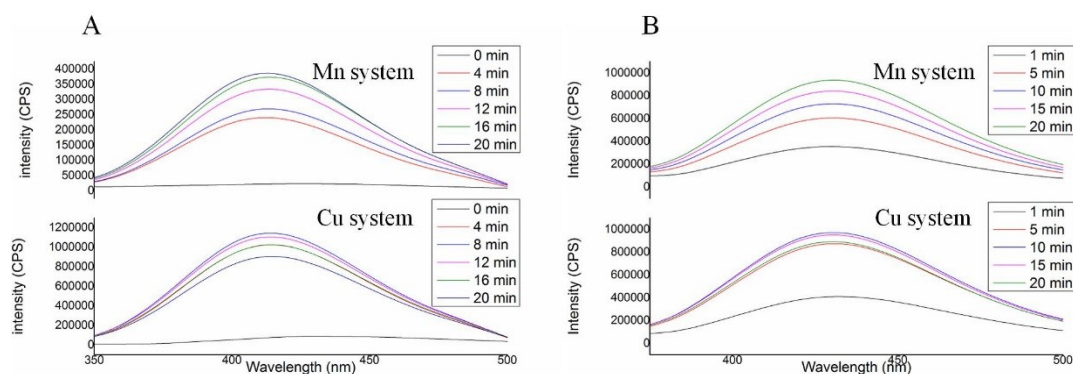


Figure S2. Fluorescence spectra analysis of (A) BA as fluorescent probe and (B) TA as fluorescent probe. Conditions:  $\text{Cu}^{2+}$  0.1 mM,  $\text{Mn}^{2+}$  0.1 mM,  $\text{HCO}_3^-$  40 mM,  $\text{H}_2\text{O}_2$  10 mM, BA 0.3 mM, TA 0.3mM. excitation wavelength 287 nm.

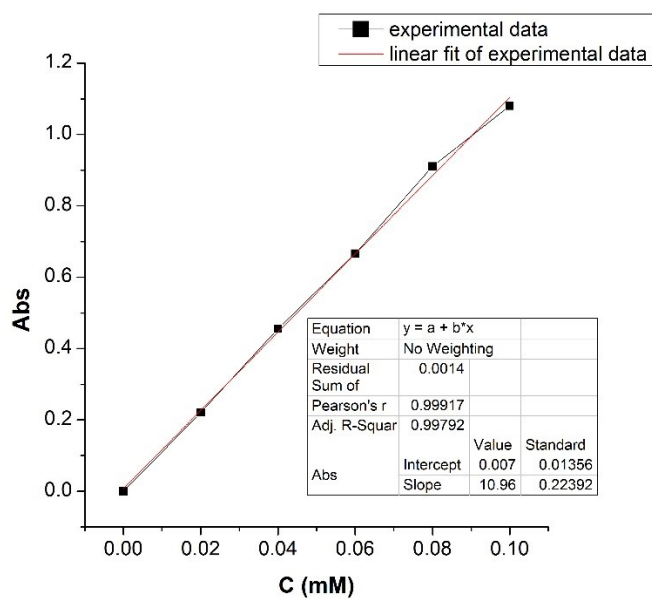


Figure S3. The linear relationship between Abs and concentration of Cu ( I ) by bathocuproine.

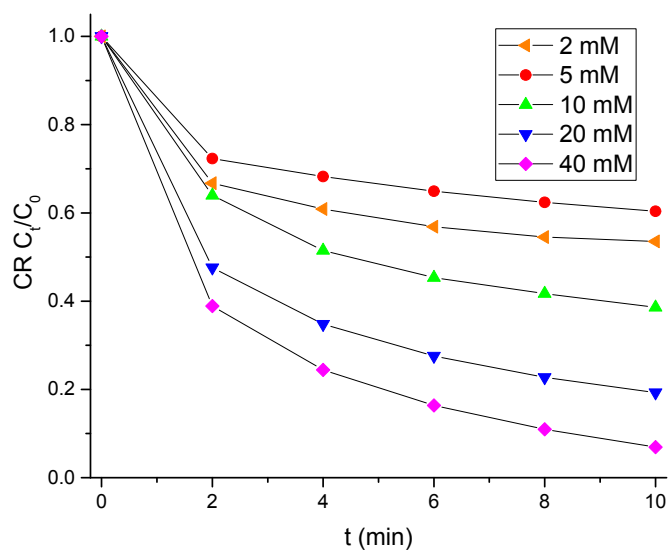


Figure S4. The dosage effect of o bicarbonate. Conditions: CR 0.02 mM,  $Cu^{2+}$  0.1 mM,  $H_2O_2$  10 mM.

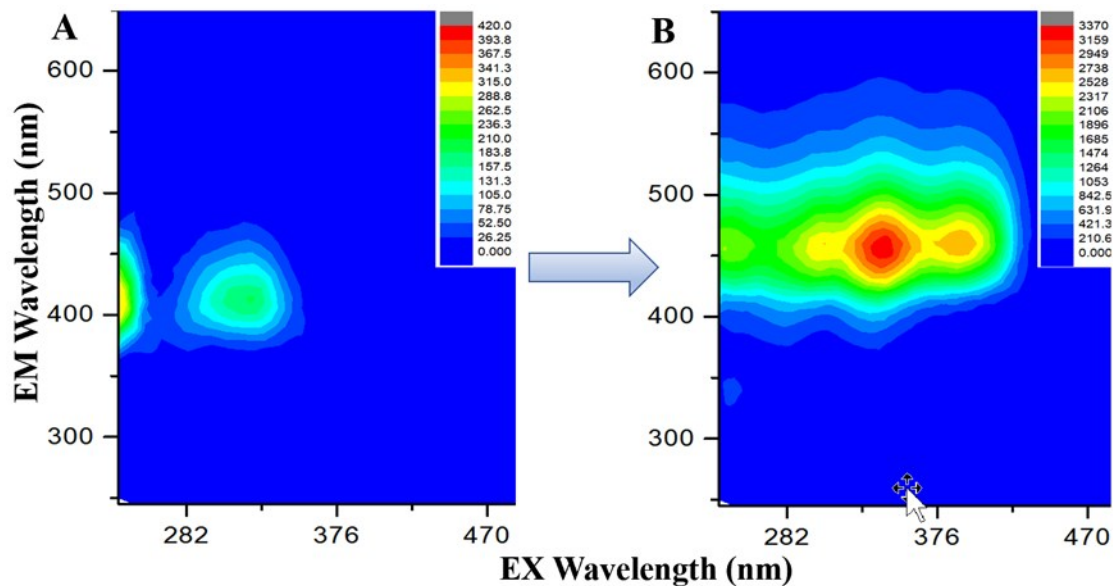


Figure S5. The 3D fluorescence spectra of CR. A: before the reaction; B: after the reaction. Conditions: CR 0.02 mM,  $Cu^{2+}$  0.1 mM,  $HCO_3^-$  40 mM,  $H_2O_2$  10 mM.