Supplementary Information

L-Arginine/L-Lysine functionalized chitosan-casein core-shell and pH-responsive nanoparticles: Fabrication, characterization and bioavailability enhancement of hydrophobic and hydrophilic bioactive compounds

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Figure S1. Schematic diagram for synthesis of Arg-CS and Lys-CS. FTIR spectra of Arginine, Lysine, Arginine-Chitosan, Lysine-Chitosan, and chitosan.

Figure S1 exhibits the characteristic FTIR spectra of CS and the arginine (Arg) or lysine (Lys) functionalized CS conjugate. The peak at 1647 cm⁻¹ (amid I) and 1600 cm⁻¹ (amid II) in CS all shift to 1630 cm⁻¹ in Arg or Lys functionalized CS, which could be ascribed to the characteristic carbonyl stretching (C=O) owing to the formation of amide bonding¹. The new peak in Arg or Lys functionalized CS appears at 1520 cm⁻¹ could be attributed to the bending vibration and stretching vibration of secondary amide. Furthermore, the peak between 3300 and 3450 cm⁻¹ can be attributed to the vibration of hydroxyl, amino and amide groups in CS². Intermolecular and intramolecular hydrogen bonding between these groups in CS

greatly limits its solubility. However, the peak becomes much narrower and weaker in Arg or Lys functionalized CS, which suggested that their conjugation to CS destroys the hydrogen bonds in CS then increases its solubility³. All these results indicates the linkage of Arg or Lys to CS.



Figure S2. Sample images of Arg-CS-CA NPs (a) and Lys-CS-CA NPs (b) under different pH values.



Figure S3. Size distribution of Arg-CS-CA NPs (a, b) and Lys-CS-CA NPs (c, d) at pH 2 and 3, respectively.



Figure S4. Interaction between Arg-CS-CA (a) and Lys-CS-CA (b) NPs and mucin at pH 5.



Figure S5. Sample images of curcumin entrapped in Arg-CS-CA NPs (a) and Lys-CS-CA NPs (b) under different pH values (containing 25% ethanol).



Figure S6. Fluorescence spectra of casein due to casein-EWDP (a)/curcumin (b) interaction.

Reference:

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