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Figure S1. X-band EPR spectra of the semiquinone anion radicals of the quercetin and its studied glycosides (1 mM), at room temperature in water-methanol solution at different methanol content (indicated in volume %).



Figure S2. X-band EPR spectra of the semiquinone anion radicals of the quercetin and its studied glycosides (1 mM), at room temperature in water-methanol, arranged as to emphasize linewidth and hyperfine coupling constants evolution as methanol concentration dependence.

Table S1. Hyperfine coupling constants (a_{Hn}) of the three protons (see Figure 3 in main text)and linewidth (LW) for quercetin and its glycosides as function of methanol content. Thevalues are determined from the experimental spectra using WinSim software.

			measured in Gauss				
%MeOH (in water)	Sample name	g value	$a_{\rm H1}$	a _{H2}	a _{H3}	LW	
40		2.0049	3.0776	0.7880	1.4992	0.3848	
50		2.0047	3.2040	0.8000	1.4328	0.3512	
60	Ouercetin	2.0048	3.5120	0.7504	0.9800	0.2816	
75	2	2.0049	3.5280	0.7728	0.9408	0.2656	
80		2.0049	3.5216	0.7888	0.9672	0.3032	
90		2.0049	3.5144	0.7560	1.0656	0.5048	
100		2.0048	3.4344	0.7656	1.1320	0.7400	
40		2.0050	2.8848	1.1320	0.9576	0.3280	
50		2.0050	2.8672	1.1568	0.9728	0.3536	
60	Hyperoside	2.0050	2.8624	1.1704	0.9736	0.3816	
75	11990100100	2.0050	2.8616	1.2208	0.9856	0.5496	
80		2.0048	2.8608	1.2560	0.9904	0.6800	
90		2.0049	2.7952	1.3104	1.1008	0.8112	
100		2.0049	2.6936	1.2632	0.8872	1.3656	
40		2.0050	2.8416	1.0720	1.1544	0.3032	
50		2.0050	2.8336	1.1040	1.1264	0.3112	
60	Ouercitrin	2.0050	2.8168	1.1016	1.2192	0.4312	
75	2	2.0049	2.8136	1.1128	1.2160	0.4752	
80		2.0050	2.8040	1.0920	1.2544	0.5280	
90		2.0050	2.7880	1.0584	1.3560	0.6280	
100		2.0050	2.8160	1.1520	1.2880	1.0808	
40		2.0050	2.8984	1.0872	0.9408	0.3568	
50		2.0049	2.8936	1.1032	0.9400	0.3824	
60	Rutin	2.0050	2.8840	1.1464	0.9464	0.4272	
75		2.0050	2.8984	1.1944	0.9344	0.4616	
80		2.0050	2.8752	1.1528	0.9384	0.4040	
90		2.0050	2.9096	1.1984	1.0008	0.8080	
100		2.0050	2.7464	1.2568	0.8088	1.2280	
40		2.0050	2.8928	0.9592	1.1408	0.2824	
50		2.0050	2.8848	0.9656	1.1792	0.3296	
60	Isoquercitrin	2.0050	2.8616	0.9808	1.1992	0.4008	
75	1	2.0050	2.8624	0.9904	1.2352	0.4424	
80		2.0050	2.8072	1.0064	1.3248	0.5648	
90		2.0050	2.8416	0.9744	1.2288	1.0472	
100		2.0049	2.7616	0.8208	1.2656	1.3104	

Compound	Torsion angle*	a _{H1}	a _{H2}	a _{H3}
Quercetin	-	4.188	0.738	0.927
Isoquercitrin	-18.2	1.980	1.732	0.939
Hyperoside	18.6	2.395	1.553	0.653
Quercitrin	-26.8	2.848	1.202	2.047
Rutin	-34	1.514	0.668	0.634

Table S2. Numerically computed parameters as described in main text. Hyperfine coupling constants (a_{Hn}) in G.

*as determined from the optimised structures shown in Figure 5 (main text).



Figure S3. *Left*: *Titration curves for quercetin, kaempferol and luteolin used for pKa determination. Right*: *Illustrative spectral changes of quercetin obtained in pH titration.*



Figure S4. pKa determination using pH titration via molecular absorption spectrophotometry following their spectral profile change at the indicated wavelengths – full circles indicating the ratio of the absorbances in the right axis title and squares indicating the ratio of the absorbances in the left axis title. The pKa values were determined after applying a sigmodal function for fitting (fitting curves indicated in magenta).



Figure S5. Modelled structures of studied compounds, in completed deprotonated state presented in three perspectives.



Figure S6. *Lipophilicity determination from logk linear dependence on methanol content () in isocratic HPLC approach as described in the main text.*



Figure S7. *Left-DPPH bleaching assay at different methanol concentrations. Right- ABTS bleaching assay of the studied compound.*



Figure S8. *Kinetic profile of liposome peroxidation, measured at 234 nm, and their inhibition by the studied glycosides in a concentration-dependent manner.*



Figure S9. Antioxidant capacity evaluation using inhibition liposomes peroxidation assay: lag time linear dependence on compounds concentration. Lag time values correspond to the inflection points in the kinetic profiles from Figure 8, main text.



Figure S10. Molecular absorption spectra recorded upon treatment of the investigated glycosides and quercetin with AlCl₃ as described in main text. Inset: Zoomed in for the spectral maxima.