

## Supporting Information

### Degradation of Poly(ethylene terephthalate) Catalyzed by Metal-free Choline-Based Ionic Liquids

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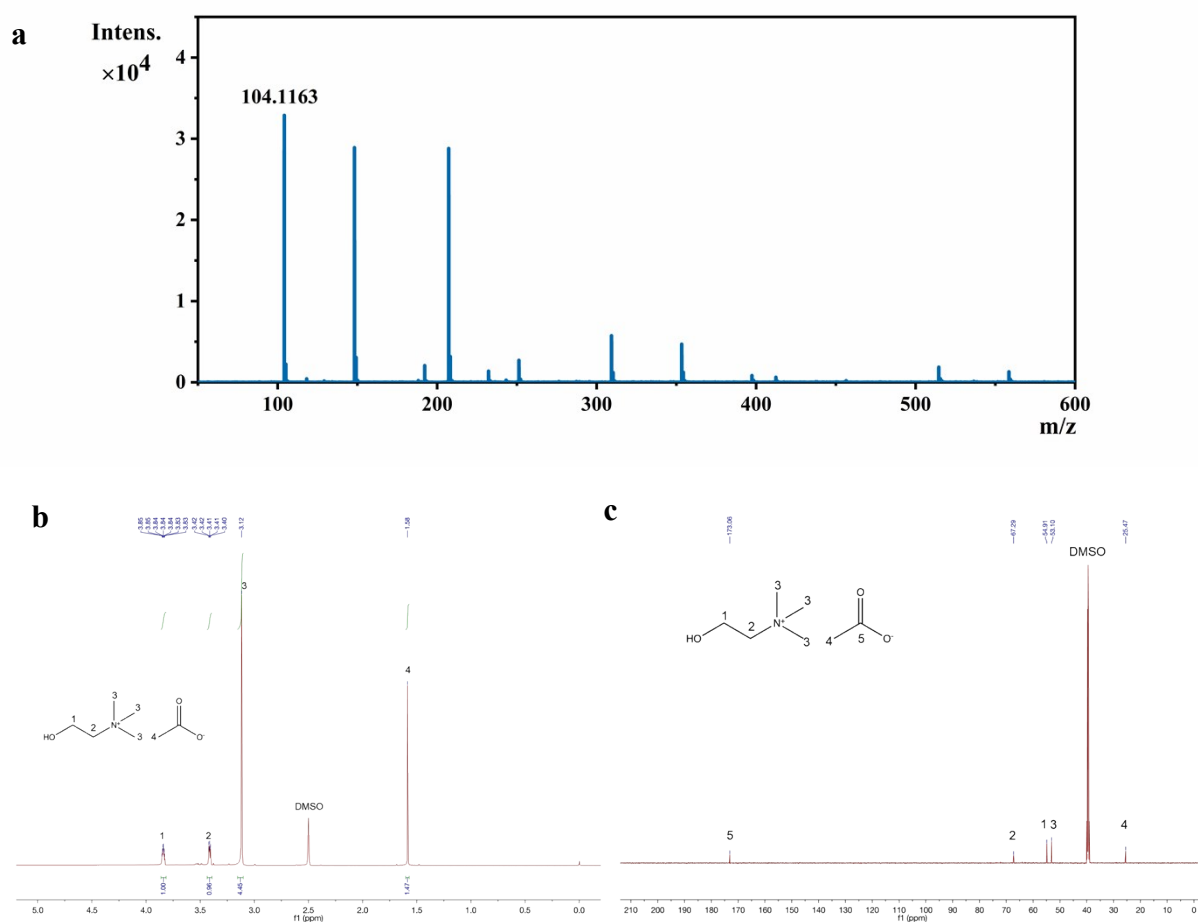
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## Characterizations of IL

ESI-MS, NMR and FT-IR, as shown in Figure S1, were carried out to identify the structures of the prepared ILs. The results of the chosen catalyst, [Ch][OAc], are mainly displayed here.



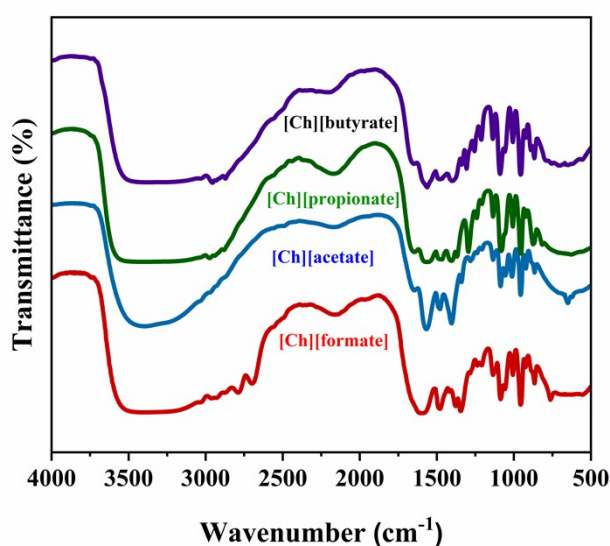
**Figure S1.** (a) ESI-MS patterns, (b)  $^1\text{H}$  NMR spectrum and (c)  $^{13}\text{C}$  NMR spectrum of [Ch][OAc].

The molecular ion peak at  $m/z = 104.1163$  of ESI-MS patterns in Figure S1. (a) is consistent with the molecular weight of  $[\text{Ch}]^+$ , which demonstrates the presence of  $[\text{Ch}]^+$  in the synthesized IL.

The  $^1\text{H}$  NMR and  $^{13}\text{C}$  NMR spectra were reproduced in Figure S1. (b) and (c). The results are as follows:

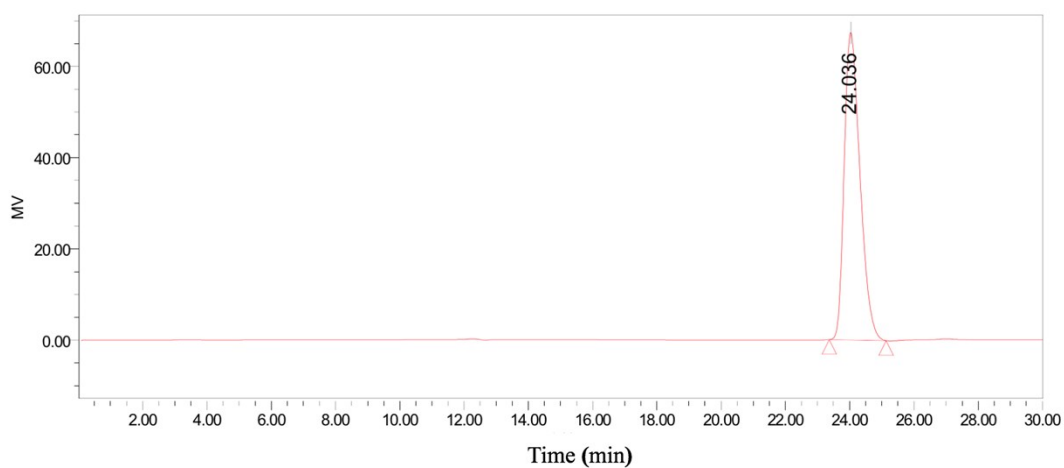
$^1\text{H}$  NMR ( $d_6$ -DMSO, 600MHz):  $\delta=3.84$  (m, 2H,  $\text{CH}_2\text{OH}$ ), 3.41 (m, 2H,  $\text{CH}_2\text{N}^+$ ), 3.12 (s, 9H,  $(\text{CH}_3)_3\text{N}^+$ ), 1.58 (s, 3H,  $\text{CH}_3\text{CO}$ );

$^{13}\text{C}$  NMR ( $d_6$ -DMSO, 500MHz):  $\delta=181.20$  ( $\text{COO}^-$ ), 67.40 ( $\text{CH}_2\text{N}^+$ ), 55.60 ( $\text{CH}_2\text{OH}$ ), 53.85 ( $(\text{CH}_3)_3\text{N}^+$ ), 23.26 ( $\text{CH}_3\text{CO}$ ).

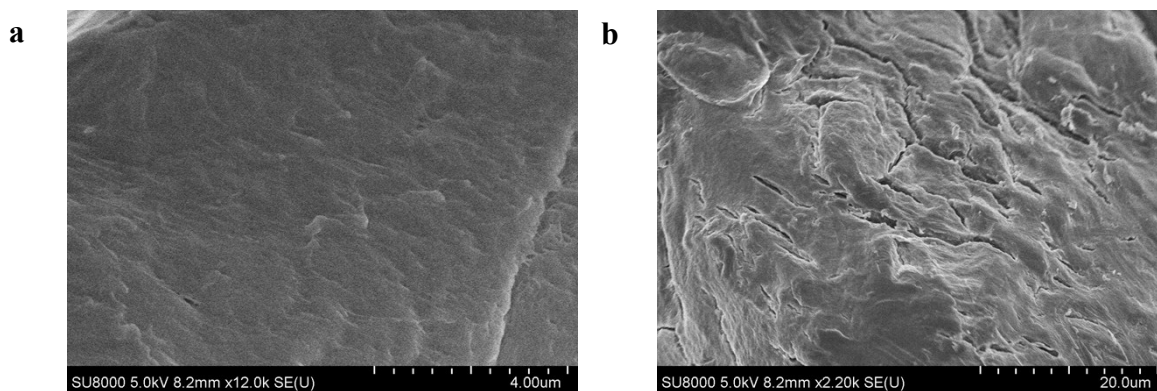


**Figure S2.** FT-IR spectra of choline-based ILs.

The FT-IR spectra are shown in Figure S2. Various ILs show similar curves in the range of 3500-500  $\text{cm}^{-1}$ , which illustrates similarity in their structures. The broad peak between 3500-3000  $\text{cm}^{-1}$  are attributed to the stretching vibration of O-H. The asymmetric stretching vibration and symmetric stretching vibration of methyl group are at 2960  $\text{cm}^{-1}$  and 2870  $\text{cm}^{-1}$ , respectively. Under cover of the O-H broad peak, the intensity of peaks of methyl group is reduced. Absorbance at 1560  $\text{cm}^{-1}$  and 1400  $\text{cm}^{-1}$  ascribe to the asymmetric stretching vibration and symmetric stretching vibration of C=O bonds in carboxylate anions, respectively.



**Figure S3.** HPLC spectrum of the main product



**Figure S4.** SEM spectra of (a) original PET and (b) residual PET

**Table S1.** Elemental analysis results of the main product (BHET)

Sample	C (%)	H (%)	O (%)
BHET standard	56.69	5.55	37.76
BHET	56.89	5.63	37.79