

Electronic supplementary information for:

**Efficient production of adipic acid from 2-methoxycyclohexanone by aerobic oxidation with phosphotungstic acid catalyst**

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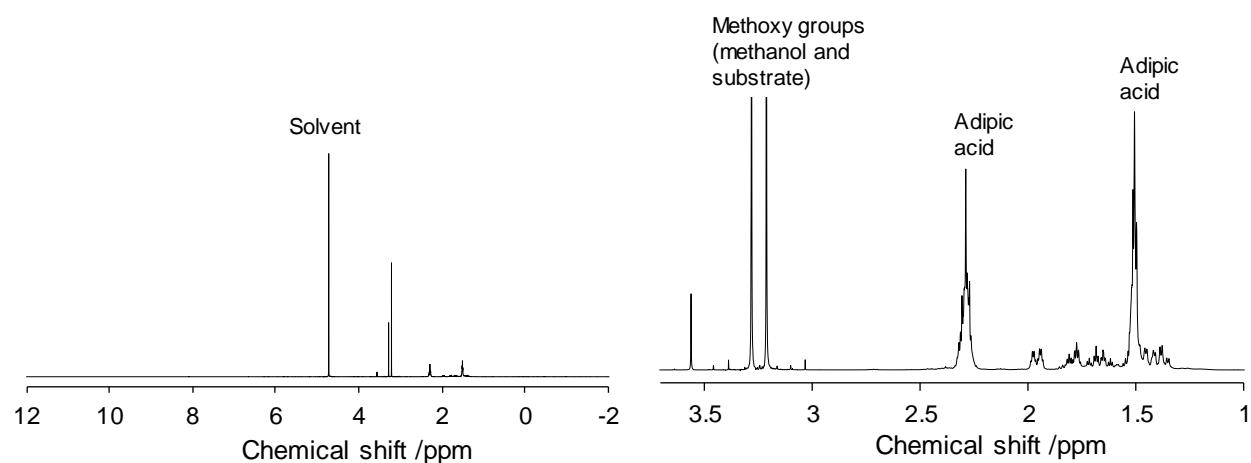


Fig. S1. <sup>1</sup>H NMR of reaction solution

Conditions: 2-methoxycyclohexanone (2-MCO) 2.1 mmol, H<sub>3</sub>PW<sub>12</sub>O<sub>40</sub> 55 μmol, D<sub>2</sub>O 5 g, O<sub>2</sub> 0.1 MPa (at r. t.), 353 K, 72 h. Reaction was carried out in glass tube with O<sub>2</sub> balloon. Left: full scale, right: enlarged spectra.

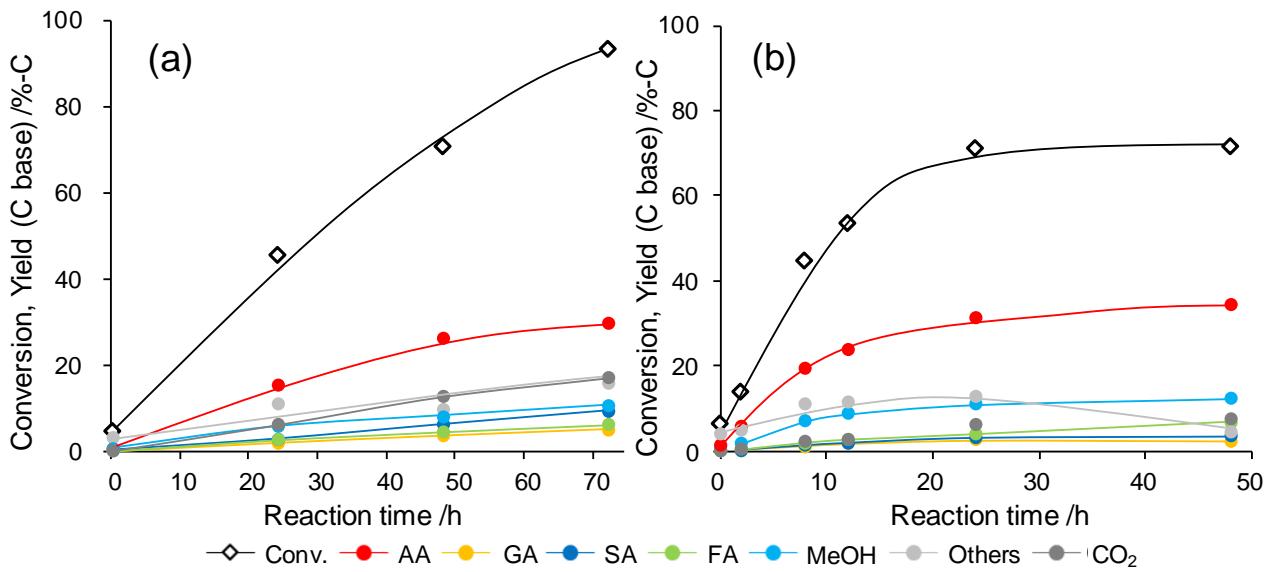


Fig. S2. Time course of the 2-MCO oxidation with typical vanadium catalyst. (a)  $\text{V}_2\text{O}_5$ . (b)  $\text{H}_5\text{PV}_2\text{Mo}_{10}\text{O}_{40}$ .

Reaction condition: 2-methoxycyclohexanone (2-MCO) 4.3 mmol, catalyst 55  $\mu\text{mol}$ , water 10 g,  $\text{O}_2$  0.8 MPa (at r. t.), 353 K, 0-72 h. AA: Adipic acid, GA: Glutaric acid, SA: Succinic acid, FA: Formic acid.

Table S1. 2-MCO oxidation with various catalysts.

Entry	Catalyst (Amount / $\mu$ mol)	Conv. /%	Yield /%-C						AA sel. /%-C	Carbon balance /%
			AA	GA	SA	FA	MeOH	Others		
1	Blank	7	1.9	0.2	<0.1	0.1	1.2	3.7	0.1	26
2	$\text{H}_5\text{PV}_2\text{Mo}_{10}\text{O}_{40}$ (55)	71	31	2.6	3.2	3.9	11	13	6.2	44
3	$\text{V}_2\text{O}_5$ (55)	31	11	1.4	1.9	1.8	3.9	7.0	3.6	37
4	$\text{V}_2\text{O}_5$ (55) + $\text{H}_3\text{PMo}_{12}\text{O}_{40}$ (110)	77	46	0.8	0.5	1.2	11	14	3.1	60
5	$\text{V}_2\text{O}_5$ (55) + $\text{H}_3\text{PW}_{12}\text{O}_{40}$ (110)	79	36	1.9	1.6	2.7	11	21	5.8	45
6	$\text{V}_2\text{O}_5$ (55) + $\text{H}_4\text{SiW}_{12}\text{O}_{40}$ (110)	84	23	3.7	9.6	8.1	12	8.7	19	28
7	$\text{H}_3\text{PW}_{12}\text{O}_{40}$ (110)	60	45	1.1	0.2	0.4	9.7	3.8	0.2	74
8	$\text{H}_3\text{PW}_{12}\text{O}_{40}$ (110) <sup>a</sup>	59	42	1.3	0.2	0.4	7.0	7.7	1.0	70
9	$\text{H}_4\text{SiW}_{12}\text{O}_{40}$ (110)	38	21	1.0	0.3	0.7	7.9	5.9	0.8	56
10	$\text{H}_3\text{PMo}_{12}\text{O}_{40}$ (110)	66	50	0.2	<0.1	1.2	9.5	5.2	0.3	75
11	$\text{H}_4\text{SiMo}_{12}\text{O}_{40}$ (110)	60	45	0.2	<0.1	1.1	8.2	4.4	0.2	76

Reaction conditions : 2-methoxycyclohexanone 4.3 mmol, catalyst 0-1320  $\mu$ mol, water 10 g,  $\text{O}_2$  0.8 MPa (at r. t.), 353 K, 24 h.

<sup>a</sup>The sheath of autoclave was covered with teflon tube. AA: Adipic acid, GA: Glutaric acid, SA: Succinic acid, FA: Formic acid, MeOH: Methanol.

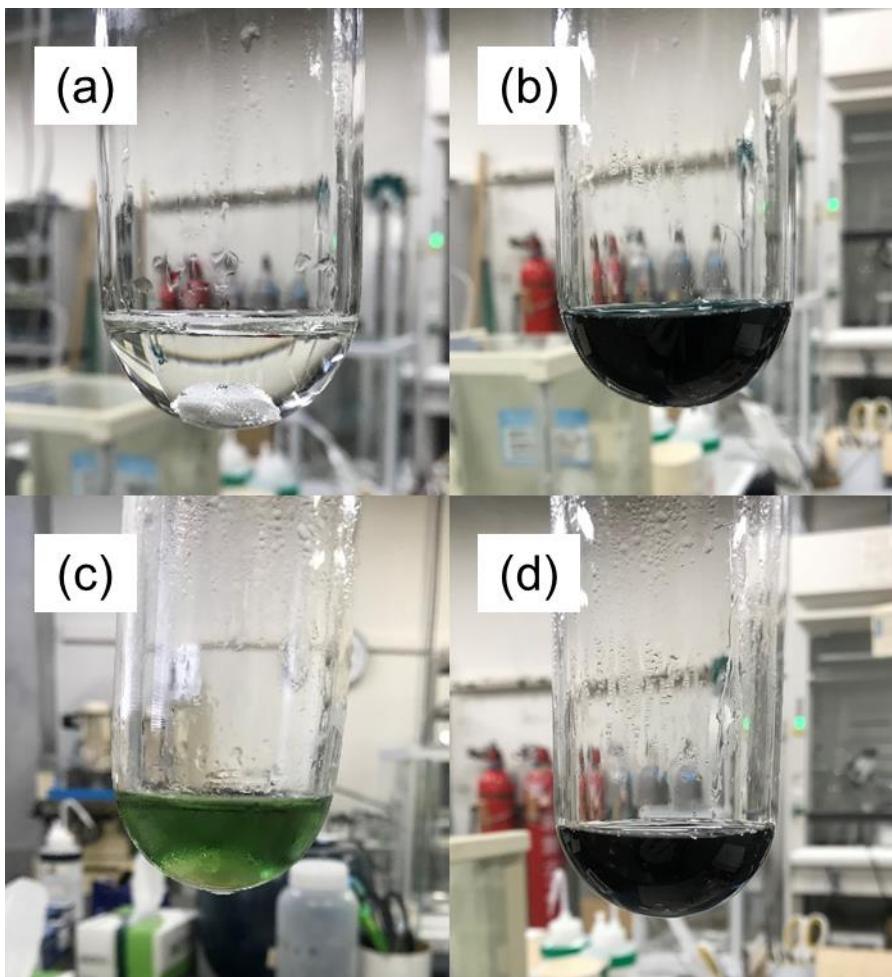


Fig. S3. The color of reaction solution with (a)  $\text{H}_3\text{PW}_{12}\text{O}_{40}$ , (b)  $\text{H}_3\text{PMo}_{12}\text{O}_{40}$ , (c)  $\text{H}_4\text{SiMo}_{12}\text{O}_{40}$  and (d)  $\text{H}_3\text{PW}_{12}\text{O}_{40}$  (at  $\text{O}_2$  0 MPa) as catalysts.

Reaction condition: 2-methoxycyclohexanone (2-MCO) 4.3 mmol, catalyst 110  $\mu\text{mol}$ , water 10 g,  $\text{O}_2$  or  $\text{N}_2$  0.8 MPa (at r. t.), 353 K, 24 h.

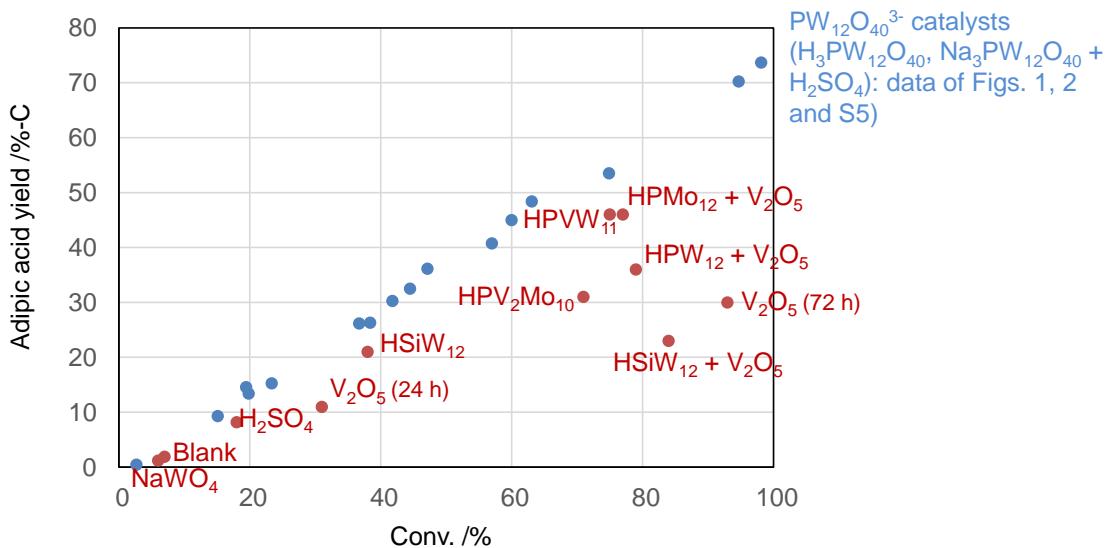


Fig. S4. Comparison of adipic acid yields between catalysts. The data are taken from Figs. 1, 2 and S5, below, for PW<sub>12</sub>O<sub>40</sub><sup>3-</sup> catalysts (blue color) and Table 1 for other catalysts (red color). The reaction temperature (353 K), 2-MCO concentration (4.3 mmol in 10 g water) and O<sub>2</sub> pressure (0.8 MPa at r.t.) were the same for all data.

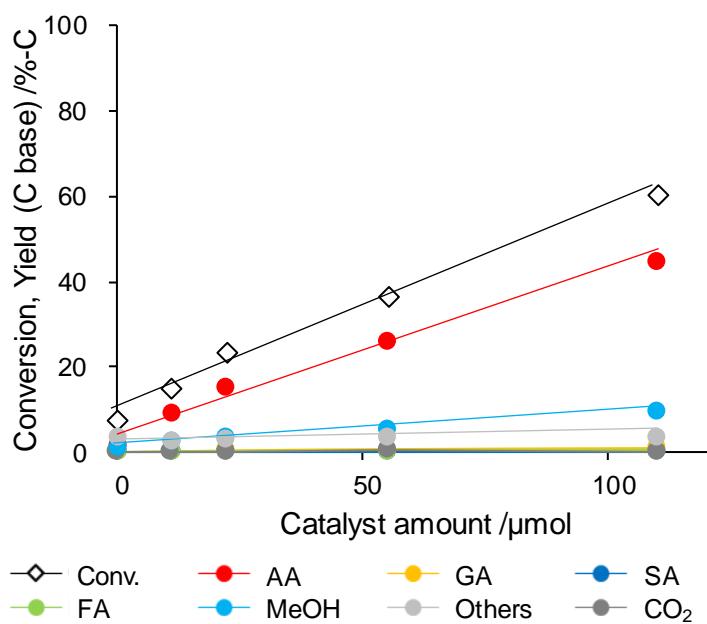


Fig. S5. Effect of amount of  $\text{H}_3\text{PW}_{12}\text{O}_{40}$  in 2-MCO oxidation.

Reaction condition: 2-methoxycyclohexanone (2-MCO) 4.3 mmol,  $\text{H}_3\text{PW}_{12}\text{O}_{40}$  0–110  $\mu\text{mol}$ , water 10 g,  $\text{O}_2$  0.8 MPa (at r. t.), 353 K, 24 h. AA: Adipic acid, GA: Glutaric acid, SA: Succinic acid, FA: Formic acid.

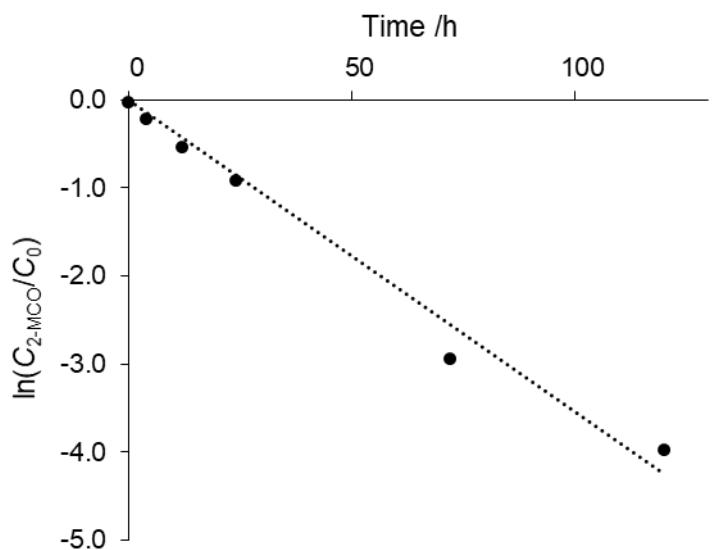


Fig. S6.  $\ln(C_{2\text{-MCO}}/C_0)$  vs Time.

Reaction condition: 2-methoxycyclohexanone (2-MCO) 4.3 mmol,  $\text{H}_3\text{PW}_{12}\text{O}_{40}$  110  $\mu\text{mol}$ , water 10 g,  $\text{O}_2$  0.8 MPa (at r. t.), 353 K, 0-120 h.