

Electronic Supporting Information

**Unlocking biomass energy: continuous high-yield production of 5-
hydroxymethylfurfural in water**

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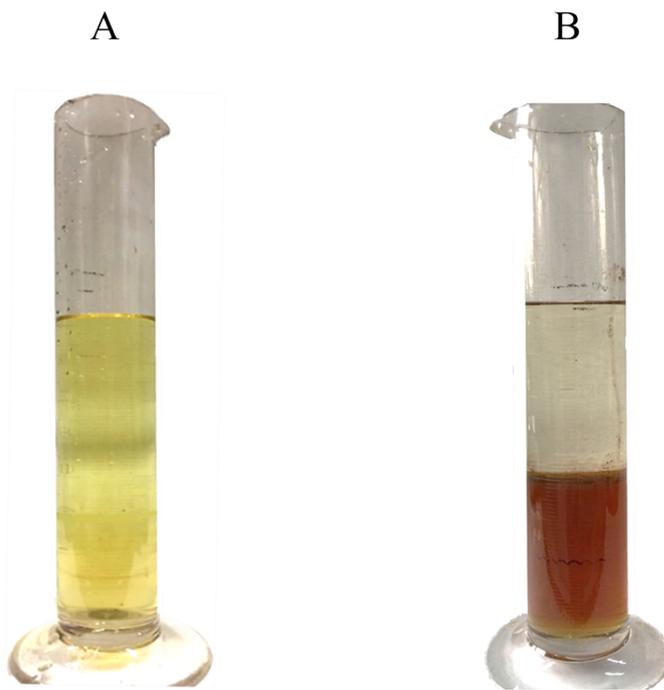


Figure S2. Aqueous and organic phases of H₂O/EAC system after reaction. (A) without addition of NaCl in the aqueous phase; (B) with saturated NaCl solution in the aqueous phase.

Table S1. Process operating data (data of feedstock dehydration at high temperature per pass).

	Fructose feedstock	Glucose feedstock
<i>Parameters</i>		
Conversion (mol%)	40.0	32.6
Selectivity (mol%)	87.5	40.2
Organic phase	2-MeTHF	MIBK
The flow rate of organic phase (mL/min)	10	20
Temperature (°C)	150	140

Table S2. Energy requirements for the HMF production (with the use of heat exchanger).

	Energy required		
	Heating, kW	Refrigerating, kW	Electricity, kW
<i>Fructose feedstock</i>			
Fructose to HMF reactor unit	446	/	/
The circulated aqueous phase system	1445	/	30.6
The circulated organic phase system	12104	-13115	4.7
Total, fructose to HMF	13995	-13115	36
<i>Glucose feedstock</i>			
Glucose to HMF reactor unit	568	/	/
The circulated aqueous phase system	7061	/	39.5
The circulated organic phase system	21911	-23612	9.2
Total, glucose to HMF	29540	-23612	49

Table S3. Project investment and operating costs.

	Fructose feedstock	Glucose feedstock
<i>Process section</i>	<i>Units (MM\$)</i>	
HMF Production system	5.910	7.520
OSBL ^a	2.370	3.010
Total project investment ^b	8.280	10.530
<i>Raw material</i>		
Feedstock	7.240	0.703
Refrigeration	2.980	5.362
Solvent makeup	0.680	0.286
Steam	1.110	2.294
Catalyst	~0	~0
Cooling water	~0	~0
Electricity	0.061	0.023
Wastewater disposal	0.003	0.006
Total variable operating costs	12.074	8.674
<i>HMF production</i>		
Production (ton/yr)	10128	10560
Purity (wt%) & recovery (%)	98.6 & >99.9	98.0 & >99.9

^a OSBL (outside battery limits of the plant) includes infrastructure costs for waste disposal, on-site storage, and utilities.

^b the total project investment including the total installed cost.

The minimum selling price of HMF is obtained as follows^{S1, S2}:

$$\text{HMF sales} = \text{Total variable operating costs} + \text{Total project investment}/20 + \text{Average income tax} + \text{Average ROI} \quad (1)$$

$$\begin{aligned} \text{Fructose feeding HMF sales (\$/ton HMF)} &= 12074000/10128 + 8280000/10128/20 + (1192 + 41) \times 0.21 / (1 - 0.21) \\ &+ (1192 + 41 + 328) \times 0.10 \\ &= 1192 + 41 + 328 + 156 \\ &= 1716 \end{aligned}$$

$$\begin{aligned} \text{Glucose feeding HMF sales (\$/ton HMF)} &= 8674000/10560 + 10530000/10560/20 + (822 + 50) \times 0.21 / (1 - 0.21) + \\ &(822 + 50 + 232) \times 0.10 \\ &= 822 + 50 + 232 + 111 \\ &= 1215 \end{aligned}$$

Table S4. List of economic parameters and assumptions for the HMF production (with the use of heat exchanger).

Fructose price (\$/ton)	450 ^a
Glucose price (\$/ton)	300 ^a
HCl price (\$/ton)	17 ^b
AlCl ₃ ·6H ₂ O price (\$/ton)	186 ^a
MIBK price (\$/ton)	1000 ^a
2-MeTHF price (\$/ton)	5000 ^a
Wastewater treatment cost (\$/ton)	0.570 ^c
Low pressure steam cost (\$/ton)	6.737 ^c
-29°C refrigerant (\$/ton)	1.300 ^c
Electricity price (\$/kWh)	0.0572 ^c
Cooling water (\$/ton)	9.85×10 ⁻³ ^c
Operating mode	Continuous
Plant life (years)	20 ^d
Plant operating hours per year (hours)	8000 ^d
Discount rate (%)	10 ^d
General plant recovery period (years)	7 ^d
Corporate income rate (%)	21 ^d
Equity financing (%)	40 ^d
Loan terms	10-years loan at 8% APR ^d
Working capital	5% of fixed capital investment ^d
Length of start-up period (weeks)	20 ^d
Revenues during start-up (%)	50 ^d
Fixed costs during start-up (%)	100 ^d

^a The price standards used refer to China's chemical market price level in 2018.

^b Taken from reference prices by *P. Desir et al.*⁵³

^c Taken from reference prices by *A. H. Motagamwala et al.*⁵⁴

^d Data were taken from a study by NREL.⁵⁵

Table S5. Results of HMF extraction ratio in biphasic system. Conditions: 0.5 g HMF; aqueous phase: 4 g H₂O, 1 g DMSO, 5mg HCl; 5 ml MIBK; T = 25°C; t = 1 h, triplicate.

Entry	HMF _{aq} (g/ml)	HMF _{org} (g/ml)	R	Average of R
1	0.052	0.048	0.90	
2	0.049	0.051	1.04	0.97
3	0.050	0.049	0.98	

Table S6. Space-time yields for different extraction models. Reaction condition: 5 wt% fructose, 28 mM HCl, 150°C, t = 210 min, 60 g aqueous phase (H₂O/DMSO = 8/2, w/w) and 80 g MIBK were added to the extraction device, the rates of aqueous and cyclic continuous flow of organic phase were set at a flow rate of 7 and 10 mL/min, respectively.

Extraction models	Fructose conversion (mol %)	HMF selectivity (mol %)	HMF yield (mol %)	Space-time yield [g/g(cat)/min]
Static extraction	91.5	42.8	39.1	6.5×10 ⁻²
Intermittent replacement extraction	95.8	65.3	62.5	1.0×10 ⁻¹
Cyclic continuous extraction	90.7	83.4	75.6	1.3×10 ⁻¹

Table S7. Results of dehydration of fructose in H₂O/MIBK circulating system. Reaction condition: 5 wt% fructose catalyzed by different kinds of acids; T = 150°C; the flow rate of aqueous phase and organic phase were 7 and 20 mL/min, respectively.

Entry	Catalyst	Conversion (mol%)	Selectivity (mol%)	Yield (mol%)	t (min)
1	28 mM HCl	99.3	87.9	87.3	360
2	28 mM H ₂ SO ₄	64.5	56.7	36.6	300
3	28 mM H ₃ PO ₄	47.2	77.8	36.7	300

Table S8. Rate constants (*k*) of acid-catalyzed dehydration of fructose in H₂O/MIBK circulating system were calculated by Berkeley Madonna software.

Parameters	<i>k</i> (min ⁻¹)	
HCl dosage (mM)	0	0.00090
	14	0.00417
	28	0.01211
	42	0.01737
Temperature (°C)	140	0.00771
	150	0.01311
	160	0.03701
	170	0.11125

Table S9. Results of acid-catalyzed dehydration of fructose in H₂O/EAC circulating system. Reaction condition: 5 wt% fructose; 28 mM HCl; T = 150°C; the flow rate of aqueous phase and organic phase were 7 and 20 mL/min, respectively.

t (min)	Conversion (mol%)	Selectivity (mol%)	Yield (mol%)
60	62.4	84.9	53.0
120	87.8	92.9	81.6

Table S10. Comparison with other reaction systems.

Type	Feedstock	Solvent	Catalyst	T (°C)	Yield (mol%)	Ref.
Biphasic circulating system	5 wt% fructose	H ₂ O/2-MeTHF	28 mM HCl	150	94.3	This work
Stirred tank	30 wt% fructose	H ₂ O/MIBK	250 mM HCl	180	54.6	28 (S6)
Continuous reactor	2 wt% fructose	(4:1 w/w) THF/H ₂ O	500 mg λ -Al ₂ O ₃ -SO ₃ H	90	69.4	27 (S7)
Biphasic continuous flow	10 wt% fructose	H ₂ O/MIBK	250 mM HCl	140	75.0	22 (S8)
Biphasic circulating system	5 wt% glucose	H ₂ O/MIBK	70 mM AlCl ₃ ·6H ₂ O and 22 mM HCl	140	69.6	This work
Biphasic slug flow capillary microreactor	2 wt% glucose	H ₂ O/MIBK	40 mM AlCl ₃ ·6H ₂ O and 40 mM HCl	160	53.0	36 (S9)

Supplemental references

- S1. Torres, A.I., Daoutidis, P., and Tsapatsis, M. (2010). Continuous production of 5-hydroxymethylfurfural from fructose: a design case study. *Energy Environ. Sci.* **3**, 1560–1572.
- S2. Torres, A.I., Tsapatsis, M., and Daoutidis, P. (2012). Biomass to chemicals: Design of an extractive-reaction process for the production of 5-hydroxymethylfurfural. *Comput. Chem. Eng.* **42**, 130–137.
- S3. Desir, P., Saha, B., and Vlachos, D.G. (2019). Ultrafast flow chemistry for the acid-catalyzed conversion of fructose. *Energy Environ. Sci.* **12**, 2463–2475.
- S4. Motagamwala, A.H., Huang, K.F., Maravelias C.T., and Dumesic, J.A. (2019). Solvent system for effective near-term production of hydroxymethylfurfural (HMF) with potential for long-term process improvement. *Energy Environ. Sci.* **12**, 2212–2222.
- S5. Davis, R., Tao, L., Tan, E.C.D., Bidy, M. J., Beckham, G. T., Scarlata, C., Jacobson, J., Cafferty, K., Ross, J., Lukas, J., and et al. (2013). Process design and economics for the conversion of lignocellulosic biomass to hydrocarbons: dilute-acid and enzymatic deconstruction of biomass to sugars and biological conversion of sugars to hydrocarbons. NREL/TP-5100-60223.
- S6. Román-Leshkov, Y., Chheda, J.N., and Dumesic, J.A. (2006). Phase modifiers promote efficient production of hydroxymethylfurfural from fructose. *Science* **312**, 1933–1937.
- S7. Morales-Leal, F.J., de la Rosa J.R., Lucio-Ortiz, C.J., De Haro-Del Rio, D.A., Maldonado, C.S., Wi, S., Casabianca, L.B., Garcia, C.D. (2019). Dehydration of fructose over thiol- and sulfonic- modified alumina in a continuous reactor for 5-HMF production: Study of catalyst stability by NMR. *Appl. Catal., B.* **244**, 250–261.
- S8. Brasholz, M., von Kanel, K., Hornung, C.H., Saubern, S., Tsanaktsidis, J. (2011). Highly efficient dehydration of carbohydrates to 5-(chloromethyl)furfural (CMF), 5-(hydroxymethyl)furfural (HMF) and levulinic acid by biphasic continuous flow processing. *Green Chem.* **5**, 1114–1117.
- S9. Guo W.Z., Heeres H.J., and Yue J. (2020). Continuous synthesis of 5-hydroxymethylfurfural from glucose using a combination of AlCl₃ and HCl as catalyst in a biphasic slug flow capillary microreactor. *Chem. Eng. J.* DOI: 10.1016/j.cej.2019.122754.