

Facile synthesis of controllable graphene-coshelled reusable Ni/NiO nanoparticles and application for synthesis of amines under mild conditions

Jianguo Liu^{1,2}, Yuting Zhu^{1,5}, Chenguang Wang¹, Thishana Singh³, Nan Wang^{1,4}, Qiying Liu¹, Zhibing Cui¹, Longlong Ma^{1,4*}

¹ CAS Key Laboratory of Renewable Energy, Guangdong Provincial Key Laboratory of New and Renewable Energy Research and Development, Guangzhou Institute of Energy Conversion, Chinese Academy of Sciences, Guangzhou 510640, P. R. China.

² Dalian National Laboratory for Clean Energy, Dalian 116023, P. R. China.

³ School of Chemistry and Physics, University of KwaZulu-Natal, Durban, 4000, South Africa.

⁴ School of Environmental Science and Engineering, Tianjin University, Tianjin 300350, P. R. China.

⁵ University of Chinese Academy of Sciences, Beijing 100049, P. R. China.

* e-mail: mall@ms.giec.ac.cn

S1. Methods and materials

All aldehydes, ketones, nitroarenes and amines were obtained commercially from various chemical companies. 1,3,5 trimethoxybenzene was purchased from Sigma Aldrich Co., Ltd.; nickel(II) nitrate hexahydrate (AR, 98%), citric acid (AR, ≥99.5%), anhydrous ethanol (99.5%) were obtained from Shanghai Macklin Biochemical Co., Ltd.; H₂SO₄ (GR, 98%) was purchased from Sinopharm Chemical Reagent Co., Ltd.; Al₂O₃ (basic, 200~300 meshes), 5 wt% Pt/C, 5 wt% Pd/C, 10 wt% Pd/C, and 5 wt% Ru/C catalysts were purchased from Aladdin Co., Ltd.; CDCl₃ (99.8% D, containing 0.03% TMS, stabilized with Ag) was applied by Meryer Chemical Technology Co., Ltd. Before using, the purity of aldehydes and ketones has been checked.

GC were recorded by GC-2014C (Shimadzu, Japan) column with HP5 (30 m × 250 mm × 0.25 μm) and FID detector. GC-MS was determined by TRACE 1300ISQ GC-MS (Thermo Fisher Scientific, America). ¹H and ¹³C NMR were recorded on a Bruker AVANCE III 400 MHz spectrometer using CDCl₃ solvents. The TEM measurements were performed on a JEM-2100F microscope operated at 200 kV. The SEM was conducted on Analytical SEM SU-70 (Hitachi, Japan) operated at 10kV. XPS data were conducted on a Escalab250 Xi photoelectron spectrometer (Thermo Fisher Scientific, America) equipped with an Al Kα X-ray source with scan number of 5. The peak position was calibrated with C 1s peak at 284.8 eV. XRD powder patterns were recorded on an X'Pert Pro MPD (PANalytical, Netherlands) using the Cu Kα radiation source. Processing and assignment of the powder patterns was done on the software Jade 6.0 and using the Powder Diffraction File (PDF) database of the International Centre of Diffraction Data (ICDD). Nitrogen adsorption isotherms were measured at -196 °C QUADRASORB SI apparatus (Quantachrome, America) apparatus by using the BET method. Raman spectra were measured at LabRAM HR800 apparatus (HORIBA JY, France) with wavelength 100~3200 cm⁻¹.

S2. General preparation of catalysts

S2.1. Preparation of Ni@C-H₂O

In a typical synthesis procedure, nickel (II) nitrate hexahydrate ($\text{Ni}(\text{NO}_3)_2 \cdot 6 \text{H}_2\text{O}$, 0.3 mol) and citric acid ($\text{C}_6\text{H}_8\text{O}_7$, 0.3 mol) were dissolved in deionized water (5 mL). The mixture was then aged at 70 °C for 8 h under stirring (300 rpm) until to obtain a transparent greenish gel, followed by drying at 100 °C for 72 h in a drying oven to remove the excess water. The obtained greenish solid was then calcining at fixed bed at 600 or 700 °C for 3 h under a high-purity N_2 (99.999 %) flow of 40 mL min^{-1} . The heating rate was controlled at 2 °C min^{-1} . The obtained black solids were treated in 1M H_2SO_4 aqueous solution at 70 °C until the solution colorless to remove the insecure and uncovered Ni particles. The black solids were then full washed with deionized water until pH of waste solution was 7. Then, the black solids were dried at -48 °C for 12 h in vacuum by using a freeze dryer. The dried black solids denoted as Ni@C-x-H₂O, where x is calcining temperature of greenish gel, such as Ni@C-600-H₂O or Ni@C-700-H₂O.

S2.2. Preparation of Ni@C-EtOH

In a typical synthesis procedure, nickel nitrate hexahydrate ($\text{Ni}(\text{NO}_3)_2 \cdot 6 \text{H}_2\text{O}$, 0.3 mol) and citric acid ($\text{C}_6\text{H}_8\text{O}_7$, 0.3 mol) were dissolved in anhydrous ethanol (5 mL). The mixture was then aged at 70 °C for 4 h under stirring (300 rpm) until to obtain a bubble-shaped greenish gel, followed by drying at 100 °C for 24 h in a drying oven to remove the excess solvent. The obtained greenish solid was then calcining at fixed bed at 600 or 700 °C for 3 h under a high-purity N_2 (99.999 %) flow of 40 mL $\cdot \text{min}^{-1}$. The heating rate was controlled at 2 °C $\cdot \text{min}^{-1}$. The obtained black solids were treated in 1M H_2SO_4 aqueous solution at 70 °C until the solution colorless to remove the insecure and uncovered Ni particles. The black solids were then full washed with deionized water until pH of waste solution was 7. Finally, the black solids were dried at -48 °C for 12 h in vacuum by using a freeze dryer. The dried black solids denoted as Ni@C-x-EtOH, where x is calcining temperature of greenish gel, such as Ni@C-600-EtOH and Ni@C-700-EtOH.

S2.3. Preparation of Ni-NiO@Cs

In a typical synthesis procedure, Ni@C-H₂O and Ni@C-EtOH were grinded and sieved to powders of 800 meshes and then were heated at fixed bed at 200 °C for 1 or 2 h under argon (containing 1 vol% O_2) flow of 30 mL $\cdot \text{min}^{-1}$. All these materials were labeled as M1/M2@C-x-y-z-s, where M1= Ni; M2=NiO; x denotes the pyrolysis temperature; y denotes the oxidation temperature; z denotes the oxidation time; and s denotes the solvent, respectively.

S3. General procedure of the synthesis of amines

S3.1. General procedure for the synthesis of primary amines

The reaction was proceeding in a stainless-steel autoclave with six wells (20 mL per well), one thermocouple and one circulating water-cooling equipment. Each well was loaded with one 10 mm magnetic stirring bar, 0.5 mmol corresponding carbonyl compound, 0.05~0.1mmol 1,3,5-trimethoxybenzene, 10 mg M1/M2@C-x-y-z-s catalyst, and 5mL ammonia solution (2 M in methanol). Then, the autoclave was sealed and flushed with hydrogen (H_2) three times at 2 MPa pressure and it was pressurized with 2 MPa H_2 . The autoclave was placed into a heating mantle and the stirring was set at 400 rpm. The autoclave was heated from room temperature to target temperature (inside temperature) at rate of 2 °C $\cdot \text{min}^{-1}$ and keep the target temperature for require time. The target temperature was used as the reaction temperature. After the reaction, the autoclave was cooled down to room temperature. The waste gas was discharged then the reaction solutions were collected with a dropper and filtered. The catalyst was immobilized on the magnetic stirring bar and can be recycled by a magnet. The catalyst was washed thoroughly with ethanol and water. The reaction products were identified by GC-MS and ^1H NMR and the yields of reaction products

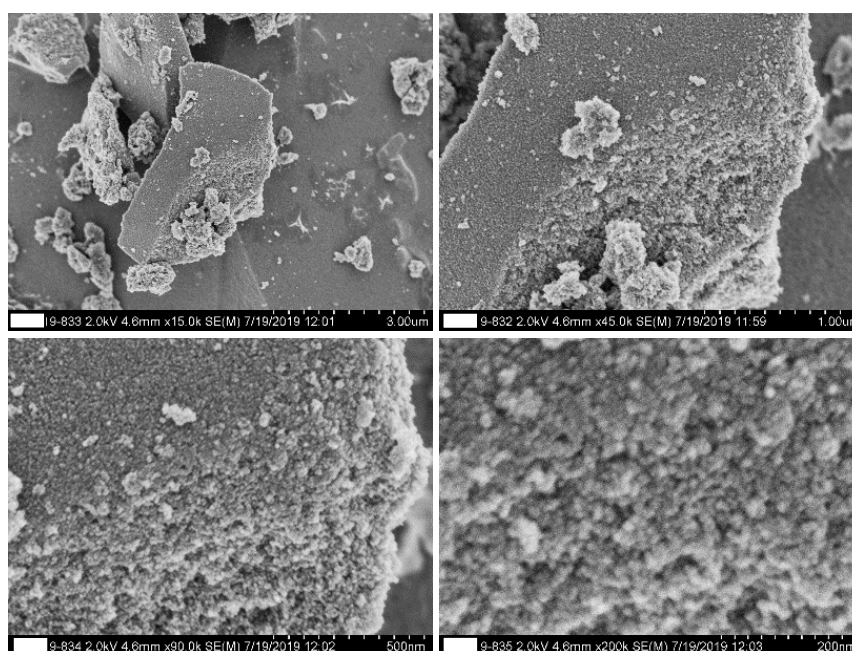
were determined by ^1H NMR. For ^1H NMR analysis, about 2 mL reaction solution was concentrated under a N_2 -flow, and then added 0.5 mL of CDCl_3 which preneutralized with basic Al_2O_3 .

1 g scale experiments were carried out in a 500 mL stainless steel autoclave with one thermocouple. In a typical experiment, 1g carbonyl compound, 0.35 mmol 1,3,5-trimethoxybenzene, one 30 mm magnetic stirring bar, and $\text{Ni/NiO}@C-700-200-2\text{h-EtOH}$ (10mg per 0.5 mmol carbonyl compound) and ammonia solution (2 M in methanol, 5 mL per 0.5 mmol carbonyl compound) were loaded into the reactor. The procedure of reaction and product analysis were similar to the 0.5 mmol scale experiments.

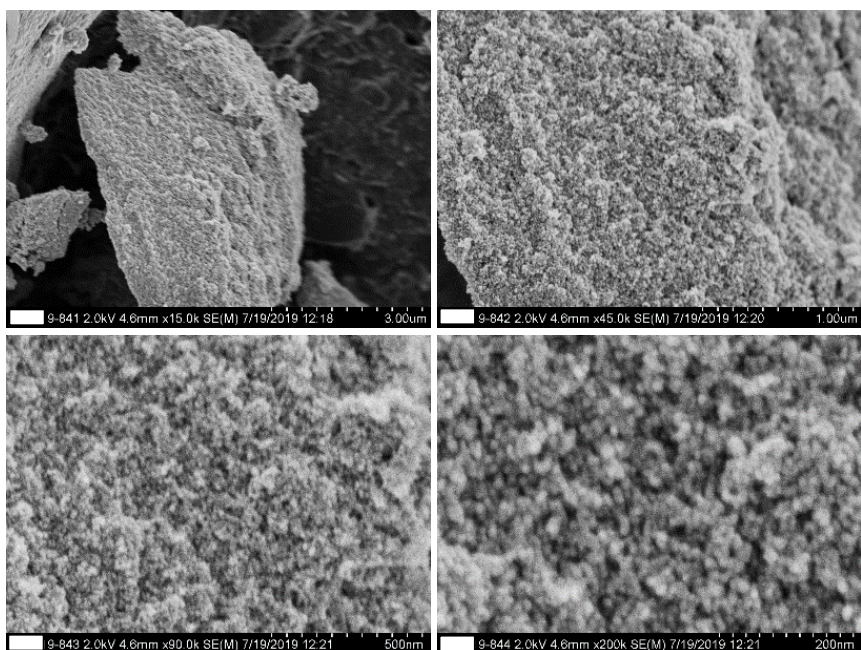
S3.2. General procedure for the synthesis of secondary amines and tertiary amines

The reaction was proceeding in a stainless steel autoclave with six wells (20 mL per well), one thermocouple and one circulating water cooling equipment. Each well was loaded with one 10 mm magnetic stirring bar, 0.5-3 mmol corresponding carbonyl compound, 0.5-0.75 mmol corresponding amine, 0.05~0.1 mmol 1,3,5-trimethoxybenzene, 10 mg $\text{Ni-NiO}@C-700-200-2\text{h-EtOH}$, and 5 mL methanol. Then, the autoclave was sealed and flushed with hydrogen (H_2) three times at 1-2MPa pressure and it was pressurized with 1-2MPa H_2 . The autoclave was placed into a heating mantle and stirring the reaction solution at 400 rpm. The autoclave was heated from room temperature to target temperature (inside temperature) at rate of $2^\circ\text{C}\cdot\text{min}^{-1}$ and keep the target temperature for require time. The target temperature was used as the reaction temperature. After the reaction, the autoclave was cooled down to room temperature. The waste gas was discharged then the reaction solutions were collected with a dropper and filtered. The catalyst was immobilized on the magnetic stirring bar and can be recycled by a magnet. The catalyst was washed thoroughly with ethanol and water. The reaction products were identified by GC-MS and ^1H NMR and the yields of reaction products were determined by ^1H NMR on a BRUKER AVANCE III 400 MHz spectrometer. For ^1H NMR analysis, about 2 mL reaction solution was concentrated under a N_2 -flow, and then added 0.5 mL of CDCl_3 which preneutralized by Al_2O_3 .

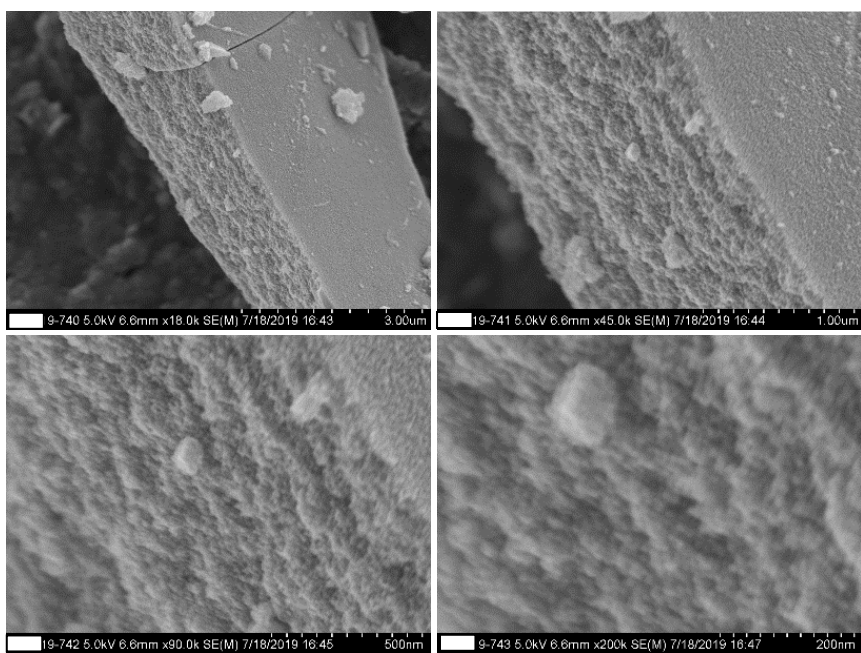
S4. TEM spectra



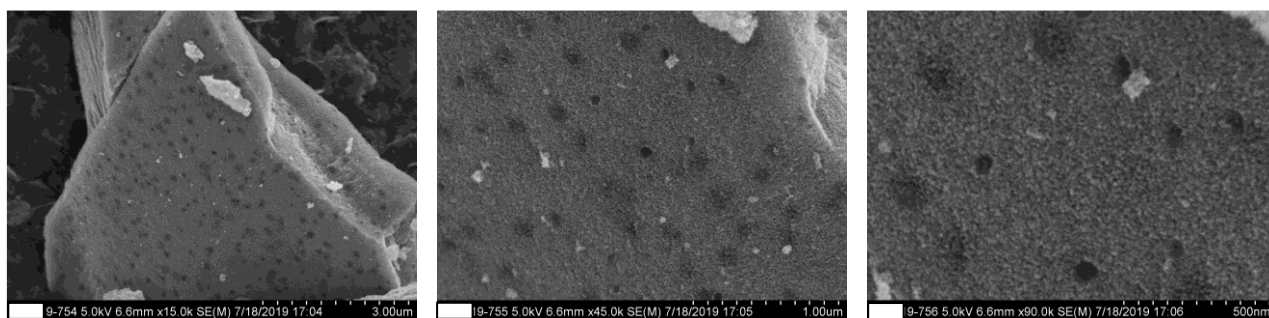
$\text{Ni}@C-600-\text{H}_2\text{O}$



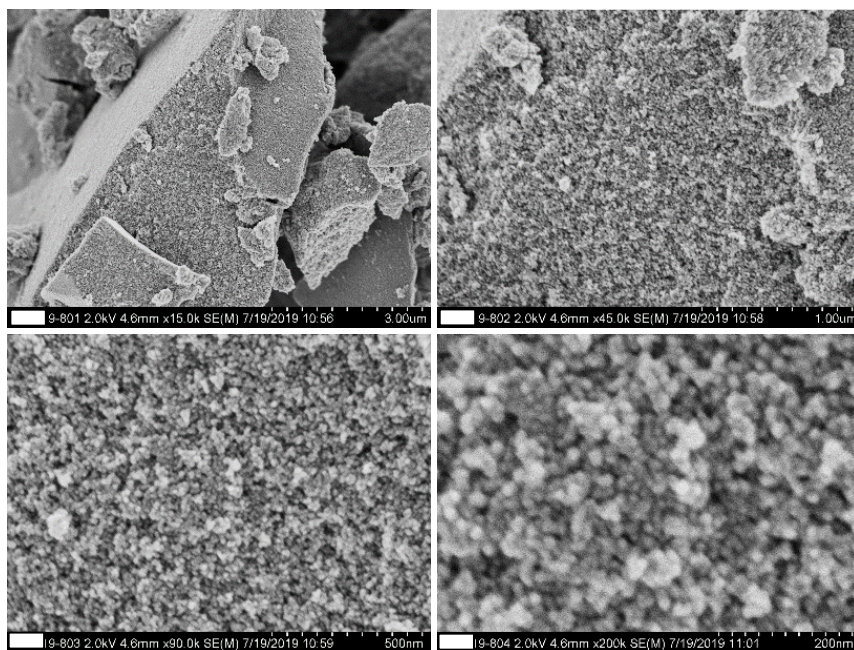
Ni/NiO@C-600-200-1h-H₂O



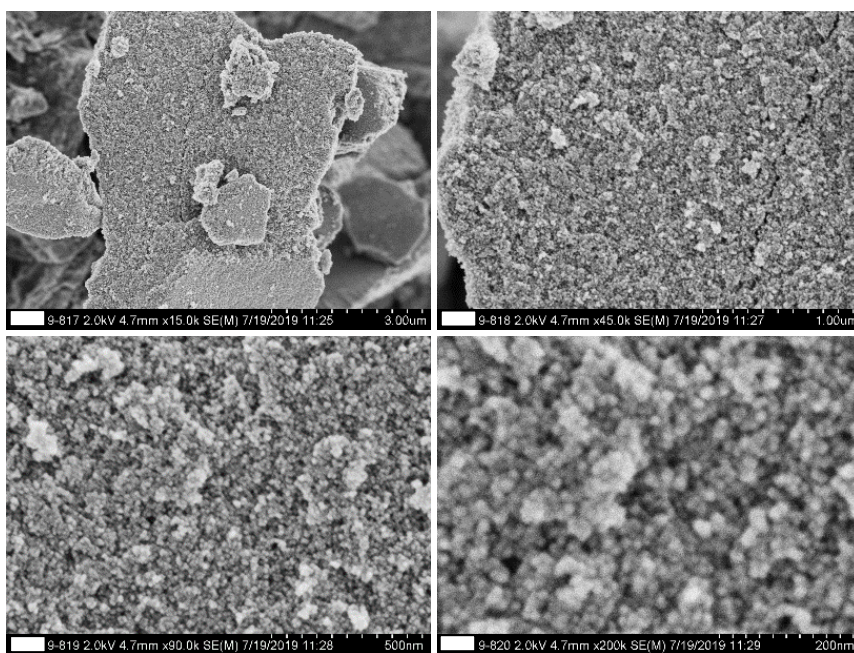
Ni@C-700-H₂O



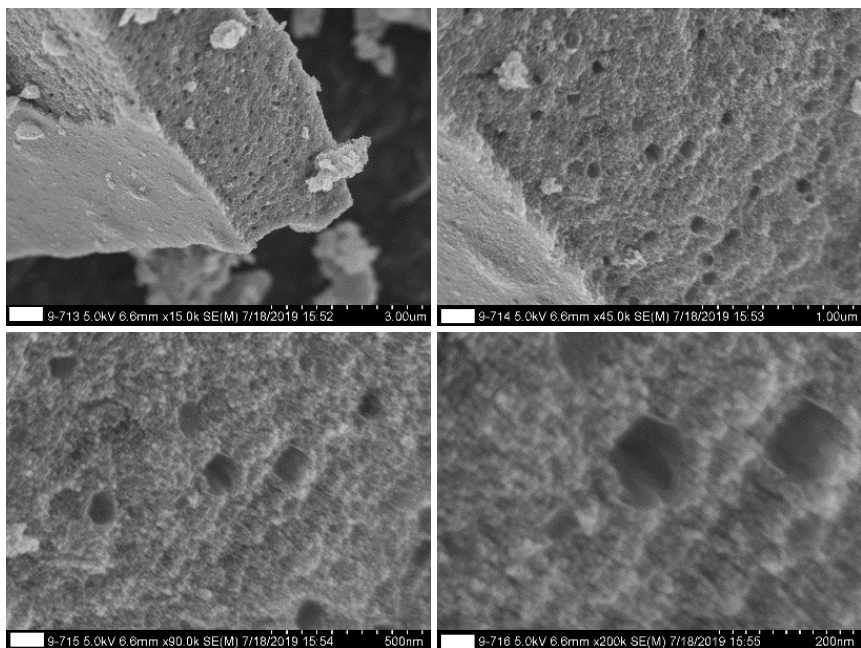
Ni/NiO@C-700-200-1h-H₂O



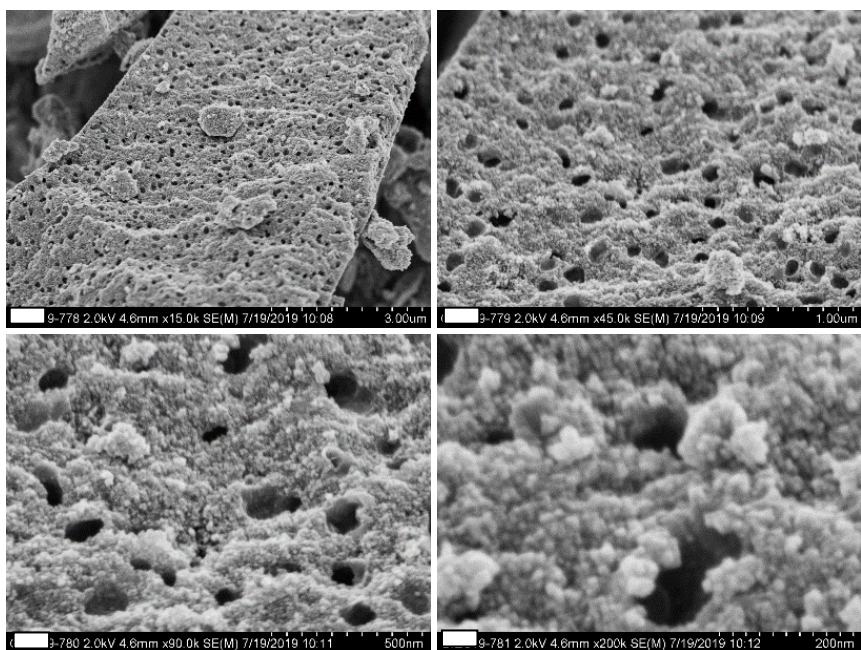
Ni@C-600-EtOH



Ni/NiO@C-600-200-1h-EtOH

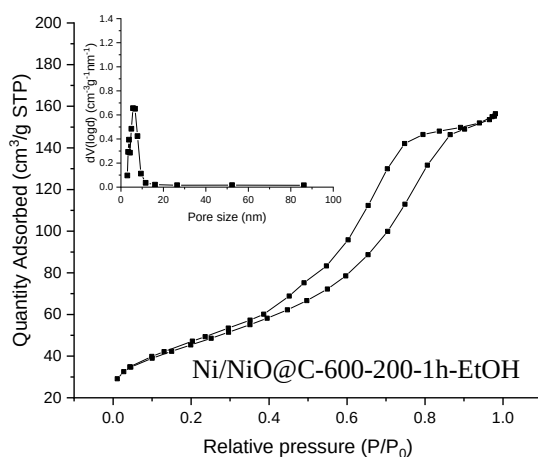
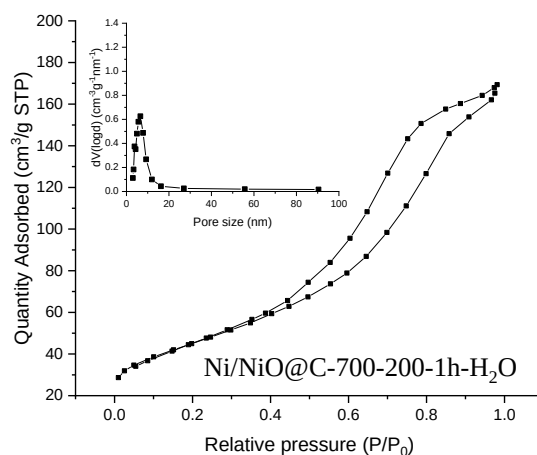
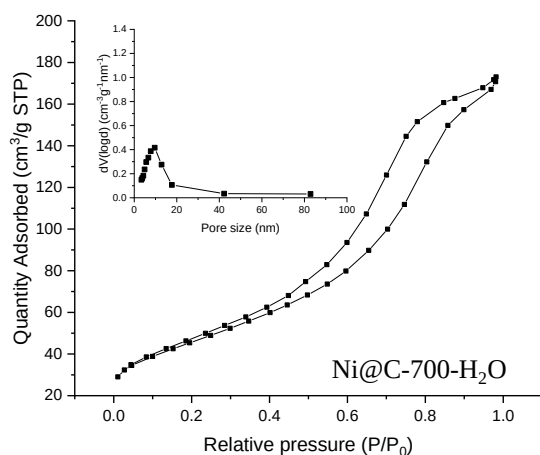
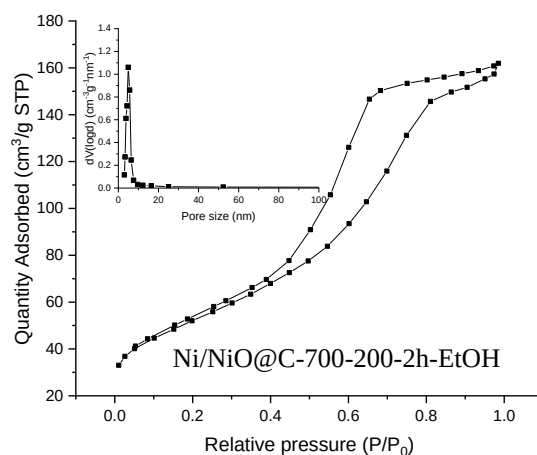
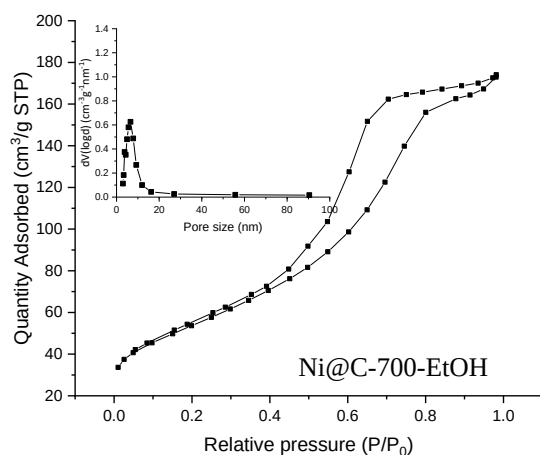


Ni@C-700-EtOH



Ni/NiO@C-700-200-2h-EtOH

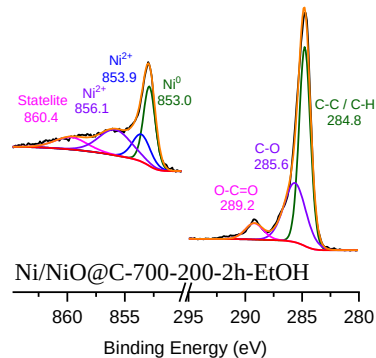
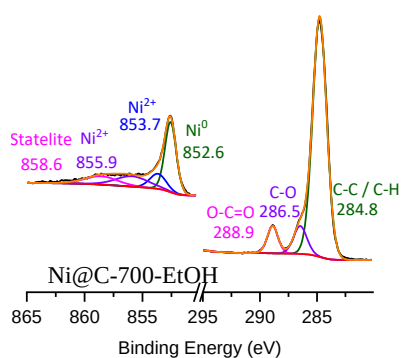
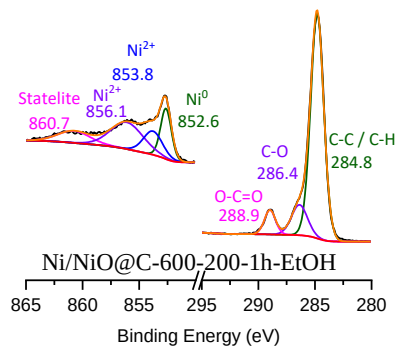
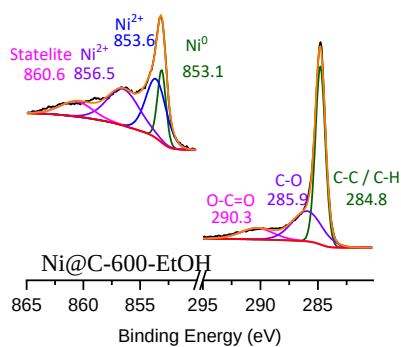
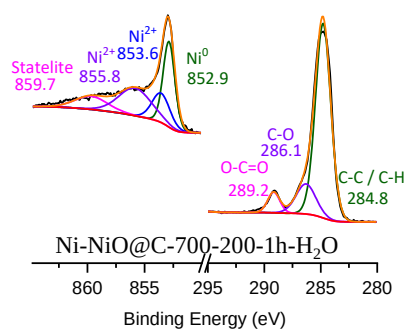
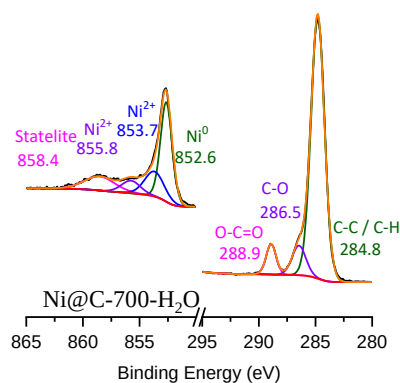
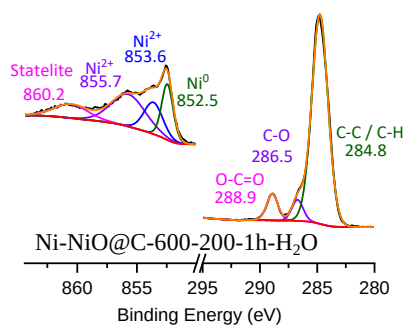
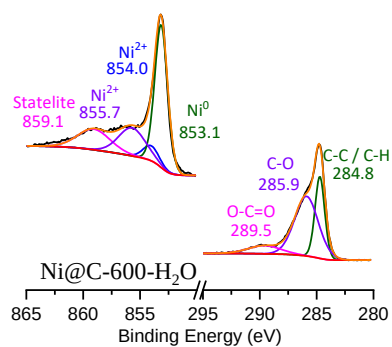
S5. BET measurements



BET table of selective catalysts

Catalyst	Pore size (nm)	Surface area (m ² /g)	Total pore volume (cm ³ /g) micro- and meso-
Ni@C-700-H ₂ O	6.543	161.937	0.268
Ni@C-700-EtOH	5.609	182.051	0.270
Ni/NiO@C-700-200-1h-H ₂ O	5.625	160.087	0.262
Ni/NiO@C-600-200-1h-EtOH	5.678	160.172	0.242
Ni/NiO@C-700-200-2h-EtOH	5.131	223.912	0.287

S6. XPS spectra

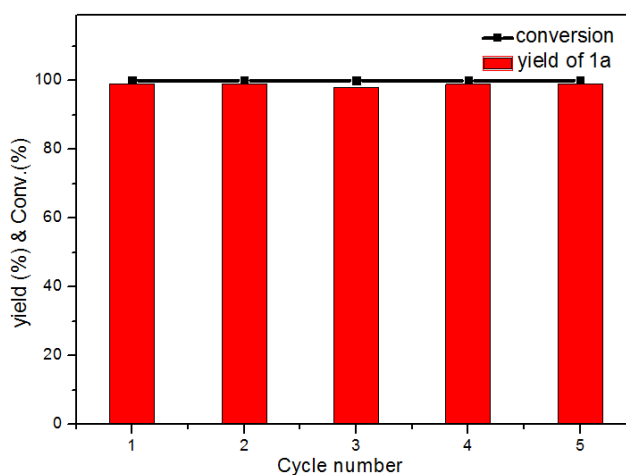


XPS peak assignment							
Catalyst	C1s			Ni2p _{3/2}			Statelite
	C-C/C-H	C-O	O-C=O	Ni ⁰	Ni ²⁺	Ni ²⁺	
Ni@C-600-H ₂ O	284.8	285.9	289.5	853.1	854.0	855.7	859.1
Ni/NiO@C-600-200-1h-H ₂ O	284.8	286.5	288.9	852.5	853.6	855.7	860.2
Ni@C-700-H ₂ O	284.8	286.5	288.9	852.6	853.7	855.8	858.4
Ni/NiO@C-700-200-1h-H ₂ O	284.8	286.1	289.2	852.9	853.6	855.8	859.7
Ni@C-600-EtOH	284.8	285.9	290.3	853.1	853.6	856.5	860.6
Ni/NiO@C-600-200-1h-EtOH	284.8	286.4	288.9	852.6	853.8	856.1	860.7
Ni@C-700-EtOH	284.8	286.5	288.9	852.6	853.7	855.9	858.6
Ni/NiO@C-700-200-2h-EtOH	284.8	286.5	289.2	853.0	853.9	856.1	860.4

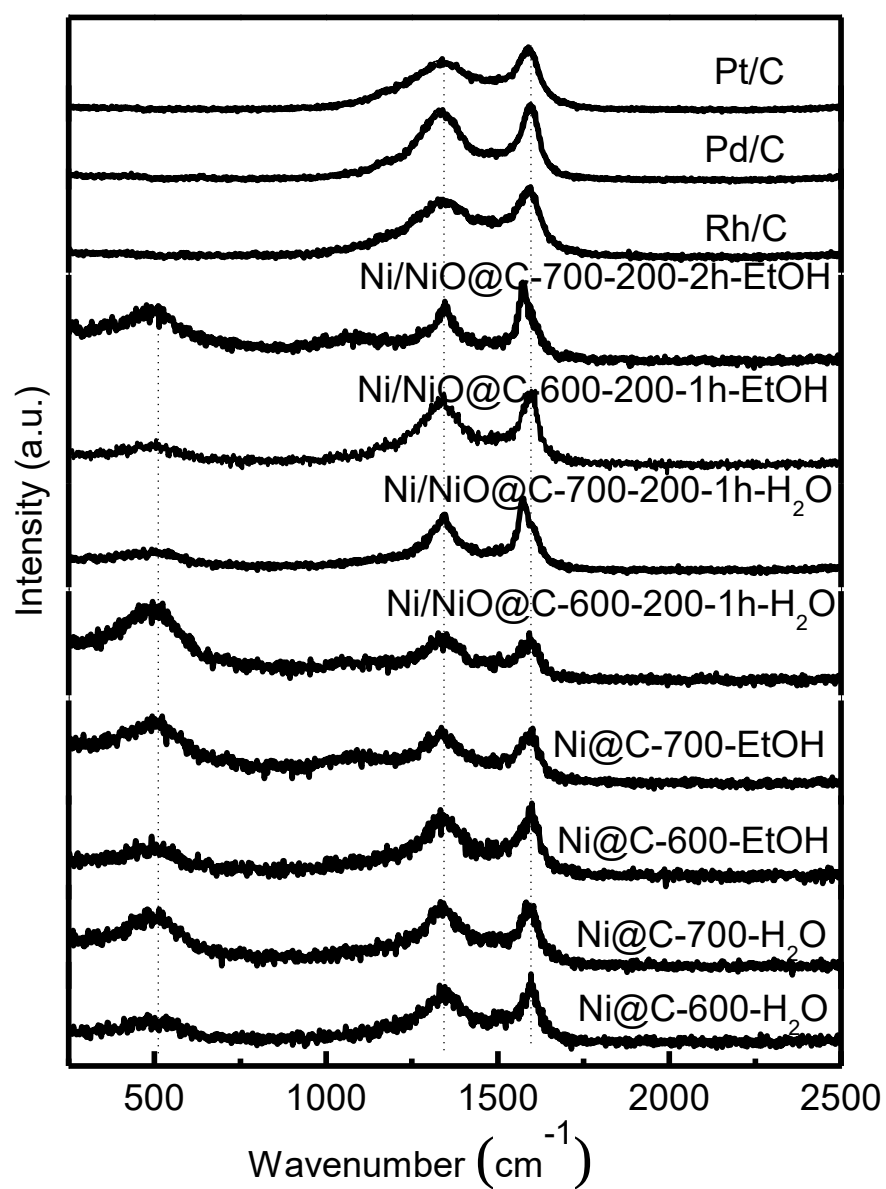
S7. EDS element

Number	Catalyst	Ni(%)	C(%)	O(%)	S(%)
A	Ni@C-600-H ₂ O	37.83	58.39	3.78	0
B	Ni@C-700-H ₂ O	33.61	60.98	5.41	0
C	Ni@C-600-EtOH	29.06	68.12	2.82	0
D	Ni@C-700-EtOH	25.35	68.6	5.32	0.73
E	Ni/NiO@C-600-200-1h-H ₂ O	30.05	60.7	9.24	0
F	Ni/NiO@C-700-200-1h-H ₂ O	22.9	72.54	4.56	0
G	Ni/NiO@C-600-200-1h-EtOH	26.91	59.24	13.84	0
H	Ni/NiO@C-700-200-2h-EtOH	20.78	64.7	14.02	0.5

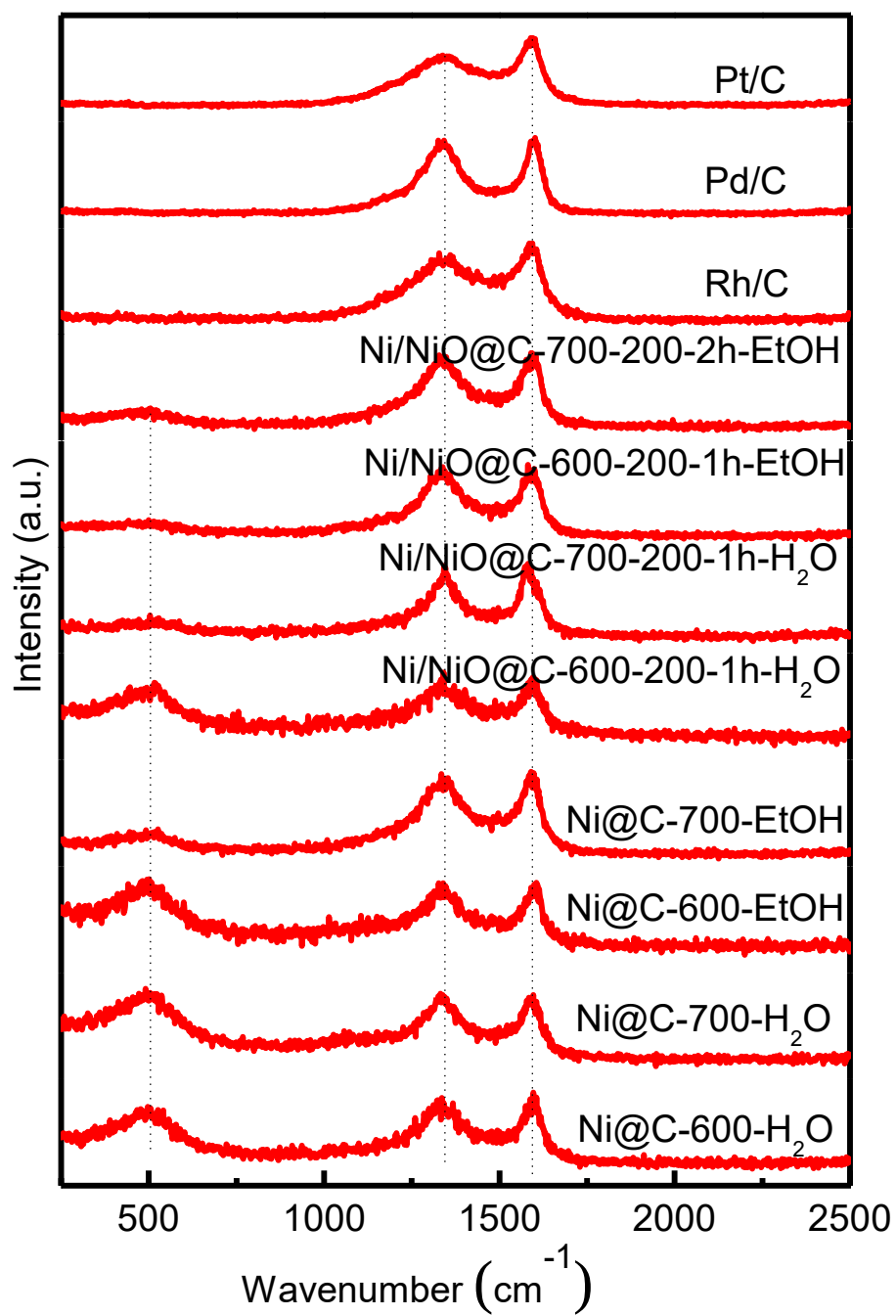
S8. Catalyst recovery using magnetic stir bar and catalytic activity recycle test



S9. Raman spectra of all fresh catalysts used in Table 1 in the manuscript

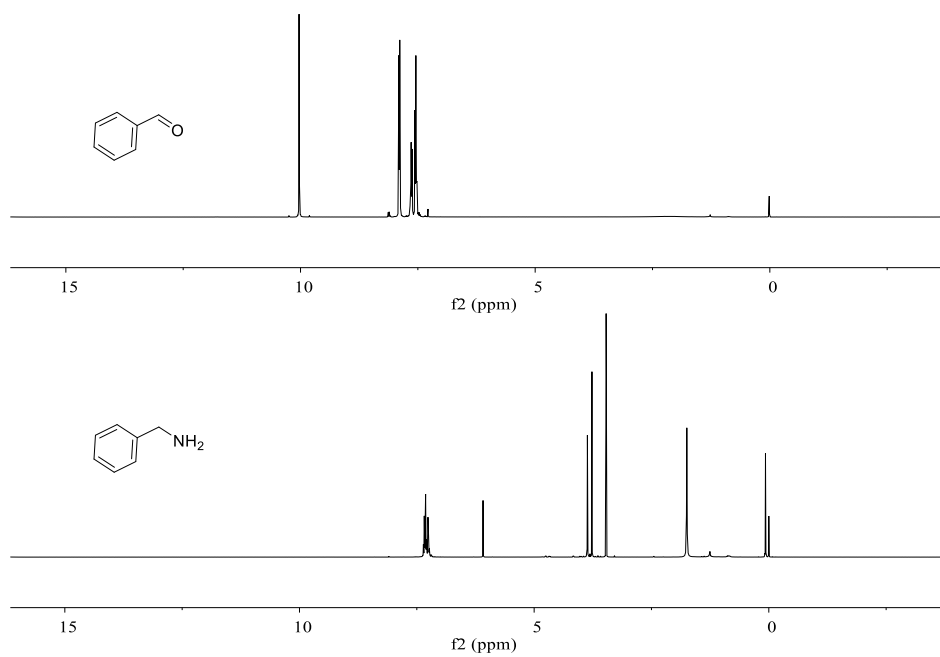


S9. Raman spectra of all catalysts after the reaction used in Table 1 in the manuscript

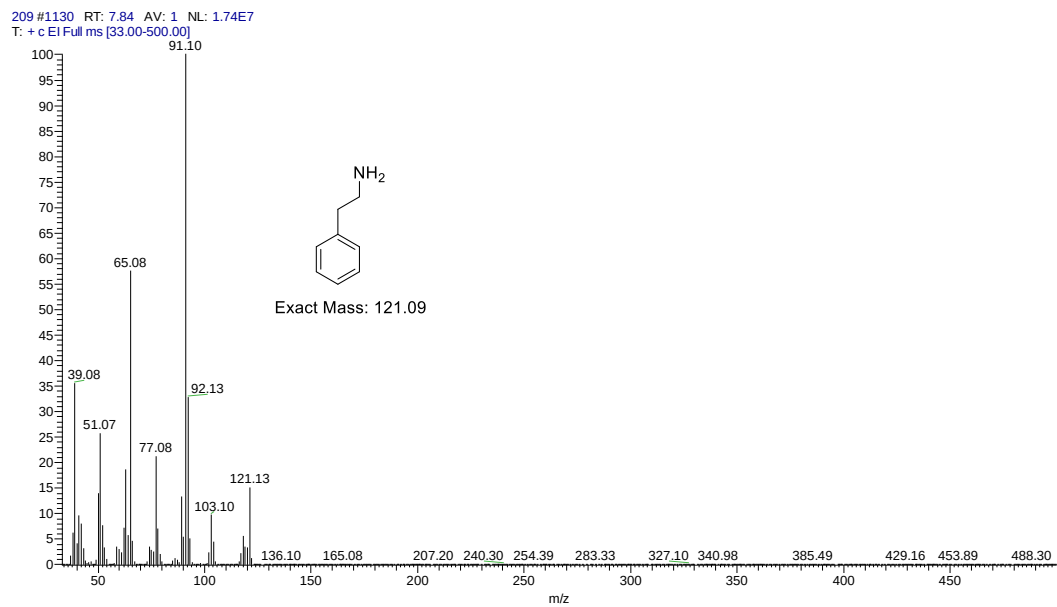
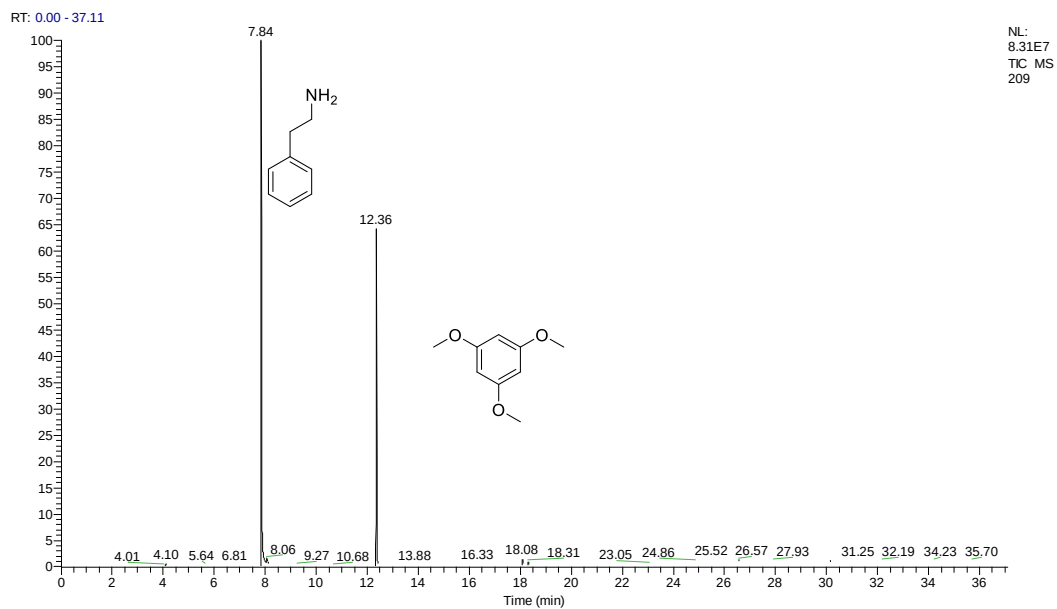
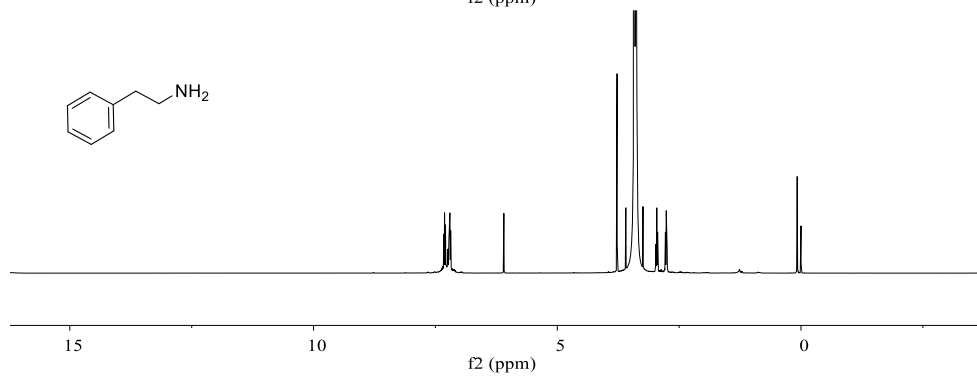
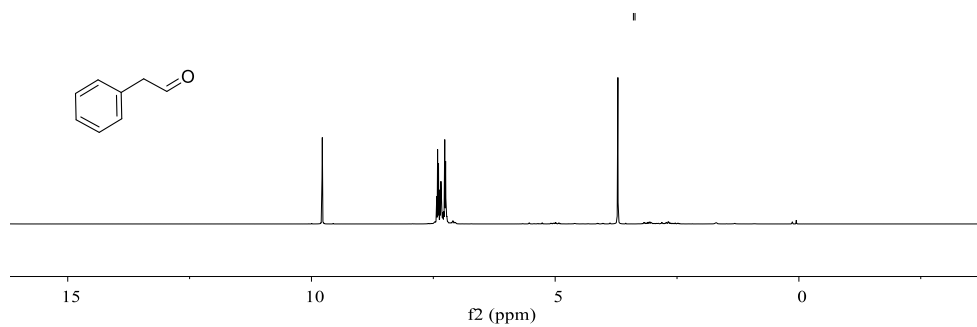


S10. NMR and GCMS spectra

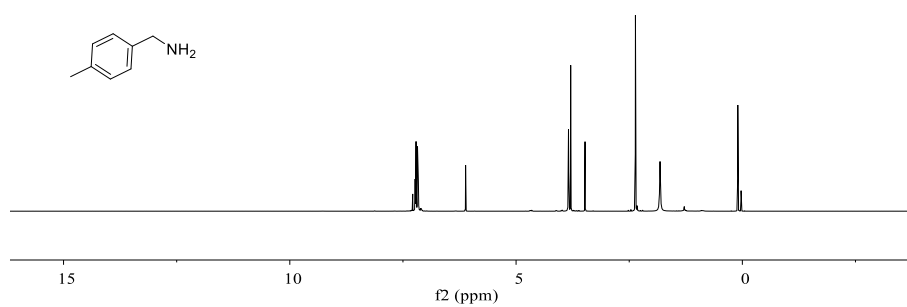
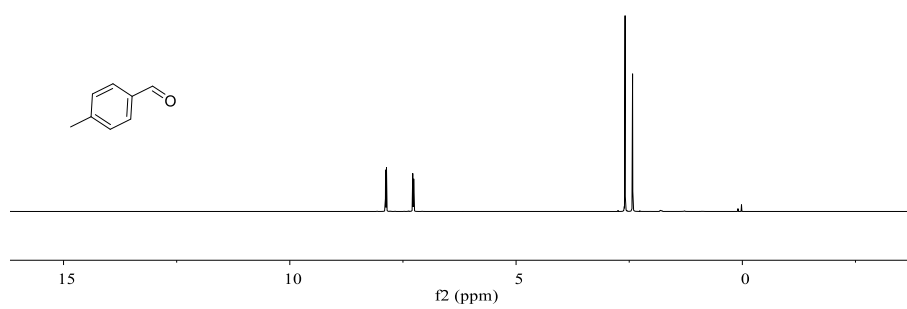
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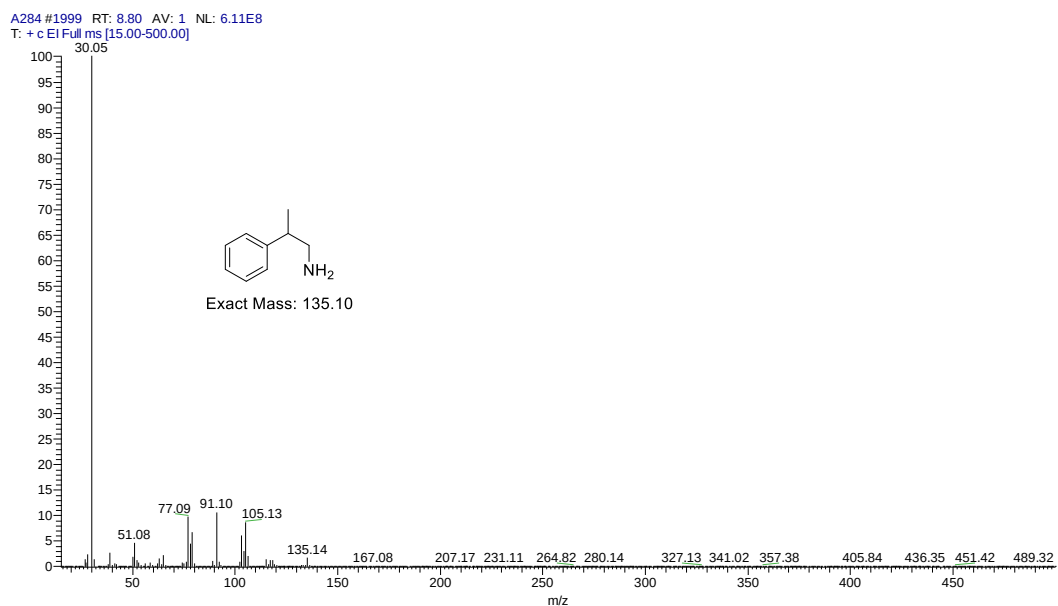
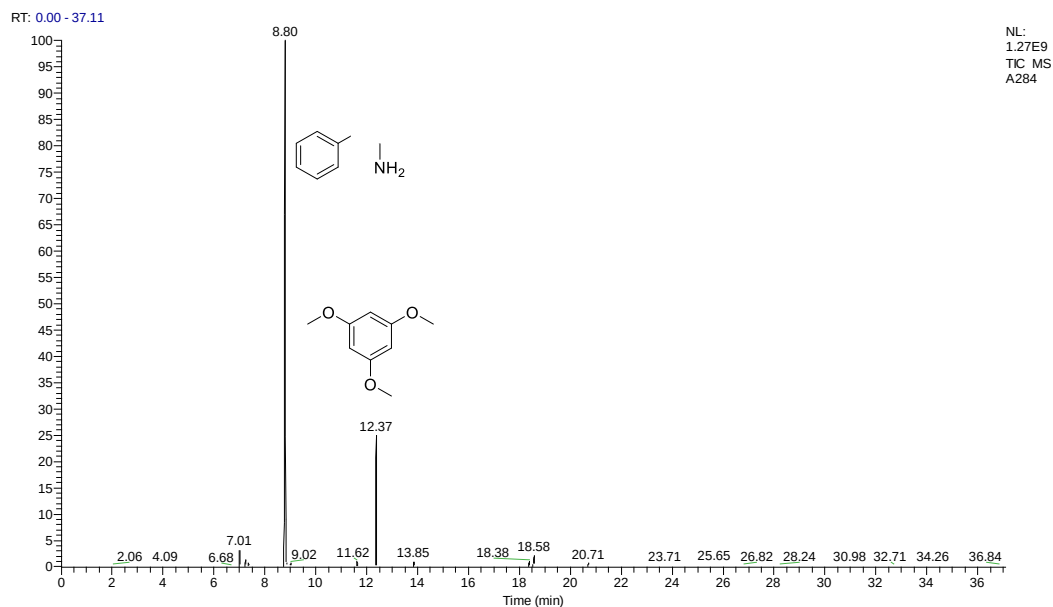
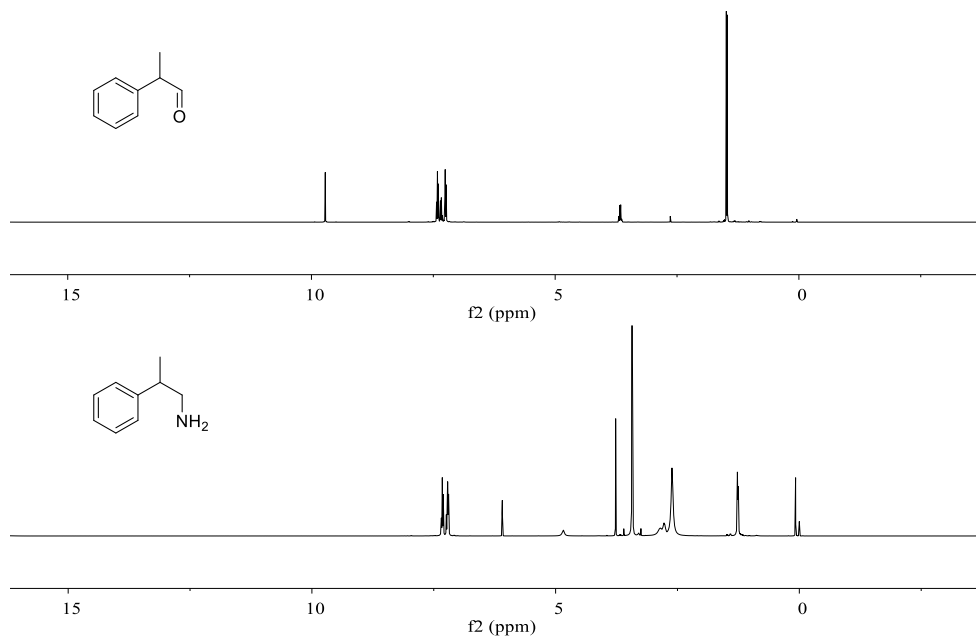
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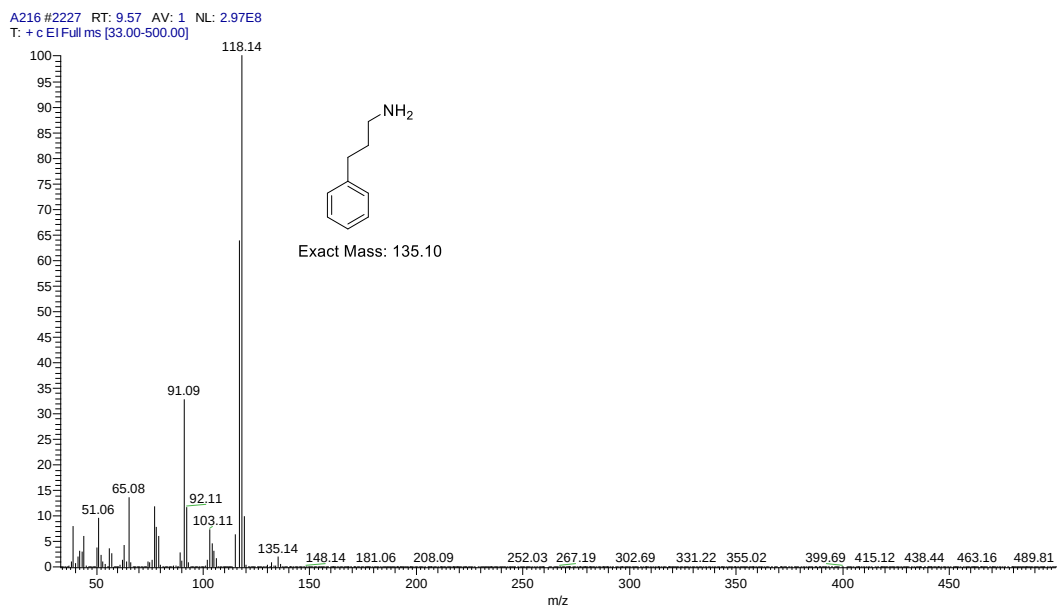
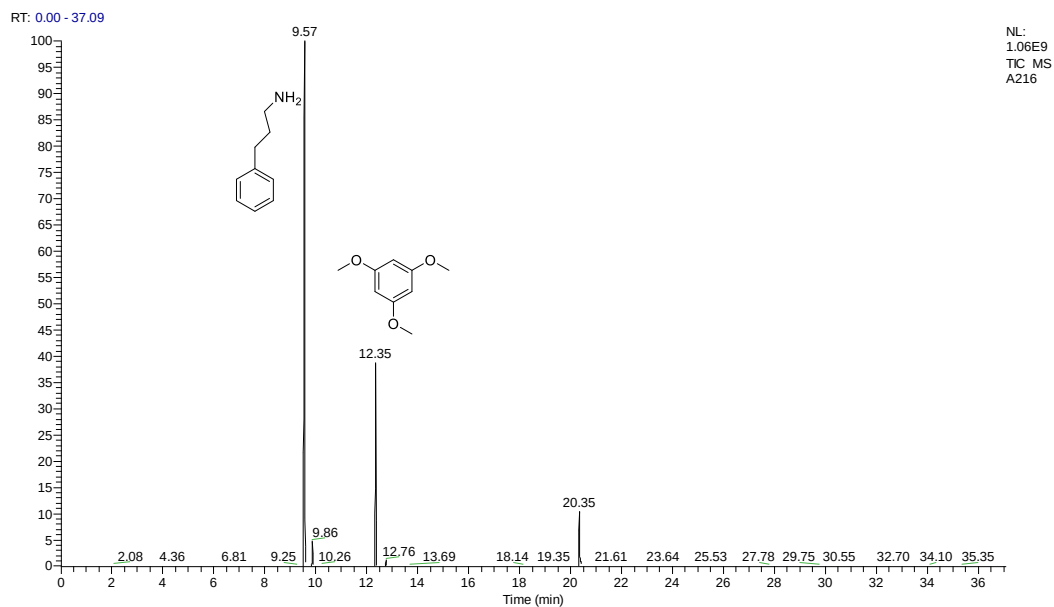
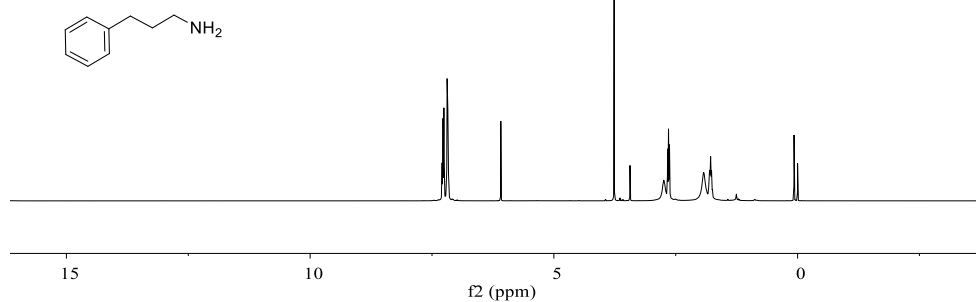
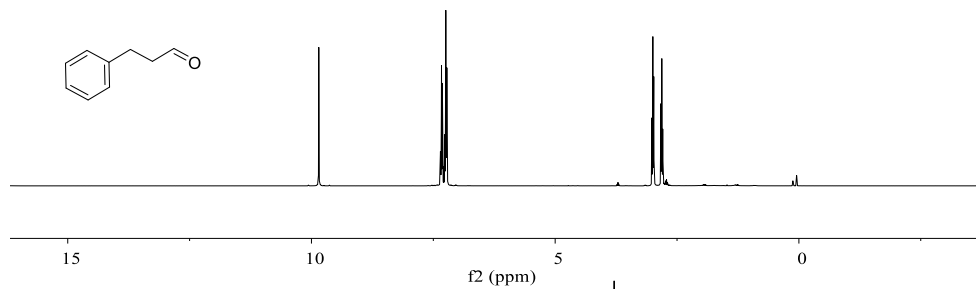
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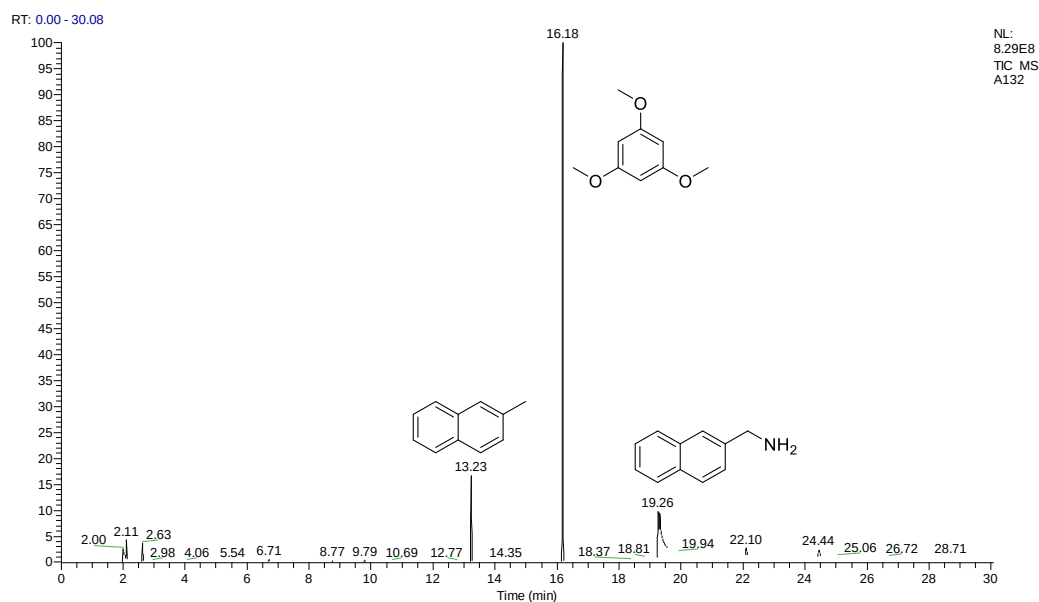
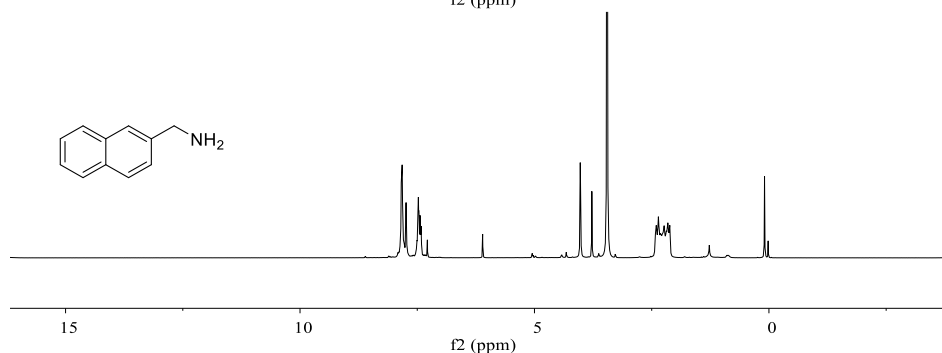
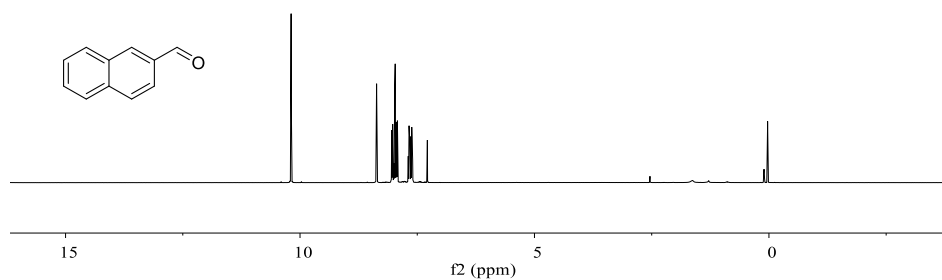
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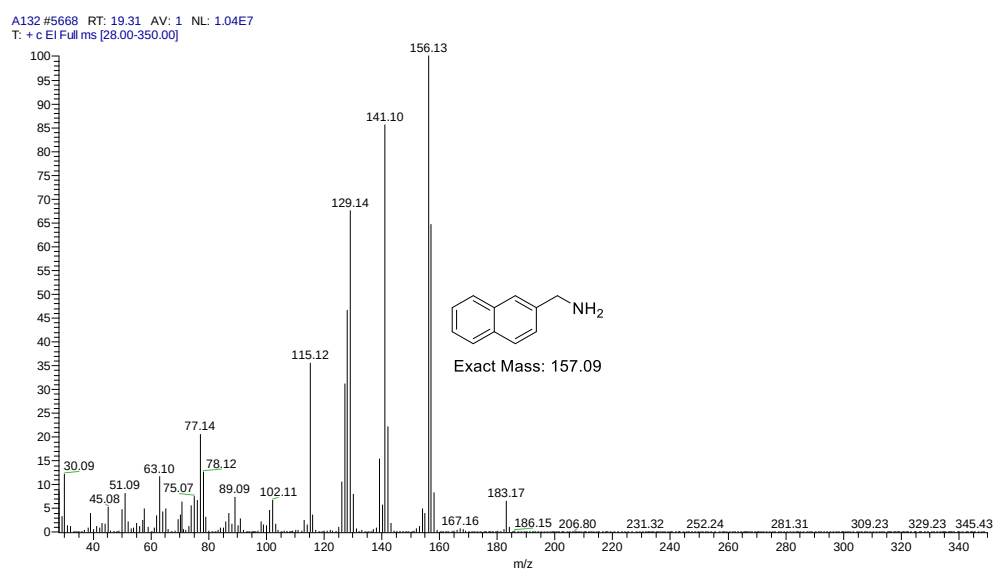
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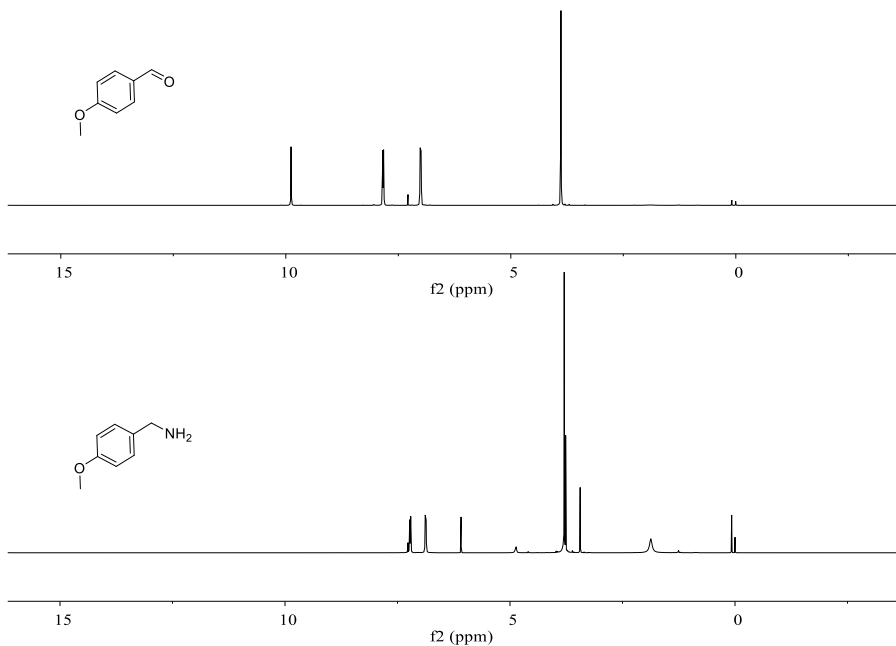
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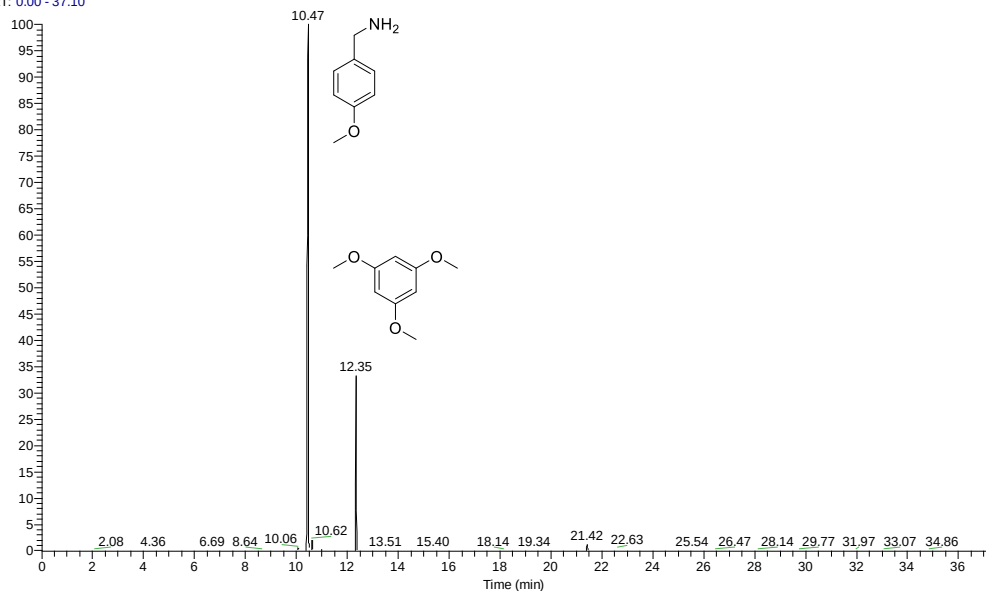
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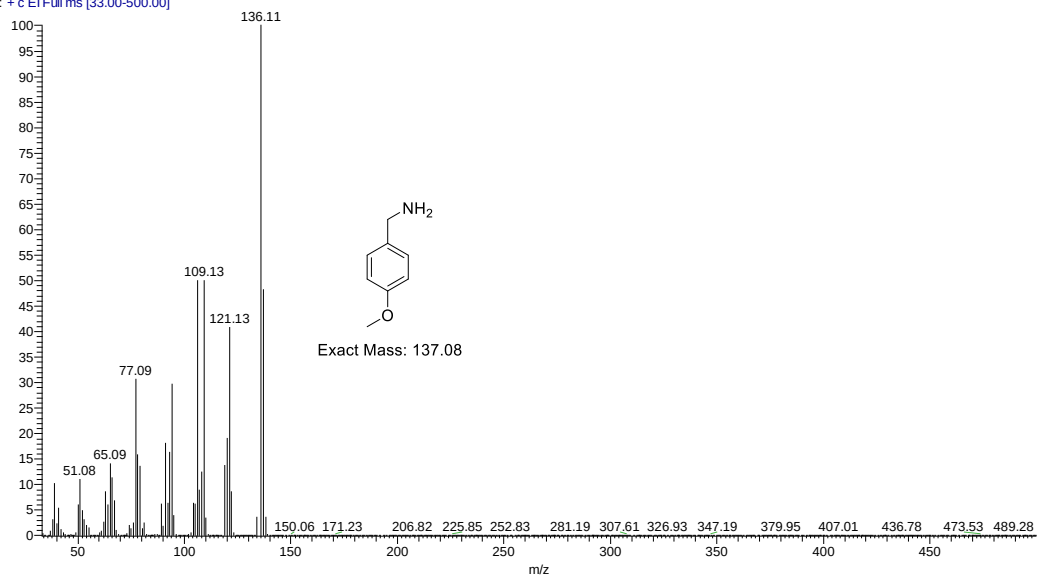


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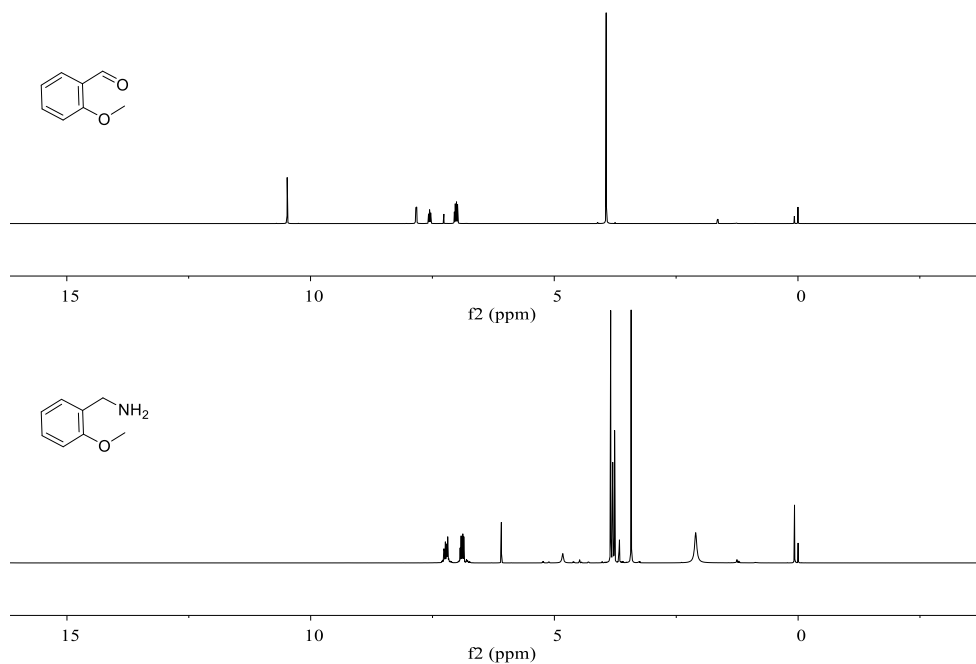


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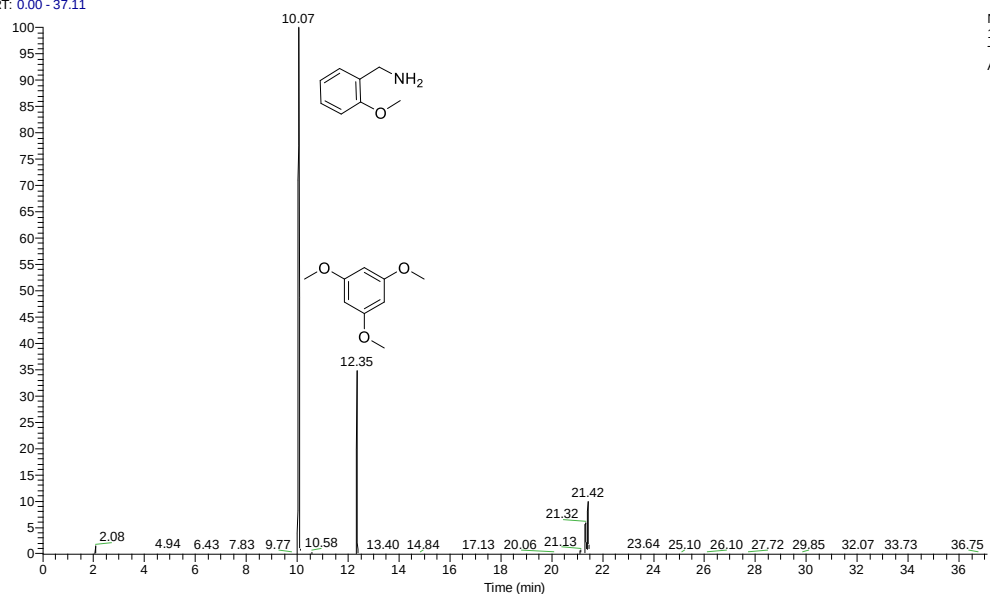
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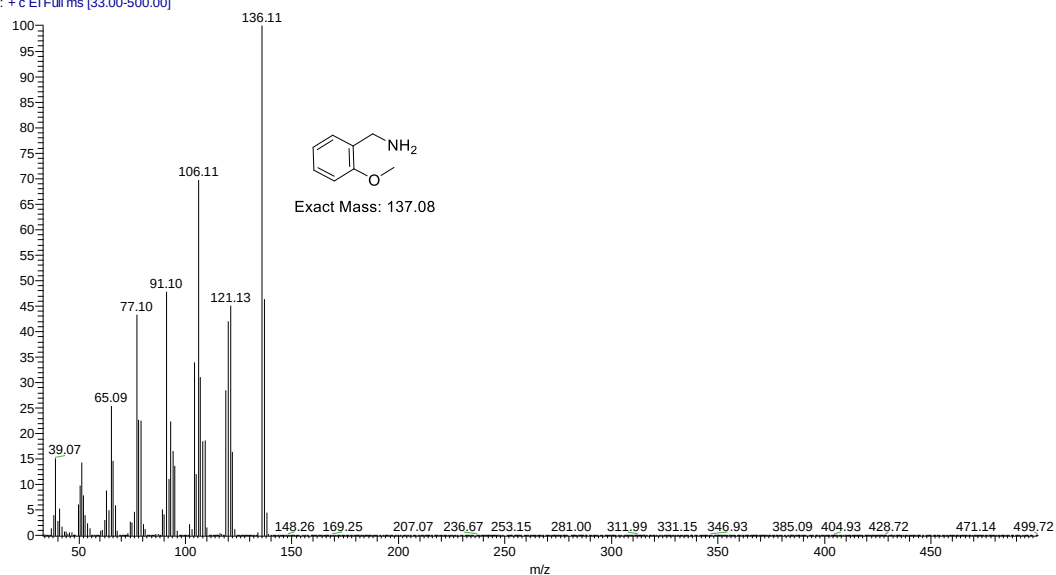


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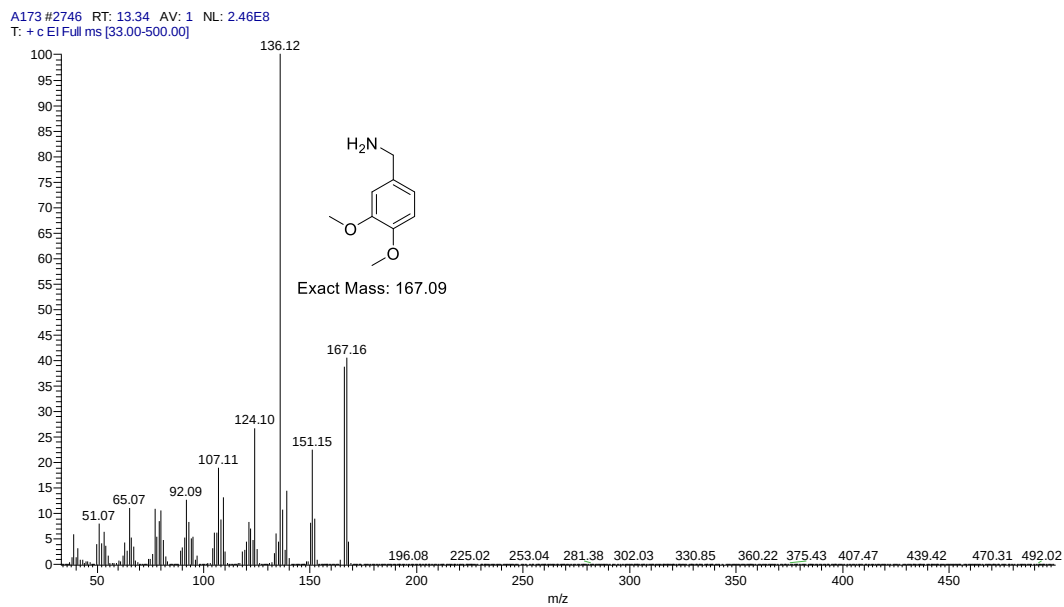
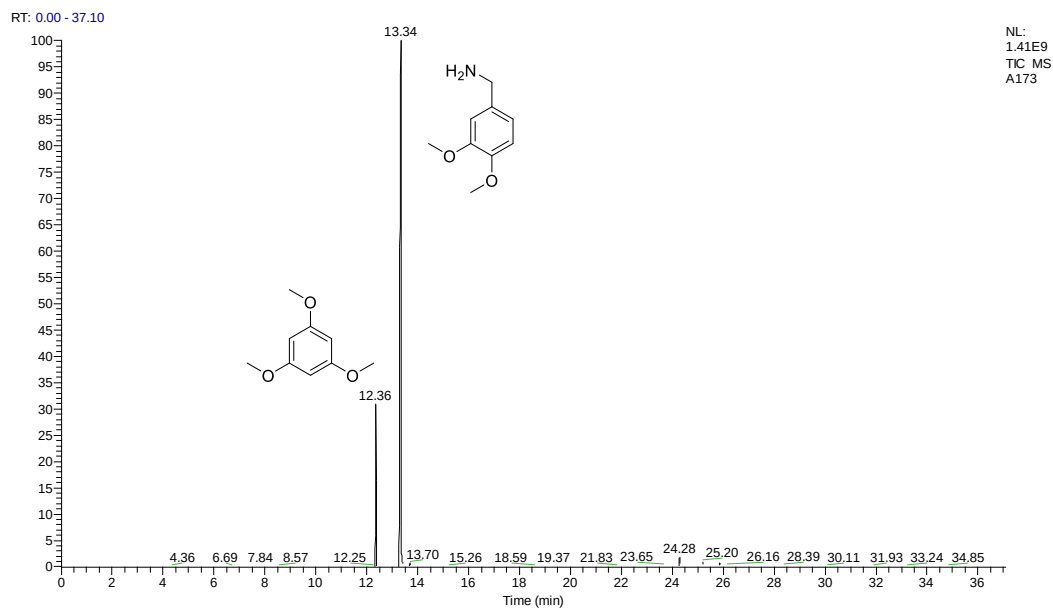
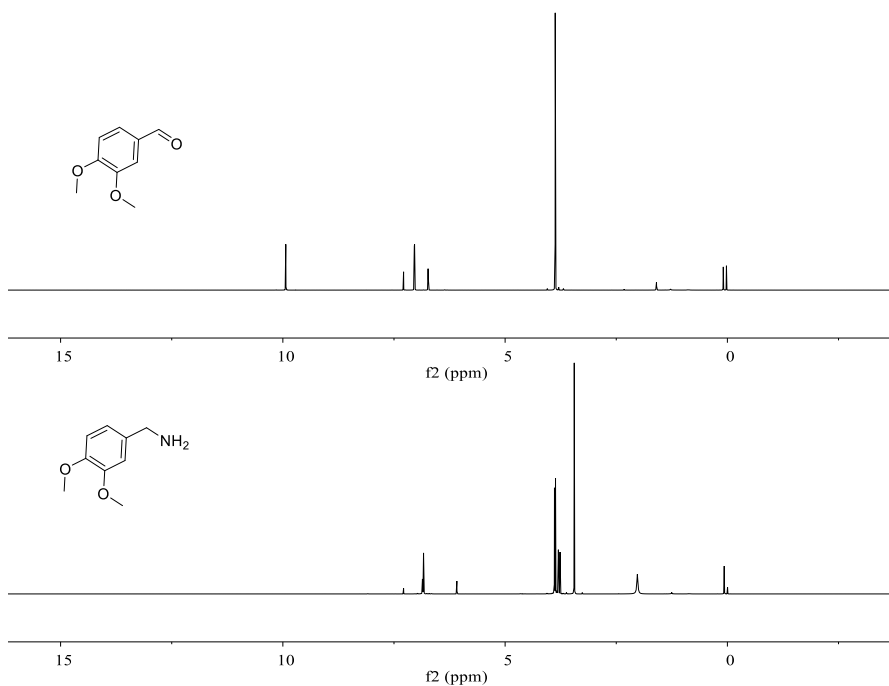


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A214

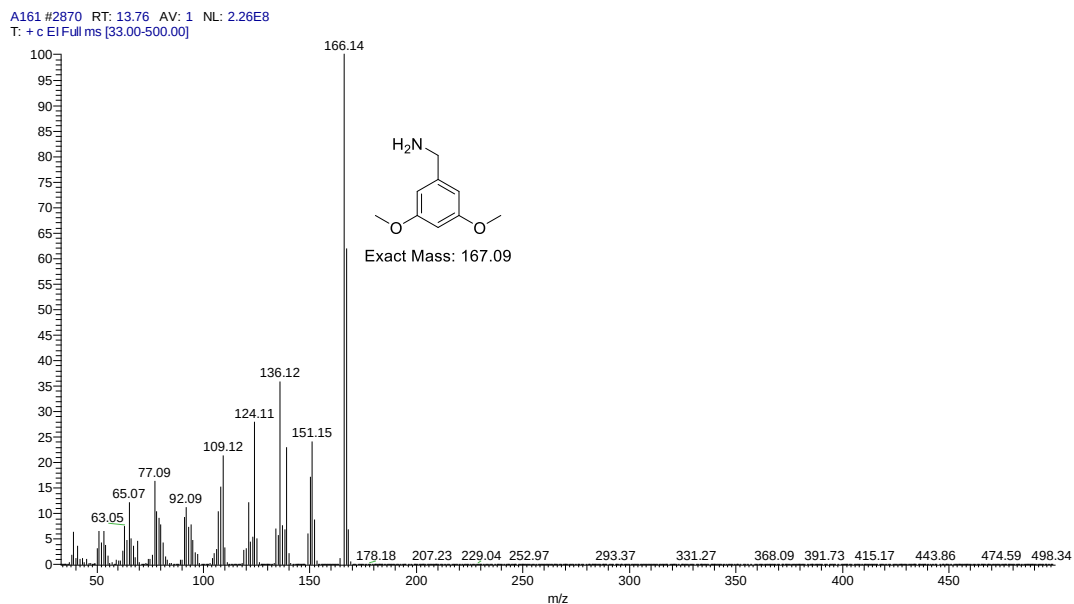
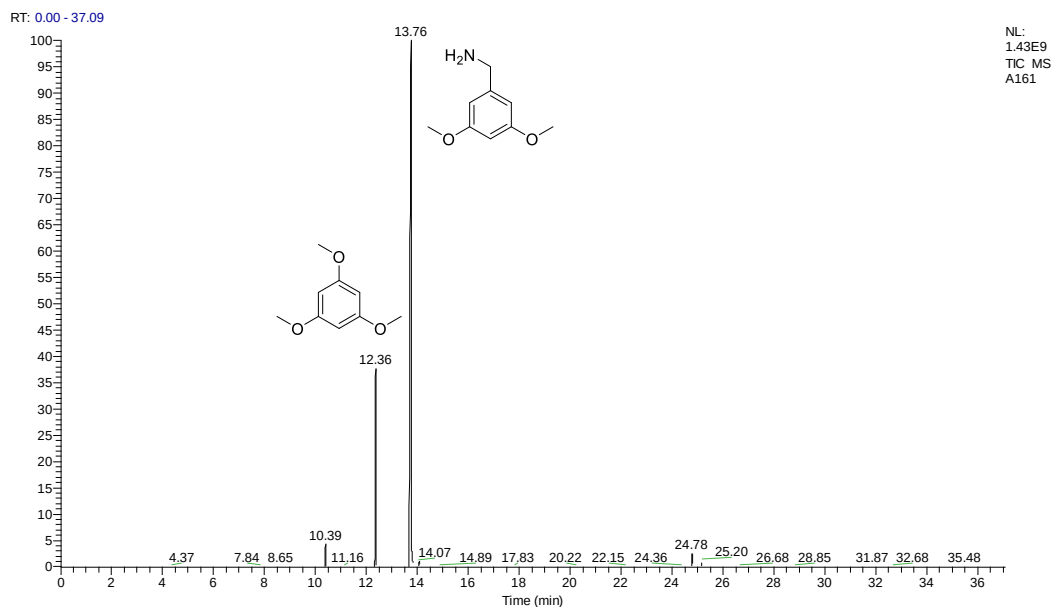
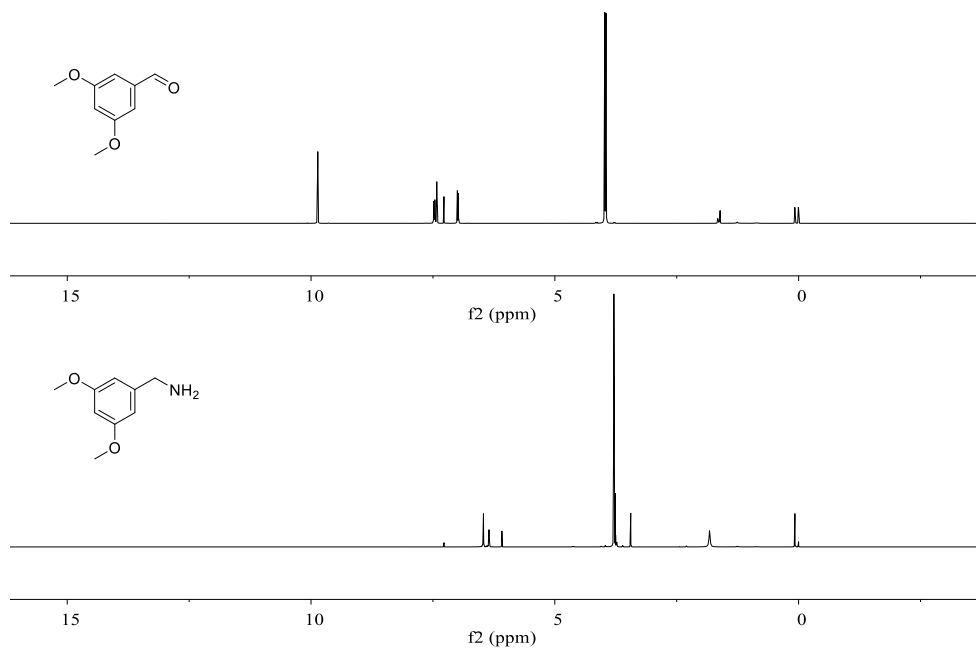
A214 #2374 RT: 10.07 AV: 1 NL: 1.34E8
T: + c EI Full ms [33.00-500.00]



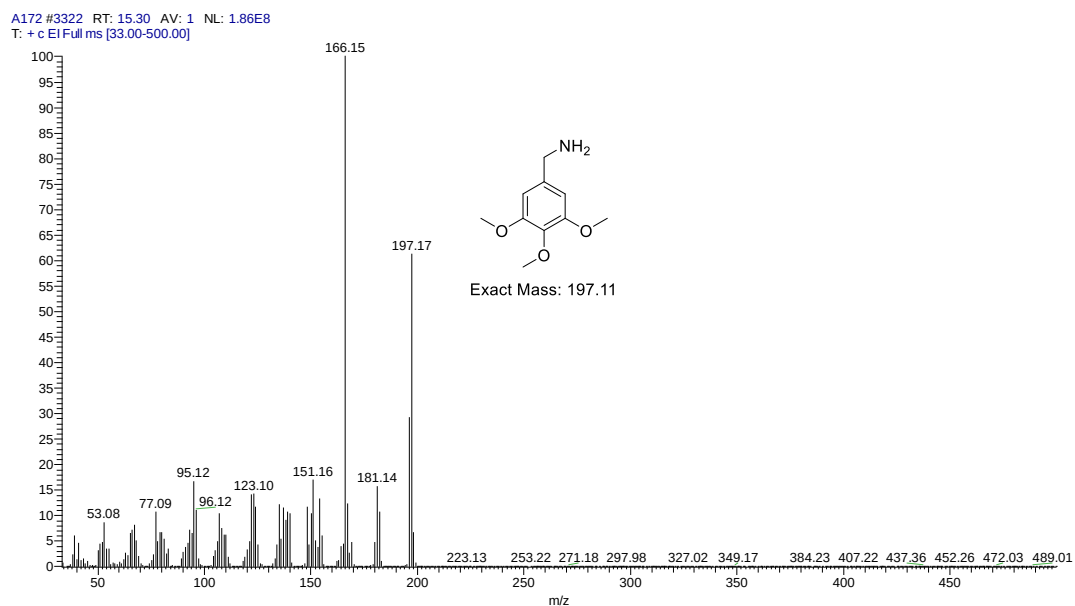
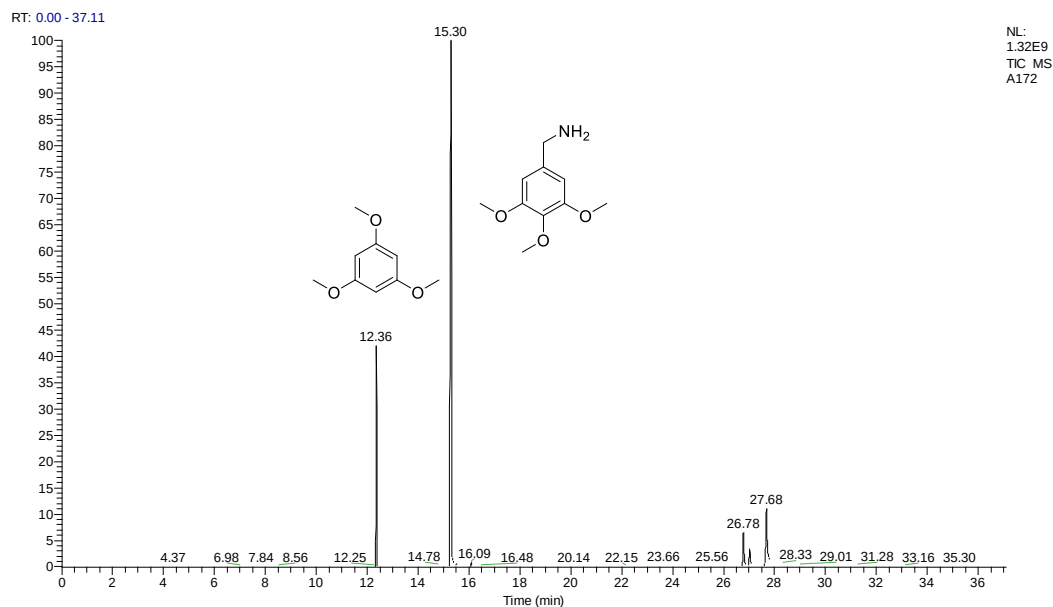
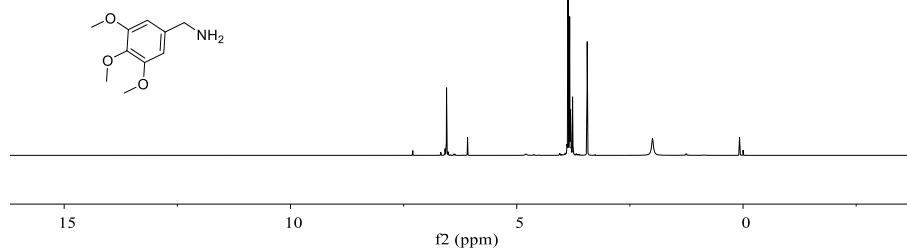
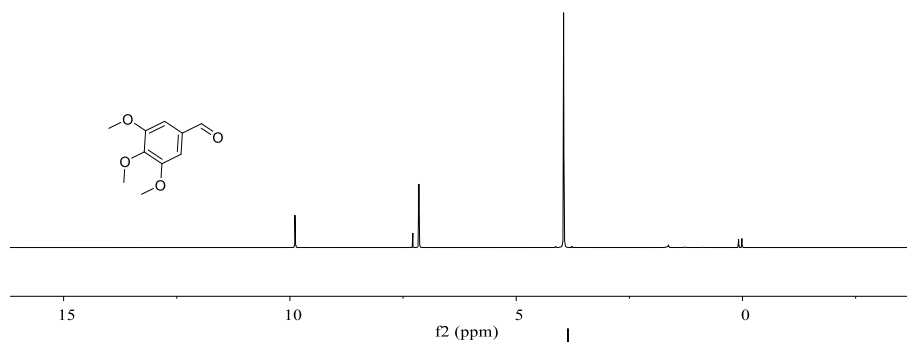
A173



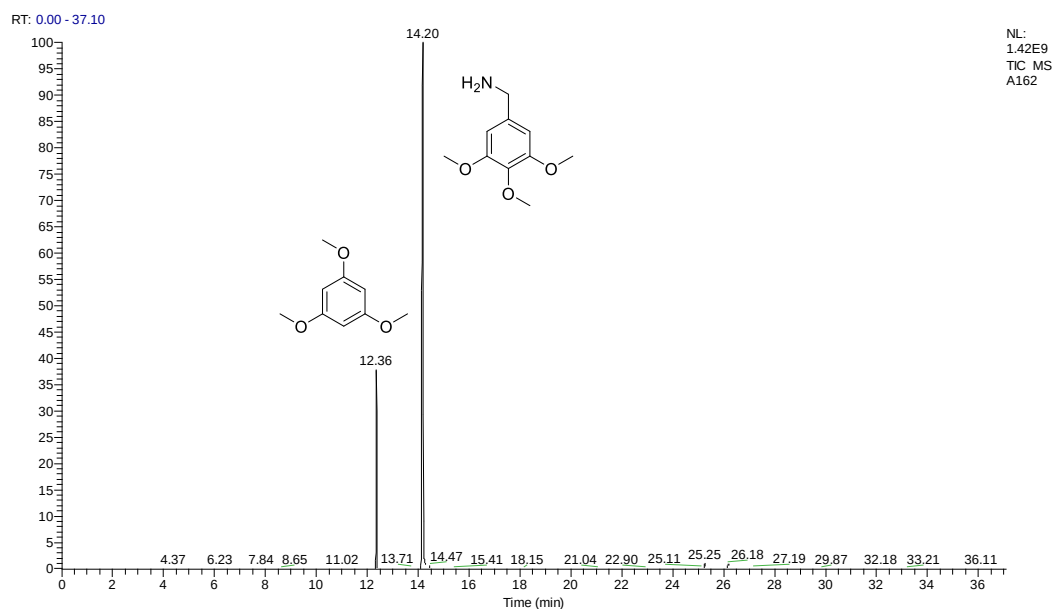
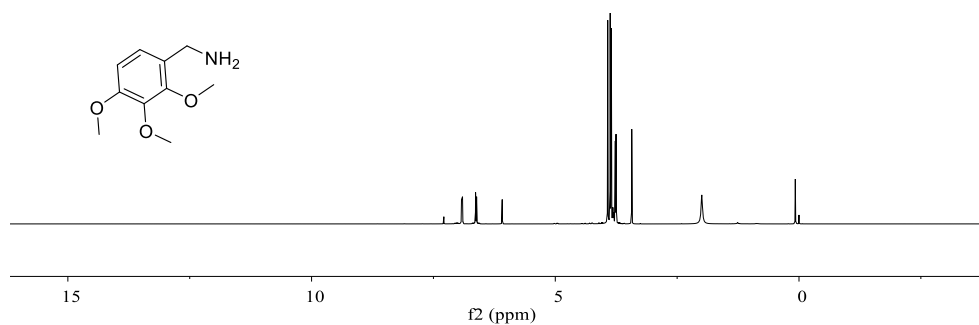
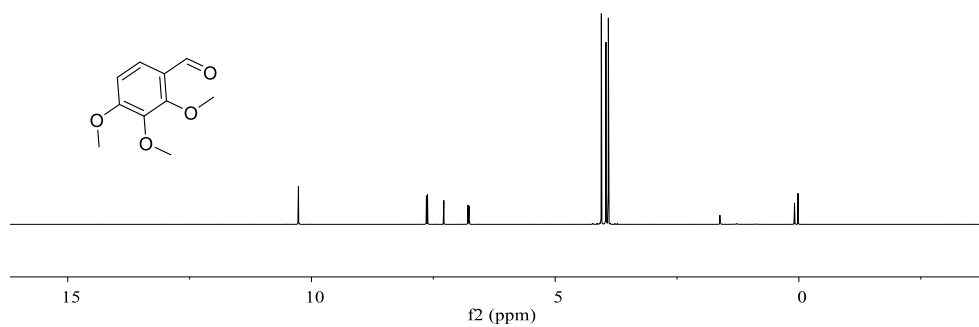
A161



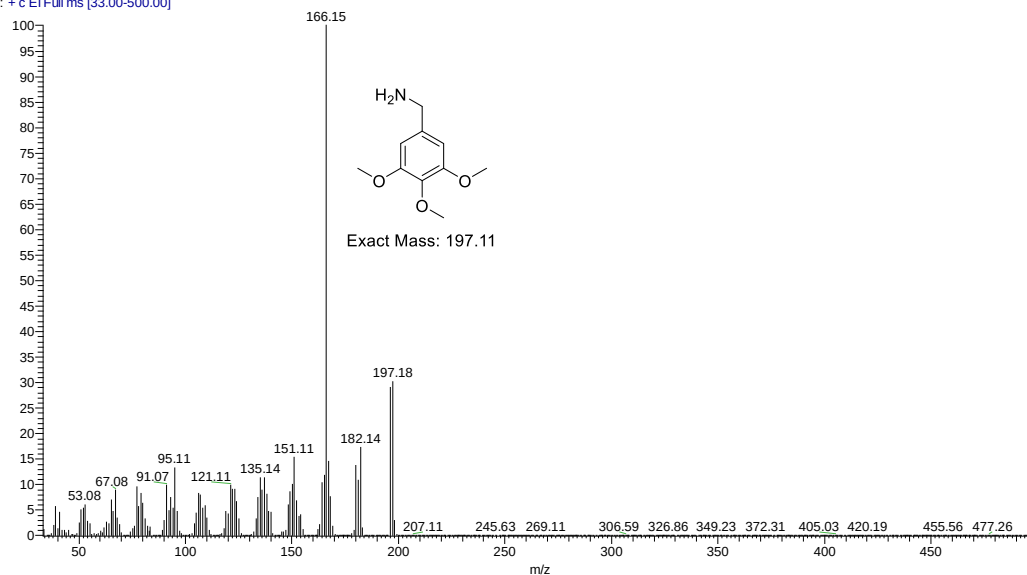
A172



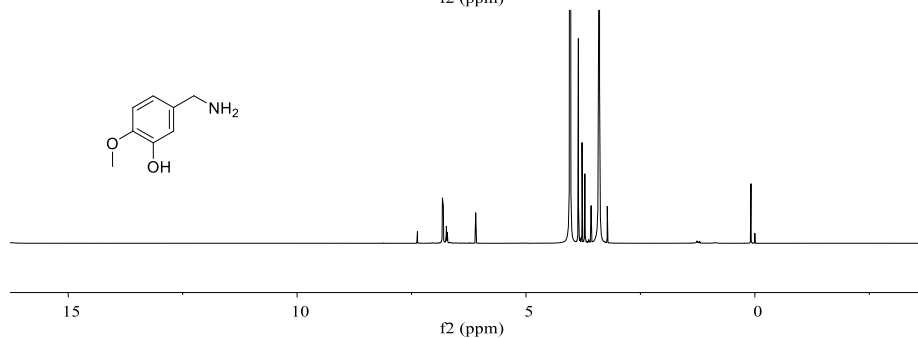
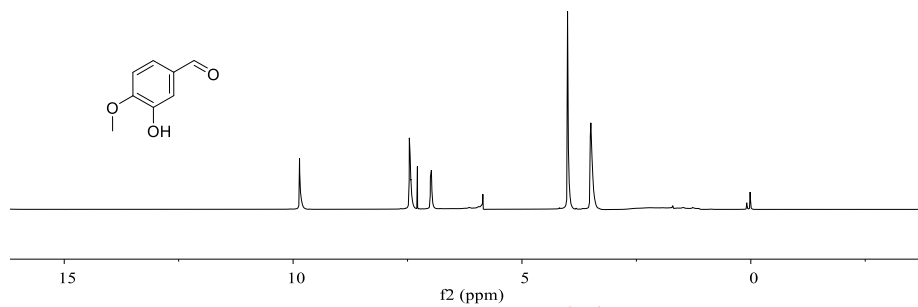
A162



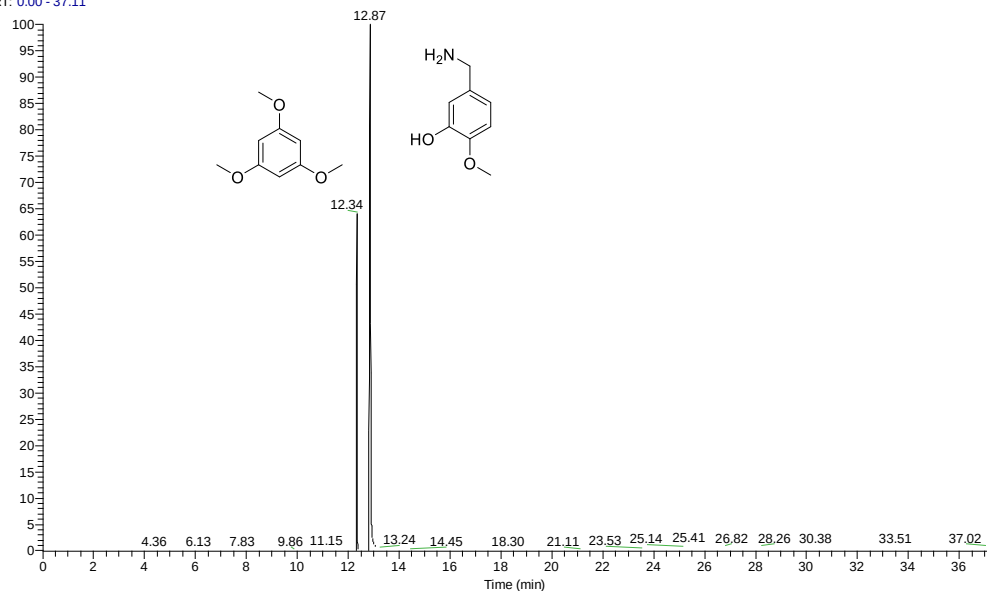
A162 #2999 RT: 14.20 AV: 1 NL: 2.21E8
T: + c EI Full ms [33.00-500.00]



A157

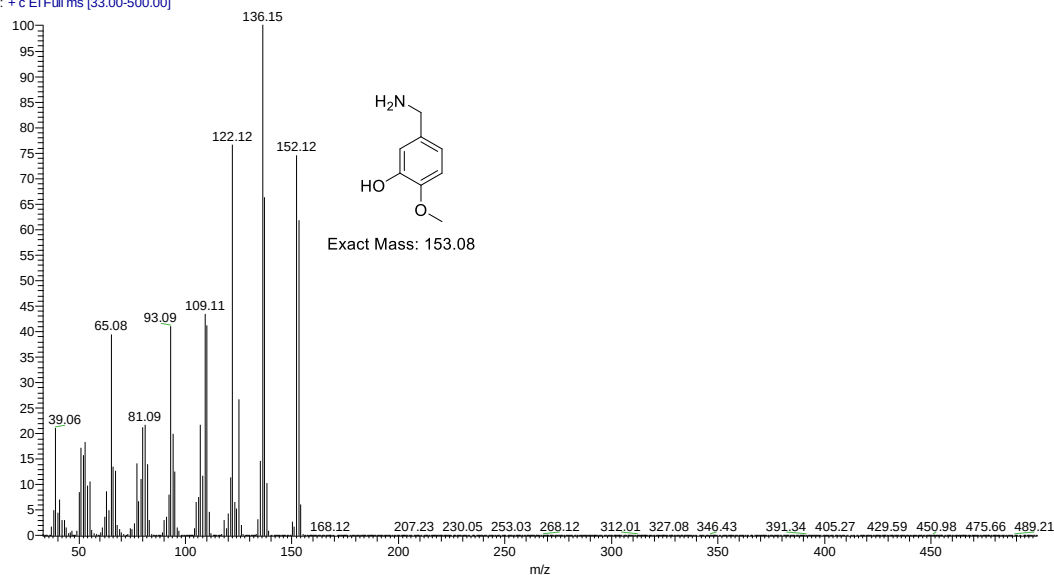


RT: 0.00 - 37.11

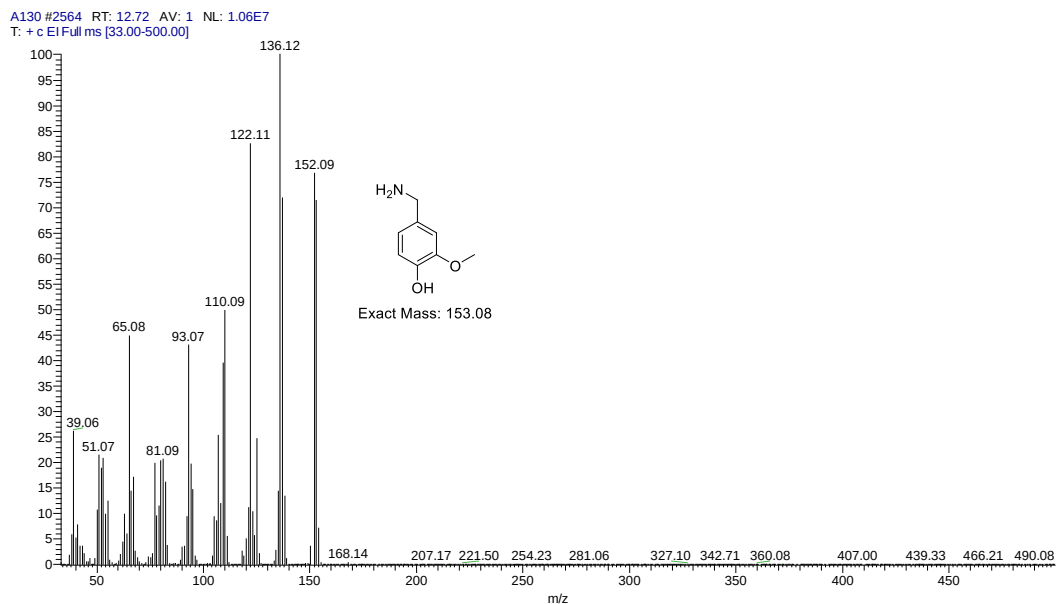
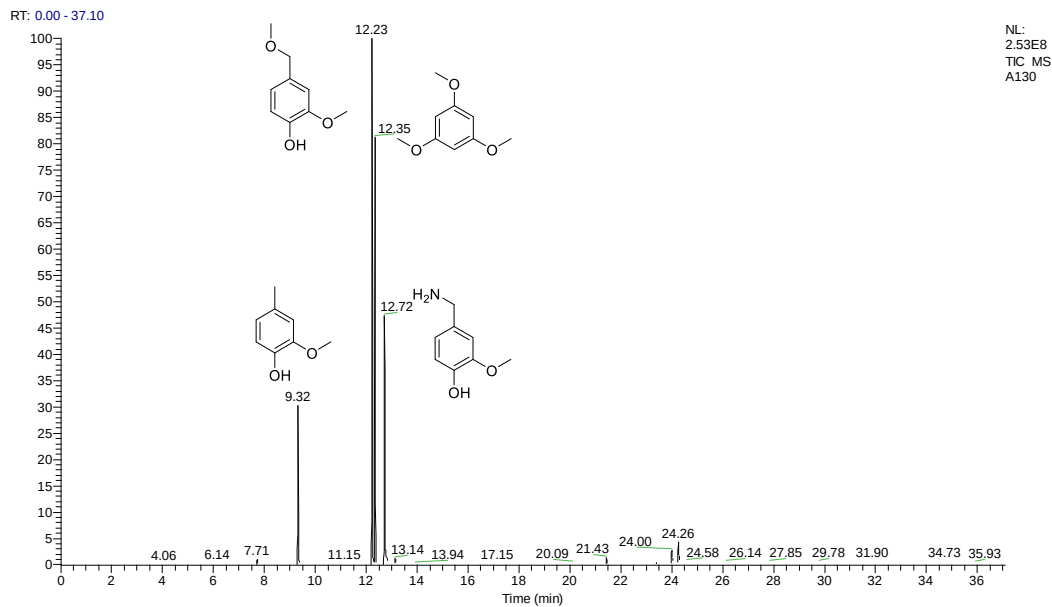
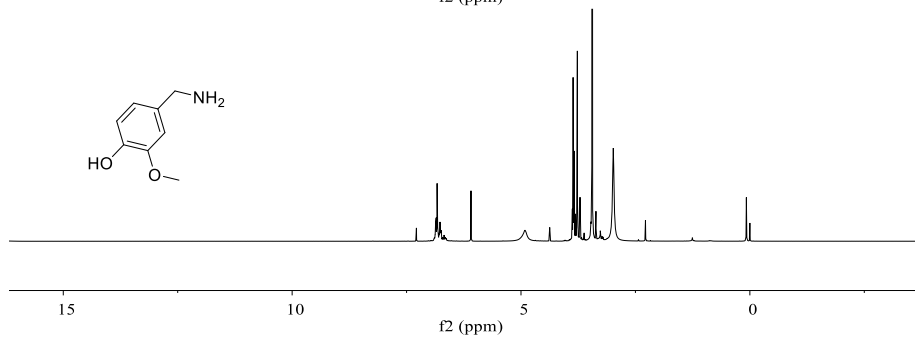
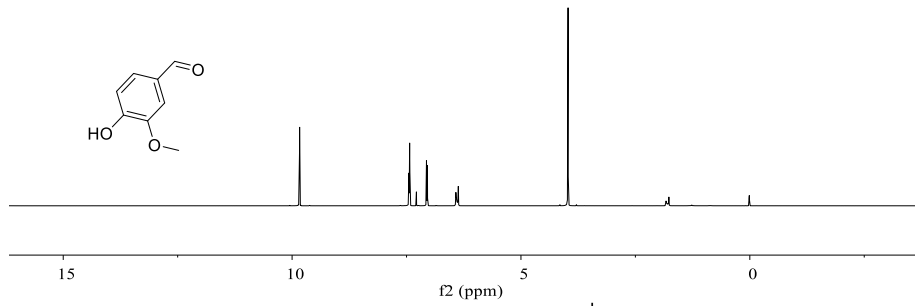


NL:
2.18E8
TIC: MS
A157

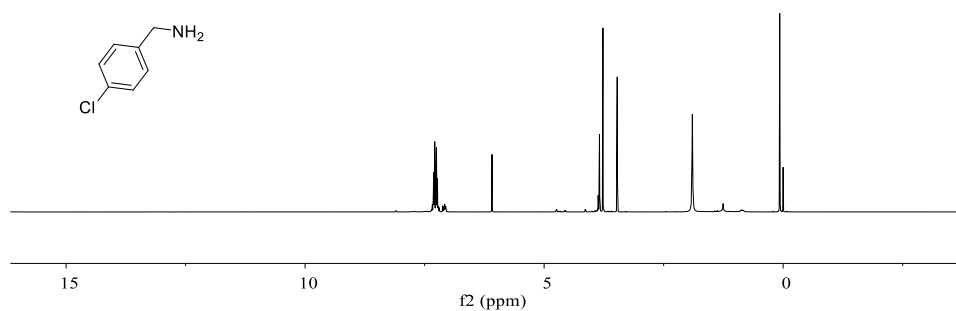
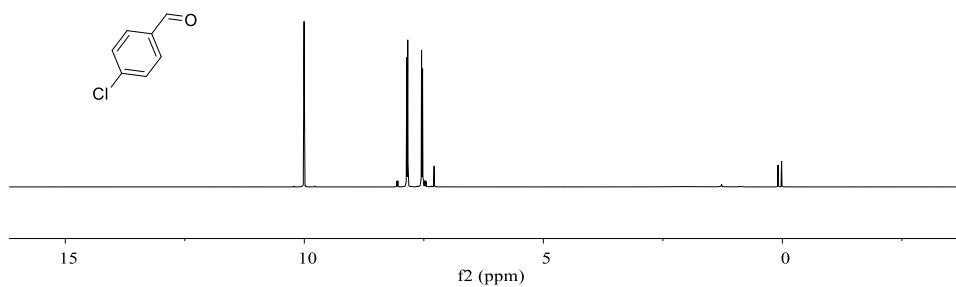
A157 #2608 RT: 12.87 AV: 1 NL: 2.12E7
T: + c EI Full ms [33.00-500.00]



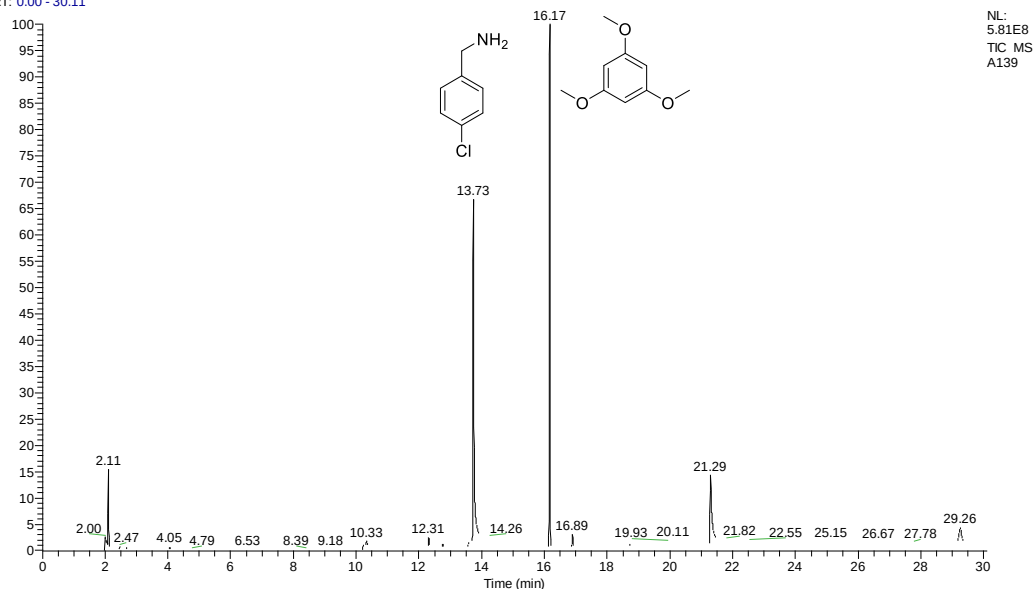
A130



A139

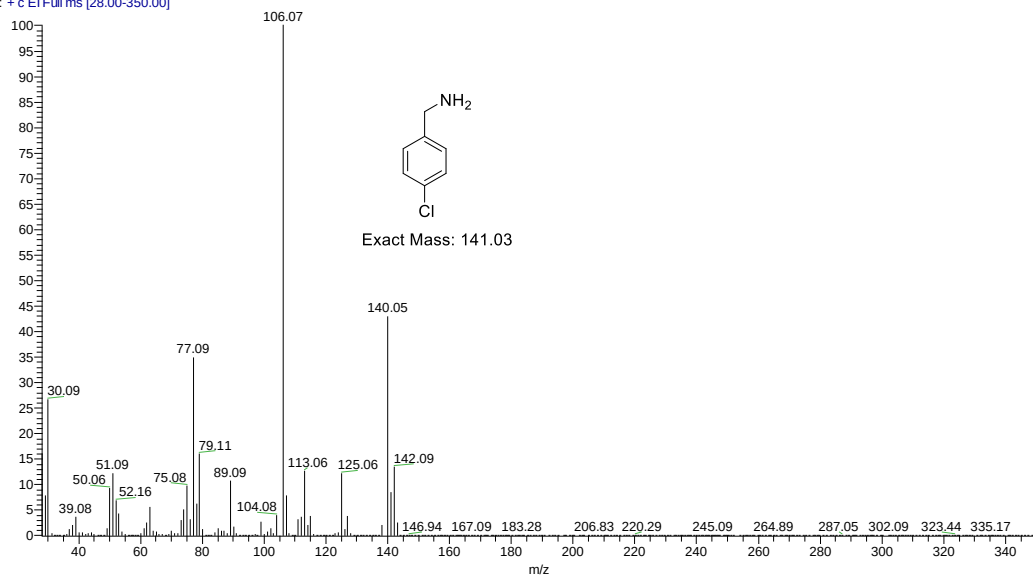


RT: 0.00 - 30.11

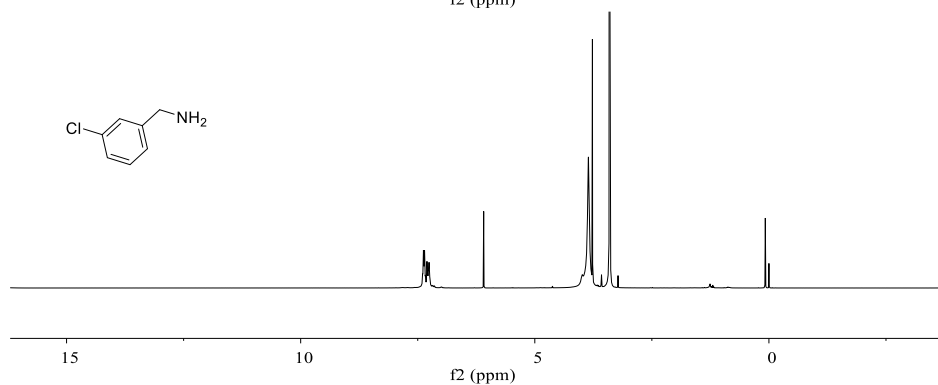
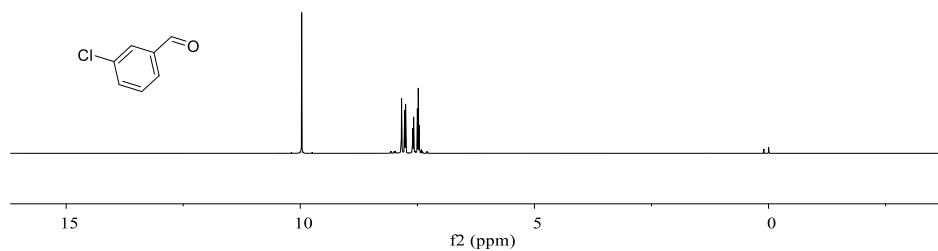


(Innovax column)

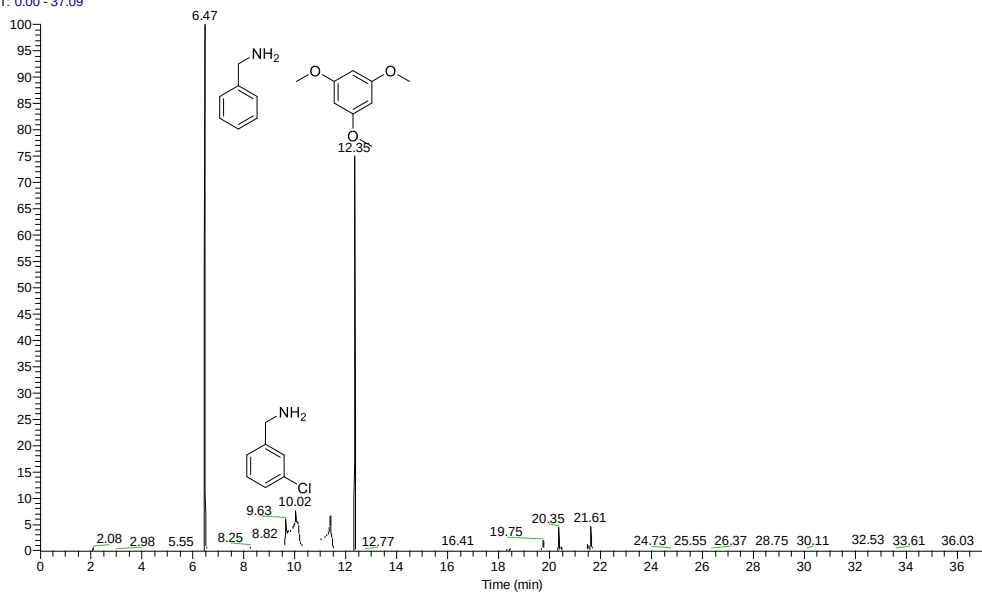
A139 #4029 RT: 13.73 AV: 1 NL: 8.86E7
T: + c EI Full ms [28.00-350.00]



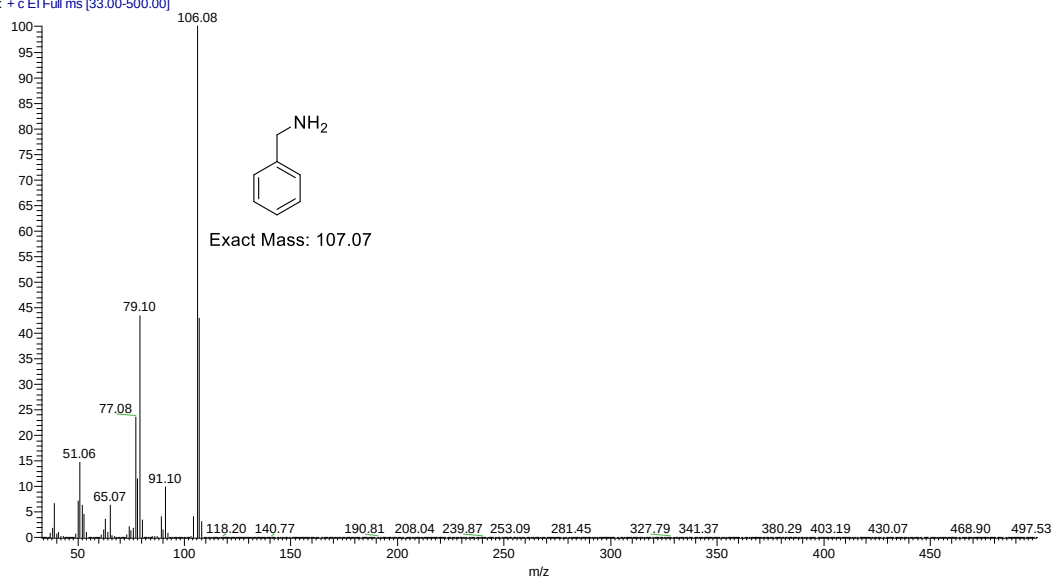
A234



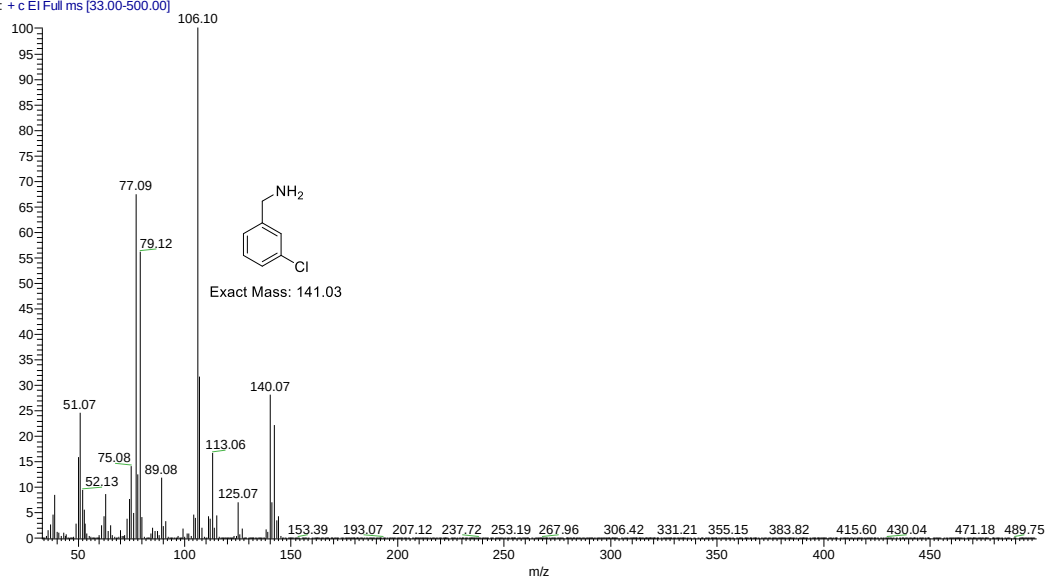
RT: 0.00 - 37.09



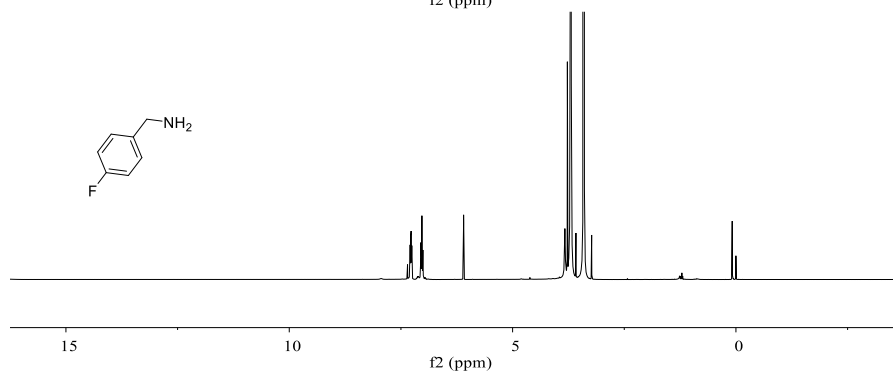
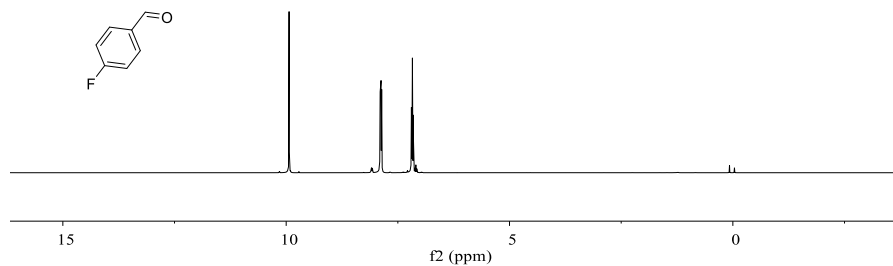
A234 #1314 RT: 6.47 AV: 1 NL: 1.91E8
T: + c EI Full ms [33.00-500.00]



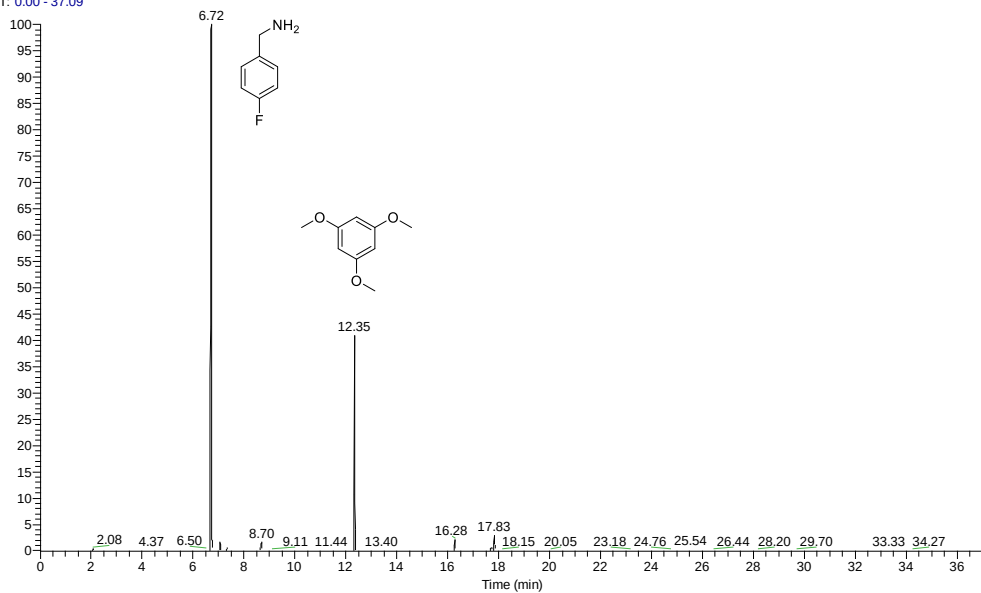
A234 #2359 RT: 10.02 AV: 1 NL: 8.00E6
T: + c EI Full ms [33.00-500.00]



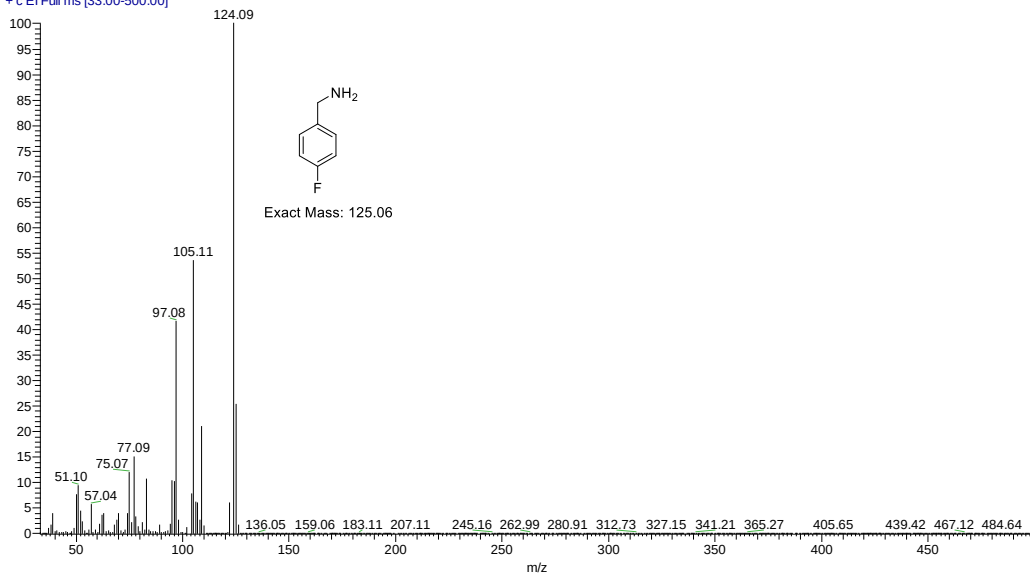
A235



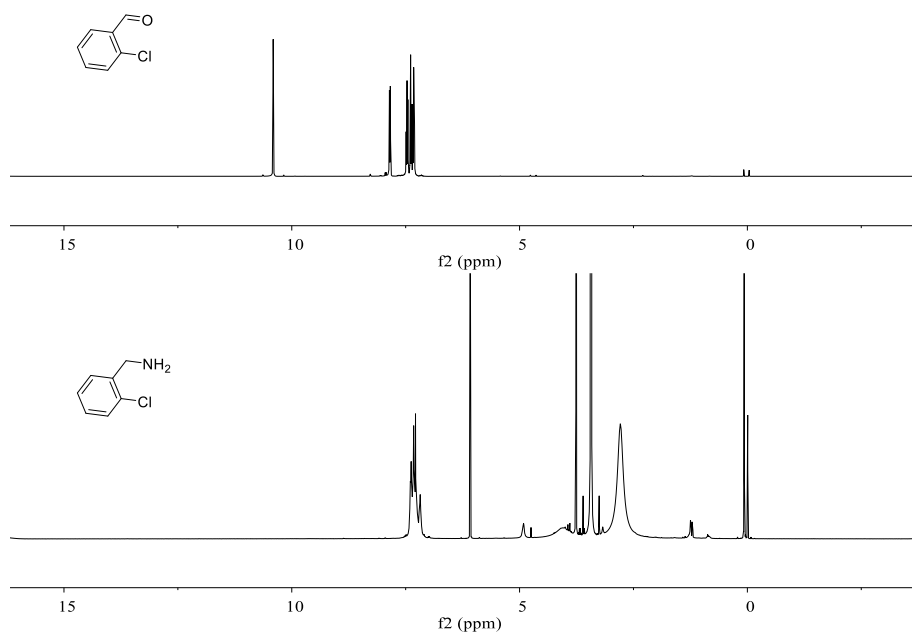
RT: 0.00 - 37.09



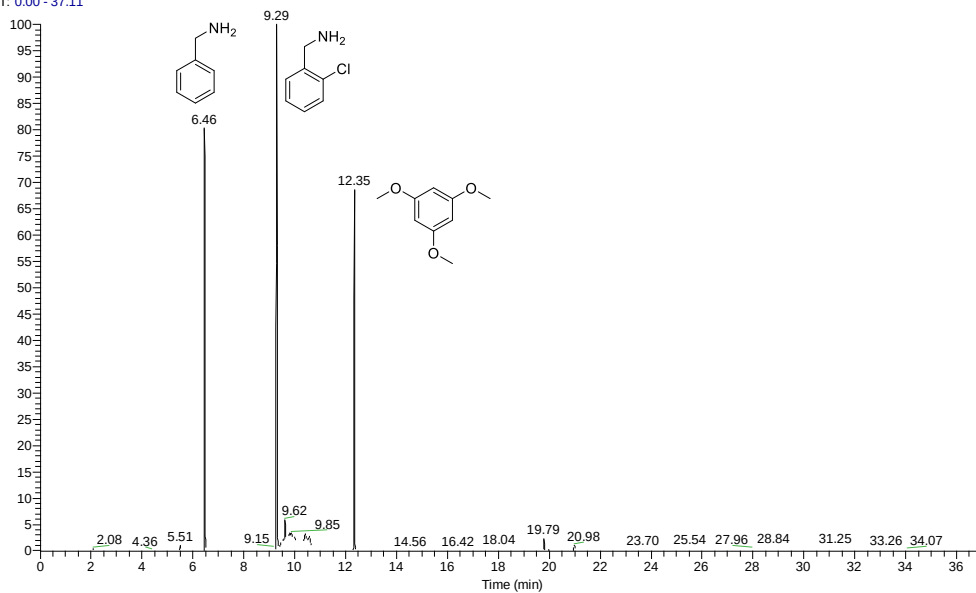
A235 #1389 RT: 6.72 AV: 1 NL: 2.10E8
T: + c EI Full ms [33.00-500.00]



A239

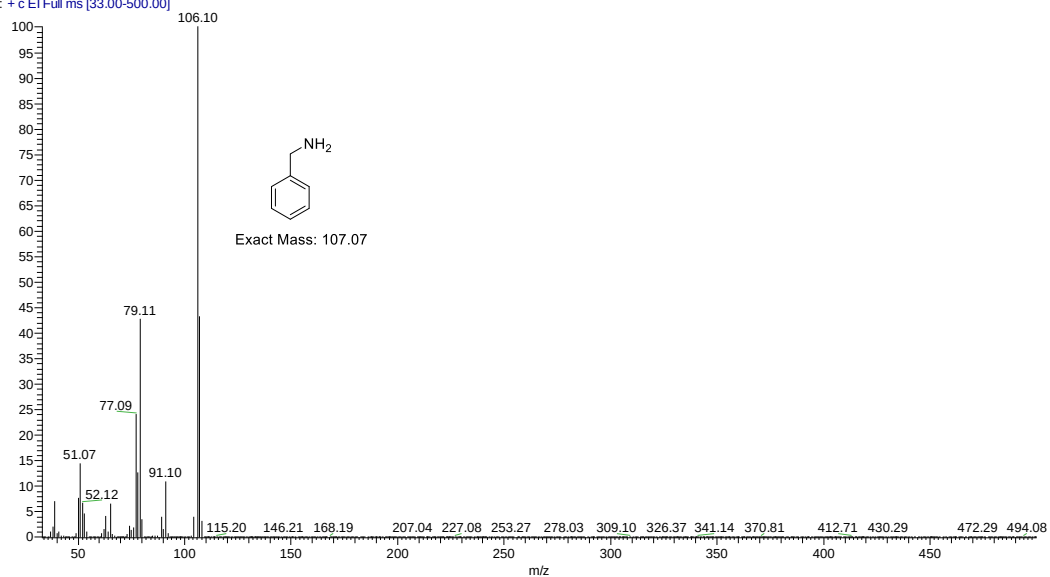


RT: 0.00 - 37.11

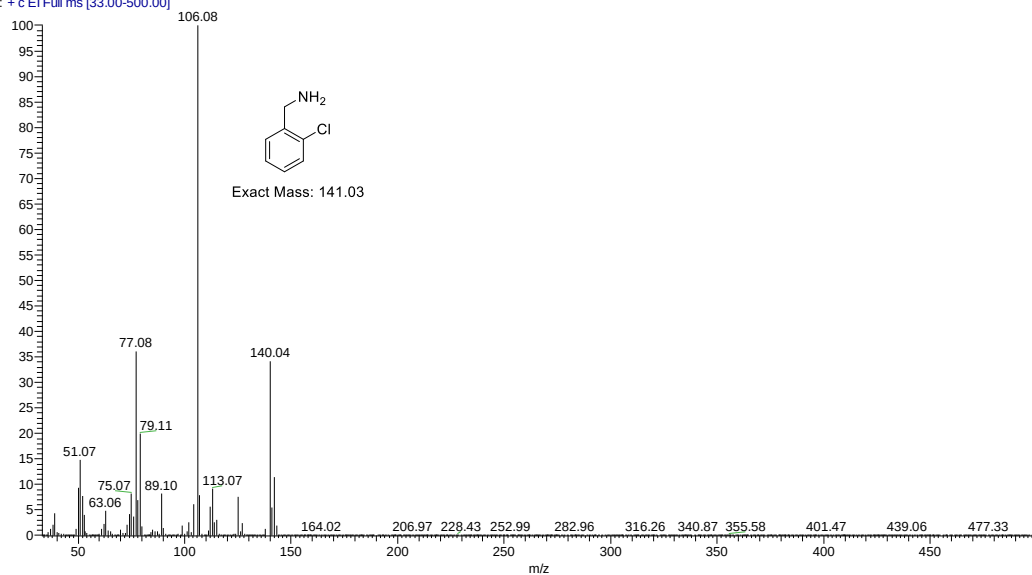


NL:
6.09E8
TIC: MS
A239

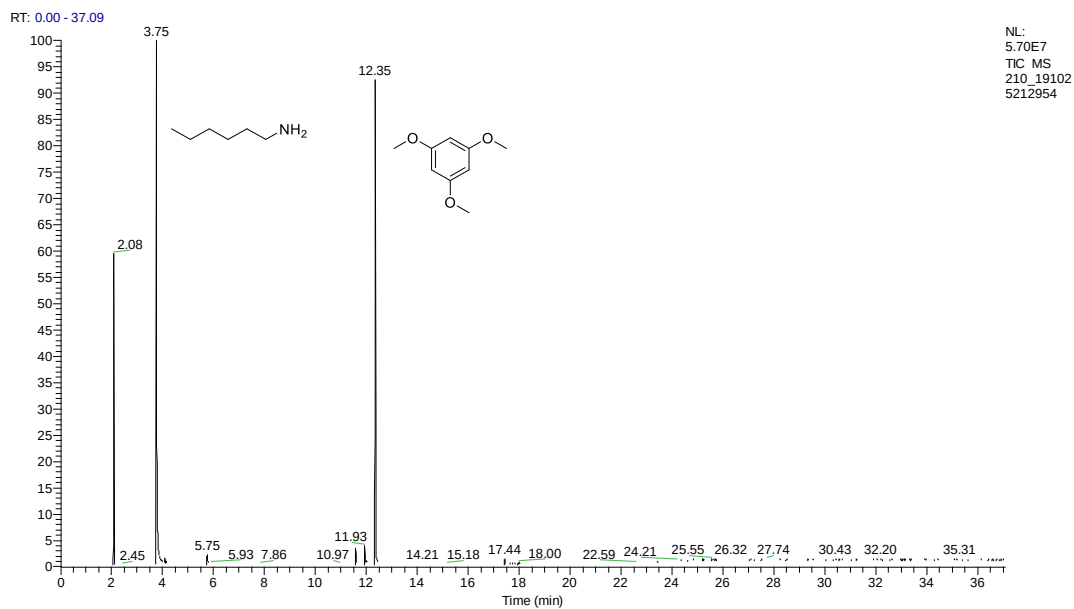
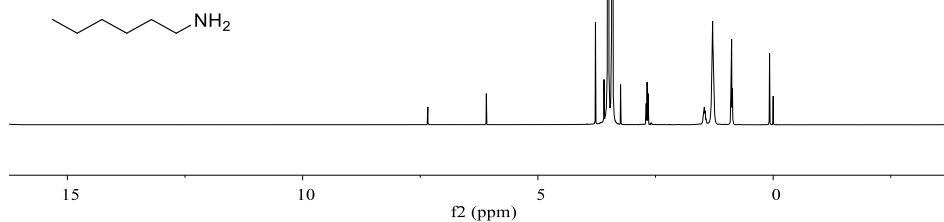
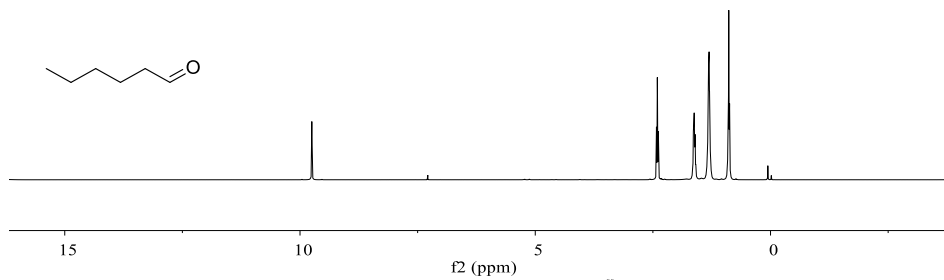
A239 #1313 RT: 6.46 AV: 1 NL: 1.42E8
T: + c EI Full ms [33.00-500.00]



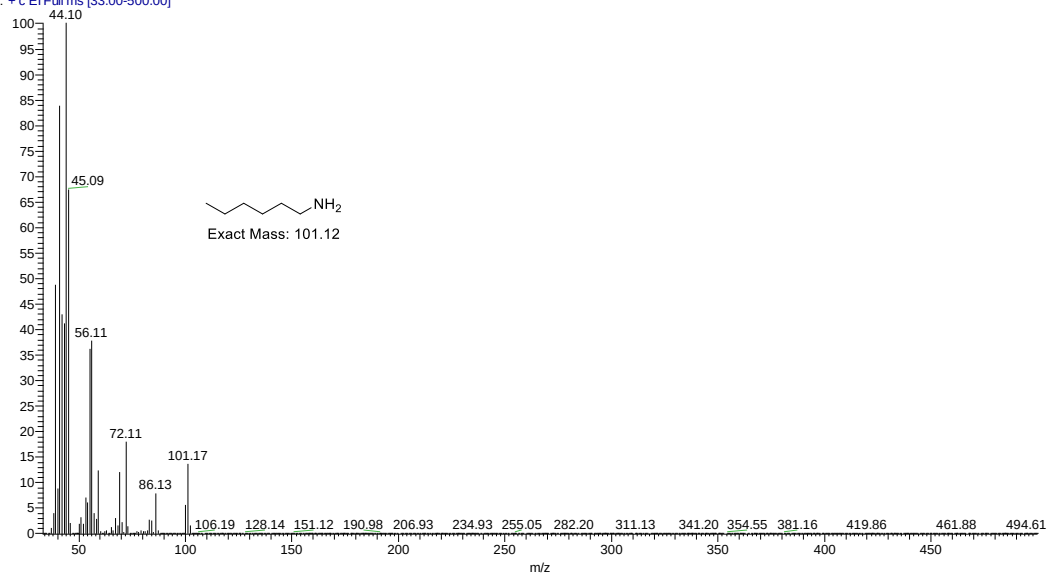
A239 #2143 RT: 9.29 AV: 1 NL: 1.60E8
T: + c EI Full ms [33.00-500.00]



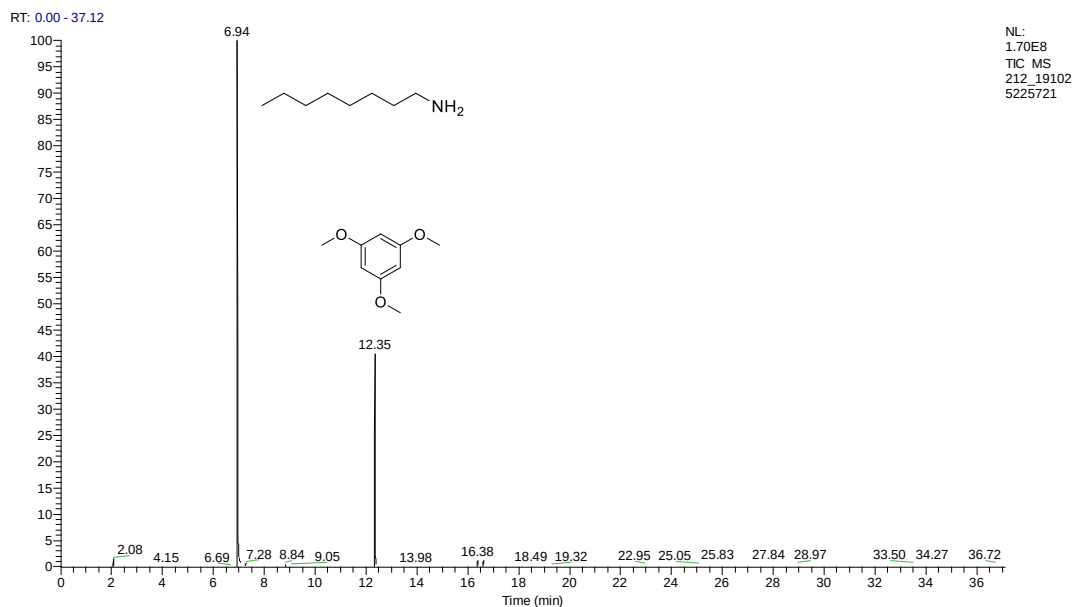
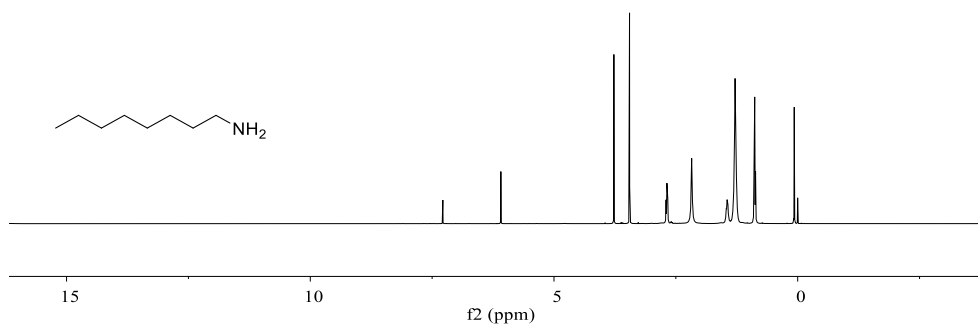
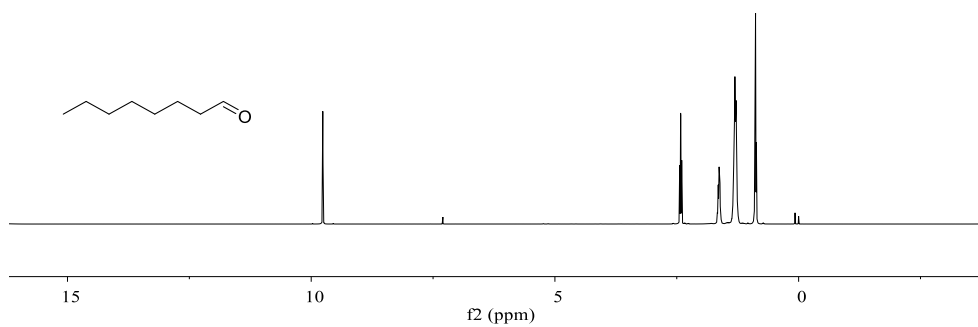
A210



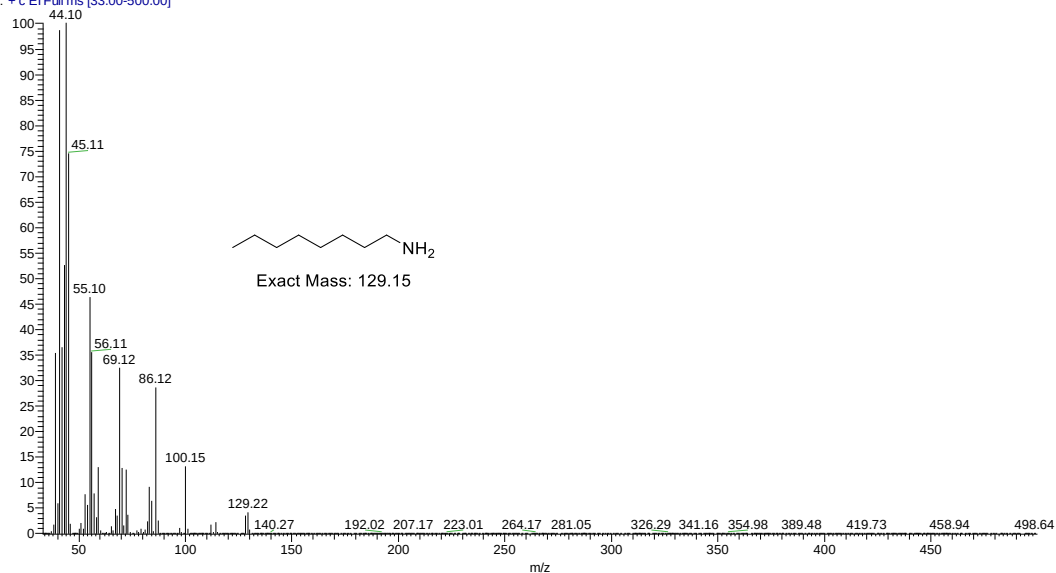
210 191025212954 #515 RT: 3.75 AV: 1 NL: 9.00E6
T: + c EI Full ms [33.00-500.00]



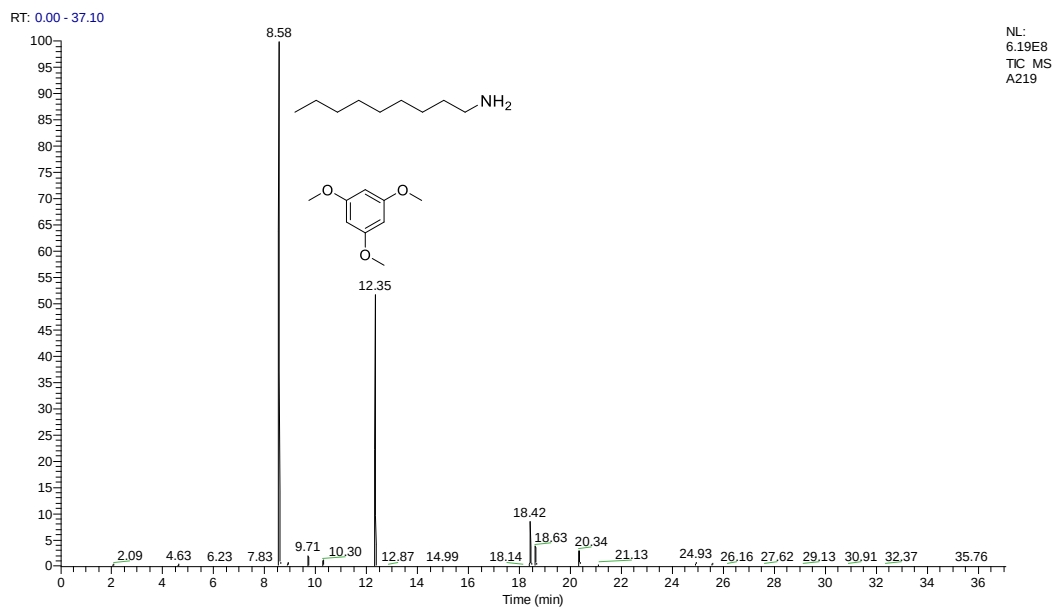
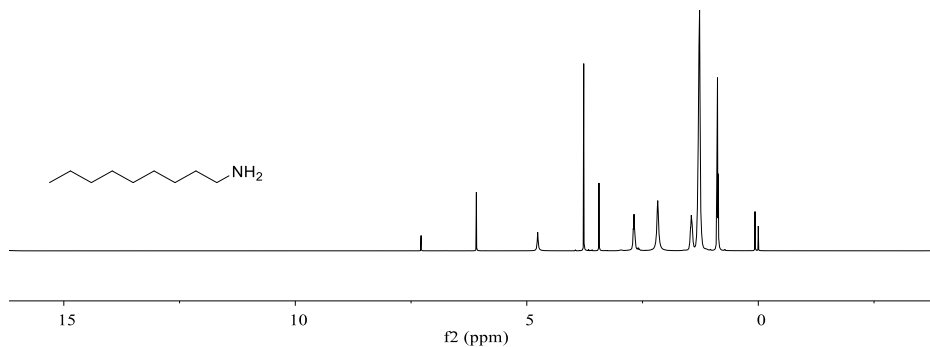
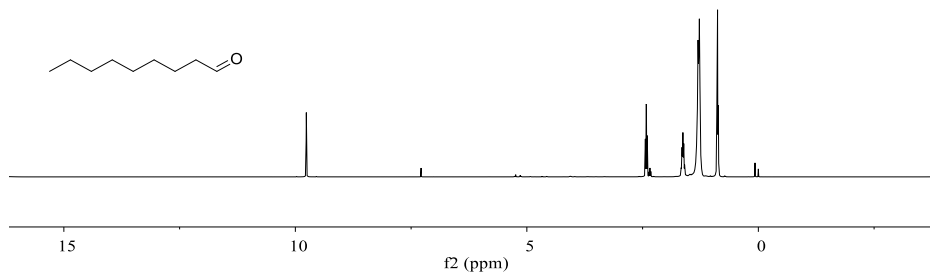
A212



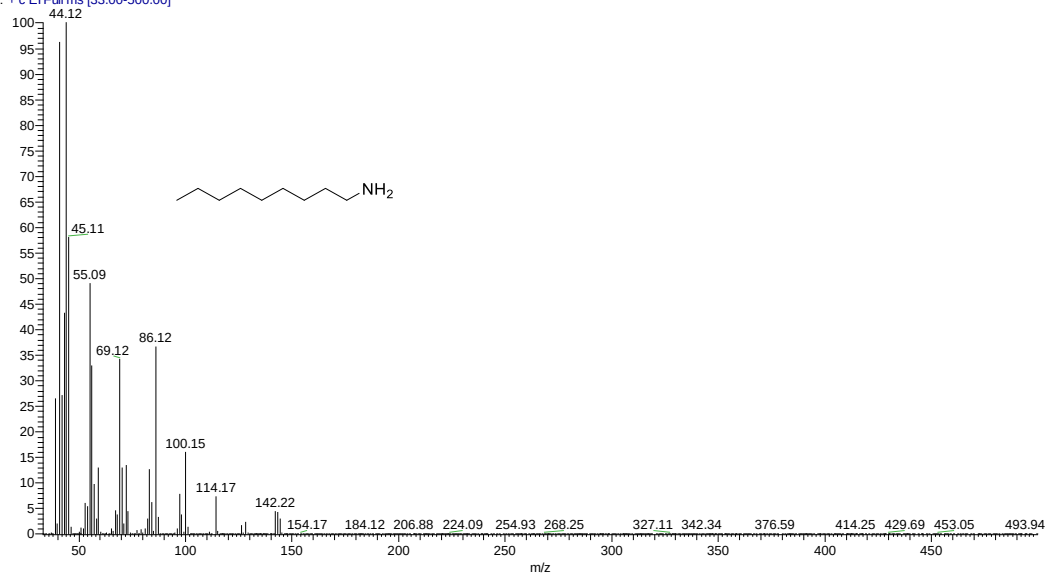
212_191025225721 #1453 RT: 6.94 AV: 1 NL: 2.28E7
T: + c EI Full ms [33.00-500.00]



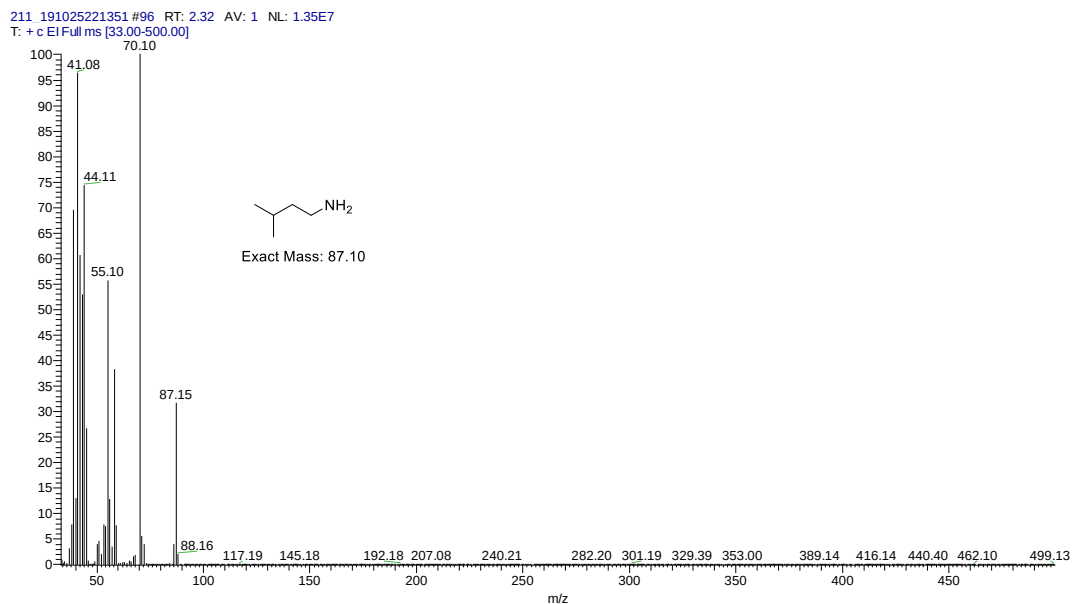
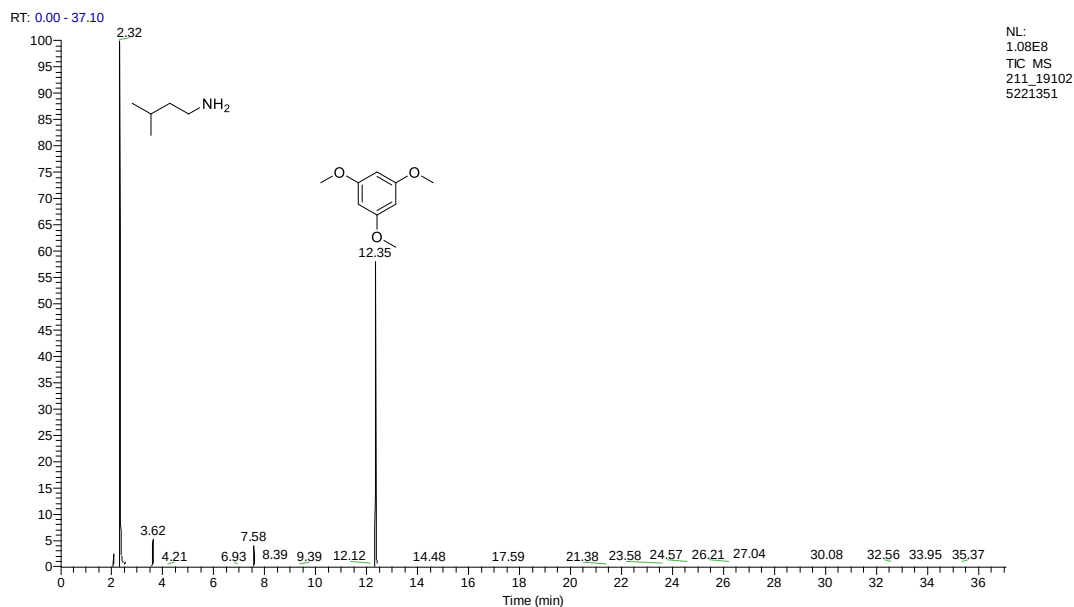
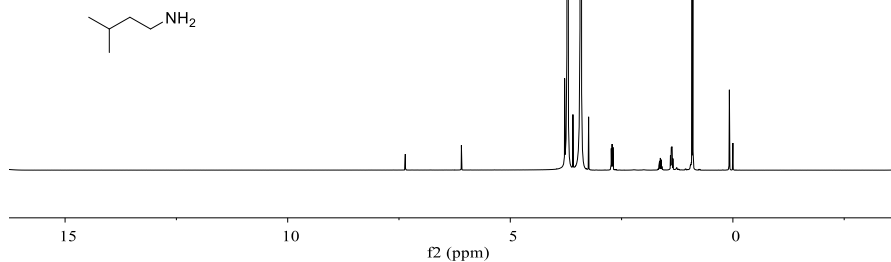
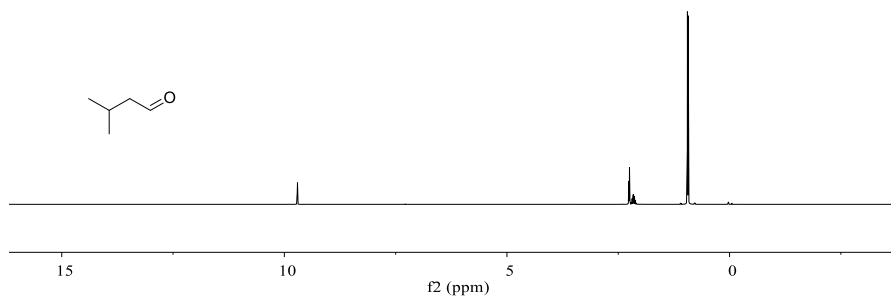
A219



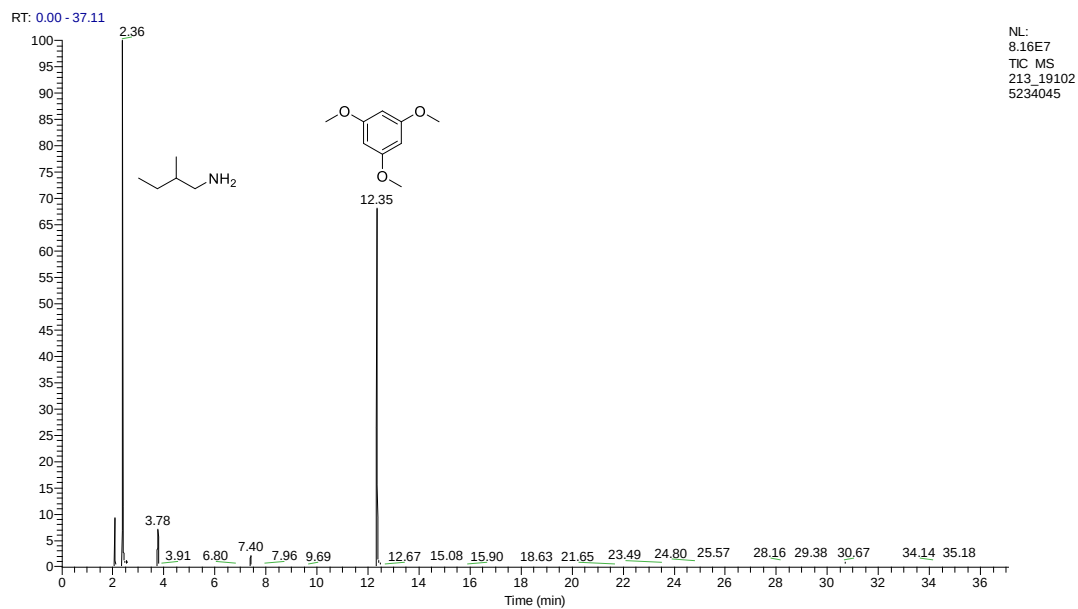
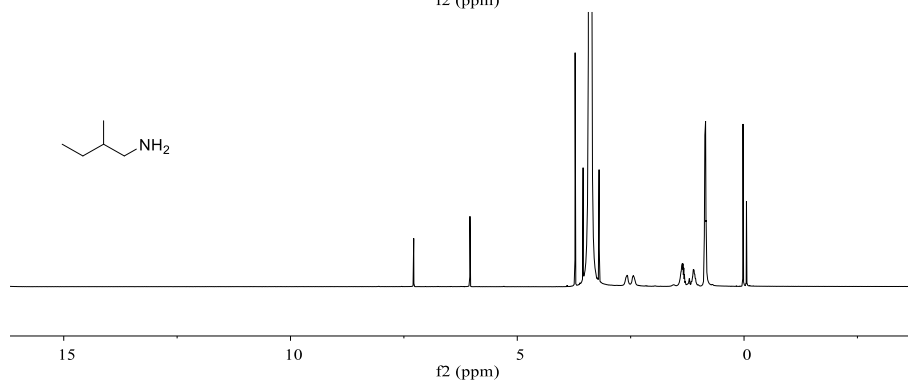
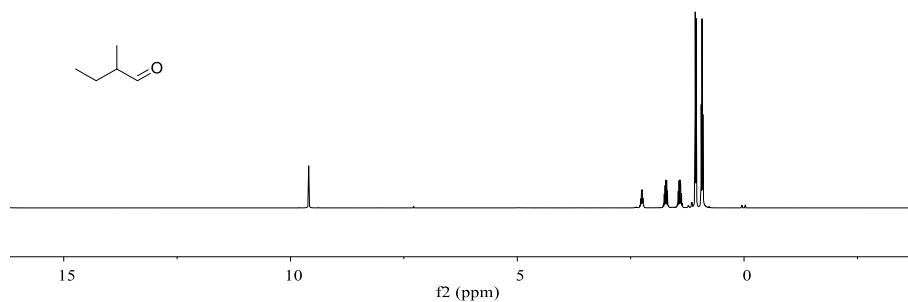
A219 #1935 RT: 8.58 AV: 1 NL: 8.57E7
T: + c EI Full ms [33.00-500.00]



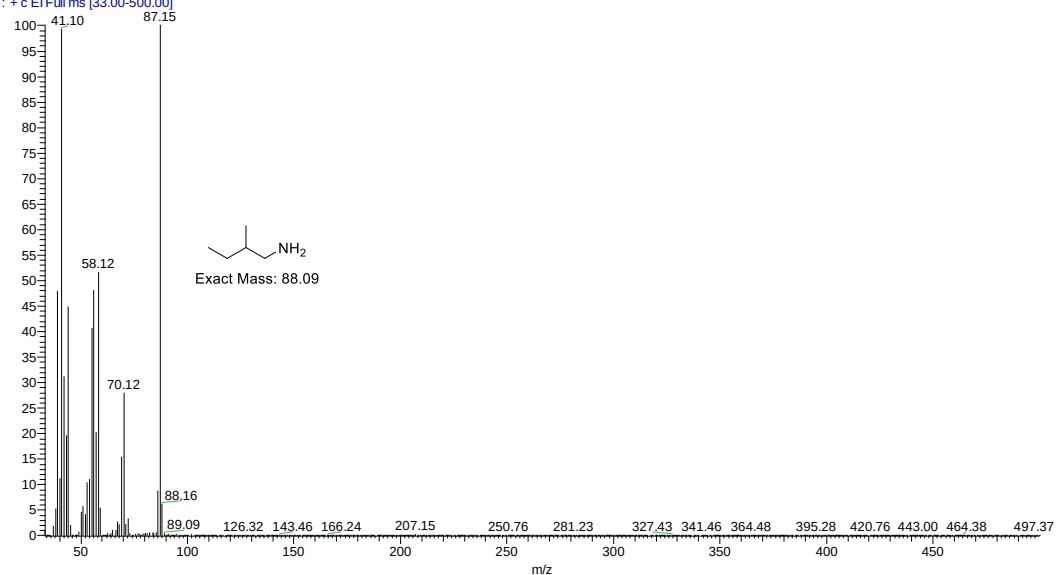
A211



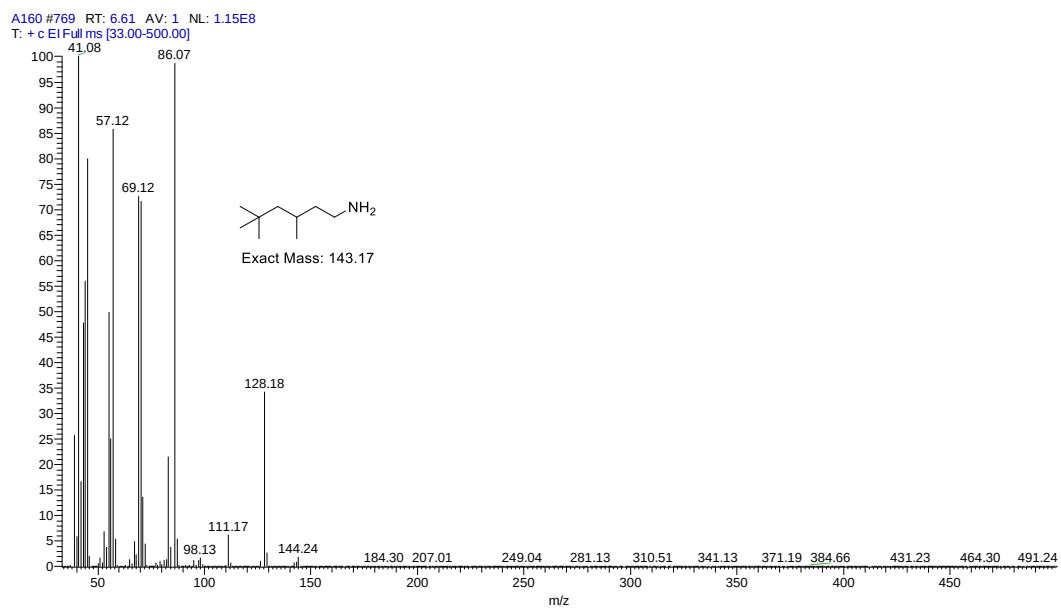
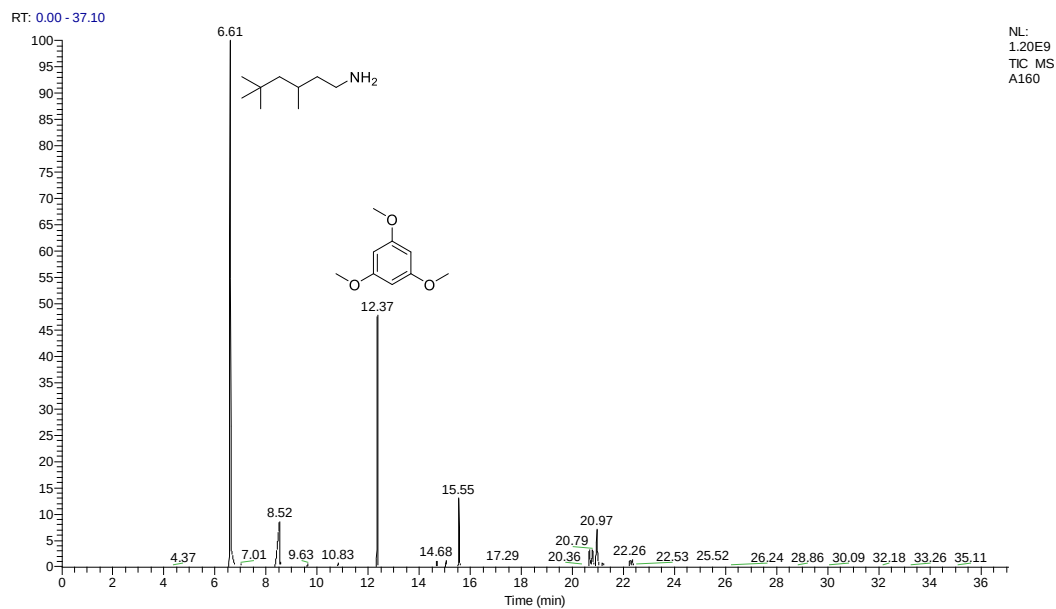
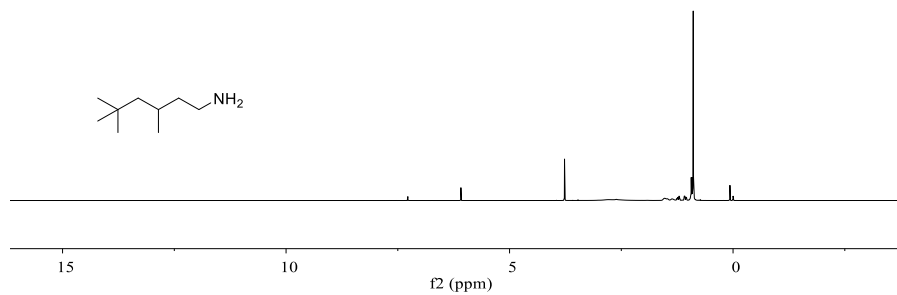
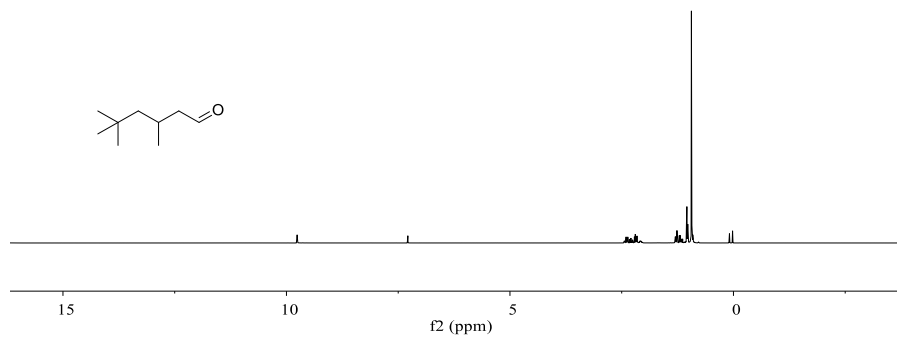
A213



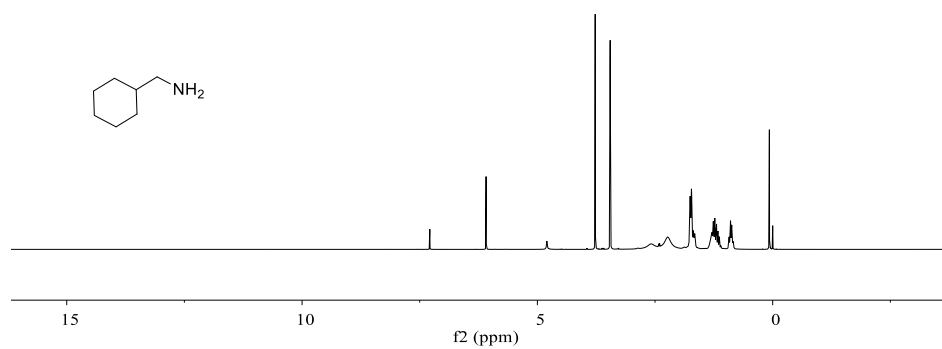
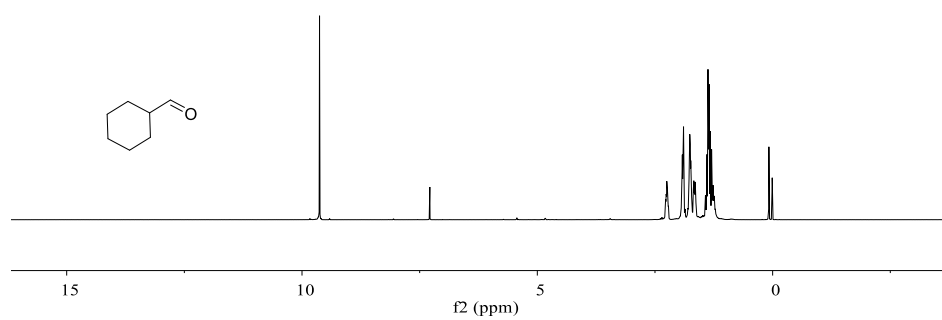
213_191025234045 #106 RT: 2.36 AV: 1 NL: 8.79E5
T: + c EI Full ms [33.00-500.00]



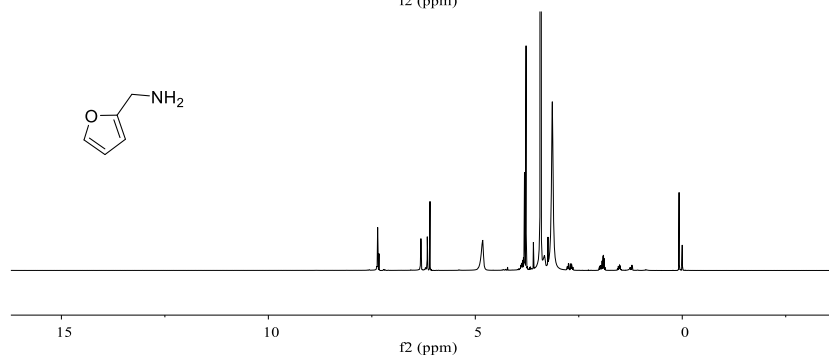
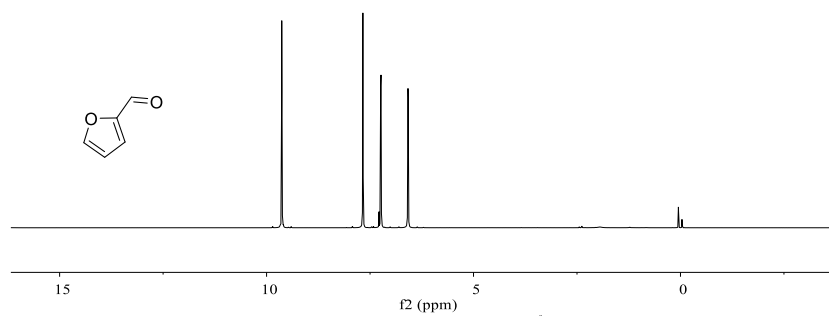
A160



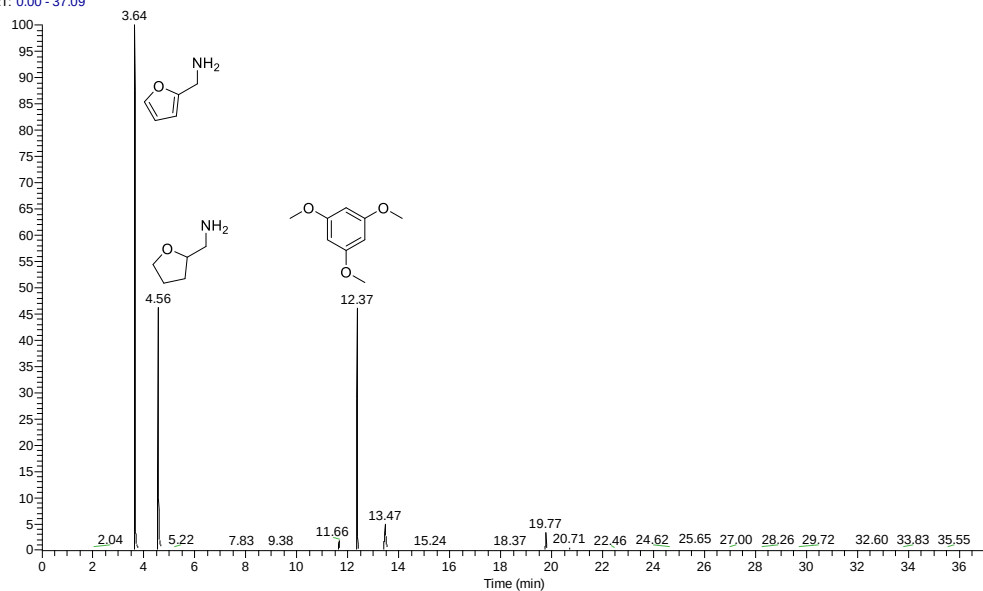
A186



A289

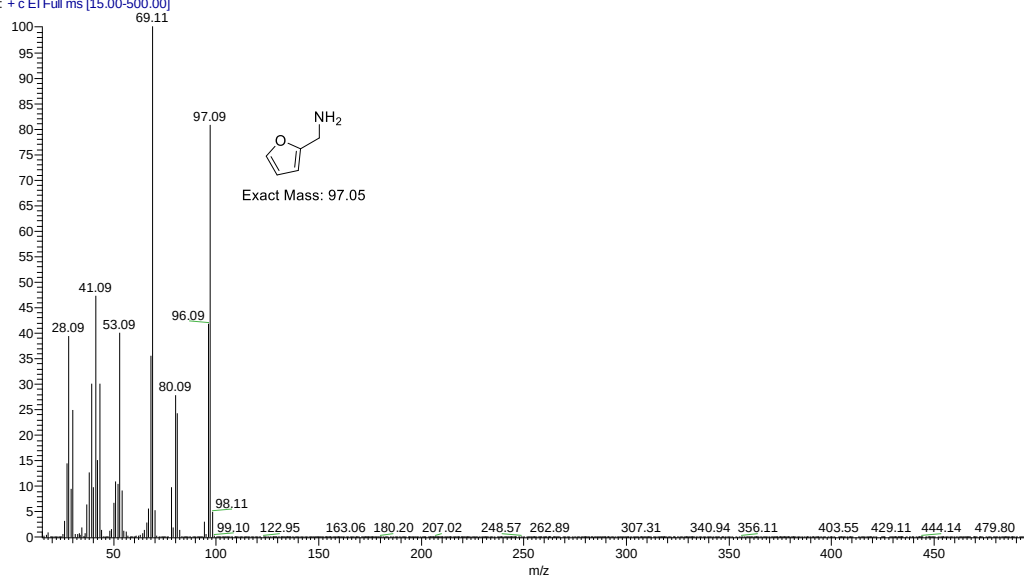


RT: 0.00 - 37.09

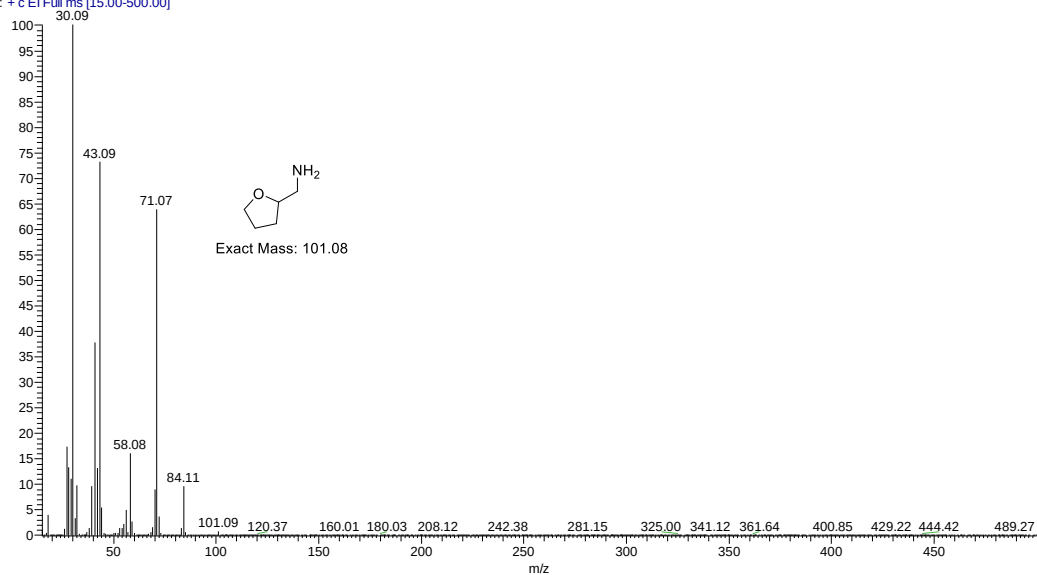


NL:
7.95E8
TIC: MS
A289

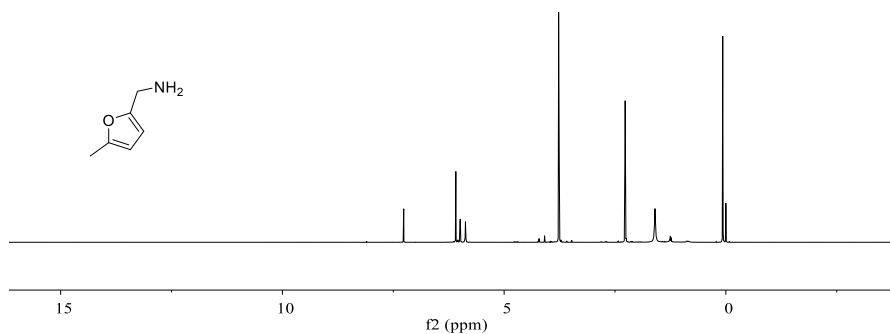
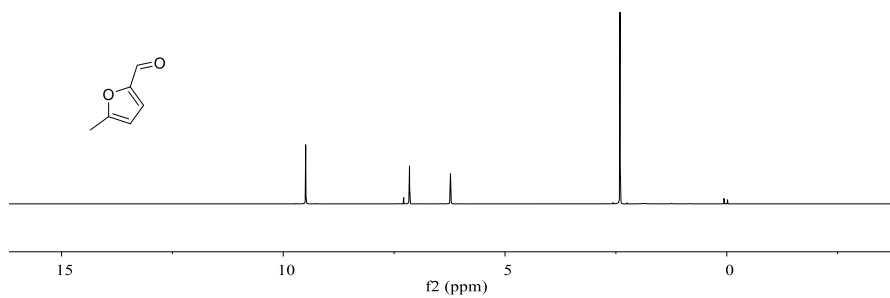
A289 #483 RT: 3.64 AV: 1 NL: 1.08E8
T: + c EI Full ms [15.00-500.00]



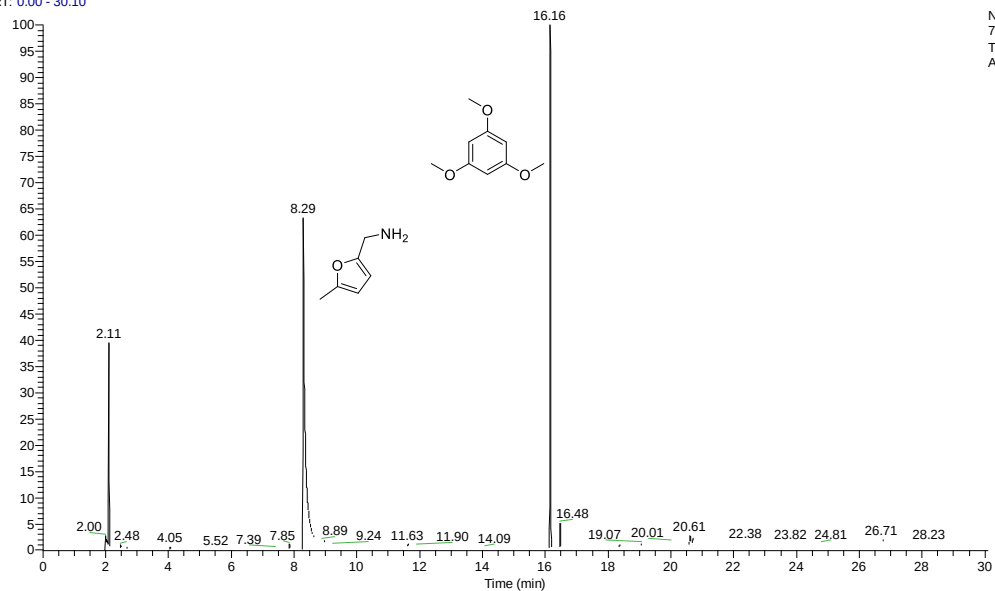
A289 #753 RT: 4.56 AV: 1 NL: 8.61E7
T: + c EI Full ms [15.00-500.00]



A145

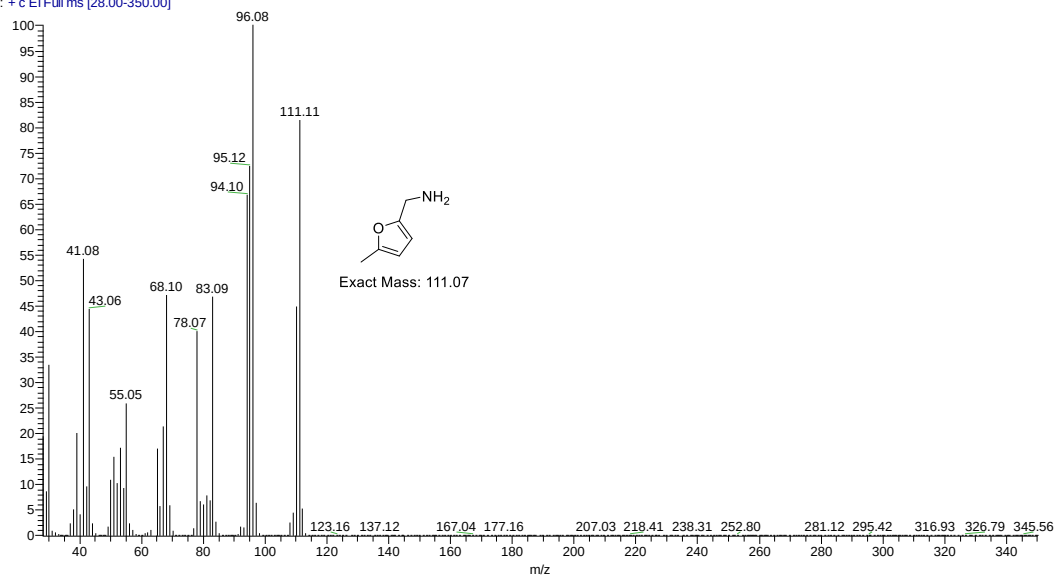


RT: 0.00 - 30.10

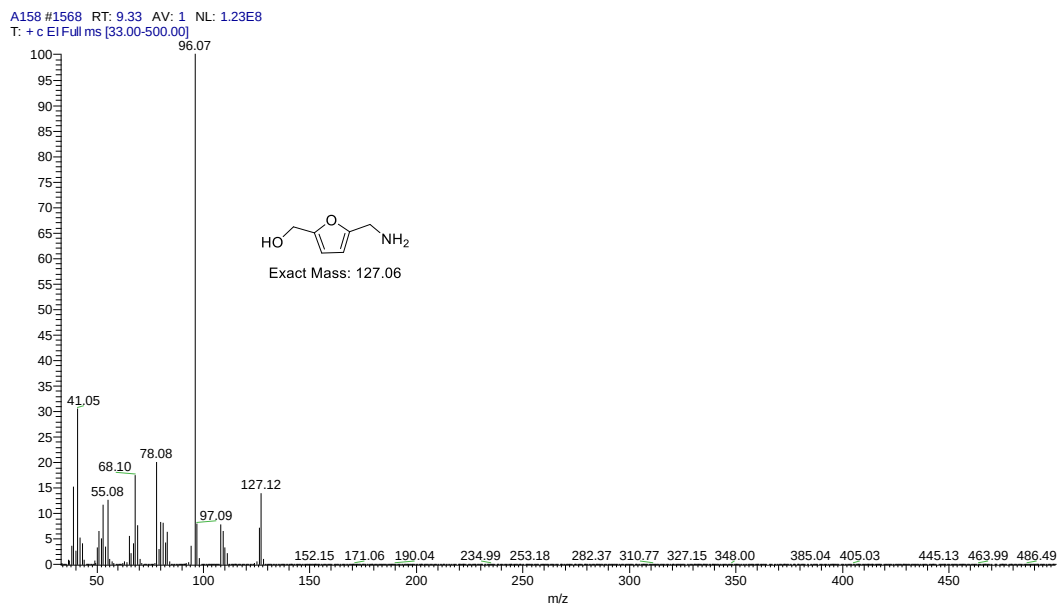
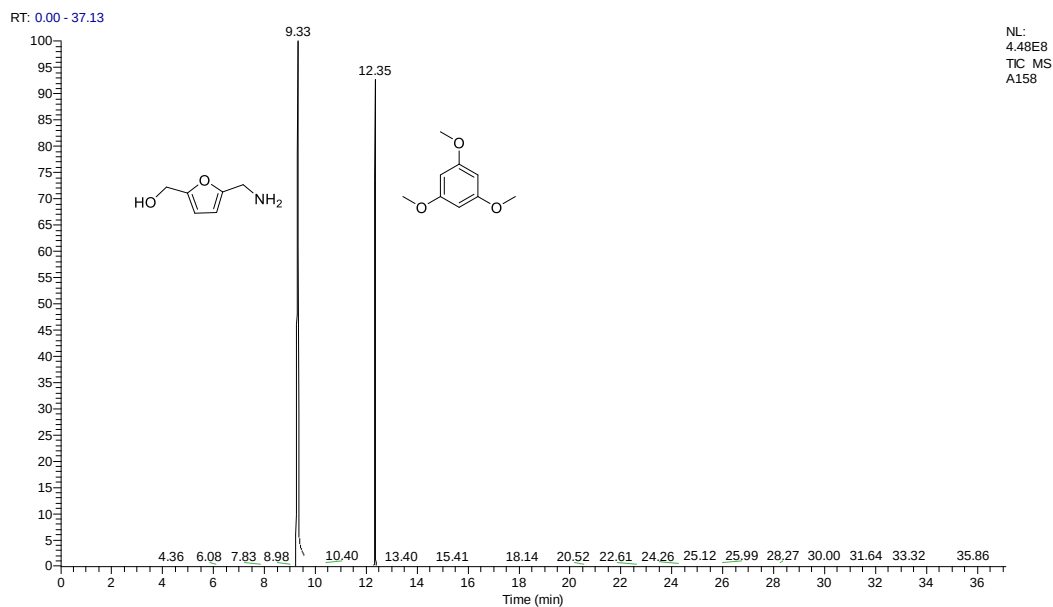
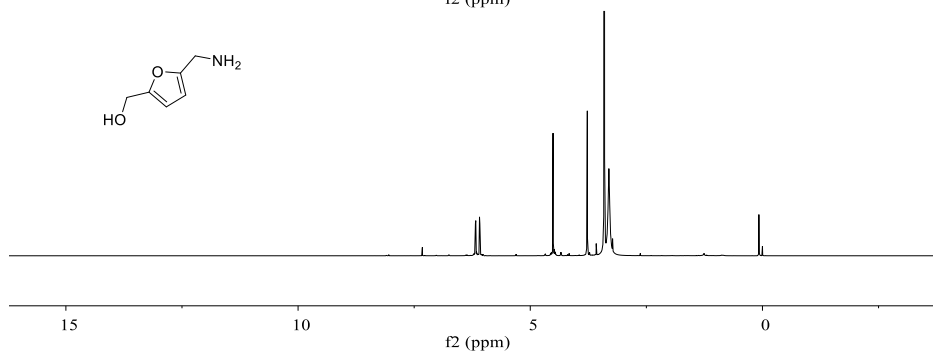
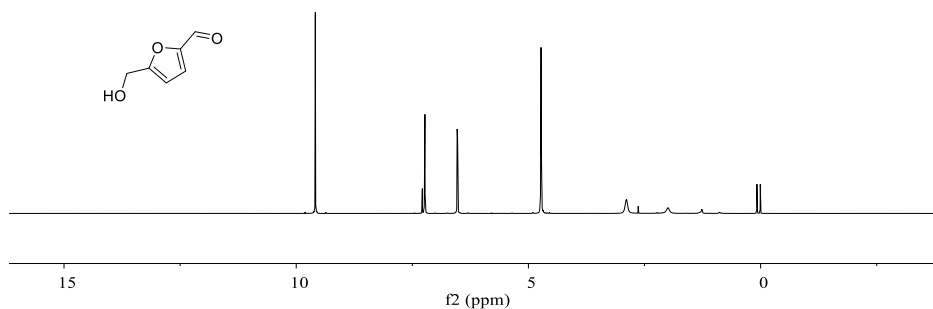


NL:
7.18E8
TIC MS
A145

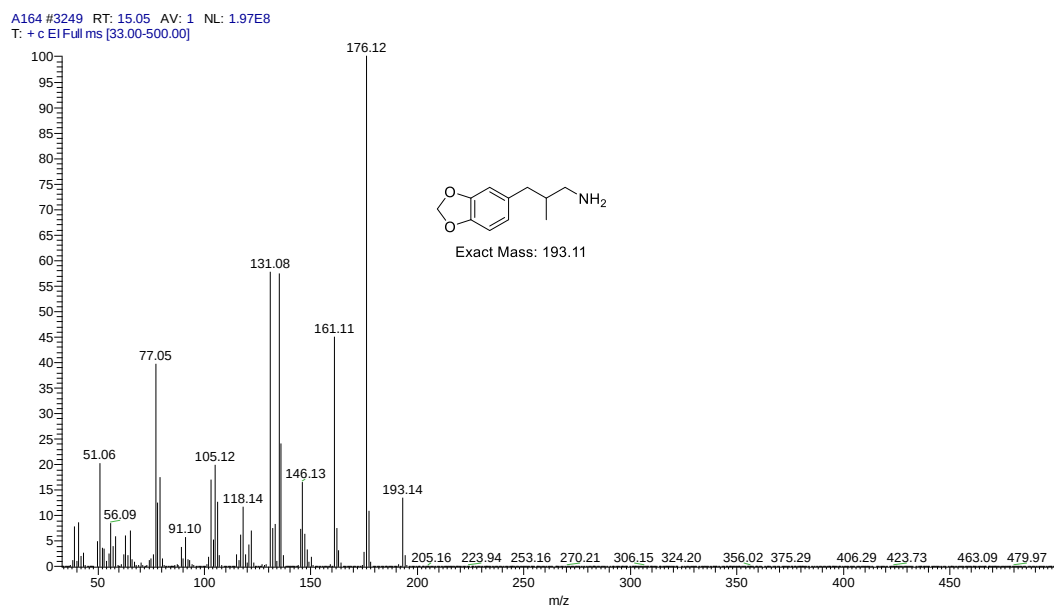
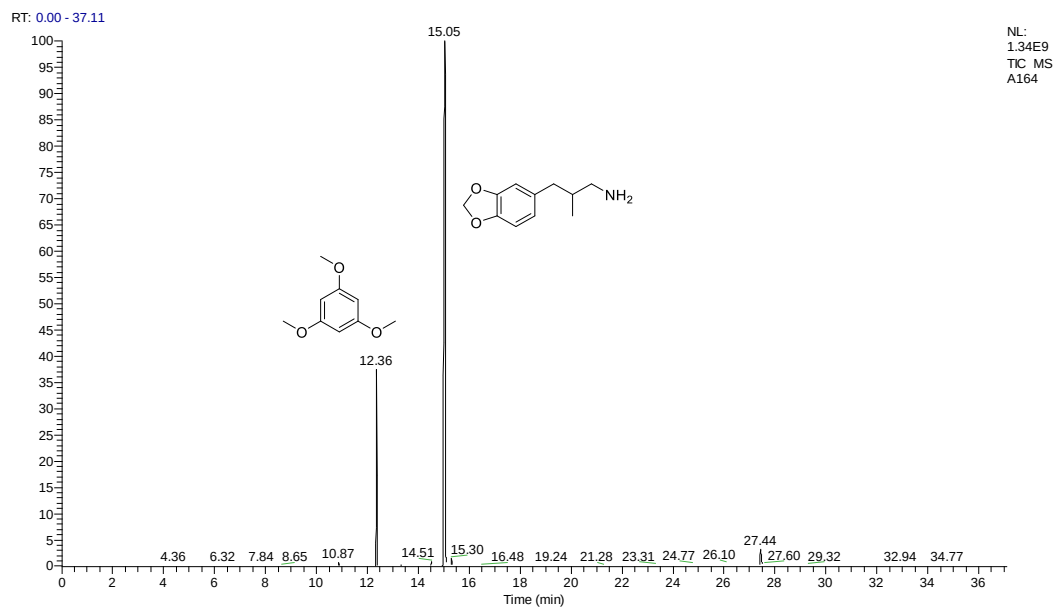
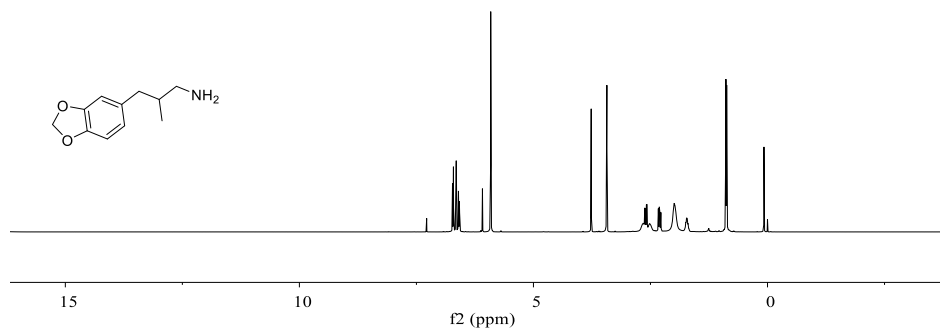
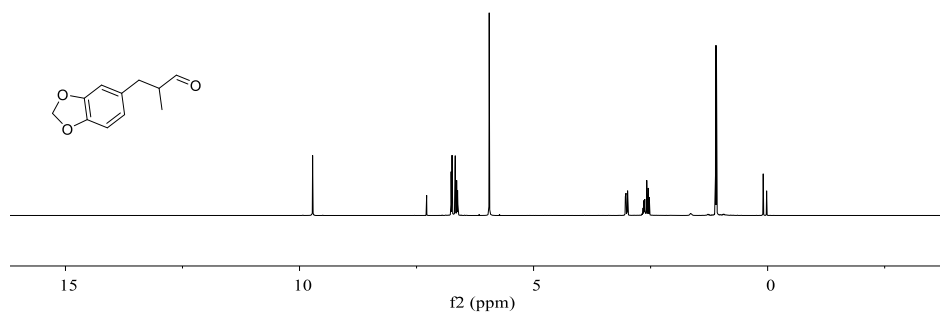
A145 #2429 RT: 8.29 AV: 1 NL: 5.00E7
T: + c EI Full ms [28.00-350.00]



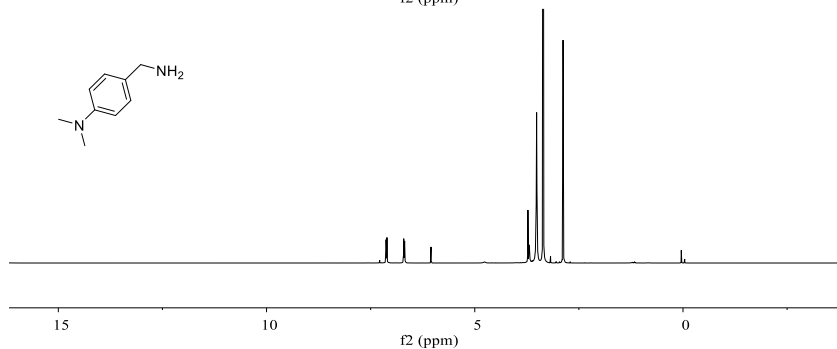
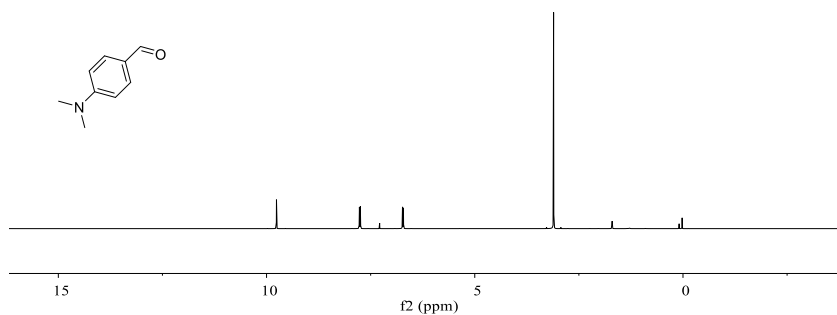
A158



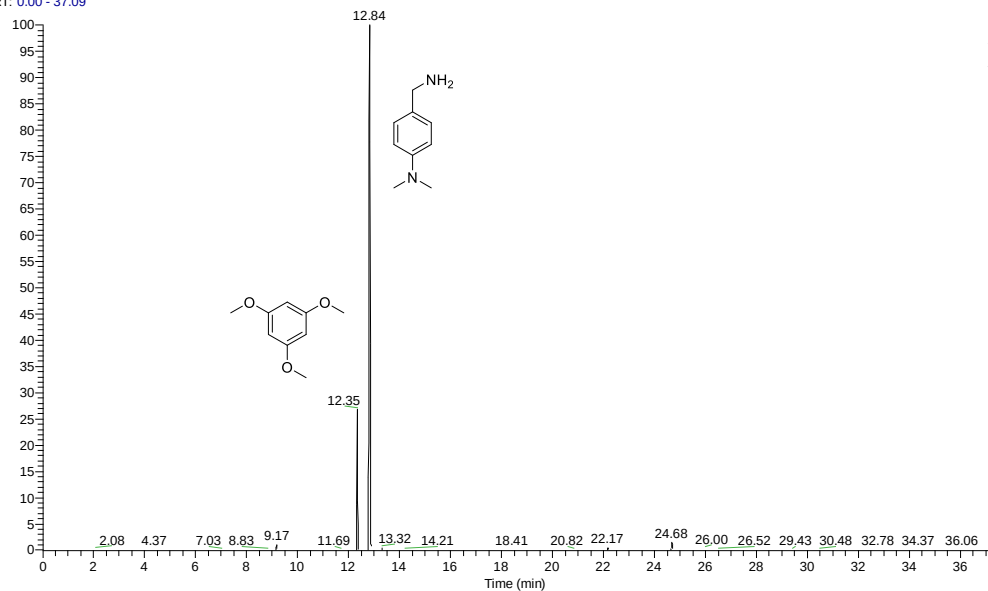
A164



A236

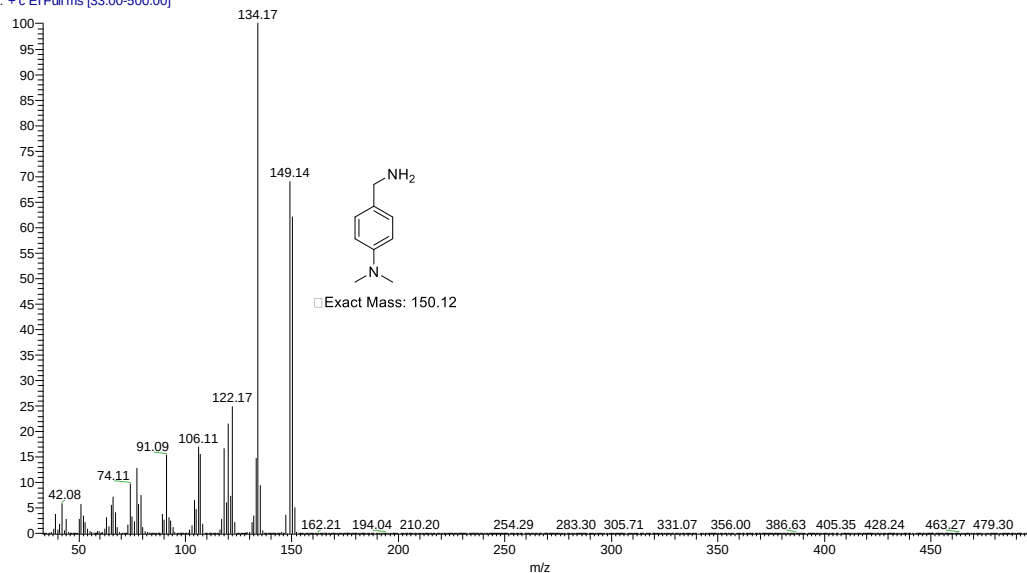


RT: 0.00 - 37.09

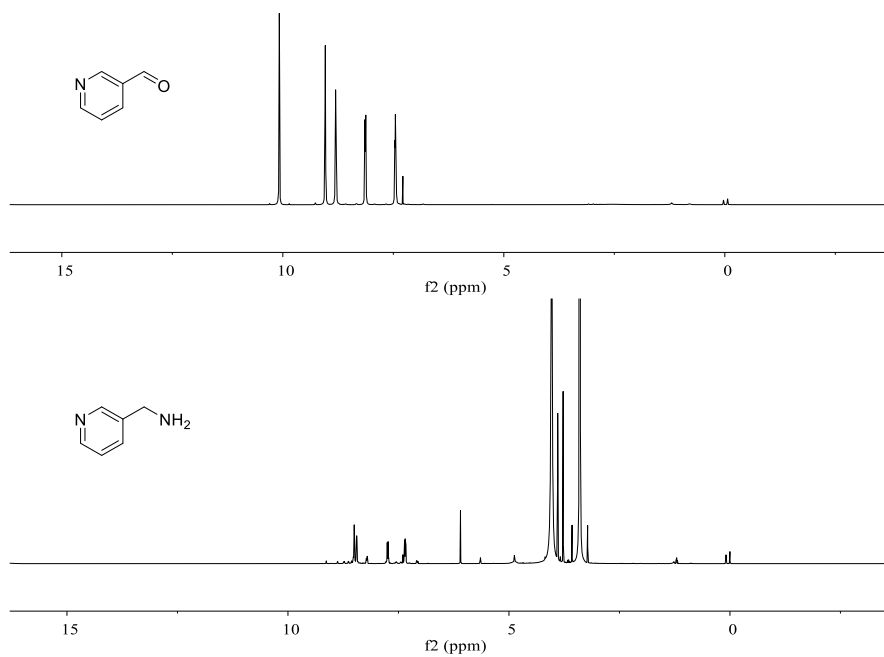


NL:
1.32E9
TIC: MS
A236

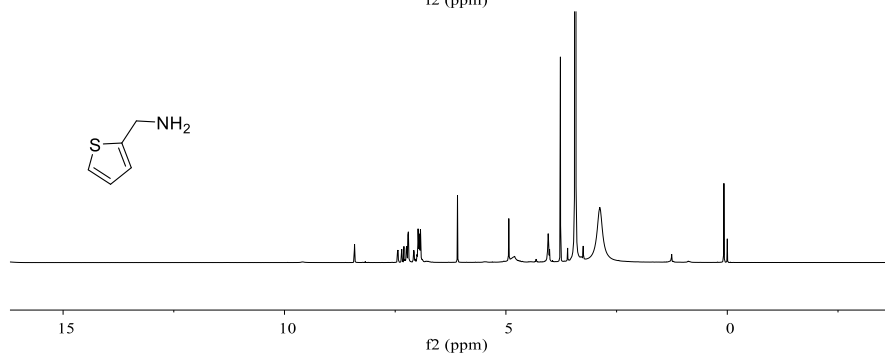
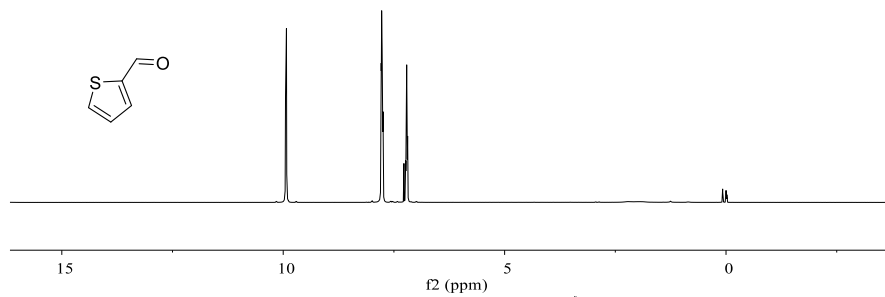
A236 #3189 RT: 12.84 AV: 1 NL: 1.91E8
T: + c EI Full ms [33.00-500.00]



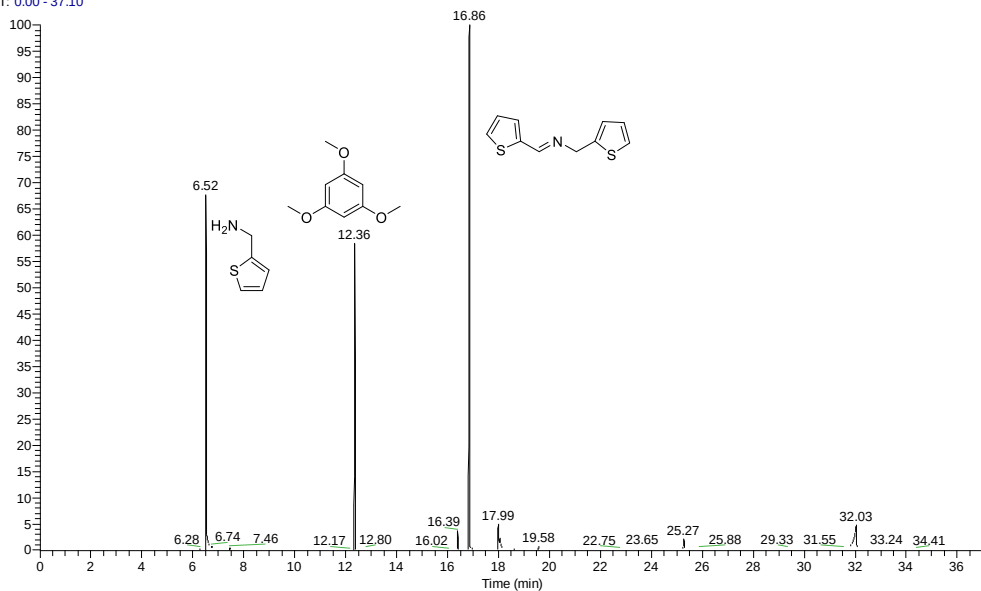
A279



A177

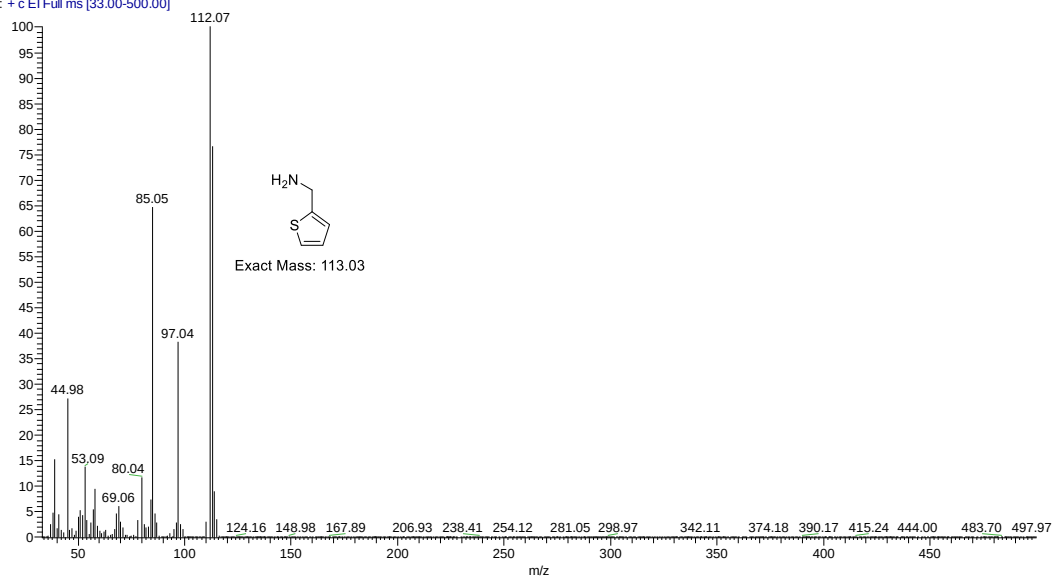


RT: 0.00 - 37.10

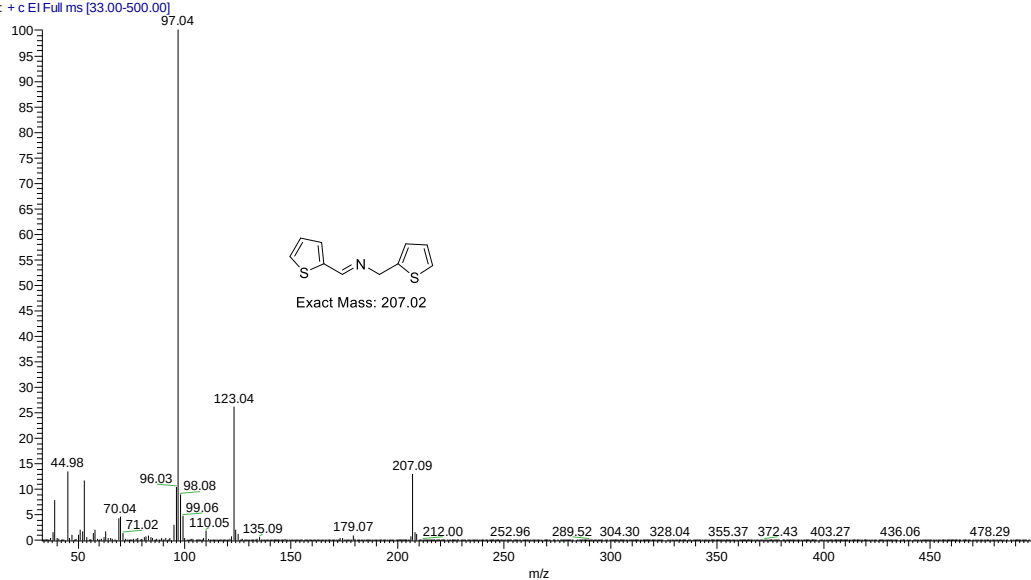


NL:
8.87E8
TIC: MS
A177

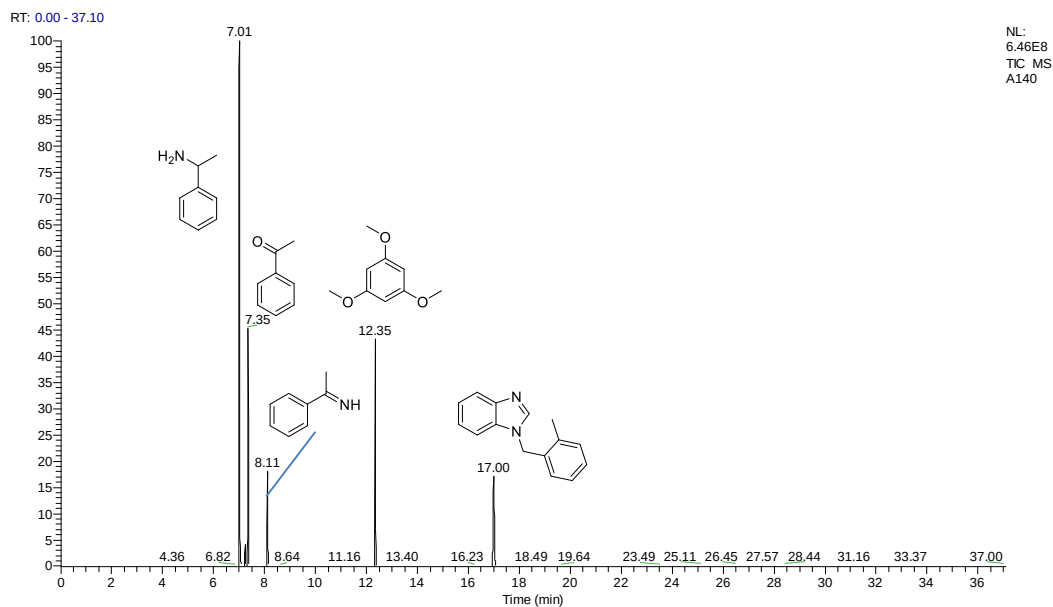
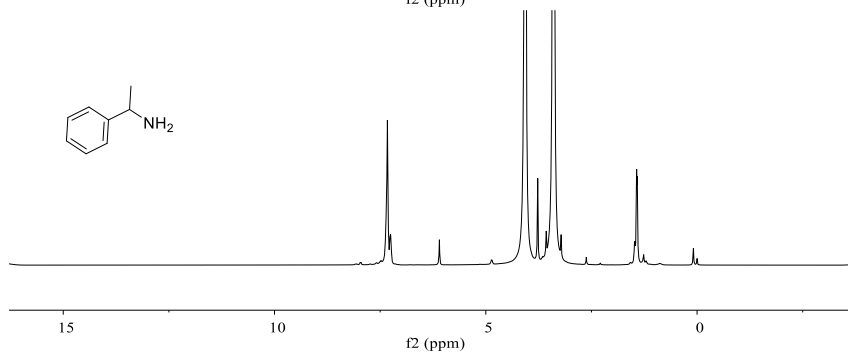
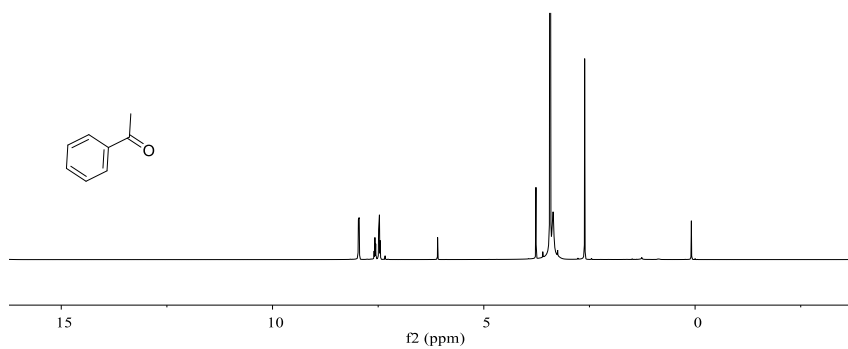
A177 #742 RT: 6.52 AV: 1 NL: 1.21E8
T: + c EI Full ms [33.00-500.00]



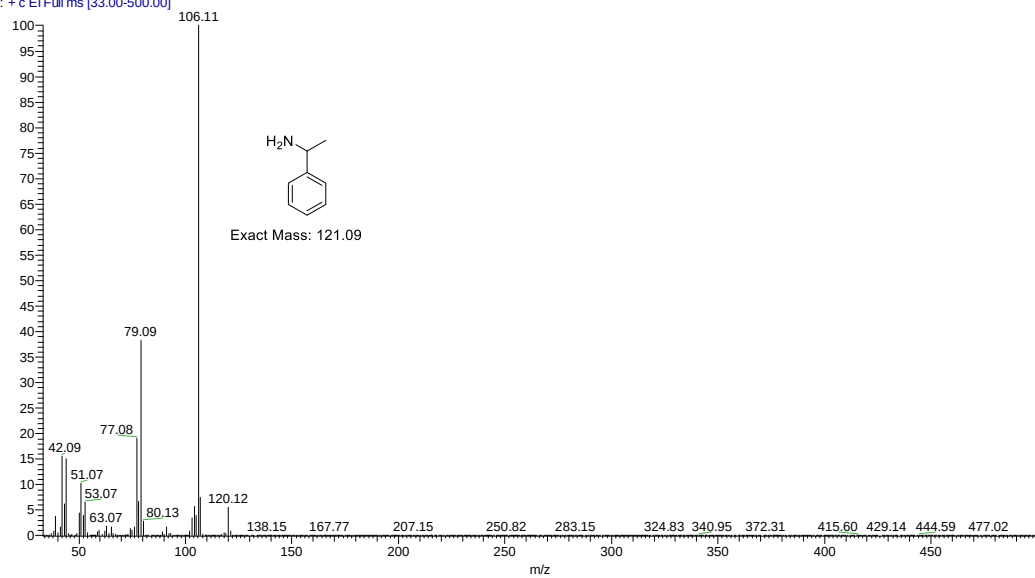
A177 #3782 RT: 16.86 AV: 1 NL: 3.21E8
T: + c EI Full ms [33.00-500.00]



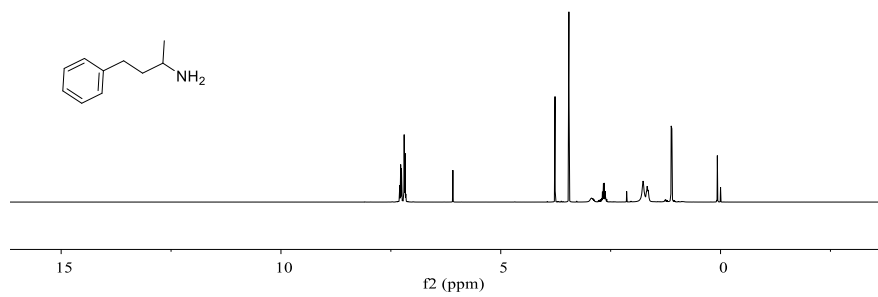
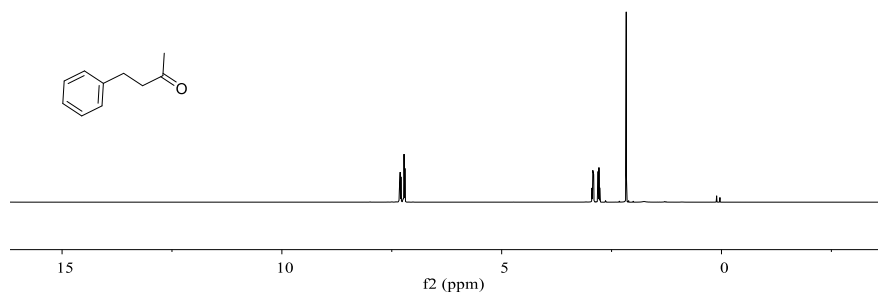
A140



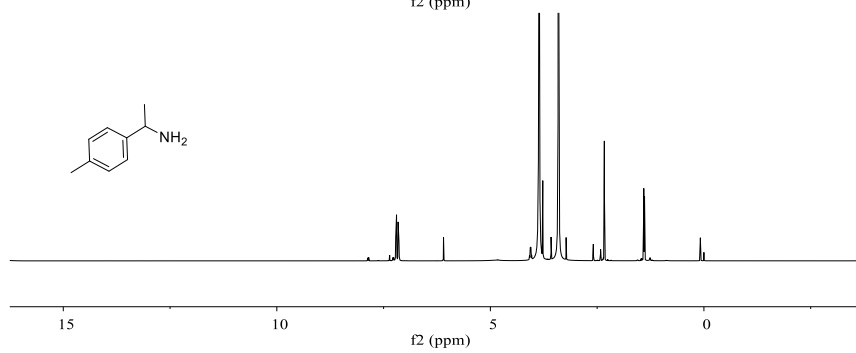
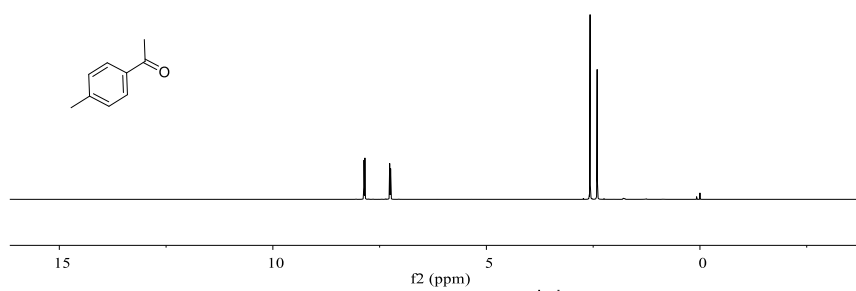
A140 #887 RT: 7.01 AV: 1 NL: 2.18E8
T: + c EI Full ms [33.00-500.00]



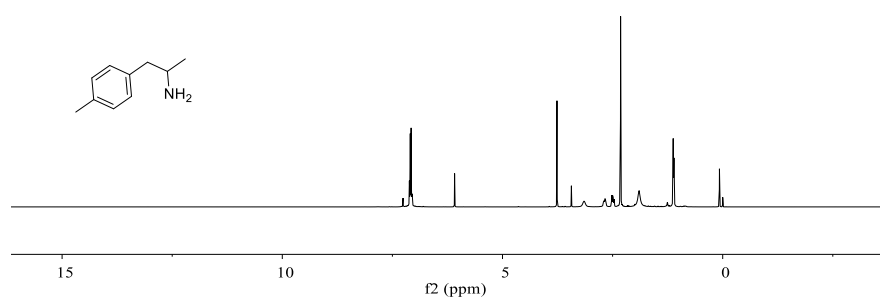
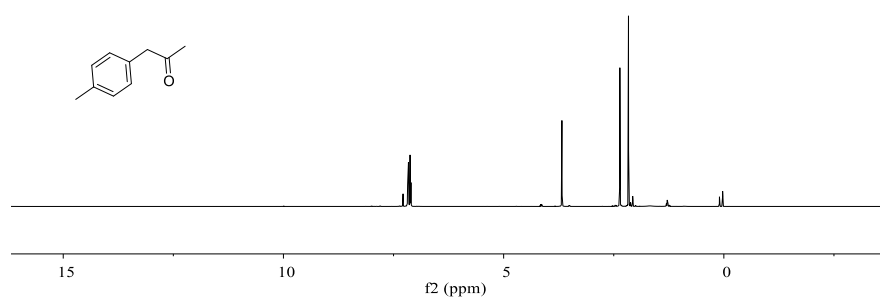
A144



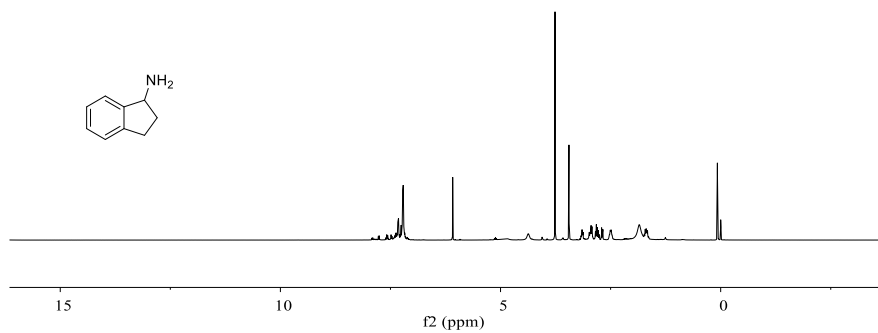
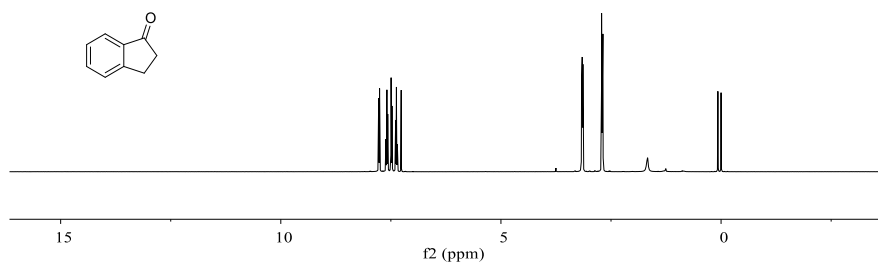
A222

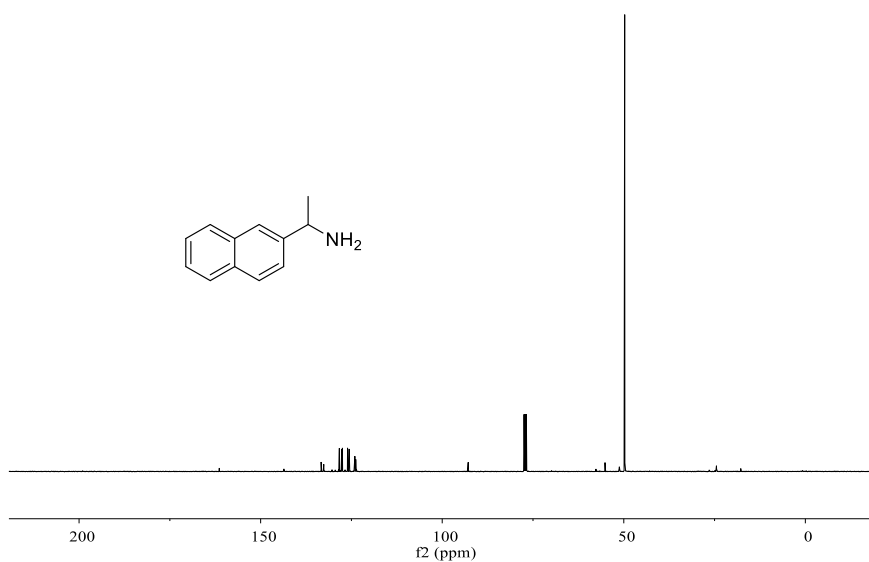
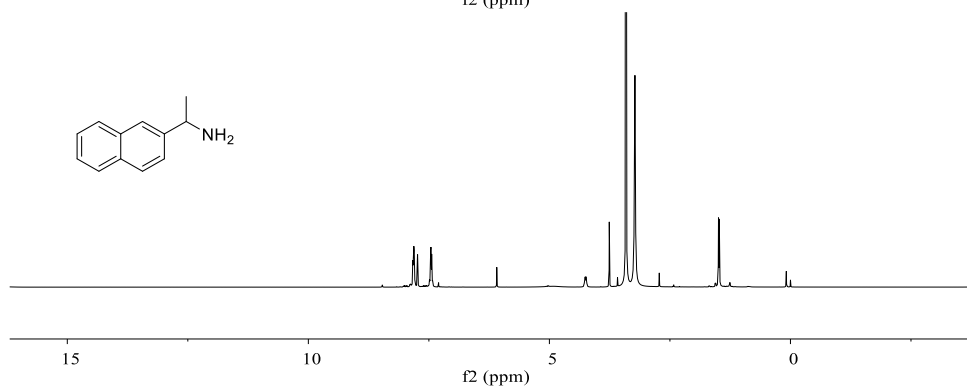
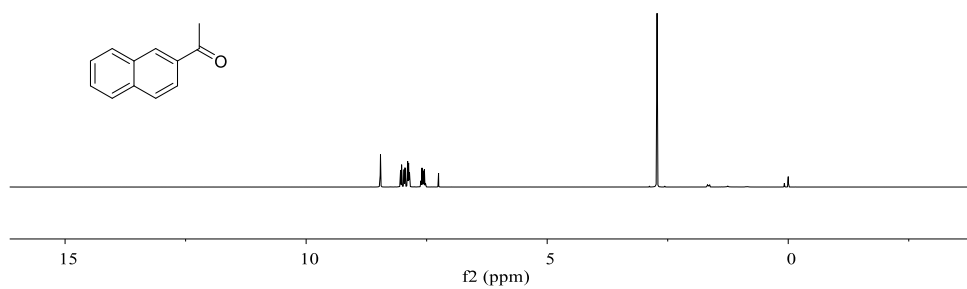


A187



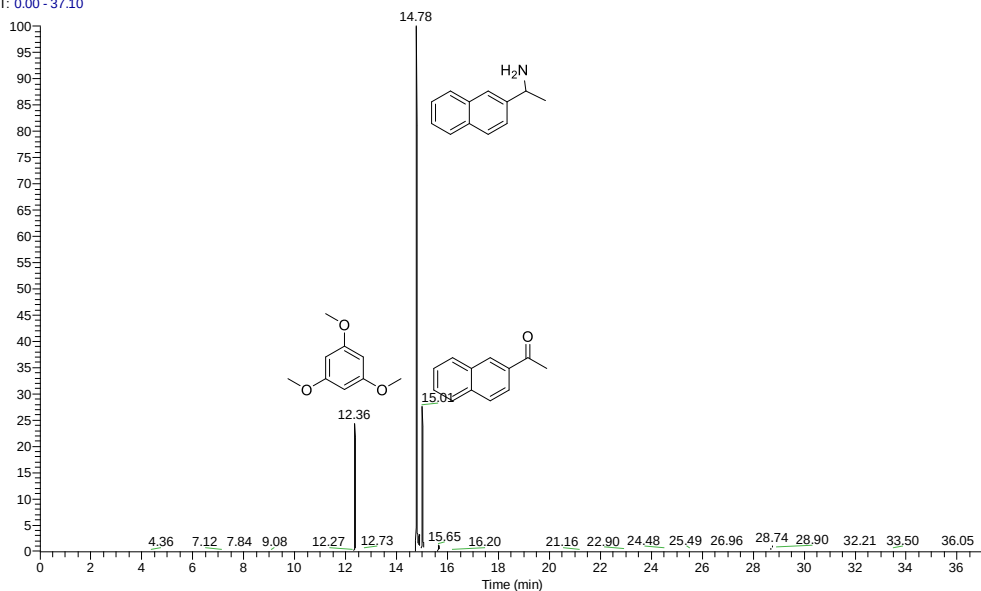
A191





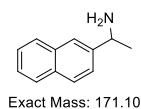
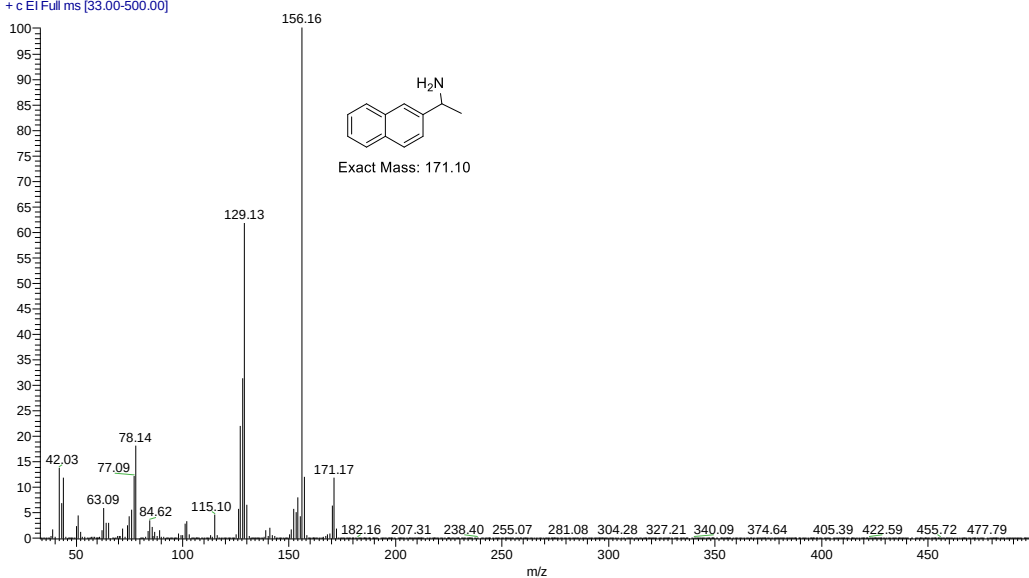
^1H NMR (400 MHz, CDCl_3) δ 7.82 (m, 3H), 7.73 (s, 1 H), 7.46 (m, 3H), 4.26 (m, 1 H), 1.48 (d, 3H). ^{13}C NMR (125MHz, CDCl_3) δ , 133.34, 132.62, 128.33, 127.66, 127.49, 126.06, 125.62, 124.05, 123.81, 51.23, 24.48.

RT: 0.00 - 37.10

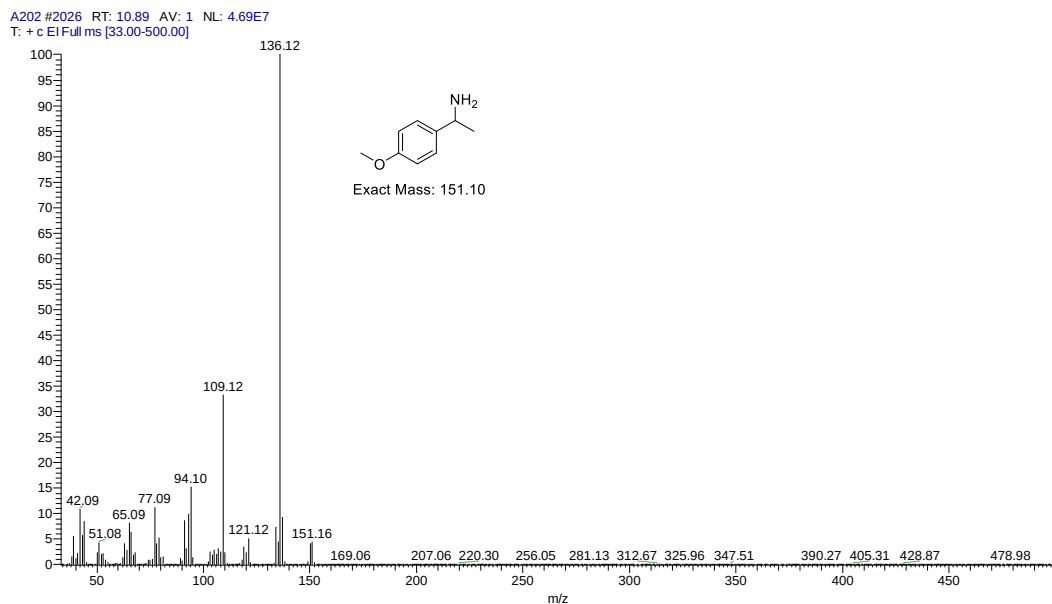
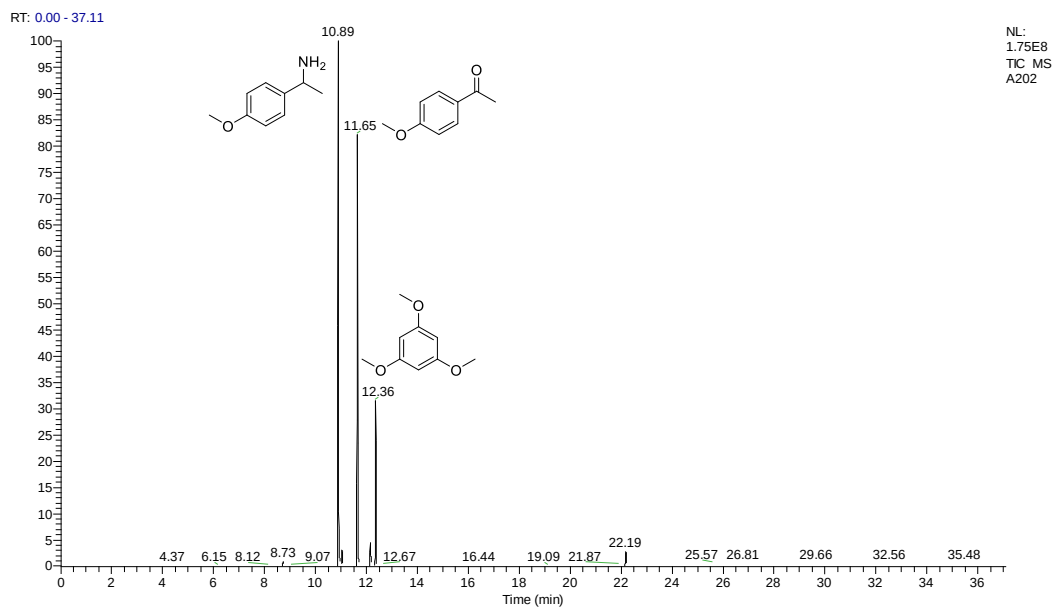
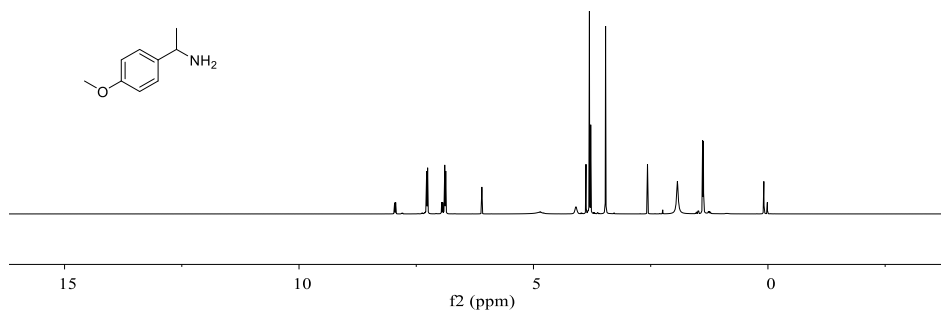
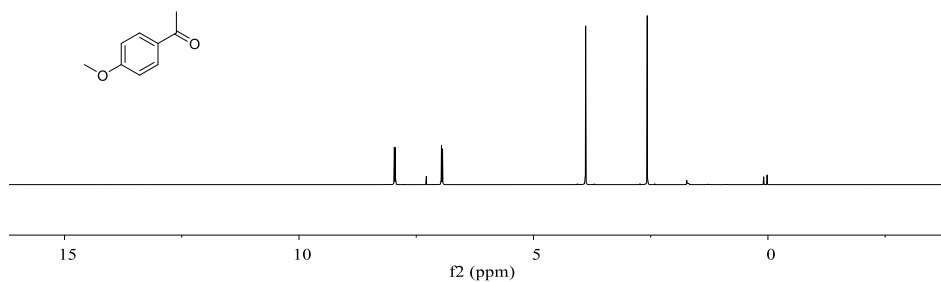


NL:
2.04E8
TIC MS
A204

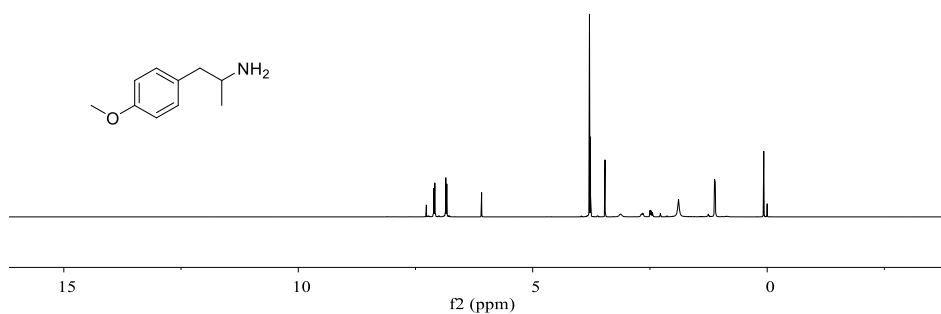
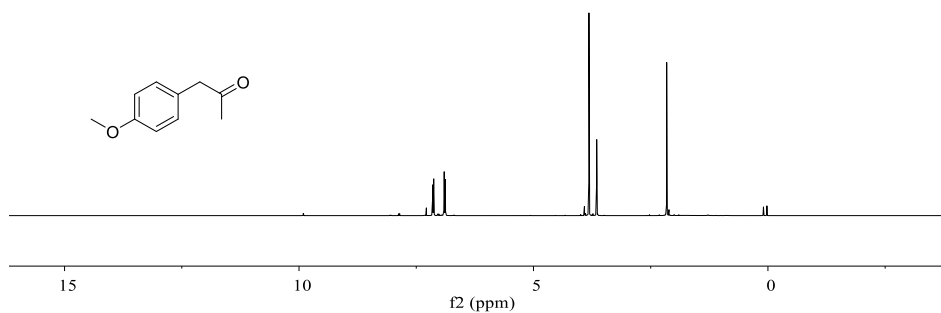
A204 #3170 RT: 14.78 AV: 1 NL: 4.32E7
T: + c EI Full ms [33.00-500.00]



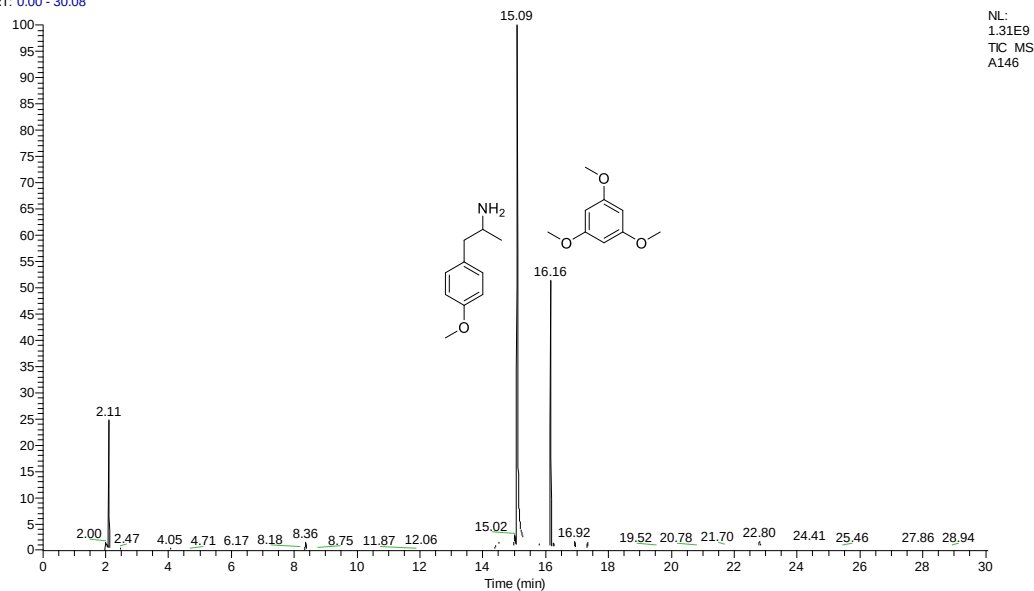
A168 (A202 GC-MS)



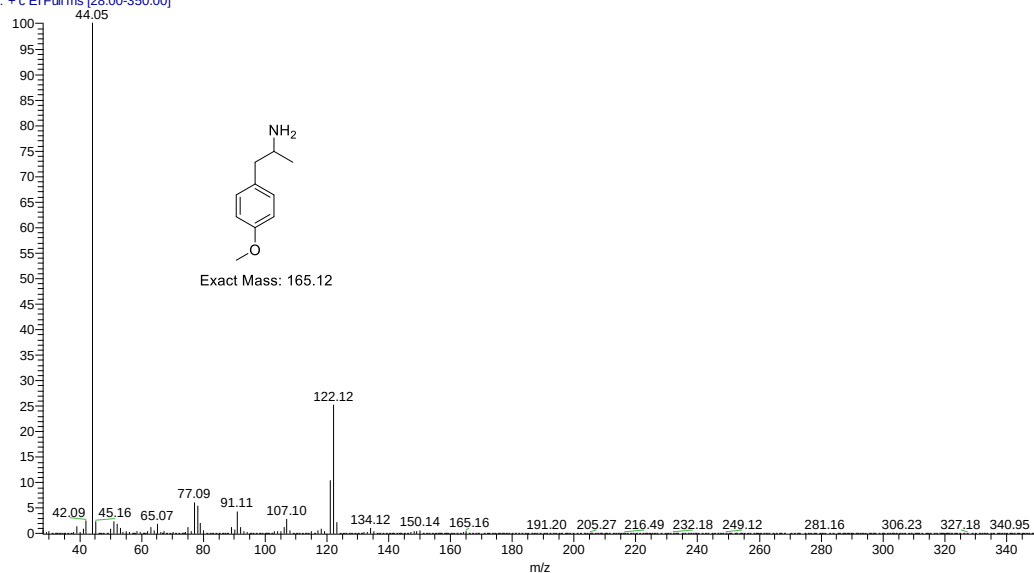
A146



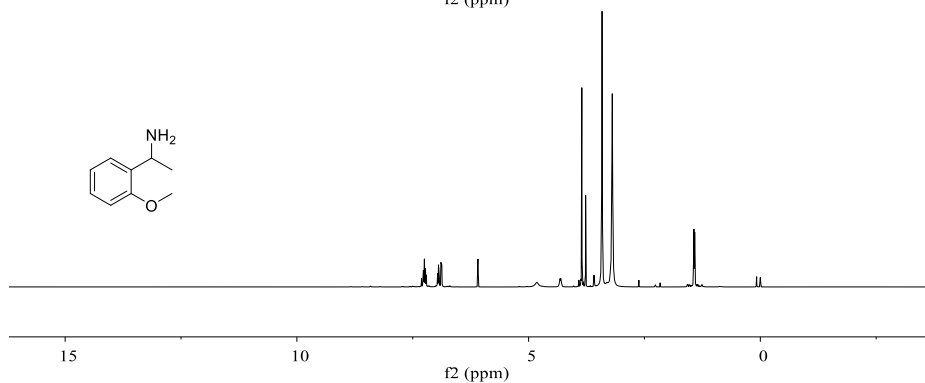
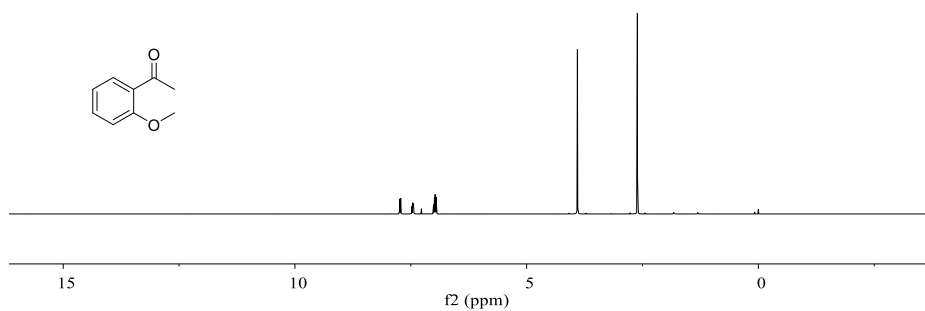
RT: 0.00 - 30.08



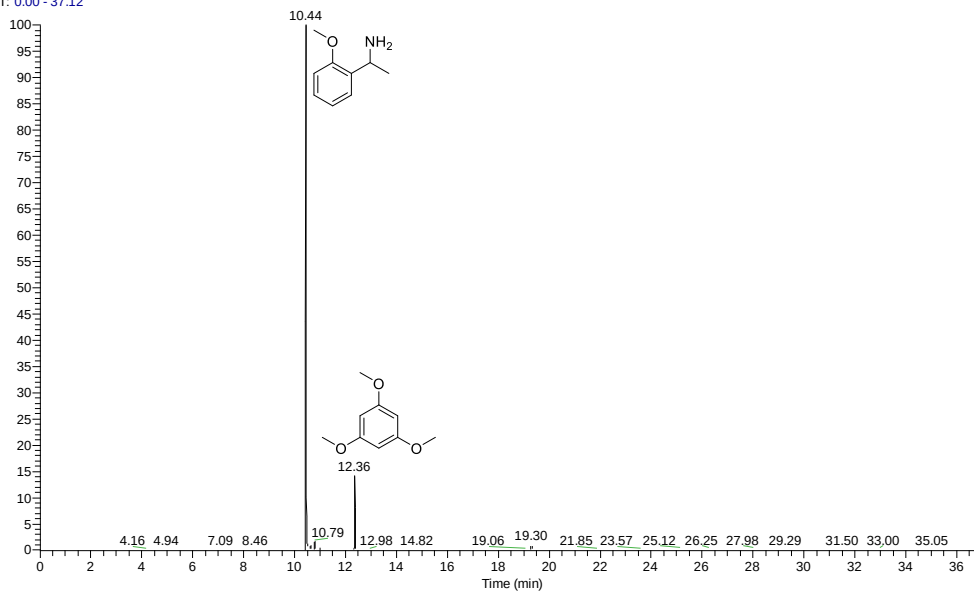
A146 #4428 RT: 15.09 AV: 1 NL: 6.65E8
T: + c EI Full ms [28.00-350.00]



A205

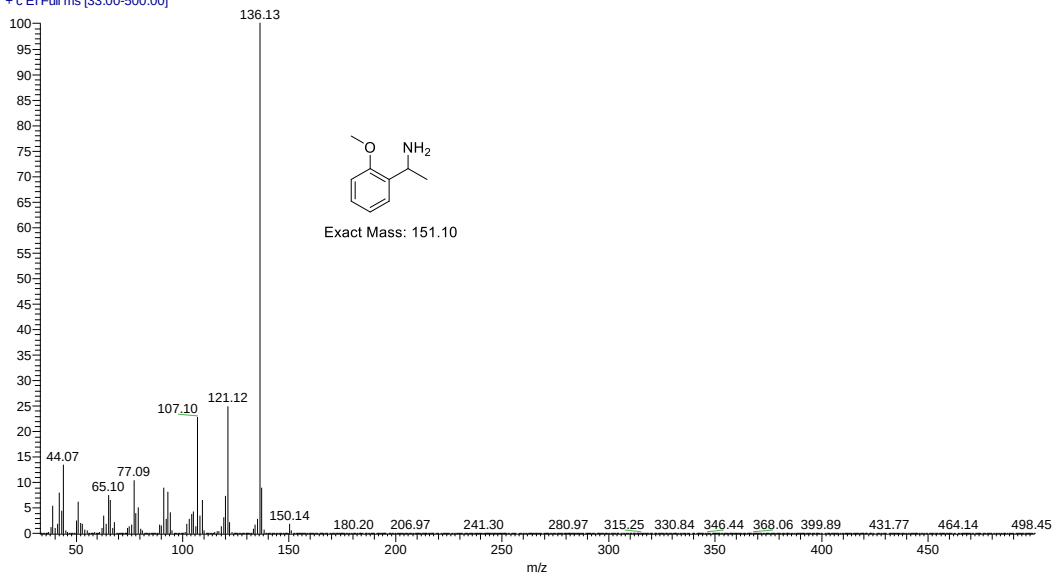


RT: 0.00 - 37.12

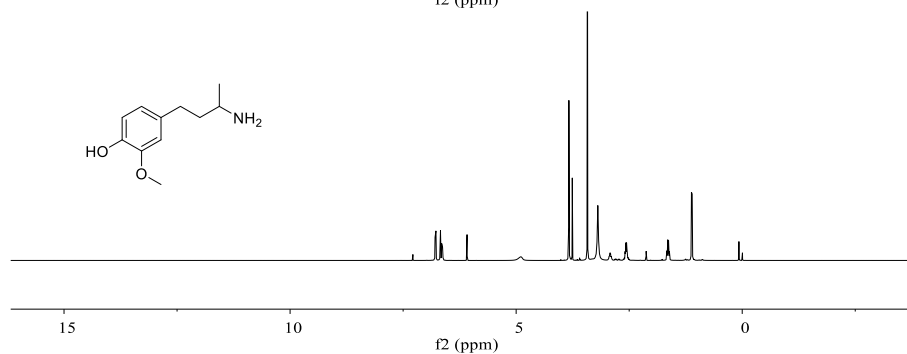
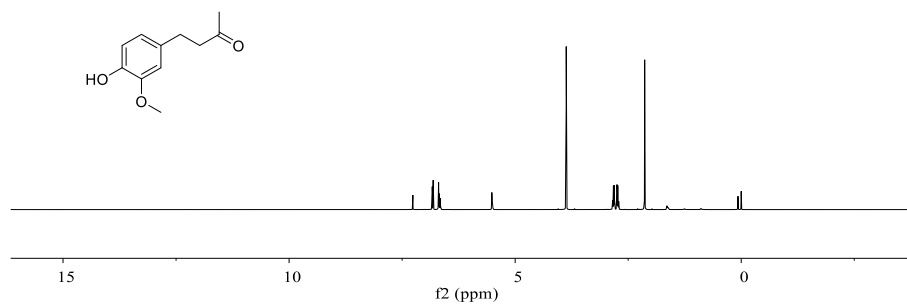


NL:
3.50E8
TIC: MS
A205

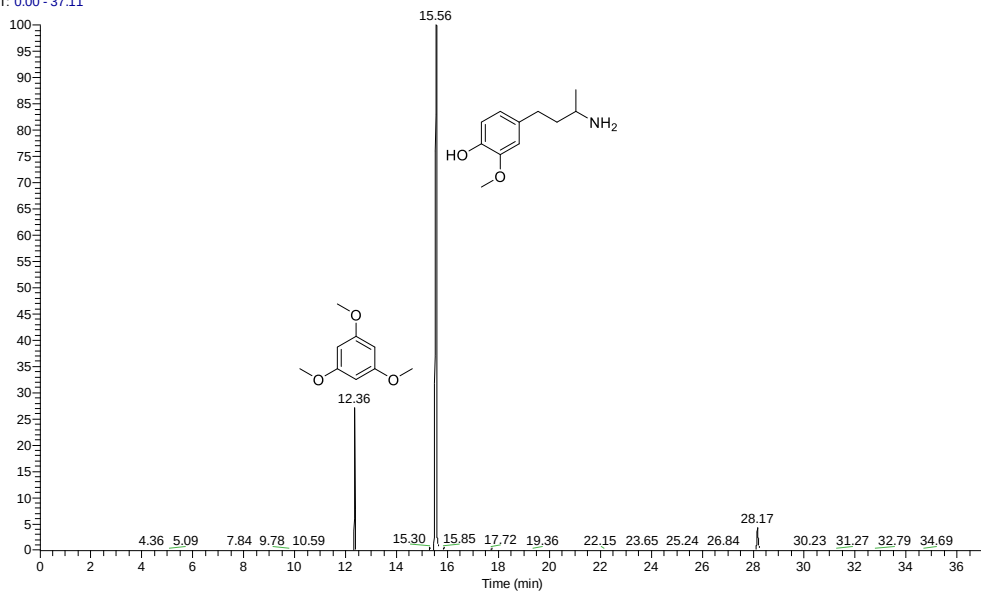
A205 #1893 RT: 10.44 AV: 1 NL: 6.89E7
T: + c EI Full ms [33.00-500.00]



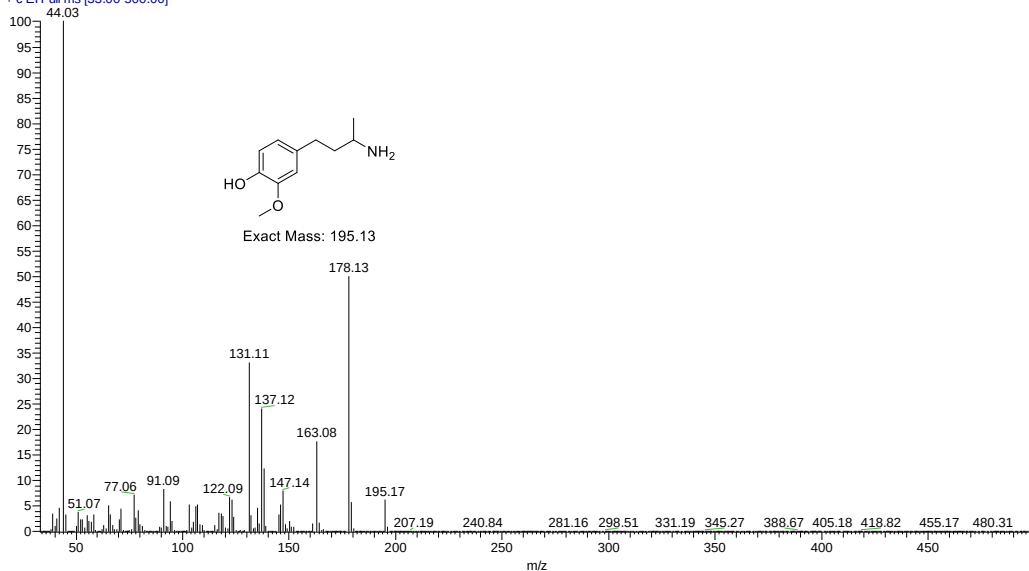
A179



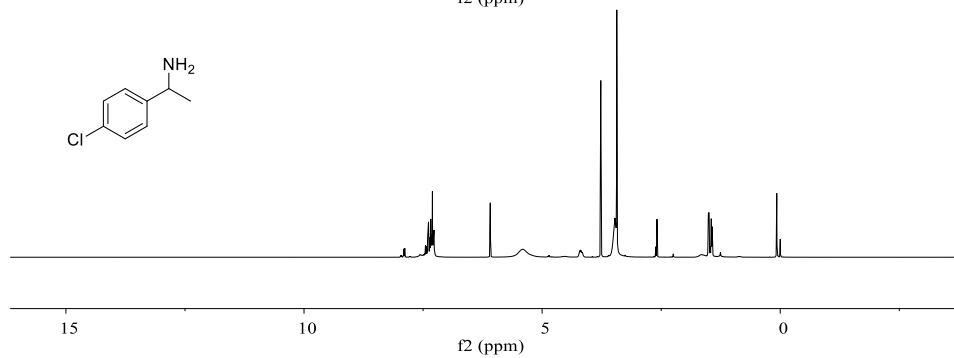
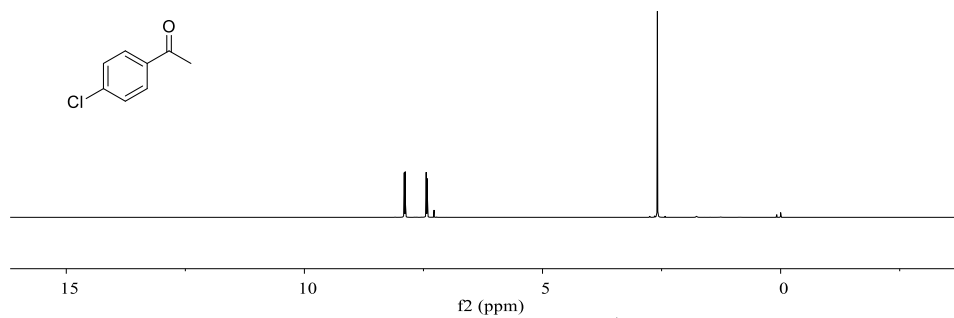
RT: 0.00 - 37.11



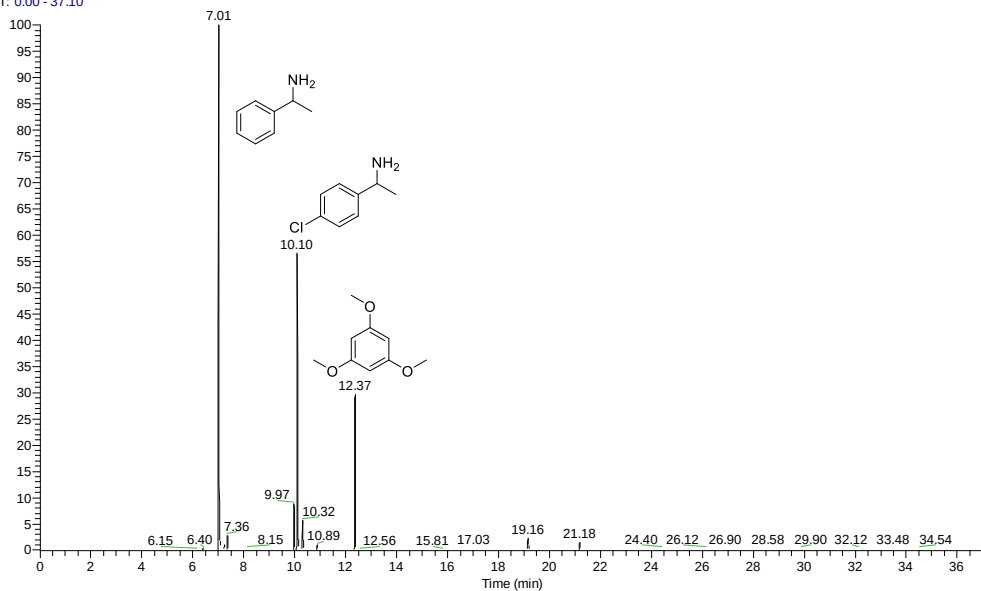
A179 #3401 RT: 15.56 AV: 1 NL: 3.28E8
T: + c EI Full ms [33.00-500.00]



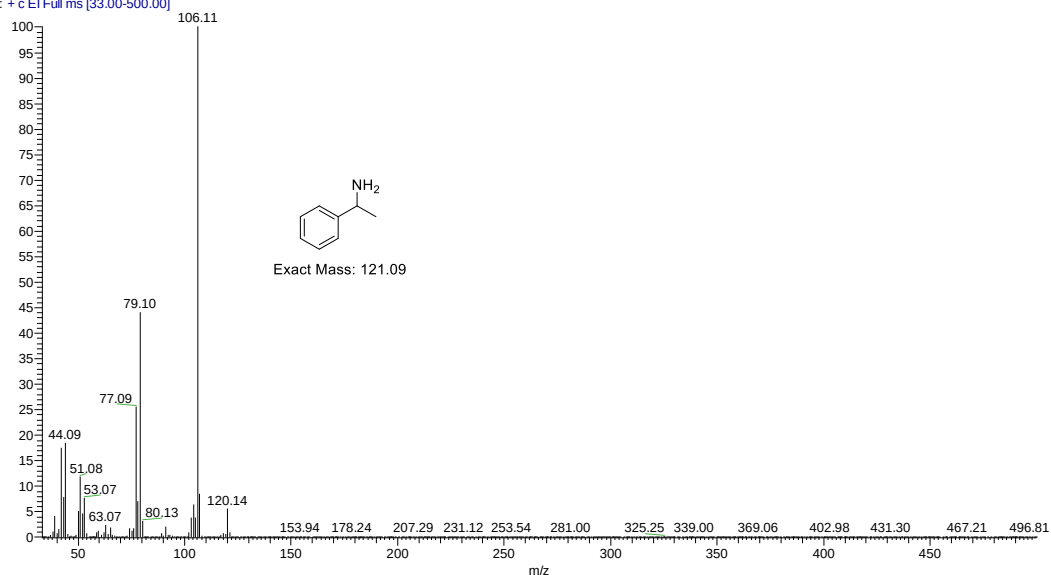
A206



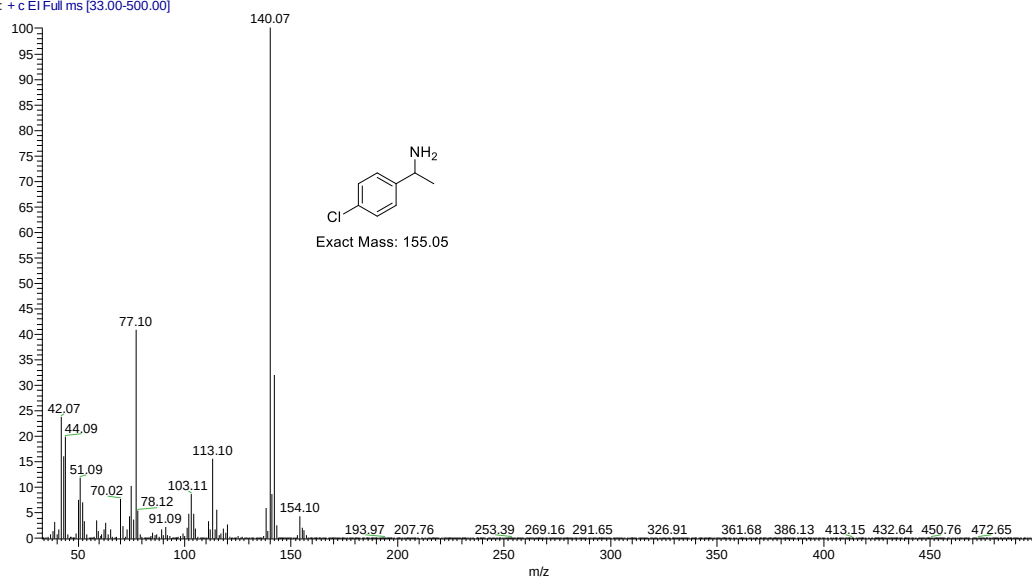
RT: 0.00 - 37.10



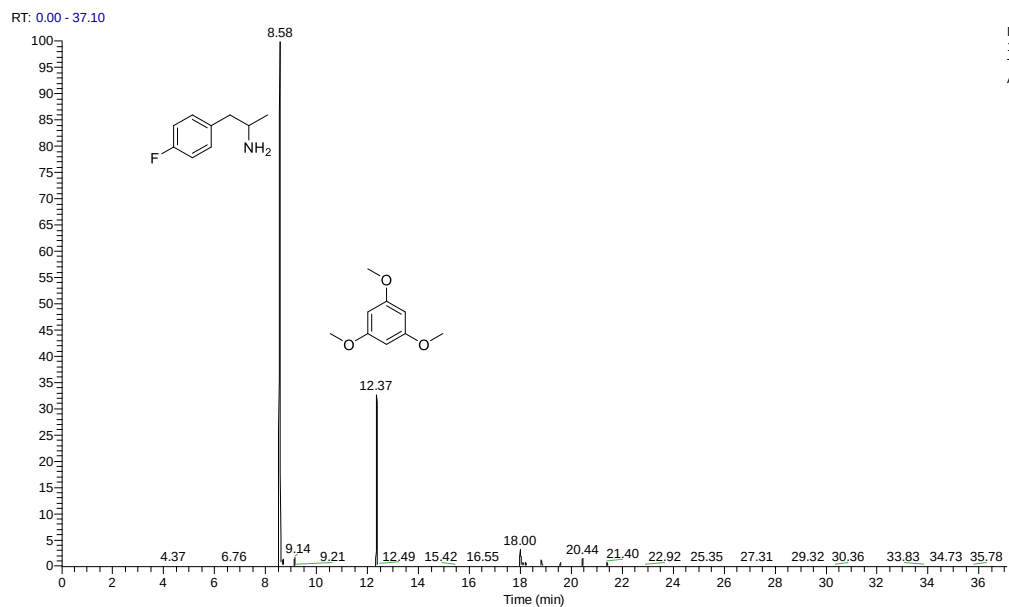
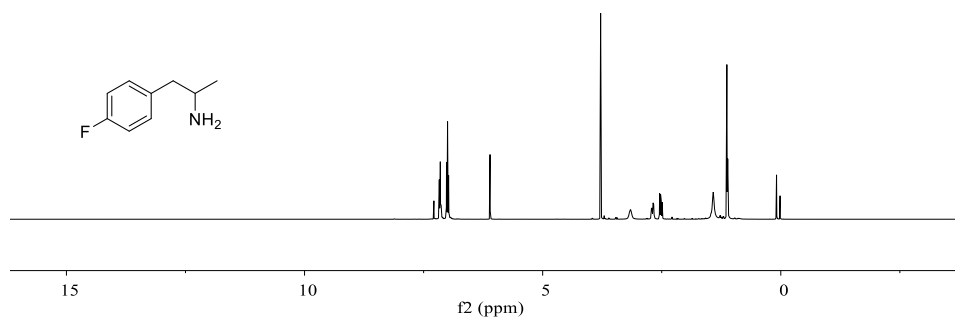
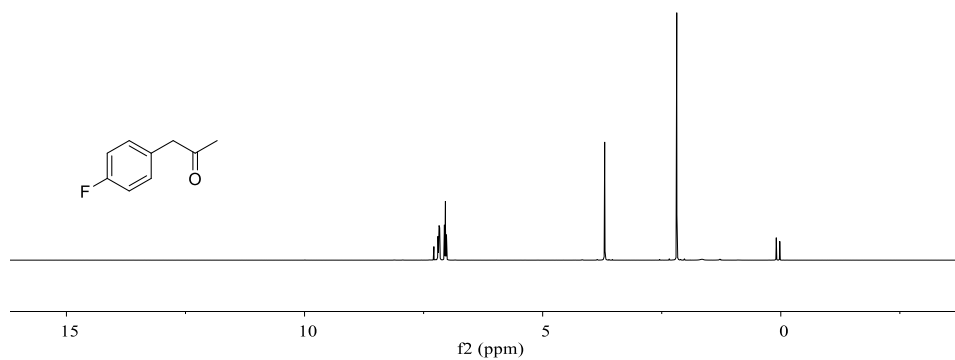
A206 #886 RT: 7.01 AV: 1 NL: 6.38E7
T: + c EI Full ms [33.00-500.00]



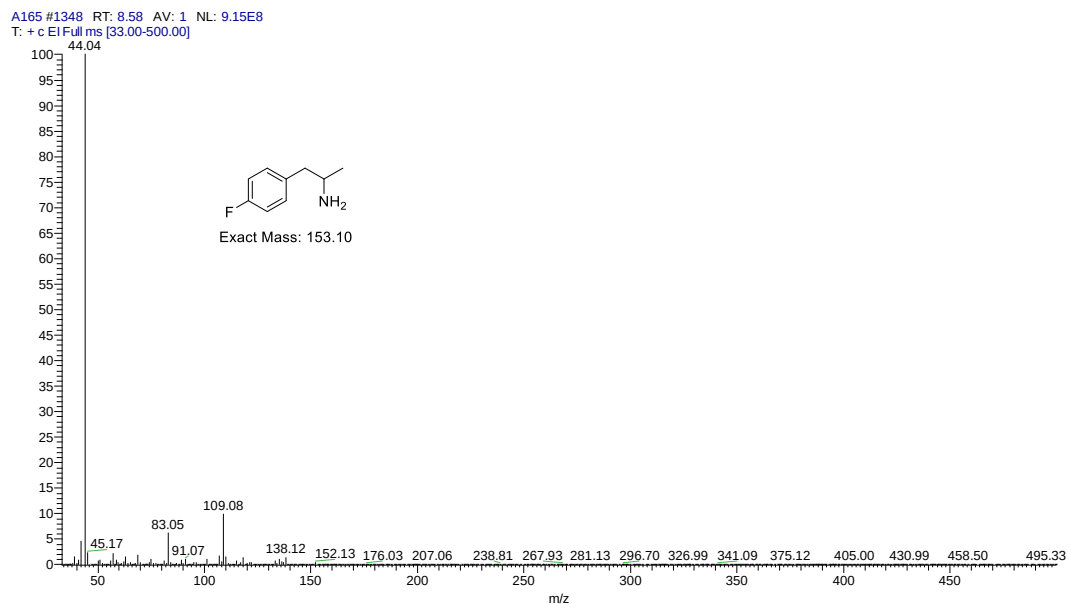
A206 #1794 RT: 10.10 AV: 1 NL: 2.93E7
T: + c EI Full ms [33.00-500.00]



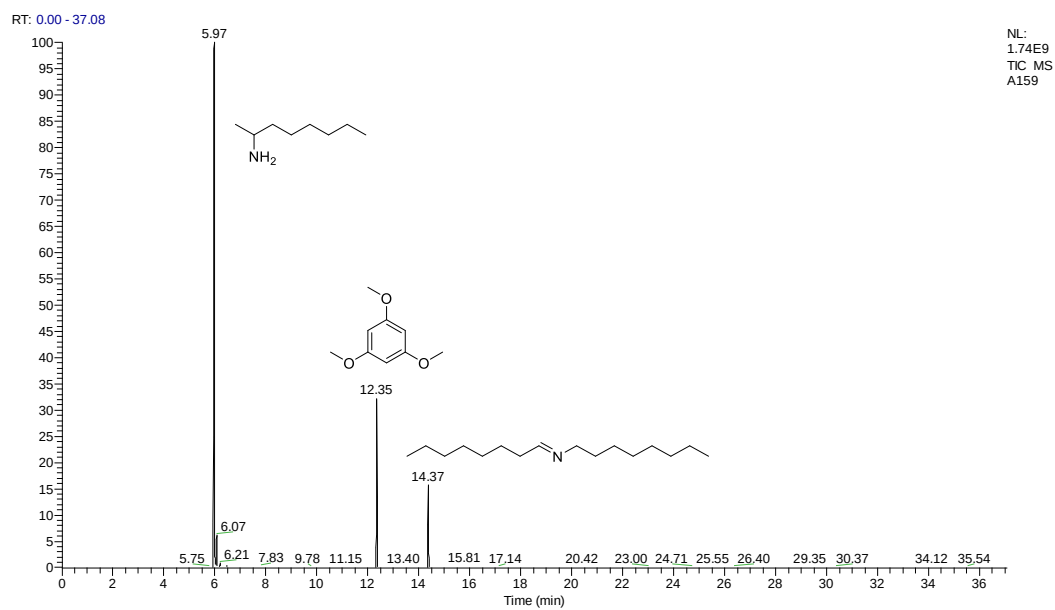
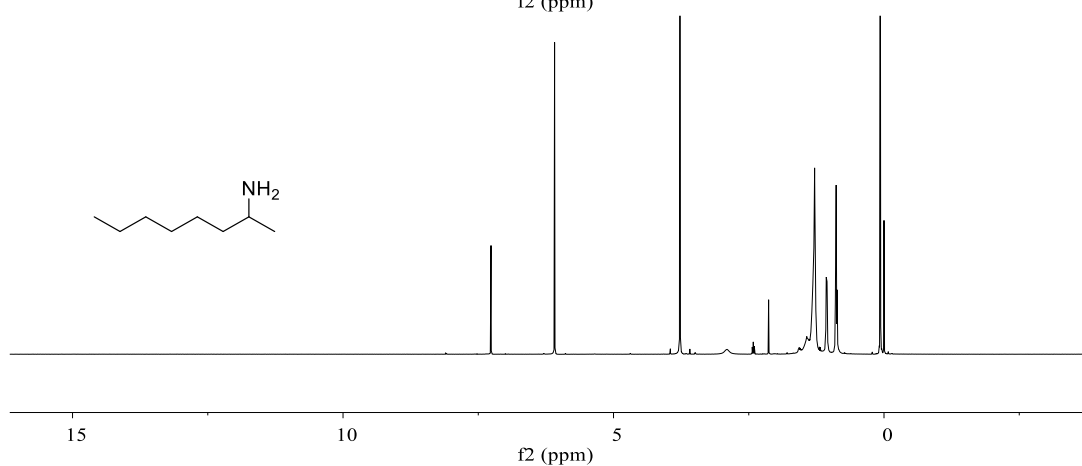
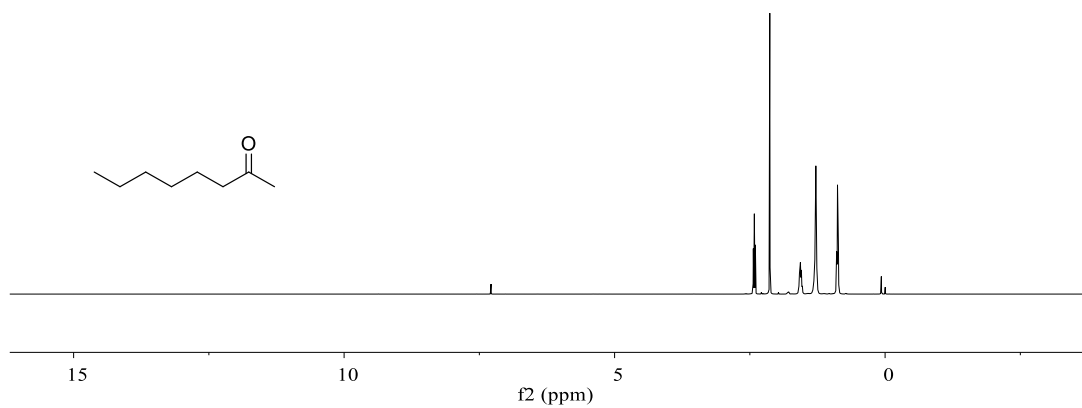
A165



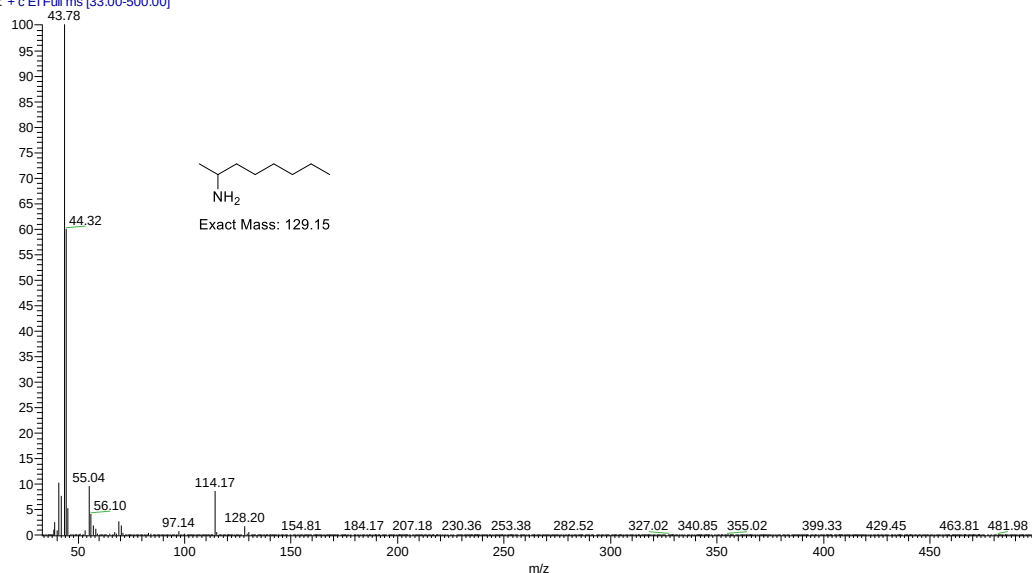
NL:
1.72E9
TIC MS
A165



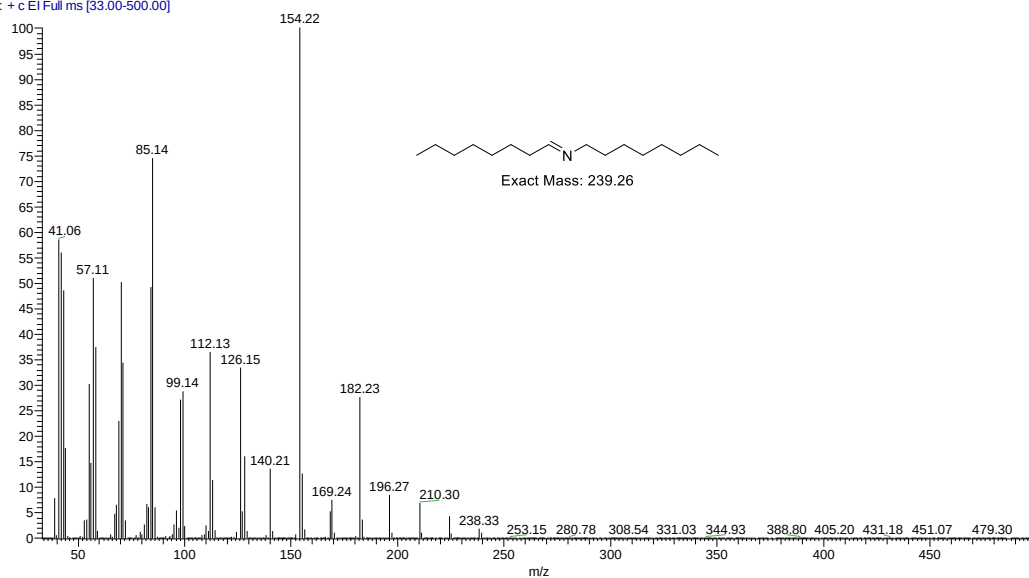
A159



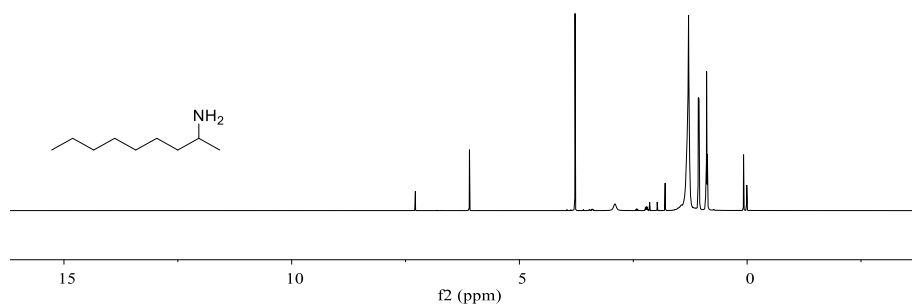
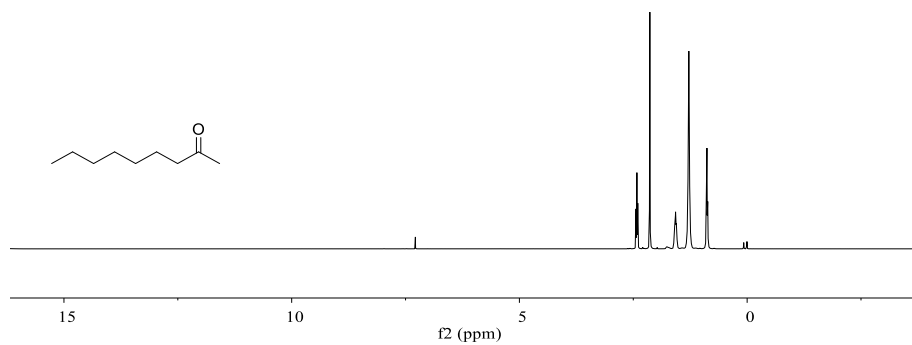
A159 #580 RT: 5.97 AV: 1 NL: 7.58E8
T: + c EI Full ms [33.00-500.00]



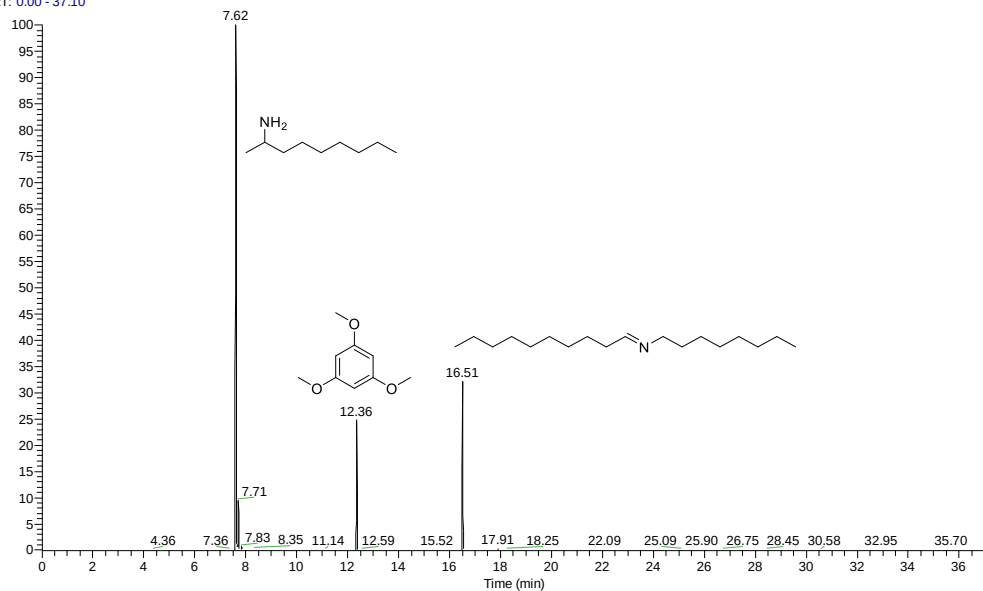
A159 #3050 RT: 14.37 AV: 1 NL: 2.77E7
T: + c EI Full ms [33.00-500.00]



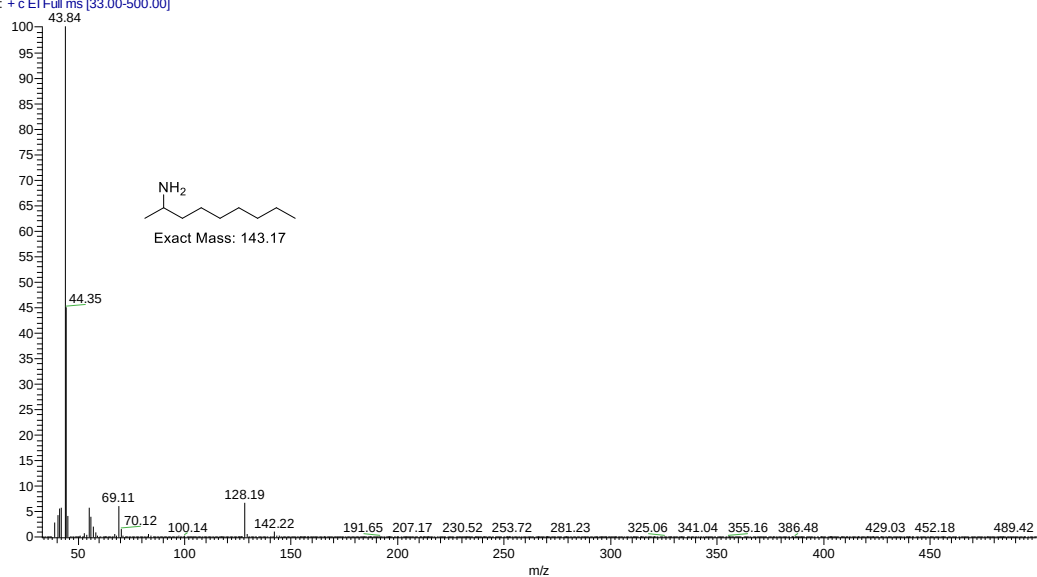
A181



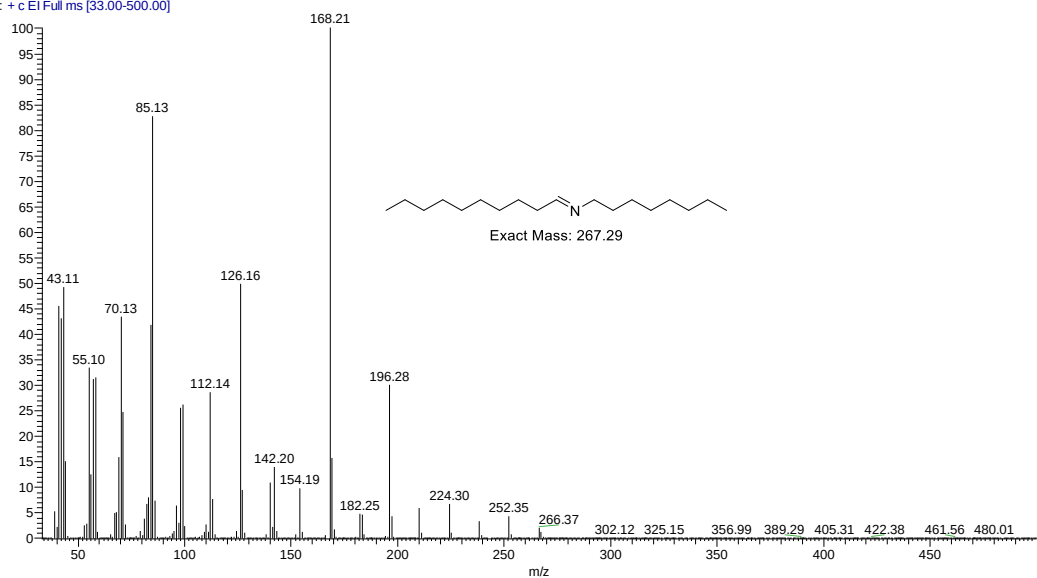
RT: 0.00 - 37.10



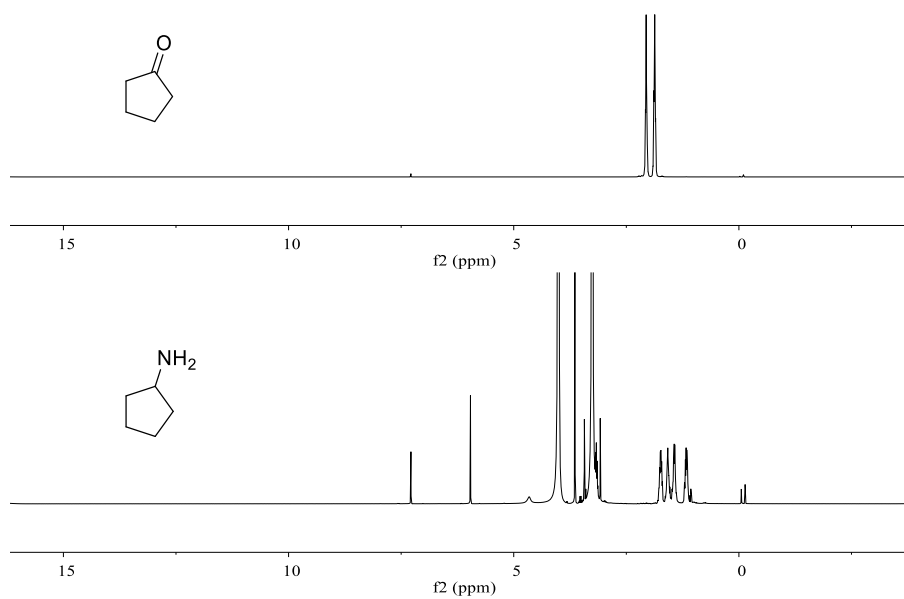
A181 #1065 RT: 7.62 AV: 1 NL: 7.87E8
T: + c EI Full ms [33.00-500.00]



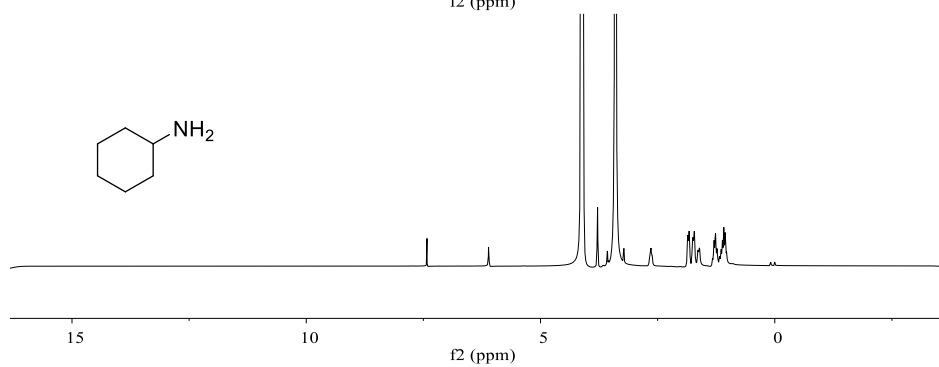
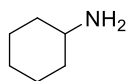
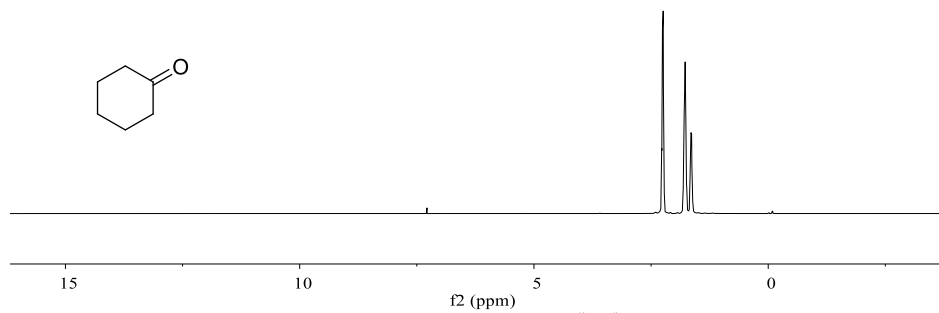
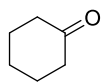
A181 #3679 RT: 16.51 AV: 1 NL: 5.65E7
T: + c EI Full ms [33.00-500.00]



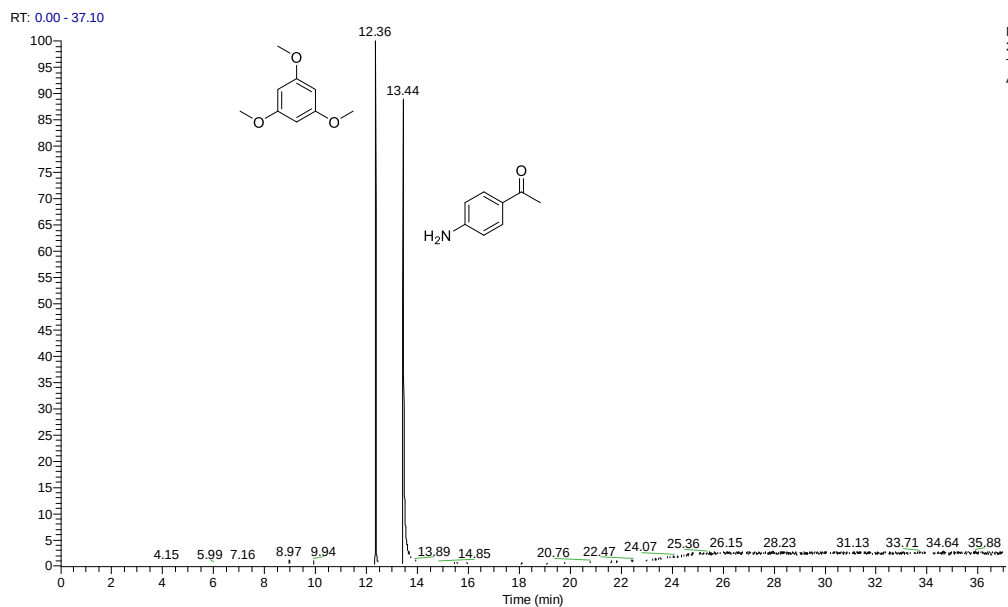
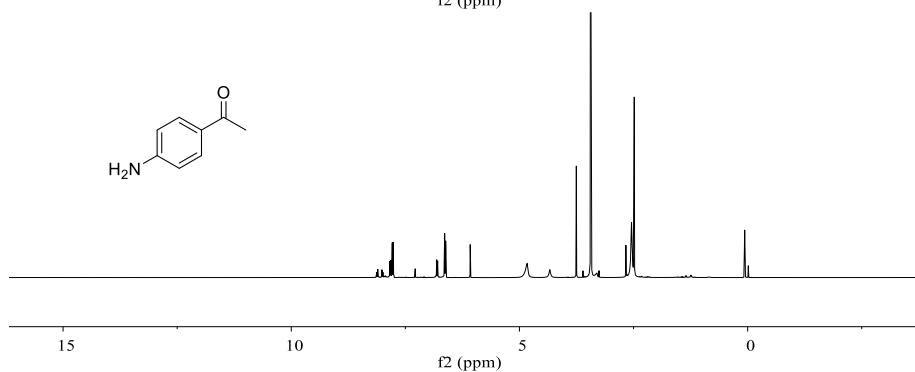
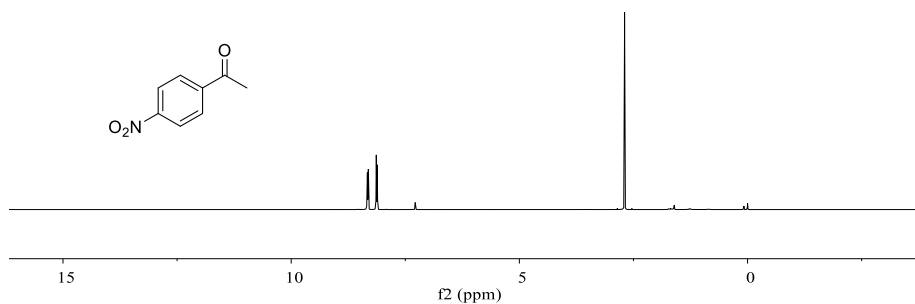
A278



A280

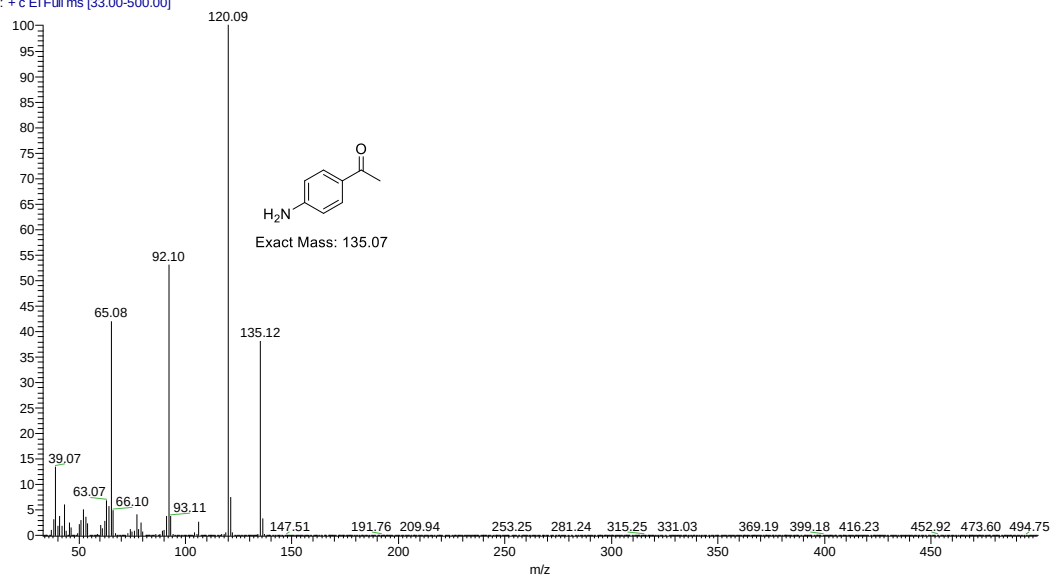


A203

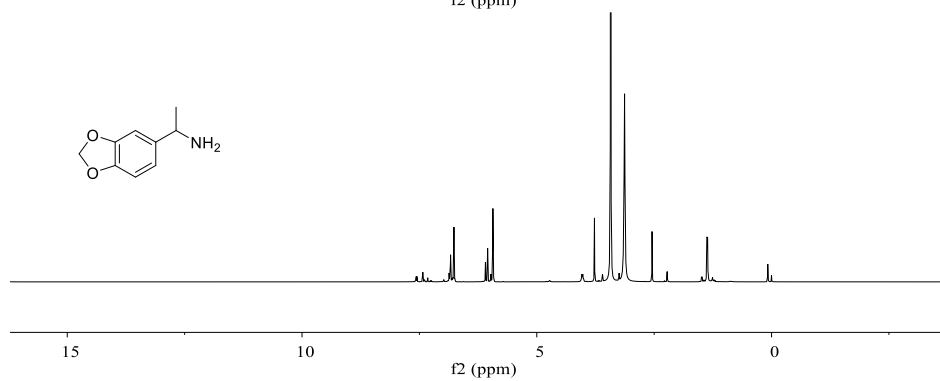
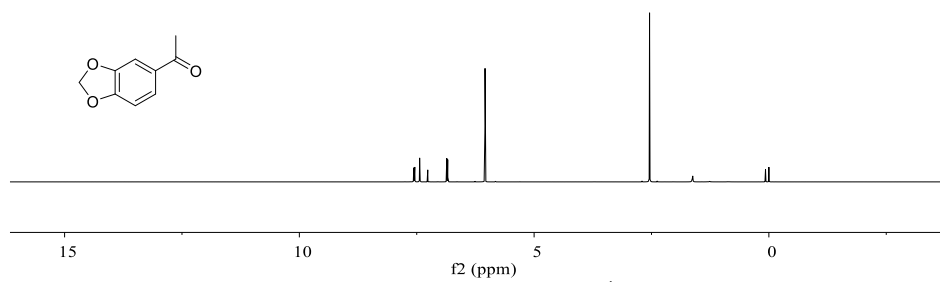


NL:
3.16E7
TIC MS
A203

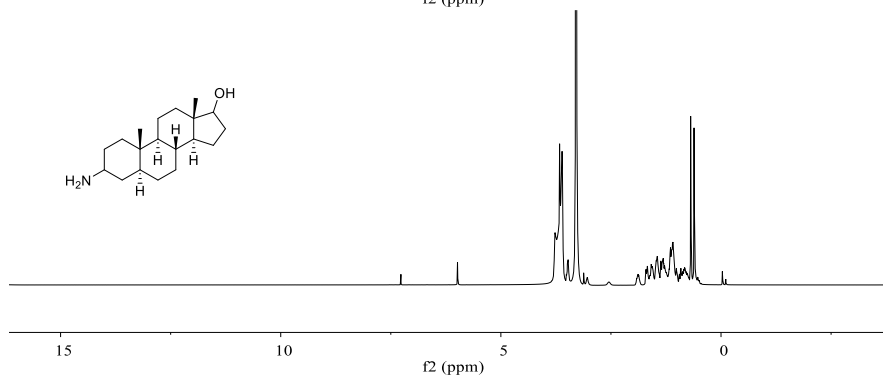
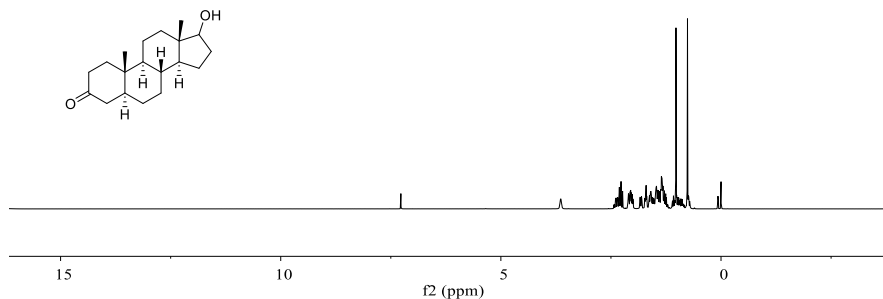
A203 #2776 RT: 13.44 AV: 1 NL: 7.65E6
T: + c EI Full ms [33.00-500.00]



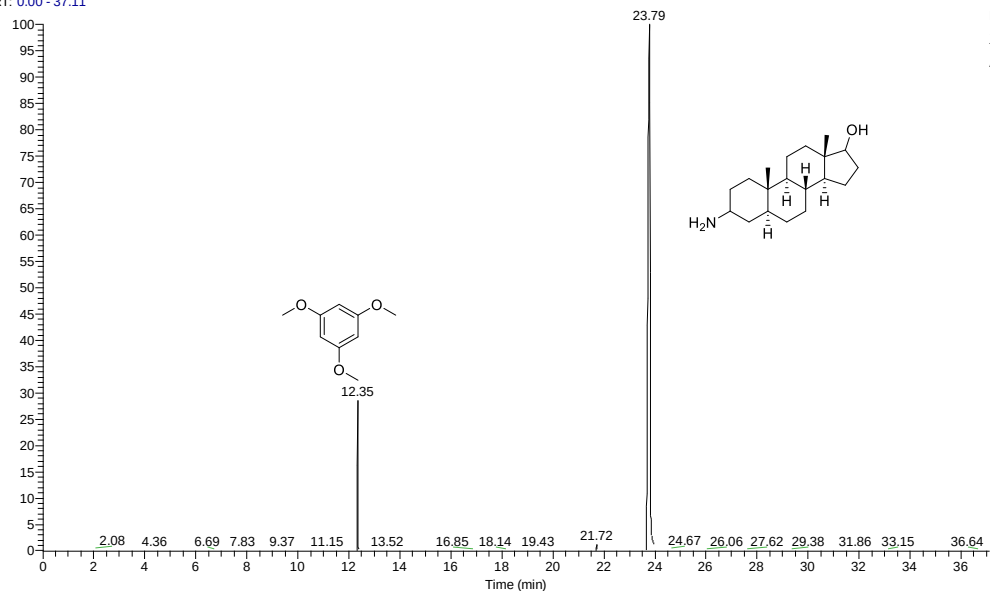
A230



A237

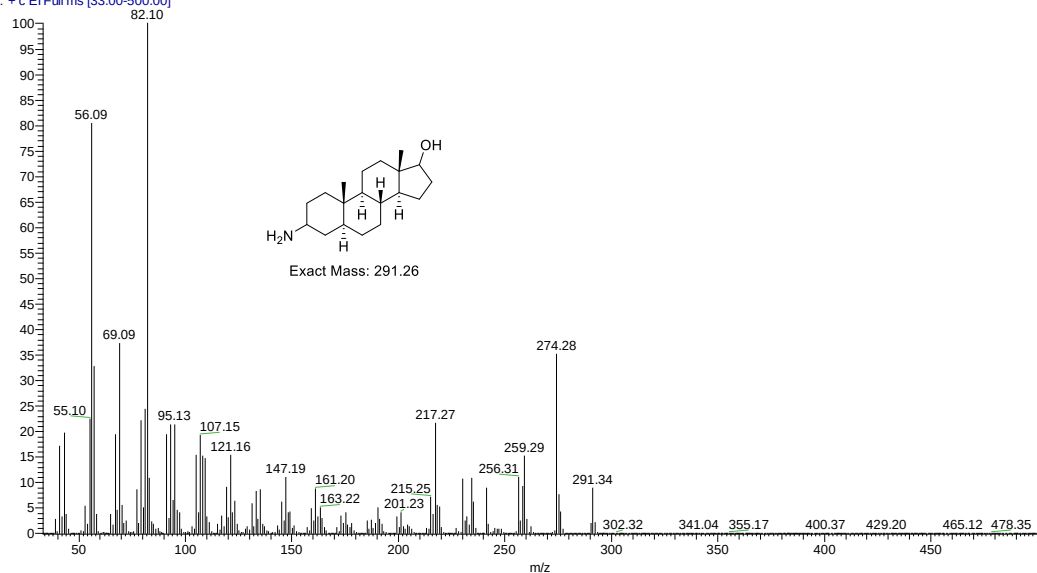


RT: 0.00 - 37.11

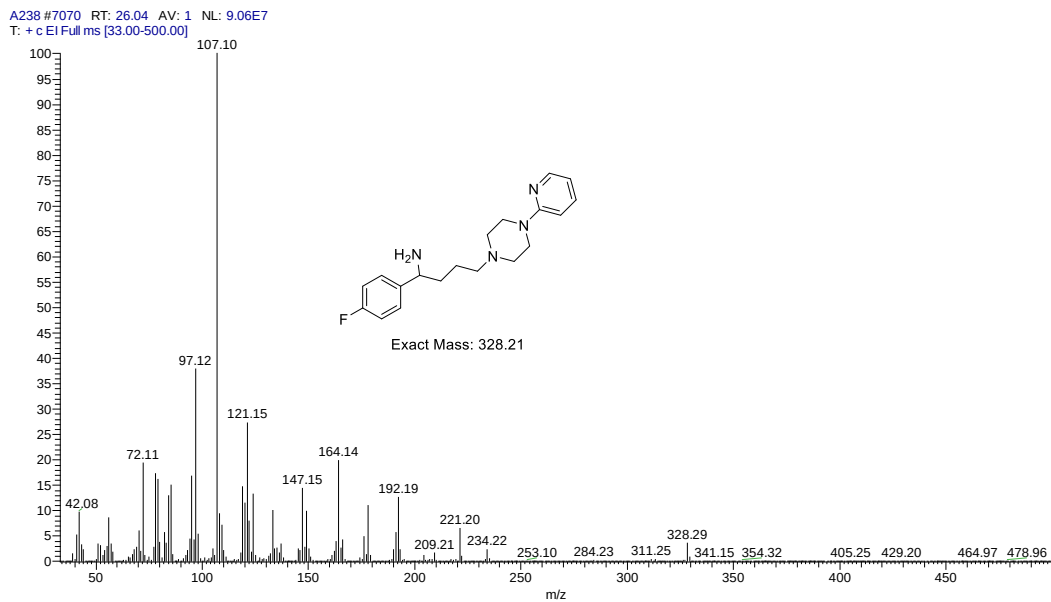
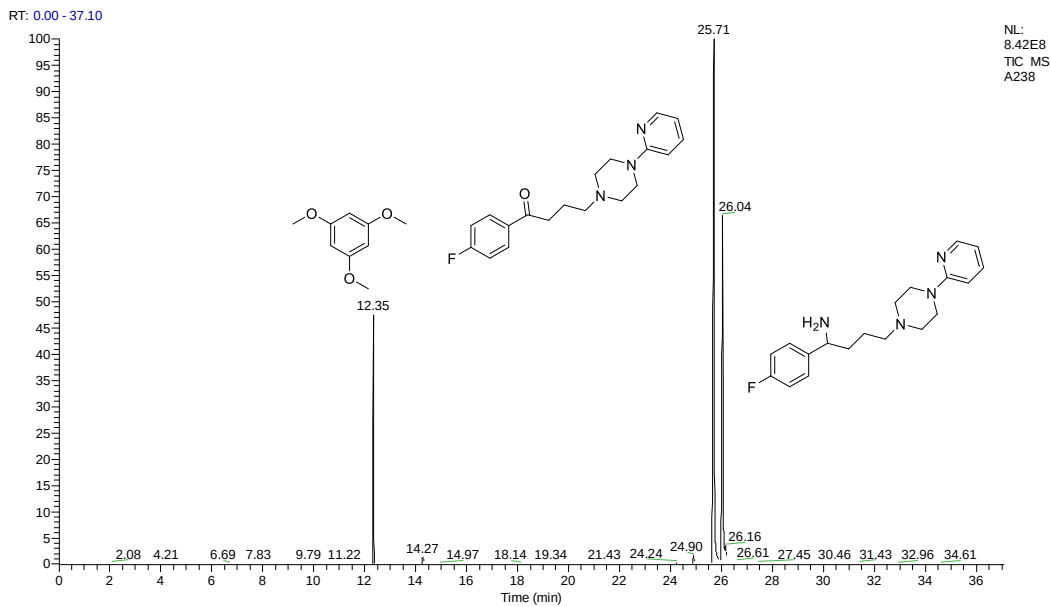
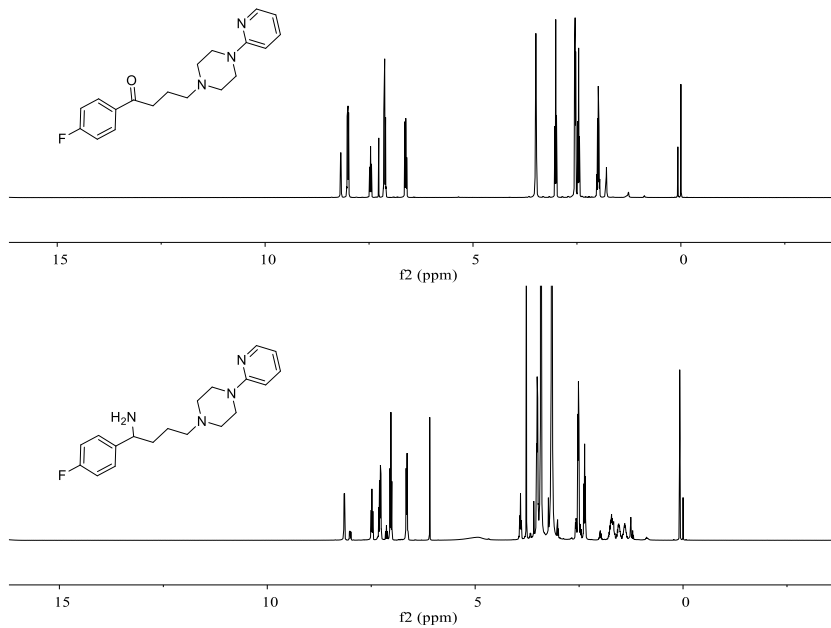


NL:
1.66E9
TIC MS
A237

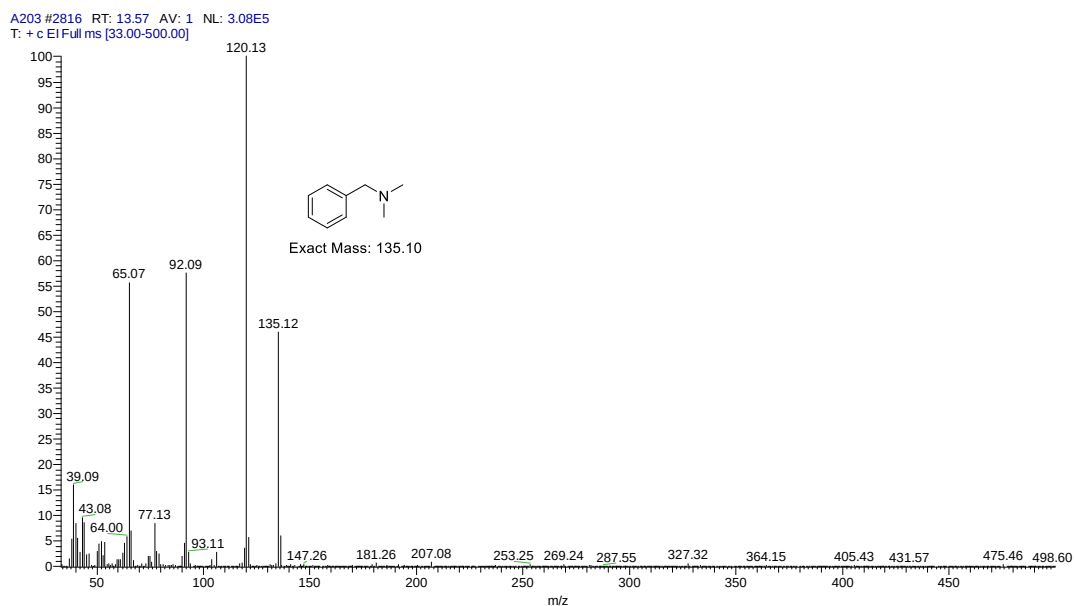
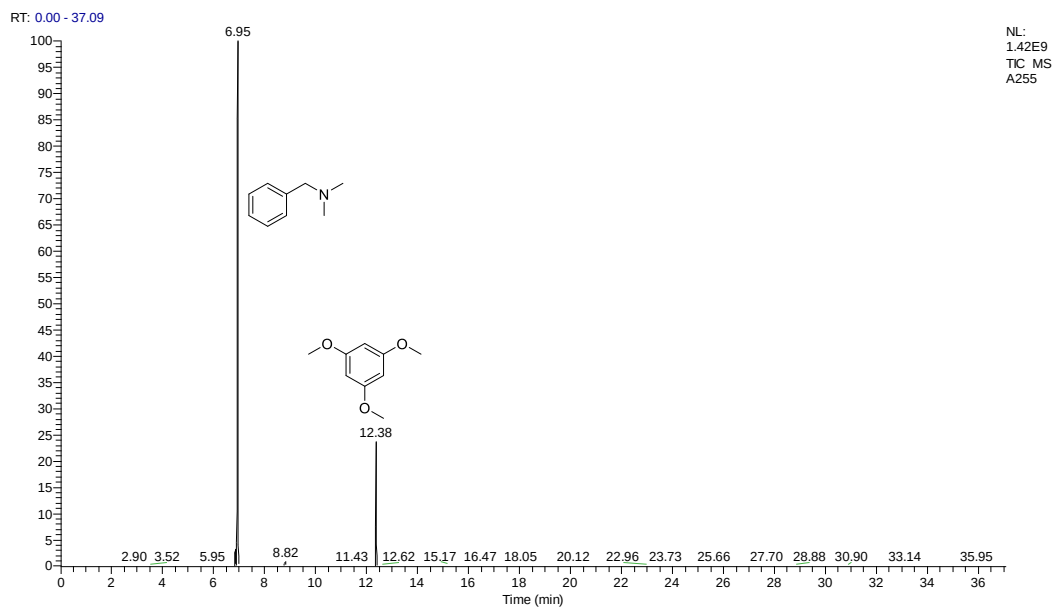
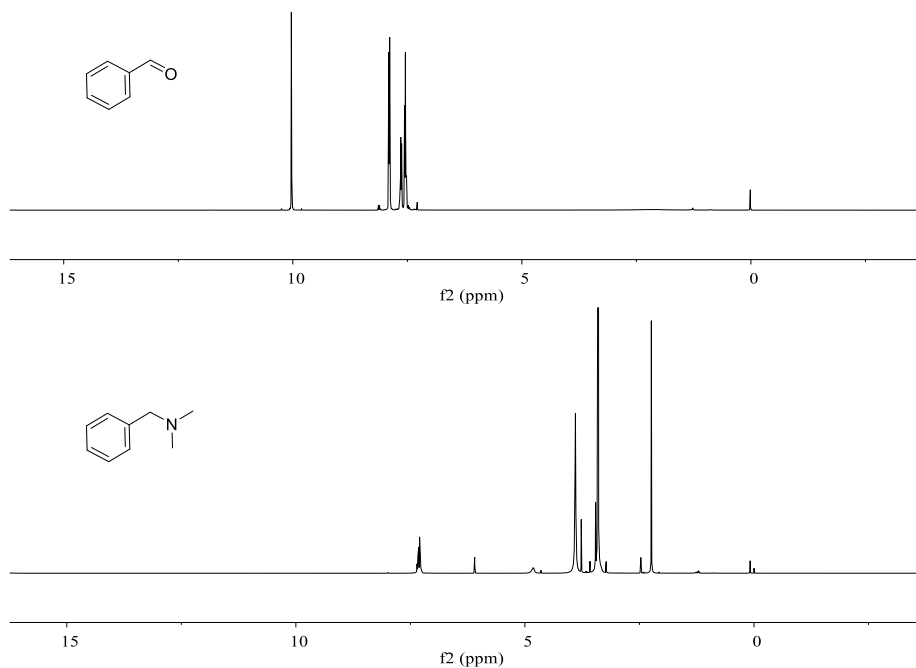
A237 #6394 RT: 23.74 AV: 1 NL: 1.34E8
T: + c EI Full ms [33.00-500.00]



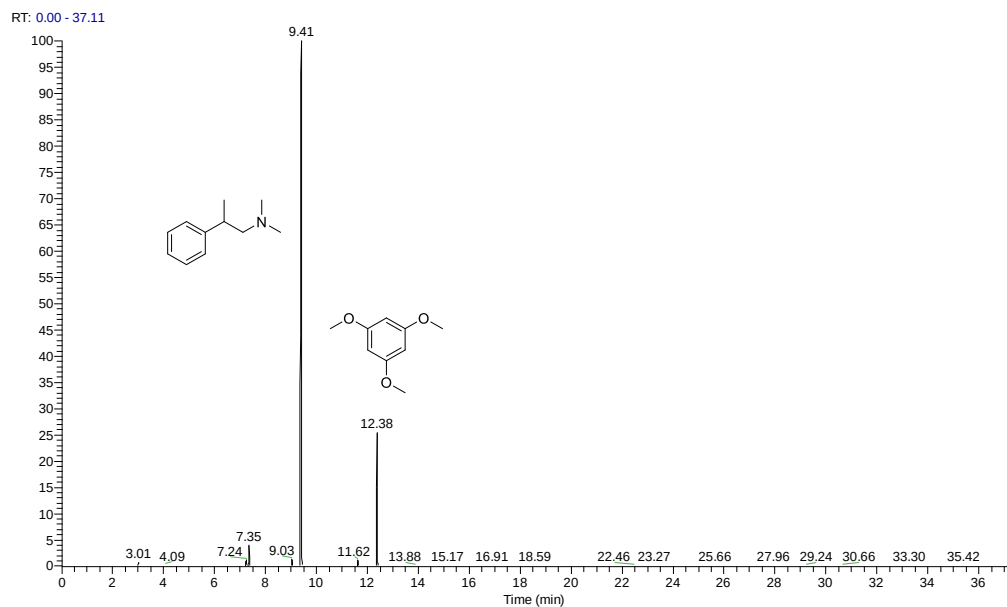
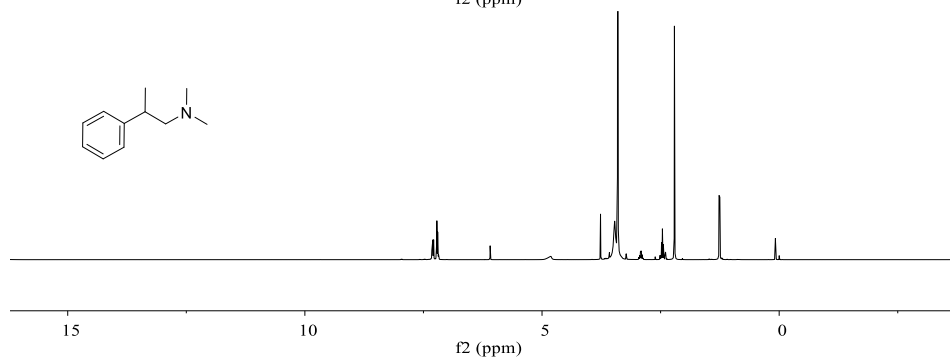
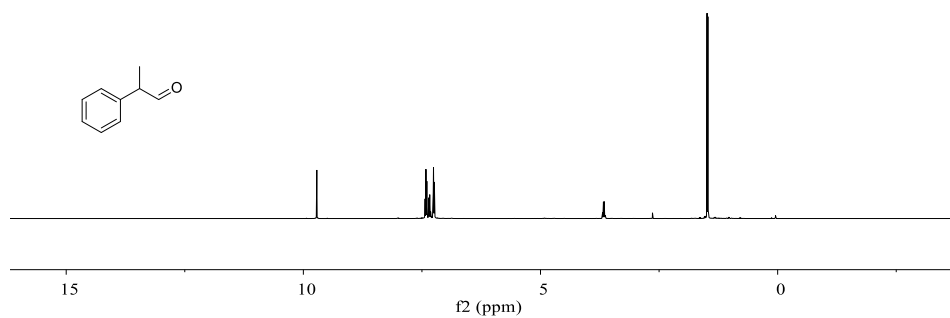
A268 (A238 GC-MS)



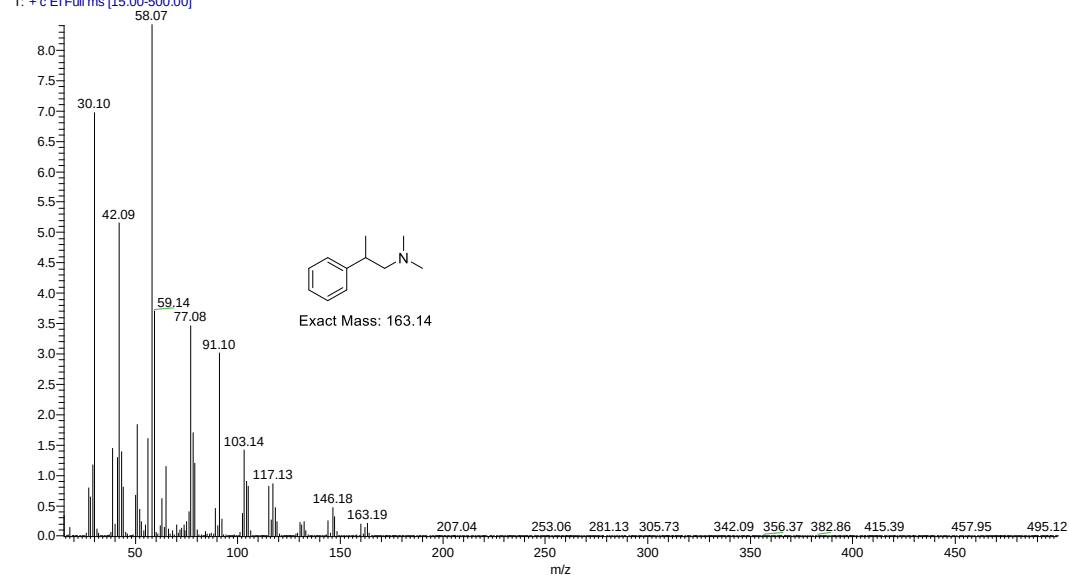
A255



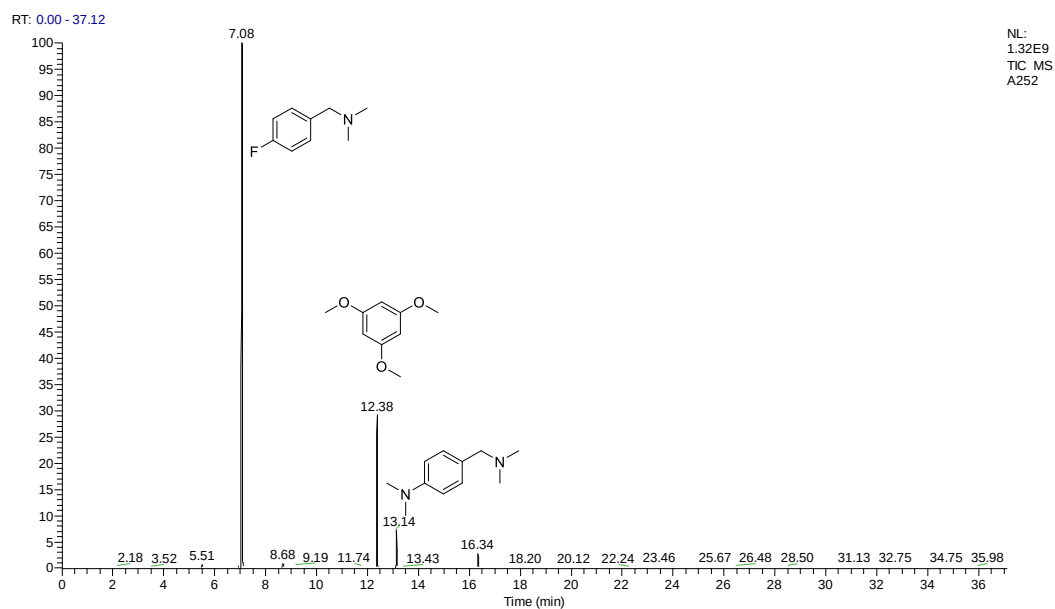
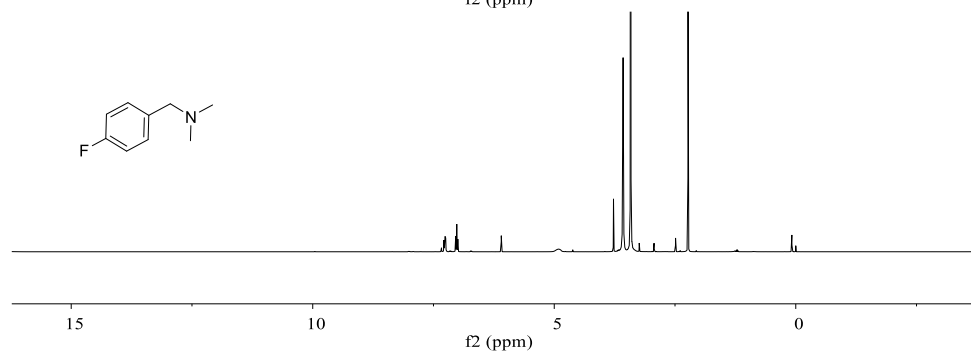
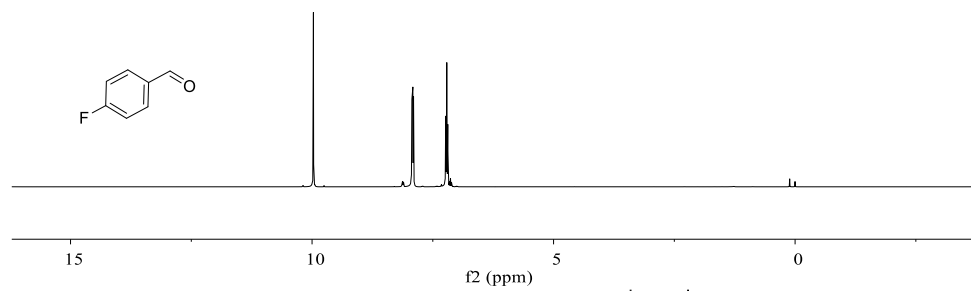
A257



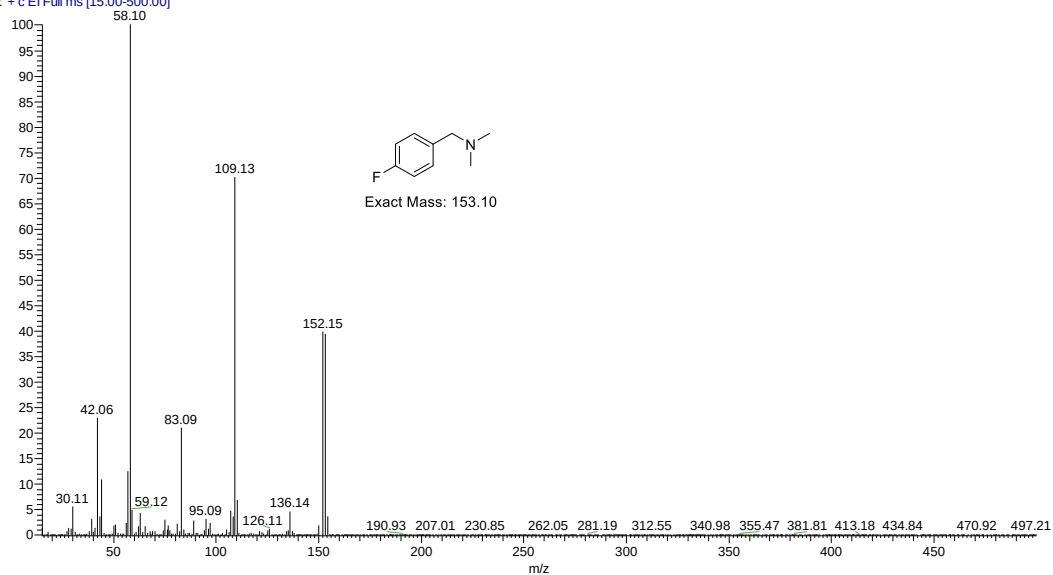
A257 #2178 RT: 9.41 AV: 1 NL: 9.14E8
T: + c EIfull ms [15.00-500.00]



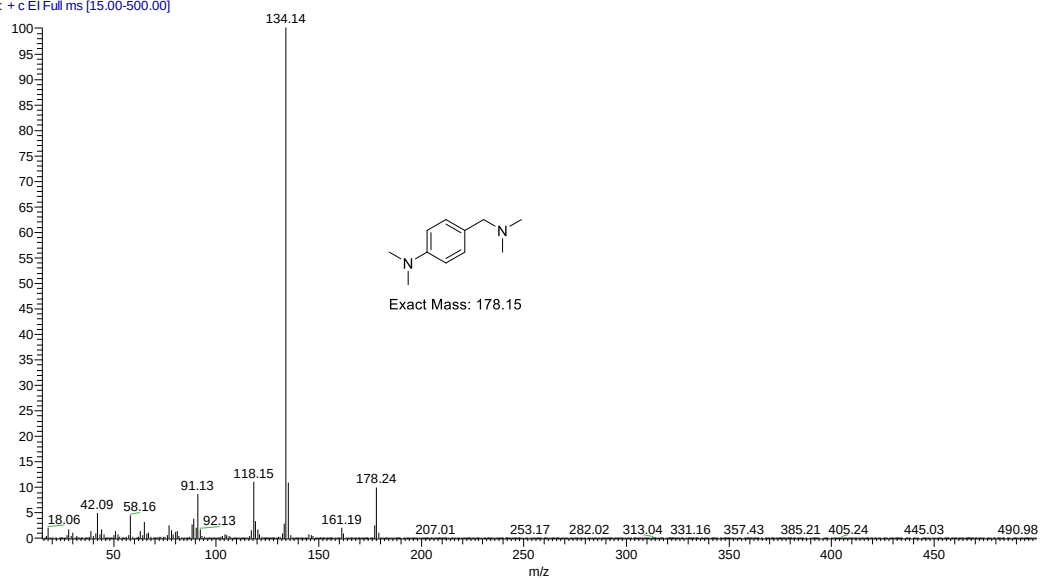
A252



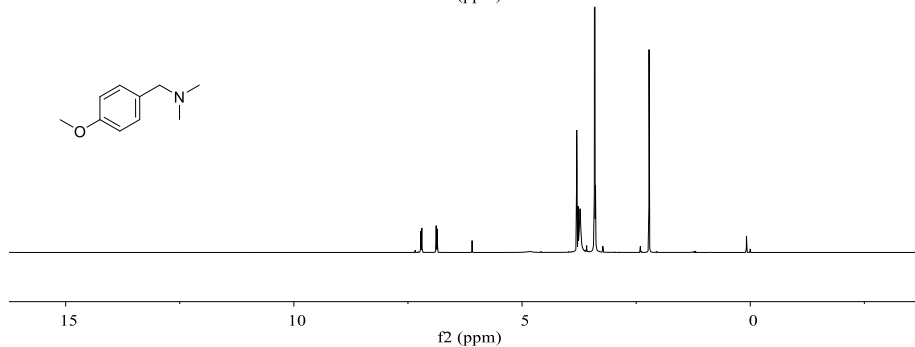
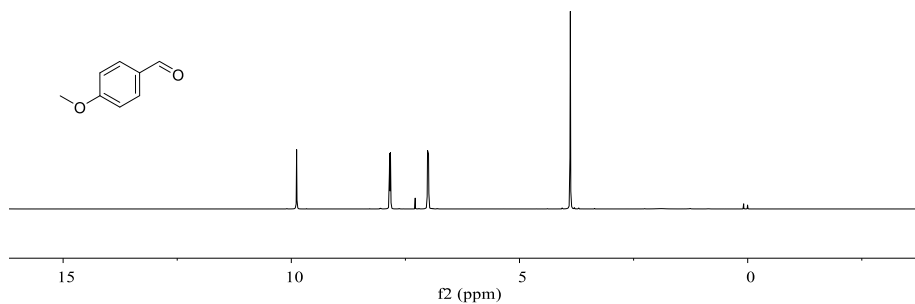
A252 #1495 RT: 7.08 AV: 1 NL: 3.05E8
T: + c EI Full ms [15.00-500.00]



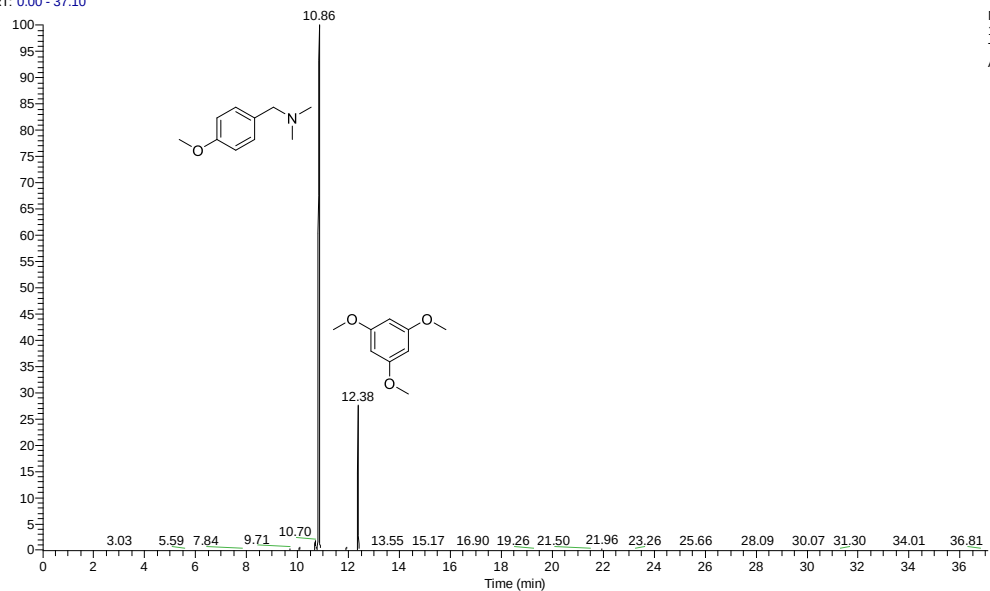
A252 #3276 RT: 13.14 AV: 1 NL: 4.40E7
T: + c EI Full ms [15.00-500.00]



A256

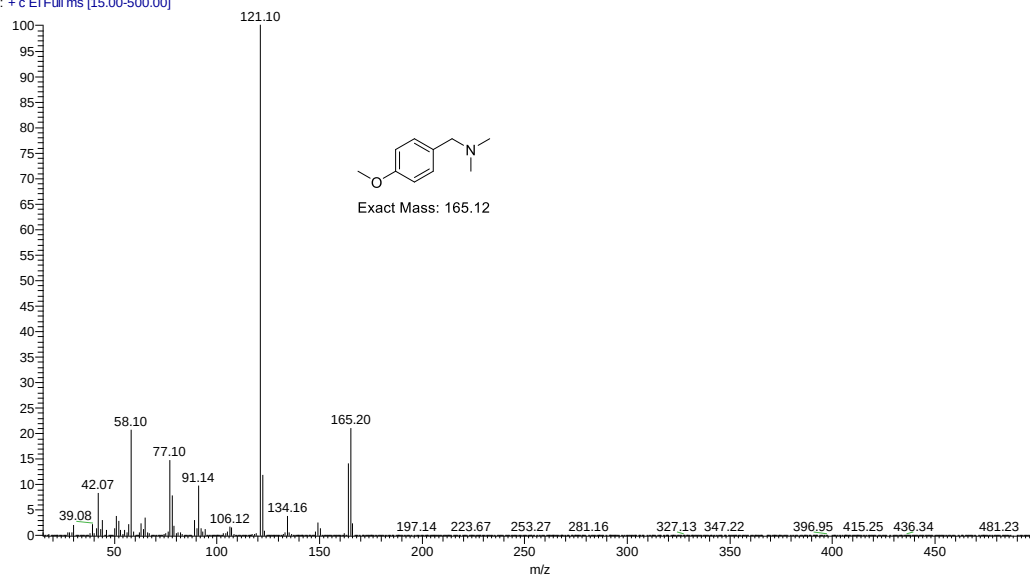


RT: 0.00 - 37.10

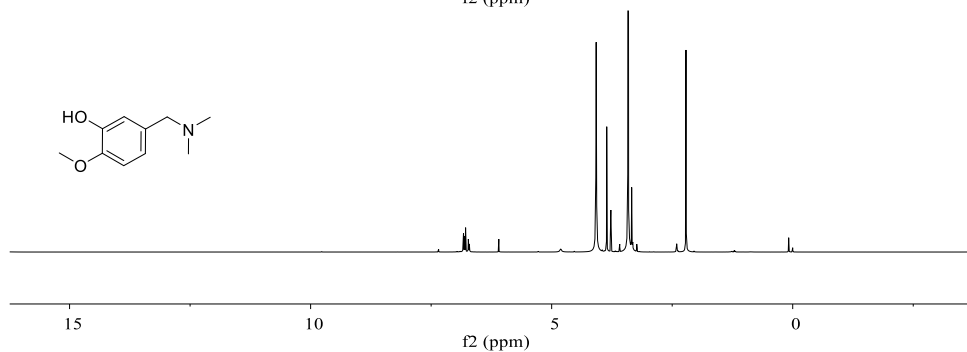
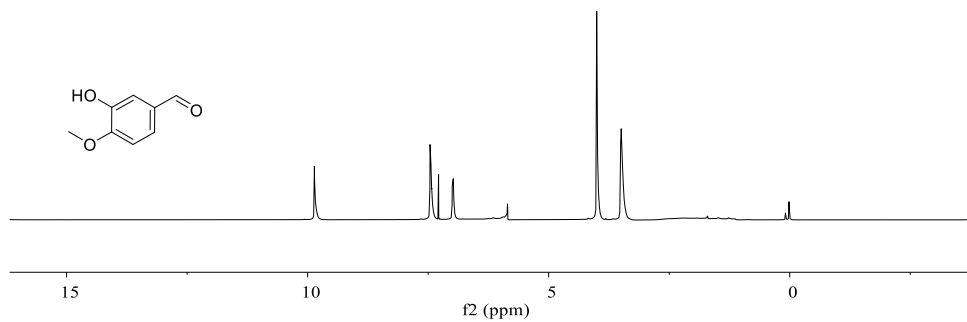


NL:
1.36E9
TIC MS
A256

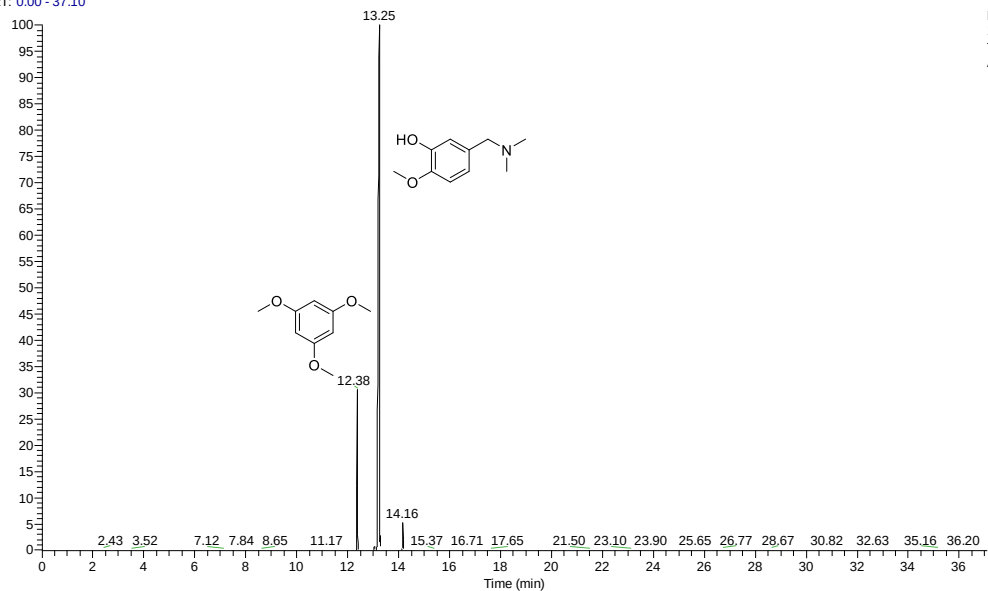
A256 #2604 RT: 10.85 AV: 1 NL: 4.64E8
T: + c EI Full ms [15.00-500.00]



A253

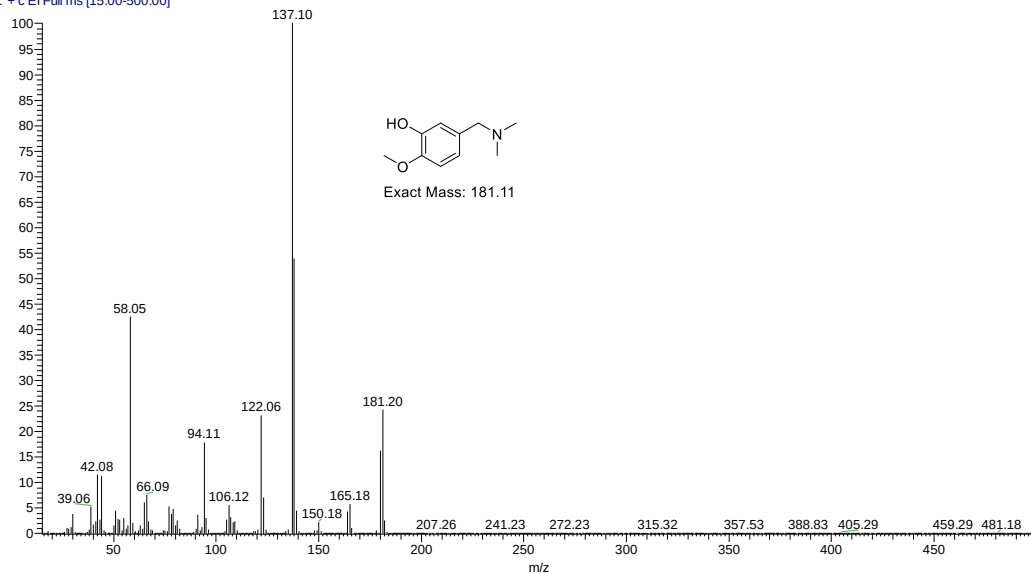


RT: 0.00 - 37.10

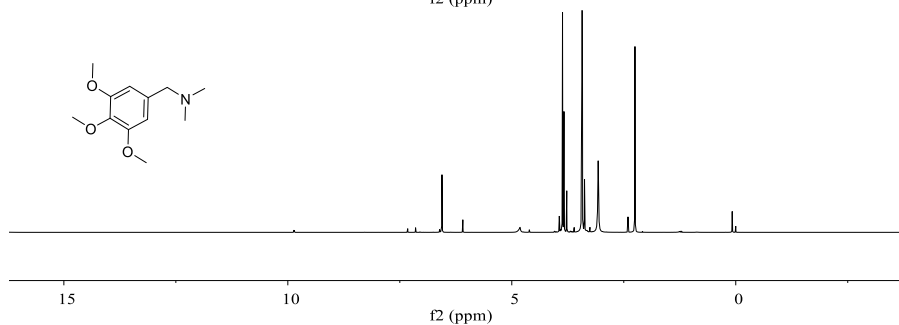
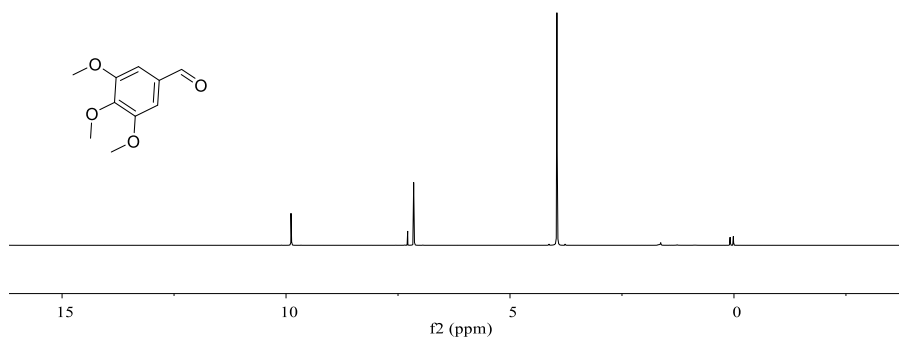


NL: 1.30E9
TIC: MS
A253

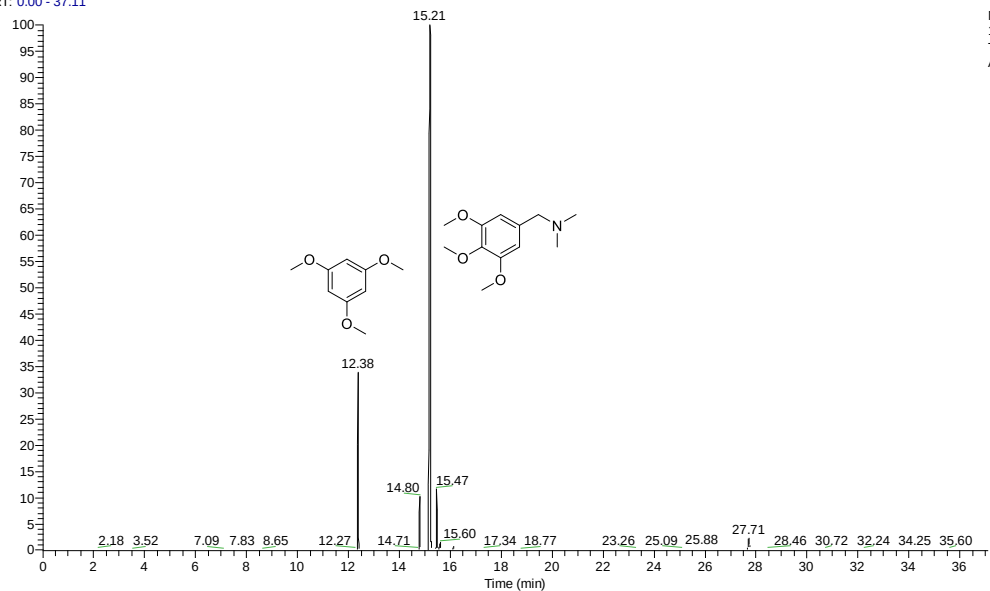
A253 #3309 RT: 13.25 AV: 1 NL: 2.90E8
T: + c EI Full ms [15.00-500.00]



A254

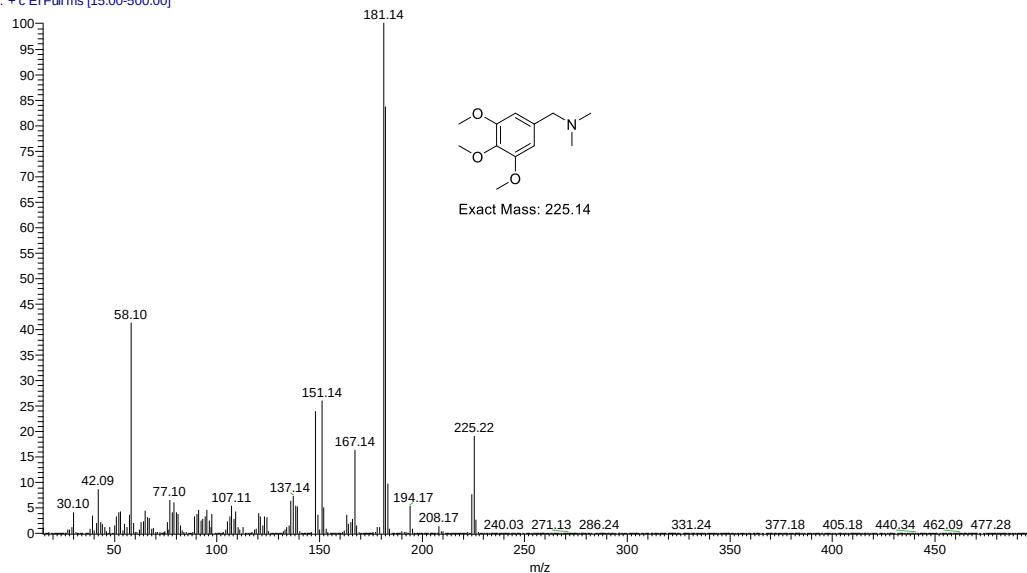


RT: 0.00 - 37.11

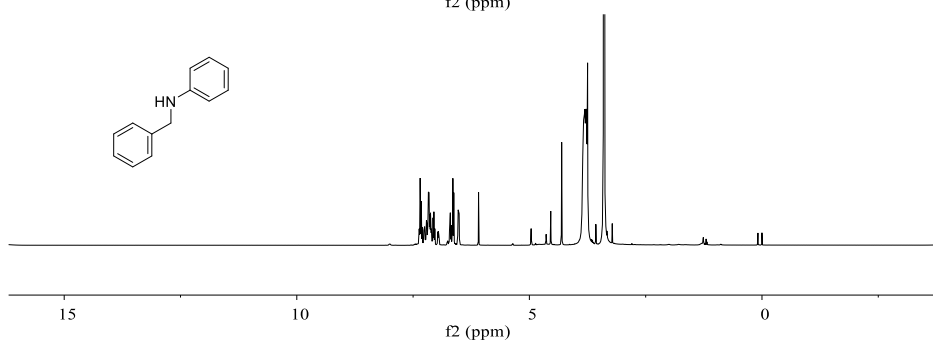
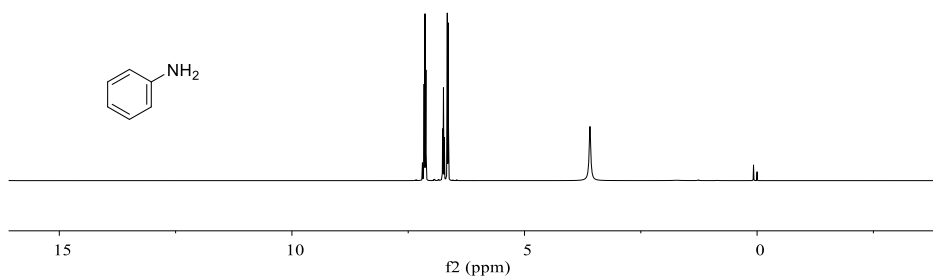
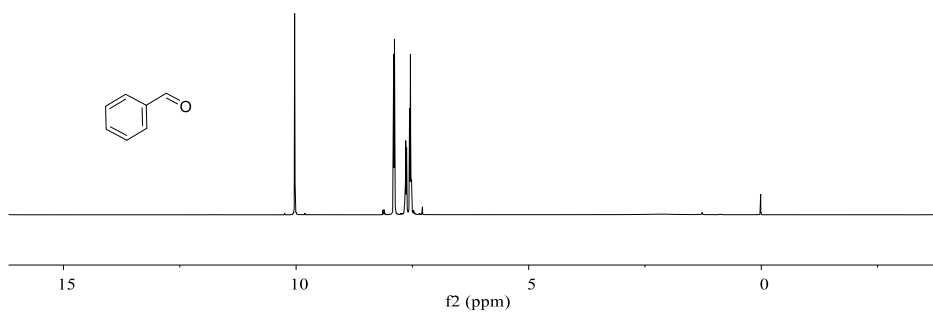


NL:
1.17E9
TIC MS
A254

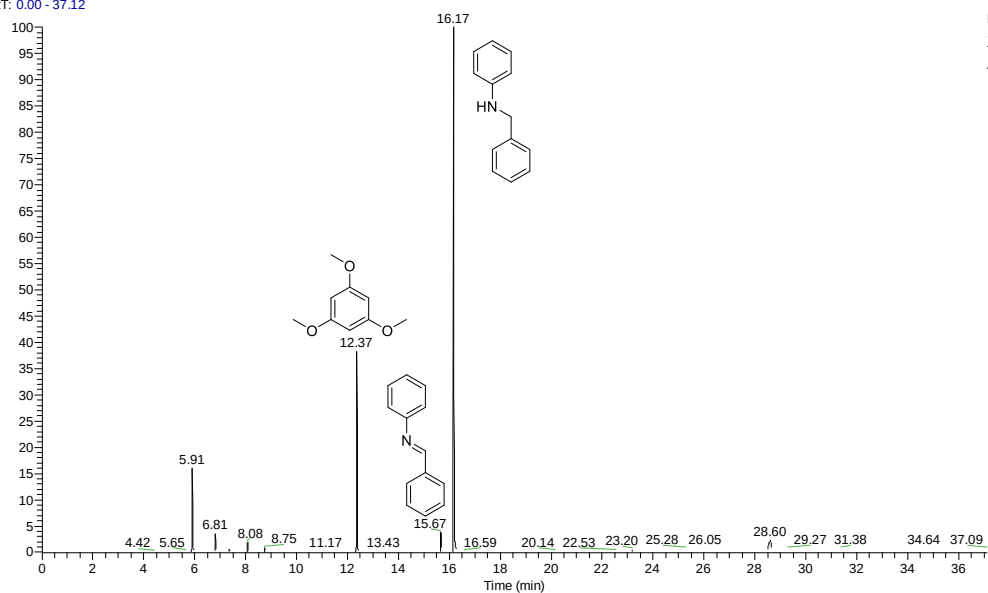
A254 #3885 RT: 15.21 AV: 1 NL: 2.03E8
T: + c EI Full ms [15.00-500.00]



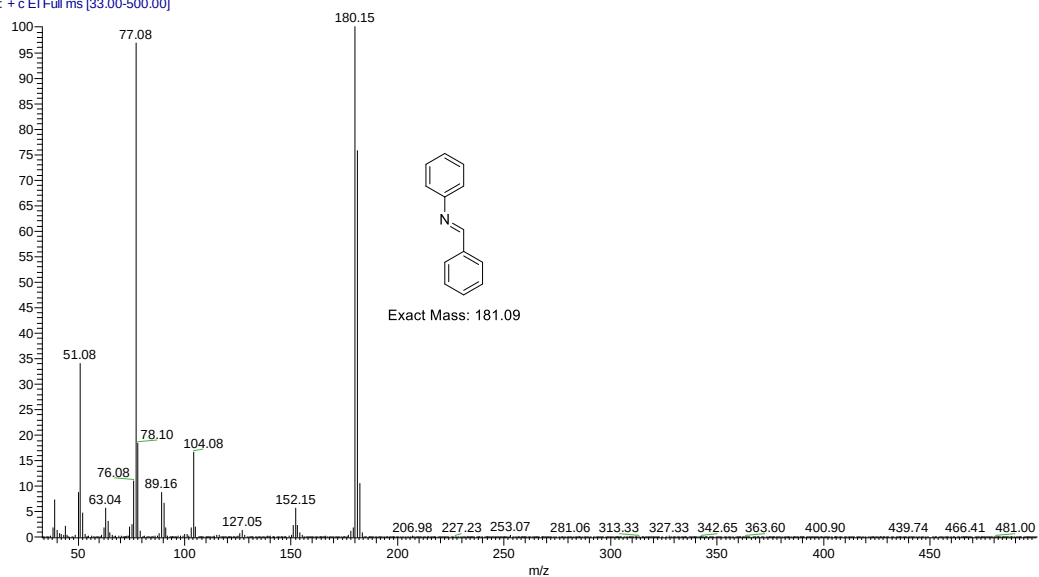
A196



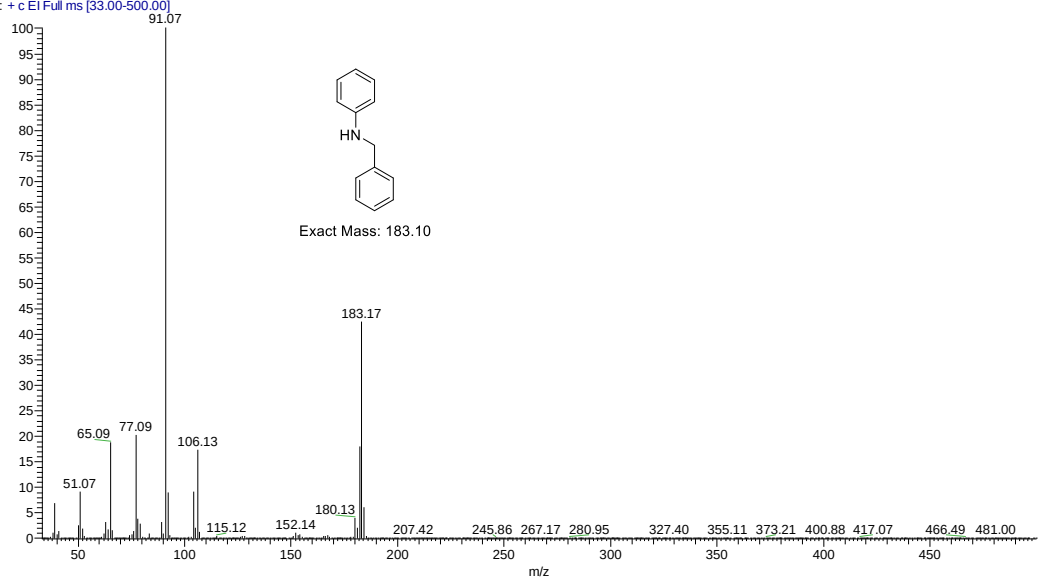
RT: 0.00 - 37.12



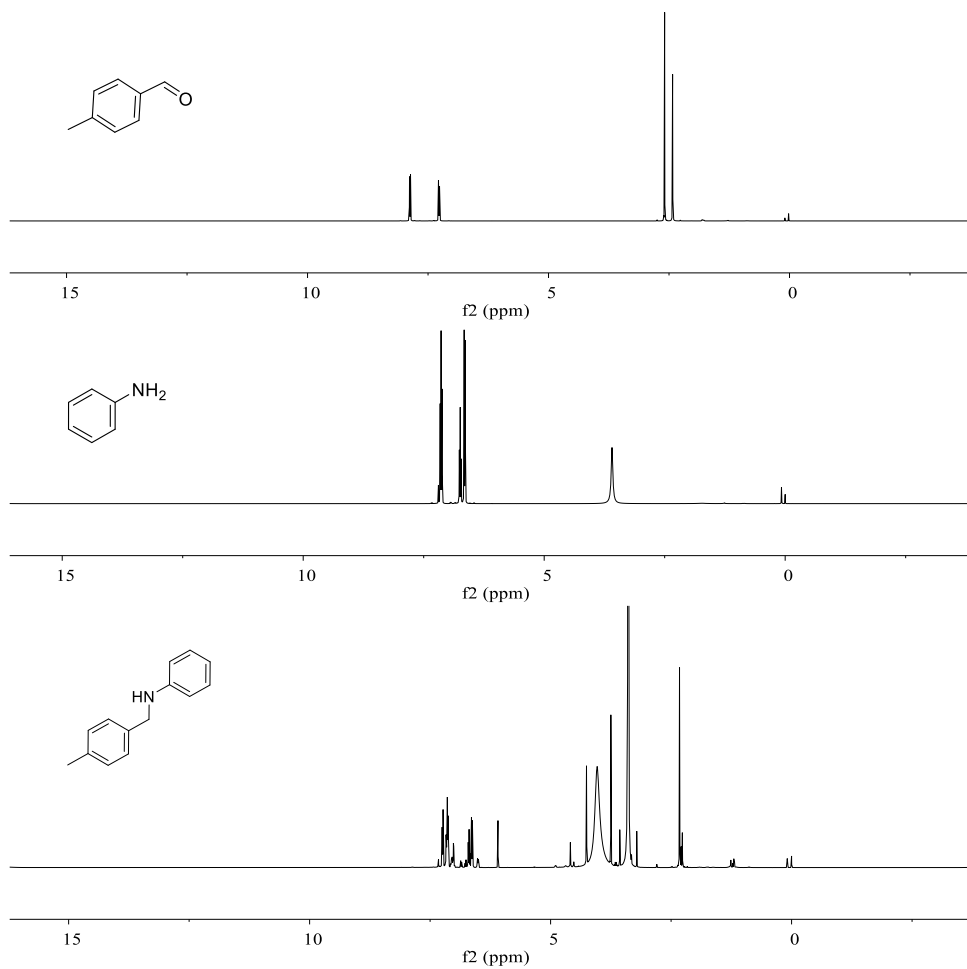
A196 #3431 RT: 15.67 AV: 1 NL: 1.22E6
T: + c EI Full ms [33.00-500.00]



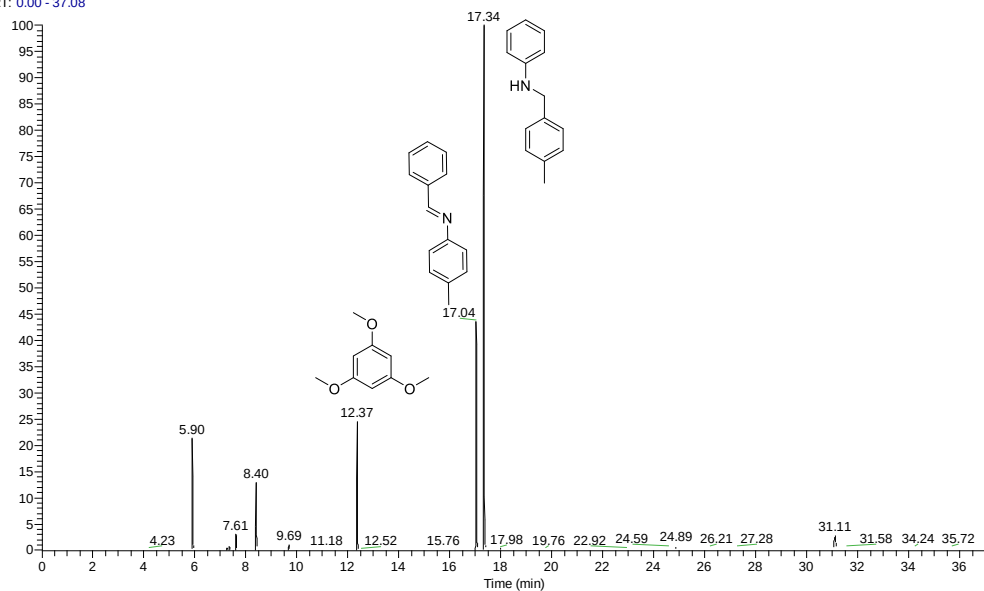
A196 #3578 RT: 16.17 AV: 1 NL: 4.39E7
T: + c EI Full ms [33.00-500.00]



A291 (A198 GC-MS)

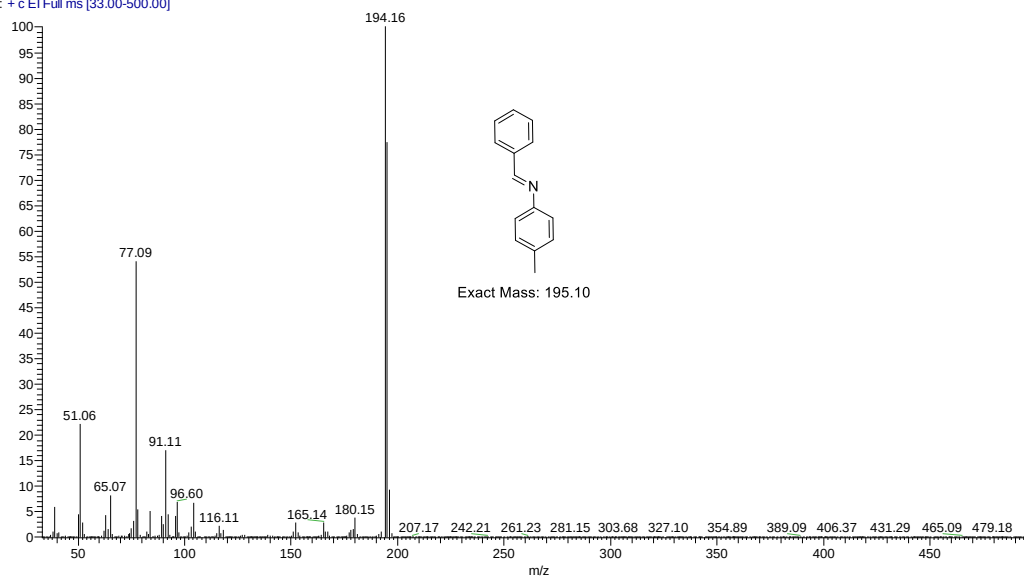


RT: 0.00 - 37.08

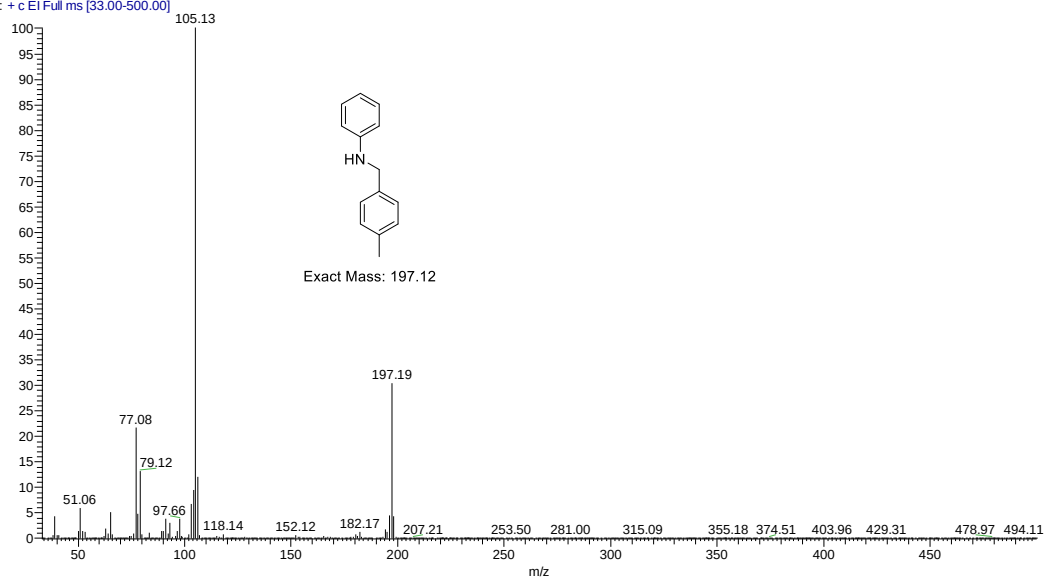


NL:
1.87E8
TIC MS
A198

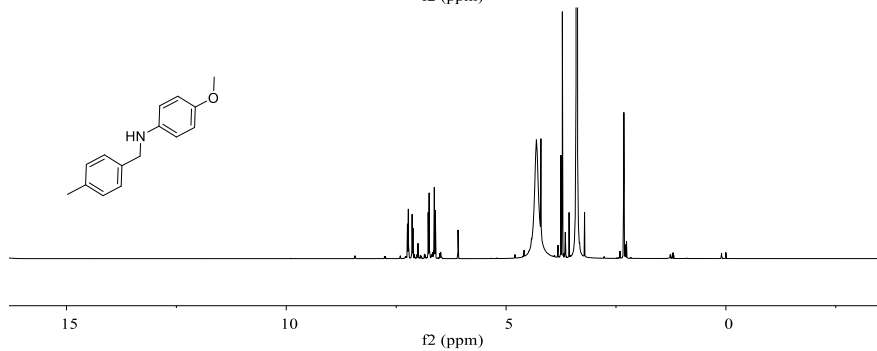
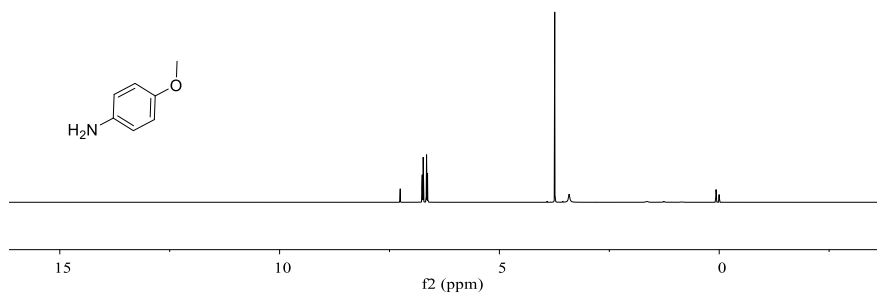
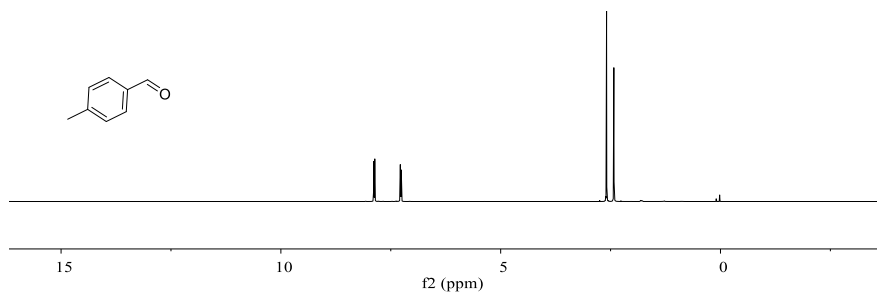
A198 #3836 RT: 17.04 AV: 1 NL: 1.56E7
T: + c EI Full ms [33.00-500.00]



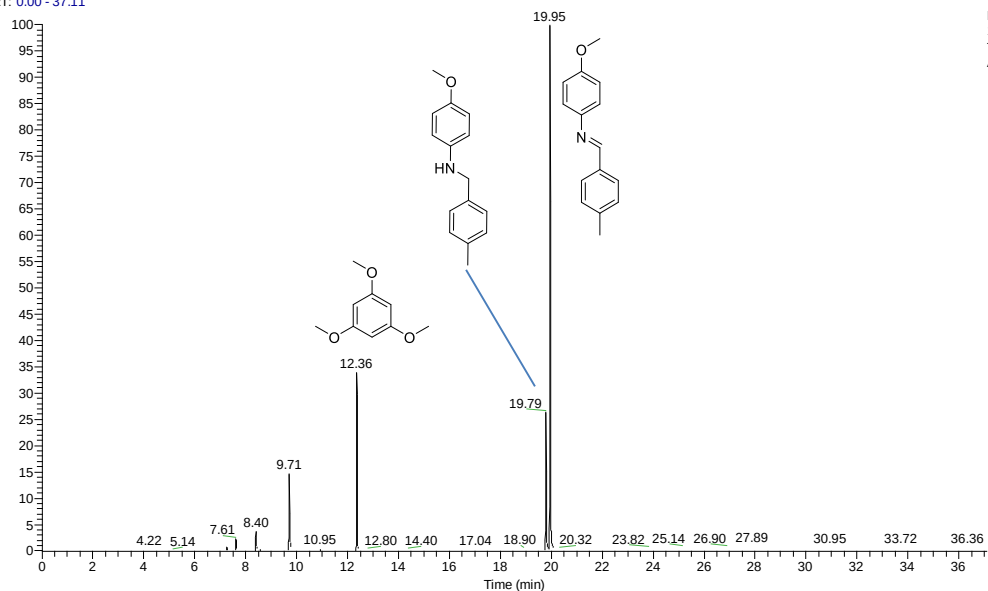
A198 #3923 RT: 17.34 AV: 1 NL: 7.00E7
T: + c EI Full ms [33.00-500.00]



A292(199 GC-MS)

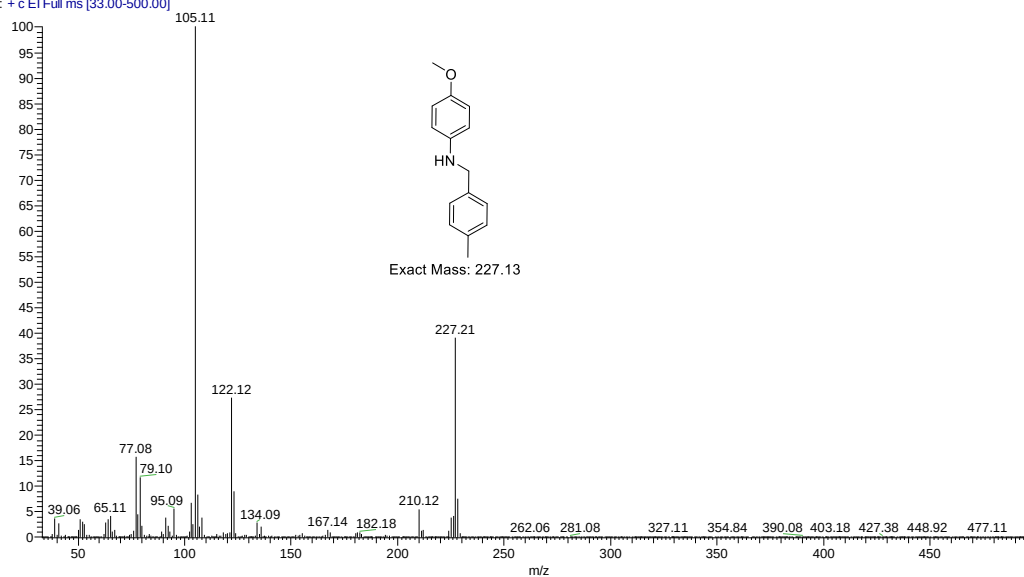


RT: 0.00 - 37.11

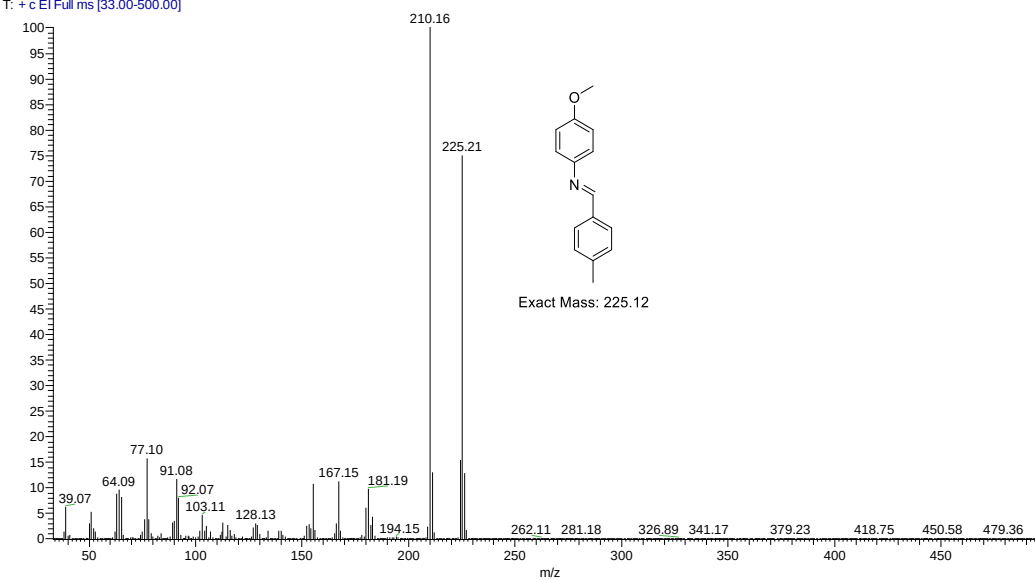


NL:
1.24E8
TIC MS
A199

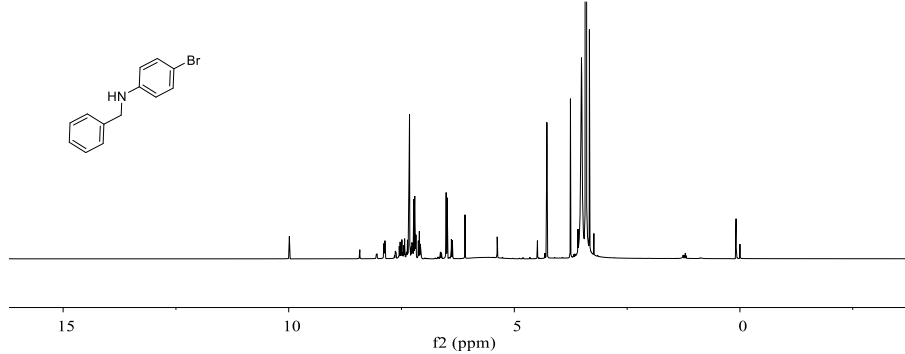
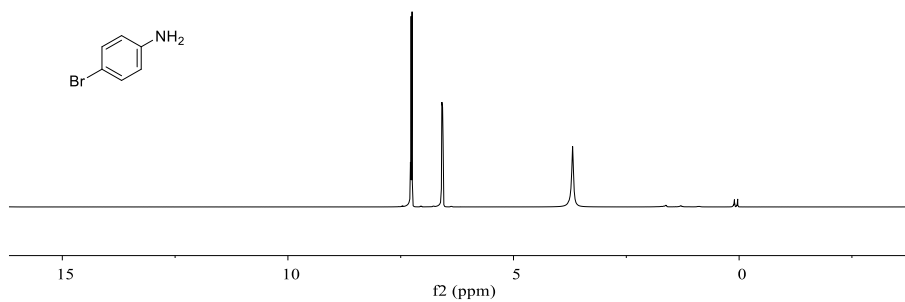
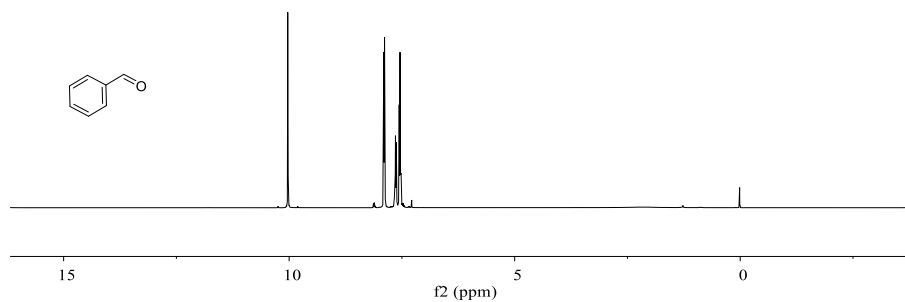
A199 #4644 RT: 19.79 AV: 1 NL: 8.65E6
T: + c EI Full ms [33.00-500.00]



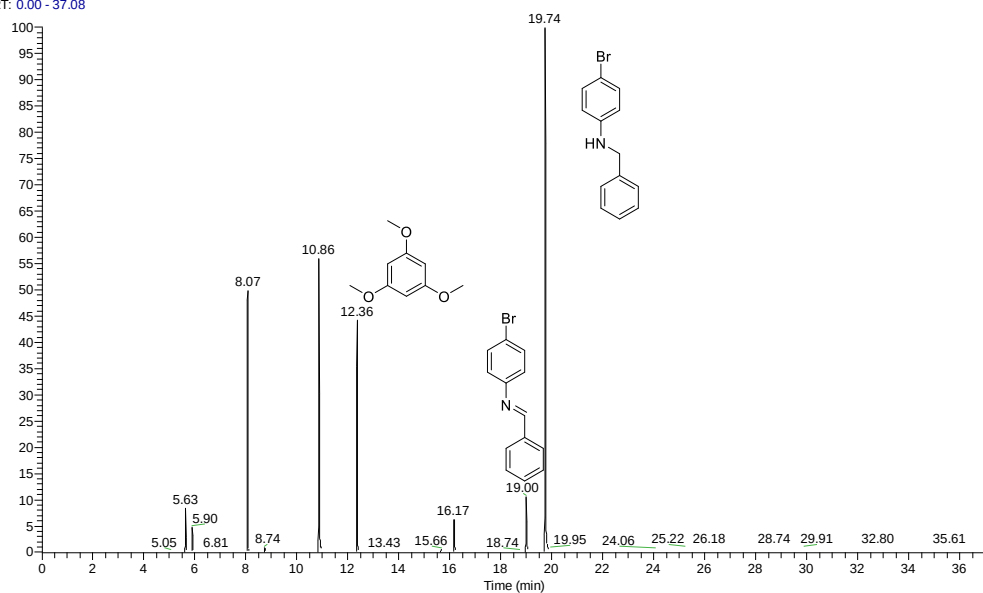
A199 #4681 RT: 19.95 AV: 1 NL: 2.83E7
T: + c EI Full ms [33.00-500.00]



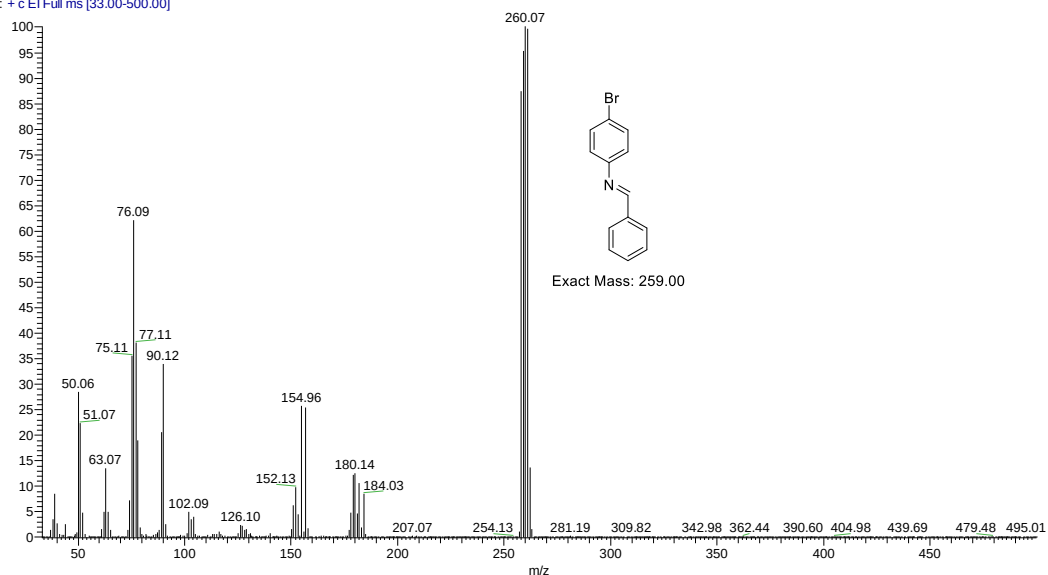
A244 (200 GC-MS)



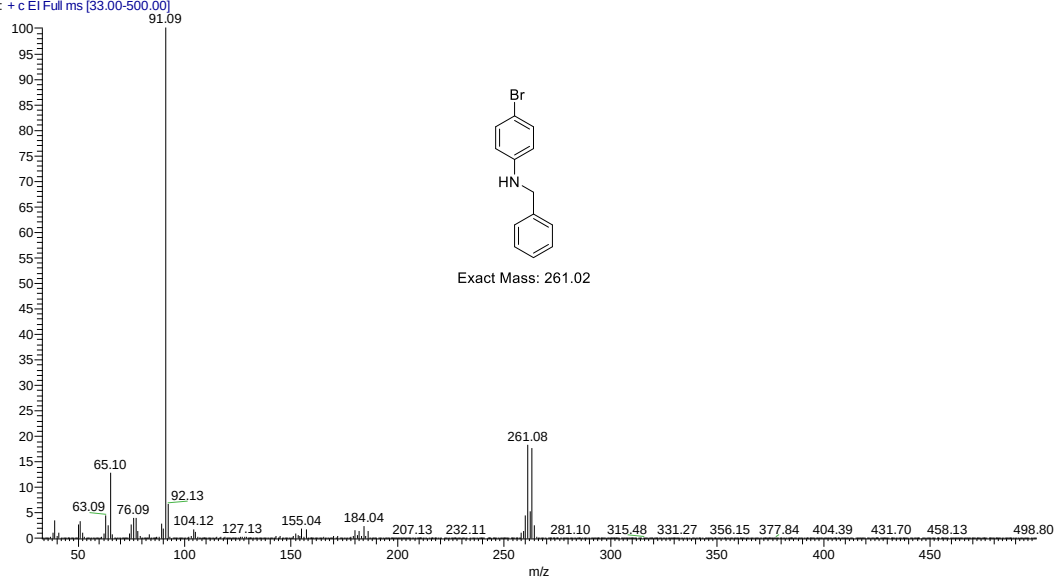
RT: 0.00 - 37.08



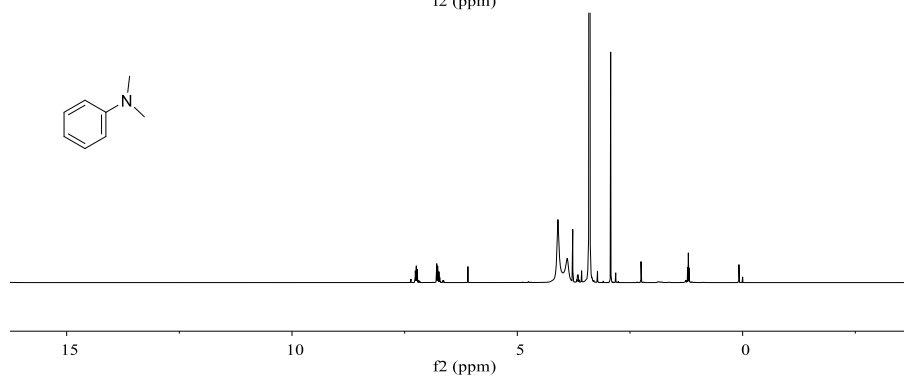
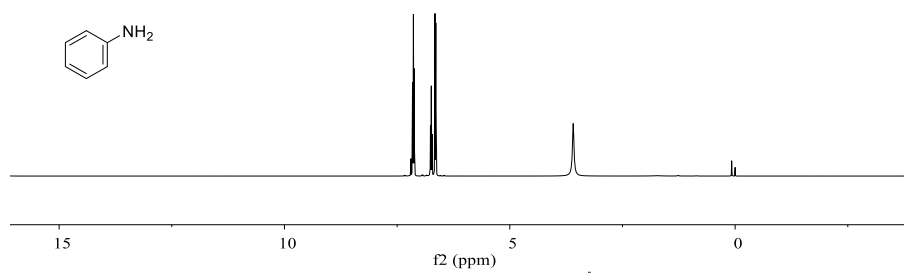
A200 #4410 RT: 19.00 AV: 1 NL: 1.21E6
T: + c EI Full ms [33.00-500.00]



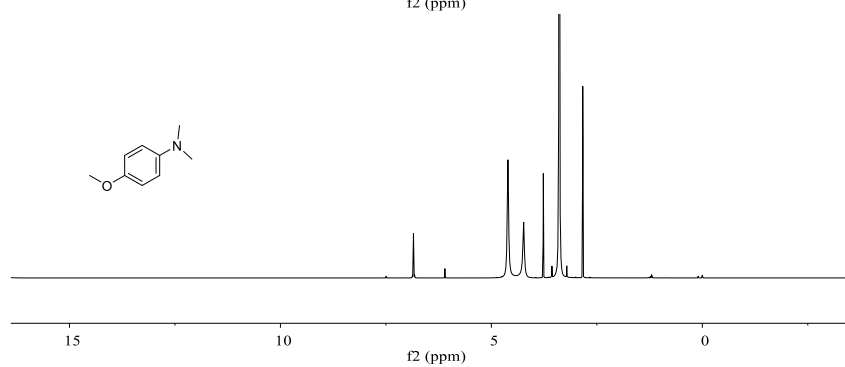
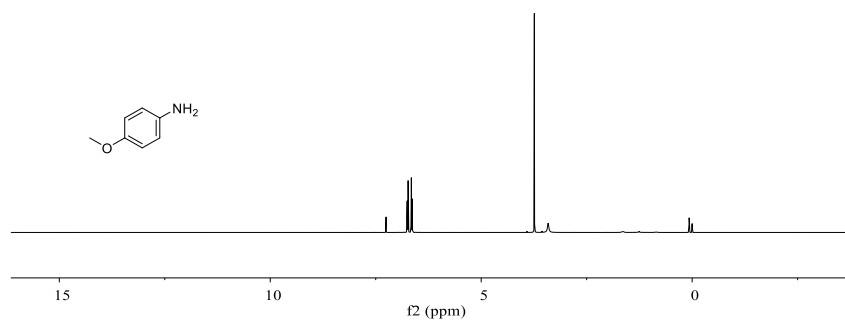
A200 #4630 RT: 19.74 AV: 1 NL: 4.65E7
T: + c EI Full ms [33.00-500.00]



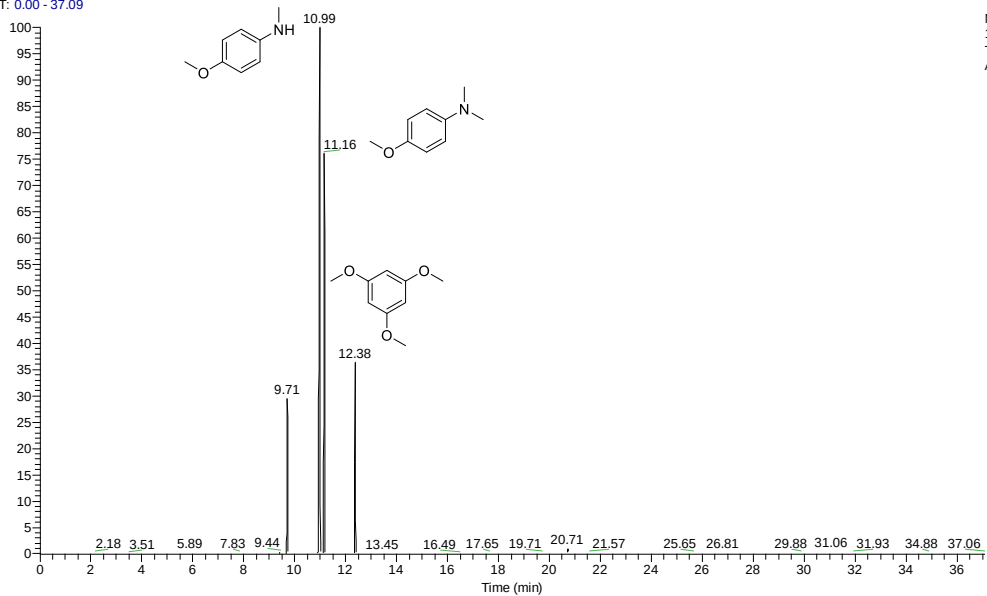
A335



A334 (277 GC-MS)

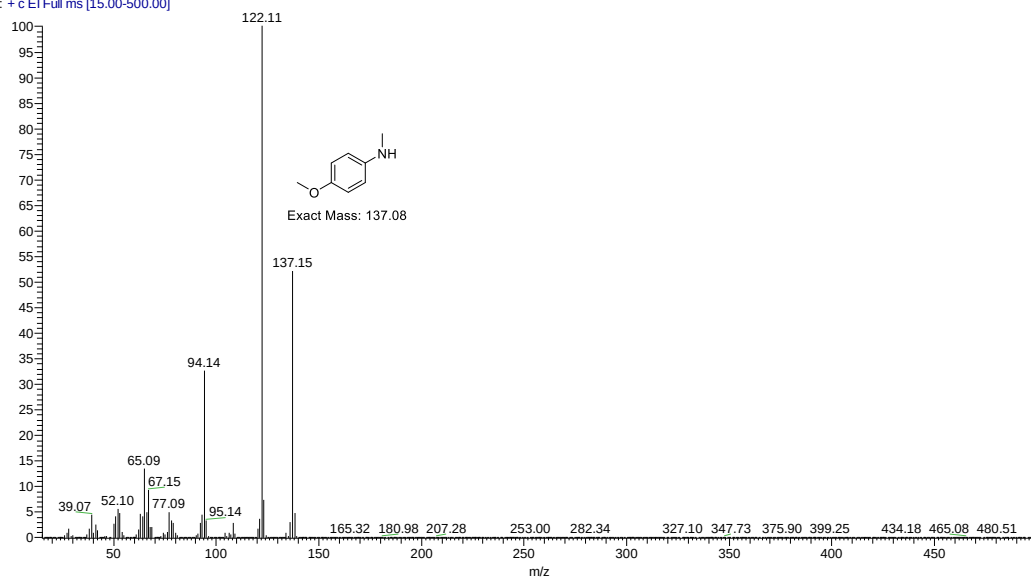


RT: 0.00 - 37.09

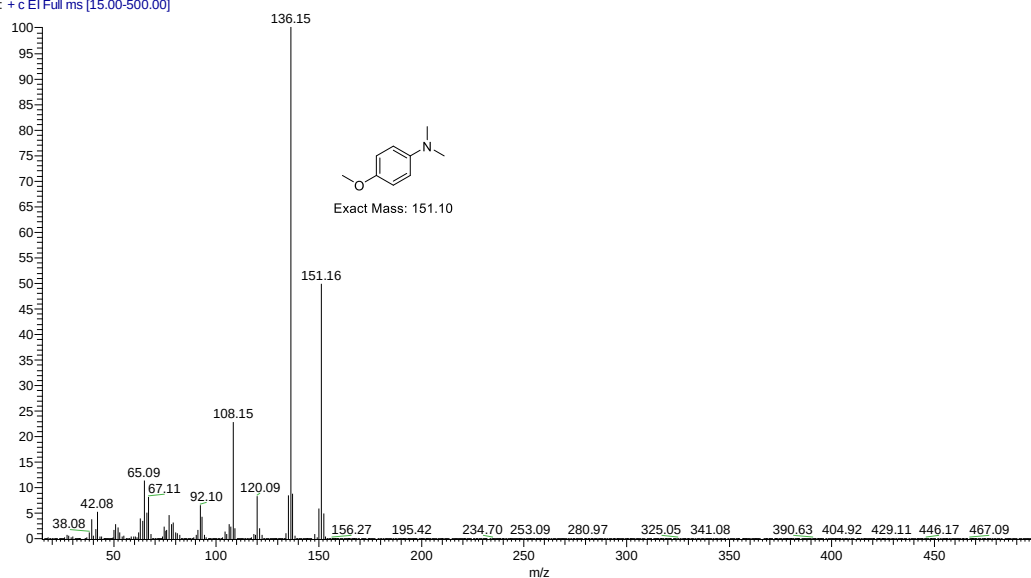


NL:
1.05E9
TIC: MS
A277

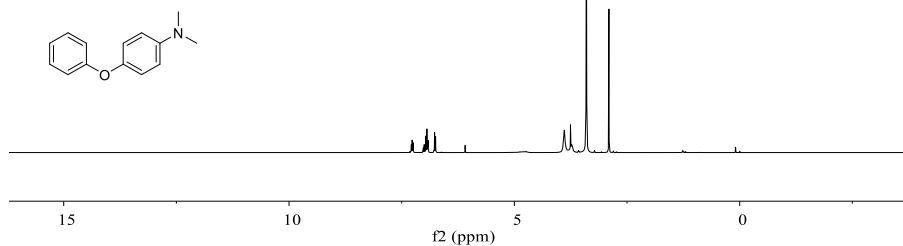
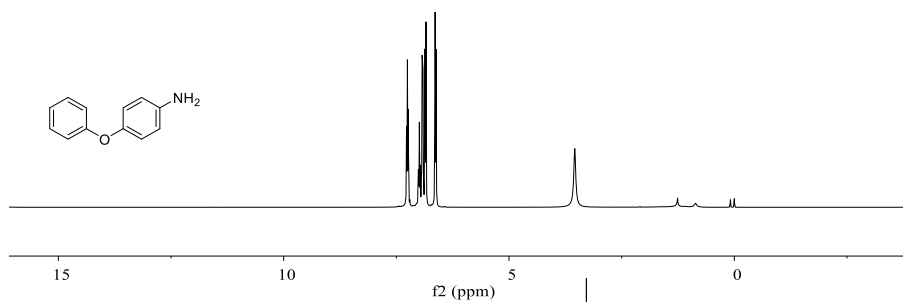
A277 #2645 RT: 10.99 AV: 1 NL: 3.27E8
T: + c EI Full ms [15.00-500.00]



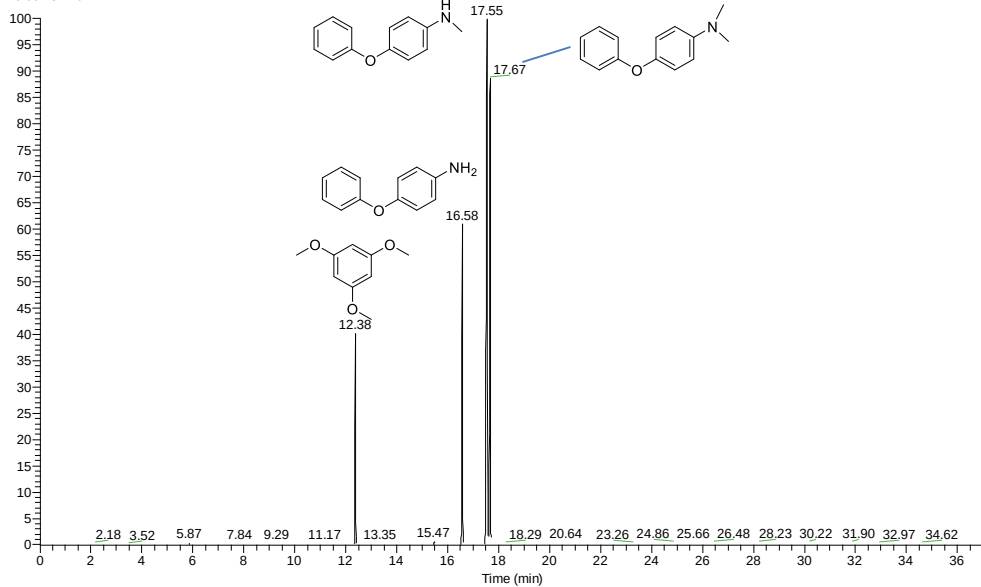
A277 #2694 RT: 11.16 AV: 1 NL: 2.35E8
T: + c EI Full ms [15.00-500.00]



A332 (A273 GC-MS)

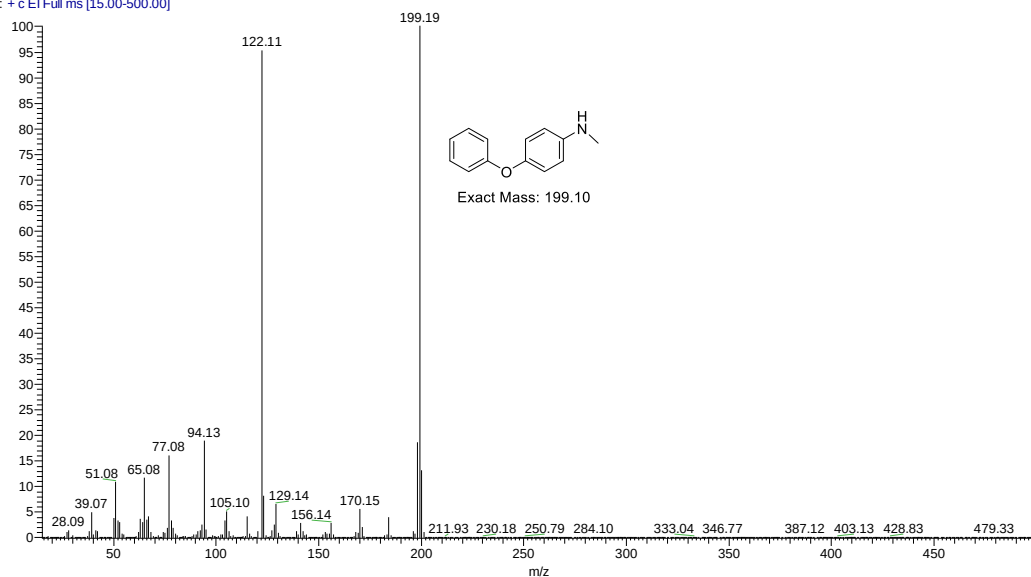


RT: 0.00 - 37.10

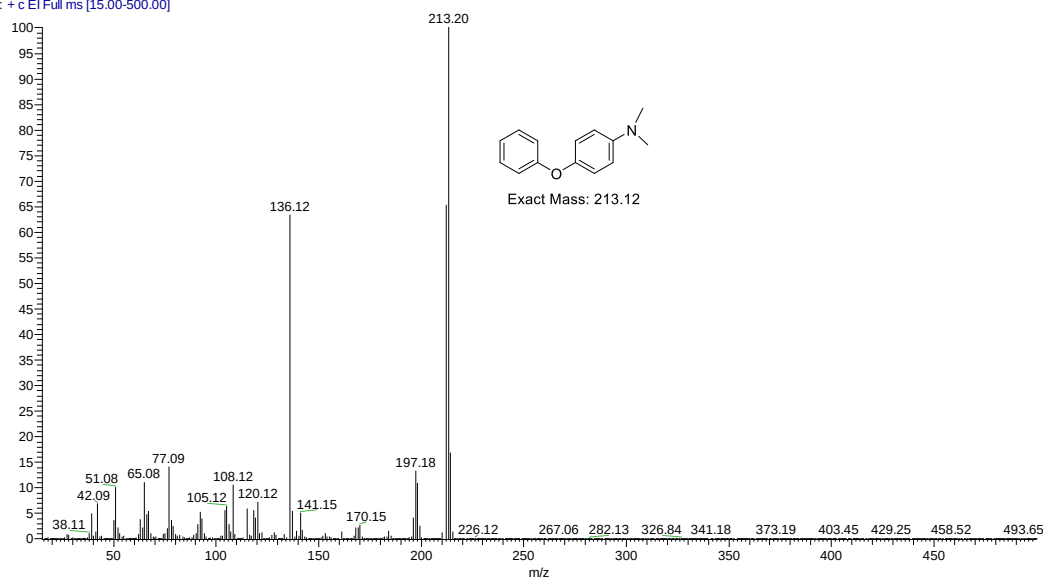


NL:
1.00E9
TIC: MS
A273

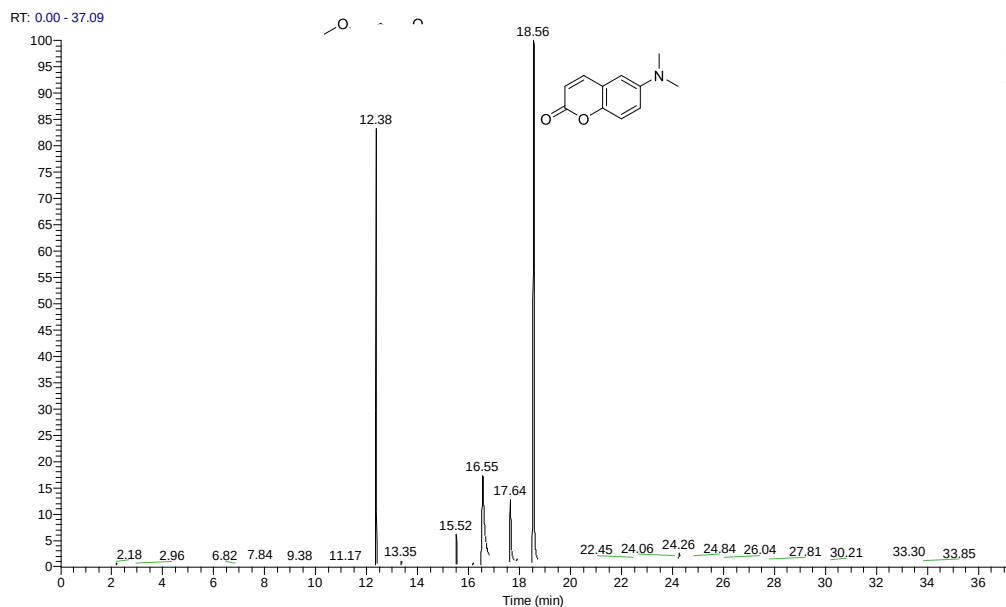
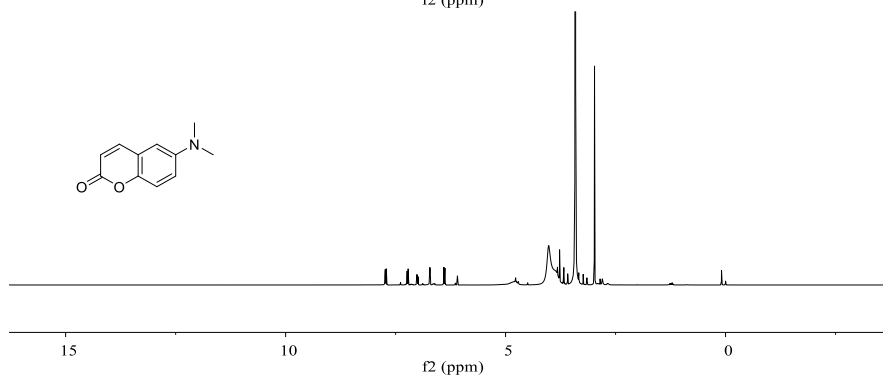
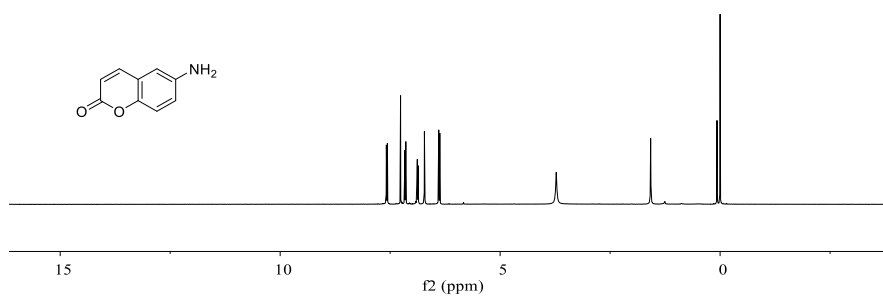
A273 #4573 RT: 17.55 AV: 1 NL: 2.36E8
T: + c EI Full ms [15.00-500.00]



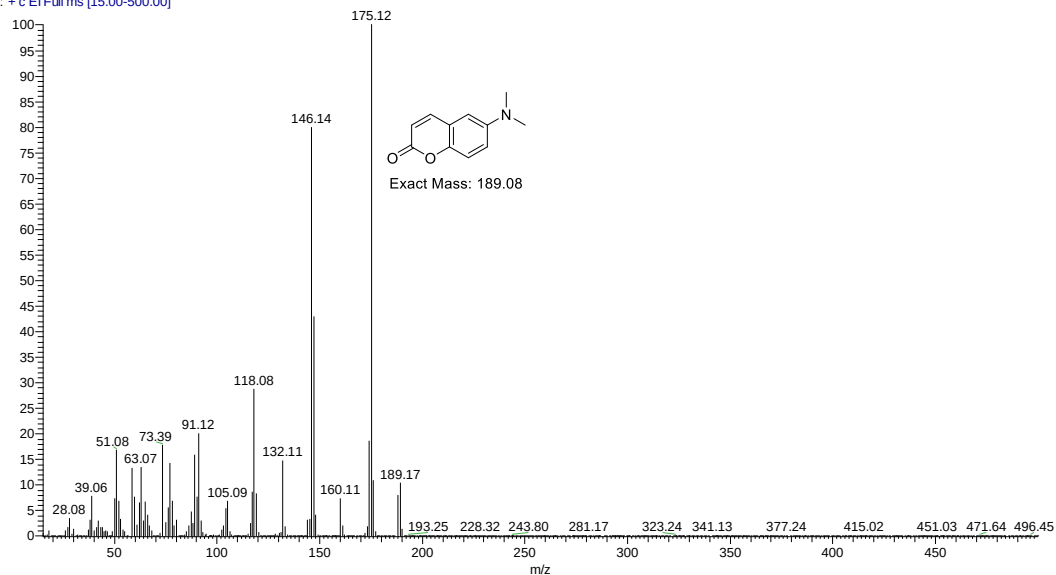
A273 #4607 RT: 17.67 AV: 1 NL: 1.84E8
T: + c EI Full ms [15.00-500.00]



A337 (A272 GC-MS)



A272 #4871 RT: 18.56 AV: 1 NL: 5.74E7
T: + c EI Full ms [15.00-500.00]



A333

