

Supporting information

One-pot compositional and structural regeneration of degraded LiCoO_2 for directly reusing as a high-performance lithium-ion battery cathode

Juan Yang^a, Wenyu Wang^a, Huimeng Yang^a and Dihua Wang^{*a}

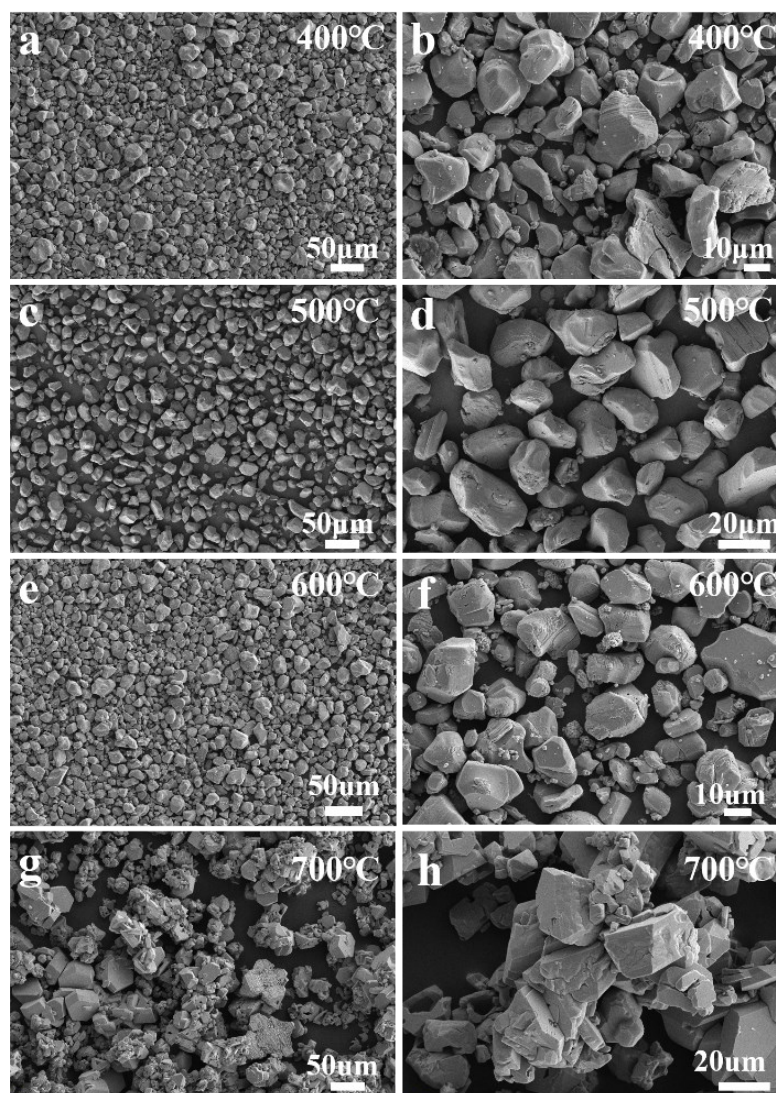


Fig. S1 SEM images of LiCoO_2 regenerated in $\text{LiOH-KOH-Li}_2\text{CO}_3$ molten salt at 400 °C (a-b), 500 °C (c-d), 600 °C (e-f) and 700 °C (g-h), respectively.

^a School of Resource and Environmental Science, International Cooperation Base for Sustainable Utilization of Resources and Energy in Hubei Province, Wuhan University, Wuhan 430072, China. E-mail: wangdh@whu.edu.cn

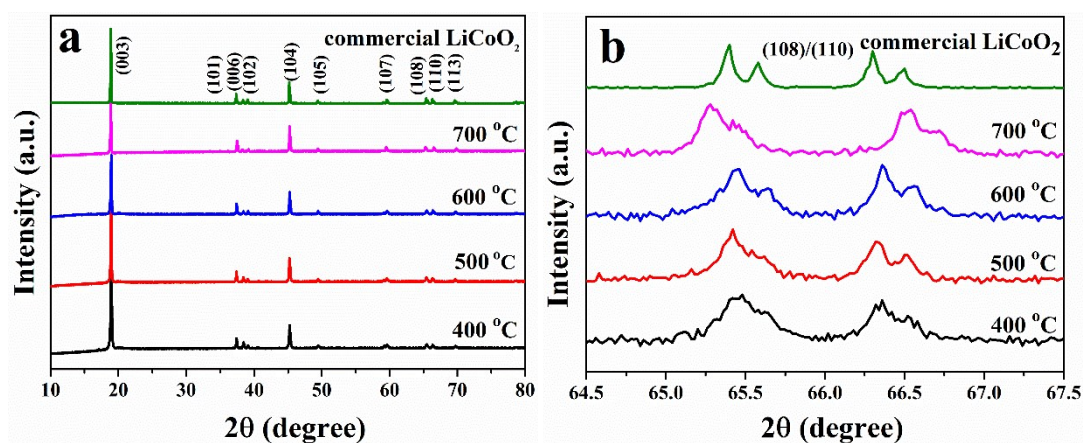


Fig. S2 (a) the XRD patterns of commercial LiCoO_2 and regenerated LiCoO_2 prepared at various temperatures; (b) The enlarged view of all (108)/(110) reflections in (a).

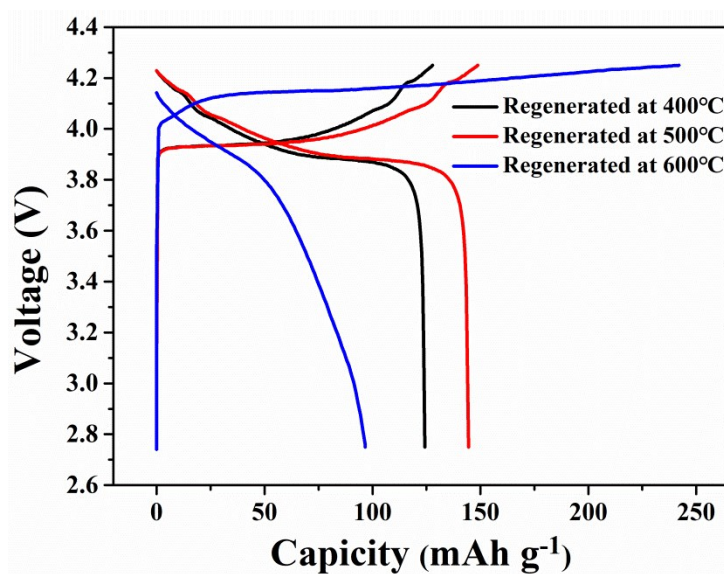


Fig. S3 Voltage–capacity profiles of different regenerated LiCoO_2 materials at 0.2 C.

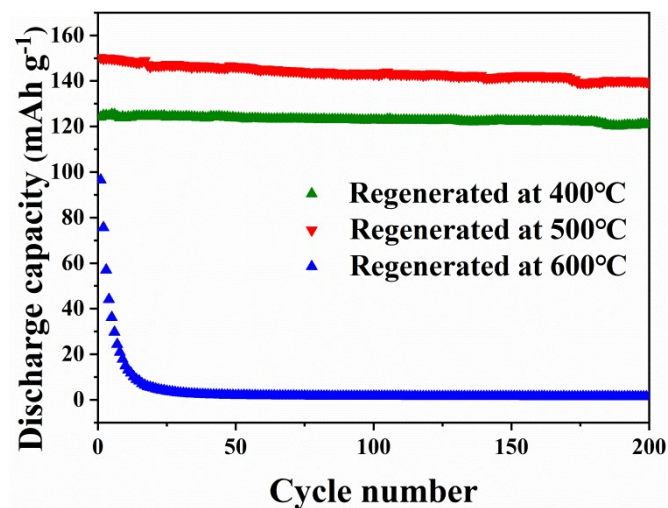


Fig. S4 Cycling performance of different regenerated LiCoO₂ materials at 0.2 C.

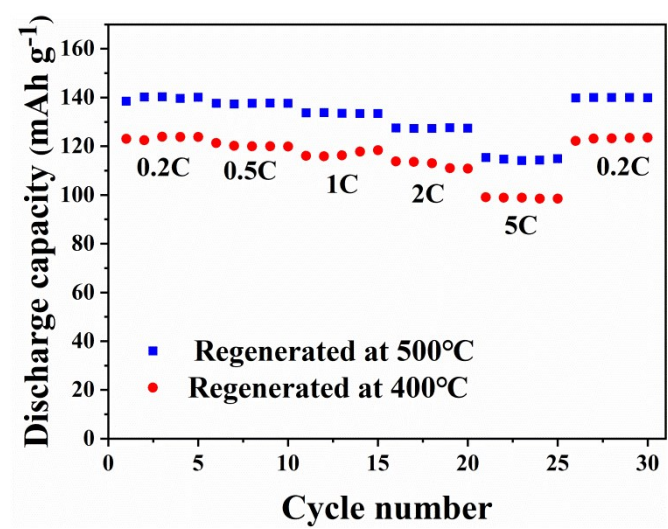


Fig. S5 Discharge specific capacity performance of different regenerated LiCoO₂ materials at different rate.