

## The elemental analysis and multi-nuclear NMR study of an alkali molten salt used to digest reference and commercial SWCNT powders

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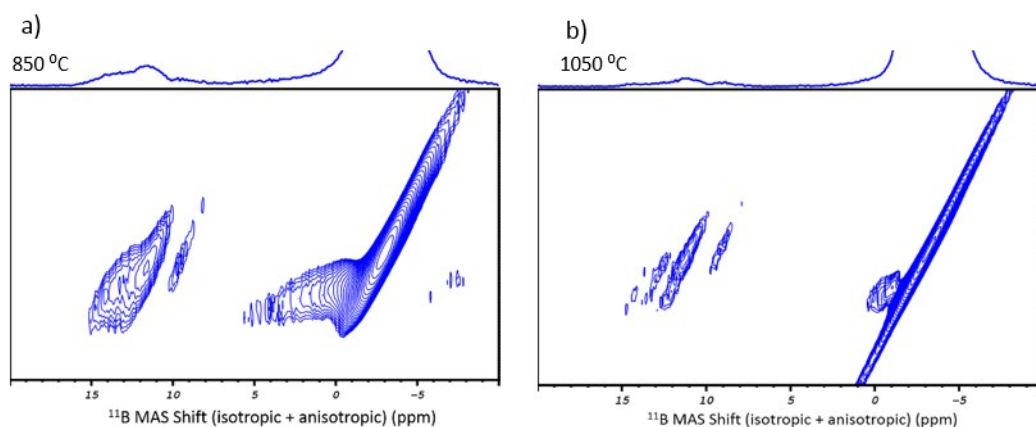
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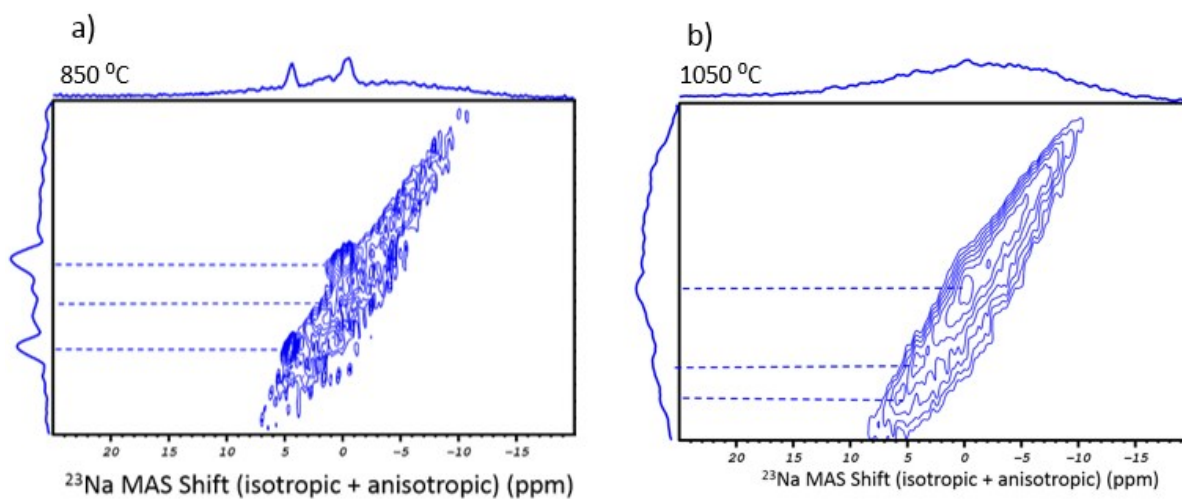
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**Table S1.** Mass fractions of selected elements obtained by INAA for SWCNT-5848V (test portion mass 30 mg, average and standard deviation for N = 5).

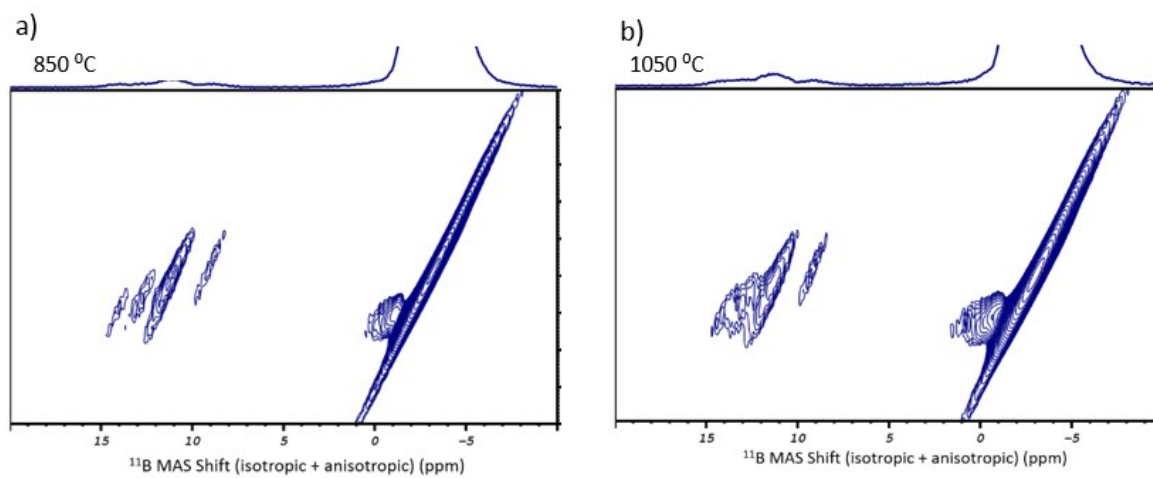
Element	avg $\pm$ std dev (mg kg <sup>-1</sup> )
Cr	604 $\pm$ 34
Fe	126700 $\pm$ 2400
Mn	43 $\pm$ 4
Ni	1110 $\pm$ 70



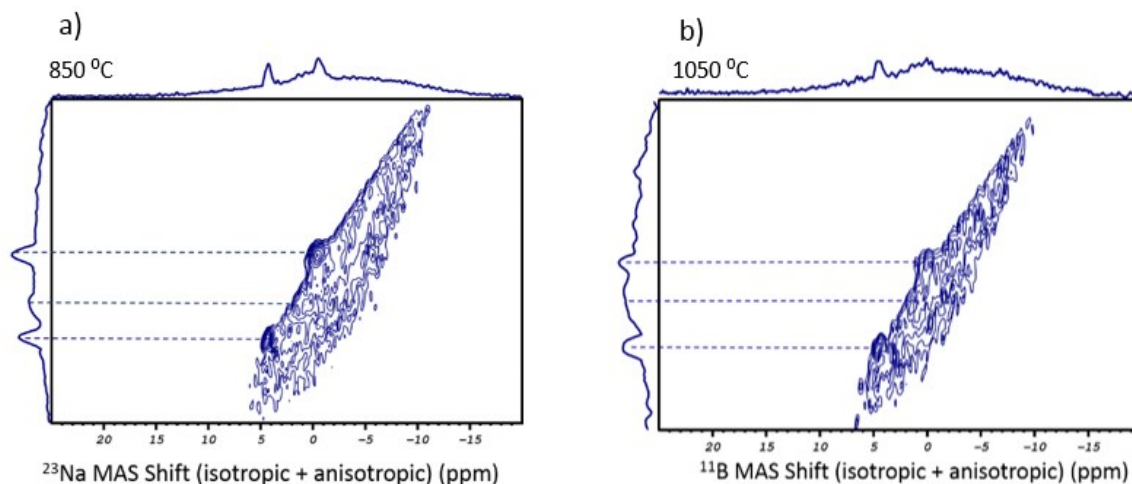
**Figure S1.** <sup>11</sup>B MAS NMR of SRM 2483 at a) 850 °C and b) 1050 °C.



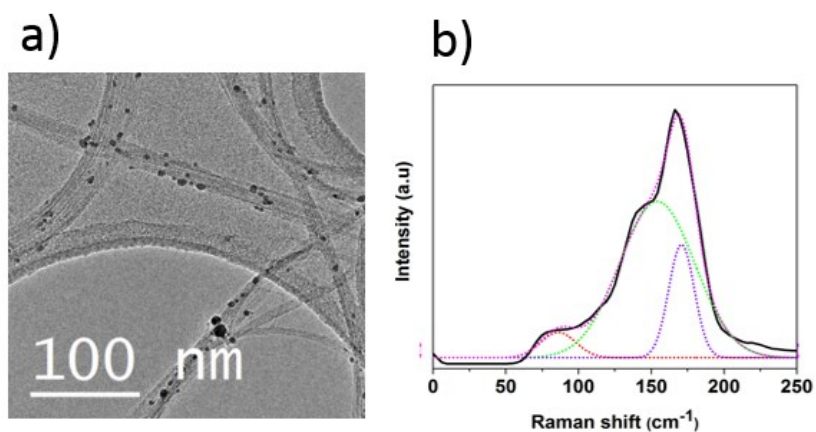
**Figure S2.**  $^{23}\text{Na}$  MAS NMR of SRM 2483 at **a)** 850 °C and **b)** 1050 °C.



**Figure S3.**  $^{11}\text{B}$  MAS NMR of SWCNT-7867V at **a)** 850 °C and **b)** 1050 °C.



**Figure S4.**  $^{23}\text{Na}$  MAS NMR of SWCNT-7867V at **a)** 850 °C and **b)** 1050 °C.



**Figure S5. a)** TEM micrograph (low mag) and **b)** Raman spectrum of the SWCNT-5848V with deconvolution of the RBM band, resulting on an average diameter of  $1.7 \pm 0.5$  nm. The diameter was calculated according to the formula (where  $\omega$  RBM is the RBM frequency,  $c_1$  and  $c_2$  are constants and  $d$ , the nanotube diameter) <sup>1</sup>:

$$\omega (RBM) = \frac{c_1}{d} + c_2$$

## References

1. Decker, J. E.; Hight Walker, A. R.; Bosnick, K.; Clifford, C. A.; Dai, L.; Fagan, J.; Hooker, S.; Jakubek, Z. J.; Kingston, C.; Makar, J.; Mansfield, E.; Postek, M. T.; Simard, B.; Sturgeon, R.; Wise, S.; Vladar, A. E.; Yang,

L.; Zeisler, R., Sample preparation protocols for realization of reproducible characterization of single-wall carbon nanotubes. *Metrologia* **2009**, *46* (6), 682-692.