

XAS study of Sn speciation in toothpaste

Supplementary Information

Morgane Desmau,[†] Marco A. Alsina,[‡] and Jean-François Gaillard^{*,†}

†Department of Civil and Environmental Engineering, Northwestern University, 2145

Sheridan Road, Evanston, IL, 60208-3109, USA

‡Department of Engineering and Construction Management, University of Talca, Camino a

Los Niches Km 1, Curicó 3340000, Chile

E-mail: jf-gaillard@northwestern.edu

Sn(II)-based compounds	Sn(IV)-based compounds
$\boxed{\text{SnF}_2 \text{ (solid/aq)}}$ $\text{SnF}_2 + \text{Carrageenan}$ $\boxed{\text{SnF}_2 + \text{Na-Gluconate}^*}$ $\text{SnF}_2 + \text{Glycerol}^*$ $\text{SnCl}_2 + \text{Na-Acetate}$ $\text{SnCl}_2 + \text{NaCl}$ $\text{SnCl}_2 + \text{Na-Gluconate}$ $\text{SnCl}_2 + \text{NaF}$ $\text{SnF}_2 + \text{NaF}$ $\text{SnF}_2 + \text{Mica}$ $\text{SnF}_2 + \text{NaF}^*$ $\text{SnF}_2 + \text{Na}_3\text{PO}_4$ $\text{SnF}_2 + \text{Polyphosphate}$ $\text{SnF}_2 + \text{TiO}_2$ $\text{SnF}_2 + \text{Zn-Lactate}$ SnO $\boxed{\text{Sn}_2\text{P}_2\text{O}_7}$ $\text{SnF}_2 + \text{Na}_2\text{HPO}_4^*$ $\text{SnF}_2 + \text{Na}_4\text{P}_2\text{O}_7$	SnO_2 $\boxed{\text{SnCl}_4, 5\text{H}_2\text{O} + \text{NaOH}^*}$

* = various ratio analyzed
█ = select references (XANES)
█ = select references (EXAFS)

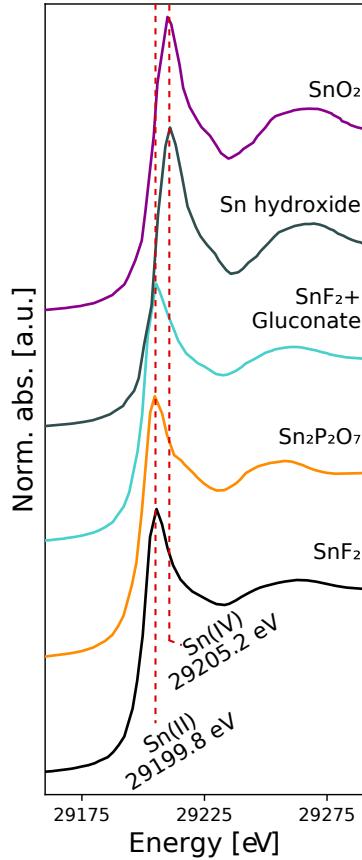


Figure S1: Sn(II)- and Sn(IV)-based compounds measured and analyzed as references for Sn K-edge XANES and EXAFS data analysis. XANES spectra of some of those compounds is also presented, with the mean whiteline values for Sn(II) and Sn(IV) compounds.

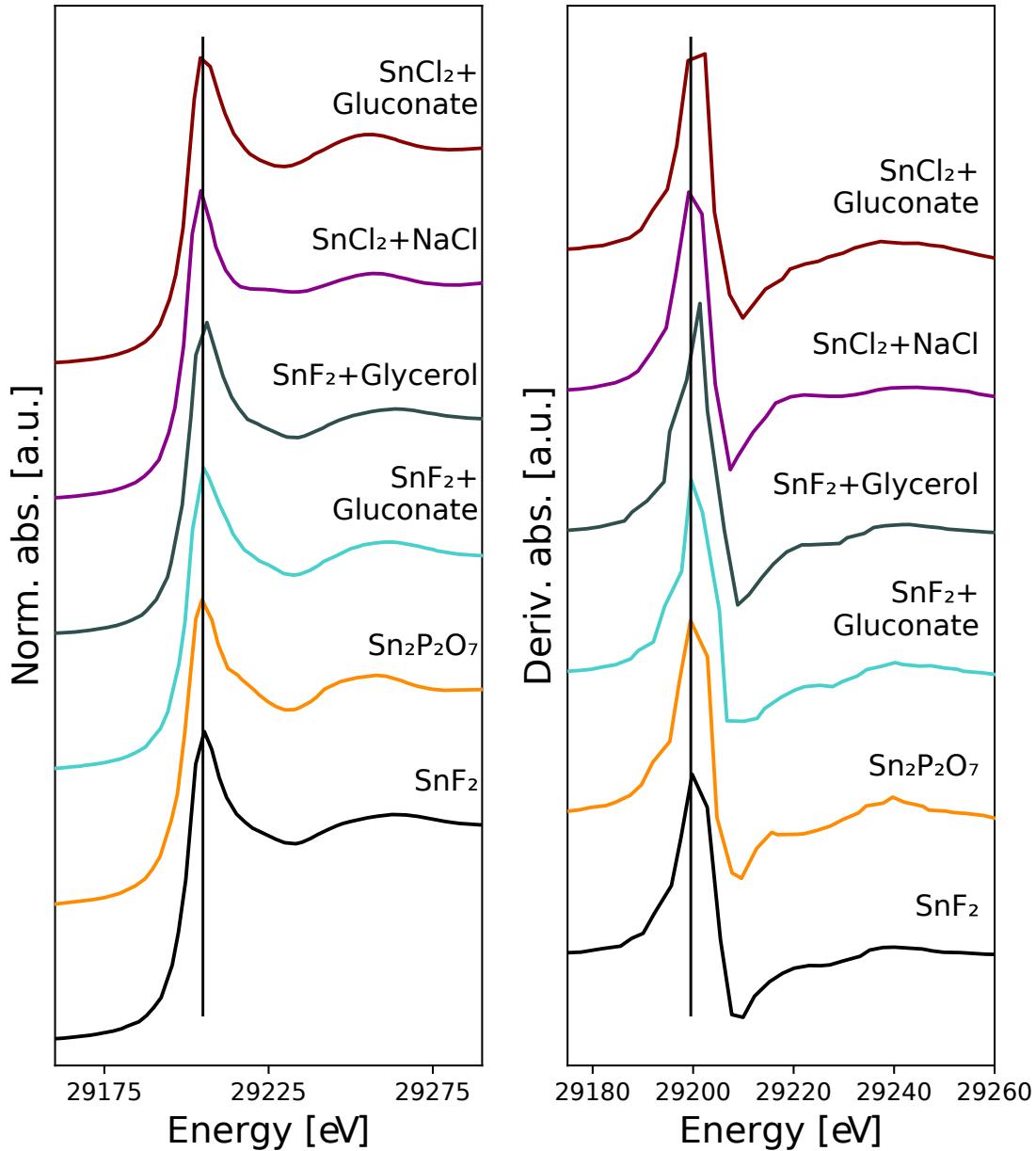


Figure S2: Sn K-edge XANES and First Derivate XANES spectra of Sn(II)-based compounds.

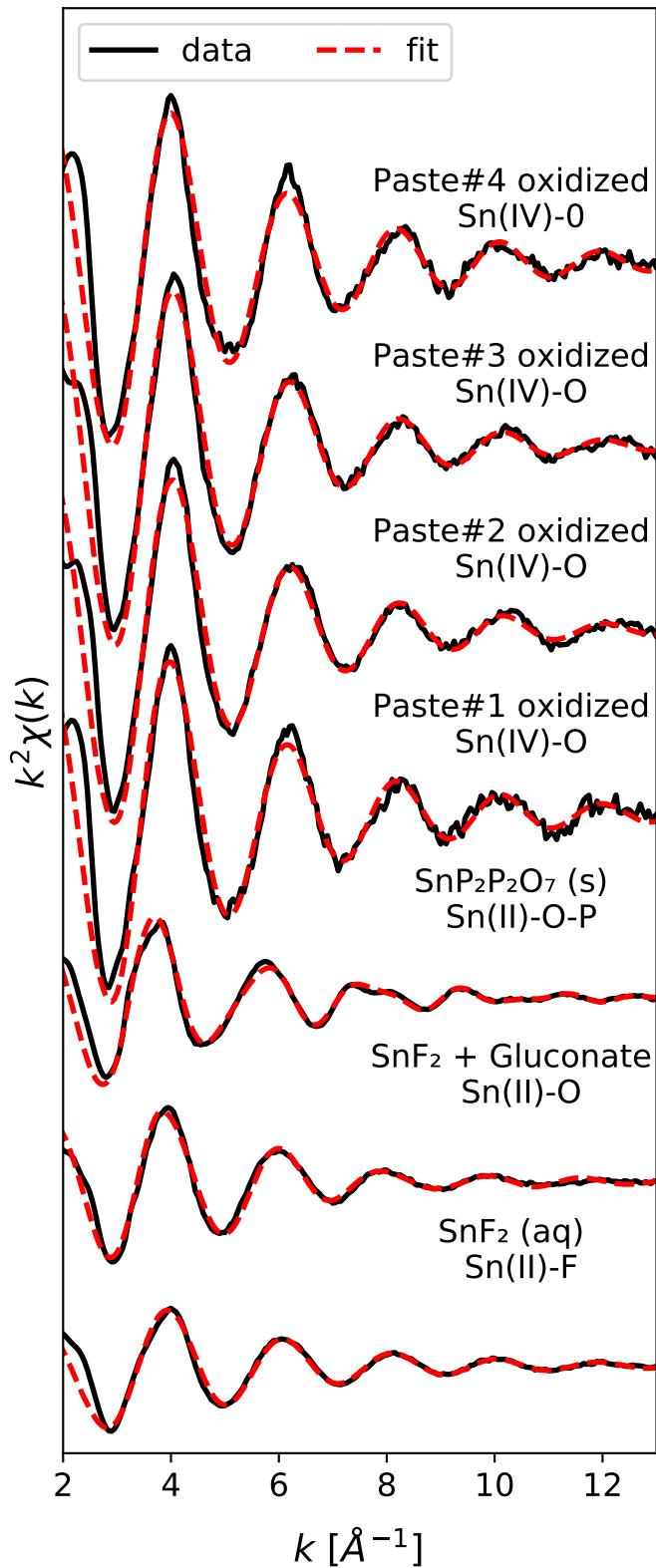


Figure S3: Results of least-squares fitting (LSF) analysis on Sn K-edge EXAFS spectra reference compounds.

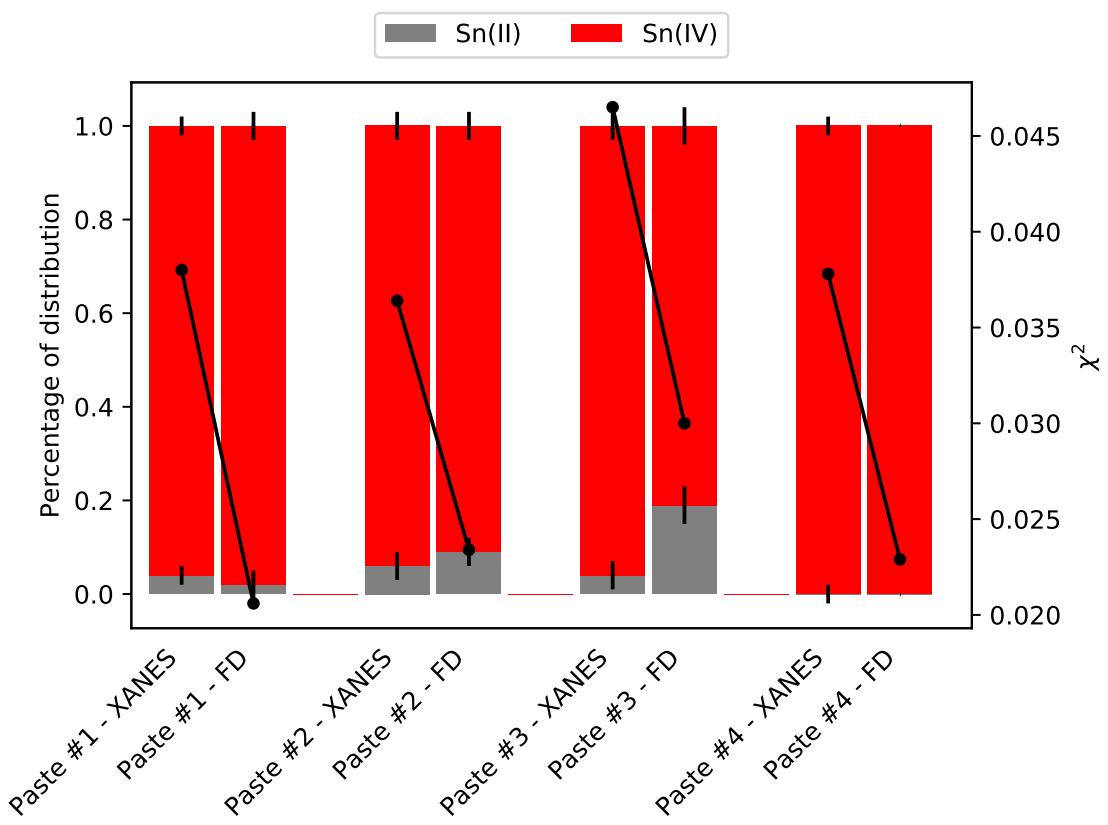


Figure S4: Results of Linear Combination Fit for the pastes oxidized, obtained by fitting the XANES and First-Derivative (FD) experimental data.

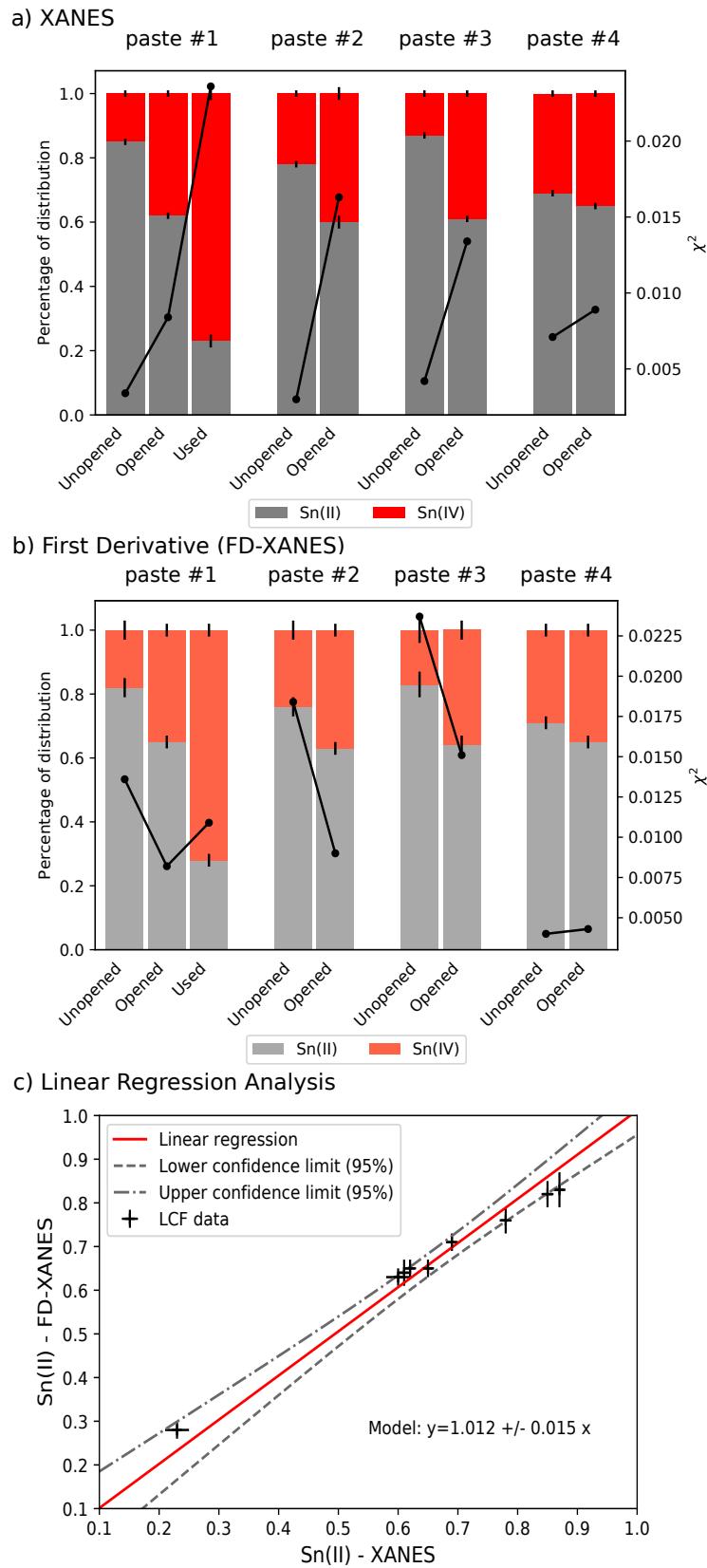


Figure S5: Results of Linear Combination Fit for XANES and XANES-FD spectra as well as the correlation between the results.

Table S1: Chemical Suppliers of the compounds used in this study

Compound	Supplier	Purity (%)	Origin
SnF ₂	Aldrich	99	USA
SnCl ₂ , 2H ₂ O	Alfa Aesar	99	USA
SnCl ₄ , 5H ₂ O	Acros	98+	India
Sn ₂ P ₂ O ₇	Aldrich	98	USA
SnO ₂	Alfa Aesar	99.9	USA
SnO	Aldrich	97	USA
Na – Acetate	Merck	99+	Germany
NaCl	Fisher Chemicals	99.9	USA
Na – Gluconate	Sigma	99	France
NaF	Fisher Scientific	99	USA
Na ₃ PO ₄ , 5H ₂ O	Sigma	98+	Germany
Na ₅ P ₃ O ₁₀	Alfa Aesar	98	USA
Na ₂ HPO ₄	Sigma	99	Japan
Na ₂ P ₂ O ₇ , 10H ₂ O	Sigma-Aldrich	98	USA
Zn – Lactate	Colgate		USA
TiO ₂	Colgate		USA
Mica	Colgate		USA
Carrageenan	Colgate		USA
Glycerol	Sigma	99	Malaysia

Table S2: Part of the compounds present in the four toothpastes analyzed

Compound	Paste #1	Paste#2	Paste#3	Paste#4
Glycerin	*		*	*
Water	*	*	*	
Sorbitol	*	*		
Hydrated Silica	*	*	*	*
Stannous Chloride		*	*	
Trisodium phosphate			*	
Sodium hexametaphosphate			*	
Pentasodium triphophaste				*
Tetrasodium pyrophosphate	*			
Zinc phosphate	*			
Sodium Gluconate		*	*	
Polyacrylic Acid				*
Sodium Citrate	*			
Cocamidopropyl Betaine				*