

## ***Supporting Information***

### **Porous carbons from sustainable and mild activation for targeted high-performance CO<sub>2</sub> capture and storage**

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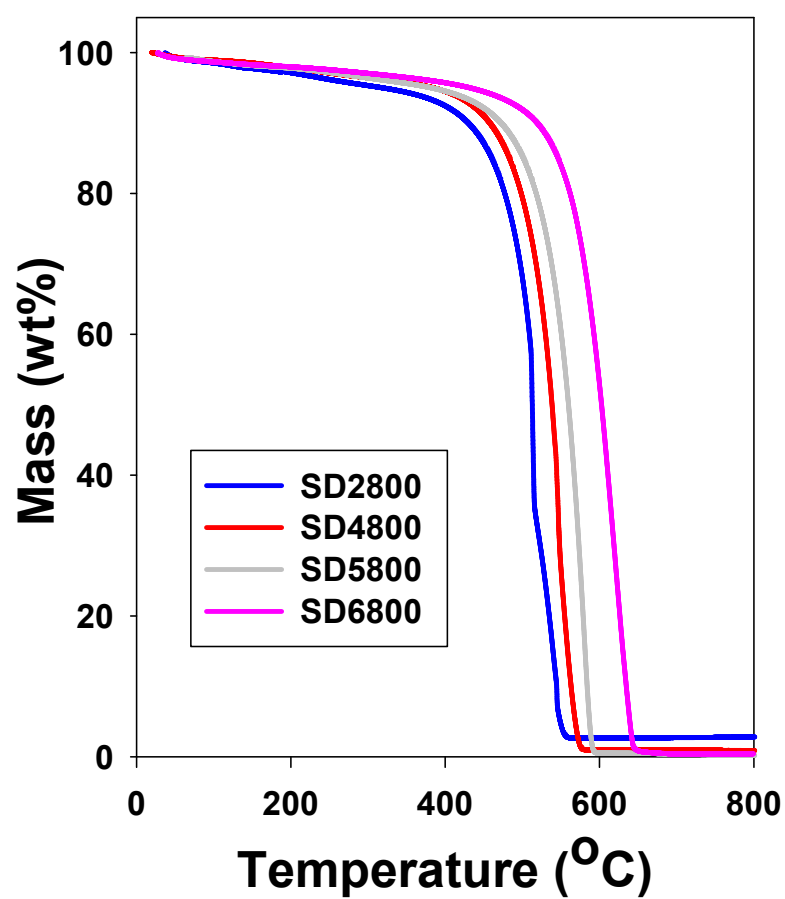
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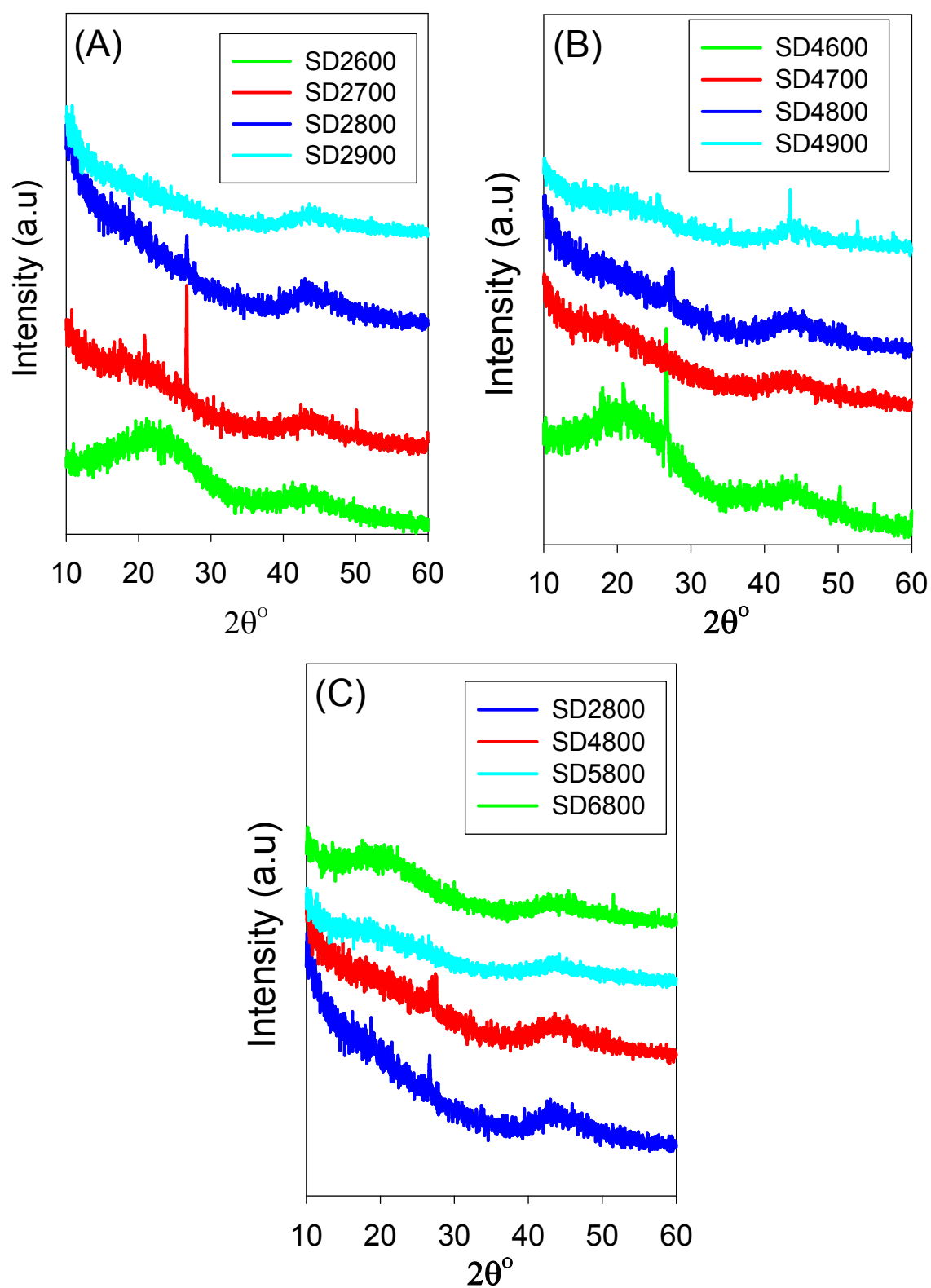
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Table S1. Yield and elemental analysis of carbons prepared at PO/precursor ratio of 2 from sawdust hydrochar (SD), polypyrrole (Ppy), and pre-mixed precursors containing sawdust hydrochar and polypyrrole (Ppy-SD).

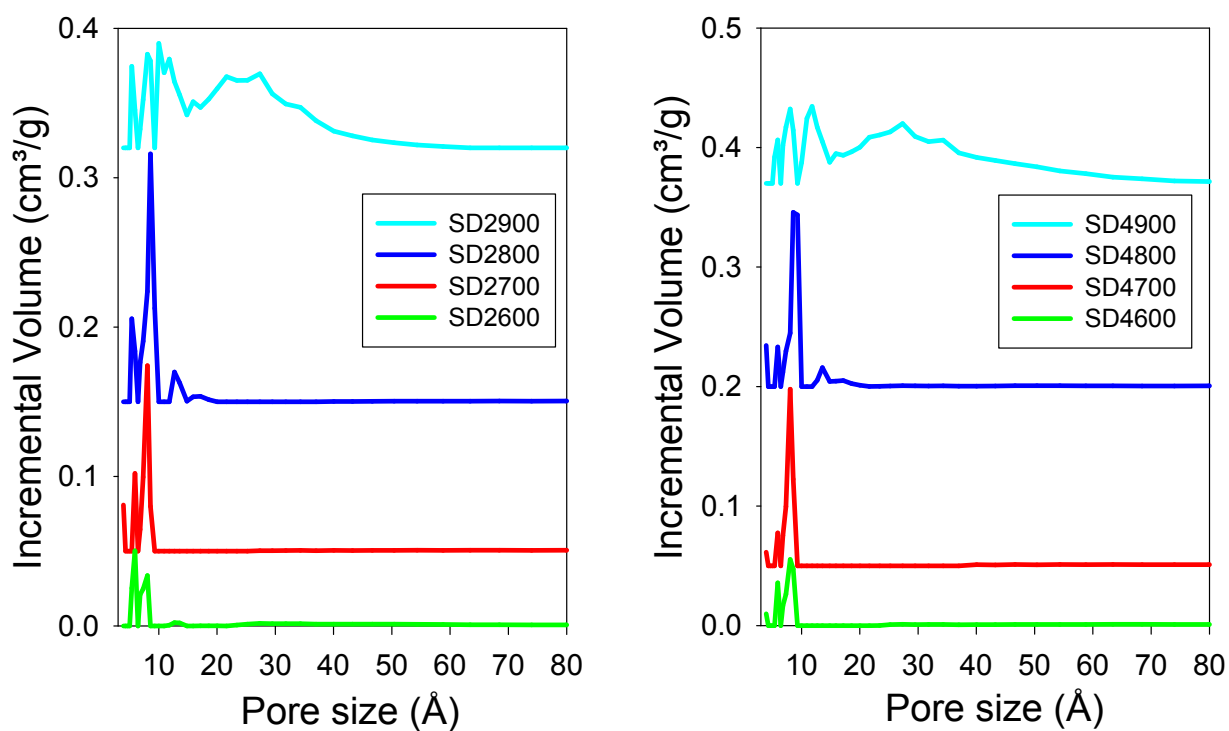
Sample	Yield (wt%)	Elemental composition (wt%)			
		C	H	N	O
SD2600	43	78.9	1.4	0.7	19.0
SD2700	41	81.1	0.2	0.2	18.5
SD2800	33	87.7	0	0.1	12.2
SD2900	21	90.7	0	0.2	9.1
Ppy-SD(1-2)2600	48	71.9	1.9	5.6	20.6
Ppy-SD(1-2)2700	27	74.0	1.8	2.9	21.3
Ppy-SD(1-2)2800	31	85.0	0.1	0.3	14.6
Ppy-SD(1-2)2900	17	91.4	0.0	0.4	8.2
Ppy-SD(1-1)2600	52	68.4	1.9	8.0	21.7
Ppy-SD(1-1)2700	39	80.2	0.3	2.0	17.5
Ppy-SD(1-1)2800	31	88.9	0.0	0.6	10.5
Ppy-SD(1-1)2900	18	91.4	0.0	0.7	7.9
Ppy-SD(2-1)2600	54	66.3	2.1	9.4	22.2
Ppy-SD(2-1)2700	39	73.7	0.3	2.3	23.7
Ppy-SD(2-1)2800	32	86.5	0.0	0.5	13.0
Ppy-SD(2-1)2900	19	90.1	0.0	0.5	9.4
Ppy2600	62	63.6	1.7	12.2	22.5
Ppy2700	43	77.6	0.8	4.0	17.6
Ppy2800	31	88.5	0.0	0.9	10.6
Ppy2900	24	91.9	0.0	1.2	6.9



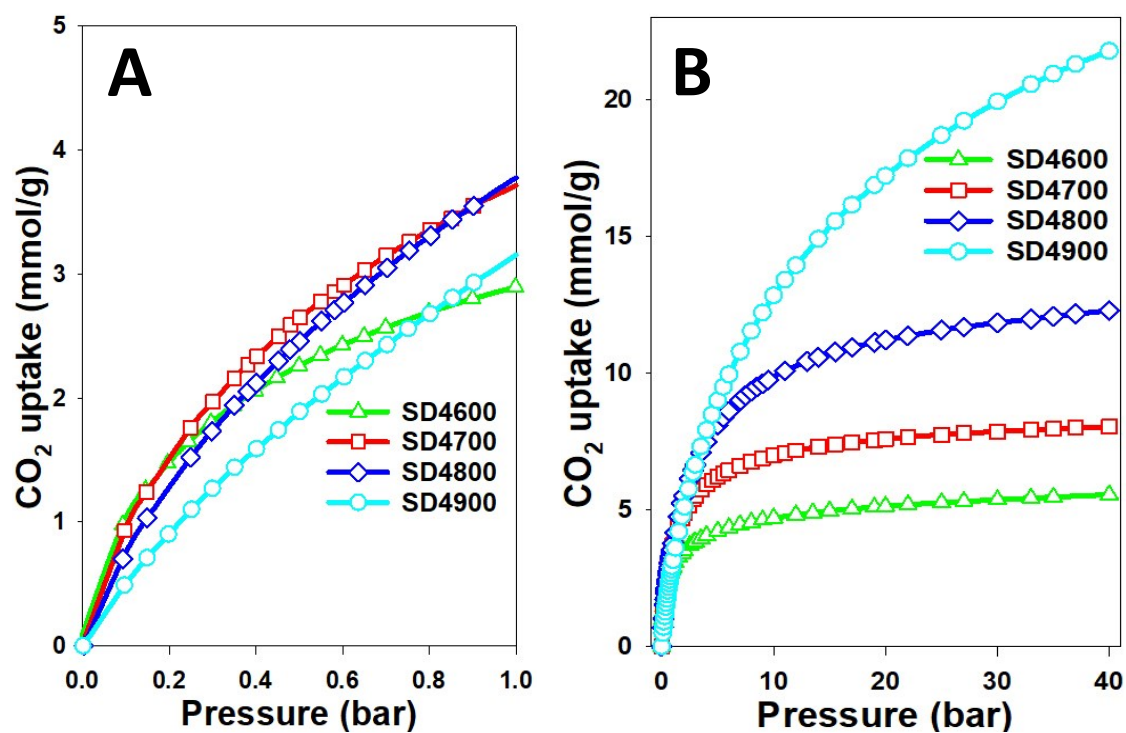
**Supporting Figure S1.** Thermogravimetric analysis of samples activated at of 800 °C and PO/SD ratio of 2, 4, 5 and 6.



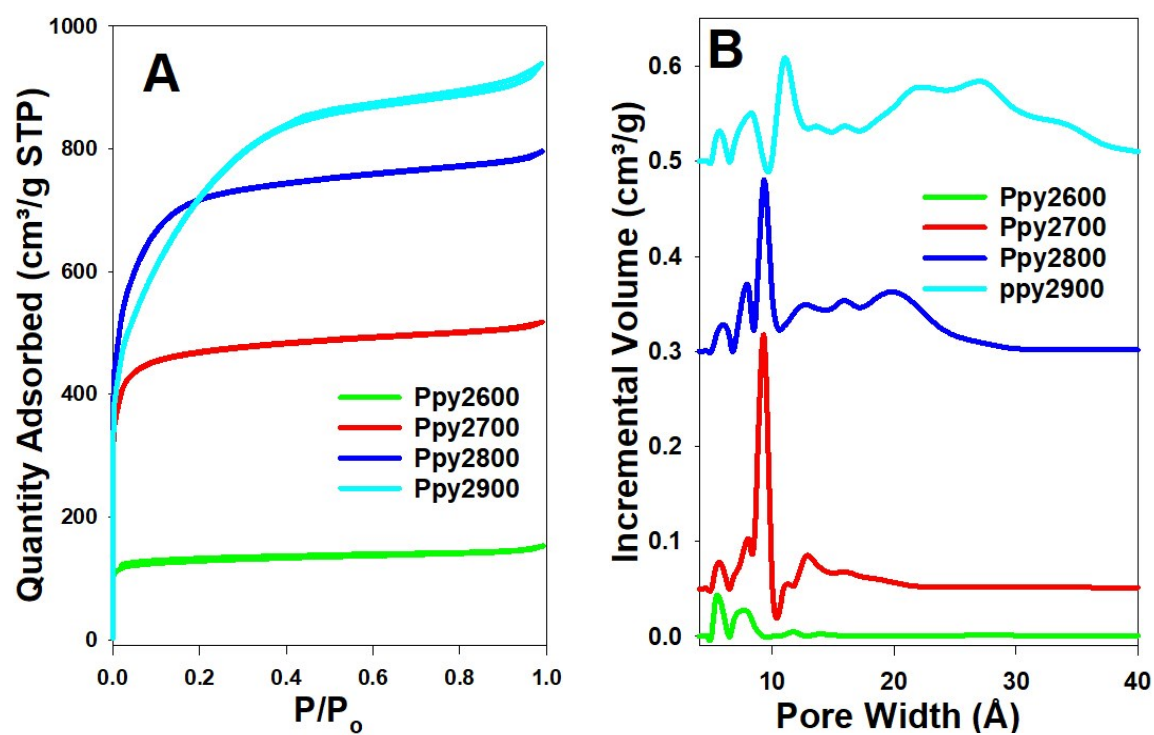
**Supporting Figure S2.** Powder XRD patterns of (A) samples activated at PO/SD ratio of 2, (B) samples activated at PO/SD ratio of 4 and (C) activated at 800 °C at various PO/SD ratios.



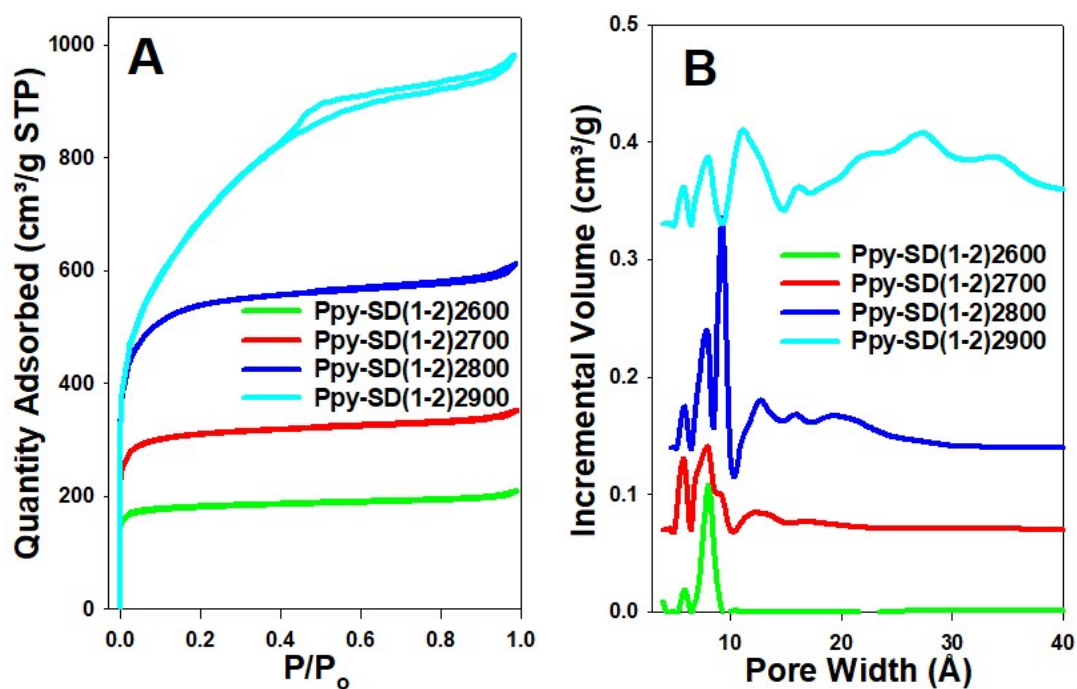
**Supporting Figure S3.** Pore size distributions curves for carbon samples activated at temperatures between 600 and 900 °C, and PO/SD ratio of 2 or 4.



**Supporting Figure S4.** Carbon dioxide uptake isotherms at 25 °C and pressure of (A) 0 – 1 bar and (B) 0 – 40 bar of carbons activated at between 600 and 900 °C and PO/SD ratio of 4.

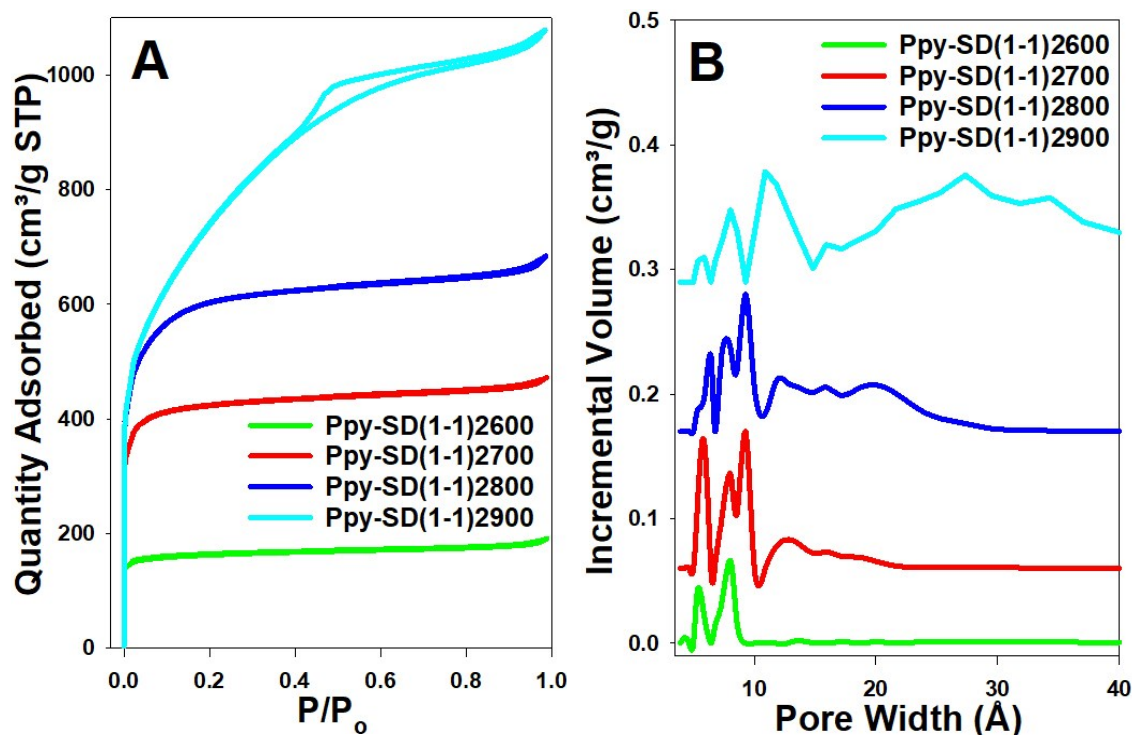


**Supporting Figure S5.** Nitrogen sorption isotherms (A) and pore size distribution curves (B) for polypyrrole (Ppy) derived carbons activated at 600 to 900 °C and PO/Ppy ratio of 2.

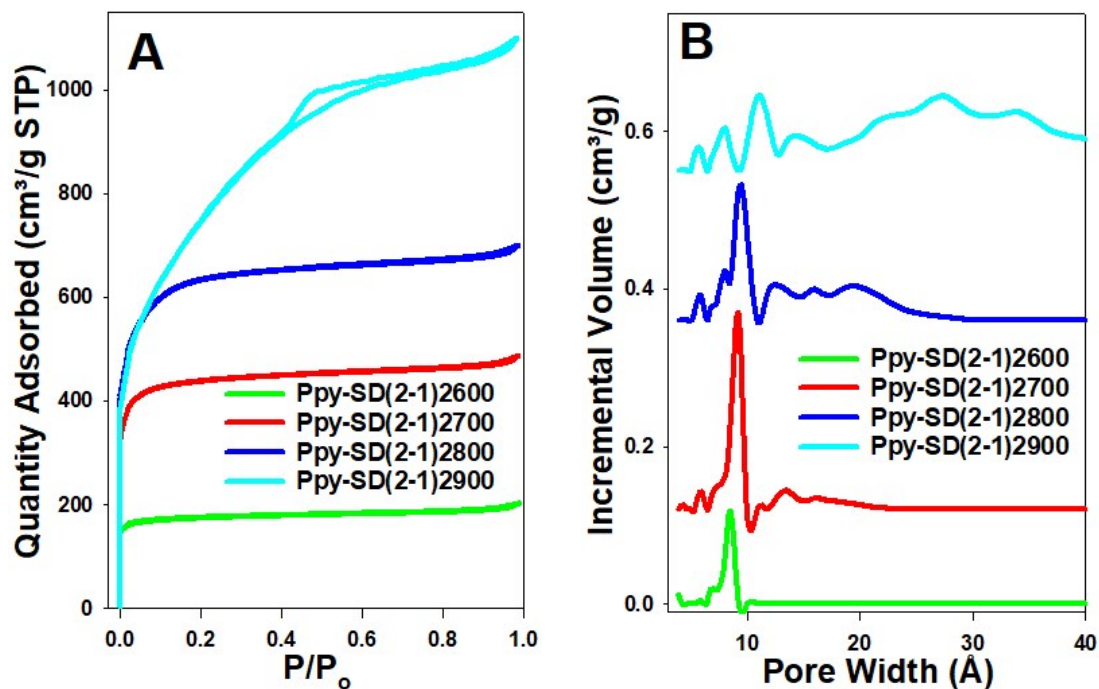


**Supporting Figure S6.** Nitrogen sorption isotherms (A) and pore size distribution curves (B) for carbons derived from pre-mixed precursors containing polypyrrole (Ppy) and sawdust hydrochar (SD) at Ppy:SD ratio of 1:2. The carbons were prepared at activation temperature of between 600 and 900 °C, and PO/precursor ratio of 2.

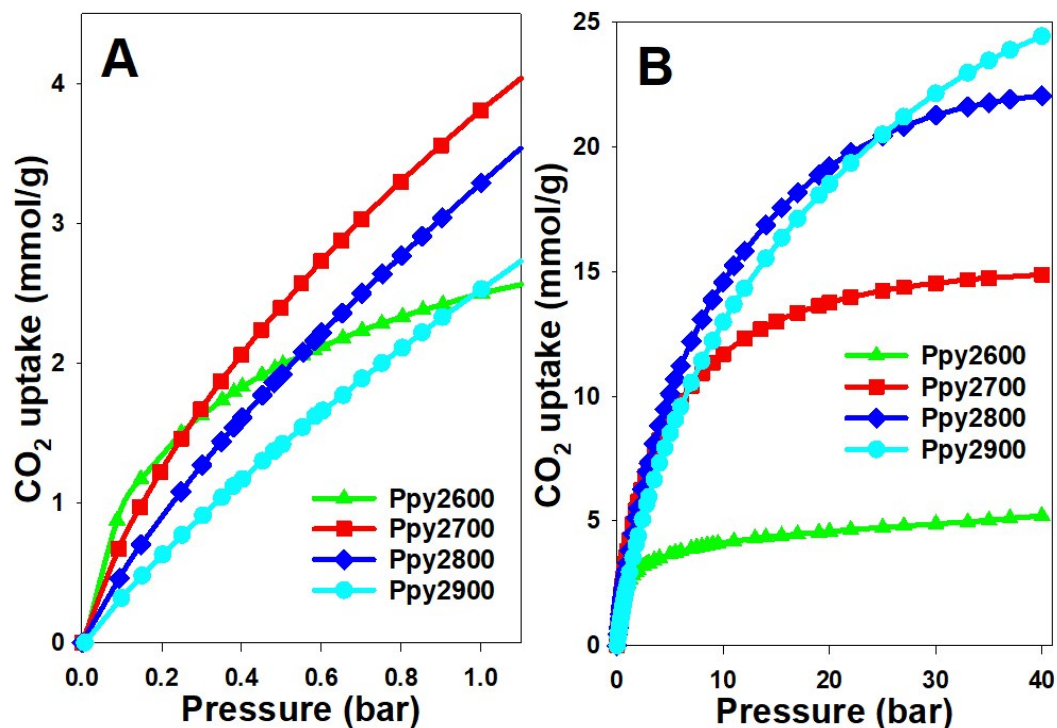




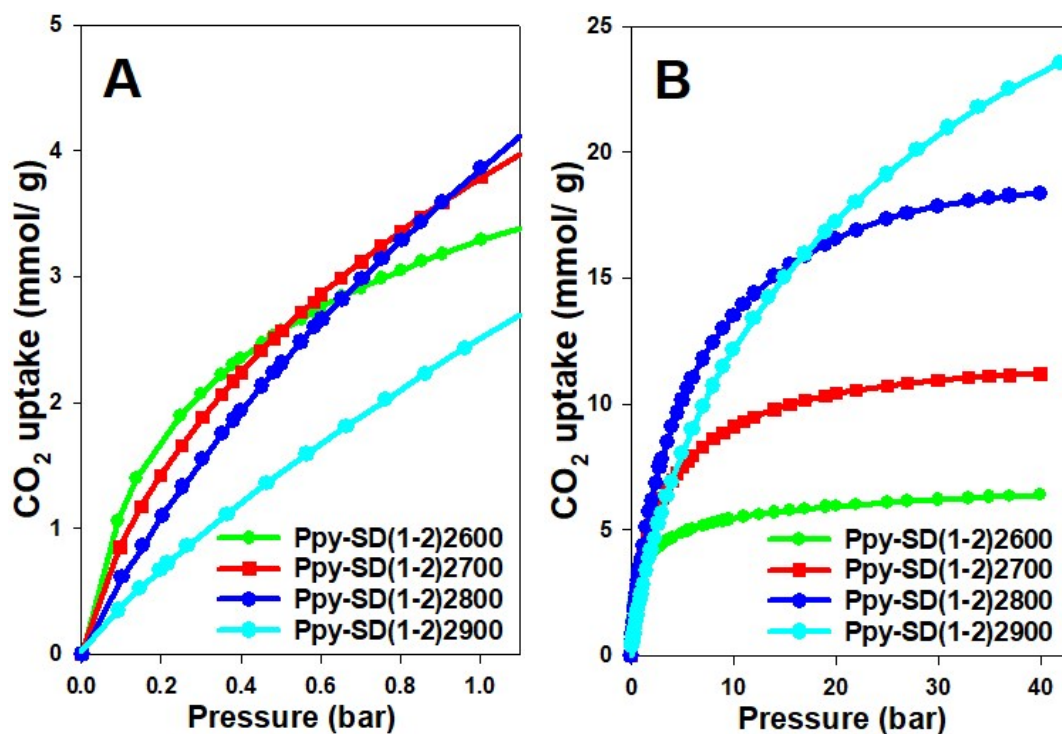
**Supporting Figure S7.** Nitrogen sorption isotherms (A) and pore size distribution curves (B) for carbons derived from pre-mixed precursors containing polypyrrole (Ppy) and sawdust hydrochar (SD) at Ppy:SD ratio of 1:1. The carbons were prepared at activation temperature of between 600 and 900 °C, and PO/precursor ratio of 2.



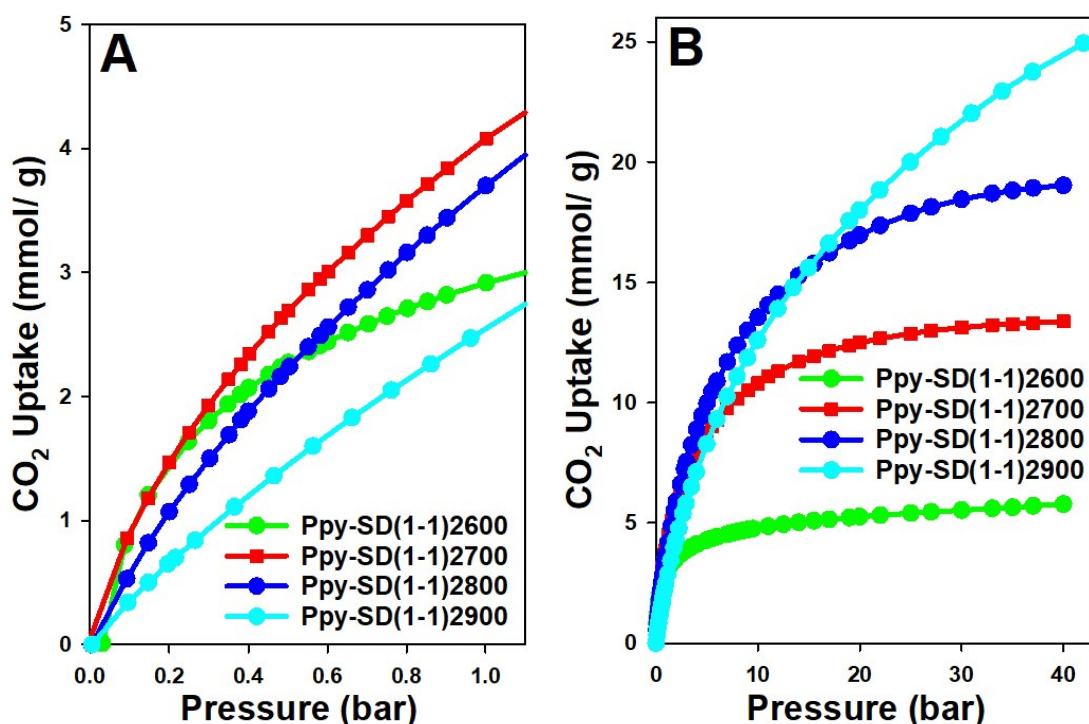
**Supporting Figure S8.** Nitrogen sorption isotherms (A) and pore size distribution curves (B) for carbons derived from pre-mixed precursors containing polypyrrole (Ppy) and sawdust hydrochar (SD) at Ppy:SD ratio of 2:1. The carbons were prepared at activation temperature of between 600 and 900 °C, and PO/precursor ratio of 2.



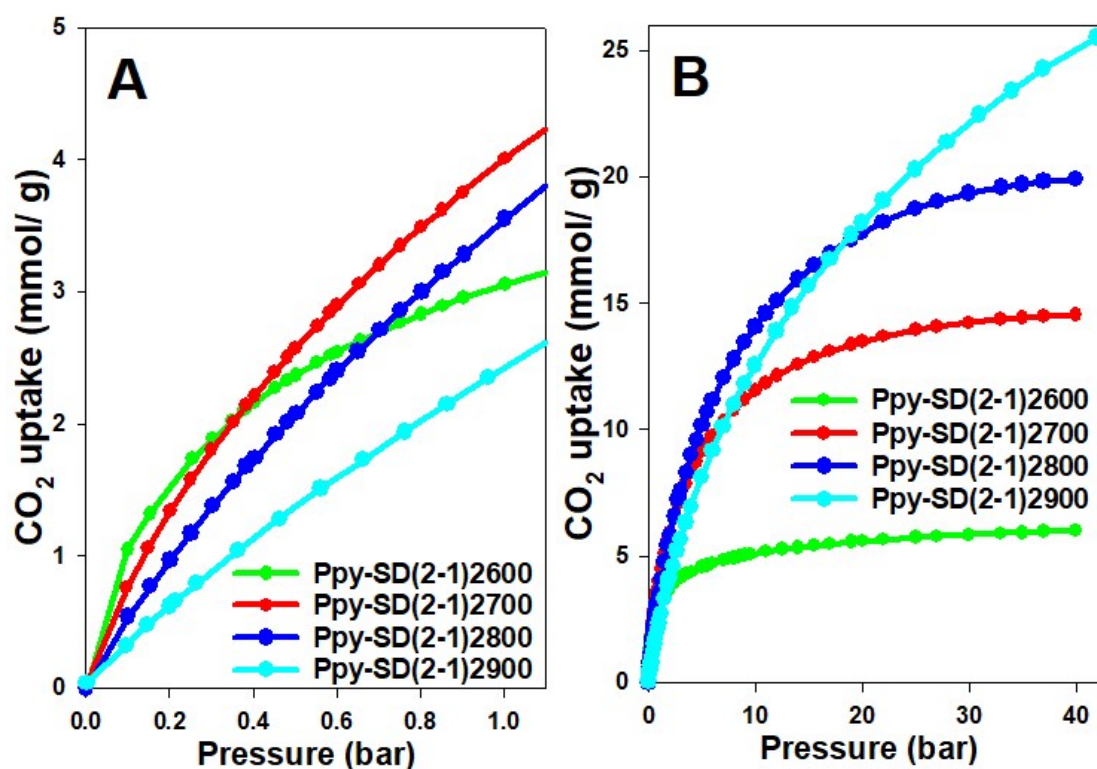
**Supporting Figure S9.** Carbon dioxide uptake isotherms at 25 °C and pressure of (A) 0 – 1 bar and (B) 0 – 40 bar of polypyrrole (Ppy) derived carbons activated at between 600 and 900 °C, and PO/Ppy ratio of 2.



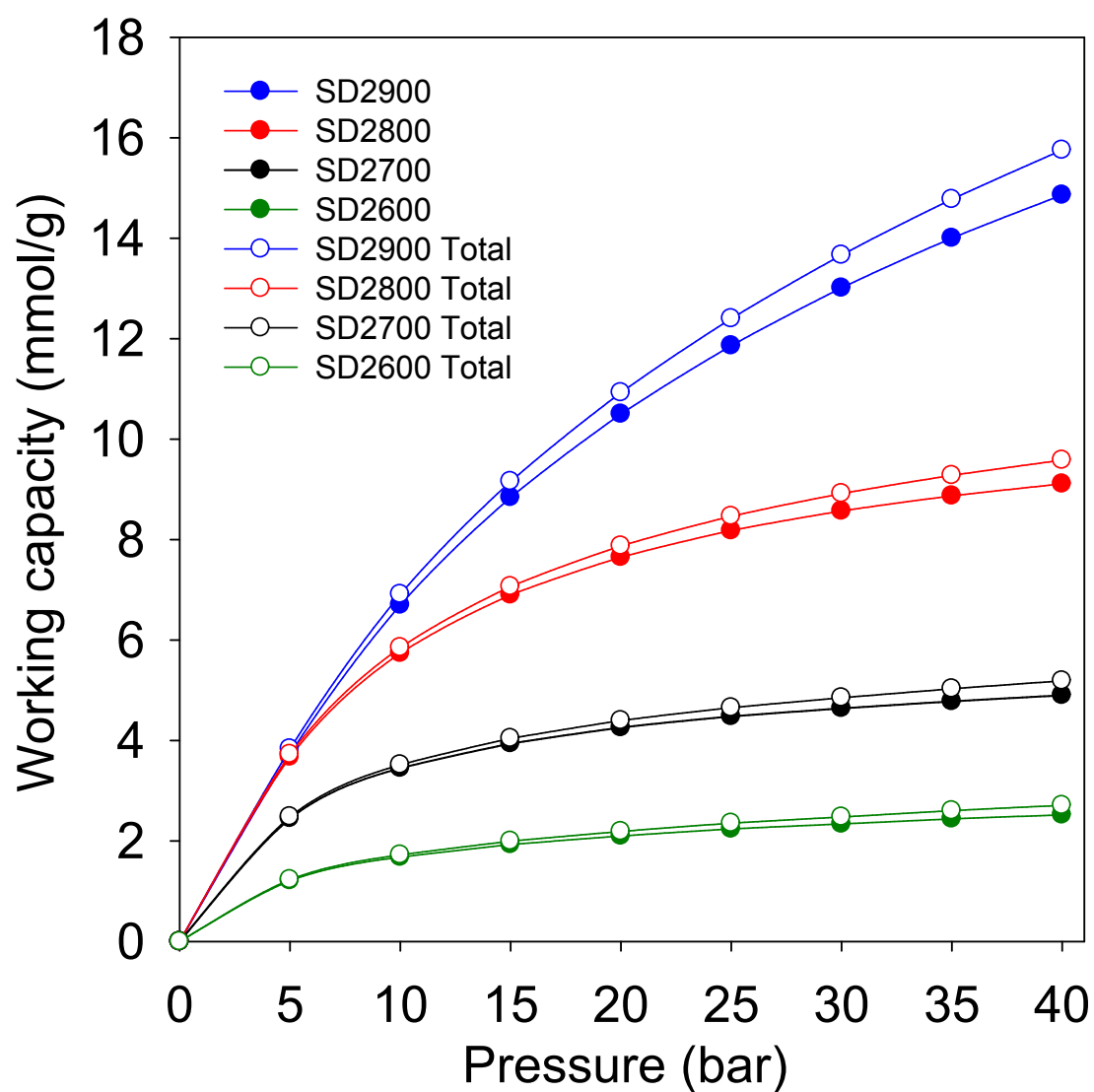
**Supporting Figure S10.** Carbon dioxide uptake isotherms at 25 °C and pressure of (A) 0 – 1 bar and (B) 0 – 40 bar of carbons derived from pre-mixed precursors containing polypyrrole (Ppy) and sawdust hydrochar (SD) at Ppy:SD ratio of 1:2. The carbons were prepared at activation temperature of between 600 and 900 °C, and PO/precursor ratio of 2.



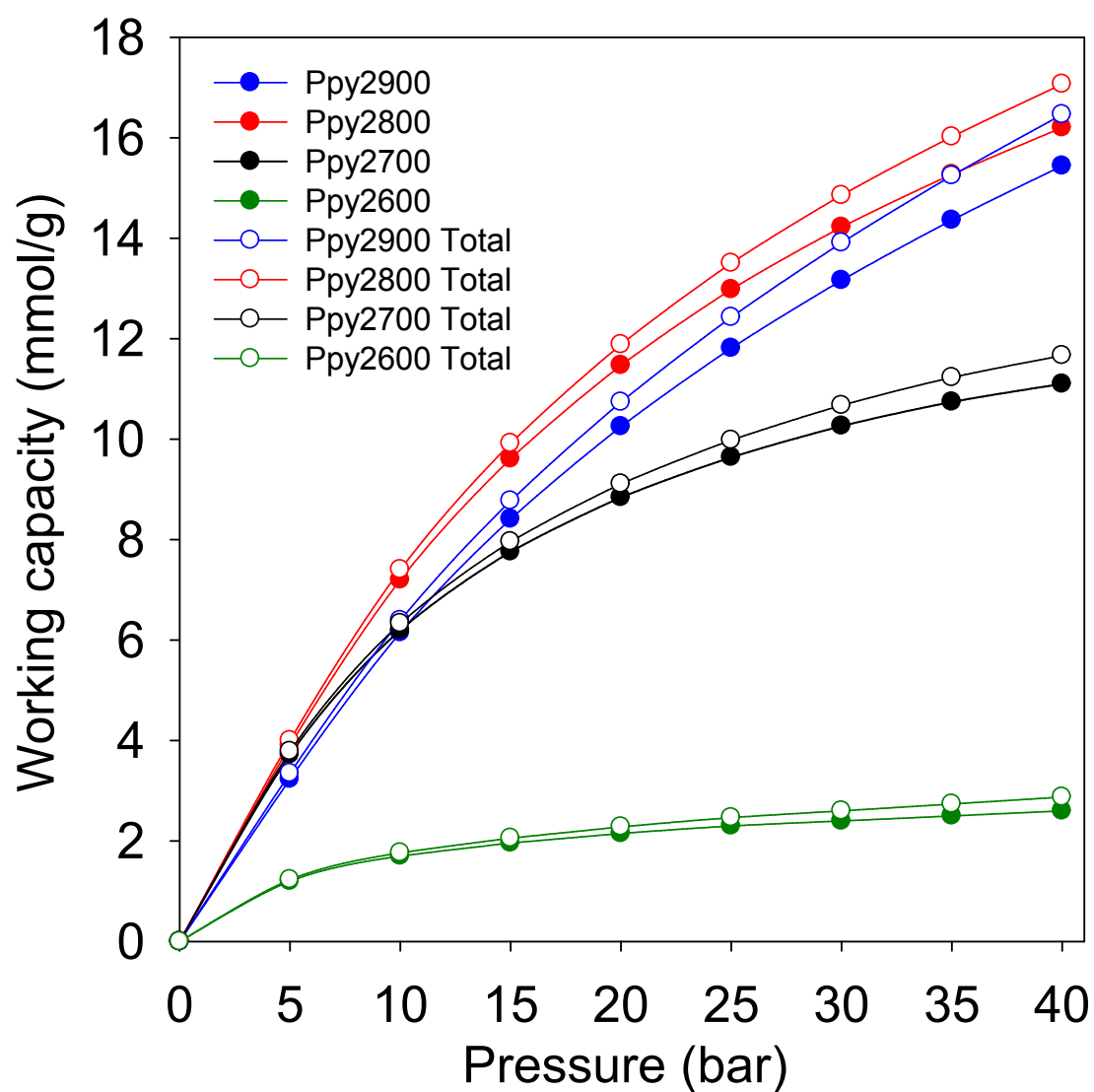
**Supporting Figure S11.** Carbon dioxide uptake isotherms at 25 °C and pressure of (A) 0 – 1 bar and (B) 0 – 40 bar of carbons derived from pre-mixed precursors containing polypyrrole (Ppy) and sawdust hydrochar (SD) at Ppy:SD ratio of 1:1. The carbons were prepared at activation temperature of between 600 and 900 °C, and PO/precursor ratio of 2.



**Supporting Figure S12.** Carbon dioxide uptake isotherms at 25 °C and pressure of (A) 0 – 1 bar and (B) 0 – 40 bar of carbons derived from pre-mixed precursors containing polypyrrole (Ppy) and sawdust hydrochar (SD) at Ppy:SD ratio of 2:1. The carbons were prepared at activation temperature of between 600 and 900 °C, and PO/precursor ratio of 2.

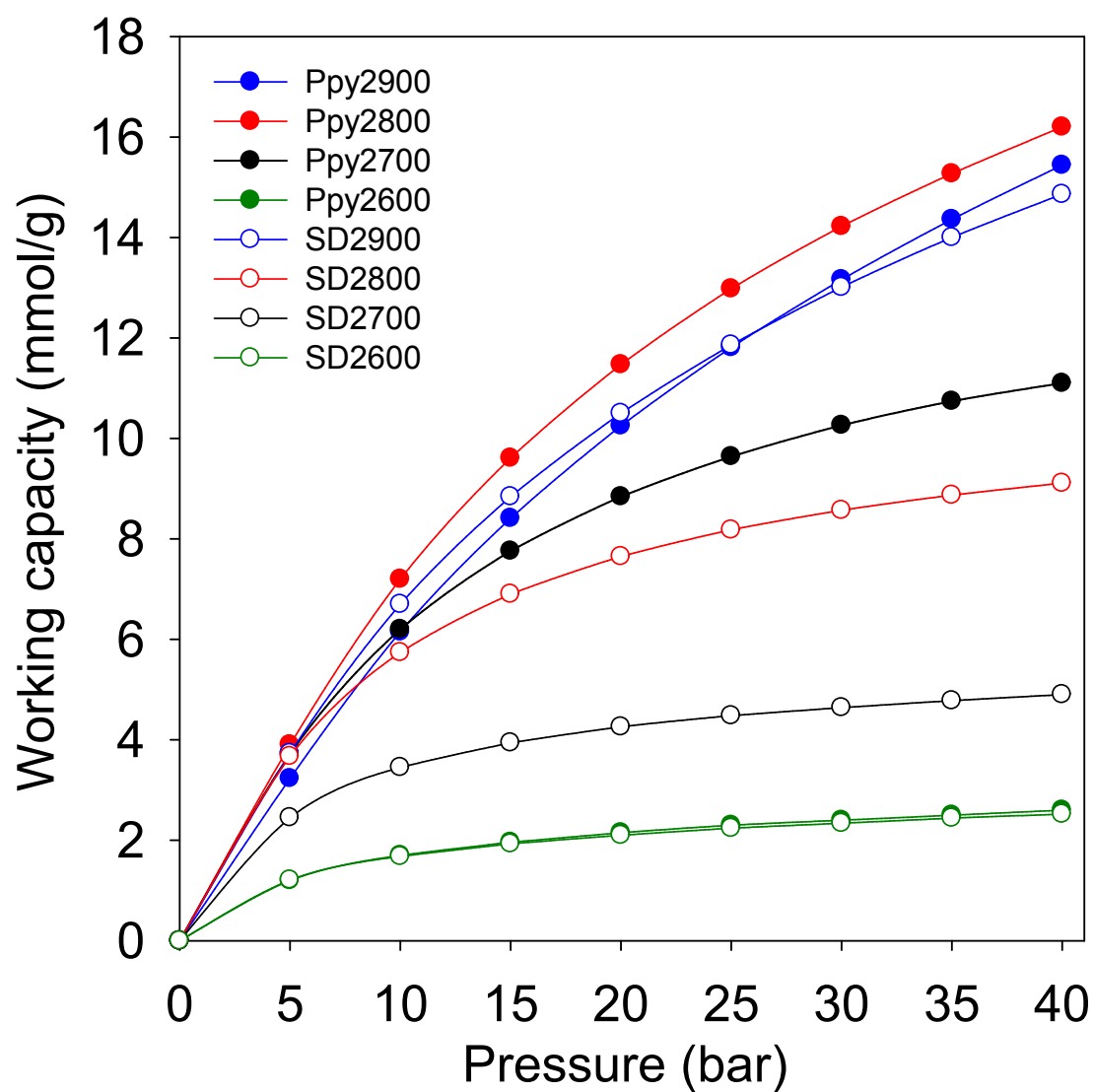


**Supporting Figure S13.** Excess and total pressure swing adsorption (PSA) CO<sub>2</sub> working capacity for a 60:40 H<sub>2</sub>/CO<sub>2</sub> mixture for sawdust hydrochar (SD) derived carbons activated at between 600 and 900 °C, and PO/SD ratio of 2.

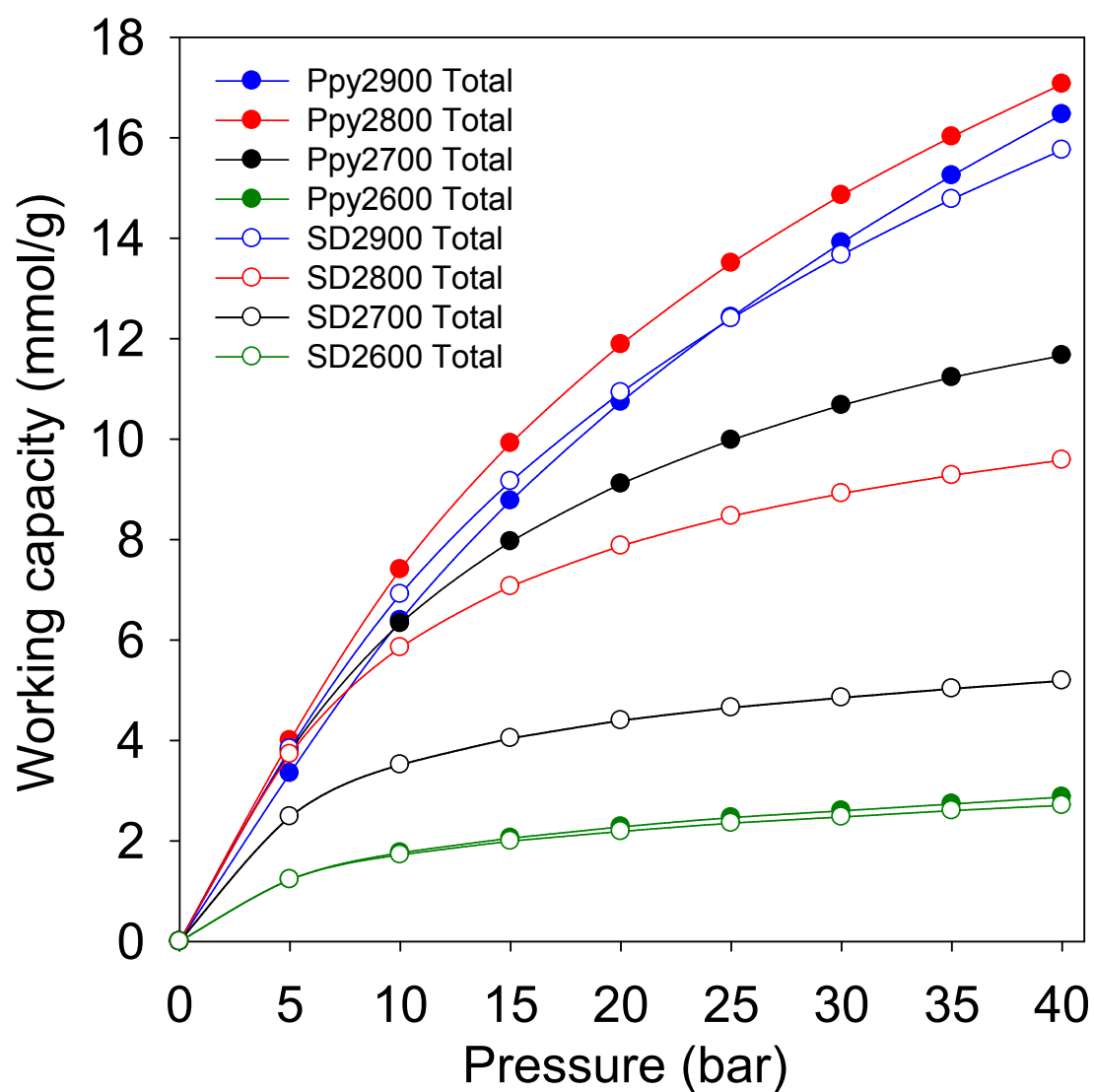


**Supporting Figure S14.** Excess and total pressure swing adsorption (PSA) CO<sub>2</sub> working capacity for a 60:40 H<sub>2</sub>/CO<sub>2</sub> mixture for polypyrrole (Ppy) derived carbons activated at between 600 and 900 °C, and PO/SD ratio of 2

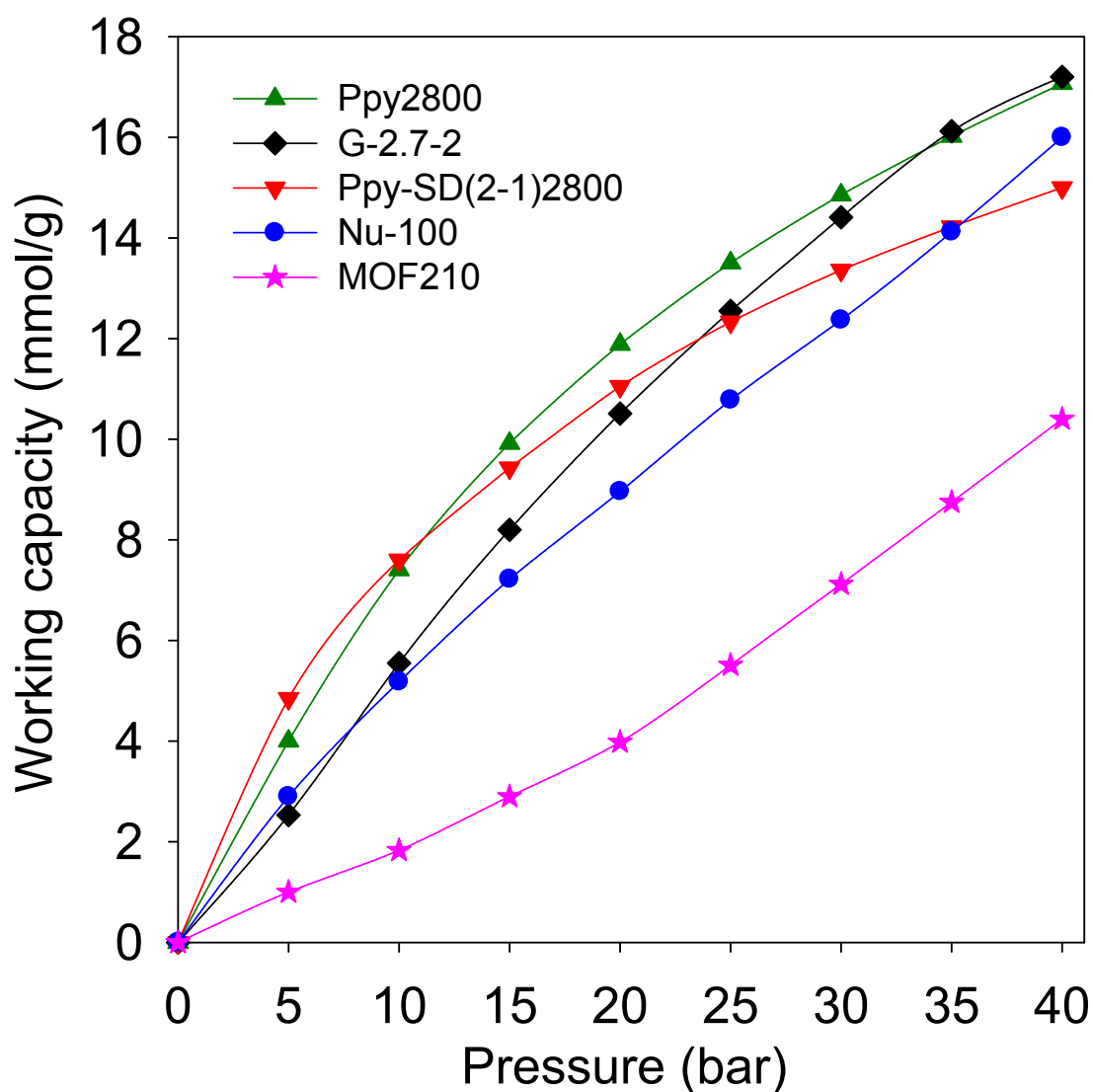




**Supporting Figure S15.** Excess pressure swing adsorption (PSA) CO<sub>2</sub> working capacity for a 60:40 H<sub>2</sub>/CO<sub>2</sub> mixture for sawdust hydrochar (SD) and polypyrrole (Ppy) derived carbons activated at between 600 and 900 °C, and PO/SD ratio of 2.



**Supporting Figure S16.** Excess pressure swing adsorption (PSA) CO<sub>2</sub> working capacity for a 60:40 H<sub>2</sub>/CO<sub>2</sub> mixture for sawdust hydrochar (SD) and polypyrrole (Ppy) derived carbons activated at between 600 and 900 °C, and PO/SD ratio of 2.



**Supporting Figure S17.** Excess pressure swing adsorption (PSA) CO<sub>2</sub> working capacity for a 60:40 H<sub>2</sub>/CO<sub>2</sub> mixture for potassium oxalate activated carbons (Ppy2800 and Ppy-SD(2-1)2800) compared to a porous carbon (G-2.7-2) and metal organic frameworks (NU-100 and MOF210) that have much higher surface area.

Data for G-2.7-2 is from the following reference: M. Sevilla, A. M. Al-Jumialy, A. B. Fuertes and R. Mokaya, *ACS Appl. Mater. Interfaces*, 2018, **10**, 1623.

Data for NU-100 and MOF210 is from the following reference: A. R. Millward and O. M. Yaghi, *J. Am. Chem. Soc.*, 2005, **127**, 17998.