

Influence of Counter Ions of Ammonium for Nitrogen Doping in Hydrothermal Carbonization: Characterization and Supercapacitor Performance

Kenneth G. Latham,^{1,2} Marveh Forghani³, Wesley M. Dose^{4,5}, Jessica A. Allen⁶, and Scott W. Donne¹*

¹Discipline of Chemistry, University of Newcastle, Callaghan, NSW 2308, Australia

²Umeå University, Department of Chemistry, SE-90 187, Umeå, Sweden

³Institute of Future Transport and Cities, Coventry University, Priory Street, Coventry CV1 5FB, UK

⁴Department of Chemistry, University of Cambridge, Lensfield Road, Cambridge, CB2 1EW, Cambridge, UK

⁵Department of Engineering, University of Cambridge, 17 Charles Babbage Road, CB3 0FS, Cambridge, UK

⁶School of Engineering, University of Newcastle, Callaghan, NSW 2308, Australia

Supplementary Information

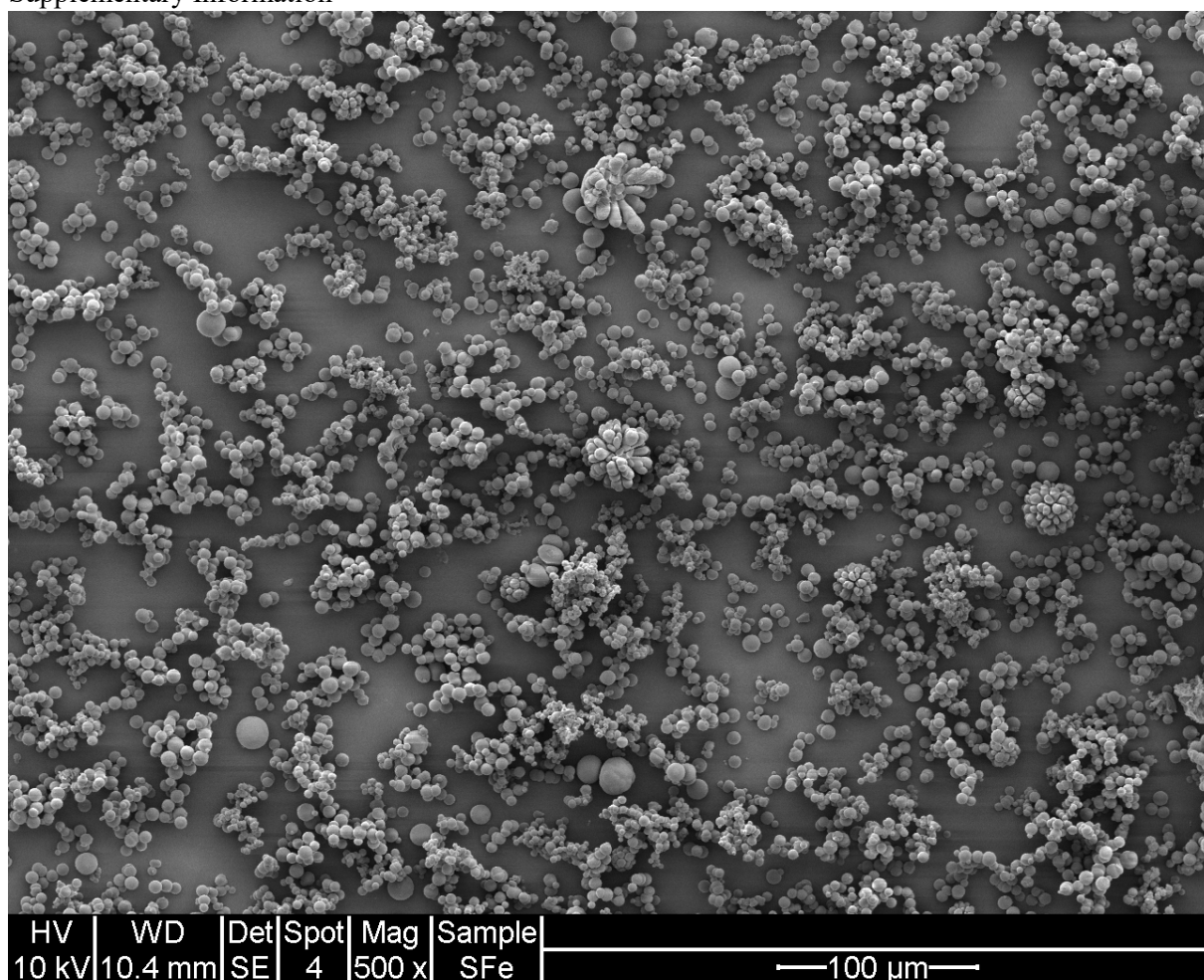


Figure S1. SEM Image of H-Fe with several "flower-like" structures at 500x.



Figure S2. 10000x SEM Image of the “flower-like” structure in H-Fe.

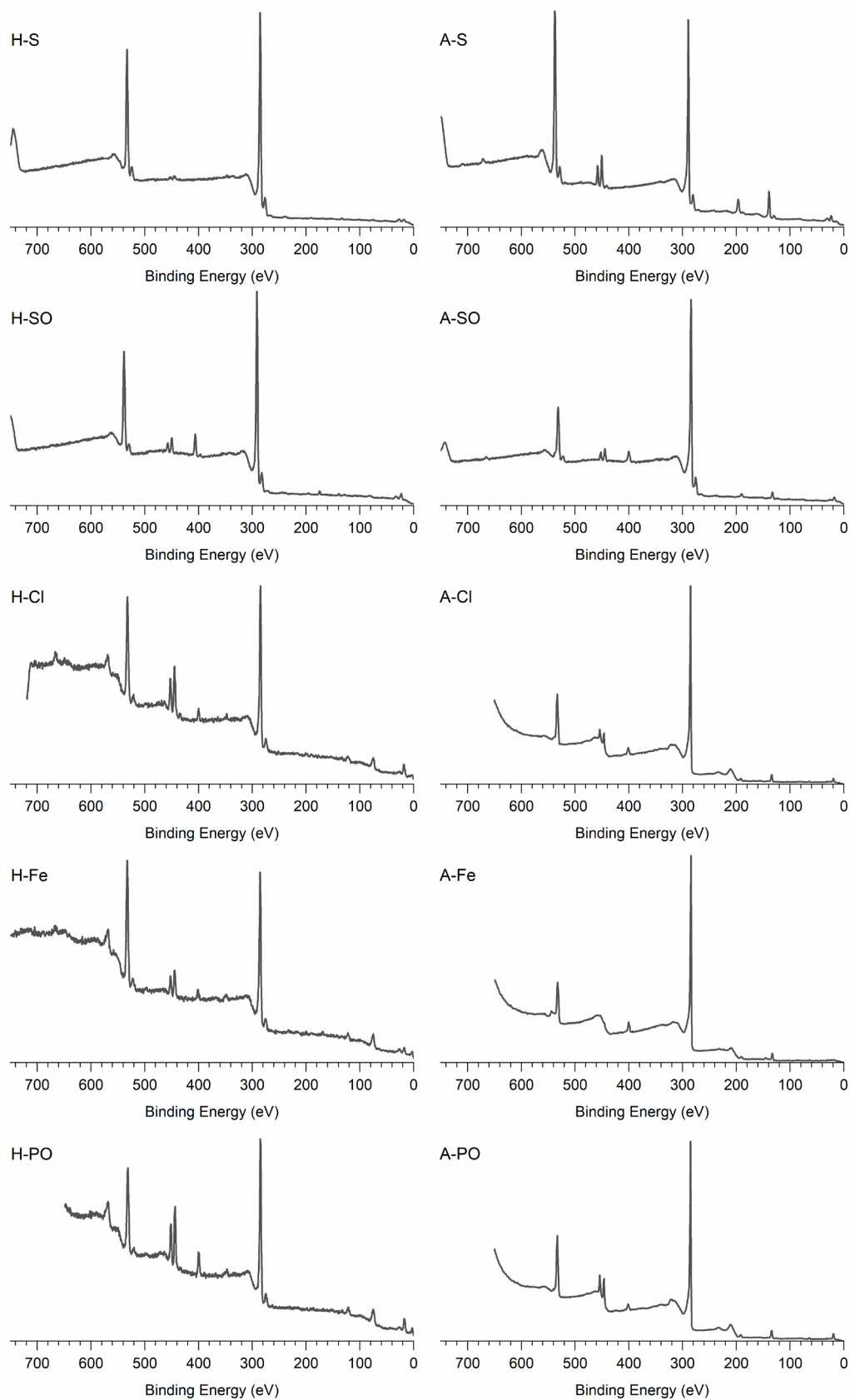
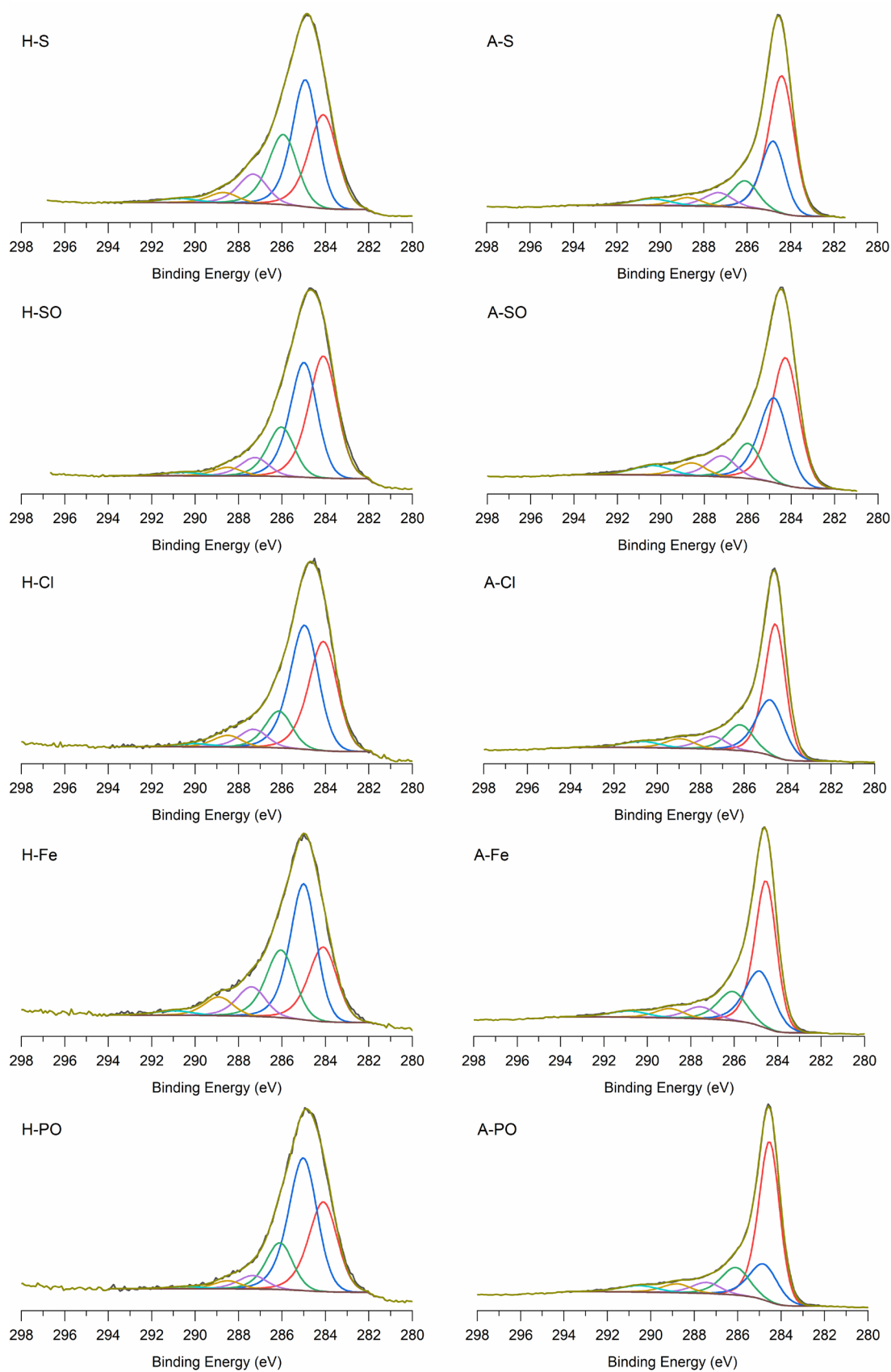
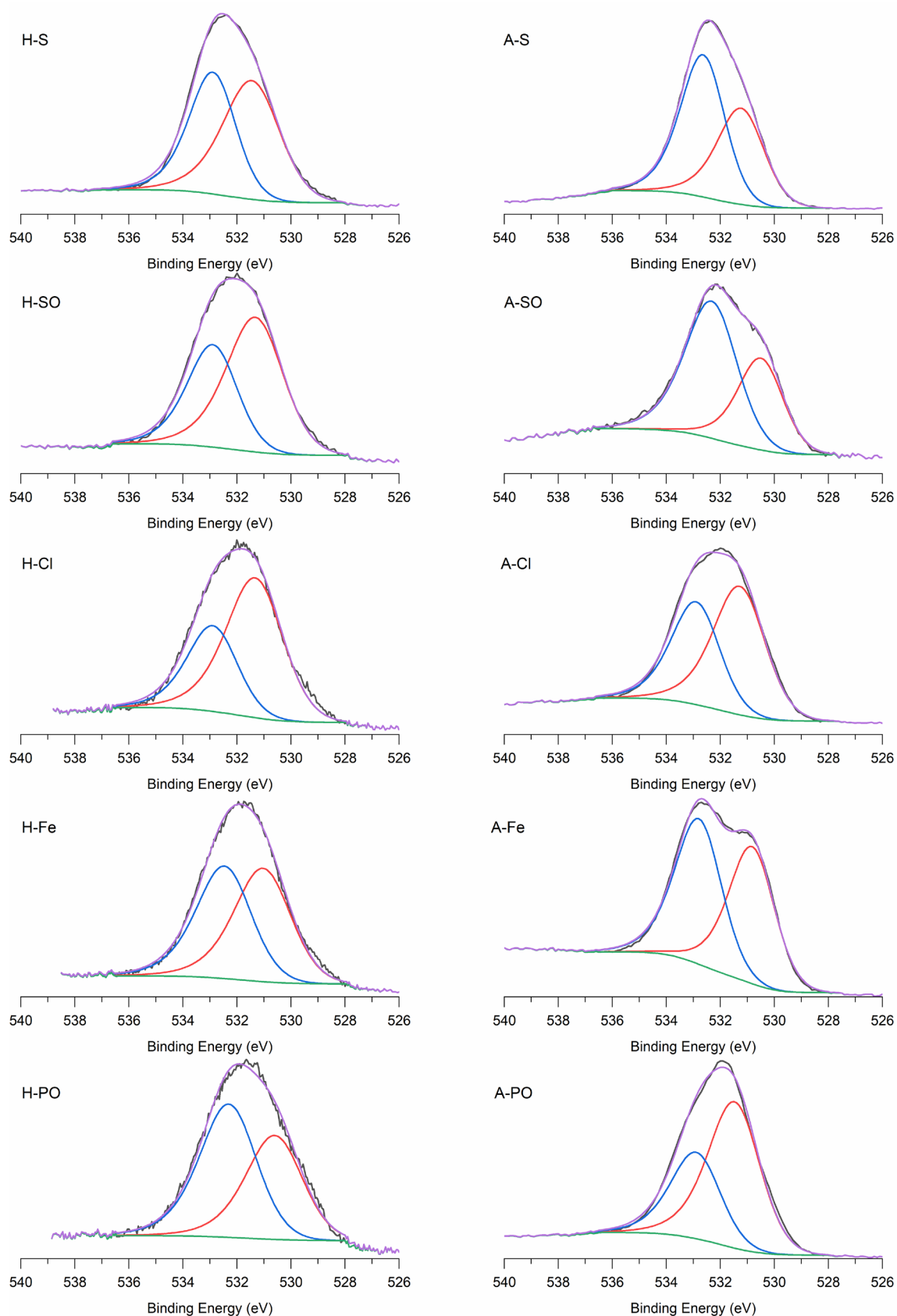


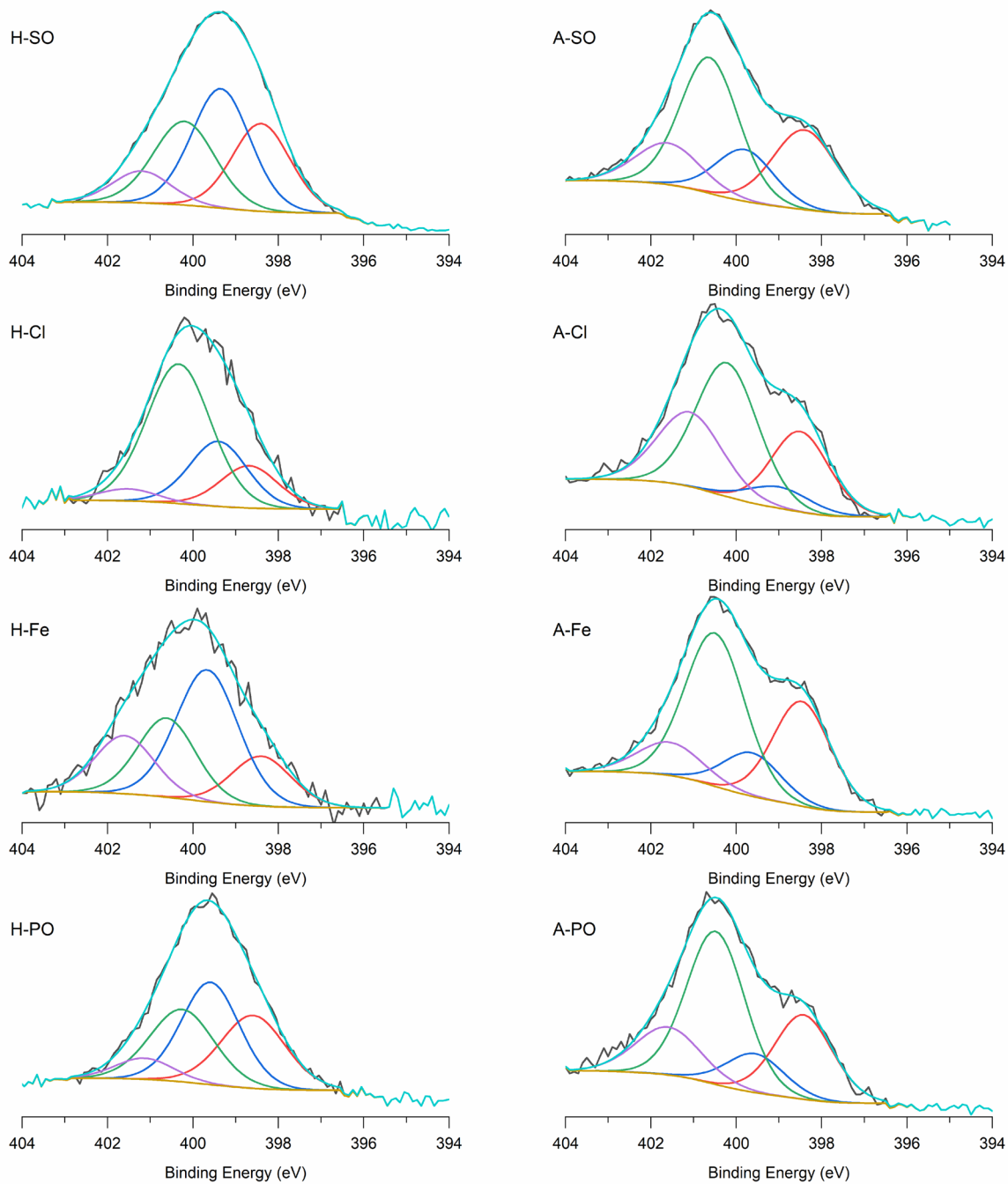
Figure S3. XPS Survey Spectra for each of the samples.



S4. XPS C1s spectra of each samples with deconvolution.



S5. XPS O1s spectra of each sample with deconvolution.



S6. XPS N1s spectra of each sample with deconvolution.

Table S1. Normalization for the C1s and N1s XPS Data

	Carbon				Nitrogen			
	C-OH	-C=O	-COOH	-CO ₃	Pyridinic	Amine	Pyrroles	Quaternary
H-Con	1.00	0.42	0.15	0.08	-	-	-	-
H-SO	1.00	0.40	0.19	0.10	1.00	1.37	1.01	0.36
H-Cl	1.00	0.53	0.34	0.13	1.00	0.67	1.60	0.13
H-Fe	1.00	0.44	0.28	0.09	1.00	1.34	2.03	0.90
H-PO	1.00	0.31	0.18	0.06	1.00	1.11	0.61	0.12
A-Con	1.00	0.57	0.34	0.37	-	-	-	-
A-SO	1.00	0.68	0.42	0.42	1.00	0.59	1.55	0.53
A-Cl	1.00	0.53	0.40	0.34	1.00	0.29	1.76	1.04
A-Fe	1.00	0.41	0.32	0.29	1.00	0.41	1.50	0.35
A-PO	1.00	0.41	0.33	0.32	1.00	0.46	1.81	0.63