

## Different electronic states in the isomorphous chiral vs racemic organic conducting salts, $\beta''$ -(BEDT-TTF)<sub>2</sub>(S- and *rac*-PROXYL-CONHCH<sub>2</sub>SO<sub>3</sub>)

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### Supplementary information

Table S1. Crystallographic data of (PPh<sub>4</sub>)*rac*-, *R*- and *S*-1

Compound	(PPh <sub>4</sub> ) <i>R</i> -1	(PPh <sub>4</sub> ) <i>S</i> -1	(PPh <sub>4</sub> ) <i>rac</i> -1*
Name	(PPh <sub>4</sub> ) <i>S</i> -PROXYL-CONHCH <sub>2</sub> SO <sub>3</sub>	(PPh <sub>4</sub> ) <i>R</i> -PROXYL-CONHCH <sub>2</sub> SO <sub>3</sub>	(PPh <sub>4</sub> ) <i>rac</i> -PROXYL-CONHCH <sub>2</sub> SO <sub>3</sub>
Formula	C <sub>34</sub> H <sub>38</sub> N <sub>2</sub> O <sub>5</sub> PS	C <sub>34</sub> H <sub>38</sub> N <sub>2</sub> O <sub>5</sub> PS	C <sub>34</sub> H <sub>38</sub> N <sub>2</sub> O <sub>5</sub> PS
Fw	617.72	617.72	617.72
T (K)	106	106	294
Space Group	<i>P</i> 1	<i>P</i> 1	<i>P</i> 1̄
<i>a</i> (Å)	11.6227(2)	11.62009(14)	11.691(4)
<i>b</i> (Å)	11.83230(18)	11.83293(15)	12.042(3)
<i>c</i> (Å)	12.1976(2)	12.19880(15)	12.306(3)
$\alpha$ (°)	83.7248(13)	83.7027(10)	83.20(2)
$\beta$ (°)	84.4992(14)	84.5080(10)	84.25(3)
$\gamma$ (°)	68.9233(15)	68.9172(11)	68.48(2)
<i>V</i> (Å <sup>3</sup> )	1552.96(5)	1552.75(4)	1597.4(8)
<i>Z</i>	2	2	2
<i>d</i> <sub>calc</sub> (g·cm <sup>-1</sup> )	1.321	1.321	1.284
$\mu$ (cm <sup>-1</sup> )	2.006	2.006	1.950
<i>F</i> (000)	654	654	654
2 <i>θ</i> range (°)	4-55	4-55	6-55
Total ref.	42137	47016	7690
Unique ref.	13469	13511	7333
<i>R</i> <sub>int</sub>	0.0561	0.0479	0.0118
Parameters	775	775	388
<i>R</i> <sub>1</sub> ( <i>I</i> > 2σ( <i>I</i> ))	0.0392	0.0347	0.0433
w <i>R</i> <sub>2</sub> (all data)	0.0976	0.0901	0.1386
<i>S</i>	1.065	1.069	1.044
Δρ <sub>max</sub> (e Å <sup>-3</sup> )	0.35	0.30	0.30
Δρ <sub>min</sub> (e Å <sup>-3</sup> )	-0.40	-0.37	-0.41
Flack parameters	0.02(4)	-0.01(3)	-
CCDC reference	2022036	2022034	685677

\*Unit cell transformation was performed from reported cells<sup>8</sup> and re-refine the data.

Table S2. Crystallographic data of *rac*- and *S*-**2**

Compound	$\beta''\text{-(BEDT-TTF)}_2S\text{-1 (S-2)}$			$\beta''\text{-(BEDT-TTF)}_2rac\text{-1 (rac-2)}$		
Name	$\beta''\text{-(BEDT-TTF)}_2S\text{-PROXYL-CONHCH}_2\text{SO}_3$			$\beta''\text{-(BEDT-TTF)}_2Rac\text{-PROXYL-CONHCH}_2\text{SO}_3$		
Formula	$C_{30}H_{34}N_2O_5S_{17}$			$C_{30}H_{34}N_2O_5S_{17}$		
Fw	1047.63			1047.63		
T (K)	290	106	29	295*	106	28
Space Group	<i>P</i> 1	<i>P</i> 1	<i>P</i> 1	<i>P</i> 1̄	<i>P</i> 1̄	<i>P</i> 1̄
<i>a</i> (Å)	9.26369(19)	9.1764(3)	9.14212(13)	9.326(3)	9.28307(18)	9.23250(8)
<i>b</i> (Å)	12.4438(2)	12.2807(3)	12.22911(18)	12.436(4)	12.2324(2)	12.13977(12)
<i>c</i> (Å)	19.5405(3)	19.4508(5)	19.4069(3)	19.528(6)	19.4799(4)	19.4474(2)
$\alpha$ (°)	83.5517(14)	84.035(2)	84.1246(13)	83.44(3)	83.644(6)	83.9780(10)
$\beta$ (°)	82.9266(16)	83.606(2)	83.7827(13)	82.94(3)	83.390(6)	83.3850(10)
$\gamma$ (°)	70.2178(17)	70.196(2)	70.2319(13)	69.78(3)	69.658(5)	69.6680(10)
<i>V</i> (Å <sup>3</sup> )	2097.45(7)	2044.35(10)	2024.80(5)	2102.9(13)	2054.34(10)	2025.33(4)
<i>Z</i>	2	2	2	2	2	2
<i>d</i> <sub>calc</sub> (g·cm <sup>-3</sup> )	1.659	1.702	1.718	1.654	1.693	1.718
$\mu$ (cm <sup>-1</sup> )	9.161	9.399	9.490	9.138	9.354	9.488
<i>F</i> (000)	1080	1080	1080	1080	1080	1080
2 <i>θ</i> range (°)	4-55	4-55	4-55	5-55	4-55	4-55
Total ref.	56662	51519	104470	10383	20214	107410
Unique ref.	18190	17612	17596	9638	9348	9276
<i>R</i> <sub>int</sub>	0.0613	0.0528	0.0831	0.0774	0.0163	0.0275
Parameters	973	973	980	493	505	512
<i>R</i> <sub>1</sub> ( <i>I</i> > 2σ( <i>I</i> ))	0.0434	0.0486	0.0413	0.0665	0.0320	0.0353
w <i>R</i> <sub>2</sub> (all data)	0.1204	0.1405	0.1088	0.2341	0.0718	0.0771
<i>S</i>	1.050	1.037	1.058	0.935	1.070	1.342
Δρ <sub>max</sub> (e Å <sup>-3</sup> )	0.61	1.13	0.96	0.64	0.66	0.61
Δρ <sub>min</sub> (e Å <sup>-3</sup> )	-0.53	-0.63	-0.64	-0.59	-0.63	-0.62
Flack parameter	0.13(6)	0.11(6)	-0.01(3)	-	-	-
CCDC reference	2022041	2022039	2022037	685678	2022046	2022042

\*Unit cell transformation was performed from reported cells<sup>8</sup> and re-refine the data.

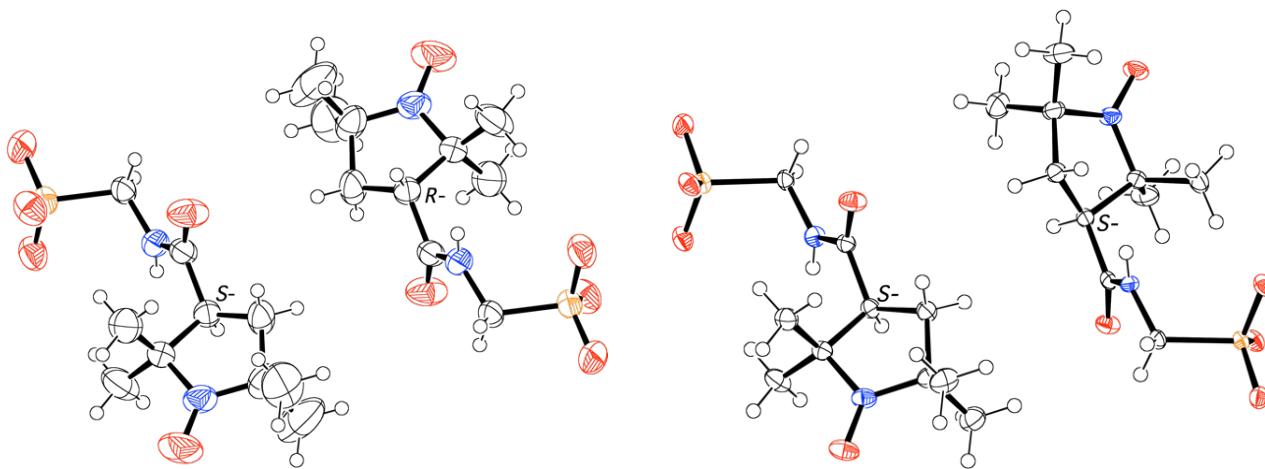


Fig. S1 Molecular structures of *rac*-**1** (left) and *S*-**1** (right) in their tetraphenylphosphonium salts at 294 and 105 K, respectively.

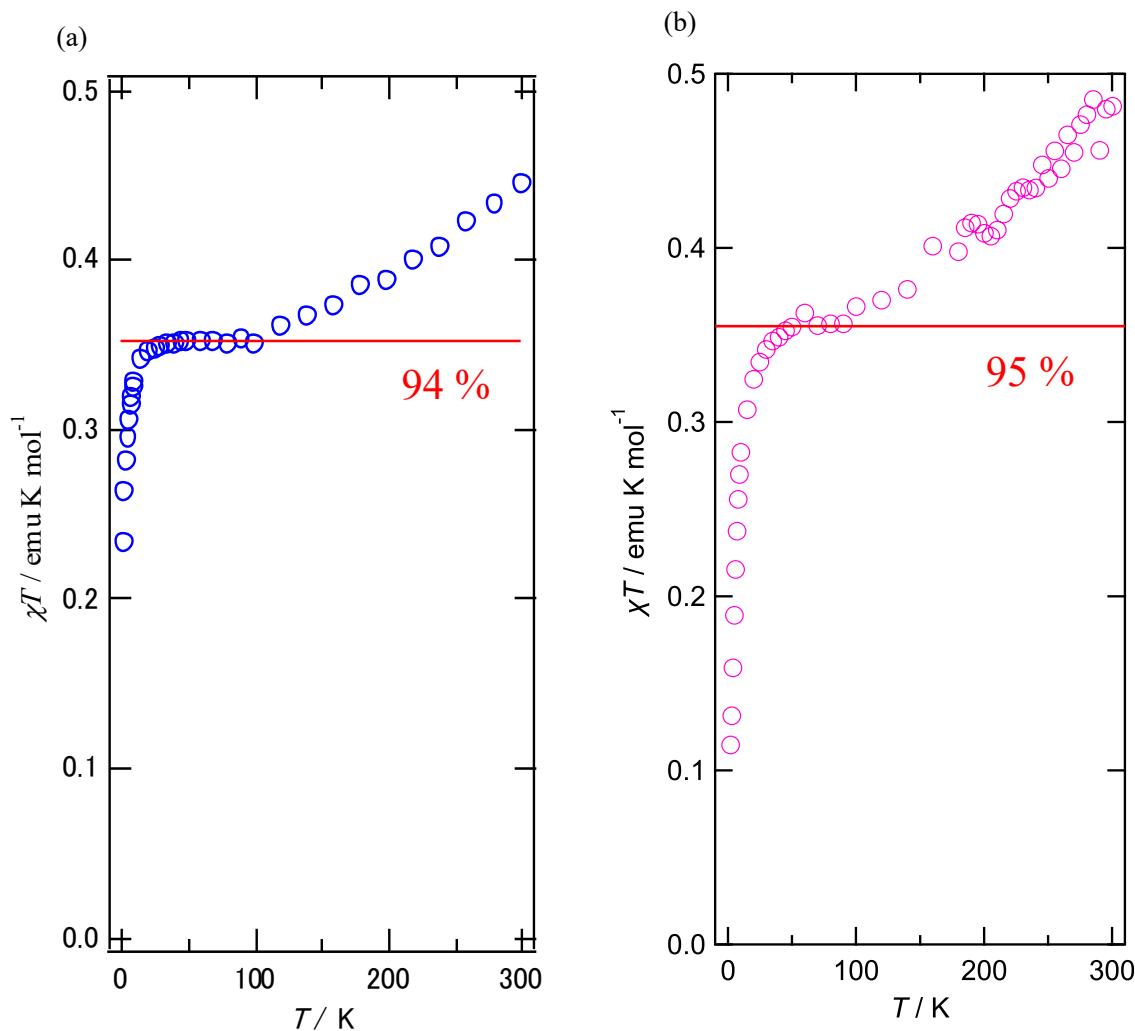
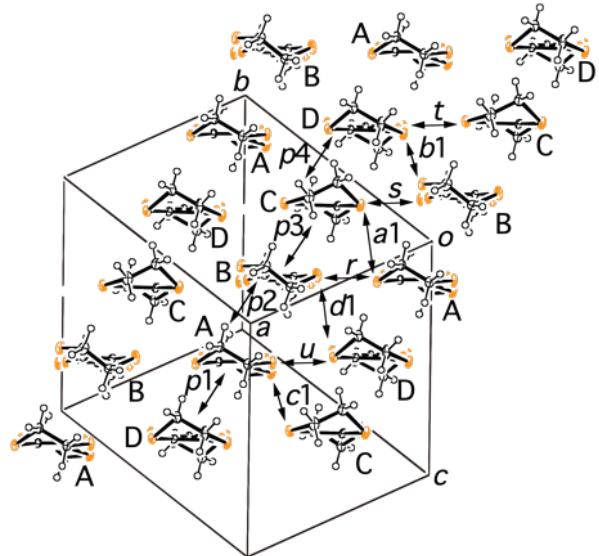


Fig. S2 Temperature-dependent magnetic susceptibility of *rac*-**2** (left)<sup>7</sup> and *S*-**2** (right). The red horizontal lines indicate the spin concentrations of  $s = 1/2$  attributed to organic free radicals of the anions. The *S*-**2** data can be fitted by the combination of a Curie-Weiss and a Singlet-Triplet models with  $C = 0.355 \text{ emu K mol}^{-1}$  ( $0.355 / 0.375 = 0.95$ ),  $\theta = -0.37 \text{ K}$  and  $J_{\text{Singlet-Triplet}} = -3.8 \text{ K}$ .

Table S3. Transfer integrals of *rac*- and *S*-**2**



$\times 10^{-3}$	<i>Racemic</i>		<i>S</i> -	
	106 K	28 K	106 K	29 K
<i>p</i> 1	-2.66	-3.27	-2.66	-2.66
<i>p</i> 2	7.82	8.23	7.32	7.81
<i>p</i> 3	-5.61	-5.66	-4.62	-4.54
<i>p</i> 4	= <i>p</i> 2	= <i>p</i> 2	6.93	7.06
<i>r</i>	-12.01	-13.14	-13.15	-14.03
<i>s</i>	10.57	11.27	11.00	11.74
<i>t</i>	-9.34	-9.61	-9.64	-9.90
<i>u</i>	= <i>s</i>	= <i>s</i>	10.70	10.83
<i>a</i> 1	17.75	18.39	17.14	17.56
<i>b</i> 1	10.76	10.97	9.82	9.98
<i>c</i> 1	= <i>b</i> 1	= <i>b</i> 1	10.47	10.86
<i>d</i> 1	= <i>a</i> 1	= <i>a</i> 1	16.66	17.08

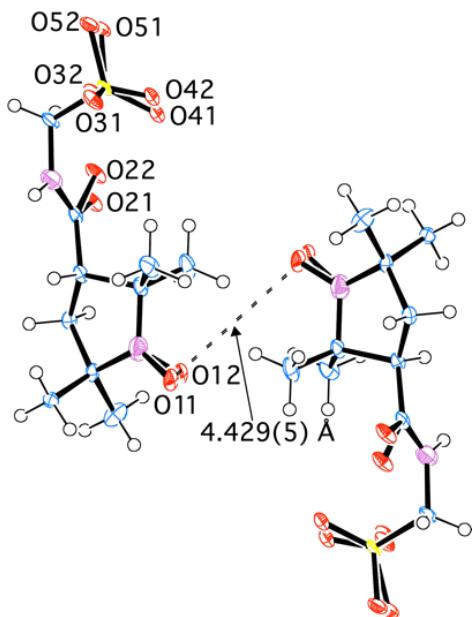


Fig. S3 Molecular structure of the *rac*-**1** anion dimer in *rac*-**2** at 28 K. Positional disorders are observed at only 28 K for the radical centre oxygen (O11 and O12), an oxygen of ketone (O21 and O22) and three oxygens of sulfo groups (O31-O51 and O32-052). The occupancies of the main atoms are 0.94(3), 0.948(10) and 0.904(17) for O11, O21 and O31-51, respectively. Those of the sub atoms are 0.06(3), 0.052(10) and 0.096(17) for O12, O22 and O32-52, respectively. The occupancy of O12 is so small therefore magnetic contribution is so small that here only the distance of O11···O11 is shown.

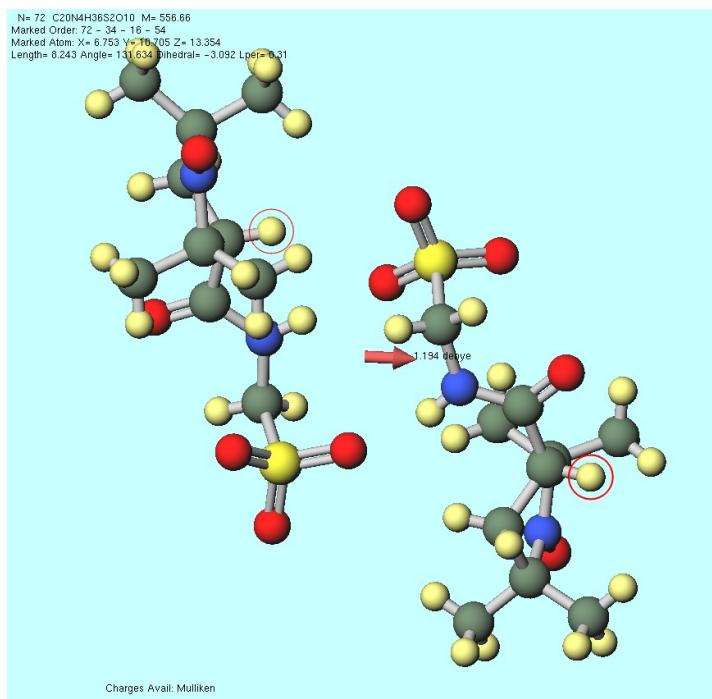


Fig. S4 Result of MOPAC2016 calculation using PM3 Hamiltonian. We calculate the dipole moment for the two anion molecules, which practically has point symmetry apart from the chiral centre C-H bonds. The resultant dipole moment of 1.194 debye is calculated. Therefore, one anion has 0.6 debye of a net dipole moment in the crystal.

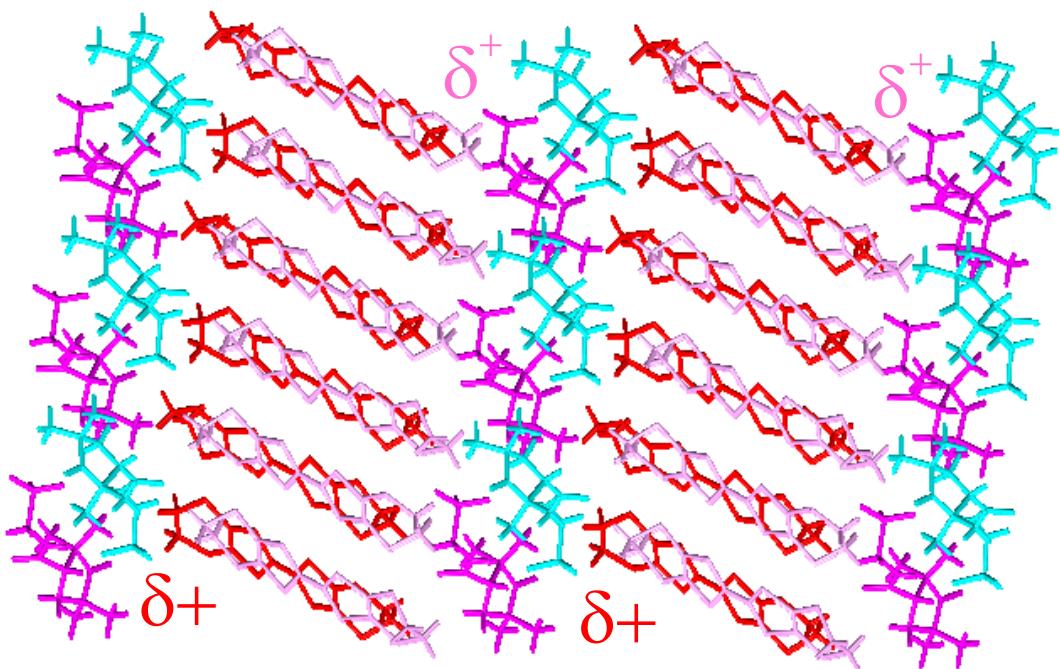


Fig. S5 Crystal structure of S-2. The red coloured BEDT-TTF molecules of B and C have larger positive charges than those of the pink coloured A and D molecule.

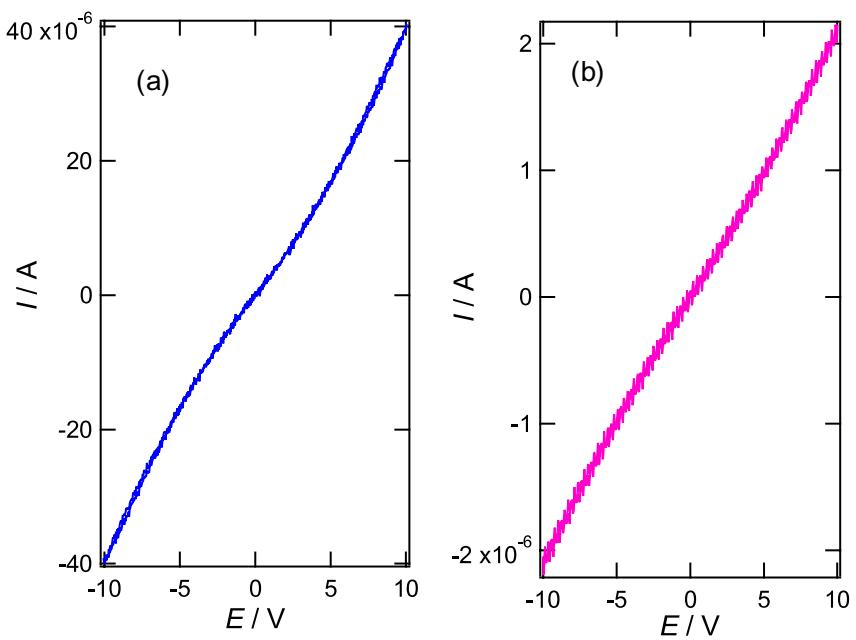


Figure S6. I-V curves of *rac*-**2** (a) and *S*-**2** (b) at 50 K at 1 Hz.