

Supporting Information

Preparation and characterization of sodium-ion conductive

Na_3BS_3 glass and glass-ceramic electrolytes

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Table S1 Atomic coordinates of the Na₃BS₃ crystal¹⁸.

Phase			Na ₃ BS ₃			
Crystal System			Monoclinic			
Space Group			<i>C</i> 2/ <i>c</i> (No. 15)			
Lattice Parameter, Volume, <i>Z</i>			$a = 11.853 \text{ \AA}, b = 6.664 \text{ \AA}, c = 8.406 \text{ \AA}, V = 585.274 \text{ \AA}^3$			
			$\alpha = 90, \beta = 118.18, \gamma = 90, Z = 4$			
Atoms	<i>x</i>	<i>y</i>	<i>z</i>	Site	Occupancy	
Na1	0.40274(6)	0.21349(9)	0.49570(8)	8f	1.0000	
Na2	0.25000	0.25000	0.00000	4c	1.0000	
S1	0.50000	0.14347(6)	0.25000	4e	1.0000	
S2	0.13550(3)	0.05790(5)	0.25321(4)	8f	1.0000	
B	0.50000	0.4217(3)	0.25000	4e	1.0000	

Table S2 Ionic conductivities at 25 °C (σ_{25}) and activation energy (E_a) for conduction of sodium-ion conducting glassy electrolytes.

Composition	$\sigma_{25} / \text{S cm}^{-1}$	$E_a / \text{kJ mol}^{-1}$	Ref.
Na_3PO_4	3.2×10^{-10}	133	20
Na_3BO_3	1.5×10^{-8}	61	13
Na_3PS_4	6.0×10^{-6}	47	5
$\text{Na}_4\text{P}_2\text{S}_7$	1.6×10^{-6}	–	21
NaPS_3	3.0×10^{-7}	–	21
$\text{Na}_4\text{B}_2\text{S}_5$	4.4×10^{-7}	48	14

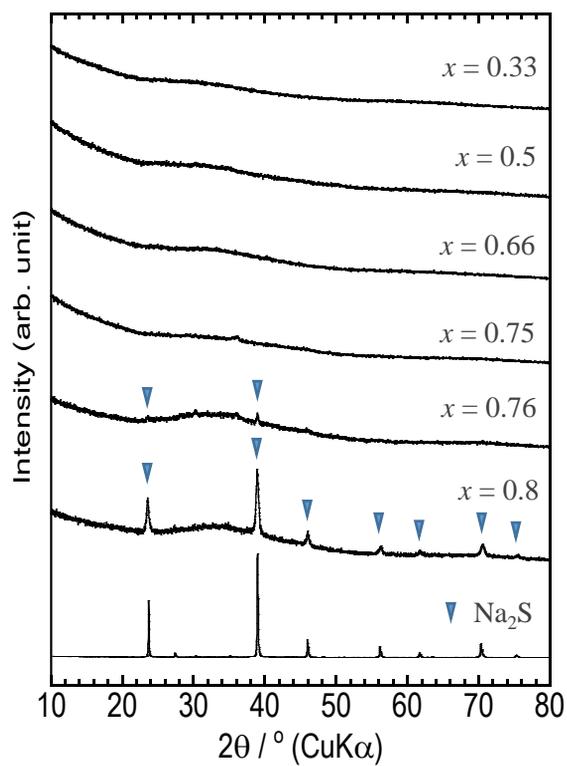


Fig. S1 XRD patterns of $x\text{Na}_2\text{S} \cdot (1-x)\text{B}_2\text{S}_3$ glass electrolytes synthesized via a mechanochemical process from the crystal corresponding to each composition.

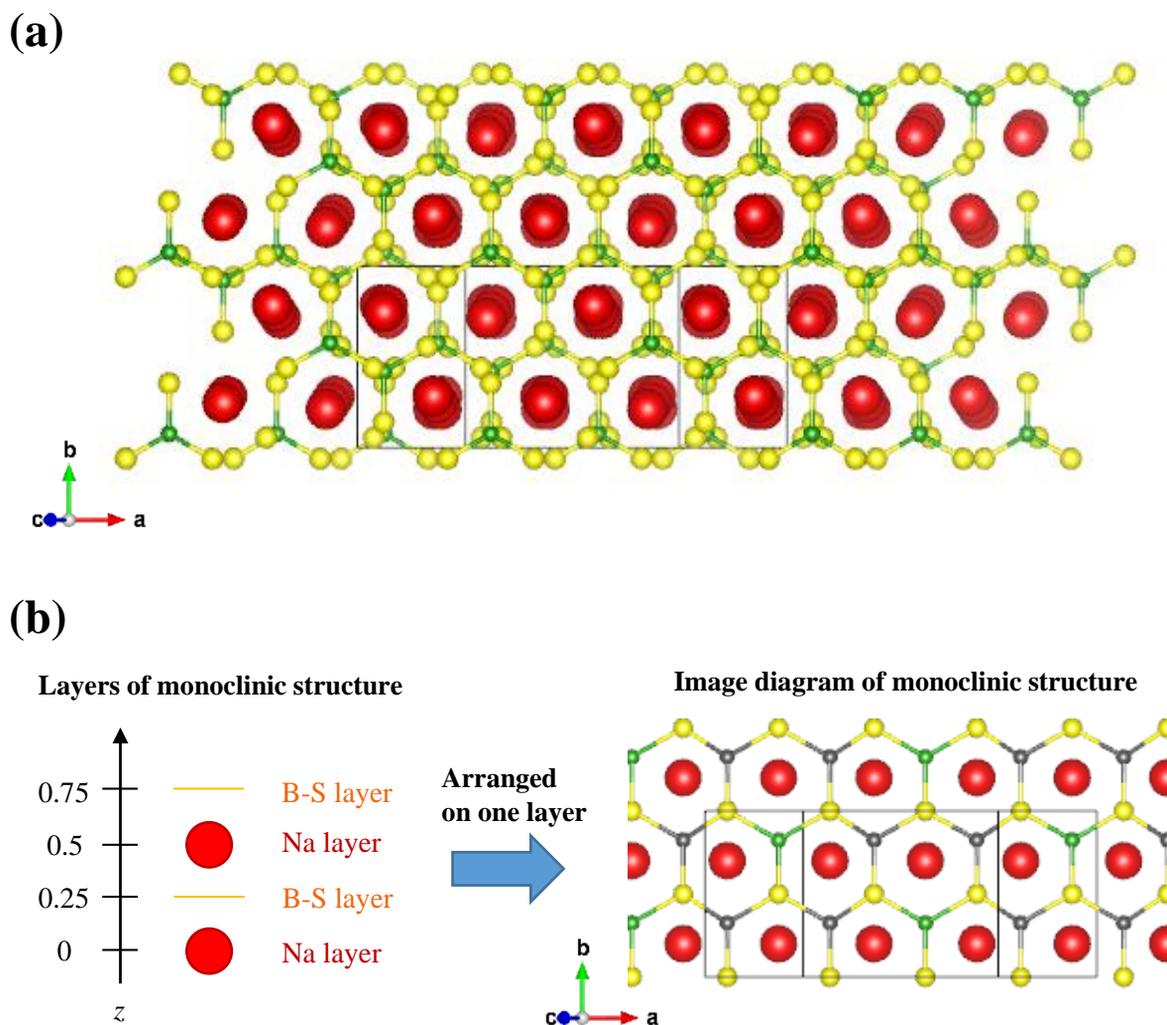


Fig. S2 (a) Schematic diagram of the crystal structure of monoclinic Na_3BS_3 .¹⁸ The red, green, and yellow spheres are Na, B, and S, respectively. (b) Image diagram of the monoclinic structure. The Na layers and B–S layers located at different z -positions are shown on the left. The right shows the arrangement of atoms in one layer. The B sites of the B–S layers are 1/3 occupied. Three-coordinated unoccupied B sites are shown as gray spheres.

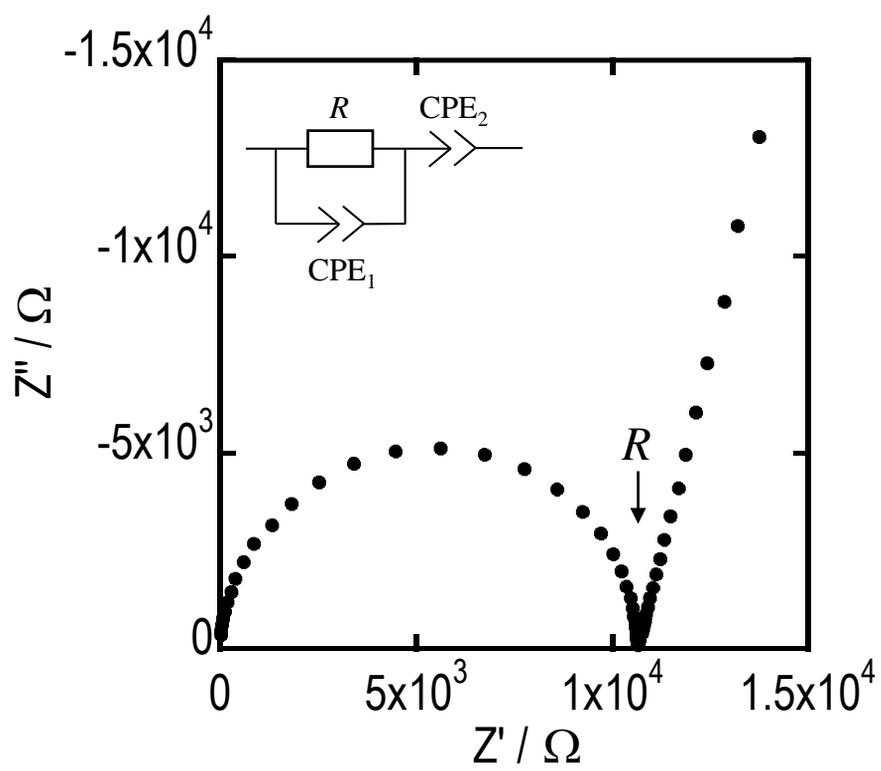


Fig. S3 Nyquist plot at $30\text{ }^\circ\text{C}$ of Na_3BS_3 glass.

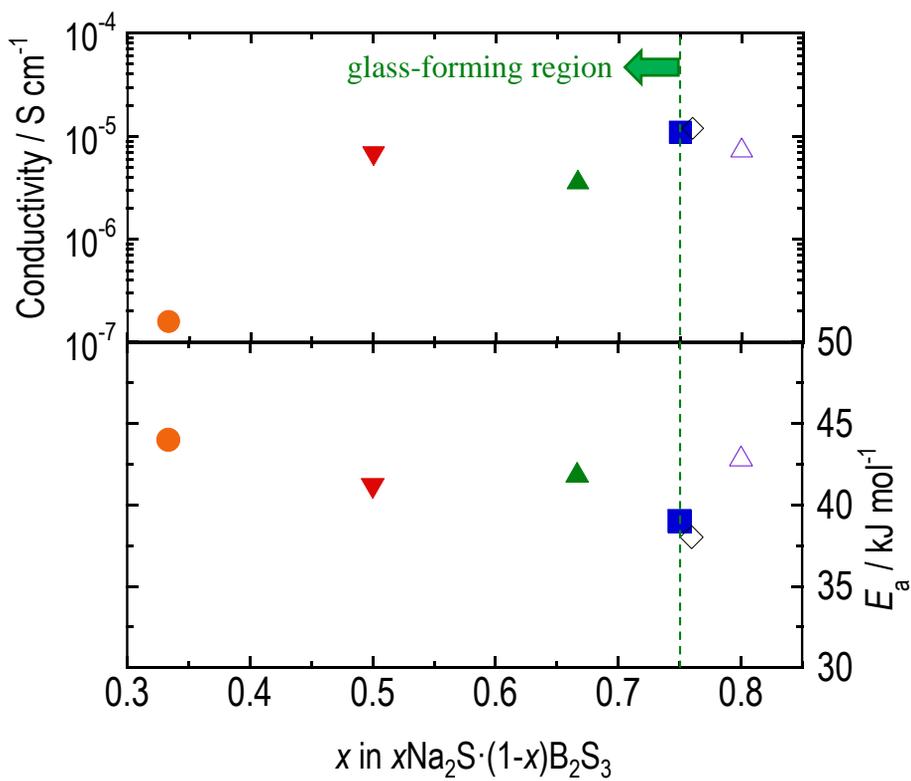
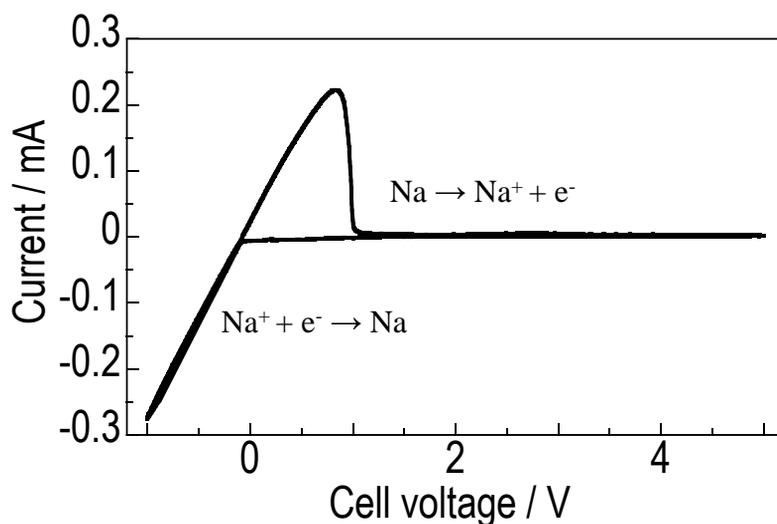


Fig. S4 Room-temperature ionic conductivities and activation energies (E_a) for conduction of $x\text{Na}_2\text{S}\cdot(1-x)\text{B}_2\text{S}_3$ glass electrolytes.

(a)



(b)

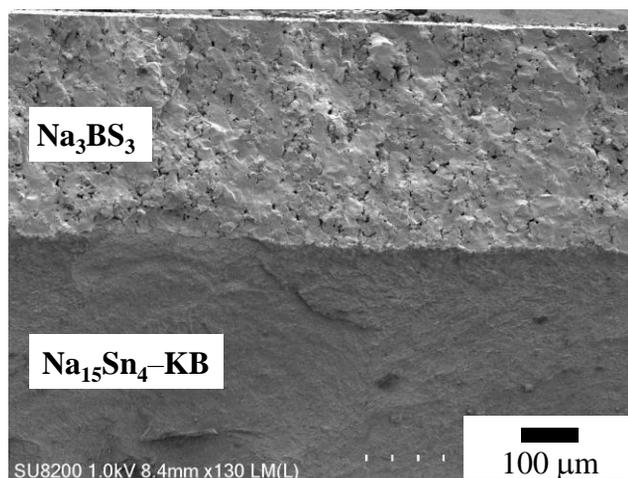


Fig. S5 (a) Cyclic voltammogram of the Na_3BS_3 glass. Stainless-steel and $\text{Na}_{15}\text{Sn}_4\text{-KB}$ was used as working and counter electrodes, respectively. The potential sweep was performed with a scanning rate of 5.0 mV min^{-1} at 60°C . (b) Cross-sectional SEM image of the cell after cyclic voltammetry. The working electrode of stainless steel was removed from the cell and two layers of $\text{Na}_3\text{BS}_3 / \text{Na}_{15}\text{Sn}_4\text{-KB}$ are observed.