Supporting Information:

# Exploring the Redox Decomposition of Ethylene Carbonate–Propylene Carbonate in Li-Ion Batteries

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**Marcus electron transfer theory**<sup>1</sup>Marcus' theory of electron transfer describes the rates of electron transfer between weakly coupled donor and acceptor states. It provides a parabolic model to calculate the activation energy of an electron-transfer reaction. The reactant and product energy surfaces were simplified representation of two parabolas. Then, the reorganization energy and change in Gibbs free energy can be obtained for the activation energy. The activation energy can be substituted into the Arrhenius or Eyring equation to calculate a rate constant for an electron-transfer reaction.

#### Adiabatic (E<sub>ad</sub>) and vertical (E<sub>vert</sub>) oxidation potentials

To calculate electrochemical stability of the representative complexes versus Li/Li<sup>+</sup>, the energy cycle shown in Fig. S1 is used. The absolute oxidation potential and vertical oxidation potential of a complex M relative to an electron at rest in vacuum is given by equations (1)-(3).<sup>2-4</sup>

$$E_{ad}(M) = \left[\Delta G_e + \Delta G_s^{\circ}(M^+) - \Delta G_s^{\circ}(M)\right]/F - 1.4$$
(1)

$$E_{vert}(M) = \left[\Delta G_e + \Delta G_s^{\circ}(M^+)_v - \Delta G_s^{\circ}(M)_v\right]/F - 1.4$$
(2)

$$\lambda = E_{vert} - E_{ad} \qquad (3)$$

Where  $E_{ad}$  and  $E_{vert}$  are the adiabatic oxidation and the neutral vertical oxidation respectively;  $\lambda$  is reorganization energy;  $\Delta G_e$  is the free energies of the electron attachment;  $\Delta G(M^+)$  and  $\Delta G(M)$  are the free energies of the oxidized solvent-phase, and neutral solvent-phase complexes; and F is the Faraday constant (F=23.061 Kcal mol<sup>-1</sup>V<sup>-1</sup>); Subscript of "s" and "v" denote the adiabatic oxidation and the vertical oxidation respectively. In the former, the optimized or relaxed geometry is required during calculation, while the complex geometry does not change during electron transfer in the latter. The Li /Li<sup>+</sup> oxidation potential is 1.4 V In this work, we used a value of 1.4 V for to convert from the absolute scale to Li<sup>+</sup>/Li.



**Fig. S1** Free-energy cycle for the redox reaction ( $M \rightarrow M^+ + e^-$ ), where M (g) denotes molecule M in gas-phase, M<sub>(S)</sub> denotes the solvated molecule, and IP denotes ionization potential.

Table S1 The dielectric constants and dipole moments of EC, PC anf FEC.

|                                     | EC                                    | PC    | FEC               |
|-------------------------------------|---------------------------------------|-------|-------------------|
| Dielectric Constant <sup>5, 6</sup> | 89                                    | 65    | 78.4              |
| Dipole Moment (Debye)               | 5.64 <sup>7</sup> , 4.81 <sup>5</sup> | 5.365 | 4.97 <sup>7</sup> |



**Reaction Coordinate** 

Fig. S2 Calculated free energy profile  $\Delta G$  of EC (left) and PC (right)

decomposition.



Fig. S3 Calculated free energy profile  $\Delta G$  of FEC decomposition.



Fig. S4 3D plots of the HOMO and LUMO orbitals of Product-F1.



Fig. S5 Ring formation of Product-P1.



Fig. S6 3D plots of the HOMO and LUMO orbitals of the lithium alkylcarbonate

complexes (dimer Product-E1 complex).



**Fig. S7** The reaction between FEC and LiCO<sub>3</sub><sup>-</sup> anion to produce inorganic LiF and less-reduced organic intermidate.



Fig. S8 Calculated possible pathways of *trans*-2,3-butylene carbonate (*t*-BC) and *cis*-2,3-butylene carbonate (*c*-BC) decomposition assisted by Li ions. The hydration energy of the solvated electron in water was -1.63 eV.



Fig. S9 Geometrical structure of dimerized trans-2,3-butylene carbonate (trans-BC).



Fig. S10 Geometrical structure of dimerized cis-2,3-butylene carbonate (cis-BC).

## TS-E1

imaginary frequency :-734.48 cm<sup>-1</sup>

| С  | 1.64927400  | -0.52113700 | 0.00029100  |
|----|-------------|-------------|-------------|
| С  | 1.44783700  | 0.96440300  | -0.00026000 |
| С  | -0.65610400 | -0.24789600 | -0.00001800 |
| Н  | 2.15581100  | -0.89262800 | -0.89350700 |
| Н  | 2.15493500  | -0.89202300 | 0.89483600  |
| Н  | 1.69012800  | 1.50379000  | 0.90992000  |
| Н  | 1.68964900  | 1.50302300  | -0.91102600 |
| 0  | -0.26495000 | 1.01207600  | 0.00016600  |
| 0  | 0.31422800  | -1.15860800 | -0.00021100 |
| 0  | -1.85174900 | -0.57144200 | -0.00004600 |
| Li | -2.63893500 | 1.11646900  | 0.00014300  |

## Int-E2

| С | 1.55273000  | -0.27237700 | -0.51160800 |
|---|-------------|-------------|-------------|
| С | 2.18162500  | 0.51004900  | 0.58283000  |
| С | -0.78770700 | -0.14575900 | -0.02082500 |
| Н | 1.32849400  | 0.35696400  | -1.37572300 |
| Н | 2.17850800  | -1.11339800 | -0.81983200 |
| Н | 2.82067400  | 0.02397400  | 1.31317900  |
| Н | 1.86677200  | 1.53052500  | 0.76910000  |

| 0  | -0.73768200 | 1.08575900  | -0.35438600 |
|----|-------------|-------------|-------------|
| 0  | 0.31126700  | -0.92256800 | -0.08401300 |
| 0  | -1.85617900 | -0.70312900 | 0.37615300  |
| Li | -2.53786300 | 0.98998800  | 0.10295400  |

## TS-E2

imaginary frequency :-163.09 cm<sup>-1</sup>

| С  | -1.99262200 | -0.11045500 | 0.62681300  |
|----|-------------|-------------|-------------|
| С  | -2.37993500 | 0.36610200  | -0.58179200 |
| С  | 0.90840700  | -0.15030600 | -0.01137600 |
| Н  | -1.44297800 | 0.51535800  | 1.31917700  |
| Н  | -2.34364900 | -1.06919500 | 0.99424900  |
| Н  | -2.98267500 | -0.22806500 | -1.26294700 |
| Н  | -2.04152900 | 1.33648100  | -0.93111600 |
| 0  | 0.74149500  | 1.10064800  | 0.19202900  |
| 0  | -0.08473500 | -1.01226400 | -0.02982500 |
| 0  | 2.07216800  | -0.65376200 | -0.20462200 |
| Li | 2.58810500  | 1.11213500  | 0.00603600  |

## TS-P1

imaginary frequency :-705.67 cm<sup>-1</sup>

| С  | 1.29854800  | -0.10690700 | 0.46323600  |
|----|-------------|-------------|-------------|
| 0  | -0.24085800 | -0.88529200 | 0.09389100  |
| С  | -1.08556800 | 0.10706000  | -0.04708900 |
| С  | 0.91356700  | 1.26521600  | -0.00995400 |
| Н  | 1.31899100  | -0.22944000 | 1.54458500  |
| Н  | 1.28913700  | 1.47343300  | -1.01726500 |
| Н  | 1.21995700  | 2.06785100  | 0.66682600  |
| 0  | -0.54612600 | 1.33317100  | -0.06223000 |
| 0  | -2.31368500 | -0.04004000 | -0.12711600 |
| С  | 2.33879700  | -0.88597500 | -0.28426000 |
| Н  | 2.17818800  | -0.82566800 | -1.36570400 |
| Н  | 3.34679400  | -0.50043200 | -0.06882500 |
| Н  | 2.32202600  | -1.93985100 | 0.00928000  |
| Li | -2.55393600 | -1.86165600 | 0.08771500  |

## Int-P2

| С | 1.22547300  | -0.53142700 | 0.22315400  |
|---|-------------|-------------|-------------|
| 0 | -0.22315800 | -0.76823000 | 0.17950000  |
| С | -1.14751000 | 0.37552900  | -0.10111800 |

| С  | 1.58365500  | 0.48753700  | 1.24733100  |
|----|-------------|-------------|-------------|
| Н  | 1.59954700  | -1.51230700 | 0.54791900  |
| Н  | 1.83150500  | 1.50254800  | 0.96145700  |
| Н  | 1.41495500  | 0.27366000  | 2.29879600  |
| 0  | -0.67017700 | 1.46686000  | -0.31163800 |
| 0  | -2.32591900 | -0.09201200 | -0.05243100 |
| С  | 1.76146200  | -0.22850000 | -1.17365000 |
| Н  | 1.37907900  | 0.73123400  | -1.52734700 |
| Н  | 2.85653000  | -0.18988400 | -1.15134800 |
| Н  | 1.45981900  | -1.01184000 | -1.87764600 |
| Li | -1.77529400 | -1.75506500 | 0.35013800  |

## TS-P2

imaginary frequency :-236.40 cm<sup>-1</sup>

| С | -1.70887400 | 0.52389600  | -0.19995600 |
|---|-------------|-------------|-------------|
| 0 | 0.44517200  | 0.69049400  | -0.41494000 |
| С | 1.22177100  | -0.34085100 | 0.11665800  |
| С | -1.65997000 | 0.70369800  | 1.15524500  |
| Н | -1.84133600 | 1.40642600  | -0.82325300 |
| Н | -1.54066500 | -0.14156600 | 1.82499500  |
| Н | -1.64839100 | 1.69857900  | 1.59084700  |
| 0 | 0.75391600  | -1.30761600 | 0.68719500  |

| 0  | 2.45463500  | -0.02063500 | -0.13570400 |
|----|-------------|-------------|-------------|
| С  | -1.91399900 | -0.80111300 | -0.86955800 |
| Н  | -1.48176700 | -1.60646200 | -0.27307200 |
| Н  | -2.98986500 | -0.98207200 | -1.00505100 |
| Н  | -1.44940800 | -0.80953600 | -1.85981700 |
| Li | 2.02936100  | 1.67430100  | -0.58713200 |

### TS-P1-1

imaginary frequency :-749.69 cm<sup>-1</sup>

| С | -1.31371200 | -0.04355400 | -0.45130900 |
|---|-------------|-------------|-------------|
| 0 | -0.11698900 | -0.92612100 | -0.46867000 |
| С | 0.97309300  | -0.28341900 | -0.05493400 |
| С | -0.80786800 | 1.35548400  | -0.23959900 |
| Н | -1.74761000 | -0.17353500 | -1.44711100 |
| Н | -1.17576600 | 1.89385600  | 0.62840500  |
| Н | -0.66216700 | 1.97482900  | -1.11942400 |
| 0 | 0.81491400  | 0.99428700  | 0.22863500  |
| 0 | 2.07053200  | -0.85291100 | 0.03132100  |
| С | -2.25815500 | -0.55532200 | 0.62613600  |
| Н | -1.81859500 | -0.42108400 | 1.62040400  |

| Н  | -3.20537900 | -0.00602000 | 0.58730100 |
|----|-------------|-------------|------------|
| Н  | -2.46750500 | -1.61869700 | 0.47943800 |
| Li | 3.12307400  | 0.59649100  | 0.54631100 |

Int-P3

| С  | 1.76527800  | -0.54020200 | -0.19917100 |
|----|-------------|-------------|-------------|
| 0  | -1.41411000 | -1.40782200 | -0.26839800 |
| С  | -1.47002200 | -0.20721700 | -0.13919500 |
| С  | 0.80713200  | -0.36270300 | 0.92402600  |
| Н  | 1.52854500  | -1.30386200 | -0.93562100 |
| Н  | 1.27039900  | 0.15077800  | 1.77411700  |
| Н  | 0.39553300  | -1.32012000 | 1.25052500  |
| 0  | -0.32708100 | 0.47581100  | 0.55222000  |
| 0  | -2.29976000 | 0.69836600  | -0.45451700 |
| С  | 2.83128300  | 0.45657500  | -0.50982100 |
| Н  | 3.20299000  | 0.95109400  | 0.39739100  |
| Н  | 3.68471000  | -0.00719400 | -1.01946900 |
| Н  | 2.46995800  | 1.25528600  | -1.18287200 |
| Li | -1.27551800 | 2.02148400  | 0.20881600  |

imaginary frequency :-156.59 cm<sup>-1</sup>

| С  | 1.87884800  | -0.41315900 | -0.25018400 |
|----|-------------|-------------|-------------|
| 0  | -1.33107200 | -1.38326900 | -0.44090900 |
| С  | -1.51541500 | -0.23022500 | -0.10414300 |
| С  | 1.29989700  | -0.66971800 | 0.94950400  |
| Н  | 1.55318000  | -1.00477300 | -1.10432100 |
| Н  | 1.59296500  | -0.12727000 | 1.84493700  |
| Н  | 0.61781500  | -1.50169000 | 1.06553800  |
| 0  | -0.59400500 | 0.53079300  | 0.58725400  |
| 0  | -2.54017000 | 0.55533200  | -0.28424000 |
| С  | 2.89199300  | 0.65580500  | -0.50723500 |
| Н  | 3.14812500  | 1.20535200  | 0.40463100  |
| Н  | 3.81581700  | 0.22934400  | -0.92266700 |
| Н  | 2.52523100  | 1.37406500  | -1.25448200 |
| Li | -1.62103200 | 2.04863800  | 0.18062600  |

## TS-F1

imaginary frequency :-743.30 cm<sup>-1</sup>

| С | 1.11690500  | 1.12106500  | 0.23108100  |
|---|-------------|-------------|-------------|
| С | 1.34276300  | -0.33755400 | 0.41813900  |
| С | -0.98720700 | 0.19530800  | -0.02203200 |

| Н  | 1.23691900  | 1.67596700  | 1.16502000  |
|----|-------------|-------------|-------------|
| Н  | 1.74905800  | 1.54358900  | -0.55360500 |
| Н  | 1.72733300  | -0.72745800 | 1.35445800  |
| 0  | -0.31900000 | -0.83541800 | 0.45353400  |
| 0  | -0.27027000 | 1.30816500  | -0.21433600 |
| 0  | -2.19388600 | 0.16082900  | -0.29039500 |
| F  | 1.86546900  | -0.97009400 | -0.64810600 |
| Li | -2.69068400 | -1.56759100 | 0.17117800  |

#### Int-F2

| С  | -0.98938400 | -0.23425700 | 0.92914500  |
|----|-------------|-------------|-------------|
| С  | -1.79235600 | 0.58763300  | 0.00348200  |
| С  | 1.21196900  | -0.15114400 | 0.00330900  |
| Н  | -0.60993900 | 0.38788700  | 1.74240000  |
| Н  | -1.59171400 | -1.05366700 | 1.32831200  |
| Н  | -1.47401800 | 1.54583500  | -0.39187300 |
| 0  | 1.21492500  | 1.09498100  | 0.28428200  |
| 0  | 0.13816800  | -0.91506000 | 0.29208400  |
| 0  | 2.19354900  | -0.73124100 | -0.54843700 |
| F  | -2.66776400 | -0.06619900 | -0.78594500 |
| Li | 2.91034600  | 0.97096600  | -0.48146200 |

## TS-F3

imaginary frequency :-151.19 cm<sup>-1</sup>

| С  | -1.39837700 | 0.00460900  | 1.07005500  |
|----|-------------|-------------|-------------|
| С  | -1.89800600 | 0.49250900  | -0.08363300 |
| С  | 1.34495400  | -0.16835100 | -0.01272000 |
| Н  | -0.71248800 | 0.61574200  | 1.63826200  |
| Н  | -1.79200400 | -0.90565600 | 1.50605000  |
| Н  | -1.55965800 | 1.39833700  | -0.57400100 |
| 0  | 1.20941100  | 1.09984600  | 0.11930300  |
| 0  | 0.36087500  | -1.01304500 | 0.20018100  |
| 0  | 2.46260100  | -0.69094800 | -0.35906200 |
| F  | -2.83220700 | -0.16464600 | -0.77592500 |
| Li | 2.99983100  | 1.07799100  | -0.37085600 |

## TS-F2

imaginary frequency :-740.97 cm<sup>-1</sup>

| С | 0.93534700  | 1.27581700  | 0.27066800  |
|---|-------------|-------------|-------------|
| С | 1.34290200  | -0.15904300 | 0.37661400  |
| С | -0.93261300 | -0.23130000 | 0.05901200  |
| Н | 0.84983600  | 1.82927200  | 1.20029900  |
| Н | 1.35323500  | 1.82920700  | -0.56249500 |

| Н  | 1.94077000  | -0.41962100 | 1.25349000  |
|----|-------------|-------------|-------------|
| 0  | 0.14234300  | -0.93930300 | 0.46956400  |
| 0  | -0.69920200 | 1.03267900  | -0.19696500 |
| 0  | -2.04475700 | -0.76153600 | -0.00913700 |
| F  | 2.01950000  | -0.55927300 | -0.74610000 |
| Li | -3.19341000 | 0.60901100  | -0.50728400 |

## Int-F3

| С  | -1.67751600 | 0.89326800  | -0.84741900 |
|----|-------------|-------------|-------------|
| С  | -1.36754600 | -0.39786100 | -0.18928800 |
| С  | 1.01876900  | -0.11946700 | -0.08480600 |
| Н  | -1.74263100 | 0.93105200  | -1.92896100 |
| Н  | -1.70555500 | 1.80488700  | -0.26707000 |
| Н  | -1.99245500 | -1.21289400 | -0.57291800 |
| 0  | -0.04816700 | -0.89844200 | -0.42119000 |
| 0  | 0.85397700  | 1.05315900  | 0.37338700  |
| 0  | 2.15698600  | -0.64350500 | -0.25596000 |
| F  | -1.58317800 | -0.30883300 | 1.16855000  |
| Li | 2.71487700  | 0.97037200  | 0.47039400  |

imaginary frequency :-287.77 cm<sup>-1</sup>

| С  | 1.90704300  | -0.13972900 | 1.11166300  |
|----|-------------|-------------|-------------|
| С  | 1.76117900  | -0.28884000 | -0.23336000 |
| С  | -1.13301500 | -0.18599200 | 0.03756600  |
| Н  | 2.11925100  | -1.01100400 | 1.71877900  |
| Н  | 1.68753700  | 0.80689900  | 1.58833700  |
| Н  | 2.03075900  | -1.18334200 | -0.78193100 |
| 0  | -0.17934200 | -1.07984700 | -0.15081700 |
| 0  | -0.90752400 | 0.98663900  | 0.49039200  |
| 0  | -2.31482800 | -0.57424900 | -0.27346100 |
| F  | 1.59444400  | 0.77020000  | -1.02246800 |
| Li | -2.72841200 | 1.16089200  | 0.21763200  |

#### REFERENCES

- 1. R. A. Marcus, *The Journal of Chemical Physics*, 1956, **24**, 966-978.
- 2. O. Borodin, *Current Opinion in Electrochemistry*, 2019, **13**, 86-93.
- 3. O. Borodin, W. Behl and T. R. Jow, *Journal of Physical Chemistry C*, 2013, **117**, 8661-8682.
- 4. O. Borodin, X. Ren, J. Vatamanu, A. von Wald Cresce, J. Knap and K. Xu, *Accounts of Chemical Research*, 2017, **50**, 2886-2894.
- 5. Y. Chernyak, *Journal of Chemical & Engineering Data*, 2006, **51**, 416-418.
- 6. K. Y. L. Y. Im, J. S. Bae, Y. T. An., Electrolyte solvent for improving safety of battery and lithium secondary battery comprising the same. WO2005024991A2005024991, 2005022004.
- T. Hou, G. Yang, N. N. Rajput, J. Self, S.-W. Park, J. Nanda and K. A. Persson, *Nano Energy*, 2019, 64, 103881.