

Supporting Information

Transition Metal Oxynitrides Catalysts for Electrochemical Reduction of Nitrogen to Ammonia

Damilola Ologunagba, and Shyam Kattel*
Department of Physics, Florida A&M University
Tallahassee, FL 32307 USA

*Corresponding author: shyam.kattel@famu.edu

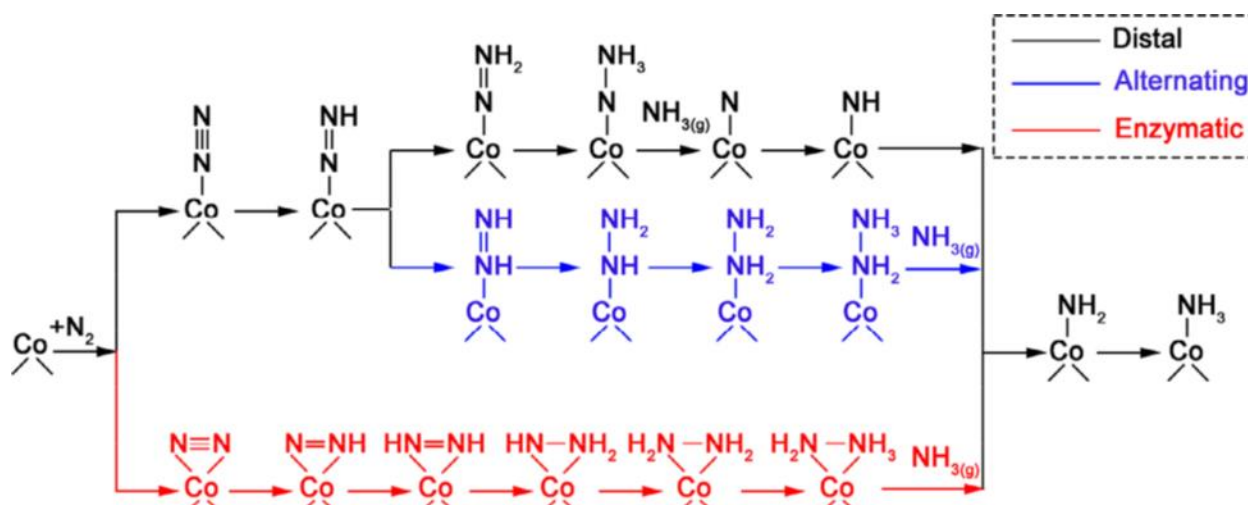


Figure S1. Three main pathways for ENRR: (i) distal, (ii) alternating, and (iii) enzymatic. (Figure adapted from ref.¹ with permission).

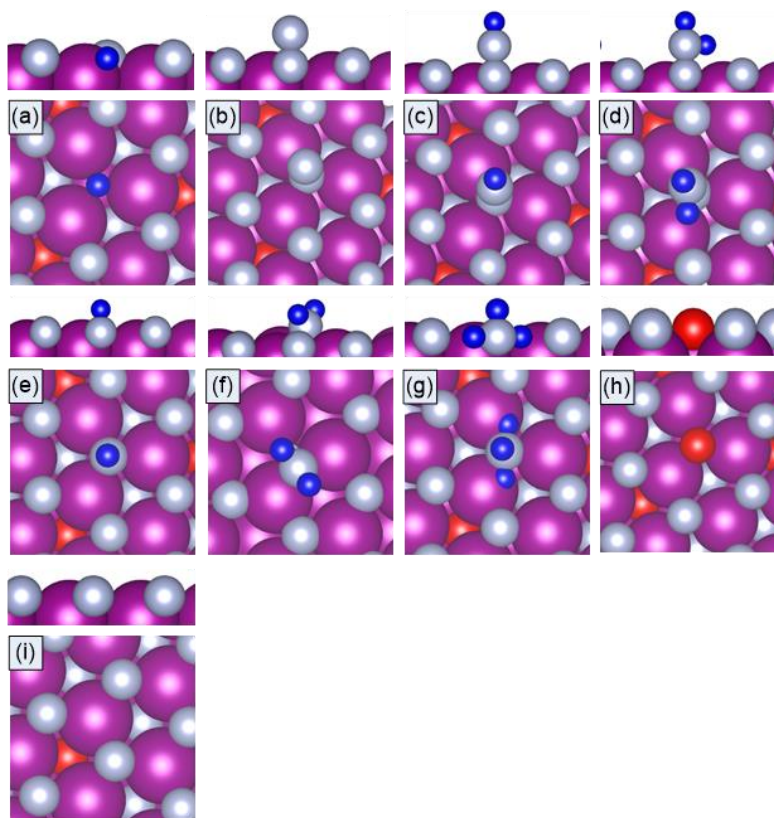
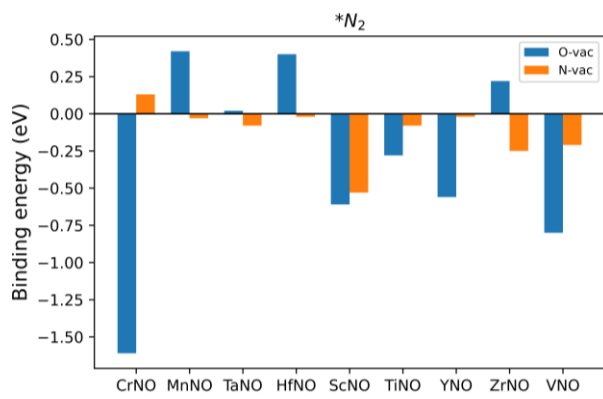


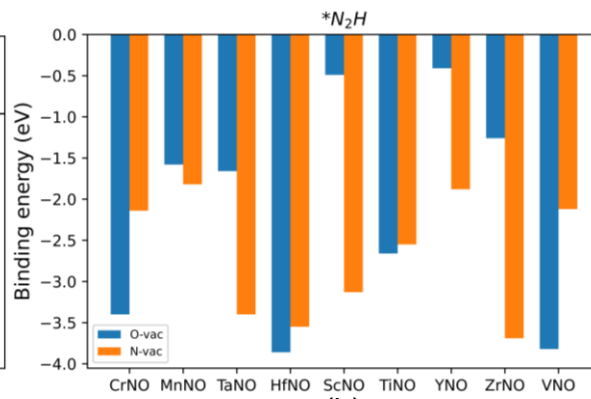
Figure S2. DFT optimized geometries of (a) *H, (b) *N₂, (c) *N₂H, (d) *N₂H₂, (e) *NH, (f) *NH₂, (g) *NH₃, (h) *O, and (i) *N adsorbed on O-vac of MnNO(111). Mn: purple, N: gray, O: red, H: blue.

Table S1. ZPE and TS values of the adsorbed species.

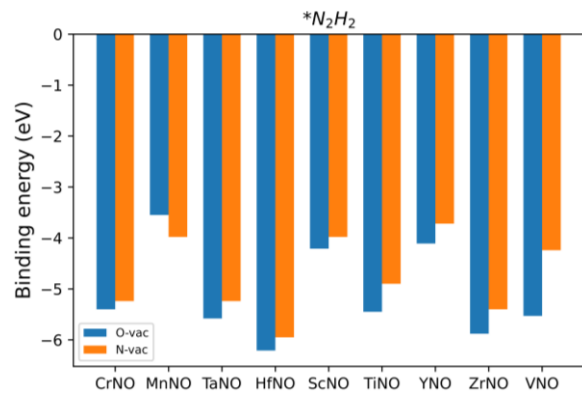
| species | ZPE/eV | TS/eV |
|--------------------------------|--------|-------|
| *N ₂ | 0.18 | 0.18 |
| *N ₂ H | 0.51 | 0.08 |
| *N ₂ H ₂ | 0.87 | 0.11 |
| *HNNH | 0.84 | 0.15 |
| *N | 0.10 | 0.02 |
| *H | 0.18 | 0.00 |
| *NH | 0.38 | 0.04 |
| *NH ₂ | 0.67 | 0.07 |
| *NH ₃ | 1.01 | 0.12 |
| H ₂ (g) | 0.27 | 0.41 |
| N ₂ (g) | 0.15 | 0.60 |
| NH ₃ (g) | 0.89 | 0.74 |
| H ₂ O(g) | 0.58 | 0.65 |



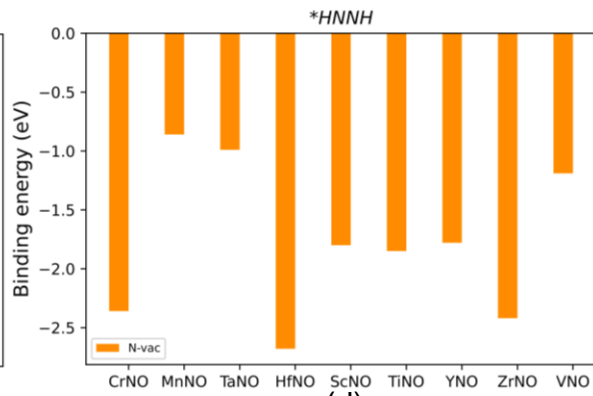
(a)



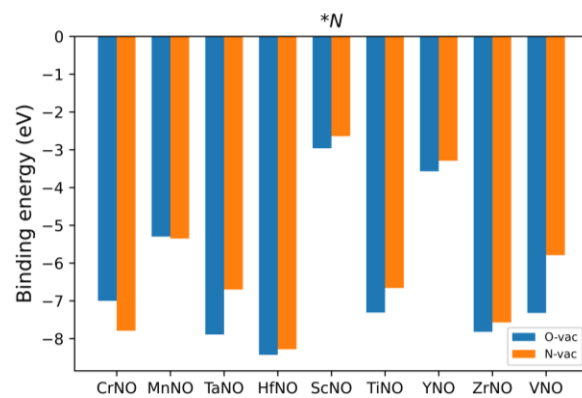
(b)



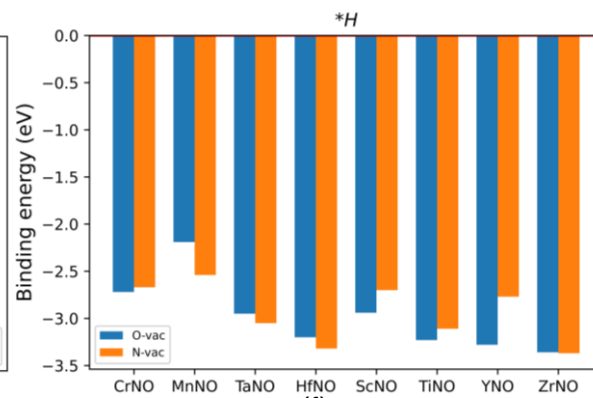
(c)



(d)



(e)



(f)

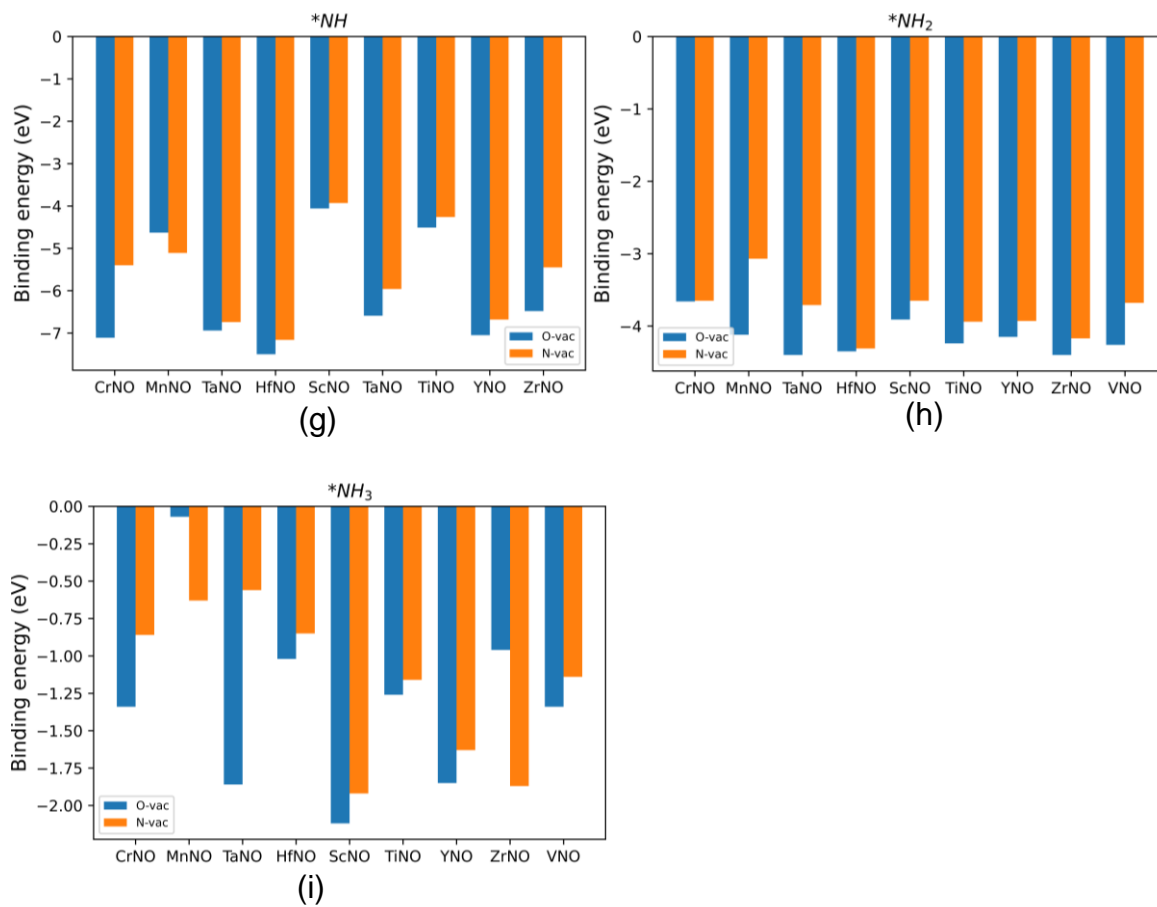


Figure S3. DFT calculated binding energies of reaction intermediates on N-vac and O-vac.

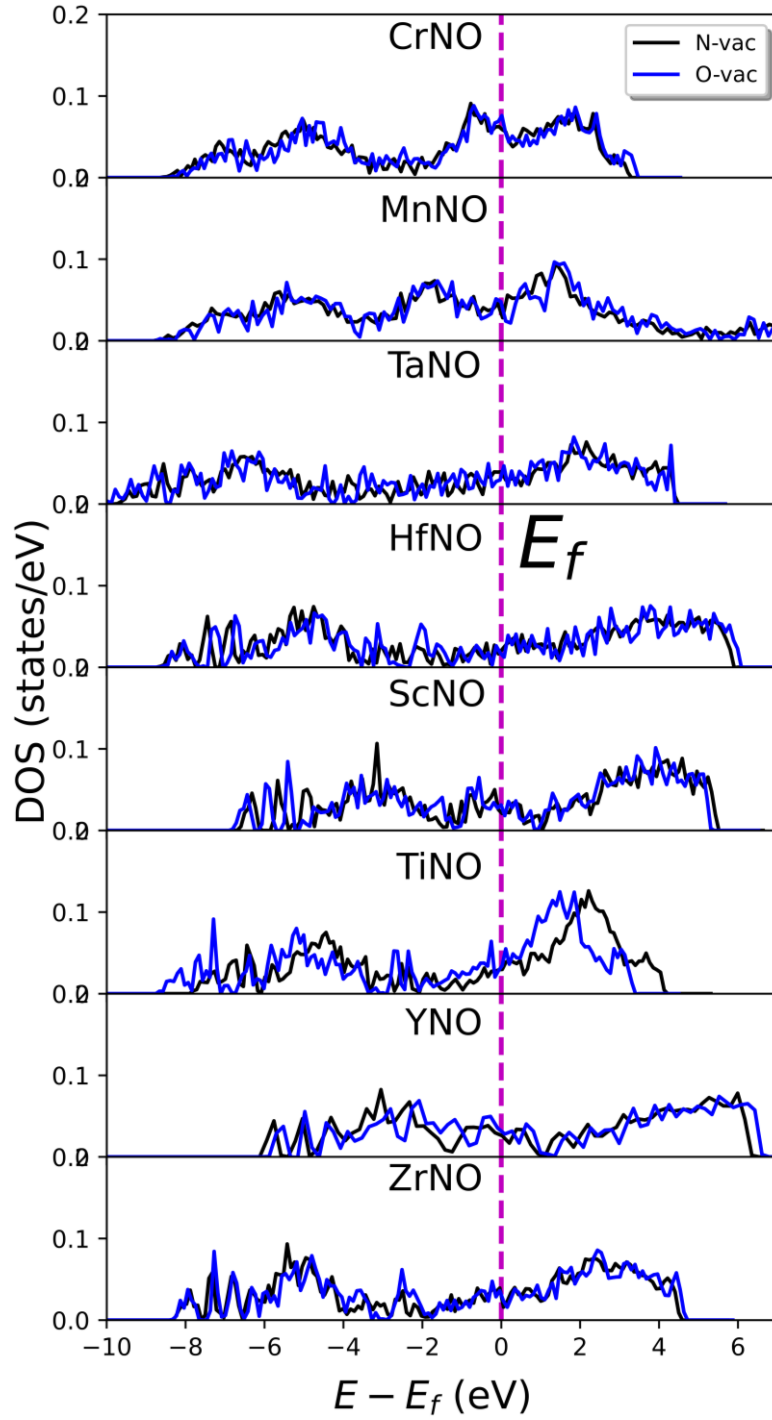


Figure S4. Density of state (DOS) plots for all the TMNO on N-vac and O-vac.

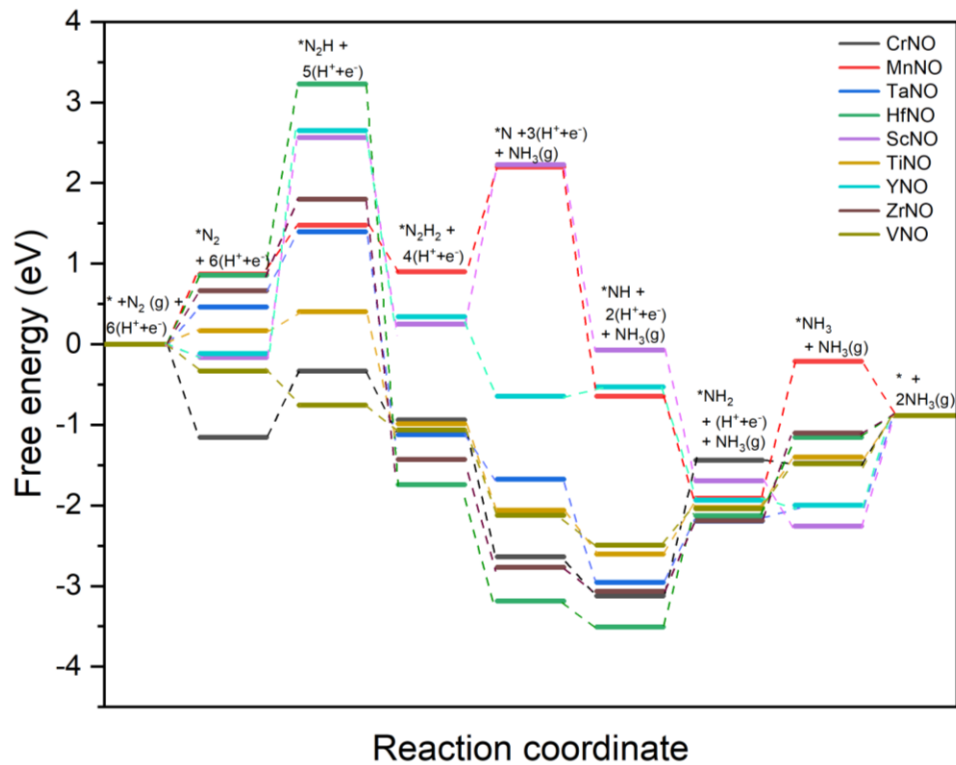


Figure S5. DFT calculated free energy diagrams at $U = 0$ V along the associative N_2H_2 pathway on O-vacancy.

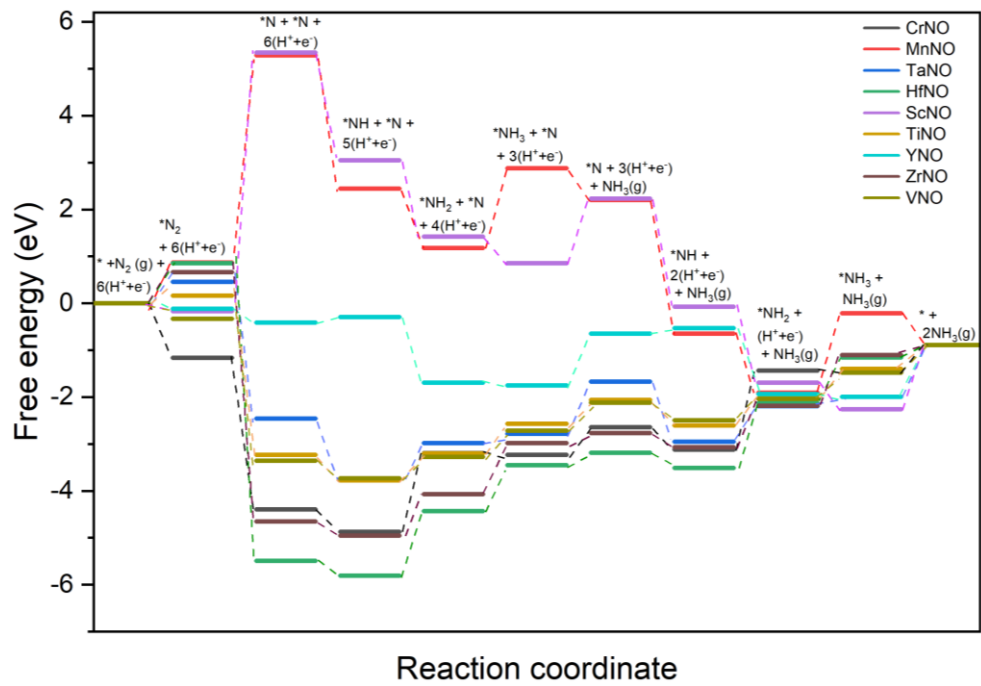


Figure S6. DFT calculated free energy diagrams of ENRR at $U = 0$ V along the dissociative pathway on O-vac.

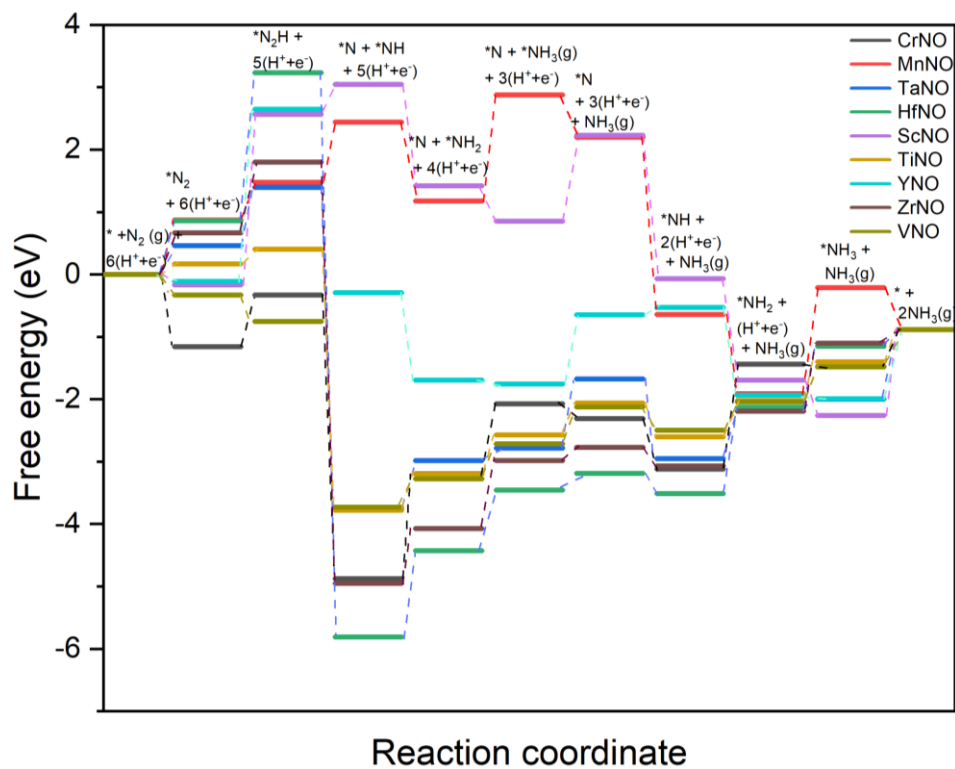


Figure S7. DFT calculated free energy diagrams of ENRR at $U = 0$ V along the associative N_2H pathway on O-vac.

REFERENCES

- 1 Z. Wei, Y. Feng and J. Ma, *J. Energy Chem.*, 2020, **48**, 322–327.